Insertion of CO₂ and CS₂ into Bi–N Bonds Enables Catalyzed CH-Activation and Light-Induced Bismuthinidene Transfer

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Experimental

General considerations

All air and moisture-sensitive manipulations were carried out using standard Schlenk techniques or in a glovebox containing purified argon. Solvents were purified by distillation using the appropriate drying agents, degassed and stored over molecular sieves (4 Å) prior to use. Deuterated solvents used for NMR spectroscopy were dried, degassed and stored over molecular sieves (4 Å) under dry argon prior to use. BiCl₃ was sublimed prior to use. CS₂ was degassed by three freeze-pump-thaw cycles and stored over molecular sieves (4 Å) under dry argon. Compounds $1-H^1$ and $1-Me^2$ were prepared according to the literature.

All NMR spectra were acquired either on a Bruker Avance 400 spectrometer or on Bruker Avance I/III 500 spectrometer. ¹H and ¹³C chemical shifts are reported relative to SiMe₄ using the residual solvent peak of the solvent as a secondary standard. ¹⁹F chemical shifts are reported relative to CFCl₃ as an external standard. Elemental analyses (C, H, N, S) were conducted on Vario Micro Cube intruments by Elementar Analysesysteme GmbH. Mass spectrometric analyses were conducted using either an AccuTOF GCv by JEOL (LIFDI) or a LTQ-FT Ultra by Thermo Fischer Scientific (APCI). UV-vis spectra were recorded on a Specord S600 diode array spectrometer by AnalytikJena. Photochemical experiments were performed using a low-pressure mercury-vapor lamp by Peschl Ultraviolet (*P* = 2.7 W, λ = 254 nm) and regular borosilicate glassware.

Single-crystals suitable for X-ray diffraction analysis were coated with perfluorinated polyether oil in a glovebox, transferred to a nylon loop and then transferred to the goniometer of a diffractometer (Bruker D8 Quest or Bruker X8-Apex II) equipped with a molybdenum X-ray tube ($\lambda = 0.71073$ Å).

[Bi(N(C₆H₅)(C₆H₄)(C(O)O))(OTf)] (2-H). [Bi₂(N(C₆H₅)(C₆H₄))₂(OTf)₂(thf)₃] (1-H) (217 mg, 0.17 mmol) was dispersed in THF (1 mL) and the atmosphere was exchanged for CO₂ (1 atm) in three freeze-pump-thaw cycles.³ A white solid started to precipitate immediately. After 30 h, when all orange solid was consumed and the reaction mixture was colorless, the solid was isolated by filtration, washed with THF and pentane and dried *in vacuo*. The product contains 0.5 equiv. THF which could not be removed under reduced pressure.

Yield: 180 mg (0.30 mmol, 87%)

¹**H NMR:** (298 K, 500 MHz, Acetonitrile-*d*₃): δ = 1.77 – 1.83 (m, 2H, β-thf), 3.63 – 3.67 (m, 2H, α-thf), 6.38 (s, br, 2H, 2,6-C₆H₅), 7.16 – 7.23 (m, 3H, 3-C₆H₄, 3,5-C₆H₅), 7.34 (d, ³*J*_{HH} = 8.2 Hz, 1H, 5-C₆H₄), 7.38 (tt, ³*J*_{HH} = 7.6 Hz, ⁴*J*_{HH} = 1.1 Hz, 1H, 4-C₆H₅), 7.67 (ddd, ³*J*_{HH} = 8.2 Hz, ³*J*_{HH} = 7.4 Hz, ⁴*J*_{HH} = 1.0 Hz, 1H, 4-C₆H₄), 8.83 (dd, ³*J*_{HH} = 7.4 Hz, ⁴*J*_{HH} = 1.2 Hz, 2-C₆H₄) ppm.

¹³**C NMR:** (298 K, 126 MHz, Acetonitrile-*d*₃): δ = 26.2 (s, β-thf), 68.4 (s, α-thf), 121.3 (q, ¹*J*_{CF} = 320 Hz, OTf), 127.4 (s, 2-C₆H₄), 128.9 (s, 4-C₆H₅), 129.3 (s, 2,6-C₆H₅), 129.8 (s, 4-C₆H₄), 130.7 (s, 3,5-C₆H₅), 131.2 (s, 3-C₆H₅), 137.7 (s, 5-C₆H₄), 141.9 (s, 1-C₆H₅), 149.4 (s, 1-C₆H₄), 160.4 (s, C=O), 223.4 (s, 6-C₆H₄) ppm. ¹⁹**F NMR:** (298 K, 471 MHz, Acetonitrile-*d*₃): δ = -78.96 (s, OTf) ppm.

ESI-MS (Pyridine): positive mode, found m/z 376.0529, calcd. for C₁₂H₉BiN⁺ m/z 376.0539. The detected fragment represents the monomeric species after elimination of CO₂ and [OTf]⁻ during the ionization process.

ESI-MS (Pyridine): negative mode, found m/z 673.9568, calcd. for $C_{14}H_9BiF_6NO_6S_2^- m/z$ 673.9584. The detected fragment represents the monomeric species after elimination of CO_2 and association with $[OTf]^-$ during the ionization process.

Elemental analysis. Anal. calc. for C₁₄H₉BiF₃NO₅S × 0.5 [C₄H₈O] (605.32 g/mol): C, 31.75; H, 2.16.; N, 2.31; S, 5.30; found: C, 31.90; H, 2.20; N, 2.26; S, 5.30.

[Bi(N(C₆H₄Me)(C₆H₃Me)(C(O)O))(OTf)] (2-Me). [Bi₂(N(C₆H₄Me)(C₆H₃Me))₂(OTf)₂(thf)] (1-Me) (120 mg, 0.11 mmol) was dissolved in THF (3 mL) and the atmosphere was exchanged for CO₂ (1 atm) in three freeze-pump-thaw cycles. The reaction mixture was stirred at ambient temperature overnight and the color changed to pale yellow. The solvent was removed under reduced pressure, and the residue was washed with pentane (3 x 5 mL). The product was obtained as white powder after drying *in vacuo* for 5 h.

Yield: 105 mg (0.18 mmol, 92%)

¹**H NMR:** (298 K, 500 MHz, Acetonitrile- d_3): δ = 2.36 (s, 3H, C₆H₃*Me*), 2.41 (s, 3H, C₆H₄*Me*), 6.36 (s, br, 2H, 2,6-C₆H₄Me), 6.98 – 7.02 (m, 3H, 3,5-C₆H₄Me, 3-C₆H₃Me), 7.32 (d, 1H, ³*J*_{HH} = 8.5 Hz, 2-C₆H₃Me), 8.62 (s, 1H, 5-C₆H₃Me) ppm.

¹³**C NMR**: (298 K, 126 MHz, Acetonitrile- d_3): δ = 21.2 (s, C₆H₃*Me*), 21.3 (s, C₆H₄*Me*), 121.1 (q, ¹*J*_{CF} = 319 Hz, OTf), 127.1 (s, 2-C₆H₃Me), 129.2 (s, 2,6-C₆H₄Me), 131.0 (s, 3,5-C₆H₄Me), 131.8 (s, 3-C₆H₃Me), 138.0 (s, 5-C₆H₃Me), 138.9 (s, 5-C₆H₄Me), 139.3 (s, 1-C₆H₄Me), 139.6 (s, 4-C₆H₃Me), 146.9 (s, 1-C₆H₃Me), 160.5 (s, C=O), 222.7 (s, 6-C₆H₃Me) ppm.

¹⁹**F NMR:** (298 K, 471 MHz, Acetonitrile-*d*₃): *δ* = – 78.89 (s, OTf) ppm.

Elemental analysis. Anal. calc. for C₁₆H₁₃BiF₃NO₅S (597.32 g/mol): C, 32.17; H, 2.19; N, 2.34; S, 5.37; found: C, 32.31; H, 2.37; N, 2.15; S, 5.14.

[Bi(N(C₆H₅)(C₆H₄)(C(S)S))(OTf)(diglyme)] (3-H). [Bi₂(N(C₆H₅)(C₆H₄))₂(OTf)₂(thf)₃] (1-H) (630 mg, 0.50 mmol) was dissolved in diglyme (5 mL), cooled to 0 °C and CS₂ (3.5 mL) was added. The reaction mixture was stirred at 0 °C for 1.5 h and a yellow solid started to precipitate. The reaction mixture was layered with *n*-hexane (5 mL) and the reaction vessel was stored at -30 °C for 3 d, and a yellow solid precipitated. It was isolated by filtration and washed with *n*-hexane (15 mL). The obtained product was dried for 2 h *in vacuo*. The number of diglyme ligands *n* in the bulk material ranges from 1.0 to 2.0 and must be determined for every batch individually.

Yield (*n* = 1): 654 mg (0.88 mmol, 89%)

¹**H NMR:** (298 K, 500 MHz, Pyridine- d_5): δ = 3.26 (s, 6H, CH₃, diglyme) 3.49 – 3.54 (m, 4H, CH₂, diglyme), 3.61 – 3.65 (m, 4H, CH₂, diglyme), 7.33 (dd, 1H, ³ J_{HH} = 7.8 Hz, ³ J_{HH} = 8.5 Hz, 3-C₆H₄), 7.50 (dd, 1H, ³ J_{HH} = 7.4 Hz, ³ J_{HH} = 8.5 Hz, 4-C₆H₄), 7.96 (d, 1H, ³ J_{HH} = 8.2 Hz, 5-C₆H₄), 8.36 (d, br, 1H, ³ J_{HH} = 8.5 Hz, 2-C₆H₄) ppm. The resonances of the phenyl group show chemical shifts between 7.10 and 7.55 ppm and could not to be assigned to individual positions due to severe signal broadening. When the number of solvent protons and the number of the protons of the phenylene unit are subtracted from the sum of the integrals of the relevant section of the spectrum (isolated doublet as calibration signal), the expected value of five is obtained, corresponding to the five protons of the phenyl group.

¹³**C** NMR: (298 K, 126 MHz, Pyridine- d_5): δ = 58.9(s, CH₃, diglyme), 71.0 (s, CH₂, diglyme), 72.6 (s, CH₂, diglyme), 122.8 (s, ¹*J*_{CF} = 322 Hz, OTf), 129.8 (s, C₆H₅), 130.0 (s, 3-C₆H₄, 130.4 (s, C₆H₅), 130.6 (3-C₆H₄), 131.3 (s, C₆H₅), 131.9 (s, 4-C₆H₄), 138.1 (s, 2-C₆H₄), 146.4 (s, 1-C₆H₅), 152.5 (s, 6-C₆H₄), 201.0 (s, *C*S₂), 210.0 (s, 1-C₆H₄) ppm. *Due to the signal broadening of the signals of the phenyl group in the* ¹*H NMR spectrum, the assignment of the exact position of the carbon atoms of the phenyl group at 129.8, 130.4 and 131.3 ppm was not possible*.

¹⁹**F NMR:** (298 K, 471 MHz, Pyridine-*d*₅): *δ* = – 77.29(s, OTf) ppm.

ESI-MS (THF): positive mode, found m/z 451.9970; 903.9970, 1052.9474 calcd. for $C_{13}H_9BiNS_2^+ m/z$ 451.9980; for $C_{26}H_{18}Bi_2N_2S_4^+ m/z$ 903.9960; for $C_{27}H_{18}Bi_2F_3N_2O_3S_5^+ m/z$ 1052.9475. The three signals correspond to the monomeric fragment without $[OTf]^-$, the dimeric fragment without $[OTf]^-$, and the dimeric fragment with one $[OTf]^-$ anion.

Elemental analysis. Anal. calc. for $C_{14}H_9Bi_1F_3NO_3S_3 \times [C_6H_{14}O_3]$ (735.56 g/mol): C, 32.66; H, 3.15; N, 1.90; S, 13.08; found: C, 32.50; H, 3.52; N, 1.95; S, 13.01.

