

Mixed valence copper oxide composites derived from metal-organic frameworks for efficient visible light fuel denitrification

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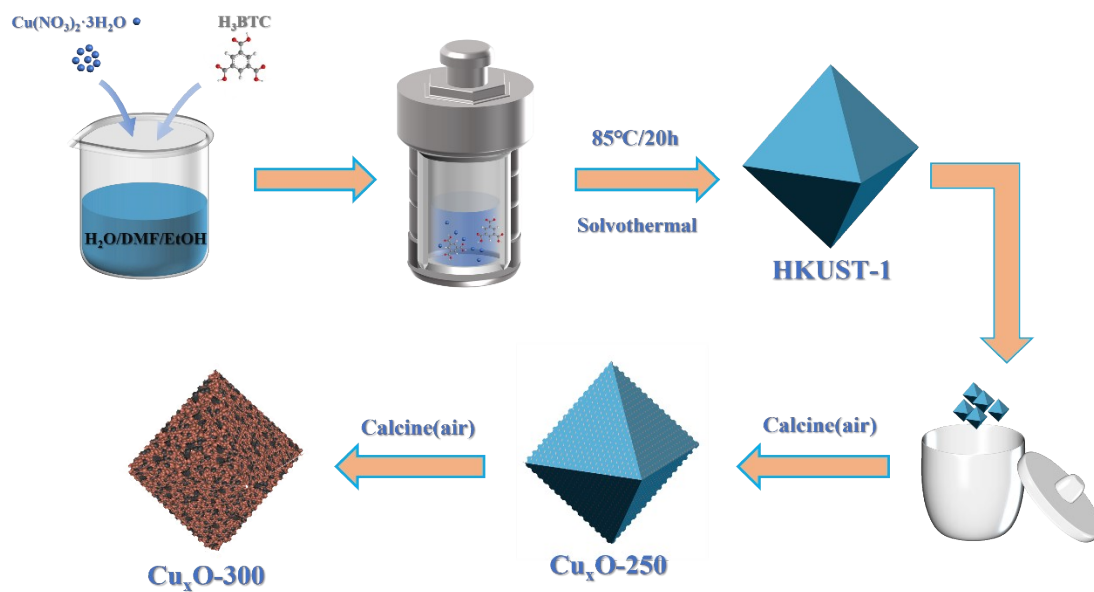


Fig. S1 Flow diagram for the fabrication of Cu_xO .

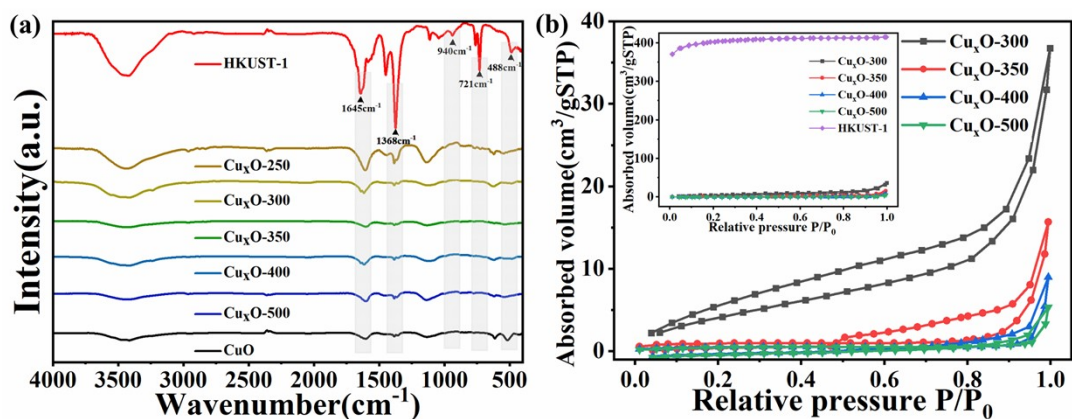


Fig. S2 (a) Fourier transform infrared reflectance spectra of samples and (b) N_2 adsorption-desorption isotherms of synthetic catalysts.

Table S1 BET specific surface area and pore volume of the synthesized catalyst.

Sample	BET surface area	Pore volume
	(m^2/g)	(cm^3/g)
$\text{Cu}_x\text{O-300}$	17.68	0.154
$\text{Cu}_x\text{O-350}$	3.45	0.024
$\text{Cu}_x\text{O-400}$	1.91	0.013
$\text{Cu}_x\text{O-500}$	1.71	0.008
HKUST-1	1525.1	0.642

