

Electronic Supplementary Information

Cluster dynamics of heterometallic trinuclear clusters during ligand substitution, redox chemistry, and group transfer processes

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Experimental Methods

Materials and methods. All manipulations involving metal complexes were carried out using standard Schlenk line or glovebox techniques under a dinitrogen atmosphere. All glassware was oven-dried for a minimum of 8 h and cooled in an evacuated antechamber prior to use in a glovebox. Celite ® 545 (J. T. Baker) was dried in a Schlenk flask for 24 h under dynamic vacuum while heating to at least 150 °C prior to use in the drybox. Benzene, diethyl ether, hexanes, and tetrahydrofuran (THF) were dried and deoxygenated on a Glass Contour System (SG Water USA, Nashua, NH) and stored over 4 Å molecular sieves (Strem) prior to use. Anhydrous pentane and pyridine were purchased from Sigma Aldrich and stored over 4 Å molecular sieves prior to use. Benzene-*d*₆ and tetrahydrofuran-*d*₈ were purchased from Cambridge Isotope Laboratories and were degassed and stored over 4 Å molecular sieves prior to use. Non-halogenated solvents were typically tested with a standard purple solution of sodium benzophenone ketyl in THF in order to confirm effective oxygen and moisture removal. Ligand (^t*bs*LH₆),¹ ligand precursors,¹ (^t*bs*LH₂)Fe₂,² and (^t*bs*L)Fe₃(thf)¹ were synthesized as previously reported. Fe₂(N(SiMe₃)₂)₄,³ Zn(N(SiMe₃)₂)₂,³ ZnCl₂py₂,⁴ KC₈,⁵ PhN₃,⁶ and (3,5-CF₃)₂C₆H₃N₃⁷ were prepared following published methods. The adducts Fe(N(SiMe₃)₂)₂(py)₂ and Zn(N(SiMe₃)₂)₂(py) were prepared by adding the appropriate number of equivalents of pyridine to the unsolvated precursors. All other reagents were purchased from commercial vendors and used without further purification unless explicitly stated.

Physical Measurements. All of the measurements for the metal complexes were made under anaerobic conditions unless explicitly stated. Elemental analysis of [NBu₄][(^t*bs*L)Zn₃(μ³-Cl)] (**8**) was performed by Complete Analysis Laboratories, Inc., Parsippany, NJ. All other elemental analyses were performed in-house on a 2400 Series II CHNS/O Elemental Analyzer manufactured by Perkin Elmer, Inc. ¹H and ¹⁹F NMR spectra were recorded on Varian Unity/Inova 500 or 600 MHz- spectrometers. Chemical shifts in ¹H NMR spectra were referenced to solvent residual resonances (C₆D₆, 7.16 ppm; THF-*d*₈, 3.58 and 1.73 ppm). Chemical shifts in ¹⁹F NMR spectra were referenced to BF₃(Oet₂) (-153 ppm) or fluorobenzene (-113 ppm).

Zero-Field ⁵⁷Fe Mössbauer Spectroscopy. ⁵⁷Fe Mössbauer spectra were measured with a constant acceleration spectrometer (SEE Co, Minneapolis, MN). Unless otherwise indicated, all ⁵⁷Fe Mössbauer spectra were recorded under zero applied field. Samples were prepared as Paratone-N mulls in a glovebox and frozen in liquid nitrogen prior to handling in air. Isomer shifts are quoted relative to α-Fe metal at room temperature. Data were processed, fitted, and analyzed using an in-house package for IGOR Pro 6 (Wavemetrics, Lake Oswego, OR) written by Evan R. King.

X-ray Fluorescence Spectroscopy. X-ray fluorescence analyses were recorded using an Olympus Delta Innov-X XRF analyzer with no additional filter (Olympus, NDT, Inc., Waltham, MA). Samples for the calibration curve were prepared by dissolving ferrocene and *bis*[4,4,4-trifluoro-1-(2-thienyl-1,3-butanedionato)]Zn·(TMEDA) in benzene, and then lyophilizing the solution to afford a homogenous mixture of solids. The areas of the Kα peaks for Fe and Zn were calculated by integration using IGOR Pro 6 (Wavemetrics, Lake Oswego, OR). Quantification was performed using the Kα peak area. Fe:Zn molar ratios in samples were determined from the peak area ratios using a linear fit to the calibration curve (Figure S13).

X-ray Diffraction Techniques. All structures were collected on a Bruker three-circle platform goniometer equipped with an Apex II CCD area detector and an Oxford cryostream cooling device at 100 K. Structures were obtained using radiation from a graphite fine-focus sealed tube Mo Kα

(0.71073 Å) source or synchrotron radiation (0.41328 Å) using the Advanced Photon Source (APS) at Argonne National Laboratory. Crystals were mounted on a cryoloop or glass fiber pin using Paratone-N oil. Data were collected as a series of ϕ and/or ω scans. Data were integrated using SAINT⁸ and scaled with either a numerical or multi-scan absorption correction using SADABS.⁸ The structures were solved by intrinsic phasing or direct methods using SHELXT/SHELXS⁹ and refined against F^2 on all data by full matrix least squares with SHELXL⁹ using the OLEX2 interface.¹⁰ All non-hydrogen atoms were refined anisotropically. Hydrogen atoms were placed at idealized positions and refined using a riding model. The isotropic displacement parameters of all hydrogen atoms were fixed to 1.2 times the atoms they are linked to (1.5 times for methyl groups). Unless otherwise indicated, all depictions of solid-state structures are displayed as 50% probability thermal ellipsoid plots. Further details on particular structures are noted in the X-Ray Diffraction Techniques section.

SQUID Magnetometry. Magnetic measurements were recorded using a Quantum Design MPMS-3 system. Samples used for magnetic characterization were prepared by carefully adding carefully weighed, finely crushed microcrystalline powder (ca. 25 mg) to the bottom of a low-paramagnetic impurity 500 MHz borosilicate NMR tube in an air-free, Argon-filled glovebox. An equal volume of n-eicosane was added to the tube (~30 mg); all weighing was performed using plastic anti-static spatulas. The NMR tube was then fitted with a gas-line adaptor, removed from the glovebox, and transferred to a Schlenk line. The tube was then flame sealed under vacuum, and the eicosane was melted using a 40-43 °C hot water bath. Upon cooling, the product-eicosane mixture formed a solid matrix. The sealed ampule was placed in a straw and suspended in the magnetometer. Dc magnetic susceptibility data were collected in the temperature range 1.8–300 K under applied fields of 0.1 and 0.5 T. Susceptibility data were corrected for the diamagnetic contribution of a blank sample consisting of eicosane, capsule and straw. The magnetic susceptibilities were adjusted for diamagnetic contributions using the constitutive corrections from Pascal's constants.¹¹ Dc susceptibility data and magnetization data were fitted using PHI.¹²

Synthesis.

(^{tb}L)Fe₃(py) (2). *Method 1:* A solution of ^{tb}LH₆ (109 mg, 0.146 mmol) and Fe(N(SiMe₃)₂)₂(py)₂ (234 mg, 0.438 mmol) in benzene (2 mL) was heated to 120 °C for 3 days. The reaction mixture was lyophilized and extracted with diethyl ether; the remaining solid was identified as pure (^{tb}L)Fe₃(py) following extraction in benzene, filtration through a glass wool plug, and lyophilization (29.2 mg). The diethyl ether fraction was filtered through a glass wool plug and concentrated, and crystallization was achieved at –33 °C to yield brown crystalline product (58.1 mg). The two fractions resulted in an overall yield of 87.3 mg of product (61%).