[Bi(N(C₆H₄Me)(C₆H₃Me)(C(S)S))(OTf)(diglyme)₂] (3-Me). [Bi₂(N(C₆H₄Me)(C₆H₃Me))₂(OTf)₂(thf)] (1-Me) (550 mg, 0.47 mmol) was dissolved in diglyme (6 mL) and CS₂ (3.00 mL) was added. The reaction mixture was stirred at room temperature for 45 min and layered with *n*-hexane (6 mL). The reaction vessel was stored at -30 °C for 3 d, and yellow crystals formed. These were isolated by filtration and washed with *n*-hexanes (3 x 15 mL). The obtained product was dried under exclusion of light for 4 h *in vacuo*. The number of diglyme ligands *n* in the bulk material ranges from 1.0 to 2.0 and must be determined for every batch individually.

Yield (n = 2): 603 mg (0.68 mmol, 72%)

¹**H NMR:** (298 K, 500 MHz, Pyridine- d_5): δ = 2.11 (s, 3H, C₆H₃*Me*), 2.21 (s, 3H, C₆H₄*Me*), 3.27 (s, 12H, CH₃, diglyme) 3.50 – 3.54 (m, 8H, CH₂, diglyme), 3.62 – 3.65 (m, 8H, CH₂, diglyme), 6.95 (s, br, 2H, 3,5-C₆H₄Me), 7.18 (d, ³*J*_{HH} = 8.4 Hz, 1H, 3-C₆H₃Me), 7.20 – 7.36 (s, br, 2H, 2,6-C₆H₄Me) 7.97 (d, ³*J*_{HH} = 8.4 Hz, 1H, 2-C₆H₃Me), 8.21 (s, 1H, 5-C₆H₃Me) ppm.

¹³**C NMR:** (298 K, 126 MHz, Pyridine-*d*₅): δ = 21.0 (s, C₆H₃*Me*) 21.3 (s, C₆H₄*Me*) 59.0 (s, CH₃, diglyme), 71.1 (s, CH₂, diglyme), 72.6 (s, CH₂, diglyme), 122.8 (s, ¹*J*_{CF} = 320 Hz, OTf), 129.1 (s, 2,6-C₆H₄Me) 130.9 (s, 3,5-C₆H₄Me, 2-C₆H₃Me), 132.1 (s, 3-C₆H₃Me), 138.5 (s, 5-C₆H₃Me), 139.8 (s, 4-C₆H₄Me), 142.2 (s, 4-C₆H₃Me), 143.9 (s, 1-C₆H₄Me), 150.3 (s, 1-C₆H₃Me), 200.2 (s, *C*S₂), 208.1 (s, 6-C₆H₃Me) ppm.

The singal at 150.3 ppm overlaps with the residual solvent signal of pyridine- d_5 and was assigned on the basis of the ¹H/¹³C HMBC spectrum.

¹⁹**F NMR:** (298 K, 471 MHz, Pyridine-*d*₅): δ = – 77.29 (s, OTf) ppm.

Elemental analysis. Anal. calc. for C₁₆H₁₈BiF₃NO₃S₃ × 2 [C₆H₁₄O₃] (897.79 g/mol): C, 37.46; H, 4.60; N, 1.56; S, 10.71; found: C, 37.64; H, 4.15; N, 1.64; S, 11.05.

UV-vis spectra of 3-R:



Figure S1. UV-vis spectra of compounds 3-R in THF.

Preparation of 3-phenylbenzo[d]thiazole-2(3H)-thione (5-H):

[Bi(N(C₆H₅)(C₆H₄)(C(S)S))(OTf)(diglyme)] (**3-H**) (270 mg, 0.37 mmol) was dissolved in THF (8 mL) and irradiated at 254 nm with a low-pressure Hg lamp for 5 h. The yellow reaction mixture turned orange and a black solid precipitated. The solvent was removed under reduced pressure, and the residue was extracted with dichloromethane. The combined organic phases were washed with water (3 x 15 mL) and dried over Mg(SO₄). The solvent was removed under reduced pressure and the crude product purified by column chromatography (DCM, 8% ethanol). After removal of the solvent, pure benzothiazolethione **5-H** was obtained as a yellow oil.

Yield: 58.4 mg (0.24 mmol, 65%)

¹**H NMR:** (298 K, 400 MHz, CDCl₃): δ = 6.73 (d, ³*J*_{HH} = 7.3 Hz, 1H), 7.28-7.33 (m, 2H), 7.39 (d, ³*J*_{HH} = 8.4 Hz, 2H), 7.51 (d, ³*J*_{HH} = 8.2 Hz, 1H), 7.59 (d, ³*J*_{HH} = 7.2 Hz, 1H), 7.62-7.67 (m, 2H) ppm.⁴ APCI-MS: positive mode, found *m/z* 244.0248, calcd. for C₁₃H₁₀NS₂⁺ *m/z* 244.0255.

Preparation of 3-(4-methylphenyl)-6-methylbenzo[d]thiazole-2(3H)-thione (5-Me):

 $[Bi(N(C_6H_4Me)(C_6H_3Me)(C(S)S))(OTf)(diglyme)_2]$ (**3-Me**) (300 mg, 0.34 mmol) was dissolved in THF (8 mL) and irradiated at 254 nm for 5 h. The yellow reaction mixture turned orange and a black solid precipitated. The solvent was removed under reduced pressure and the residue was extracted with dichloromethane and washed with water (3 x 15 mL). After drying over Mg(SO₄), the solvent was removed and the crude product purified by column chromatography (DCM, 5% ethanol). After removal of the solvent, pure benzothiazolethione **5-Me** was obtained as yellow powder.

Yield: 73.4 mg (0.27 mmol, 81%)

¹**H NMR:** (298 K, 500 MHz, Benzene- d_6): δ = 1.95 (s, 3H, C₆H₃*Me*), 2.02 (s, 3H, C₆H₄*Me*), 6.32 (d, ³*J*_{HH} = 8.4 Hz, 1H, 6-C₆H₃Me), 6.44 (s, 1H, 3-C₆H₃Me), 6.59 (d, ³*J*_{HH} = 8.4 Hz, 1H, 5-C₆H₃Me), 6.87 (d, ³*J*_{HH} = 8.4 Hz, 2H, 2,6-C₆H₄Me), 6.90 (d, ³*J*_{HH} = 8.4 Hz, 2H, 3,5-C₆H₄Me) ppm.

¹³**C NMR:** (298 K, 126 MHz, Benzene- d_6): δ = 20.8 (s, C₆H₃*Me*), 21.1 (s, C₆H₄*Me*), 113.0 (s, 6-C₆H₃Me), 121.4 (s, 3-C₆H₃Me), 127.5 (s, 5-C₆H₃Me), 128.0 (s, 2-C₆H₃Me), 128.5 (s, 2,6-C₆H₄Me), 130.7 (s, 3,5-C₆H₄Me), 134.4 (s, 4-C₆H₃Me), 134.9 (s, 1-C₆H₄Me), 139.3 (s, 4-C₆H₄Me), 142,1 (s, 1-C₆H₃Me), 190.1 (s, C=S) ppm.

APCI-MS: positive mode, found *m/z* 272.0564, calcd for C₁₅H₁₄NS₂⁺ *m/z* 272.0567.

Trapping of Bi^I(OTf) to yield compound 7

 $[Bi(N(C_6H_5)(C_6H_4)(C(S)S))(OTf)(diglyme)]$ (3-H) (220 mg, 0.15 mmol) and 3,5-di-*tert*-butyl-obenzoquinone (6) (62 mg, 0.28 mmol) were dissolved in THF (5 mL) and irradiated at 254 nm for 16 h. Traces of a black precipitate formed and the reaction mixture turned orange. The solvent was removed and the residue was washed with pentane (3 x 8 mL). THF (1.5 mL) was added, the suspension containing minor amounts of a dark solid was filtered, and the filtrate was layered with *n*-hexanes (0.5 mL) and stored at -30 °C for two weeks. Crystals had formed, were isolated by filtration, washed with pentane (3 x 5 mL), dried *in vacuo* for 3 h to give compound 7 with two equiv. thf per bismuth atom, which could not be removed under reduced pressure.

Yield: 91 mg (0.13 mmol, 42%)

¹H NMR spectroscopic reaction monitoring indicated the formation of **7** in quantitative spectroscopic yield after 16 h reaction time.

¹**H NMR:** (298 K, 500 MHz, THF-*d*₈): δ = 1.28 (s, 9H, 3-*t*Bu), 1.45 (s, 9H, 5-*t*Bu), 1.75 – 1.81 (m, 8H, β-THF), 3.60 – 3.66 (m, 8H, α-THF), 6.53 (d, ⁴*J*_{HH} = 2.23 Hz, 1H, 6-*CH*), 6.58 (d, ⁴*J*_{HH} = 2.23 Hz, 1H, 4-*CH*) ppm.

¹³**C NMR:** (298 K, 126 MHz, THF-*d*₈): δ = 26.4 (s, β-THF), 30.3 (s, *CH*₃ (5-*t*Bu)), 32.5 (s, *CH*₃ (3-*t*Bu), 34.6 (s, *C*_q (3-*t*Bu)), 35.1 (*C*_q (5-*t*Bu), 68.3 (s, α-THF), 114.0 (s, 4-C₆H₂(*t*Bu)₂) 119.2 (s, 6-C₆H₂(*t*Bu)₂), 121.4 (q, ¹*J*_{CF} = 320 Hz, OTf), 141.8 (s, 3-C₆H₂(*t*Bu)₂), 142.3 (s, 5-C₆H₂(*t*Bu)₂), 153.8 (s, 2-C₆H₂(*t*Bu)₂), 157.8 (s, 1-C₆H₂(*t*Bu)₂) ppm.

¹⁹**F NMR:** (298 K, 282 MHz, THF-*d*₈): *δ* = -79.26 (s, OTf) ppm.

LIFDI-MS: positive mode, found *m*/*z* 578.0791, calcd. for C₁₅H₂₀BiF₃O₅S⁺ *m*/*z* 578.0787.

Elemental analysis (the sample was dried in a stream of argon). Anal. calc. for C₁₅H₂₀BiF₃O₅S × 3 [C₄H₈O] (794.68 g/mol): C, 40.81; H, 5.58; S, 4.03; found: C, 40.56; H, 5.51; S, 3.64.

Reaction monitoring: formation of 5-H from 3-H

General procedure: a defined amount of **3-H** was dissolved in THF-d₈ (0.5 mL) and the progress of the reaction was monitored by ¹H NMR spectroscopy. The reaction conditions are summarized in Table S1.

#	Conditions	Concentration	Full Conversion to 5-H
1	THF, rt	4.08 [·] 10 ⁻³ mmol/mL	5d
2	THF, rt	13.6 [.] 10 ⁻³ mmol/mL	5d
3	THF, rt	47.6 [·] 10 ⁻³ mmol/mL	5d
4	THF, rt, dark	13.6 [.] 10 ⁻³ mmol/mL	Stable
5	THF, 60 °C, dark	13.6 [.] 10 ⁻³ mmol/mL	Stable
6	THF, hv	13.6 [·] 10 ⁻³ mmol/mL	45 min

Table S1. Overview of the conducted reactions to assess the different factors of the transformation of **3-H** to yield **5-H**.

NMR spectroscopic data of 5-H has been previously described.⁴

Test reactions of 4-H with 6:

NMR spectroscopic monitoring of reactions of equimolar amounts of **4-H** and **6** in THF indicated coordination of **6** in the dark. After irradiation of the reaction mixture with a low-pressure Hg lamp for up to 16 h, no reaction could be observed, except for minor decomposition reactions. The formation of **7** and carbozole (as the corresponding C–C coupling product) could not be detected. This is in line with Bi–S bonds showing a higher tendency towards photochemical activation than Bi–C bonds.⁵

Investigations of Reactions under Catalytic Conditions: Transformation of 1-R to 4-R

Blind measurement:

In a J. Young-type NMR tube, **1-H** (20.0 mg, 15.8 μ mol) was dissolved in THF- d_8 (0.5 mL) and was heated to 60 °C for 4 d. The ¹H NMR spectrum revealed that **1-H** is stable under these conditions and the formation of **4-H** was not observed.

