

Development of a surrogate artificial neural network for microkinetic modeling: case study with the methanol synthesis

Bruno Lacerda de Oliveira Campos^a, Esly Ferreira da Costa Junior^{b}, Karla Herrera Delgado^{a*},
Andréa Oliveira Souza da Costa^b, Stephan Pitter^a, and Jörg Sauer^a*

a. Institute for Catalysis Research and Technology (IKFT), Karlsruhe Institute of Technology (KIT), Hermann-von-Helmholtz-Platz 1, 76344 Eggenstein-Leopoldshafen, Germany. karla.herrera@kit.edu

b. Chemical Engineering Department, Universidade Federal de Minas Gerais (UFMG), Av. Presidente Antônio Carlos 6627, Pampulha 31270-901, Belo Horizonte (MG), Brazil. esly@deq.ufmg.br

Electronic Supplementary Information (ESI)

S1. Microkinetic model

Table S1. Three-site field extended reaction mechanism for the methanol synthesis and the water-gas shift reaction over Cu (211) and Cu/Zn (211), thermodynamically consistent. (a): Cu (211), (b): Cu/Zn (211), (c): special Cu (211) site for hydrogen and water adsorption. Reproduced with permission from Campos et al.¹ Copyright 2021 Royal Society of Chemistry.

| No. | Reaction | Forward reaction | | | Reverse reaction | | |
|---|---|---|------------------------------------|------------------|---|------------------------------------|------------------|
| | | $\Delta S^{\ddagger,+} \cdot 10^3$ [kJ·(mol·K) ⁻¹] | E_A^+ [kJ·mol ⁻¹] | β^+ [-] | $\Delta S^{\ddagger,-} \cdot 10^3$ [kJ·(mol·K) ⁻¹] | E_A^- [kJ·mol ⁻¹] | β^- [-] |
| R1 | $H_{2(g)} + 2 \cdot (c) \rightleftharpoons 2 \cdot H_{(c)}$ | -119.24 | 69.57 | 0.000 | 1.72 | 93.01 | 0.000 |
| R2 | $CO_{(g)} + (a) \rightleftharpoons CO_{(a)}$ | -158.23 | 0.00 | 0.000 | 0.00 | 58.37 | 0.000 |
| R3 | $CO_{(g)} + (b) \rightleftharpoons CO_{(b)}$ | -151.60 | 8.90 | -0.119 | -6.64 | 0.00 | 0.119 |
| R4 | $CO_{2(g)} + (a) \rightleftharpoons CO_{2(a)}$ | -144.74 | 0.00 | -0.129 | -7.22 | 52.29 | 0.129 |
| R5 | $CO_{2(g)} + (b) \rightleftharpoons CO_{2(b)}$ | -138.11 | 0.00 | -0.249 | -13.86 | 51.59 | 0.249 |
| R6 | $H_3CO_{(a)} + H_{(c)} \rightleftharpoons CH_3OH_{(g)} + (a) + (c)$ | 56.99 | 81.22 | -0.547 | -181.26 | 32.93 | 0.547 |
| R7 | $H_3CO_{(b)} + H_{(c)} \rightleftharpoons CH_3OH_{(g)} + (b) + (c)$ | 47.03 | 92.56 | -0.368 | -171.30 | 23.61 | 0.368 |
| R8 | $H_2O_{(g)} + (c) \rightleftharpoons H_2O_{(c)}$ | -177.19 | 0.00 | 0.378 | 21.08 | 31.21 | -0.378 |
| R9 | $H_2O_{(c)} + (c) \rightleftharpoons OH_{(c)} + H_{(c)}$ | 5.99 | 79.84 | 0.140 | 21.60 | 105.17 | -0.140 |
| R10 | $CO_{(a)} + H_{(c)} \rightleftharpoons HCO_{(a)} + (c)$ | 25.38 | 84.62 | -0.299 | -21.90 | 20.55 | 0.299 |
| R11 | $HCO_{(a)} + H_{(c)} \rightleftharpoons H_2CO_{(a)} + (c)$ | 15.73 | 56.26 | -0.299 | -8.10 | 92.81 | 0.299 |
| R12 | $CO_{2(a)} + H_{(c)} \rightleftharpoons HCOO_{(a)} + (c)$ | 36.44 | 77.74 | -0.249 | 8.72 | 109.64 | 0.249 |
| R13 | $CO_{2(b)} + H_{(c)} \rightleftharpoons HCOO_{(b)} + (c)$ | 43.07 | 60.24 | -0.368 | 2.08 | 105.43 | 0.368 |
| R14 | $HCOO_{(a)} + H_{(c)} \rightleftharpoons HCOOH_{(a)} + (c)$ | 10.42 | 128.23 | 0.000 | -153.70 | 29.81 | 0.000 |
| R15 | $HCOO_{(b)} + H_{(c)} \rightleftharpoons HCOOH_{(b)} + (c)$ | 10.42 | 136.53 | 0.000 | -153.70 | 7.72 | 0.000 |
| R16 | $HCOOH_{(a)} + H_{(c)} \rightleftharpoons H_2COOH_{(a)} + (c)$ | -150.42 | 46.89 | 0.000 | -25.96 | 62.43 | 0.000 |
| R17 | $HCOOH_{(b)} + H_{(c)} \rightleftharpoons H_2COOH_{(b)} + (c)$ | -150.42 | 1.54 | 0.000 | -25.96 | 48.92 | 0.000 |
| R18 | $H_2COOH_{(a)} + (c) \rightleftharpoons H_2CO_{(a)} + OH_{(c)}$ | -20.59 | 20.52 | -0.249 | -15.21 | 38.77 | 0.249 |
| R19 | $H_2COOH_{(b)} + (c) \rightleftharpoons H_2CO_{(b)} + OH_{(c)}$ | -13.95 | 11.22 | -0.368 | -21.85 | 9.96 | 0.368 |
| R20 | $H_2CO_{(a)} + H_{(c)} \rightleftharpoons H_3CO_{(a)} + (c)$ | 42.42 | 52.66 | -0.547 | -13.47 | 107.12 | 0.547 |
| R21 | $H_2CO_{(b)} + H_{(c)} \rightleftharpoons H_3CO_{(b)} + (c)$ | 32.46 | 51.46 | -0.368 | -3.51 | 132.06 | 0.368 |
| R22 | $CO_{(a)} + OH_{(c)} \rightleftharpoons COOH_{(a)} + (c)$ | -11.31 | 79.90 | -0.119 | -3.74 | 16.05 | 0.119 |
| R23 | $CO_{(b)} + OH_{(c)} \rightleftharpoons COOH_{(b)} + (c)$ | -11.31 | 13.33 | -0.119 | -3.74 | 16.05 | 0.119 |
| R24 | $COOH_{(a)} + OH_{(c)} \rightleftharpoons CO_{2(a)} + H_2O_{(c)}$ | 9.92 | 17.09 | -0.119 | -22.36 | 60.82 | 0.119 |
| R25 | $COOH_{(b)} + OH_{(c)} \rightleftharpoons CO_{2(b)} + H_2O_{(c)}$ | 9.92 | 17.09 | -0.119 | -22.37 | 60.82 | 0.119 |
| If the CO₂/CO_x ratio in feed is higher than 0.65, the activation energies of reactions 24 and 25 are adjusted: | | | | | | | |
| R24 | $COOH_{(a)} + OH_{(c)} \rightleftharpoons CO_{2(a)} + H_2O_{(c)}$ | 9.92 | 32.53 | -0.119 | -22.36 | 76.22 | 0.119 |
| R25 | $COOH_{(b)} + OH_{(c)} \rightleftharpoons CO_{2(b)} + H_2O_{(c)}$ | 9.92 | 32.53 | -0.119 | -22.37 | 76.22 | 0.119 |