Method 2: To (^{tb}L)Fe₃(thf) (115 mg, 0.117 mmol) in benzene (2 mL) was added a solution of pyridine (14.4 mg, 0.182 mmol) in benzene (2 mL). The reaction mixture was stirred at room temperature for 1 h, then concentrated *in vacuo* and crystallized from concentrated diethyl ether. Crystallization at –33 °C yielded 86.4 mg of product (74%) ¹H NMR (500 MHz, C₆D₆): δ 222.5, 113.7, 66.8, 55.7, 54.1, 49.3, 46.0, 43.9, 31.4, 26.2, 23.2, 3.3, 1.1, –1.1, –5.9, –17.0, –23.0, –53.0, –77.9, –95.5, –105.1, –197.9, –208.1. ⁵⁷Fe Mössbauer (90 K) (δ , $|\Delta E_Q|$ (mm/s)): component 1 (38%): 0.33, 1.84 (Γ = 0.18 mm/s); component 2 (28%): 0.55, 1.76 (Γ = 0.16 mm/s), component 3 (33%): 0.74, 1.39 (Γ = 0.29 mm/s). Calc. for C₄₇H₇₁Fe₃N₇Si₃: C, 57.26; H, 7.26; N, 9.94. Found: C, 57.61; H, 7.21; N, 10.09.

[Na(thf)][(t^{bs}L)Fe₂Na(thf)] (4). To a frozen solution of NaN(SiMe₃)₂ (166.7 mg, 0.909 mmol) in THF (3 mL) was added (t^{bs}LH₂)Fe₂ (387.8 mg, 0.454 mmol) in THF (3 mL). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers, and stirred at room temperature over the course of 2 h. The volatile components of the reaction mixture were removed *in vacuo*, then washed with hexanes. The remaining solid was collected with THF and filtered through a glass wool plug. Removal of the volatile components from the filtrate *in vacuo* yielded 467.7 mg of dark brown solid (99%). It is noted that crystallization of **4** from ether solution at -33 °C yields a composition differing in the coordinated solvent; refer to the X-Ray Diffraction Techniques section. ¹H NMR (500 MHz, THF-*d*₈): δ 136.5, 89.8, 77.8, 48.7, 45.0, 41.3, 37.4, 31.2, 24.8, 23.8, 14.1, 7.3, 1.3, 0.9, -1.3, -3.6, -5.5, -9.1, -11.2, -15.3, -19.8, -34.6, -38.5, -39.2, -43.5, -44.2, -65.2, -73.4. ⁵⁷Fe Mössbauer (90 K) (δ, |ΔE_Q| (mm/s)): component 1 (50%): 0.54, 0.87 (Γ = 0.16 mm/s); component 2 (50%): 0.47, 1.51 (Γ = 0.16 mm/s). Calc. for C₅₀H₈₂Fe₂N₆Na₂O₂Si₃: C, 57.68; H, 7.94; N, 8.07. Found: C, 57.68; H, 7.90; N, 8.20.

(t^{bs}L)Fe₂Zn(thf) (5). To ZnCl₂ (19.2 mg, 0.14 mmol) frozen in THF (3 mL) was added (t^{bs}L)Fe₂Na₂(thf)₂ (**4**) (146.3 mg, 0.14 mmol) in THF (3 mL). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers, and stirred at room temperature overnight. The volatile components of the reaction mixture were removed *in vacuo*, and the resulting dark brown solid was extracted in hexanes with drops of benzene. The dark brown extracts were filtered through a Celite plug, and maintenance of the filtrate at -33 °C yielded 117 mg of dark brown crystalline product (84%). ¹H NMR (600 MHz, C₆D₆): δ 143.0, 108.7, 45.3, 41.0, 27.4, 25.6, 23.7, 22.6, 19.1, 17.0, 14.1, 9.5, 8.3, 6.0, 5.4, 4.1, 3.8, 3.1, 1.8, 1.7, 1.6, 1.5, 1.2, 1.0, 1.0, 0.9, -0.6, -1.1, -2.0, -2.2, -5.4, -8.5, -10.4, -11.2, -15.4, -17.8, -23.8, -28.1, -41.4, -42.0, -54.3. ⁵⁷Fe Mössbauer (90 K) (δ, |ΔE_Q| (mm/s)): component 1 (50%): 0.6, 1.83; component 2 (50%): 0.84, 1.55. Calc. for C₄₆H₇₄Fe₂N₆Si₃Zn: C, 55.90; H, 7.55; N, 8.50. Found: C, 56.30; H, 7.89; N, 8.29. Fe/Zn ratio by XRF: 2.2:1.

(t^{bs}L)Fe₂Zn(py) (6). To ZnCl₂(py) (30.8 mg, 0.10 mmol) frozen in benzene (2 mL) was added (t^{bs}L)Fe₂Na₂(thf)₂ (**4**) (99 mg, 0.095 mmol) in benzene (2 mL). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers, and stirred at room temperature overnight. The volatile components of the reaction mixture were removed *in vacuo*, and the brown solids were extracted in diethyl ether. Filtration of the extracts through a Celite plug, and maintenance of the filtrate at -33 °C yielded 70.7 mg of dark brown crystalline product (75%). ¹H NMR (600 MHz, C₆D₆): δ 129.8, 97.8, 37.9, 27.3, 25.3, 24.9, 19.9, 19.7, 16.0, 14.2, 10.5, 4.3, 3.6, 3.3, 1.7, 1.6, 1.5, 1.2, 1.1, 1.0, 0.9, 0.1, -0.2, -0.6, -0.81, -0.84, -2.0, -3.0, -7.8, -10.9, -16.7, -21.9, -26.0, -36.0, -40.3. ⁵⁷Fe Mössbauer (90 K) (δ, |ΔE_Q| (mm/s)): component 1 (50%): 0.61, 1.68; component 2 (50%): 0.79, 1.63. Calc. for C₄₇H₇₁Fe₂N₇Si₃Zn: C, 56.71; H, 7.19; N, 9.85. Found: C, 57.09; H, 7.48; N, 9.78. Fe/Zn ratio by XRF: 2.1:1.

[NBu₄][(t^{bs}L)Fe₃(μ³-Cl)] (7). To [NBu₄][Cl] (28.2 mg, 0.101 mmol) frozen in THF (2 mL) was added (t^{bs}L)Fe₃(thf) (**1**) (99.6 mg, 0.102 mmol) in THF (2 mL). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers, and stirred at room temperature over the course of 3 hours. The volatile components of the reaction mixture were removed *in vacuo*, and the brown solids were extracted in diethyl ether. Filtration of the extracts through a Celite plug, and maintenance of the filtrate at -33 °C yielded 100.9 mg of dark brown crystals (>99%). ¹H NMR (500 MHz, C₆D₆): δ 184.7, 56.9, 36.1, 32.7, 31.1, 7.3, 5.8, 3.4, 3.1, 1.6, 1.1, 1.0, 0.7, 0.2, -36.9, -41.4. ⁵⁷Fe Mössbauer (90 K) (δ, |ΔE_Q| (mm/s)): 0.72, 1.31 (Γ = 0.24 mm/s). Calc. for C₅₈H₁₀₂ClFe₃N₇Si₃: C, 58.80; H, 8.68; N, 8.28. Found: C, 58.98; H, 8.60; N, 8.66.