Blind measurement:

In a J. Young-type NMR tube, **2-H** (13.0 mg, 0.02 mmol) was dispersed in THF- d_8 (0.5 mL) and was heated to 60 °C for 14 d. A yellow reaction solution was formed and **4-H** was identified by ¹H NMR spectroscopy with near-quantitative yield.

Reaction under catalytic conditions:

In a J. Young-type NMR tube, THF- d_8 (0.5 mL) was added to **1-H** (20.0 mg, 15.8 µmol) and **2-H** (0.9 mg, 1.58 µmol, 10 mol%) and the reaction mixture was heated to 60 °C for 4 d. **4-H** was spectroscopically identified with 91% yield.

To **1-H** (20.0 mg, 15.8 μ mol) and **2-H** (0.09 mg, 0.16 μ mol, 1 mol%) THF- d_8 (0.5 mL) was added and the reaction mixture was heated to 60 °C for 17 d. **4-H** was spectroscopically identified with 84% yield. Full conversion of **1-H** was not achieved at this point.

Single-Crystal X-ray Analyses

Deposition Numbers 2223276-2223278 contain the supplementary crystallographic information for this work. These data are provided free of charge by the joint Cambridge Crystallographic Data Centre and Fachinformationszentrum Karlsruhe Access Structures service under <u>www.ccdc.cam.ac.uk/structures</u>.

Compound 2-H

Compound 2-H crystallizes from a saturated solution in acetonitrile which was layered with diethyl ether at -30 °C in the orthorhombic space group $P2_12_12_1$ with Z = 4. Its molecular structure in the solid state is depicted in the main part Scheme 1b, and to further clarify the bonding situation also in Figure S2. The discussion of the bonding modes and parameters is presented in the main part. However, it should be noted that the distances between Bi1 and O3 (2.976(5) Å), Bi2 and O5 (2.939(5) Å), as well as Bi3 and O1 (3.029(4) Å) fulfill the distance criteria for Bi–O bond formation (, $\Sigma r_{van der Waals}$ (Bi, O) = 3.59 Å),⁶ which could thus be taken into account. In this case, the bismuth centers Bi1 and Bi2 would have a coordination number of six and adopt a distorted octahedral coordination geometry, while Bi3 would show a coordination number of seven (as recently reported for dihalo bismuth cations⁷) and a strongly distorted pentagonal bipyramidal coordination geometry. The CO₂ moieties would show a bridging μ_2 - η^2 : η^1 coordination mode. We argue that this coordination mode is less likely to be present in this compound, and the proximity of Bi1/O3, Bi2/O5, and Bi3/O1 is rather a result of geometric constraints than of tight binding interactions in this compound, because the chelating bonding mode of the CO₂ moieties would be highly unsymmetric. Thus, while a μ_2 - η^2 : η^1 coordination mode can formally not be ruled out, we prefer the description in terms of μ_2 - η^1 : η^1 coordination of the CO₂ moieties in **2-H**. All the Bi–C bonds in **2-H** are in the same range (\emptyset = 2.203 Å) and thus equal within limits of error to those in **1-H** (2.207(4) Å).¹ In contrast to **3-Me**, insertion of CO₂ does not result in the elongation of the Bi–C bond (*vide infra*).



Figure S2. Representation of the mononuclear subunit **2-H** in the solid state. Displacement ellipsoids are drawn at the 50% probability level. Hydrogen atoms are omitted for clarity and atoms exceeding one subunit are drawn as white ellipsoids.

Compound 2-Me

Compound **2-Me** crystallizes from a saturated diglyme solution which was layered with *n*-hexanes at – 30 °C in the monoclinic space group $P2_1$ with Z = 2. The unit cell parameters are: a = 10.178(4) Å, b = 27.160(8) Å, c = 13.043(4) Å, $\alpha = 90^\circ$, $\beta = 106.577(15)^\circ$, $\gamma = 90^\circ$, V = 3456(2) Å³. The quality of the obtained data is not sufficient for a detailed discussion of the bonding parameters but provides a proof of connectivity. The molecular structure in the solid state is shown in Figure S3. The CO₂ moieties are probably best described in a μ_2 - η^1 : η^1 coordination mode, as observed and discussed for **2-H**. Compound

2-Me exists similarly to **3-H** as a dinuclear complex, but the coordination modes of the CE₂ units differ ($(\mu_2 - \eta^1: \eta^1 \text{ in } 2-H \text{ and } 2-Me$, but terminal $\eta^1: \eta^1$ in **3-H**). Due to one bridging triflate unit, both bismuth atoms adopt a distorted pentagonal bipyramidal coordination geometry and are coordinatively saturated by one diglyme ligand each. The second triflate unit does not interact directly with any bismuth atom and can be considered as weakly coordinating anion.



Figure S3. Molecular structure of compound **2-Me** in the solid state. Displacement ellipsoids are drawn at the 50% probability level. Hydrogen atoms are omitted for clarity, diglyme ligands are depicted in the wireframe model.

Compound 3-H

Compound **3-H** crystallizes from a saturated dimethoxyethane (DME) solution, which was layered with *n*-hexane at ambient temperature, in the triclinic space group $P\overline{1}$ with Z = 2. The unit cell parameters are: a = 10.2847(6) Å, b = 14.4870 (8) Å, c = 16.8468(9) Å, $\alpha = 69.701(2)^\circ$, $\beta = 79.652(2)^\circ$, $\gamma = 89.710(2)^\circ$, V = 2311.4(2) Å³. The quality of the obtained data is not sufficient for detailed discussion of the bonding parameters, but provides a proof of connectivity. The molecular structure of **3-H** in the solid state is shown in Figure S4. It reveals a tetranuclear complex with bridging μ_2 - η^1 : η^1 -coordinated CS₂ units between Bi1 and Bi2 and bridging triflate ligands between Bi1 and Bi1' as well as Bi2 and Bi2'. All bismuth centers adopt a square pyramidal coordination geometry.



Figure S4. Molecular structure of **3-H** in the solid state. Displacement ellipsoids are drawn at the 50% probability level. Hydrogen atoms are omitted for clarity, DME ligands and phenyl groups are depicted as wireframe.

Compound 3-Me

Compound **3-Me** crystallizes from a saturated diglyme solution, which was layered with *n*-hexane at – 30 °C, in the orthorhombic space group *Pnna* with Z = 4, and its molecular structure in the solid state is shown in the main part (Scheme 1c). The unit cell contains multiple disordered molecules of diglyme and *n*-hexane (in superposition) which have been treated as a diffuse contribution to the overall scattering without specific atom positions by Squeeze/Platon. A detailed discussion of the structural parameters is presented in the main part.

Compound 7

Compound **7** crystallizes from a THF solution, which was layered with *n*-hexane at -30 °C, over the period of two weeks in the monoclinic space group $P2_1/c$ with Z = 2. The unit cell parameters are: a = 10.2720(19) Å, b = 17.948(3) Å, c = 16.609(3) Å, $\alpha = 90^\circ$, $\beta = 105.482(7)^\circ$, $\gamma = 90^\circ$, V = 2950.9(9) Å³. The quality of the obtained data is not sufficient for detailed discussion of the bonding parameters, but provides a proof of connectivity. The molecular structure in the solid state is shown in Figure S5 and reveals a dinuclear complex. The bismuth atom Bi1 adopts a distorted octahedral coordination geometry. The hydroquinolate moieties act as bridging ligands and appear to be in plane with the two bismuth atoms and the oxygen atoms of two thf ligands. A second thf ligand and a triflate unit bound in the axial positions saturate the coordination sphere of each bismuth atom.



Figure S5. Molecular structure of compound **7** in the solid state. Displacement ellipsoids are drawn at the 50% probability level. Hydrogen atoms and lattice bound THF are omitted for clarity, THF ligands are depicted in the wireframe model.

Compound 4-Me

4-Me crystallized from a THF solution layered with *n*-hexane at -30 °C in the monoclinic space group $P2_1/c$ with Z = 4. The molecular structure of a compound closely related to **4-Me** has previously been described: in the previous example, two pyridine ligands were bound to the bismuth center instead of the two thf ligands that are observed in **4-Me**.² This has no significant effect on the bonding parameter and the coordination geometry around bismuth: Bi1 adopts a bisphenoidal coordination geometry with the two thf ligands in the axial positions (O4–Bi1–O5, 167.61(10)°) and the triflate anion forms a hydrogen bond with the NH moiety, while showing a weak interaction with the bismuth center of an adjacent formula unit (not displayed in Figure S6).



Figure S6. Molecular structure of compound **4-Me** in the solid state. Displacement ellipsoids are drawn at the 50% probability level. Hydrogen atoms but H1 are omitted for clarity, THF ligands are depicted as wireframe. Selected bond lengths [Å] and angles [°]: Bi1–C1, 2.207(4); Bi1–C8, 2.190(4); C1–C2, 1.397(6); C8–C9, 1.408(5); H1…O1, 2.190; C1–Bi1–C8, 90.21(15); C1–Bi1–O4, 84.25(12); C1–Bi1–O5, 85.23(13); O4–Bi1–O5, 167.61(10).

NMR Spectra of Isolated Compounds



Figure S7. ¹H and ¹³C NMR spectra of $[Bi(N(C_6H_5)(C_6H_4)(C(O)O))(OTf)]$ (2-H) in acetonitrile- d_3 .



Figure S8. ¹⁹F NMR spectrum of $[Bi(N(C_6H_5)(C_6H_4)(C(O)O))(OTf)]$ (2-H) in acetonitrile- d_3 .



Figure S9. ¹H and ¹³C NMR spectra of $[Bi(N(C_6H_4Me)(C_6H_3Me)(C(O)O))(OTf)]$ (2-Me) in acetonitrile- d_3 .



Figure S10. ¹⁹F NMR spectrum of $[Bi(N(C_6H_4Me)(C_6H_3Me)(C(O)O))(OTf)]$ (**2-Me**) in acetonitrile- d_3 .



Figure S11. ¹H and ¹³C NMR spectra of $[Bi(N(C_6H_5)(C_6H_4)(C(S)S))(OTf)(diglyme)]$ (3-H) in pyridine- d_5 .



Figure S12. ¹⁹F NMR spectrum of $[Bi(N(C_6H_5)(C_6H_4)(C(S)S))(OTf)(diglyme)]$ (3-H) in pyridine- d_5 .



pyridine-d₅.



Figure S14. ¹⁹F NMR spectrum of $[Bi(N(C_6H_4Me)(C_6H_3Me)(C(S)S))(OTf)(diglyme)_2]$ (3-Me) in pyridine- d_5 .



Figure S15. ¹H and ¹³C NMR spectra of the benzothiazolethione **5-Me** in benzene- d_6 .



Figure S16. ¹H and ¹³C NMR spectra of compound 7 in THF- d_8 .



Figure S17. ¹⁹F NMR spectrum of compound **7** in THF- d_8 .

Diffusion-Ordered NMR Spectroscopy

Samples for DOSY measurements were prepared by dissolving 4 mg of the compounds of interest (**2-H** or **3-H**) in 0.5 mL of a deuterated solvent (see Table S2). NMR spectroscopic measurements were performed on a Bruker AVIII 500 instrument with a cryo probe Prodigy BB-H&F with z-gradient at 298 K. The temperature was calibrated with a Bruker standard (4% methanol in methanol- d_4). DOSY experiments were done using pulse sequences with a long delay for the recovery of the eddy current and bipolar gradients, and with the method of double stimulated echo for compensating the effect of thermal convection on diffusion.⁸ Pulse sequences were ledbpgp2s and dstegp3s from the Bruker pulse program library. The gradient shape used was smoothed square SMSQ10.100 and the gradient length was between 2 and 3 ms. The optimized diffusion delays (Δ) were between 30 and 50 ms. Diffusion attenuation was realized with 9 steps in gradient ramp. The DOSY 2D spectra were recorded with 9 experiments and each with 16 to 32 transients. Relaxation delays were 3 s.