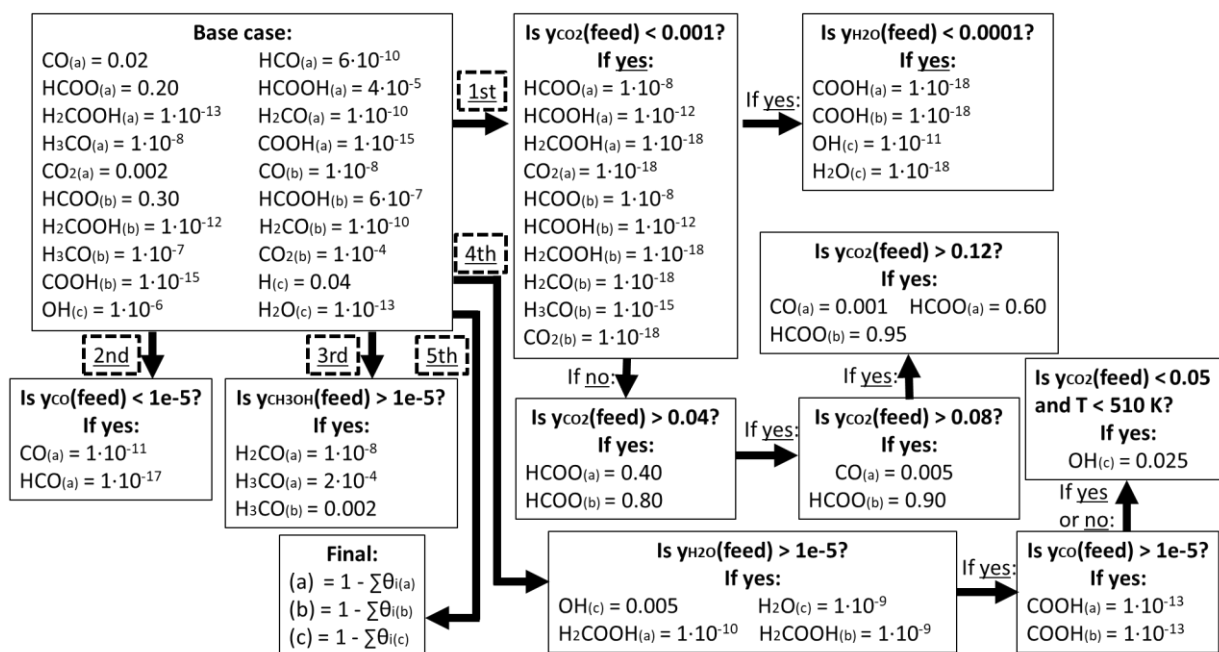


Figure S1. Initial guess method: Finding an educated initial guess to calculate the coverages. Reproduced with permission from Campos et al.¹ Copyright 2021 Royal Society of Chemistry.

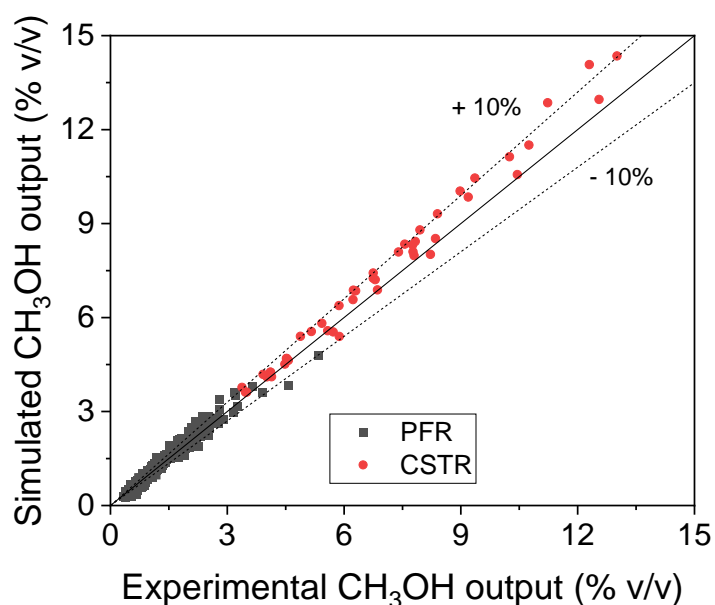


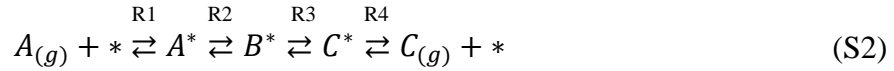
Figure S2. Parity plot of the methanol output concentration (% v/v) from steady-state experiments of the methanol synthesis performed in a PFR¹ and in a CSTR.² MKM simulations vs. experimental data.

S2. Expliciting the kinetic and thermodynamic terms in a microkinetic model

To develop the artificial neural network (ANN) in this work, one of the main assumptions was that the rate of a global reaction (r) consists in a kinetic term (r^K) and a thermodynamic term (r^T) which are multiplying each other. That is:

$$r = r^K \cdot r^T \quad (\text{S1})$$

While in formal kinetic models these terms are easily identified in the equations, they can also be separated in microkinetic models containing a single global reaction. Let's take a hypothetical system as an example: A adsorbs (R1), forms B (R2), which forms C (R3), and desorbs (R4).



Where * denotes a free active site.