[NBu₄][(t^{bs}L)Zn₃(μ³-Cl)] (8). In a thick-walled bomb flask with a Teflon screw cap, Zn(N(SiMe₃)₂)₂(py) (464 mg, 1.0 mmol) and t^{bs}LH₆ (247.7 mg, 0.33 mmol) were combined in benzene (10 mL). The reaction mixture was heated to 110 °C for five hours. This solution was added to a vial containing [NBu₄][Cl] (93 mg, 0.33 mmol), and the resulting mixture stirred at room temperature overnight. A white precipitate formed which was filtered and washed with benzene. The remaining solid was extracted in THF and filtered through a glass wool plug. The filtrate was layered with hexanes, and maintenance of the mixture at -33 °C yielded 287.2 mg of white crystalline product (71% yield). ¹H NMR (500 MHz, THF-*d*₈): δ 6.97 (d, 3H), 6.76 (d, 3H), 6.49 (t, 3H), 6.23 (t, 3H), 3.77 (br, s, 3H), 3.22 (t, 8H), 2.28 (d, 3H), 2.09 (d, 3H), 1.66 (m, 8H), 1.39 (m, 8H), 0.99 (t, 12H), 0.96 (s, 27H), -0.10 (s, 9H), -0.20 (s, 9H). Calc. for C₅₈H₁₀₂ClN₇Si₃Zn₃: C, 57.41; H, 8.47; N, 8.08. Found: C, 57.25; H, 8.47; N, 7.87.

[NBu₄][(t^{bs}L)Fe_{*n*}Zn_{3-*n*}(μ³-Cl)] (9; *n* = 1,2). To [NBu₄][Cl] (11 mg, 0.040 mmol) frozen in benzene (1 mL) was added (t^{bs}L)Fe₂Zn(thf) (**4**) (39.2 mg, 0.040 mmol) in benzene (1 mL). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers, and stirred at room temperature overnight. The remaining solid was extracted in THF and filtered through a glass wool plug. The filtrate was layered with hexanes, and maintenance of the mixture at -33 °C yielded 33.6 mg of dark brown crystalline material (71%, based on recovery of mass). ¹H NMR (600 MHz, THF-*d*₈): δ 223.7, 220.9, 210.4, 191.1, 189.5, 130.7, 124.1, 122.0, 57.8, 50.4, 50.5, 48.3, 47.5, 42.7, 41.8, 39.9, 36.3, 36.2, 35.9, 33.2, 32.0, 31.3, 26.7, 26.6, 25.5, 21.5, 21.2, 16.1, 15.7, 13.7, 11.6, 10.9, 7.3, 7.1, 5.6, 5.5, 5.0, 2.7, 1.9, 1.3, 1.1, 1.0, 0.9, 0.8, 0.2, 0.0, -0.7, -1.5, -2.2, -3.1, -3.6, -5.3, -6.3, -7.5, -9.7, -11.1, -28.0, -29.2, -33.4, -37.7, -42.2, -43.6 -59.7, -65.3. ⁵⁷Fe Mössbauer (90 K) (δ, |ΔE_Q| (mm/s)): 0.77, 1.13. Fe/Zn ratio by XRF: 2.1:1.

(t^{bs}L)Fe₂Zn(μ³-NPh) (11). To phenyl azide (3.8 mg, 0.032 mmol) frozen in benzene (1 mL) was added a benzene solution (1 mL) of (t^{bs}L)Fe₂Zn(thf) (**5**) (30.1 mg, 0.030 mmol). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers, and stirred at room temperature over the course of 1 h. The reaction mixture was lyophilized to yield 28.6 mg of dark brown powder, which represented a mixture of (t^{bs}L)Fe₃(μ³-NPh) and (t^{bs}L)Fe₂Zn(μ³-NPh) (95% mass recovery). Crystals suitable for X-ray diffraction were obtained from a saturated hexanes solution of **11** at -33 °C. Crude ⁵⁷Fe Mössbauer (90 K) (δ, |ΔE_Q| (mm/s)): 0.28, 2.31.

(t^{bs}L)Fe₂Zn(μ³-NPh)(py) (12). To phenyl azide (5.9 mg, 0.050 mmol) frozen in benzene (1 mL) was added (t^{bs}L)Fe₂Zn(py) (**6**) (41.5 mg, 0.042 mmol) in benzene (1 mL). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers, and stirred at room temperature over the course of 1 hour. The reaction mixture was lyophilized, and then extracted in hexanes. Following filtration of the hexanes extracts through a glass wool plug, maintenance of the filtrate at -33 °C yielded 26.4 mg of dark brown crystalline product (58% yield). ¹H NMR (500 MHz, C₆D₆): δ 176.4, 110.7, 74.8, 54.6, 27.1, 25.6, 22.7, 21.3, 16.2, 14.5, 11.9, 9.7, 9.1, 8.5, 8.2, 7.8, 6.7, 4.4, 3.6, 2.5, 1.8, 1.4, 1.2, 1.0, 0.9, 0.3, -0.2, -1.9, -2.8, -3.6, -5.4, -7.4, -8.7, -14.6, -16.9, -21.6, -23.8. ⁵⁷Fe Mössbauer (90 K) (δ, |ΔE_Q| (mm/s)): component 1 (51%): 0.33, 1.80; component 2 (49%): 0.35, 2.76. Calc. for C₅₃H₇₆Fe₂N₈Si₃Zn: C, 58.59; H, 7.05; N, 10.31. Found: C, 58.40; H, 6.99; N, 10.33. Fe/Zn ratio by XRF: 2.1:1.

(t^{bs}L)Fe₃(μ³-N(3,5-C₆H₃(CF₃)₂)) (13). To a frozen solution of 3,5-(CF₃)₂C₆H₃N₃ (27.6 mg, 0.11 mmol) in benzene (2 mL) was added (t^{bs}L)Fe₃(thf) (**1**) (100.1 mg, 0.10 mmol) in benzene (2 mL). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers, and stirred at room temperature over the course of 1 h. The reaction mixture was

lyophilized and washed with cold pentane. Crystals were obtained upon cooling of the pentane wash to $-33\text{ }^{\circ}\text{C}$. The remaining solid extracted in hexanes and filtered through a glass wool plug. Maintenance of the hexanes filtrate at $-33\text{ }^{\circ}\text{C}$ yielded a second crystalline crop. Crystals harvested from the two fractions gave an overall yield of 61.8 mg of product (53% yield). ^1H NMR (500 MHz, C_6D_6): δ 194.7, 60.1, 9.8, 8.0, 5.6, 3.0, 2.5, 1.9, -5.1 , -5.5 , -8.4 , -21.7 . ^{19}F NMR (500 MHz, C_6D_6): δ -69.4 . ^{57}Fe Mössbauer (90 K) (δ , $|\Delta E_Q|$) (mm/s): component 1 (33%): 0.37, 1.8; component 2 (33%): 0.38, 1.13; component 3 (33%): 0.45, 2.61. Calc. for $\text{C}_{50}\text{H}_{69}\text{F}_6\text{Fe}_3\text{N}_7\text{Si}_3$: C, 52.96; H, 6.13; N, 8.65. Found: C, 53.30; H, 6.28; N, 8.28.

$(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))$. To a frozen solution of $3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3\text{N}_3$ (5.3 mg, 0.028 mmol) in benzene (1 mL) was added a benzene solution (1 mL) of $(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\text{thf})$ (**5**) (27.1 mg, 0.027 mmol). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers, and stirred at room temperature over the course of 1 h. The reaction mixture was lyophilized to yield 27.6 mg of dark brown powder, which represented a mixture of $(^{\text{tbs}}\text{L})\text{Fe}_3(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))$ and $(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))$ (86% mass recovery).