Based on the experimentally determined diffusion coefficients, the hydrodynamic radii of **2-H** and **3-H** were calculated. Through a comparison between the experimentally determined hydrodynamic radii and those obtained by the method of hard-sphere increments,⁹ the nuclearity of the compounds was estimated. Mononuclear model compounds were used as a starting point and the results are summarized in Table S2.

Compound	Solvent	Monomer (model species)	D [×10 ⁻⁹ m ² ·s ⁻¹]	<i>R</i> н [Å]	R _H ^{HSI} [Å]	Oligmer (n)
2-H	Acetonitrile	$[Bi(NC_6H_4Ph)CO_2(MeCN)_2(O_3SCF_3)]$	0.85	6.9	4.7	trimer(3.2)
3-H	THF	$[Bi(NC_6H_4Ph)CS_2(THF)_2(O_3SCF_3)]$	0.65	7.0	4.9	trimer (2.8)

Table S2. Diffusion coefficients, hydrodynamic radii and extent of aggregation of 2-H and 3-H 298 K.

a: When pyridine was used as the solvent, two species were detected in the ¹H NMR spectrum, which were also separated in the diffusion dimension.

Quantum Chemical Analyses

All DFT calculations were performed with the Amsterdam Density Functional (ADF) program¹⁰ using relativistic, dispersion-corrected density functional theory (DFT) at the ZORA-BLYP-D3BJ/TZ2P level of theory for geometry optimizations and energy calculations, with the full electron model for all atoms (no frozen core).¹¹ Solvation in THF and pyridine was simulated by using the conductor-like screening model (COSMO).¹² All stationary points were verified to be minima on the potential energy surface through vibrational analysis.

Table S3. Cartesian coordinates and ADF total electronic energy (in parentheses and in kcal mol⁻¹) of compounds **2-H** and **3-H** and their dimeric species under analysis computed at the ZORA-BLYP-D3(BJ)/TZ2P level of theory in pyridine.

2-H (-8065.7)

(`		• /		
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10		14.93990292	10.50555415	3.41160210
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45	C	0.14520575	3 67037743	1 50231/00
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40	U T	10.38081061	2.01293093	1.39484337
47	Н	10.13357791	1.6/33/653	1.10656736
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54	С	15.29313798	0.94925214	3.58750477
55	Н	15.30696719	0.03246123	4.17076582
56	С	16.22918603	1.15978670	2.57253199
57	С	16.21698381	2.36166797	1.85267729
58	Н	16.97247076	2.51282451	1.08475166
59	C	15.26305605	3.36055359	2.11135959
60	Ĥ	16.97705940	0.40218104	2.34860114
61	0	18 08984650	5 19928770	1 70345736
62	Š	18 92398640	4 71308374	0 57228458
63	õ	20 15833726	4 01963331	0.95819463
64	F	20.15055720	7 05237157	0.55746208
65	Ċ	19 52128015	6 32722814	-0.22/138923
66	õ	18 12032715	4 07530503	0.40743082
67	E	18.12032713	7.08202201	0.61211277
69	Г	10.43391347	6.07449320	1 21854622
60	Г N	20.26367032	6 29016059	-1.51654052
09 70	N	17.13030041	0.38010938	0.15000512
70	C	1/.86/3186/	6.19560663	1.20053390
71	C	17.71182913	6.2/8/0663	4.93263761
72	C	19.22/1233/	5.90542473	7.20515224
73	C	19.06388703	5.9/29/413	4.80059905
74	С	19.83447192	5.78634880	5.95116207
75	Н	17.34432602	6.27365598	8.21575291
76	Η	17.07466663	6.43789396	4.06892381
77	Η	19.79001231	5.76826335	8.12304764
78	Η	19.48321257	5.87500688	3.80485247
79	Η	20.89154156	5.54763465	5.87390494
80	Ν	15.21024829	3.90325955	7.20409490
81	С	14.17312663	3.29429174	7.81035200
82	С	16.20205863	3.13198361	6.71951111
83	С	14.08879056	1.90770018	7.96180390
84	С	16.20885167	1.74196372	6.83805980
85	С	15.13097102	1.11689587	7.47082441
86	Η	13.38217527	3.94307563	8.18465516
87	Н	17.00934007	3.64866907	6.20678573
88	Н	13.22702211	1.46547197	8.45400086
89	Н	17.03389615	1.16642779	6.42872542
90	Н	15.09982481	0.03514824	7.57220695
91	Ν	14.83219730	7.72084131	-0.89068251
92	С	15.26066498	8.87490341	-0.33988811
93	Č	13,78800338	7.78049690	-1.74078935
94	Č	14.66314695	10.11035697	-0.59997626
95	Č	13.12691509	8.97004123	-2.05594126
96	Č	13 57098184	10 15895383	-1 47014302
97	н	16 09986508	8 79592739	0.34755500
98	н	13 46875320	6 83735196	-2 17916601
90	н	15 04694554	11 010135/3	-0 12727062
100	и П	12 28/67605	8 05040223	-0.12727002
100	и П	12.28407003	0.93940223	1 69720009
101	п N	12 54122505	11.10202495	-1.00/30090
102		13.34133393	4.29005651	-0.43809733
103	C	13./1//8/03	3.12993020	-1.09458202
104	C	12.33/38013	4.90/09455	-0.48062915
105	C	12.69631806	2.53/35965	-1.85151027
106	C	11.2/36/389	4.3/504282	-1.202/3085
107	C	11.45395532	3.1/398262	-1.891/56/2
108	H	14.69896934	2.66801999	-1.01728647
109	H	12.24426188	5.82099571	0.09395931
110	Η	12.87775131	1.59842340	-2.34505546
111	Η	10.31940858	4.89072570	-1.20814115
112	Н	10.63770313	2.73586333	-2.45910329

3-H dimer (-15850.5)

1	Bi	15.03688894	6.68716158	6.05151478
2	Ν	13.47346211	8.36989280	3.00165572
3	С	13.95511269	7.45379576	2.13353547
4	S	15.65433061	7.33305291	1.88777046
5	S	12.87451530	6.38388884	1.27276546
6	С	12.06912442	8.68403802	3.13812071
7	С	11.31182089	9.04348884	2.01698020
8	Н	11.77422333	9.06699684	1.03654815
9	С	9.96307384	9.37304494	2.17518261
10	н	9 37393821	9 65231908	1 30524250
11	C	9 37910084	9 35847639	3 44783957
12	c	10 15244184	9.02162295	A 56553176
12	н	9 70927102	9.01902370	5 55809936
13	C	11 50021725	9.01902370	1 41622206
14		12,00007020	8.00043691	5 29242226
15	п	12.09907929	0.43440092	2.20342320
10	H	8.3303/349	9.6186/811	3.30830/30
1/	C	14.39/1322/	9.20826610	3.76185924
18	C	14.51011336	10.53641701	3.33259661
19	Н	13.92395097	10.87056246	2.48176647
20	С	15.37376367	11.41207503	3.99040539
21	Н	15.46554620	12.44108113	3.65213095
22	С	16.11582118	10.95144594	5.08219931
23	С	15.97828046	9.62652649	5.50807660
24	Η	16.54923726	9.30552638	6.37610291
25	С	15.11881567	8.72095565	4.86018472
26	Η	16.79148153	11.62176088	5.60876950
27	0	12.62927645	8.22696619	9.81269895
28	F	11.32935971	10.29260850	8.03768931
29	S	13.26827147	8.48639001	8.51439499
30	0	14.74350727	8.58379456	8.52529444
31	Õ	12.74414444	7 66819601	7 38644980
32	Č	12.68949894	10 23937754	8 07255307
33	F	13 16093065	10.62111498	6 8601 1917
34	F	13.10079614	11 13116886	9.00306803
35	Ri	15 05998188	5 3/998060	0.09111263
36	N	13 62185356	3 58465505	3 1373/710
27	C	13.02103330	<i>1.50</i> 40 <i>555</i>	2 00/1/089
20	C c	14.011/0323	4.33931910	3.99444900 1 0005076
20	с С	13.70947230	4.82203173	4.23003370
39	S C	12.03304192	2.12002056	4.81340314
40	C	12.25215972	3.12093056	3.0/818696
41	C	11.6412/160	2.604996/1	4.22691565
42	H	12.18693097	2.58061318	5.16362833
43	C	10.33010089	2.13001746	4.15244653
44	H	9.85299231	1.72914042	5.04332593
45	С	9.63647882	2.16130760	2.93536570
46	С	10.26111762	2.66740618	1.78949319
47	Н	9.73013622	2.69240500	0.84129925
48	С	11.57148325	3.14830318	1.85932736
49	Η	12.06288866	3.54525684	0.98213991
50	Η	8.61659733	1.78894564	2.88061896
51	С	14.61472308	2.82247836	2.39200338
52	С	14.83770711	1.50680510	2.81927159
53	Η	14.26232062	1.11508651	3.65417509
54	С	15.79813182	0.71718596	2.18487307
55	H	15.97100752	-0.30159147	2.52209379
56	С	16.54141686	1.25312256	1.12879597
57	Č	16.30342386	2.56575506	0.70553564
58	Ĥ	16.89936046	2.96409919	-0.11324735
59	C	15.32974030	3.37485511	1.31763115
60	й	17 30285131	0 65442004	0 63427905
61	0	17 74162126	5 41050225	-0.15630988
62	ç	18 70583804	5 71//51/0	0.9465/1861
63	0	18 82085451	7 1/500060	1 2587/8/5
00	U	10.00000401	1.14000009	1.23074043

64	F	20.56305519	6.05076405	-0.99283583
65	С	20.35616364	5.28674494	0.11578277
66	0	18.63289683	4.79935944	2.08962585
67	F	20.39105953	3.97830182	-0.26095428
68	F	21.39107136	5.50467445	0.97639118
69	Ν	17.63095943	6.86103819	5.88200140
70	С	18.37906180	6.69343567	6.99219741
71	С	18.24465540	7.05782858	4.69589011
72	С	19.77205062	6.72332742	6.95731013
73	С	19.63250427	7.09096794	4.57988083
74	С	20.40962693	6.92461241	5.72906350
75	Η	17.83974683	6.53611049	7.92358268
76	Η	17.60474063	7.18575862	3.82878860
77	Η	20.33852897	6.58967354	7.87409392
78	Η	20.07496667	7.23660513	3.60004866
79	Н	21.49455804	6.94821957	5.67015547
80	Ν	14.37184966	4.11367665	7.60293751
81	С	13.14022863	4.17959012	8.14649898
82	С	14.79491622	2.91900429	7.14606151
83	С	12.29626277	3.07192769	8.25907273
84	С	14.01882952	1.75912374	7.21210503
85	С	12.74388034	1.83731998	7.78073892
86	Н	12.81646030	5.16222819	8.48457962
87	Н	15.78534958	2.89942910	6.69661087
88	Н	11.31072449	3.18127305	8.70341329
89	Η	14.40674876	0.82085760	6.82458060
90	Η	12.11111863	0.95567260	7.84449957
91	Ν	15.07825950	8.10751024	-1.55544498
92	С	16.35664510	8.47720085	-1.32871927
93	С	14.10602811	8.92372657	-1.09901963
94	С	16.70809155	9.64905460	-0.65543917
95	С	14.36306641	10.11185639	-0.41301370
96	С	15.69130471	10.48156162	-0.18493933
97	Η	17.12529924	7.78929080	-1.67530759
98	Η	13.08273007	8.60132138	-1.28349011
99	Η	17.75505678	9.88264477	-0.48556229
100	Η	13.53893416	10.72907020	-0.06564298
101	Η	15.92708754	11.39217329	0.35904088
102	Ν	13.15873159	3.95195235	-1.15738366
103	С	13.21621340	2.60942315	-1.27610948
104	С	12.11351472	4.61120683	-1.69945853
105	С	12.22466711	1.87791011	-1.92765205
106	С	11.07866780	3.95122380	-2.36026881
107	С	11.13363596	2.55921851	-2.47410711
108	Н	14.07217036	2.12133318	-0.81834774
109	Н	12.10978128	5.69220623	-1.58380221
110	Н	12.30781299	0.79739292	-1.99601050
111	Η	10.25059591	4.52032404	-2.77183498
112	Н	10.34046121	2.01542613	-2.97971371

Table S4. Thermodynamic data (in kcal mol⁻¹) of the dimerization of **2-H** and **3-H** computed at the ZORA-BLYP-D3(BJ)/TZ2P level of theory in pyridine and in vacuo.

	pyridine			vacuo
—	ΔΕ	ΔG	ΔΕ	ΔG
2-H dimer	-5.3	10.9	-15.8	-4.5
3-H dimer	-23.1	-8.8	-29.6	-20.4

When using a pyridine solvent model (COSMO), a monomeric species is **2-H** is favored, while a dinuclear species of **3-H** is slightly preferred. Since a higher reactivity of mononuclear species has been suggested for closely related systems based on combined experimental and theoretical approaches,^{1,13} the following mechanistic investigations were performed with mononuclear species not only for **2-H** (and related compounds), but also for **3-H** (and related compounds).