The following rate equations can be written:

$$r_1 = k_1^+ \cdot f_A \cdot \theta_* - k_1^- \cdot \theta_{A^*} \quad (\text{S3})$$

$$r_2 = k_2^+ \cdot \theta_{A^*} - k_2^- \cdot \theta_{B^*} \quad (\text{S4})$$

$$r_3 = k_3^+ \cdot \theta_{B^*} - k_3^- \cdot \theta_{C^*} \quad (\text{S5})$$

$$r_4 = k_4^+ \cdot \theta_{C^*} - k_4^- \cdot f_C \cdot \theta_* \quad (\text{S6})$$

Here, k is the reaction constant, f_j is the fugacity of gas species j (in bar), θ_i is the coverage of surface species i , and the superscripts + and – refer to forward and reverse reaction, respectively.

The equilibrium constant of an elementary step (K) is defined as follows:

$$K = \frac{k^+}{k^-} \quad \rightarrow \quad k^- = \frac{k^+}{K} \quad (\text{S7})$$

Considering that the catalyst surface is in steady-state, all four elementary steps must have the same rate. That is:

$$r_1 = r_2 = r_3 = r_4 = r \quad (\text{S8})$$

Substituting Eq. (S7) and (S8) into Eqs. (S3-S6):

$$r = k_1^+ \cdot f_A \cdot \theta_* - \frac{k_1^+}{K_1} \cdot \theta_{A^*} \quad (\text{S9})$$

$$r = k_2^+ \cdot \theta_{A^*} - \frac{k_2^+}{K_2} \cdot \theta_{B^*} \quad (\text{S10})$$

$$r = k_3^+ \cdot \theta_{B^*} - \frac{k_3^+}{K_3} \cdot \theta_{C^*} \quad (\text{S11})$$

$$r = k_4^+ \cdot \theta_{C^*} - \frac{k_4^+}{K_4} \cdot f_C \cdot \theta_* \quad (\text{S12})$$

Isolating θ_{A^*} from Eq. (S9):

$$\theta_{A^*} = K_1 \cdot f_A \cdot \theta_* - \frac{K_1}{k_1^+} \cdot r \quad (\text{S13})$$

Isolating θ_{B^*} from Eq. (S10) and substituting θ_{A^*} from Eq. (S13):

$$\theta_{B^*} = K_1 \cdot K_2 \cdot f_A \cdot \theta_* - r \cdot \left(\frac{K_1 \cdot K_2}{k_1^+} + \frac{K_2}{k_2^+} \right) \quad (\text{S14})$$

Isolating θ_{C^*} from Eq. (S11) and substituting θ_{B^*} from Eq. (S14):

$$\theta_{C^*} = K_1 \cdot K_2 \cdot K_3 \cdot f_A \cdot \theta_* - r \cdot \left(\frac{K_1 \cdot K_2 \cdot K_3}{k_1^+} + \frac{K_2 \cdot K_3}{k_2^+} + \frac{K_3}{k_3^+} \right) \quad (\text{S15})$$

Substituting Eq. (S15) in Eq. (S12):

$$r = k_4^+ \cdot \left[K_1 \cdot K_2 \cdot K_3 \cdot f_A \cdot \theta_* - r \cdot \left(\frac{K_1 \cdot K_2 \cdot K_3}{k_1^+} + \frac{K_2 \cdot K_3}{k_2^+} + \frac{K_3}{k_3^+} \right) \right] - \frac{k_4^+}{K_4} \cdot f_C \cdot \theta_* \quad (\text{S16})$$

Multiplying both sides by K_4/k_4^+ and grouping the terms with r on the left side:

$$r \cdot \left(\frac{K_1 \cdot K_2 \cdot K_3 \cdot K_4}{k_1^+} + \frac{K_2 \cdot K_3 \cdot K_4}{k_2^+} + \frac{K_3 \cdot K_4}{k_3^+} + \frac{K_4}{k_4^+} \right) = K_1 \cdot K_2 \cdot K_3 \cdot K_4 \cdot f_A \cdot \theta_* - f_C \cdot \theta_* \quad (\text{S17})$$

The equilibrium constant of the global reaction $A \rightleftharpoons C$ (K_P^0) is the product of the elementary equilibrium constants:

$$K_P^0 = K_1 \cdot K_2 \cdot K_3 \cdot K_4 \quad (\text{S18})$$

Substituting Eq. (S18) in Eq. (S17):

$$r \cdot \left(\frac{K_P^0}{k_1^+} + \frac{K_2 \cdot K_3 \cdot K_4}{k_2^+} + \frac{K_3 \cdot K_4}{k_3^+} + \frac{K_4}{k_4^+} \right) = K_P^0 \cdot f_A \cdot \theta_* - f_C \cdot \theta_* \quad (\text{S19})$$

Isolating r and reorganizing:

$$r = \frac{K_P^0 \cdot f_A \cdot \theta_*}{\left(\frac{K_P^0}{k_1^+} + \frac{K_2 \cdot K_3 \cdot K_4}{k_2^+} + \frac{K_3 \cdot K_4}{k_3^+} + \frac{K_4}{k_4^+} \right)} \cdot \left(1 - \frac{f_C}{f_A \cdot K_P^0} \right) \quad (\text{S20})$$

Here, we can define k as the effective global reaction constant:

$$k = \frac{K_P^0}{\left(\frac{K_P^0}{k_1^+} + \frac{K_2 \cdot K_3 \cdot K_4}{k_2^+} + \frac{K_3 \cdot K_4}{k_3^+} + \frac{K_4}{k_4^+} \right)} \quad (\text{S21})$$

Finally, substituting Eq. (S21) in Eq. (S20):

$$r = [k \cdot f_A \cdot \theta_*] \cdot \left(1 - \frac{f_C}{f_A \cdot K_P^0}\right) \quad (\text{S22})$$

In Eq. (S22), the kinetic term (r^K) and thermodynamic term (r^T) are now identifiable:

$$r^K = [k \cdot f_A \cdot \theta_*] \quad (\text{S23})$$

$$r^T = \left(1 - \frac{f_C}{f_A \cdot K_P^0}\right) \quad (\text{S24})$$

The majority of the reactive systems have multiple global reactions, in which this explicit separation of r^K and r^T is not possible, as an algebraic system with multiple variables would be reached. Still, the concept of Eq. (S1) should still be true, and one can calculate the kinetic term indirectly by dividing the reaction rate with the thermodynamic part, i.e.:

$$r^K = \frac{r}{r^T} \quad (\text{S25})$$

S3. Calculation of the hydrogen equilibrium conversion

At chemical equilibrium in the methanol synthesis, the following equations must hold:^{1, 3-4}

$$K_{P,CO\text{ hyd.}}^0 = T^{-3.384} \cdot \exp(10092.4 \cdot T^{-1} - 4.200) = \frac{f_{CH_3OH}}{f_{H_2}^2 \cdot f_{CO}} \quad (S26)$$

$$K_{P,CO_2\text{ hyd.}}^0 = T^{-4.481} \cdot \exp(4755.7 \cdot T^{-1} + 8.369) = \frac{f_{CH_3OH} \cdot f_{H_2O}}{f_{H_2}^3 \cdot f_{CO_2}} \quad (S27)$$

$$K_{P,rWGS}^0 = T^{-1.097} \cdot \exp(-5337.4 \cdot T^{-1} + 12.569) = \frac{f_{CO} \cdot f_{H_2O}}{f_{H_2} \cdot f_{CO_2}} \quad (S28)$$