$(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))(\text{py})$ (14**)**. To a frozen solution of $3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3\text{N}_3$ (12 mg, 0.047 mmol) in benzene (1 mL) was added crystalline **5** (47.7 mg, 0.048 mmol) in benzene (1 mL). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers, and stirred at room temperature over the course of 1 h. The volatile components of the reaction mixture were removed *in vacuo*, and the resulting solid was extracted in hexanes with drops of benzene. The dark brown extracts were filtered through a Celite plug, and maintenance of the filtrate at $-33\text{ }^{\circ}\text{C}$ yielded 30 mg of dark brown crystalline product (51%). ^1H NMR (500 MHz, C_6D_6): δ 216.0, 197.4, 135.1, 64.1, 60.6, 50.3, 32.2, 31.3, 25.8, 24.8, 20.9, 18.8, 13.3, 10.8, 8.8, 6.7, 5.4, 2.4, 1.2, 0.9, -0.7 , -1.9 , -3.3 , -5.3 , -9.1 , -11.1 , -14.9 , -15.8 , -20.0 , -21.6 , -23.5 , -24.4 , -43.5 , -45.4 , -48.6 . ^{19}F NMR (500 MHz, C_6D_6): δ -68.5 , -71.2 . ^{57}Fe Mössbauer (90 K) (δ , $|\Delta E_Q|$) (mm/s): component 1 (55%): 0.33, 1.76; component 2 (45%): 0.40, 2.92. Calc. for $\text{C}_{55}\text{H}_{74}\text{F}_6\text{Fe}_2\text{N}_8\text{Si}_3\text{Zn}$: C, 54.03; H, 6.10; N, 9.17. Found: C, 54.21; H, 5.79; N, 8.85. Fe/Zn ratio by XRF: 2.1:1.

$[\text{CoCp}^*_2][(^{\text{tbs}}\text{L})\text{Fe}_3(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))]$ (15**)**. To CoCp^*_2 (14.3 mg, 0.043 mmol) frozen in THF (1 mL) was added $(^{\text{tbs}}\text{L})\text{Fe}_3(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))$ (**13**) (49.1 mg, 0.043 mmol) in THF (1 mL). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers. The reaction mixture was stirred at room temperature, and the volatile components were removed *in vacuo* 1 h after thawing. The resulting solid was extracted in THF and filtered through a glass wool plug. Diffusion of hexanes into the filtrate solution at $-33\text{ }^{\circ}\text{C}$ yielded 44.8 mg of dark brown crystalline product (71%). ^1H NMR (500 MHz, $\text{THF-}d_8$): δ 56.1, 12.1, 9.7, 7.3, 7.0, 6.8, 5.8, 5.4, 2.2, 1.3, 1.0, 0.9, 0.4, -3.8 , -10.1 , -29.4 , -55.9 , -70.9 . ^{19}F NMR (500 MHz, $\text{THF-}d_8$): δ -85.9 . ^{57}Fe Mössbauer (90 K) (δ , $|\Delta E_Q|$) (mm/s): component 1 (67%): 0.49, 1.82 ($\Gamma = 0.35$ mm/s); component 2 (33%): 0.59, 1.20 ($\Gamma = 0.16$ mm/s). Calc. for $\text{C}_{70}\text{H}_{99}\text{CoF}_6\text{Fe}_3\text{N}_7\text{Si}_3$: C, 57.46; H, 6.82; N, 6.70. Found: C, 57.19; H, 6.87; N, 7.02.

$[(2,2,2\text{-cryptand})\text{K}][(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))]$ (16**)**. To a frozen suspension of KC_8 (3.8 mg, 0.028 mmol) in THF (1 mL) was added a solution of crystalline $(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))(\text{py})$ (**14**) (34.3 mg, 0.028 mmol) in THF (1 mL). The reaction mixture was frozen to keep the layers separate, then thawed to allow slow diffusion of the layers, and stirred at room temperature over the course of 1 h. The reaction mixture was filtered through Celite to a vial containing 2,2,2-cryptand (10.6 mg, 0.028 mmol), and the resulting mixture was stirred at room

temperature for 30 min. The volatile components of the reaction mixture were then removed *in vacuo*, and the resulting solid was extracted in diethyl ether. Filtration of the ethereal extracts, followed by maintenance of the filtrate at $-33\text{ }^{\circ}\text{C}$ yielded 30 mg of dark brown crystalline product (69% yield). ^1H NMR (500 MHz, THF- d_8): δ 237.7, 156.3, 126.9, 47.7, 45.9, 23.5, 20.8, 16.9, 14.8, 13.8, 10.3, 9.2, 7.0, 5.2, 2.6, 1.3, 0.9, -0.3 , -1.2 , -2.5 , -2.8 , -4.4 , -5.6 , -7.5 , -9.4 , -18.0 , -19.2 , -23.2 , -41.8 , -52.7 . ^{19}F NMR (500 MHz, THF- d_8): δ -79.4 . ^{57}Fe Mössbauer (90 K) (δ , $|\Delta E_Q|$) (mm/s): component 1 (50%): 0.35, 1.73 ($\Gamma = 0.29$ mm/s); component 2 (%): 0.59, 3.01 ($\Gamma = 0.20$ mm/s). Calc. for $\text{C}_{68}\text{H}_{105}\text{F}_6\text{Fe}_2\text{KN}_9\text{O}_6\text{Si}_3\text{Zn}$: C, 52.39; H, 6.79; N, 8.09. Found: C, 52.24; H, 6.73; N, 8.51. Fe/Zn ratio by XRF: 1.9:1.

Reaction of $[\text{NBu}_4][(\text{tbsL})\text{Fe}_3(\mu^3\text{-Cl})]$ (7) and $[\text{NBu}_4][(\text{tbsL})\text{Zn}_3(\mu^3\text{-Cl})]$ (8). Crystalline $[\text{NBu}_4][(\text{tbsL})\text{Fe}_3(\mu^3\text{-Cl})]$ (7) (10.1 mg, 0.00853 mmol) and $[\text{NBu}_4][(\text{tbsL})\text{Zn}_3(\mu^3\text{-Cl})]$ (8) (10.3 mg, 0.00853 mmol) were separately dissolved in THF (0.2 mL) and combined in a J. Young NMR tube. Three 0.05 mL aliquots of THF were used to rinse the vials used in sample preparation, and the aliquots were added to the reaction mixture. ^1H NMR spectra collected within 1 h of sample preparation and after 24 h were identical. The sample was then subjected to heating at $40\text{ }^{\circ}\text{C}$ for 24 h and then at $60\text{ }^{\circ}\text{C}$ for 24 h; ^1H NMR spectra recorded after each heating period showed no change. Heating at $80\text{ }^{\circ}\text{C}$ for 24 h resulted in formation of small quantities of the metal-atom redistribution product $[\text{NBu}_4][(\text{tbsL})\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl})]$ (9). These ^1H NMR spectra are presented in Figure S21.

Reaction of $(\text{tbsL})\text{Fe}_2\text{Zn}(\text{thf})$ (5) and $[\text{NBu}_4][(\text{tbsL})\text{Fe}_3(\mu^3\text{-Cl})]$ (7). Crystalline $(\text{tbsL})\text{Fe}_2\text{Zn}(\text{thf})$ (5) (10.2 mg, 0.0103 mmol) and $[\text{NBu}_4][(\text{tbsL})\text{Fe}_3(\mu^3\text{-Cl})]$ (7) (12.2 mg, 0.0103 mmol) were separately dissolved in THF/ C_6D_6 (25% v/v; 0.2 mL) and combined in a J. Young NMR tube. Three 0.05 mL aliquots of the solvent mixture were used to rinse the vials used in sample preparation, and the aliquots were added to the reaction mixture. ^1H NMR spectra were recorded after 4 h and 24 h of maintenance at room temperature, and after heating at $60\text{ }^{\circ}\text{C}$ for 24 h. These spectra indicate that the metal-atom redistribution product $[\text{NBu}_4][(\text{tbsL})\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl})]$ (9) was generated rapidly at room temperature after 4 h, with 5 and 7 being no longer observable. No further change was observed after heating at $60\text{ }^{\circ}\text{C}$ for 24 h. These ^1H NMR spectra are presented in Figure S22.