Table S5. Cartesian coordinates and ADF total electronic energy (in parentheses and in kcal mol⁻¹) of compounds involved in the mechanism for the extrusion of CE_2 (E = O, S) from **2-H** and **3-H** to give **4-H**, computed at the ZORA-BLYP-D3(BJ)/TZ2P level of theory in THF.

2-H (-8089.1)

1	С	-2.59996800	-0.17561305	-3.48615642
2	C	-4.12966884	5.46602302	-2.32706667
3	Ĥ	-3.26231860	-0.27921828	-4.34233698
4	C	-4 29149996	3 57052792	-0.81993405
5	н	-4 58118284	6 33757903	-2 79479366
6	Ri	0 22332603	0.07593138	0 70706340
7	S	3 14381444	-1 15536502	-0 56177091
8	N	-2 35//27/7	2 05771357	-0.51698022
9	C	3 393950/6	-0.93697660	-0.51070022
10	0	3 10508165	0.23447560	-0.05171832
11	C	-1 86799399	1 69252527	-0.03171032 -1.42195712
12	C	-2 0/093683	0.07030701	-1.42179712
12	C	0.03266171	0.10/11611	1 2/3/6805
13	с u	-0.93200171 5 80547613	4 05008656	1 18703000
14	0	-3.89347013	4.93998030	-1.18703900
15	0	4.27012290	-1.98509008	-0.14182338
10	C	1.60004670	-1.01003217	-0.40922101
1/		-2.90890212	3.22000087	-1.11/42/48
18	F	4.60388661	-0.3/304285	-2.6//24//2
19	U U	-2.22948166	3.99252315	-2.02407078
20	H	-1.21126/13	3.70448112	-2.26353349
21	H	-4.85//18/8	2.96450565	-0.11903989
22	C	-2.80995266	5.11118004	-2.62952353
23	Н	-2.23058024	5.70438985	-3.33291277
24	F	3.33673893	-2.13588400	-3.06485487
25	С	-2.87192739	0.80085509	-2.52918011
26	Н	-3.73568376	1.44536346	-2.65435752
27	С	-0.67906365	-0.87571245	-2.21474567
28	Η	0.16941236	-1.53921211	-2.08635821
29	С	-1.49212994	-1.01610295	-3.34162360
30	Η	-1.26961888	-1.77586483	-4.08675345
31	F	2.43097168	-0.12934596	-2.94968262
32	С	0.91578442	3.48735914	1.12614495
33	С	-0.45979154	-3.25955060	1.30262553
34	С	1.46380862	4.34492373	-1.04941268
35	Η	2.25745052	2.29574089	-1.32202335
36	Η	0.57964647	2.48799743	-1.89637536
37	С	0.72087888	4.72330749	0.25662535
38	Η	0.14592469	3.33968499	1.88496408
39	Η	1.91477166	3.44103786	1.57673939
40	Н	1.04496240	4.85143103	-1.92311023
41	Н	2.52225795	4.61284822	-0.97650563
42	Н	1.13041640	5.62364672	0.72292969
43	С	-2.41138377	-3.22945290	-0.16276095
44	Н	-3.07092849	-2.38534231	1.76653044
45	Н	-2.80589381	-1.14158261	0.51377513
46	C	-0.99689716	-3.86641701	0.00433375
47	Н	0.62063067	-3.11573873	1.32310640
48	Н	-0.78562536	-3.81423220	2.19048252
49	Н	-2.53633299	-2.82456375	-1.16928056
50	Н	-3.20521883	-3.95999708	0.01541574
51	н	-1 03578751	-4 95772314	0.06153147
52	н	-0 34828770	-3 58867875	-0.83048090
52	ц	-0 3/600170	4 87508278	0.07036685
55	\cap	-2 53/81/06	3 088871/2	1 525/10/00
55	C C	-2.55-61400	2 09718194	0 87963262
55	\sim	2.1/11///2	2.07/1017 1	0.01703202

56) 0	-1.62112852	1.04032697	1.46425403
57	0	0.78011749	2.38302789	0.15950698
58	3 C	1.31123093	2.81423523	-1.15415539
59) ()	-1.08338682	-1 92327222	1 36858583
60		-2 /7201886	-2 10623048	0.89332244
00	, C	-2.47201000	-2.100230+0	0.07552244
2 11 /	7027	1)		
3-Н (-1937.	1)	0.04400700	0 1000 (010
1	C	-3.05124202	-0.26693723	-3.19826219
2	C	-3.66260695	5.55185759	-2.91239796
3	Н	-3.77834391	-0.43836633	-3.98788084
4	С	-4.27803277	3.83895334	-1.30803465
5	Н	-3.94073623	6.39350183	-3.54192911
6	Bi	0.05654282	0.10023800	0.77621814
7	S	3.04890059	-1.08808933	-0.72621630
8	N	-2 56701394	2 24090251	-0.48662570
0	C	3 12/8/760	0.60120441	2 55860412
9 10		2 12040691	-0.00129441	-2.33600412
11		5.12040081	0.20823007	-0.01842200
11	C	-4.63090597	4.91/81891	-2.12113986
12	2 C	-2.32960857	1.03923670	-1.26799013
13	8 C	-1.21503444	0.18783787	-1.11408903
14	I H	-5.66127759	5.26424749	-2.13691358
15	5 0	4.19216656	-1.99112717	-0.55086319
16	5 0	1.71205648	-1.73550792	-0.61647852
17	7 C	-2.95207693	3.39704300	-1.29281856
18	S F	4.28981076	0.04155761	-2.83538408
19) C	-1 98490850	4 00975333	-2.08878645
20) н	-0.96815696	3 63587023	-2 07934264
20	, П Ц	5.01602524	2 24557066	0.68420242
21		-3.01093324	5.54557000	-0.06429542
22		-2.3405/045	5.09528809	-2.890/0331
23) H	-1.58630892	5.57749127	-3.51352/44
24		3.04086237	-1.70059820	-3.35298769
25) <u>C</u>	-3.24383214	0.78977218	-2.30836955
26	5 H	-4.10850248	1.43408931	-2.42432035
27	C C	-1.02959950	-0.86089238	-2.02904495
28	8 H	-0.16692126	-1.51026147	-1.92022090
29) C	-1.92823197	-1.08807595	-3.07477499
30) H	-1.75535693	-1.90096711	-3.77577446
31	F	2.09382242	0.23111122	-2.87534469
32	2 C	0.85185951	3.43408913	1.34121177
33	3 C	-0.21980183	-3.34188800	1.29407065
34	F C	1.52565331	4.37133168	-0.78701034
35	5 Н	2.31907190	2.31004978	-0.88085005
36	5 Н	0.78539532	2.50243153	-1.76713852
37		0.84846605	4 74474235	0 56250022
38	2 н	0.04805483	3 33781039	2 07222873
30	, п) п	1 82406447	3 21/58/73	1 80014450
35 AC	, П	1.02400447	1 20205477	1.60014450
40	, п п	1.03393323	4.69303477	-1.02402550
41		2.36730900	4.05505550	-0.77695505
42		1.38508113	5.556/1062	1.09205985
43		-2.15924162	-3./1806319	-0.10913623
44	H H	-2.99234223	-2.66237803	1.64408239
45	• Н	-2.73923470	-1.59576595	0.23792849
46	5 C	-0.64925011	-4.06925201	0.02326138
47	/ Н	0.83056030	-3.05149178	1.31551734
48	8 H	-0.48201023	-3.89543326	2.20418074
49) Н	-2.42891694	-3.51617344	-1.14855100
50) H	-2.78640705	-4.53687758	0.25466591
51	H	-0.47616554	-5.14691372	0.09296087
52	2 H	-0.08206542	-3.68100230	-0.82675757
53	8 H	-0.18603461	5.06410073	0.40325704
54	S	-2.86073218	3.85921590	1.63911799
55	5 C	-2.50851646	2.39688323	0.87303113
56	5 S	-2.09592511	1.03127790	1.93715051