Since this system of reactions has only two degrees of freedom, two reactions are sufficient to describe it. To calculate H₂ equilibrium conversion, we chose the hydrogenation of CO and the hydrogenation of CO₂. The reaction extensions are defined as follows, with CO and CO₂ as references (X_{CO} and X_{CO_2}):

$$X_{CO} = \frac{\dot{n}_{CO}^{in} - \dot{n}_{CO}}{\dot{n}_{CO}^{in}} \quad (S29)$$

$$X_{CO_2} = \frac{\dot{n}_{CO_2}^{in} - \dot{n}_{CO_2}}{\dot{n}_{CO_2}^{in}} \quad (S30)$$

Where \dot{n}_{CO}^{in} and $\dot{n}_{CO_2}^{in}$ are the inlet mole flow of CO and CO₂, respectively, and \dot{n}_{CO} and \dot{n}_{CO_2} are the mole flow of CO and CO₂ after the reaction.

The mole flow of each component after the reaction is calculated according to its respective stoichiometric coefficients and the reaction extensions, as follows:

$$\dot{n}_{H_2} = \dot{n}_{H_2}^{in} - 2 \cdot X_{CO} \cdot \dot{n}_{CO}^{in} - 3 \cdot X_{CO_2} \cdot \dot{n}_{CO_2}^{in} \quad (S31)$$

$$\dot{n}_{CO} = \dot{n}_{CO}^{in} \cdot (1 - X_{CO}) \quad (S32)$$

$$\dot{n}_{CO_2} = \dot{n}_{CO_2}^{in} \cdot (1 - X_{CO_2}) \quad (S33)$$

$$\dot{n}_{CH_3OH} = \dot{n}_{CH_3OH}^{in} + X_{CO} \cdot \dot{n}_{CO}^{in} + X_{CO_2} \cdot \dot{n}_{CO_2}^{in} \quad (S34)$$

$$\dot{n}_{H_2O} = \dot{n}_{H_2O}^{in} + X_{CO_2} \cdot \dot{n}_{CO_2}^{in} \quad (S35)$$

The mole fraction of each component (y_j) after the reaction is calculated as follows:

$$y_j = \frac{\dot{n}_j}{\dot{n}} \quad (S36)$$

$$\dot{n} = \sum_{j=1}^5 \dot{n}_j \quad (\text{S37})$$

Here, \dot{n} is the total mole flow after the reaction.

With the information of the total pressure, the reaction temperature, and the mole fractions, the fugacity of the components (f_j) were estimated with the Peng-Robinson equation of state.⁵ Binary interaction parameters and other necessary information were taken from literature,⁶⁻⁷ including an effective hydrogen acentric factor of -0.05 .⁸

The aforementioned equations (Eqs. S26-S27, S29-S37, and the Peng-Robinson equations) form a non-linear algebraic system with two variables (X_{CO} , X_{CO_2}). This system is solved with the Matlab function *fsolve*, with the function and step tolerances set to 10^{-6} . With the final $X_{CO,eq}$ and $X_{CO_2,eq}$, the equilibrium H_2 conversion ($X_{H_2,eq}$) is calculated as follows:

$$X_{H_2,eq} = \frac{(\dot{n}_{H_2}^{in} - 2 \cdot X_{CO,eq} \cdot \dot{n}_{CO}^{in} - 3 \cdot X_{CO_2,eq} \cdot \dot{n}_{CO_2}^{in})}{\dot{n}_{H_2}^{in}} \quad (\text{S38})$$

S4. Parameters of the developed ANNs

Table S2. ANN₁₀ parameters of the hidden layer (p_{wj}). w refers to the cell index while j refers to the input index.

| p_{wj} | $j = 0$ | $j = 1$ | $j = 2$ | $j = 3$ | $j = 4$ | $j = 5$ | $j = 6$ |
|----------|------------|------------|------------|------------|------------|------------|------------|
| $w = 1$ | -0.9716898 | 0.0041664 | -0.0061756 | 0.0100854 | -0.0101530 | 0.0003118 | -0.2135094 |
| $w = 2$ | -0.9959330 | 0.0240174 | 0.0051838 | -0.0024755 | -0.3549400 | 0.0002915 | -0.0028441 |
| $w = 3$ | -0.9748353 | 0.0192092 | 0.0140785 | -0.0291657 | -0.0107911 | 0.0007449 | -0.0138335 |
| $w = 4$ | -0.9605484 | 0.0259080 | 0.0053966 | 0.0075902 | -0.0970261 | 0.0068462 | -0.0081281 |
| $w = 5$ | 0.9384540 | -0.0468252 | -0.0046727 | -0.0112876 | 0.1626274 | -0.0126964 | 0.0115955 |
| $w = 6$ | 0.8449822 | -0.0413623 | 0.0264285 | -0.0291187 | 0.0202620 | 0.0017597 | -0.3306908 |
| $w = 7$ | -1.0459629 | 0.0638154 | 0.0261562 | -0.0030565 | 0.0170041 | -0.0035466 | -0.0047106 |
| $w = 8$ | 0.8054555 | -0.0534534 | 0.0582256 | 0.5045525 | -0.0113396 | -0.0154450 | -0.9923904 |
| $w = 9$ | 0.9736794 | -0.0229409 | 0.0358447 | -0.0001882 | -0.0047470 | 0.0027775 | -0.0288284 |
| $w = 10$ | 0.9566228 | -0.0345442 | -0.0065016 | -0.0005030 | 0.0005019 | 0.0001845 | 0.0061815 |

Table S3. ANN₁₀ parameters of the output layer (\hat{p}_{nw}). n refers to the output index while w refers to the cell index. The table was transposed because of space.