Reaction of $(\text{tbsL})\text{Fe}_2\text{Zn}(\text{thf})$ (5) and $[\text{NBu}_4][(\text{tbsL})\text{Zn}_3(\mu^3\text{-Cl})]$ (8). Crystalline $(\text{tbsL})\text{Fe}_2\text{Zn}(\text{thf})$ (5) (10.7 mg, 0.0108 mmol) and $[\text{NBu}_4][(\text{tbsL})\text{Zn}_3(\mu^3\text{-Cl})]$ (8) (13.1 mg, 0.0108 mmol) were separately dissolved in THF/ C_6D_6 (25% v/v; 0.2 mL) and combined in a J. Young NMR tube. Three 0.05 mL aliquots of the solvent mixture were used to rinse the vials used in sample preparation, and the aliquots were added to the reaction mixture. ^1H NMR spectra were recorded after 4 h and 24 h of maintenance at room temperature, and after heating at $60\text{ }^{\circ}\text{C}$ for 24 h. These spectra indicate that the metal-atom redistribution product $[\text{NBu}_4][(\text{tbsL})\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl})]$ (9) was generated slowly at room temperature, and the reaction mixture is predominantly composed of 5 and 8. At elevated temperatures, conversion to 9 is increased, but 5 and 8 persist in solution. These ^1H NMR spectra are presented in Figure S23.

Reaction of $(\text{tbsL})\text{Fe}_2\text{Zn}(\text{thf})$ (5) with FeCl_2 . A suspension of FeCl_2 (1.3 mg, 0.010 mmol) in THF (0.2 mL) was added to a solution of $(\text{tbsL})\text{Fe}_2\text{Zn}(\text{thf})$ (5) (10.1 mg; 0.0102 mmol) in THF (0.2 mL). Three 0.05 mL aliquots of THF were used to rinse the vials used in sample preparation, and the aliquots were added to the reaction mixture. ^1H NMR spectra were recorded after 4 h and 24 h of

maintenance at room temperature. These spectra indicate that metal-atom redistribution product $[\text{NBu}_4][(\text{t}^{\text{bs}}\text{L})\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl})]$ (**9**) and tri-ferrous $[\text{NBu}_4][(\text{t}^{\text{bs}}\text{L})\text{Fe}_3(\mu^3\text{-Cl})]$ **7** were generated at room temperature, although small quantities of **5** remain present. Several unidentified resonances attributed to paramagnetic species are also present in the ^1H NMR spectra, and increase in intensity from 4 h to 24 h. Over this period, the color of the reaction mixture changed from dark brown to purple; this process was further accompanied by the deposition of a precipitate. These ^1H NMR spectra are presented in Figure S24.

Reaction of $(\text{t}^{\text{bs}}\text{L})\text{Fe}_2\text{Zn}(\text{thf})$ (5**) with ZnCl_2 .** A solution of ZnCl_2 (1.1 mg, 0.0081 mmol) in THF (0.2 mL) was added to a solution of $(\text{t}^{\text{bs}}\text{L})\text{Fe}_2\text{Zn}(\text{thf})$ (**5**) (8.0 mg, 0.0081 mmol) in THF (0.2 mL). Three 0.05 mL aliquots of THF were used to rinse the vials used in sample preparation, and the aliquots were added to the reaction mixture. ^1H NMR spectra were recorded after 4 h and 24 h of maintenance at room temperature. These spectra indicate that no metal-scrambling products were generated at room temperature, and **5** persisted in solution. However, the intensity of all resonances diminished noticeably after 24 h, and this was accompanied by a change in the color of the reaction mixture from dark brown to purple, although no precipitation was evident. These ^1H NMR spectra are presented in Figure S25.

Zero Field ^{57}Fe Mössbauer Spectroscopy

Table S1. ^{57}Fe Mössbauer Parameters for Selected Complexes

Compound	δ (mms/s)	$ \Delta E_Q $ (mm/s)	%
$(^{\text{tbs}}\text{L})\text{Fe}_3(\text{thf})^1$ (1)	0.37	1.99	36
	0.46	1.52	36
	0.88	1.29	28
$(^{\text{tbs}}\text{L})\text{Fe}_3(\text{py})$ (2)	0.33	1.84	38
	0.55	1.76	28
	0.74	1.39	33
$[\text{Na}(\text{thf})][(^{\text{tbs}}\text{L})\text{Fe}_2\text{Na}(\text{thf})]$ (4)	0.47	1.51	50
	0.54	0.87	50
$(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\text{thf})$ (5)	0.60	1.83	50
	0.84	1.55	50
$(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\text{py})$ (6)	0.61	1.68	50
	0.79	1.63	50
$[\text{NBu}_4][(^{\text{tbs}}\text{L})\text{Fe}_3(\mu^3\text{-Cl})]$ (7)	0.72	1.31	100
$[\text{NBu}_4][(^{\text{tbs}}\text{L})\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl})]$ (9)	0.77	1.13	100
$(^{\text{tbs}}\text{L})\text{Fe}_3(\mu^3\text{-NPh})^{1b}$ (10)	0.42	1.97	67
	0.42	1.09	33
$(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\mu^3\text{-NPh})$ (11)	0.28	2.31	100
$(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\mu^3\text{-NPh})(\text{py})$ (12)	0.33	1.80	51
	0.35	2.76	49
$(^{\text{tbs}}\text{L})\text{Fe}_3(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))$ (13)	0.37	1.80	33
	0.38	1.13	33
	0.45	2.61	33
$(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))(\text{py})$ (14)	0.33	1.76	55
	0.40	2.92	45
$[\text{CoCp}^*_2][(^{\text{tbs}}\text{L})\text{Fe}_3(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))]$ (15)	0.49	1.82	67
	0.59	1.20	33
$[2,2,2\text{-crypt}(\text{K})][(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))]$ (16)	0.35	1.73	50
	0.59	3.01	50