57	0	0.58554461	2.43790734	0.28804089
58	С	1.36402213	2.83860310	-0.90472561
59	0	-1.01780797	-2.10100306	1.28321137
60	С	-2.35898299	-2.46479658	0.77375510
Int-1-C) (-80	071.4)		
1	Ċ	-3.56421815	1.91100499	-3.07127428
2	Ĉ	-1.38230151	1.99770620	3.87945981
3	Ĥ	-4 21062476	2 30960999	-3 84965090
4	C	-2 29786264	2 54292625	1 69123862
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8	N	-3.80/0680/	0.99258398	0.53832872
9	C	3.26521441	-1.21590742	-1.15132775
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12	С	-3.25546896	1.22937816	-0.75800748
13	С	-1.91080919	0.90840646	-1.05336891
14	Η	-0.93596643	3.80123392	2.77381410
15	0	1.48623368	-3.20118752	-1.49683554
16	0	0.62837591	-0.91720171	-1.05798032
17	С	-3.03712461	1.32899588	1.66782190
18	F	4.28156648	-2.05850899	-0.86032861
19	C	-2.90803201	0 45216058	2 77924850
20	н	-3 45791900	-0.48033821	2 77848203
20	н	2 38/86120	3 22804534	0.853/0103
21	C	2.30+00122	0.70870115	3 87024258
22		-2.10336302	0.13073113	J.87024238
25	п	-2.02004242	0.11203494	4.70930330
24	Г	3.27701400	-0.94590022	-2.4/02/82/
25	C	-4.06809351	1.74440582	-1./8013224
26	H	-5.09850499	1.999996818	-1.55313092
27	C	-1.41339566	1.0/352994	-2.35150392
28	Н	-0.38916434	0.79662624	-2.58460156
29	С	-2.23841442	1.57071642	-3.36639042
30	Η	-1.84789802	1.69207780	-4.37379452
31	F	3.44080341	-0.05884166	-0.46572933
32	С	1.83268288	2.42635057	1.35689885
33	С	-2.52838122	-2.33553933	-0.74264886
34	Η	2.90187333	4.30692041	1.62538295
35	Н	1.13644263	4.44042287	1.76008603
36	С	0.74724988	3.15550892	-0.69888660
37	Н	1.51932800	5,18736654	-0.54820288
38	Н	2.76256224	3.94290530	-0.80255628
39	0	0 93730904	2.01605165	0 23961299
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40	н	2 78613760	1 01107604	1 20001618
42	н	0.85500640	2 75883740	1 70858170
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43	11	-2.733353434	-4.37/10143	-1.40732424
44	П	-3.77079077	-3.90101930	-0.05150177
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47	Н	-0./15/35/0	-4.34538249	0.00288593
48	0	-1.91039126	-1.83580960	0.53192223
49	Н	-3.42727848	-1.73837624	-0.90222270
50	Η	-1.79793613	-2.16611233	-1.53797172
51	Η	-0.62464005	-2.80573327	1.87364131
52	Н	-2.37839347	-2.99517772	2.20603532
53	Η	-0.26895412	3.53560414	-0.55768586
54	0	-5.57465303	0.14514122	1.80112936
55	С	-5.03772073	0.13093307	0.66618338
56	0	-5.36604243	-0.47951472	-0.37846869
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58	С	1.81711292	4.16605056	-0.29616279
59	С	-2.78260430	-3.81762367	-0.48157831
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Int-1-S (-	-7915.5)
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1	C	-3.52878780	1.99028077	-3.06393894
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8	Ν	-3.84317616	1.05752008	0.53511957
9	С	3.27640277	-1.23306406	-1.13296561
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12	С	-3.27003017	1.24335421	-0.76832185
13	С	-1.94321023	0.85809418	-1.05156244
14	Η	-1.00072446	3.88206151	2.81186686
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17	С	-3.04361513	1.39977953	1.66308266
18	F	4.28812316	-2.07123285	-0.81747255
19	С	-2.86603385	0.47386657	2.72134856
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21	Н	-2.47850940	3.33703208	0.89297730
22	С	-2.04137787	0.79379559	3.80568047
23	Η	-1.92044390	0.07623722	4.61254276
24	F	3.31457560	-0.96673366	-2.45756340
25	С	-4.04729705	1.82889025	-1.77983055
26	Н	-5.05904500	2.14663861	-1.54896868
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28	Η	-0.42355033	0.69222238	-2.58110361
29	С	-2.22525113	1.57055381	-3.35864733
30	Η	-1.82266319	1.68440061	-4.36208895
31	F	3.43202934	-0.07329676	-0.44657448
32	С	1.78458429	2.39646993	1.36222723
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34	Η	2.88863364	4.25980965	1.60759962
35	Η	1.12783832	4.42092958	1.77410424
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37	Η	1.48160559	5.15861895	-0.54229788
38	Η	2.69927485	3.89309487	-0.81641859
39	0	0.85364147	2.00323964	0.26870476
40	Η	1.34006784	2.07321137	2.30644496
41	Η	2.72573457	1.86631841	1.19204118
42	Η	0.76241240	2.73482272	-1.68487599
43	Η	-2.59261061	-4.53747192	-1.41303770
44	Н	-3.69603119	-4.05256166	-0.10543675
45	С	-1.60059353	-3.04845289	1.43903234
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47	Н	-0.64832035	-4.44994827	0.09387728
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50	Η	-1.62403561	-2.31680027	-1.54956262
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5	Н	-1.62423509	1.72137587	5.24105371
6	Bi	-0.46292271	0.22593376	0.78759032
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8	N	-3.77976010	1.09859421	0.41812318
9	С	3.11659381	-0.79034689	-1.83313249
10	0	2.54483895	-1.18095021	0.75357414
11	Č	-1 50222120	2 26031160	3 15986024
12	Č	-3.08550461	1.28226014	-0.77863540
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14	H	-0.65537737	2.93216543	3 28302365
15	0	2 31068559	-3 13014669	-0.81142801
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17	Č	-3 21140337	1 25444584	1 63109766
18	F	4 43458305	-1 04462011	-1 64323198
10	Ċ	-3 73021430	0 58124993	2 78885712
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20	н	-1 73054162	2 75111739	1.06323664
21	C	-3 16268490	0.75309259	4 03676443
22	н	-3 58610423	0.21564611	4.89070445
23	F	2 7720/058	-1 18026346	-3 08/183165
25	C	-3 79195773	1 75780731	-1 907/9052
25	н	-4 82452195	2 07198104	-1.70747032
20 27	C	-1 15379464	0.92284523	-2 25057340
27	н	-0.13265180	0.52264525	-2.25057540
20	C	-0.13203100	1 38707862	-3 35571579
30	н	-1 41912380	1.30707002	-4 34014721
31	F	2 916/11582	0.55186764	-1.73033612
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32	C	-1 26038474	-2.87172651	1 71318011
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35	н	0 57809544	2 87980121	-1 52409371
36	ц	0.06772303	3 8000/072	0.14001824
37	C	2 60833825	3.0004072	0.04207556
38	н	1 94580214	2 07588890	2 24000927
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40	Ц	2.70050751	5 02216/00	0.00277007
40	Ц	2.14221051	3 10317131	1 22200162
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45 46	C	1 26552183	4 05045120	0.74653526
40	н	-0.26431067	-7 56435140	2 04362217
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54	C	5 66171101	0.04722043	1.04704/0/
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4	С	-1.92559656	1.69516883	1.95680107
5	Η	-1.81062099	0.99954135	5.31750511
6	Bi	-0.27627910	0.15921974	0.71831309
7	S	2.36569687	-1.55296306	-0.83229399
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9	С	3.11050242	-0.44988149	-2.18450573
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12	С	-2.92942889	1.27879547	-0.81795471
13	С	-1.57923223	0.85075138	-1.00940366
14	Η	-0.59218438	2.28809605	3.56311533
15	0	2.82795669	-2.90278744	-1.16080050
16	0	0.89124457	-1.34641756	-1.04545554
17	С	-3.16391529	1.03085404	1.59014388
18	F	4.46458200	-0.47155158	-2.12207749
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21	Η	-1.57800195	2.48940578	1.29827808
22	С	-3.34774212	0.31526300	3.92741839
23	Η	-3.88331539	-0.24602470	4.69038552
24	F	2.72401593	-0.86658494	-3.41558719
25	С	-3.60757356	1.82678414	-1.93292112
26	Н	-4.63093264	2.16458841	-1.79758481
27	С	-0.99275494	0.92820704	-2.27823564
28	Н	0.02291011	0.57232312	-2.42440404
29	С	-1.68992638	1.45900287	-3.37067413
30	Н	-1.21496365	1.51491599	-4.34694969
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32	С	2.34085558	2.38732205	1.09689744
33	С	-1.65094596	-2.66101528	2.05388369
34	С	2.09477148	4.05785988	-0.62007394
35	H	0.23209800	3.05970861	-1.29003739
36	Н	0.04503579	4.10495263	0.14979013
37	C	2.81834657	3.78990115	0.71840396
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40	Н	2.01987286	5.12366064	-0.85302688
41	Н	2.60924472	3.55436442	-1.44555331
42	Н	3.90714991	3.83083580	0.62670648
43	C	-2.76168533	-3 77067402	0 23077425
44	Ĥ	-2.83531829	-1.85225921	-0.86585456
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46	C	-1 89369733	-4.05558136	1 47747344
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51	н	-2 39077876	-4 70757267	2 20085282
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52 53	ц	2 50/100071	4 5106/286	1.10595057
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51	C	-1./14032/8	-1.14202009	0.001/22/2
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1	C	2 63987619	-0.88093257	1 17799028
-	с ц	4 34768477	0.28077552	3.03518005
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8	N	3.38089723	-0.6/1/0504	-1.18495800
9	C	-3.64862860	1.62036632	-0.60/33240
10	0	-2.15635964	2.1/14/801	1.52653949
11	С	2.99830123	-0.65505542	2.56336175
12	С	2.16946737	-1.03400862	-1.75551224
13	С	0.89551715	-0.52198791	-1.34975317
14	Η	2.34885029	-1.06573672	3.33418584
15	0	-2.55642938	4.00971089	-0.12590044
16	0	-1.04777060	2.11378412	-0.69274595
17	С	3.55866204	-0.45454052	0.10868503
18	F	-4.80469212	1.90977824	0.04203522
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20	н	2 08250581	-1 79812084	0.97629/38
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22	с u	4.97433903	1 1 1 5 0 1 0 2 0	1.05507002
23	п	3.83977430	1.11391039	2.12/02423
24	F	-3.8133/105	1.89592178	-1.9264/01/
25	C	2.20522401	-1.815/2125	-2.938/4480
26	H	3.16423690	-2.22507527	-3.24976945
27	С	-0.23180567	-0.70614251	-2.16213543
28	Н	-1.18226262	-0.26526145	-1.87277306
29	С	-0.16355710	-1.46378522	-3.33684747
30	Η	-1.05057262	-1.60872334	-3.94869646
31	F	-3.41540893	0.27917852	-0.48146617
32	С	-1.88029061	-1.53100735	2.04600162
33	С	2.21939733	2.61069765	-1.31573215
34	Н	-3.81375167	-2.10837204	1.19629443
35	Н	-2.98201766	-3.40766698	2.07022780
36	С	-0.63981546	-2.94051263	0.53478204
37	Ĥ	-2.40104394	-3 99713875	-0 21043490
38	н	-2 28345794	-2 29331978	-0 70268461
30	0	-0 55478864	-1 7/221805	1 39555832
40	н	-1 74401762	-1 68077268	3 12001281
40	ц	2 17017718	0.40420011	1 85611582
42	ц	0.06120480	2 70603330	0.28832070
42	C	1 42700270	4 62162520	-0.28632079
43		1.43/092/9	4.02102339	-0.20020200
44	п	0.70851054	3.40212304	1.42588025
45	Н	2.452/1156	3./93119/3	1.49/31491
46	C	2.40468945	4.12355237	-1.362/0563
47	Н	3.09396497	2.02119901	-1.58881031
48	Н	1.35201445	2.28231412	-1.89353682
49	Η	1.68050137	5.62627481	0.09024515
50	Η	0.40612379	4.61635749	-0.63361455
51	Η	2.16229873	4.52529046	-2.35073926
52	Η	3.43812066	4.39537721	-1.11900195
53	Η	-0.34267343	-3.80733331	1.13672210
54	С	-2.10363587	-2.99692958	0.11719118
55	0	1.94699899	2.31327429	0.12090590
56	С	1.60988889	3.57760081	0.83349582
57	Ċ	-2.83638230	-2.55168449	1.40339084
58	Č	5.08990203	0.21261527	-3.30714019
59	õ	4.26753652	0.96349846	-3.67516664
60	õ	5.95615968	-0.51391393	-2.99409761
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2	С	4.02710502	0.12575388	2.99563481
3	Н	1.18442967	-2.93841747	-4.43330366
4	С	2.58711145	-0.91903728	1.30718365
5	Н	4.24411013	0.40144569	4.02406699
6	Bi	0.56376236	0.34699643	0.69017490
7	S	-2.04216025	2.70263547	-0.11469350
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9	C	-3 51083228	1 70560301	-0 78231059
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12	C	2.20208444	-1.252/0580	-1.0554/055
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18	F	-4.66914649	2.09239409	-0.19014504
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20	Н	5.43803293	0.40428595	-0.13453122
21	Η	2.02134104	-1.84123811	1.15573044
22	С	4.93906633	0.49838868	1.94983278
23	Η	5.82883282	1.06910844	2.20918914
24	F	-3.63777353	1.86539394	-2.12467342
25	С	2.23658449	-2.08540491	-2.76713048
26	Н	3.18383015	-2.55891679	-3.01652749
27	С	-0.15407665	-0.78168518	-2.16745282
28	Η	-1.08625952	-0.26847913	-1.94570624
29	С	-0.08322800	-1.60539836	-3.29604606
30	Н	-0.94938795	-1.72644268	-3.94200171
31	F	-3.32923133	0.37389405	-0.53087703
32	С	-1.94877878	-1.53918550	2.01089791
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8	Ν	3.21702309	-0.41096267	-1.23101037
9	С	-4.02209494	1.16036377	-0.91385282
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30	Н	-0.95886706	-1.44718439	-4.35356469
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24	F	-4.52132495	2.22734125	-0.88435287
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26	Н	2.59499467	-2.10032502	-3.92263098
27	С	-0.49130826	-0.19068986	-2.57990571
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29	С	-0.51337222	-0.67868969	-3.88172726
30	Η	-1.38101545	-0.52012553	-4.51816094
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32	С	-3.16985742	-2.28832397	1.33455792
33	С	1.74669227	-3.95190738	1.50706244
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36	Н	-3.45556473	-3.11261499	-0.69823991
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51	Η	3.08557423	-5.30683568	2.53360974
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54	Η	0.80096977	3.42456126	1.30342344
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66	С	-1.36495713	-2.75162273	-0.15796434
67	С	-1.85994003	-1.77288473	1.96523335
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70	Η	-1.21821872	-3.81729618	0.06341884
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9	С	2.90385845	-0.23034522	-2.60888555
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29	C	-2.1/281293	0.73590946	-3.1/808424
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4	С	-1.59662343	0.66311392	2.41019649
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6	Bi	-0.11202749	-0.24782902	0.98081308
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33	Г С Н	-3.33235154 -0.41474176	-3.50254484 2.72645262	0.36271827 -0.88420302
33 34	C H H	-3.33235154 -0.41474176 -3.11221458	-3.50254484 2.72645262 -2.55354590	0.36271827 -0.88420302 2.35525582