| \hat{p}_{nw}' | $n = 1$ | $n = 2$ | $n = 3$ |
|-----------------|-------------|--------------|-------------|
| $w = 0$ | 11068.3626 | -94142.7103 | -29624.1371 |
| $w = 1$ | 709.9192 | -282.5021 | -56164.3427 |
| $w = 2$ | 649.2352 | -101175.0807 | 4744.4682 |
| $w = 3$ | -15995.9726 | -3.6427 | 1353.2367 |
| $w = 4$ | 34472.1413 | 5405.3482 | 18589.0573 |
| $w = 5$ | 5102.2267 | 998.5848 | 2790.5874 |
| $w = 6$ | -4.3013 | 2.0209 | -152.1544 |
| $w = 7$ | 577.4606 | 11890.9206 | 1277.1276 |
| $w = 8$ | 0.7117 | -0.2048 | 13.0161 |
| $w = 9$ | 7507.0901 | 12992.4854 | -5848.5115 |
| $w = 10$ | -3262.2452 | -4016.0001 | 2619.7030 |

Table S4. ANN₂₀ parameters of the hidden layer (p_{wj}). w refers to the cell index while j refers to the input index.

| p_{wj} | $j = 0$ | $j = 1$ | $j = 2$ | $j = 3$ | $j = 4$ | $j = 5$ | $j = 6$ |
|----------|------------|------------|------------|------------|------------|------------|------------|
| $w = 1$ | 0.4107634 | -0.3476371 | -0.0343069 | -0.4910167 | 0.3421876 | 0.0587555 | 0.3425156 |
| $w = 2$ | 0.4788869 | -0.5265364 | -0.0705651 | 0.0468037 | -0.0430378 | 0.0112581 | 0.0226196 |
| $w = 3$ | 0.5235582 | -0.0012607 | 0.1937560 | 0.4779385 | 0.1020426 | -0.0414237 | -1.3779806 |
| $w = 4$ | -0.4101937 | 0.0879208 | -0.1416665 | -0.0958177 | 1.1418159 | -0.0091950 | 0.0577283 |
| $w = 5$ | 1.0299699 | -0.0610526 | -0.0215066 | 0.0006823 | -0.0074083 | -0.0004268 | 0.0119220 |
| $w = 6$ | -0.0667126 | -0.0922277 | -0.1715939 | -0.0862231 | 0.5347287 | -0.0446638 | 0.9247602 |
| $w = 7$ | 0.4684067 | -0.0953478 | 0.1615514 | 0.1012457 | -1.3064510 | 0.0100775 | -0.0753646 |
| $w = 8$ | -0.9865289 | 0.0214754 | 0.0044386 | -0.0040830 | -0.8306544 | -0.0012574 | 0.0120926 |
| $w = 9$ | 0.1672123 | -0.1619341 | 0.0224200 | 0.0914151 | -0.1196893 | 0.0058689 | 0.0063903 |
| $w = 10$ | 0.1316965 | -0.5871169 | 0.6962471 | 0.1545037 | -1.0756671 | 0.1769251 | -0.2042572 |
| $w = 11$ | -0.9663492 | 0.0023345 | -0.0048524 | 0.0006847 | -0.0033376 | 0.0003150 | -0.6501090 |
| $w = 12$ | 0.2377170 | -0.1813779 | 0.0092137 | -0.3371499 | 0.2234223 | 0.0387758 | 0.1925004 |
| $w = 13$ | -0.9496994 | 0.0345588 | -0.0588820 | -0.0009962 | -0.0107111 | 0.0057568 | 0.0347929 |
| $w = 14$ | -0.0548285 | -0.0008430 | 0.1472877 | -0.1680355 | 0.0992056 | 0.0230020 | -0.6062469 |
| $w = 15$ | -0.2002659 | 0.0226395 | 0.0513328 | 0.2348420 | 0.0364116 | -0.0330605 | -1.0627367 |
| $w = 16$ | -1.0033942 | 0.0376572 | 0.0057459 | 0.0014855 | -0.1792533 | 0.0078282 | -0.0097843 |
| $w = 17$ | -0.1562959 | 0.0498811 | 0.0054537 | 0.0517360 | 0.0080279 | -0.0135540 | -0.9118531 |
| $w = 18$ | 0.0752436 | 0.0109804 | 0.0557756 | -0.2336342 | 0.0568813 | 0.0205622 | -0.1238210 |
| $w = 19$ | 0.9272197 | 0.0160861 | 0.0387003 | -0.0102710 | 0.0799524 | -0.0009957 | -0.1313046 |
| $w = 20$ | 0.2602767 | -0.0104805 | -0.0757907 | -0.3215158 | -0.0402635 | 0.0363001 | 1.0610396 |

Table S5. ANN₂₀ parameters of the output layer (\hat{p}_{nw}). n refers to the output index while w refers to the cell index. The table was transposed because of space.

| \hat{p}_{nw}' | $n = 1$ | $n = 2$ | $n = 3$ |
|-----------------|------------|-------------|-------------|
| $w = 0$ | -3192.6495 | -48814.6635 | -23609.9722 |
| $w = 1$ | 19.1794 | 0.3520 | 5.1972 |
| $w = 2$ | -5.9923 | -1.8207 | -1.2544 |
| $w = 3$ | -9.7435 | -0.3226 | -37.5713 |
| $w = 4$ | 64.1003 | 3.1109 | 20.3288 |
| $w = 5$ | -4512.9794 | -11740.4489 | 1189.2378 |
| $w = 6$ | 0.2324 | 0.0888 | -5.9175 |
| $w = 7$ | 42.6292 | 2.2433 | 12.4537 |
| $w = 8$ | 236.6448 | -35073.4100 | 1479.1315 |
| $w = 9$ | 80.5643 | 5.2245 | 6.9051 |
| $w = 10$ | 0.2542 | -0.0543 | 0.3699 |
| $w = 11$ | 275.8564 | 206.0867 | -19976.7228 |
| $w = 12$ | -80.8444 | -1.8040 | -22.7775 |
| $w = 13$ | -1658.9701 | -1800.4704 | -1332.2755 |
| $w = 14$ | 0.6853 | -0.0314 | 129.9186 |
| $w = 15$ | 102.0798 | 1.5156 | 460.0508 |
| $w = 16$ | -4947.7715 | -23877.7532 | 2314.5123 |
| $w = 17$ | -24.5357 | 0.5622 | -133.4187 |
| $w = 18$ | 61.3209 | 1.8288 | -202.0073 |
| $w = 19$ | 1608.1333 | 9.3080 | 4925.4788 |
| $w = 20$ | 96.7751 | 2.1706 | 411.3621 |

Table S6. ANN₃₀ parameters of the hidden layer (p_{wj}). w refers to the cell index while j refers to the input index.