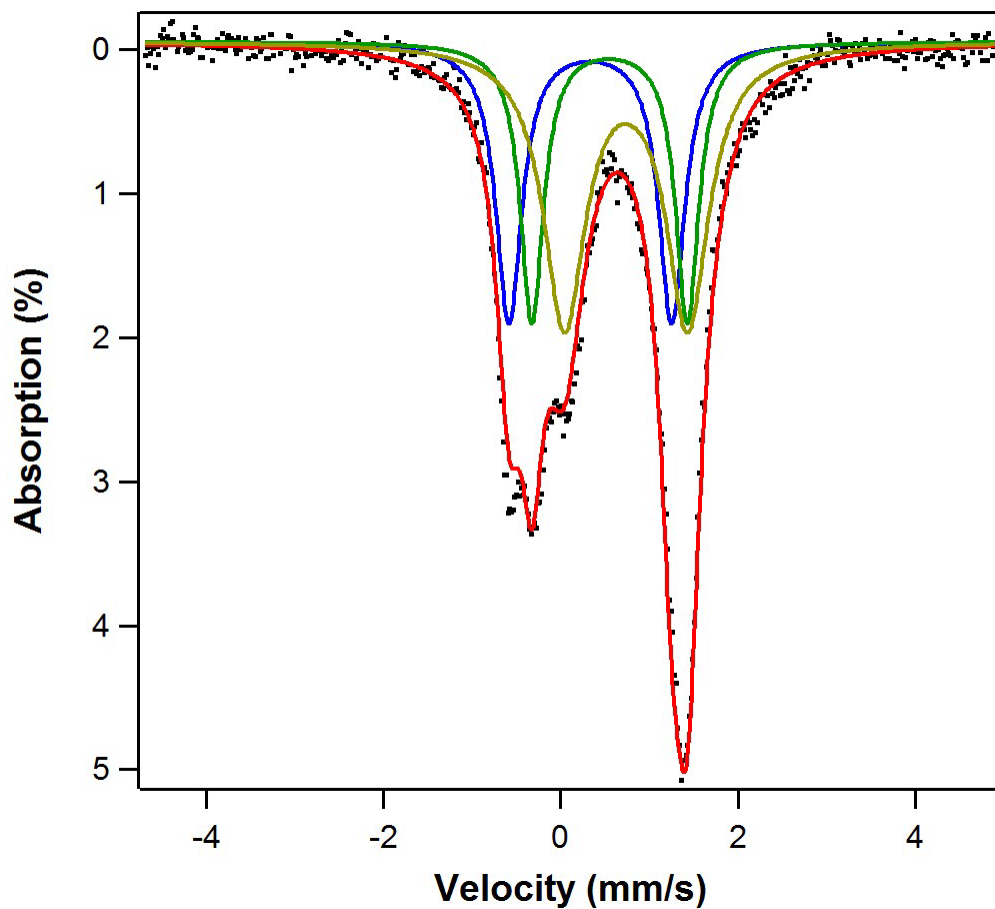


Figure S1. Zero-field ^{57}Fe Mössbauer spectrum of $(^{\text{tbsL}})\text{Fe}_3(\text{py})$ (**2**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) component 1 (blue, 38%) 0.33, 1.84 ($\Gamma = 0.18 \text{ mm/s}$); component 2 (green, 28%) 0.55, 1.76 ($\Gamma = 0.16 \text{ mm/s}$); component 3 (yellow, 33%) 0.74, 1.39 ($\Gamma = 0.29 \text{ mm/s}$).

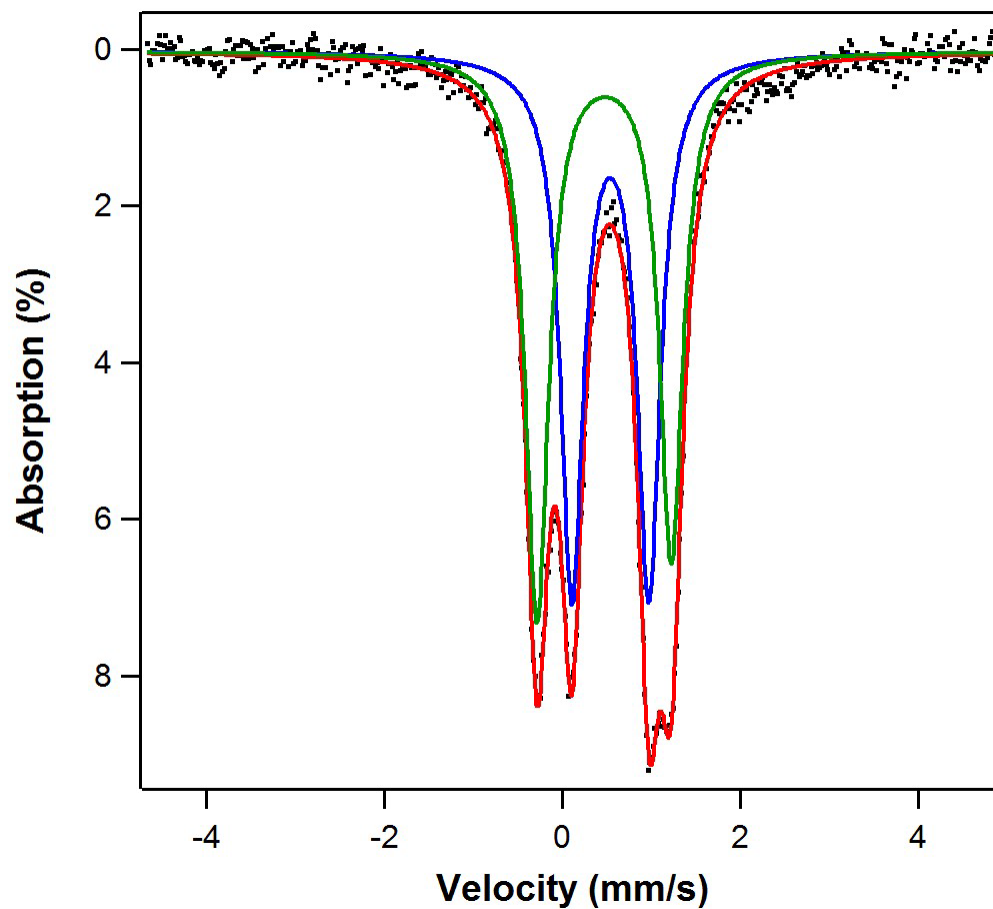


Figure S2. Zero-field ^{57}Fe Mössbauer spectrum of $[\text{Na}(\text{thf})][(\text{tbsL})\text{Fe}_2\text{Na}(\text{thf})]$ (**4**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) component 1 (blue, 50%) 0.54, 0.87 ($\Gamma = 0.16 \text{ mm/s}$); component 2 (green, 50%) 0.47, 1.51 ($\Gamma = 0.16 \text{ mm/s}$).

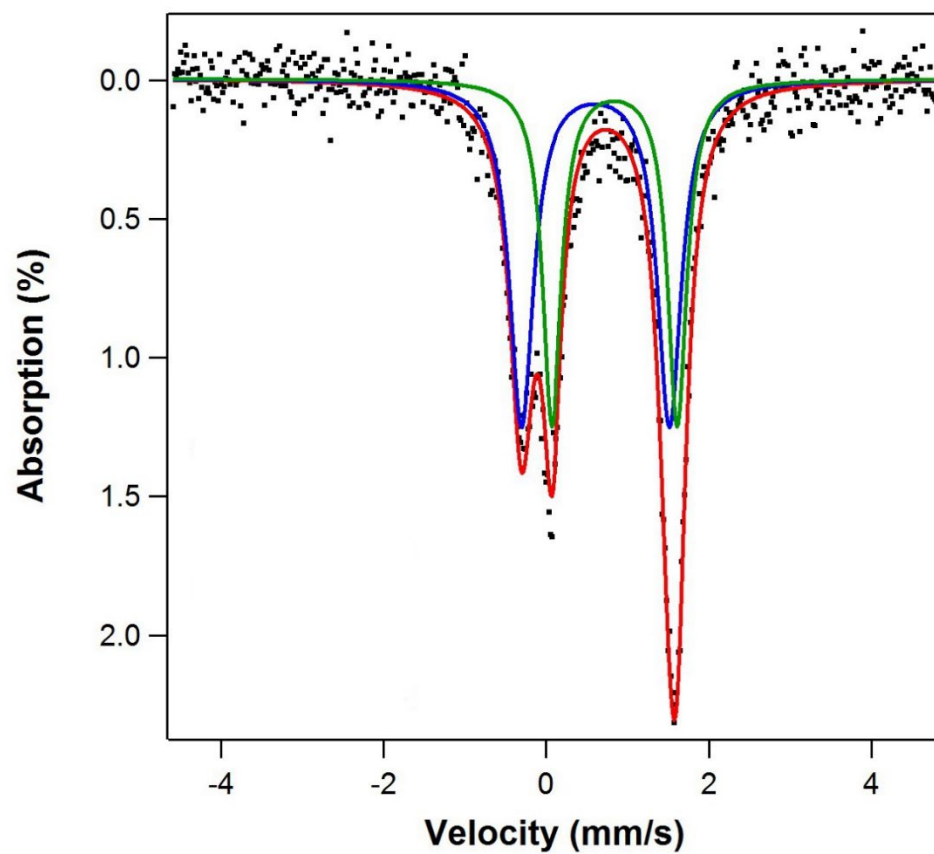


Figure S3. Zero-field ^{57}Fe Mössbauer spectrum of $(\text{tbsL})\text{Fe}_2\text{Zn}(\text{thf})$ (**5**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) component 1 (blue, 50%) 0.60, 1.83 ($\Gamma = 0.18 \text{ mm/s}$); component 2 (green, 50%) 0.84, 1.55 ($\Gamma = 0.14 \text{ mm/s}$).