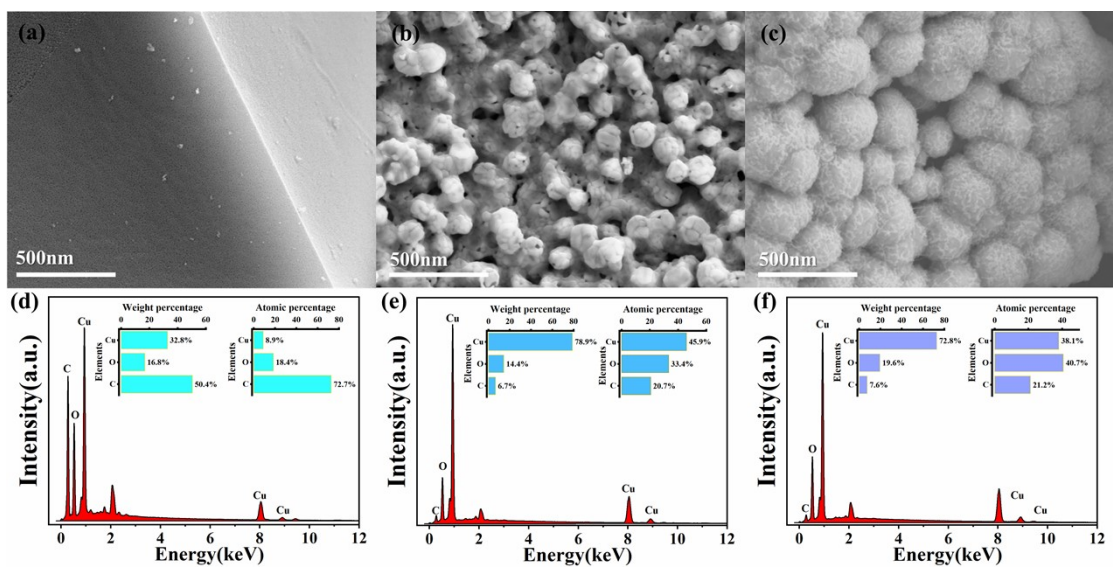


Fig. S3 (a-c) SEM image of Cu_xO-250, Cu_xO-300 and Cu_xO-500, (d-f) EDS of Cu_xO-250, Cu_xO-300 and Cu_xO-500

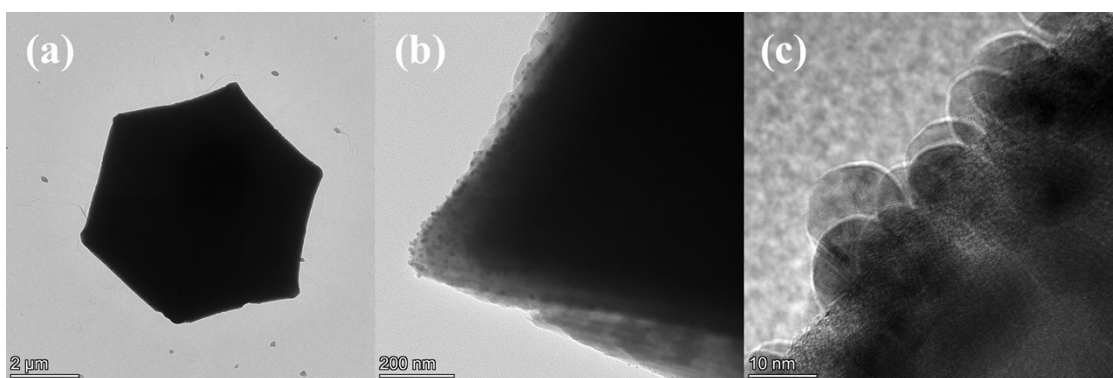


Fig. S4 (a-c) TEM image of HKUST-1.

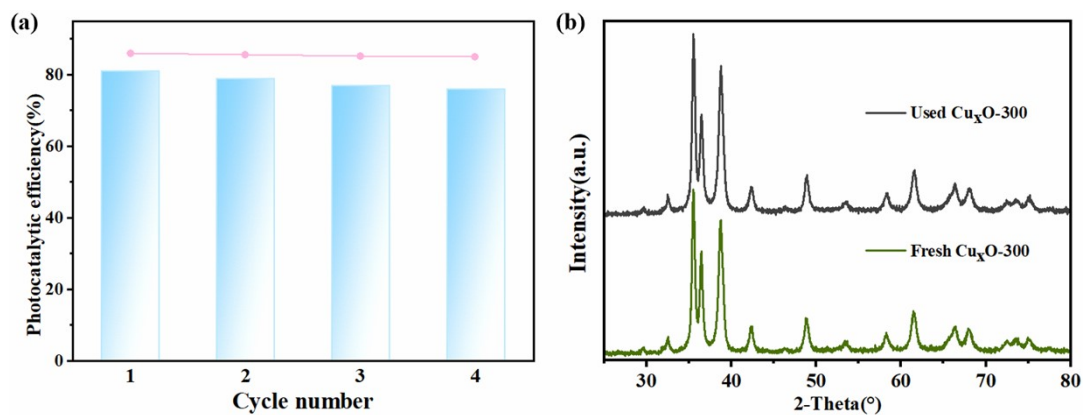


Fig. S5 (a) The cyclic experimental results of Cu_xO-300 and (b) The XRD patterns of Cu_xO-300 after four reaction cycles.