| p_{wj} | $j = 0$ | $j = 1$ | $j = 2$ | $j = 3$ | $j = 4$ | $j = 5$ | $j = 6$ |
|----------|------------|------------|------------|------------|------------|------------|------------|
| $w = 1$ | 0.9210260 | 0.0144556 | -0.0055219 | -0.0254152 | 0.0265035 | 0.0012437 | 0.7429898 |
| $w = 2$ | -0.9764701 | 0.0370388 | -0.0669420 | -0.0983673 | 0.0187385 | 0.0166996 | 0.1093835 |
| $w = 3$ | -0.9514059 | 0.0003606 | -0.0140501 | -0.1407724 | -0.0163624 | 0.0081432 | 0.0657403 |
| $w = 4$ | -0.9676759 | 0.0011831 | -0.0200063 | 0.0013056 | -0.0232506 | 0.0025903 | -1.0061496 |
| $w = 5$ | -0.9860803 | 0.0153332 | 0.0020657 | -0.0040916 | -0.7942445 | -0.0006715 | 0.0210091 |
| $w = 6$ | -1.0314882 | 0.0454929 | 0.0179973 | -0.0002921 | 0.0078623 | -0.0008390 | -0.0048101 |
| $w = 7$ | 0.9484849 | 0.0042774 | -0.0191207 | -0.0020081 | 0.0181528 | 0.0001254 | -0.0308138 |
| $w = 8$ | -0.9820565 | -0.0081976 | 0.0098745 | 0.0063298 | -0.1588040 | -0.0018979 | 0.0043109 |
| $w = 9$ | 0.9800557 | -0.0200801 | 0.0256699 | -0.0011864 | 0.0120047 | -0.0036895 | -0.0071408 |
| $w = 10$ | -1.0237492 | 0.1225330 | 0.0721624 | 0.0169683 | -0.8349192 | 0.0280220 | -0.1447876 |
| $w = 11$ | -0.2962458 | -0.0197919 | 0.1882503 | -0.0428655 | -0.0017384 | 0.0123888 | -1.0239027 |
| $w = 12$ | -0.8529406 | -0.0309410 | 0.0117282 | 0.0510897 | -0.0522445 | -0.0022420 | -1.3906473 |
| $w = 13$ | -0.0412349 | -0.0239703 | -0.0166151 | -0.0425768 | 0.2393774 | -0.0089242 | 0.0314645 |
| $w = 14$ | -1.0136615 | 0.0805500 | 0.0448732 | 0.0104937 | -0.5435348 | 0.0179463 | -0.0907573 |
| $w = 15$ | 0.0630118 | 0.1521816 | 0.1063422 | 0.0638993 | -0.1034713 | 0.0127883 | -0.2630422 |
| $w = 16$ | -0.0208821 | -0.0524536 | 0.0550701 | 0.0802684 | 0.0609775 | 0.0010868 | 0.0847652 |
| $w = 17$ | -0.1048916 | -0.0497711 | -0.0234816 | 0.1069471 | 0.0761395 | 0.0004229 | 0.1142789 |
| $w = 18$ | 0.3777372 | -0.0152203 | -0.0987326 | -0.2421954 | -0.0604080 | -0.0216252 | 0.8987170 |
| $w = 19$ | 0.8877308 | -0.5220655 | 0.1144222 | 0.1379238 | -0.2481600 | 0.0105050 | 0.5708149 |
| $w = 20$ | 0.0315558 | 0.0517333 | -0.0794345 | 0.0396991 | -0.1372819 | 0.0049044 | -0.0271347 |
| $w = 21$ | 0.5344184 | -0.5093209 | -0.0224921 | 0.3218643 | 0.5777068 | -0.0424688 | -0.1385745 |
| $w = 22$ | 0.9566903 | -0.2122409 | -0.2471810 | -0.4637741 | -0.1823084 | -0.0257392 | -0.3850709 |
| $w = 23$ | -0.7326411 | -0.0041719 | -0.0927623 | -0.7648524 | -0.0931389 | 0.0395052 | 0.3862078 |
| $w = 24$ | 0.8701789 | -0.0895655 | -0.2855420 | -0.5321962 | -0.1703353 | -0.0436570 | 0.9563571 |
| $w = 25$ | 0.9039345 | 0.1360256 | -0.8335025 | -0.0212982 | 0.2684881 | 0.0329665 | 0.8461650 |
| $w = 26$ | -0.9189061 | -0.1482228 | 0.8775237 | 0.0217204 | -0.2768904 | -0.0450017 | -0.7943776 |
| $w = 27$ | 0.3367264 | 0.6245309 | -0.1381385 | -0.2949816 | -0.3098113 | 0.0680996 | -1.1056290 |
| $w = 28$ | 0.9425932 | -0.1788266 | -0.2369277 | -0.4742727 | -0.1979102 | -0.0250730 | -0.3491069 |
| $w = 29$ | -0.0055630 | 0.2531751 | -0.1584149 | -0.0627589 | 0.2618731 | -0.0104989 | -0.1385452 |
| $w = 30$ | -0.9658869 | 0.0534001 | -0.0954103 | -0.1401058 | 0.0273488 | -0.0006017 | 0.1378509 |

Table S7. ANN₃₀ parameters of the output layer (\hat{p}_{nw}). n refers to the output index while w refers to the cell index. The table was transposed because of space.

| \hat{p}_{nw}' | $n = 1$ | $n = 2$ | $n = 3$ |
|-----------------|-------------|-------------|-------------|
| $w = 0$ | 5014.6684 | -67218.7139 | -24640.1621 |
| $w = 1$ | -141.8928 | -356.0612 | 12208.2237 |
| $w = 2$ | -4537.3650 | -1618.7976 | -33857.4346 |
| $w = 3$ | -15669.8679 | -2035.7802 | -34751.8883 |
| $w = 4$ | -1698.7260 | 74.0943 | -23834.3334 |
| $w = 5$ | -2897.8441 | -67366.3029 | 2648.7725 |
| $w = 6$ | -4253.7984 | 51194.8479 | 1622.2464 |
| $w = 7$ | 1738.9680 | 3965.0607 | -25216.0578 |
| $w = 8$ | 47509.0643 | -3933.6909 | 21278.2679 |
| $w = 9$ | 11449.9315 | 31428.3834 | -17823.7016 |
| $w = 10$ | 316.2938 | 2112.2209 | -587.5362 |
| $w = 11$ | 2.5557 | -0.6271 | 54.5921 |
| $w = 12$ | -10.0019 | -33.2223 | 1099.8649 |
| $w = 13$ | 105.3360 | -24.9122 | 54.2074 |
| $w = 14$ | -1602.1186 | -10959.1007 | 2929.0181 |
| $w = 15$ | -40.8261 | -6.9090 | 5.7855 |
| $w = 16$ | 314.5280 | -1.1459 | 126.6244 |
| $w = 17$ | -255.9859 | 5.5620 | 6.4420 |
| $w = 18$ | -4.4710 | -2.1267 | 62.6596 |
| $w = 19$ | -1.5361 | -1.6503 | -1.2003 |
| $w = 20$ | 217.8041 | -22.7268 | 131.8603 |
| $w = 21$ | -1.2949 | -0.3805 | -0.9806 |
| $w = 22$ | 32.6347 | -3.2432 | -3.7043 |
| $w = 23$ | 23.2024 | 3.1948 | 59.0229 |
| $w = 24$ | 3.6723 | 0.6332 | -11.5088 |
| $w = 25$ | 4.3456 | 2.7947 | 24.1452 |
| $w = 26$ | 3.9460 | 2.6198 | 19.2900 |
| $w = 27$ | 1.2927 | 0.3668 | 1.5258 |
| $w = 28$ | -36.7011 | 2.8556 | 9.0313 |
| $w = 29$ | 2.1071 | 6.4978 | 5.4788 |
| $w = 30$ | 917.3079 | 377.9798 | 7931.9928 |