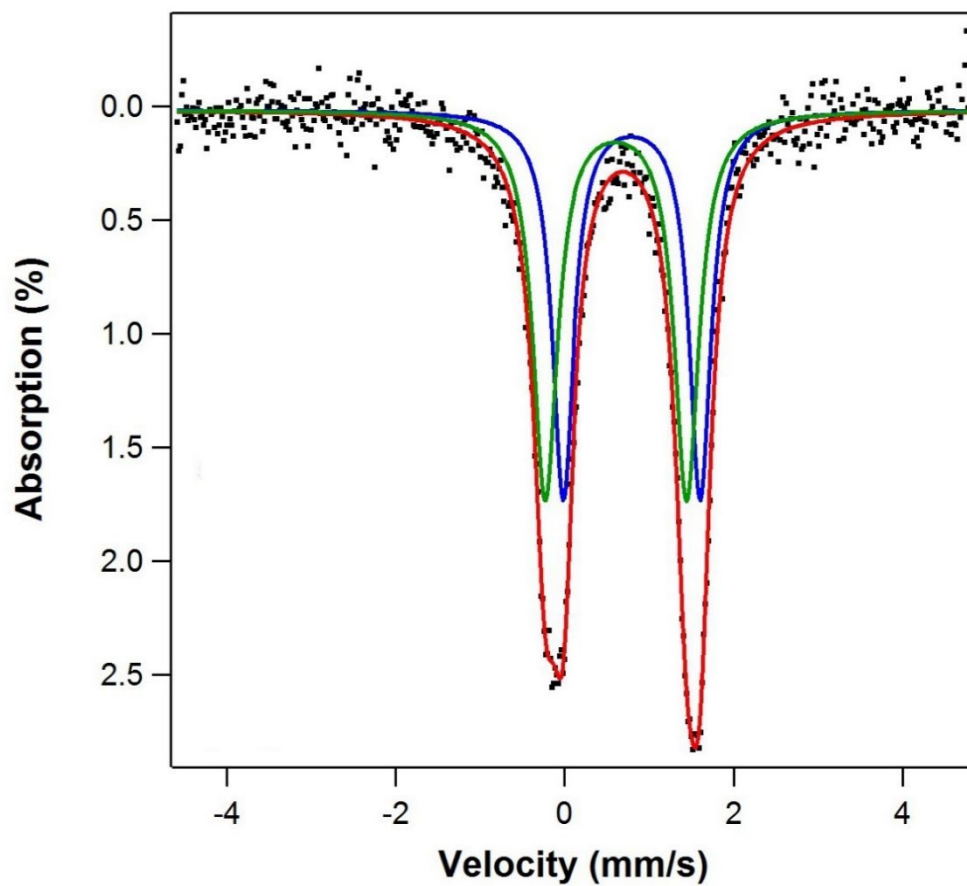


Figure S4. Zero-field ^{57}Fe Mössbauer spectrum of $(^{\text{tbsL}})\text{Fe}_2\text{Zn}(\text{py})$ (**6**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) component 1 (blue, 50%) 0.79, 1.63 ($\Gamma = 0.16 \text{ mm/s}$); component 2 (green, 50%) 0.61, 1.68 ($\Gamma = 0.18 \text{ mm/s}$).

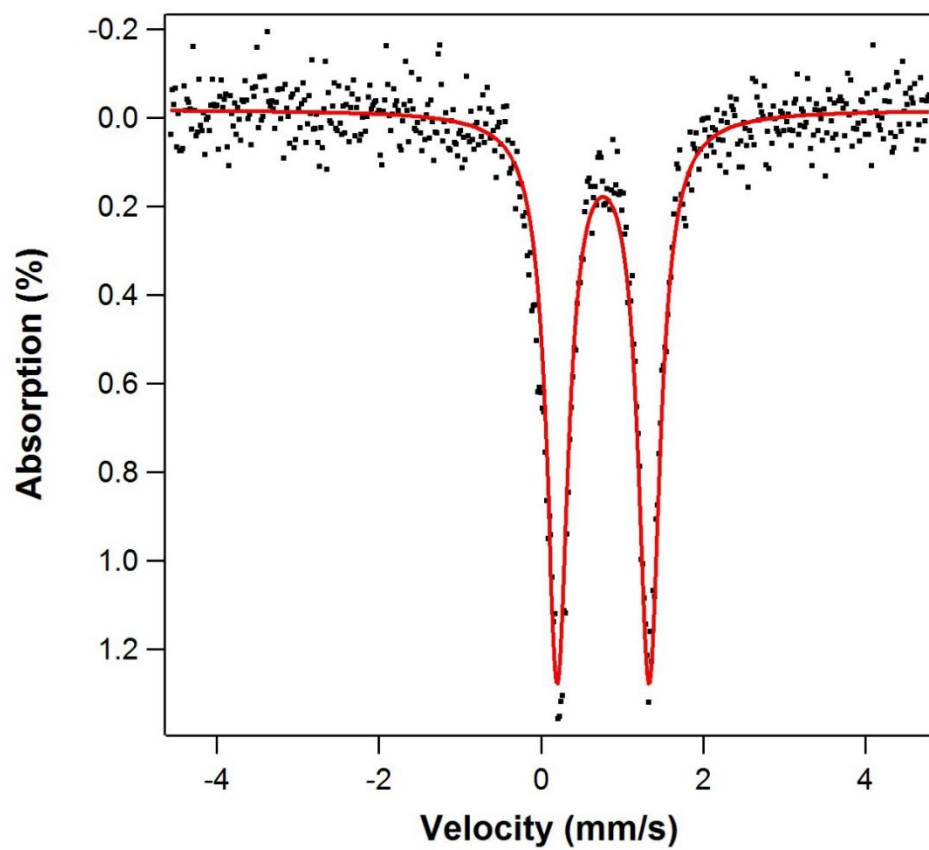


Figure S5. Zero-field ^{57}Fe Mössbauer spectrum of $[\text{NBu}_4][\{(\text{tbs})\text{L}\}\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl})]$ (**9**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) 0.77, 1.13 ($\Gamma = 0.16 \text{ mm/s}$).

