Fungal chitosan-copper as a rational and sustainable nanozyme with intrinsic laccase activity

for robust degradation of phenolic pollutants

Efrata Getachew Mekonnen^a, Kassie Nigus Shitaw^b, Bing-Joe Hwang^{b,c}, Yitayal Admassu

Workie, de Ebrahim M. Abda*a Menbere Leul Mekonnen*d,e

^a Biotechnology Department, Addis Ababa Science, and Technology University, Addis Ababa, P.O.Box 1647, Ethiopia.

^b Department of Chemical Engineering, National Taiwan University of Science and Technology, Taipei 106, Taiwan

^cNational Synchrotron Radiation Research Center, Hsinchu, Taiwan

^d Industrial Chemistry Department, Addis Ababa Science and Technology University, Addis Ababa, P.O.

Box 1647, Ethiopia.

^eNanotechnology Center of Excellence, Addis Ababa Science and Technology University, Addis Ababa,

P.O.Box 1647, Ethiopia.

*Corresponding authors:

Menbere Leul Mekonnen (<u>menbere.leul@aastu.edu.et</u>) Ebrahim M. Abda <u>(ebrahim.mama@aastu.edu.et</u>)

Electronic Supplementary Material

Chitosan extraction procedure

The method used for chitin and chitosan extraction in this work followed the procedures described in elsewhere.¹ Briefly, the mycelia recovered by filtration through Whatman filter paper was subjected to repeated washing with distilled water and dried at 65 °C to a constant weight. Finely powdered dry fungal mycelia was then suspended in a 1 mol/L sodium hydroxide (1: 30 w/v), autoclaved for 20 minutes at 121 °C, and then centrifuged (6000 g ,15 min.) to separate the alkali-insoluble fraction (AIF). The AIF was then washed with distilled water, and centrifuged to a neutral pH. The residues were further extracted using 2 % acetic acid (1: 40 w/v) at 95 °C for 8 h. The extracted slurry was centrifuged at (6000 g ,15 min.) and the insoluble acid fraction was discarded. The solution was centrifuged at (12000 g ,15 min.) and the pH of the supernatant was corrected to 10 with 2 mol/L sodium hydroxide. The precipitated chitosan was then washed with distilled water, 95% ethanol, and acetone, respectively, and dried at 60 °C to a constant weight.

The resulting chitosan was checked for its degree of deactylation and presence of important functional groups using Infrared absorption technique (FTIR Nicolet XX). First a 2 mg sample of chitosan was subjected to a KBr pellet (100 mg KBr) followed by scanning the absorbance from 400- 4000 cm⁻¹ Then the ratio of the absorbance at 1655 and 3450 cm⁻¹ (A1655/A3450.) after appropriate baseline subtraction was taken for calculation of the degree of deactylation (DDA) according to equation 1

$$DA = \frac{A_{1655}}{A_{3450}} \times 115$$
 Equation (S1)

In these equations 1655 and 3450 cm⁻¹ represent the absorbance of chitosan at 1655, 3450,cm⁻¹ Further the average molecular weight of the obtained chitosan was estimated from its intrinsic viscosity using rotating viscometer (NDJ -8s, Shanghai) The intrinsic viscosity was determined from the y-intercept of the plot of specific viscosity vs concentration of chitosan solution in 0.2 M Sodium acetate and 0.3 M acetic acid. The molecular weight of the chitosan was then estimated according to the Mark–Houwink–Sakurada equation

$[\eta] = KM^a$ Equation (S2)

Where, M is viscosity molecular weight, K = 0.076 mL/g and a = 0.76 in this chitosan-solvent system

Characterization of fungal chitosan

Fungal chitosan produced by the potent isolate was further characterized using FTIR and SEM. The chitosan that was dried in a powder form was investigated using Fourier transform infrared (FT-IR) spectroscopy at a 4 cm-1 resolution at wave numbers ranging from 4000 to 400 cm-1. To analyze the surface morphology, scanning electron microscope were used. The sample was processed by self-coting with gold.



Figure S1 A representative fungal isolates cultured on PDA plates. (A) AWK1; (B) AWK2 ;(C) AWK7; (D) AWK3 and (F) AWK6

| isolate | Potato dextrose broth | | | | Minimal salt medium | | | |
|---------|-----------------------|---------|---------------|-------|---------------------|---------|---------------|-------|
| Code | Fungal | biomass | Chitosan | yield | Fungal | biomass | Chitosan | yield |
| | (g/100mL) | | (mg/ g/100mL) | | (g/100mL) | | (mg/ g/100mL) | |
| AWK1 | 0.4 | | 0.0018 | | 0.734 | | 0.0065 | |
| AWK2 | 0.46 | | 0.0031 | | 0.561 | | 0.0121 | |
| AWK3 | 0.41 | | 0.0058 | | 0.654 | | 0.0124 | |
| AWK6 | 0.354 | | 0.0034 | | 0.449 | | 0.0078 | |
| AWK7 | 0.37 | | 0.0038 | | 0.654 | | 0.127 | |
| AWK8 | 0.37 | | 0.0008 | | | | | |
| AWK9 | 0.369 | | 0.0004 | | | | | |
| AWK12 | 0.467 | | 0.0014 | | | | | |
| AWK15 | 0.343 | | 0.004 | | | | | |

Table S1 chitosan yield by the isolated fungi in potato dextrose broth and minimal salt medium

Table S2: Taxonomic affiliation of the selected chitosan-producing fungi based on ITS rDNA sequences analysis.

| Isolate Code | Top-hit Taxon | Identity (%) | Taxonomy |
|--------------|-----------------------------------|--------------|---------------|
| AWK2 | Irpex sp. isolate FS16 | >99 | Basidiomycota |
| AWK1 | Schizophyllum commune strain C77P | >99 | Basidiomycota |
| AWK7 | Diaporthe sp. isolate F255 | >97 | Ascomycota |

| | | | | | | , | Spectrum 13 |
|-----|---|---|----|----|---------|---------|-------------|
| | | | | | Element | Weight% | 6 Atomic% |
| | | | | | CK | 6.52 | 16.20 |
| | | | | | NK | 0.72 | 1.52 |
| | | | | | OK | 27.74 | 51.74 |
| | | | | | Cu L | 65.03 | 30.54 |
| | | | | | Totals | 100.00 | |
| 3 | | | | | | | |
| | | | 79 | Þ | | | |
| 0 2 | 4 | 6 | 8 | 10 | 12 1 | 4 16 | 18 |

Figure S2: EDX spectrum of CsCu nanozymes



Figure S3: Activity of the CsCu nanozyme

[1] P. Pochanavanich, W. Suntornsuk, Fungal chitosan production and its characterization, Letters in Applied Microbiology 35(1) (2002) 17-21.