36	С	-2.17399450	-3.64635664	-0.64921967
37	Η	-0.05644206	-3.10983582	-0.26677138
38	Η	-0.71432318	-4.38855416	0.79933653
39	Η	-4.27866973	-3.22582566	-0.11073961
40	Η	-3.47910969	-4.44224146	0.90655153
41	Η	-2.17634596	-4.61221627	-1.16245847
42	Η	-2.21170785	-2.84730223	-1.39763091
43	Η	-0.71389918	3.52615629	0.68772365
44	С	2.11097096	3.99555018	0.91345089
45	Η	2.20926848	2.18886629	2.18572513
46	Η	2.83663889	1.95123740	0.52946695
47	Η	0.74070702	5.18918088	-0.34760133
48	Η	1.65370371	3.93949922	-1.22315337
49	Η	3.11874254	4.36496214	0.70359629
50	Η	1.72582999	4.51845390	1.79628887
51	0	-1.35071758	-2.44142612	1.23404005
52	С	-2.83605569	-2.40268145	1.30904484
53	Η	-4.12260487	1.88506759	0.59700069
54	0	0.75813226	2.04361364	0.69256220
55	С	0.05623973	3.14535323	0.00662544
56	С	2.10011064	2.48209674	1.13796926
57	С	1.14699447	4.17587706	-0.28112619



Figure S18. Thermodynamic data for the potential steps in the formation of **4-H** from **2-H** or **3-H**. Left values for Δ H and right values for Δ G (in kcal mol⁻¹).

Table S6. Cartesian coordinates and ADF total electronic energy (in parentheses and in kcal mol⁻¹) of compounds involved in the mechanism from **3-H** to **5-E** under analysis computed at the ZORA-BLYP-D3(BJ)/TZ2P level of theory in THF.

TS-3-S (-7899.6)

1	С	-4.30287198	-0.38082083	-3.24961128
2	С	-2.41758746	5.53063526	-4.27491662
3	Н	-4.74633939	-0.62998140	-4.20970439
4	С	-3.81670378	4.56776050	-2.54244744
5	Н	-2.25275593	6.28906513	-5.03624658
6	Bi	-0.93791041	0.13174347	0.58296366
7	S	2.56324350	0.11750680	-0.25936627
8	Ν	-3.04431037	2.55886326	-1.33270086
9	С	3.72418836	-1.30327910	-0.74407543
10	0	3.45355311	1.11360084	0.34890026
11	С	-3.59981969	5.53912554	-3.52417701
12	С	-3.32555239	1.24071254	-1.75702411
13	С	-3.10632086	0.23307731	-0.80263997
14	Н	-4.35536126	6.29979479	-3.70326728
15	0	1.63514335	-0.55073653	0.70681568
16	0	1.91854370	0.49711256	-1.52780159
17	С	-2.83954472	3.59461899	-2.32173671
18	F	4.33072658	-1.82822052	0.35391142
19	С	-1.66100023	3.56815036	-3.07249335
20	Ĥ	-0.92421343	2.79102815	-2.89096523
21	Н	-4.72703645	4.56198792	-1.95197793
22	C	-1.45012658	4.54359587	-4.05103692
23	Ĥ	-0 53388649	4 53065394	-4 63558488
24	F	3 03695671	-2 29979216	-1 36845857
25	C	-3 88543365	0.93091816	-3.00233004
26	н	-4 02081372	1 71065044	-3 74605190
27	C	-3 64000791	-1.04909372	-1 00169579
$\frac{27}{28}$	н	-3 54585380	-1 81593244	-0 23805070
29	C	-4 20564072	-1 35749569	-2 24298859
30	н	-4 56556025	-2 36737808	-2 42712805
31	F	4 68640533	-0.85522836	-1 59265530
32	Ċ	0 18293479	2 94286663	2 21363124
33	C	-0 47087104	-2 71317337	-1 22573426
34	C	1 46735641	4 20105494	0 59165505
35	н	0.99472321	2 68060060	-0.94888893
36	н	-0.35079783	3 808/971/	-0 5530/282
37	C	0.76828585	4 33855243	1 96482236
38	н	-0.81576665	2 96258740	2 65769235
30	н	0.84805701	2.90290740	2.03707233
40	н	1 58739929	5 16366622	0.08586552
41	н	2 44879944	3 72709342	0.69729626
42	н	1 45912846	4 63050504	2 76050559
72 //3	C	0.62957472	-4 04421895	0.44760662
43 AA	н	-0.16/5/1883	-2 68015928	1 99685516
45 1	ц	1 3373003/	3 00860068	1 /100800/
45 46	C	0 15363037	1 0038//38	1.41998904
40	с u	1 21016255	-4.09504450	-1.02107243
47 18	и П	-1.31910333	1.06623102	-1.91343130
40	и П	0.20954154	-1.90023102	-1.52009170
49 50	и П	1 57203501	-3.03801703	0.88521508
50	п U	0.07197200	-3.4743/00/	1 77204020
51	П II	0.9/10/200	-4.2/200420	-1.72394020
52 52	п u	-0.27804408	-4.0/930490	-1.13408194
55 51	п с	-0.03023284	J.07049111	1.90622394
54 55	2	-2.9900/014	4.3001/240	0.7103480/
55	C C	-3.10343923	2.0441/199	0.01308001
30	S	-3.33907023	1.4000/612	1.10910526

58	С	0.52395324	3.27674714	-0.16662013	
59	0	-0.99217366	-2.32101674	0.10978571	
60	С	-0.49027082	-3.26781890	1.13556382	
Int-5-S	(trip	olet, -7899.1)			
1	С	-3.39541168	-0.55137093	-3.20850952	
2	С	-2.24528951	5.70683509	-2.60158391	
3	Н	-4.08768845	-0.77569615	-4.01565767	
4	С	-3.56586419	4.41917136	-1.02059682	
5	Н	-2.04698261	6.64084208	-3.12070435	
6	Bi	-0 18622515	0 24617145	0 70497989	
7	S	2 95/13/1502	-0.96823964	-1 37653398	
8	N	2.22434302	2 05162415	0 57223100	
0	C	2 5 5 5 0 6 5 6 0	0.11202122	2 02461787	
9	C	2.33300309	-0.11392122	-3.02401787	
10	0	2.988/1096	0.15589248	-0.41696/4/	
11	C	-3.30026576	5.61919203	-1.68/93/01	
12	С	-2.72675579	0.84088699	-1.34978411	
13	С	-1.60312325	0.03414126	-1.09895857	
14	Η	-3.93736014	6.48025613	-1.50329212	
15	0	4.25924746	-1.60701751	-1.63773544	
16	0	1.82208042	-1.89985344	-1.20484681	
17	С	-2.74490614	3.30455788	-1.24753096	
18	F	3.48597266	0.83595978	-3.32316428	
19	С	-1.70600515	3.37986588	-2.18739635	
20	н	-1 09965468	2 50539791	-2 38837610	
20	н	-1 / 1519610	4 34850629	-0 35225653	
21	C	1 46018375	4.57624854	2 85063003	
22		-1.40010373	4.57024054	-2.03903093	
23	п	-0.04978525	4.02550542	-3.38211883	
24	F	2.53001293	-1.01491838	-4.04639767	
25	С	-3.615/0048	0.56109806	-2.39366350	
26	Н	-4.46782136	1.21503071	-2.56014844	
27	С	-1.39200840	-1.07010372	-1.94201790	
28	Η	-0.51434967	-1.69308867	-1.79518308	
29	С	-2.27772316	-1.36396428	-2.98494503	
30	Η	-2.09027091	-2.22155339	-3.62698636	
31	F	1.33564581	0.49607228	-2.97814773	
32	С	0.13686458	3.60675196	1.19440351	
33	С	0.20765413	-3.14448594	1.30839440	
34	Č	1.44360369	4.58389451	-0.58359347	
35	н	2 28325957	2 53492352	-0 48594487	
36	ц	1.07307607	2.33472332	1 7738//70	
30	C	0.33516315	4 02160065	0 45113207	
20	с u	0.35510315	4.92109003	1 62805022	
20	п П	-0.85518807	2 42427500	1.02803922	
39	п	0.91009038	5.45457590	1.93372007	
40	H	1.27958261	5.10997236	-1.52/24624	
41	H	2.42927772	4.864/0584	-0.201/4361	
42	Н	0.62824225	5.73073695	1.12594016	
43	С	-1.79162755	-4.07799603	0.31848800	
44	Η	-2.65067394	-2.84423475	1.94006749	
45	Η	-2.66812556	-2.03761911	0.34873558	
46	С	-0.24219588	-4.11264492	0.21972635	
47	Η	1.16743416	-2.66595725	1.10968038	
48	Н	0.20197106	-3.60028690	2.30617409	
49	Н	-2.25719308	-4.15275540	-0.66794505	
50	H	-2.16133807	-4.90366339	0.93343313	
51	н	0 16140447	-5 11725475	0 37567205	
52	н	0.09909859	-3 74796199	-0 75250995	
52	ц	-0 50760080	5 20/68110	_0 05380013	
55	11 C	-0.37207760 2 50122475	2 17605100	-0.03307713	
34 55	ъ С	-3.371324/3	3.17083108	1.00029439	
55	C	-3.41321890	1.93523240	0.70427083	
56	S	-3.81638150	0.42881246	1.416/9802	

57 O 0.06160715 2.32691248 0.86894946

57	0	0.27716856	2.59765674	0.12901638				
58	С	1.36494648	3.05258233	-0.77067378				
59	0	-0.81908358	-2.08468434	1.28873418				
60	60 C -2.11903212		-2.72597269	0.99090745				
TS-4-S (triplet, -7889.7)								
1	C	-4 30183105	-0 75202793	-2 55017391				
2	C	-1 /16/3690	1 538198//	-4 53461015				
23	ч	1 60/812/0	1 16835337	3 17311161				
1	C	2 21780608	4 15126269	2 05602026				
4		-5.21/60006	4.13120306	-2.93008980				
5	П П	-1.01901/14	5.0/114904	-5.59500859				
6	B1	-0.96133425	0.22660319	0.76379689				
7	S	2.3/431627	0.32031998	-0.78546898				
8	Ν	-2.94277392	2.44076964	-1.19624391				
9	С	3.51129976	-0.88320202	-1.71605928				
10	0	3.28034381	1.40233865	-0.36944945				
11	С	-2.70218005	4.83188070	-4.06310212				
12	С	-3.28769611	1.07466075	-1.35231484				
13	С	-3.21294388	0.28698183	-0.16349224				
14	Н	-3.30587546	5.58881906	-4.55750185				
15	0	1.84616263	-0.51019583	0.32968061				
16	0	1.35975325	0.68522168	-1.79394531				
17	Ċ	-2.43286337	3.18091015	-2.32709436				
18	F	4 46309014	-1 39124595	-0.88730856				
19	Ċ	-1 15197885	2 86983236	-2 79523104				
20	н	-0 557/3698	2.00705250	-2 30107935				
20	и П	4 21280065	4 36683063	2.50107955				
21	П	-4.21260003	4.30063003	-2.36033149				
22		-0.04440890	3.33373230	-3.90209145				
23	H	0.35306331	3.32264277	-4.26585986				
24	F	2.8010/425	-1.92556219	-2.233/4215				
25	C	-3.77665079	0.546/1519	-2.54690195				
26	Н	-3.77937227	1.1515/129	-3.44893480				
27	С	-3.84042373	-0.99026941	-0.17250353				
28	Н	-3.84625550	-1.59564256	0.72985668				
29	С	-4.36453151	-1.49876527	-1.35626364				
30	Н	-4.80276486	-2.49421781	-1.36511355				
31	F	4.13571256	-0.24887165	-2.74437917				
32	С	-0.49273166	3.20156173	2.60764526				
33	С	-0.71781717	-2.26304660	-1.50651048				
34	С	1.23776074	4.33970666	1.35186330				
35	Н	1.04250709	2.84557219	-0.26705136				
36	Н	-0.27914086	4.05798673	-0.19691326				
37	C	0.24167402	4.54564324	2.51626297				
38	Ĥ	-1.56476514	3.30188181	2,79382467				
39	Н	-0.04375378	2 51914823	3 33616900				
40	н	1 54599315	5 28314758	0.89166364				
40	ц	2 132/0688	3 80380834	1 68850836				
41	и П	2.13249088	1 70700135	2 45745504				
42	C II	0.73017739	2 9460900	0.22216705				
43		0.05150550	-3.04090092	-0.33210703				
44	н	0.45225854	-2.50950985	1.45391859				
45	H	-0.90694188	-3./1480233	1.21650352				
46	C	-0.19044908	-3.68964088	-1.61859052				
47	H	-1.68655948	-2.09484638	-1.98042941				
48	Н	0.00781186	-1.52027277	-1.84987212				
49	Η	0.79500267	-4.89342202	-0.04894973				
50	Н	1.63063643	-3.37796974	-0.45877659				
51	Н	0.40237794	-3.82826593	-2.52704588				
52	Η	-1.01977626	-4.40599875	-1.62910040				
53	Η	-0.46735775	5.34233170	2.26935794				
54	S	-3.05731696	4.67431522	0.35369064				
55	С	-3.21321976	3.03600450	0.02665673				
56	S	-3.77441526	1.94627801	1.27601596				
57	0	-0.33159655	2.57291046	1.27348256				