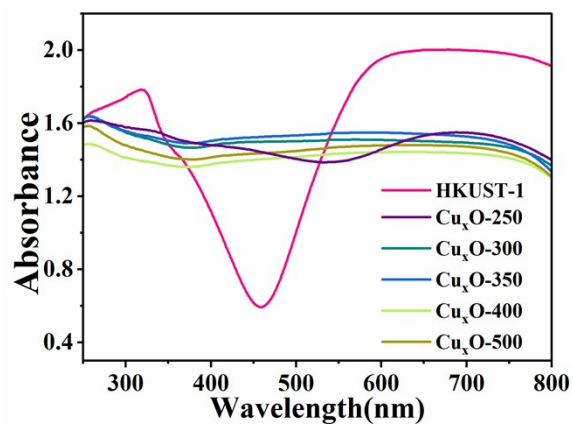


Fig. S6 UV-vis diffuse reflectance spectra of HKUST-1 and $\text{Cu}_x\text{O-T}$.

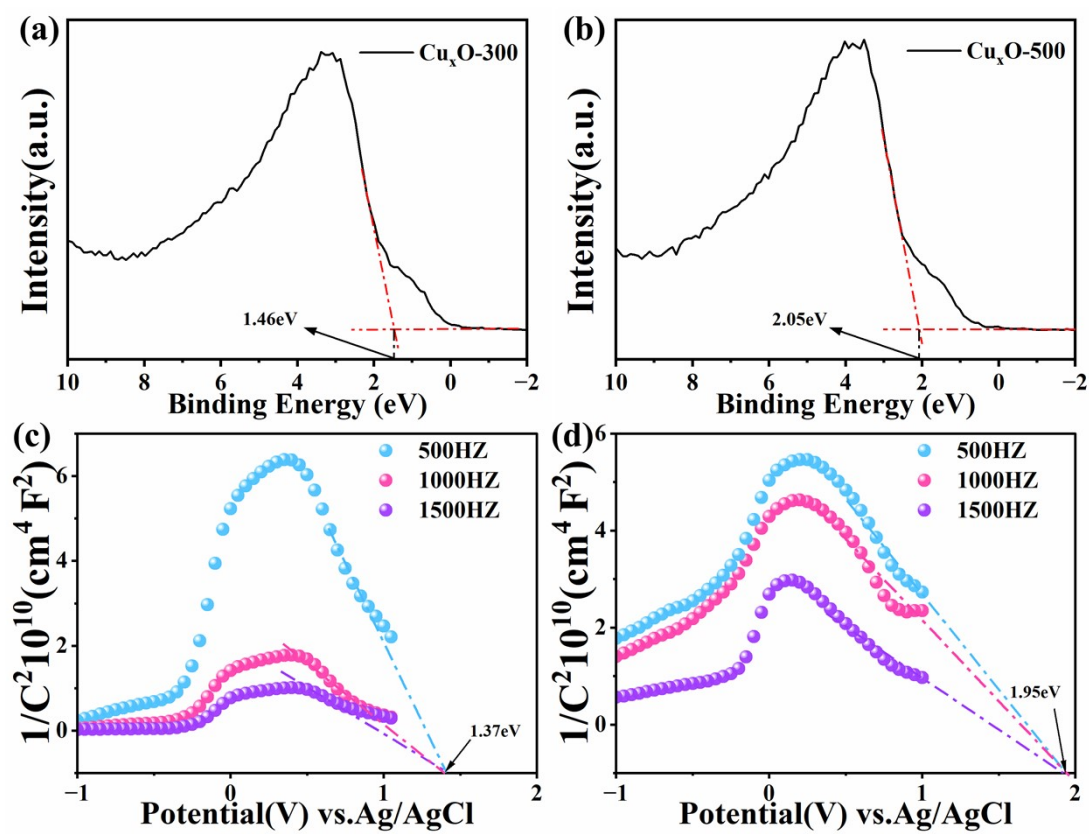


Fig. S7 (a, b) Valence band curves of $\text{Cu}_x\text{O-300}$ and $\text{Cu}_x\text{O-500}$; (c, d) Mott-Schottky plots of $\text{Cu}_x\text{O-300}$ and $\text{Cu}_x\text{O-500}$.

Table S2 Comparison between the photocatalytic activity of Cu_xO-300 and that of other reported catalysts for pyridine denitrogenation.

Photocatalysts	C pyridine (μg/g)	C cat. (mg/mL)	Light Source	Denitrogenation efficiency (%)	Ref.
Cu _x O	100	1.0	300W(λ >420nm)	4.0h,81%	This work
Pd/ZnIn ₂ S ₄	100	1.0	300W(λ >420nm)	4.0h,80%	[R1]
TiO ₂ @MIL-101(Cr)	100	1.0	300W(λ >420nm)	4.0h,70%	[R2]
Bi ₂ MoO ₆ /CdS	100	1.0	300W(λ >420nm)	4.0h,81%	[R3]
CoCu-ZIF	100	1.0	300W(λ >420nm)	4.0h,80%	[R4]

References

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