Table S8. ANN₄₀ parameters of the hidden layer (p_{wj}). w refers to the cell index while j refers to the input index.

| p_{wj} | $j = 0$ | $j = 1$ | $j = 2$ | $j = 3$ | $j = 4$ | $j = 5$ | $j = 6$ |
|----------|------------|------------|------------|------------|------------|------------|------------|
| $w = 1$ | 0.9200367 | 0.0076764 | -0.0009857 | -0.0165760 | 0.0175379 | 0.0036085 | 1.1262685 |
| $w = 2$ | -0.9827898 | 0.0496525 | -0.1002516 | -0.0668408 | 0.0047046 | 0.0415199 | 0.1402700 |
| $w = 3$ | -0.9308891 | -0.0001035 | -0.0269745 | -0.4010498 | -0.0368052 | 0.0032319 | 0.1173029 |
| $w = 4$ | -0.9687155 | -0.0009855 | -0.0243023 | 0.0029107 | -0.0203690 | 0.0033600 | -1.3451289 |
| $w = 5$ | -0.9852336 | 0.0142400 | 0.0014441 | -0.0031647 | -0.8427927 | 0.0003946 | 0.0189562 |
| $w = 6$ | -1.0577440 | 0.0645960 | 0.0225978 | -0.0035011 | 0.0133353 | -0.0033700 | -0.0141970 |
| $w = 7$ | 0.9672956 | -0.0006928 | -0.0020129 | 0.0020031 | 0.0492340 | 0.0005421 | -0.0786190 |
| $w = 8$ | -0.9820317 | -0.0094197 | 0.0094059 | 0.0063725 | -0.1994111 | -0.0020577 | 0.0111338 |
| $w = 9$ | 0.9335614 | -0.0199485 | 0.0194790 | -0.0012524 | 0.0097728 | -0.0032763 | -0.0050180 |
| $w = 10$ | -1.0206142 | 0.1171118 | 0.0694301 | 0.0106954 | -0.9272709 | 0.0216159 | -0.0984454 |
| $w = 11$ | -0.2021453 | -0.0663709 | 0.3816407 | -0.0399833 | -0.0261569 | -0.0198502 | -1.1507226 |
| $w = 12$ | -0.8642980 | -0.0142196 | 0.0024998 | 0.0284852 | -0.0300141 | -0.0066122 | -1.9375957 |
| $w = 13$ | -0.3665186 | -0.0618812 | -0.0245794 | -0.0813862 | 0.2434990 | -0.0155301 | 0.0221331 |
| $w = 14$ | -1.0105644 | 0.0696289 | 0.0388431 | 0.0058528 | -0.5436944 | 0.0125731 | -0.0555272 |
| $w = 15$ | 0.1173194 | 0.1951941 | -0.1354525 | 0.0246064 | -0.1158351 | 0.0259153 | 0.0048485 |
| $w = 16$ | -0.0292299 | -0.0397383 | 0.0363245 | 0.0751064 | 0.0140031 | 0.0035132 | 0.0213155 |
| $w = 17$ | -0.1151793 | -0.0272537 | -0.0379548 | 0.1299361 | 0.0129362 | 0.0044727 | 0.0567924 |
| $w = 18$ | 0.2381513 | 0.0069968 | -0.0690950 | -0.3235856 | 0.0025524 | -0.0123455 | 0.9082660 |
| $w = 19$ | 0.6764443 | -0.3708194 | 0.0610742 | -0.0138102 | -0.0098356 | -0.0233576 | 0.0042944 |
| $w = 20$ | 0.1776015 | 0.0447696 | -0.0705152 | 0.0920699 | -0.1000310 | 0.0024711 | 0.0430449 |
| $w = 21$ | 0.4755637 | -0.3271537 | -0.0750362 | 0.3348838 | 0.6578049 | -0.0353007 | -0.0507947 |
| $w = 22$ | 0.8001454 | -0.2364633 | -0.0887444 | -0.3893658 | 0.0442344 | 0.0264606 | -0.2680961 |
| $w = 23$ | -0.8155336 | -0.0009335 | -0.0734646 | -1.0857503 | -0.1055481 | 0.0064367 | 0.3174450 |
| $w = 24$ | 0.7063370 | -0.0240587 | -0.2451499 | -0.9691192 | 0.0643380 | -0.0601396 | 2.2032006 |
| $w = 25$ | 0.6289627 | 0.3954138 | -1.2483782 | 0.0063962 | 0.1141701 | 0.1673217 | 1.5710404 |
| $w = 26$ | -0.6302366 | -0.3982644 | 1.2502565 | -0.0051650 | -0.1157481 | -0.1688652 | -1.5453826 |
| $w = 27$ | 0.4735275 | 0.2978224 | -0.1135825 | -0.0780181 | -0.4178516 | 0.0576710 | -0.2308728 |
| $w = 28$ | 0.8221592 | -0.1029569 | 0.0060872 | -0.5483422 | -0.0782219 | 0.0694359 | -1.4717670 |
| $w = 29$ | 0.1352859 | 0.1493739 | -0.1644704 | -0.0052121 | 0.1976783 | 0.0194770 | 0.0758168 |
| $w = 30$ | -0.9755096 | 0.0670431 | -0.1349110 | -0.0906471 | 0.0062291 | 0.0325506 | 0.1668614 |
| $w = 31$ | 1.6862808 | -0.6694639 | -1.2833303 | -1.0903861 | 2.3292406 | -0.0465565 | 1.2805244 |