58	С	0.43868049	3.47292611	0.38787853			
59	0	-0.92686311	-2.06213408	-0.04659335			
60	С	-0.16264720	-3.08815315	0.71548436			
Int-6-S	(trip	olet, -7915.5)					
1	С	15.69816682	5.62701468	2.19682131			
2	С	17.22744133	5.84829100	4.97471590			
3	Ν	16.24870448	6.65341724	4.35741431			
4	С	18.14011201	4.99579017	4.34608361			
5	Н	18.13983615	4.88209986	3.26785890			
6	S	14.21298505	8.43552847	4.80202889			
7	C	19.05059733	4.29724277	5.14498325			
8	Н	19.76676153	3.63112260	4.67188376			
9	S	15.95404109	7.09317541	6.88201974			
10	C	19.06213049	4.45442195	6.54016421			
11	H	19.79276291	3.91/90308	7.13898216			
12	C	18.14511974	5.30014266	7.17102625			
13	H	18.1608/309	5.43/21103	8.24714562			
14	C	17.22271814	5.98418557	6.37848473			
15	C	15.45414499	7.39210878	5.21313430			
10	C	16.413/8540	8.00033729	0.88501251			
1/	C	15.63445929	5.70447538	0.80121634			
18	C	15.99326536	6.88/60638	0.14565281			
19	H	16./99/28//	8.78392625	2.86839664			
20	H	15.41843866	4./1682900	2./194//60			
21	H	16.68853192	8.92197701	0.37846182			
22	H	15.30200922	4.84150404	0.23024077			
23	H	15.94229845	6.94450170	-0.938/6091			
24	C	16.13183082	0./3959601	2.92076520			
25	C D	16.48409372	7.92851459	2.27754051			
26	B1	18.49013775	9.74260518	5.49031046			
27	п	22.10340804	0.82092700	2.73842295			
28	П	15.24804524	11.21/51001 8 co007007	9.0/108203			
29	C	10.41420333	8.02007227 8.68885710	0.30409194 2.51622522			
21	с u	20.04032463	0.00003719	2.51055522			
31	п u	20.30978392	12 42057455	8 30765326			
32	0	17.02005026	12.43937433	6 86840056			
33 34	e e	10.80725035	8 15604237	0.80840030 8 73577547			
34	ы Ц	19.00725055	6 75070221	2 10656242			
36	0	20.06030775	8 50134572	2.10050242			
30	ч	20.00939773	7 70/27108	1 01836340			
38	н	20 18738042	6 81338017	5 06380435			
39	н	18 78537234	12 08969408	7 59378338			
40	н	17 47333419	13 20224175	7.09568209			
41	н	15 11252955	10 40367395	7 02836973			
42	н	16 32067079	9 66876397	8 13426600			
43	0	20.59659937	8.08605403	7.48977230			
44	Č	20.61182093	9.60031736	9.67004934			
45	Õ	19.97833823	7.01870981	9.65868966			
46	Н	19.11130424	9.19267657	2.24427658			
47	Н	20.89687719	9.29449293	2.19536981			
48	С	21.13327367	6.63304968	3.07153663			
49	Ċ	17.10907361	12.21815989	9.02623576			
50	C	15.69279226	11.65572671	8.77335196			
51	С	20.82764143	7.37449473	4.38318559			
52	F	20.55734610	10.74989428	8.93264720			
53	F	21.92207246	9.34343413	9.94200635			
54	F	19.97532193	9.83605664	10.85267463			
55	С	20.14691863	7.24314457	2.04795967			
56	Н	21.00116927	5.55416371	3.18468640			
57	С	17.70858658	12.27117272	7.62128262			
58	Н	17.09860896	13.20474671	9.49788328			

59	Н	17.68309350	11.53138837	9.65694158			
60	С	15.94778663	10.60004563	7.70408512			
Int-6-S (singlet, -7916.1)							
1	С	15.54930893	5.80209889	2.63722755			
2	С	17.45703142	5.59518606	5.23519698			
3	Ν	16.34214374	6.37962346	4.89068928			
4	C	18.34104381	4.95906036	4.35814989			
5	н	18 20388905	5 03378009	3 28505548			
6	S	1/ 1038003/	7 76823005	5 86157478			
7	C	10.20641701	4 22102140	1 00252252			
/		19.39041701	4.22193140	4.90233333			
8	Н	20.0916/4//	3./18/0/8/	4.23620139			
9	S	16.34412893	6.35914/28	/.4681299/			
10	C	19.57124370	4.12553550	6.29257935			
11	Н	20.40600236	3.55686765	6.69370717			
12	С	18.68290986	4.75375760	7.17050720			
13	Η	18.81931949	4.69613411	8.24487319			
14	С	17.62092761	5.48165899	6.63069181			
15	С	15.59799996	6.86161827	5.94668002			
16	С	16.12773977	8.40953562	1.77808112			
17	Ċ	15.32768525	6.16770928	1.30450802			
18	Č	15 61669940	7 46819212	0.87525751			
10	н	16 76446082	8 75/255/6	3 82838642			
20	и П	15 22612110	4 70821750	2.02752100			
20	п	16.252012119	4.79621730	2.98733190			
21	н	10.35283085	9.41905829	1.44505559			
22	H	14.92809693	5.43000/50	0.60641804			
23	Н	15.44220355	7.74905189	-0.16044077			
24	С	16.06558748	6.74864972	3.52340399			
25	С	16.35583372	8.05041547	3.10804600			
26	Bi	19.17485925	9.46372425	6.11398042			
27	Η	22.06156371	7.12124169	2.31937159			
28	Η	14.83633456	11.31472379	8.17124521			
29	0	18.35783163	9.14083541	8.63822111			
30	С	20.01555527	8.98906291	2.78430080			
31	Н	19.98609054	7.60231544	1.09590019			
32	Н	14.99779432	12.58211071	6.93748429			
33	0	17 50824871	11 23138718	6 20782852			
34	Š	19 30695579	8 03840467	8 96684847			
35	н	18 99029203	7 109/5927	2 / 8023606			
36	0	20 63400552	8 8205/3/6	4 12057020			
27	П	20.03499552	7 49009701	4.12037029			
20	п	22.12040360	7.40900791	4.72171100			
38 20	н	20.44457275	0.880/8390	4.91005325			
39	H	18.83350996	12.39193059	/.31690993			
40	H	17.61141824	13.31580180	6.38333916			
41	Н	15.59185750	10./19/1220	5.52984340			
42	Н	16.18520800	9.76733684	6.91421853			
43	0	20.09202239	7.64987653	7.75569268			
44	С	20.59137385	8.86456650	10.09898679			
45	0	18.78122162	6.93976261	9.77377874			
46	Η	19.03936490	9.46364769	2.92395961			
47	Η	20.66834409	9.65438334	2.20968251			
48	С	21.13132186	6.86432296	2.83807357			
49	С	16.78451921	12.36200751	8.17164660			
50	С	15.53130038	11.78816636	7.47196666			
51	Ĉ	21.13377016	7,42874679	4.25733654			
52	F	21 17122711	9 92227304	9 48040588			
52	F	21 55894400	7 97282187	10 43305/70			
55	F	10 0075/1920	9 30238051	11 22500761			
54	r C	10.021044009	7 58170207	11.23370701 11.23370701			
33 EC		17.75100450	1.30419021	2.10/90280			
56	H	21.02234/34	5.//00880/	2.838/3443			
57	C	17.78521756	12.4434/36/	7.02486272			
58	H	16.60314186	13.33860322	8.62974284			
59	Н	17.14980185	11.67040516	8.93873172			

60 C 16.11171783 10.	6.48836417 6900920
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5-H (-3659.5)

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1	С	13.95536401	6.05830388	0.93097317
2	С	16.22377759	5.86972282	3.35863499
3	Ν	16.06819910	5.46075540	2.02302336
4	С	15.21932606	5.96585144	4.32680374
5	Н	14.19238021	5.71255949	4.08277979
6	S	17.34246710	5.00987210	-0.35293789
7	С	15.57840004	6.39447269	5.60823187
8	Η	14.81238104	6.47588388	6.37476887
9	S	18.58424705	5.97355583	2.24888239
10	С	16.91002881	6.72029405	5.91741438
11	Η	17.16613576	7.05179138	6.92032070
12	С	17.91479999	6.62361471	4.94820235
13	Η	18.94494333	6.87487193	5.18353991
14	С	17.56025751	6.19524650	3.66700987
15	С	17.21408989	5.44661094	1.25951141
16	С	13.11844513	3.38450232	1.07487963
17	С	12.69993754	5.68942557	0.43746871
18	С	12.28248949	4.35474340	0.50870726
19	Η	15.03914256	3.00656271	2.01077281
20	Η	14.29798590	7.08808905	0.88220314
21	Η	12.79445017	2.34847662	1.13110981
22	Η	12.05104128	6.44296971	-0.00140651
23	Η	11.30567007	4.07122993	0.12459892
24	С	14.77915715	5.07928094	1.48970961
25	С	14.37527654	3.74531488	1.57064615

Table S7. Dipole moments of selected compounds as determined by experimental methods (literature) or by DFT calculations the ZORA-BLYP-D3(BJ)/TZ2P level (this work).

Compound X	CH_2Cl_2	THF	Pyridine	1,2-C ₆ F ₂ H ₄	1-H	2-H
Dipole moment of X [D]	1.6 (exp.) ¹⁴	1.8 (exp.) ¹⁵	2.3 (exp.) ¹⁶	2.6 (exp.) ¹⁷	-	-
po.eo	-	1.8 (theo.)	2.3 (theo.)	-	4.8 (theo.)	6.1 (theo).
Reaction $1-H \rightarrow 4-H$ proceeds,	No	No	Voc	Voc	No	Voc
when X is used as solvent / catalyst	INU	INU	162	162	INU	162

In the analysis of the dipole moments, it has to be taken into consideration that the solvents are present in excess, while the starting material **1-H** is present in stoichiometric amounts and compound **2-H** was present in catalytic amounts.



Figure S19. Thermodynamic data for the potential steps in the formation of **5-H** from **3-H**. Left values for Δ H and right values for Δ G (in kcal mol⁻¹).

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