| | | | | | | | |
|---------------|------------|------------|------------|------------|------------|------------|------------|
| w = 32 | -0.4655295 | -0.1729073 | 0.0404600 | -0.3131722 | 0.2125989 | -0.0497864 | 1.9471693 |
| w = 33 | 0.5020775 | -0.2096806 | 0.0905325 | -0.0383210 | 0.5208753 | -0.0528558 | 0.2703349 |
| w = 34 | 0.2849129 | 0.0954594 | -0.0380159 | 0.5053490 | 0.1045337 | -0.0474973 | 0.0986259 |
| w = 35 | 0.8222148 | -0.7026919 | -0.1287898 | -0.0123962 | -0.0646418 | -0.0159605 | 0.0724903 |
| w = 36 | -0.0989402 | -0.0397708 | -0.5586013 | -0.0453288 | 0.0873285 | 0.0487099 | 0.2711086 |
| w = 37 | 0.2411174 | 0.0070856 | 0.0771951 | -0.0647974 | 0.0479239 | -0.0473074 | 1.5814832 |
| w = 38 | -0.5713111 | 0.1007126 | 0.1058824 | -0.0806702 | -0.4140267 | 0.0331325 | -0.4111302 |
| w = 39 | -0.3920248 | -0.1036485 | -0.0980156 | 0.1339449 | 0.2666204 | -0.0394928 | 1.0521331 |
| w = 40 | 0.4839421 | -0.1284074 | -0.0375403 | -0.0579197 | 0.5687728 | -0.0458338 | 0.2178815 |

Table S9. ANN₄₀ parameters of the output layer (\hat{p}_{nw}). n refers to the output index while w refers to the cell index. The table was transposed because of space.

| \hat{p}_{nw}' | $n = 1$ | $n = 2$ | $n = 3$ |
|-----------------|------------|-------------|-------------|
| $w = 0$ | 9507.3185 | -66056.0886 | -26159.9740 |
| $w = 1$ | 1522.8483 | -103.7176 | 14027.6114 |
| $w = 2$ | -8147.1965 | -1610.6743 | -33520.0646 |
| $w = 3$ | -8364.9046 | 110.7577 | -35273.7378 |
| $w = 4$ | -432.1208 | 328.3294 | -25438.0037 |
| $w = 5$ | -2605.5484 | -67054.3050 | 3217.6946 |
| $w = 6$ | -6243.7633 | 48481.3382 | 1503.8741 |
| $w = 7$ | -2631.4782 | 1147.4086 | -24189.5422 |
| $w = 8$ | 43947.8750 | -4943.5291 | 17180.0172 |
| $w = 9$ | 12730.2588 | 26530.2238 | -20943.9385 |
| $w = 10$ | -55.7499 | 2112.2702 | -391.8940 |
| $w = 11$ | 6.4570 | 1.9100 | 57.3262 |
| $w = 12$ | 207.1185 | -14.2433 | 1791.6267 |
| $w = 13$ | 146.7510 | -22.4728 | 70.4588 |
| $w = 14$ | 246.3041 | -16426.9455 | 2902.5883 |
| $w = 15$ | -40.3999 | -14.6354 | 2.5303 |
| $w = 16$ | 353.2149 | -30.2512 | 159.6581 |
| $w = 17$ | -257.6934 | 15.8314 | 3.7484 |
| $w = 18$ | 4.6366 | -0.2838 | 67.7121 |
| $w = 19$ | -15.4434 | -17.1775 | -0.1232 |
| $w = 20$ | 283.2293 | -11.2045 | 161.0353 |
| $w = 21$ | -3.5059 | -0.8316 | -2.6006 |
| $w = 22$ | 12.4337 | 0.1920 | -0.1678 |
| $w = 23$ | 185.8430 | -3.1862 | 824.5165 |
| $w = 24$ | -0.1302 | 0.0448 | -2.4149 |
| $w = 25$ | 15.7102 | 10.2271 | 101.9132 |
| $w = 26$ | 15.5747 | 10.2622 | 99.8888 |
| $w = 27$ | 35.0183 | 1.8037 | 18.6600 |
| $w = 28$ | -1.8163 | -0.3499 | 7.8325 |
| $w = 29$ | 14.1689 | 10.2108 | 15.5896 |
| $w = 30$ | 2433.1696 | 529.2693 | 9965.6195 |

| | | | |
|---------------|----------|---------|----------|
| w = 31 | 0.0477 | -0.0012 | 0.0022 |
| w = 32 | -0.8329 | 0.1420 | -7.9189 |
| w = 33 | -50.2144 | -3.2071 | -15.8676 |
| w = 34 | -9.2777 | 0.4421 | -12.1313 |
| w = 35 | 0.9162 | 2.7283 | -0.3383 |
| w = 36 | 2.9042 | 2.4056 | -1.2588 |
| w = 37 | 0.9017 | 0.4573 | -11.5757 |
| w = 38 | 4.5984 | -8.1404 | -30.9170 |
| w = 39 | 1.8114 | 1.1330 | 28.8090 |
| w = 40 | 48.5507 | -5.4538 | 4.9950 |

References

1. Lacerda de Oliveira Campos, B.; Herrera Delgado, K.; Wild, S.; Studt, F.; Pitter, S.; Sauer, J., Surface reaction kinetics of the methanol synthesis and the water gas shift reaction on Cu/ZnO/Al₂O₃. *Reaction Chemistry & Engineering* **2021**, *6* (5), 868-887.
2. Seidel, C.; Jörke, A.; Vollbrecht, B.; Seidel-Morgenstern, A.; Kienle, A., Kinetic modeling of methanol synthesis from renewable resources. *Chemical Engineering Science* **2018**, *175*, 130-138.
3. Goos, E.; Burcat, A.; Ruscic, B. New NASA thermodynamic polynomials database. Available at: <http://garfield.chem.elte.hu/Burcat/THERM.DAT> (Access in Mar. 2022).
4. Lacerda de Oliveira Campos, B.; Herrera Delgado, K.; Pitter, S.; Sauer, J., Development of Consistent Kinetic Models Derived from a Microkinetic Model of the Methanol Synthesis. *Industrial & Engineering Chemistry Research* **2021**.
5. Peng, D.-Y.; Robinson, D. B., A New Two-Constant Equation of State. *Industrial & Engineering Chemistry Fundamentals* **1976**, *15* (1), 59-64.
6. Meng, L.; Duan, Y.-Y., Prediction of the second cross virial coefficients of nonpolar binary mixtures. *Fluid Phase Equilibria* **2005**, *238* (2), 229-238.
7. Meng, L.; Duan, Y.-Y.; Wang, X.-D., Binary interaction parameter *k*_{ij} for calculating the second cross-virial coefficients of mixtures. *Fluid Phase Equilibria* **2007**, *260* (2), 354-358.
8. Deiters, U. K., Comments on the modeling of hydrogen and hydrogen-containing mixtures with cubic equations of state. *Fluid Phase Equilibria* **2013**, *352*, 93-96.