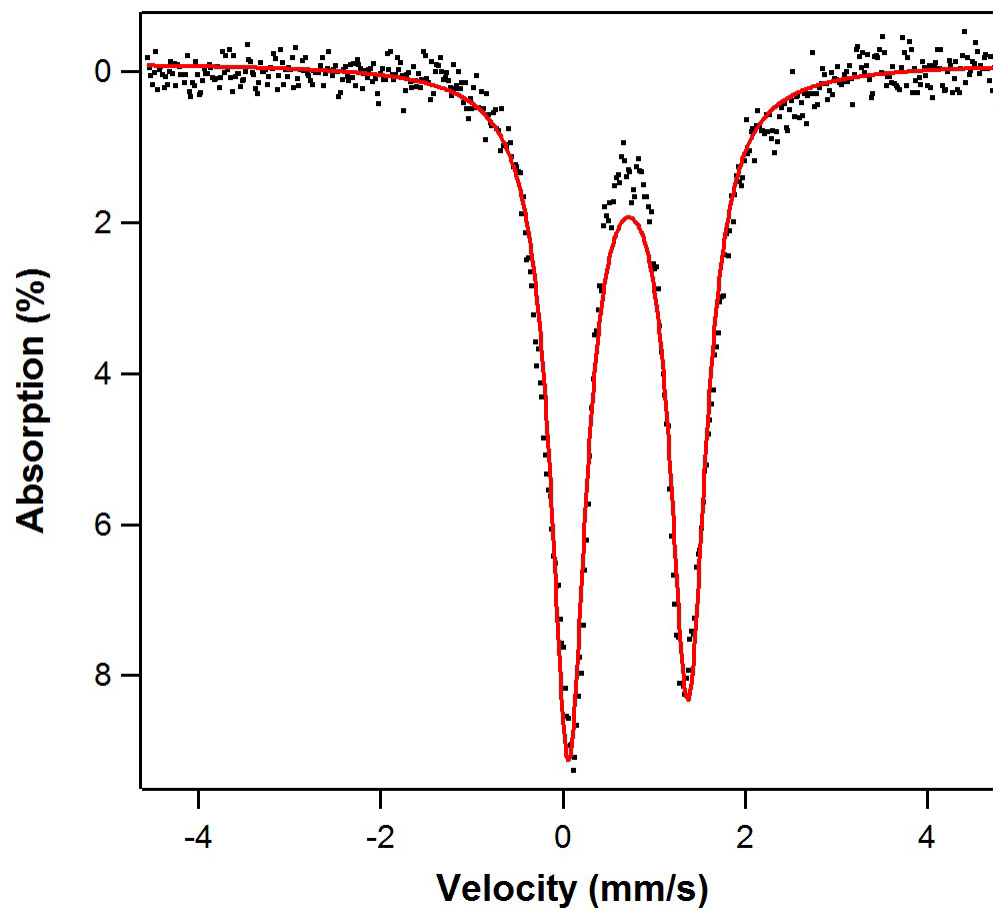


Figure S6. Zero-field ^{57}Fe Mössbauer spectrum of $[\text{NBu}_4][[(\text{tb}^{\text{SL}})\text{Fe}_3(\mu^3\text{-Cl})]$ (**7**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) 0.72, 1.31 ($\Gamma = 0.24 \text{ mm/s}$).

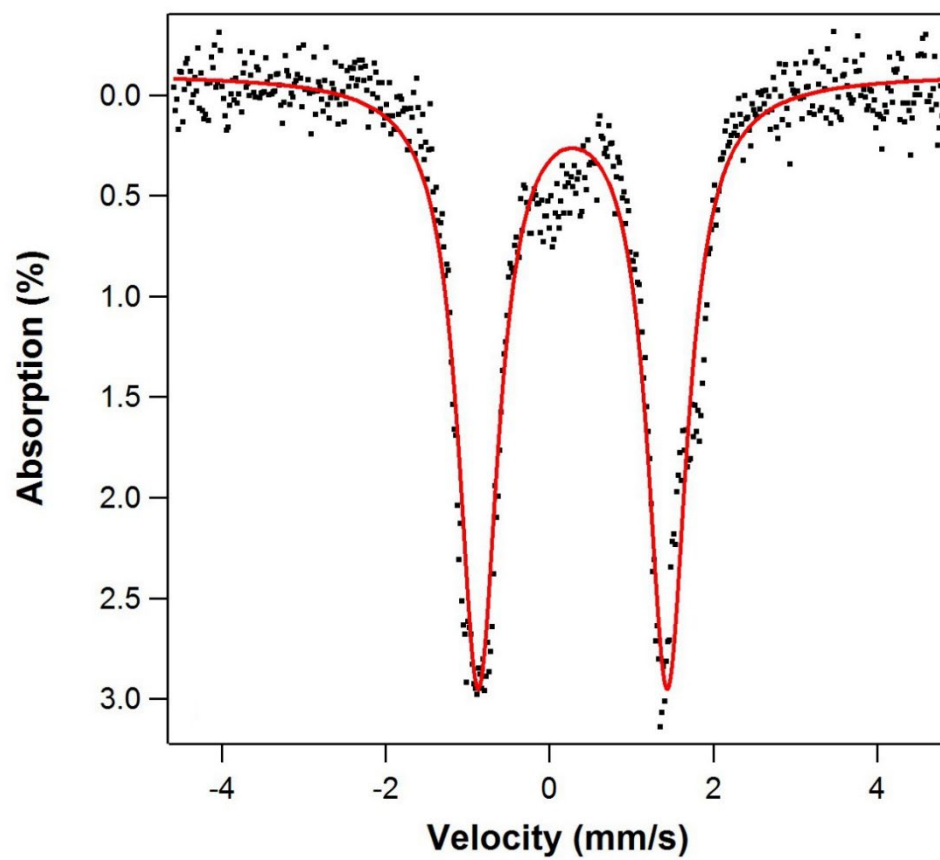


Figure S7. Zero-field ^{57}Fe Mössbauer spectrum of crude $(^{tbSL})\text{Fe}_2\text{Zn}(\mu^3\text{-NPh})$ (**11**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) 0.28, 2.31 ($\Gamma = 0.30 \text{ mm/s}$).

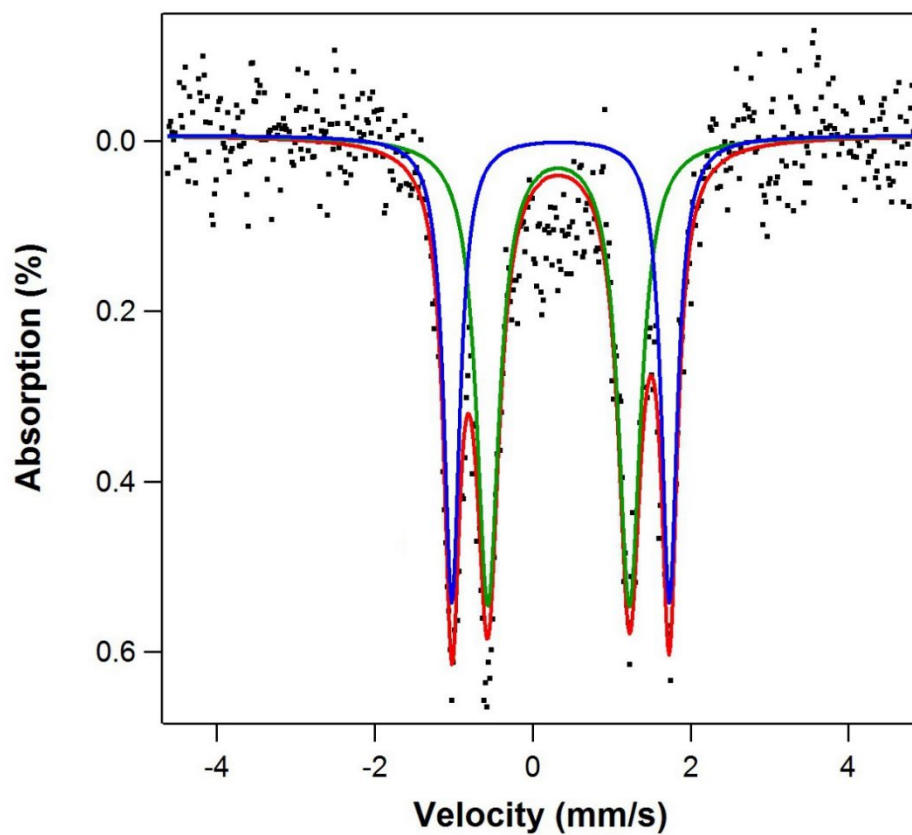


Figure S8. Zero-field ^{57}Fe Mössbauer Spectrum of $(\text{tbsL})\text{Fe}_2\text{Zn}(\mu^3\text{-NPh})(\text{py})$ (**12**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) component 1 (blue, 49%) 0.35, 2.76 ($\Gamma = 0.12 \text{ mm/s}$); component 2 (green, 51%) 0.33, 1.80 ($\Gamma = 0.17 \text{ mm/s}$).

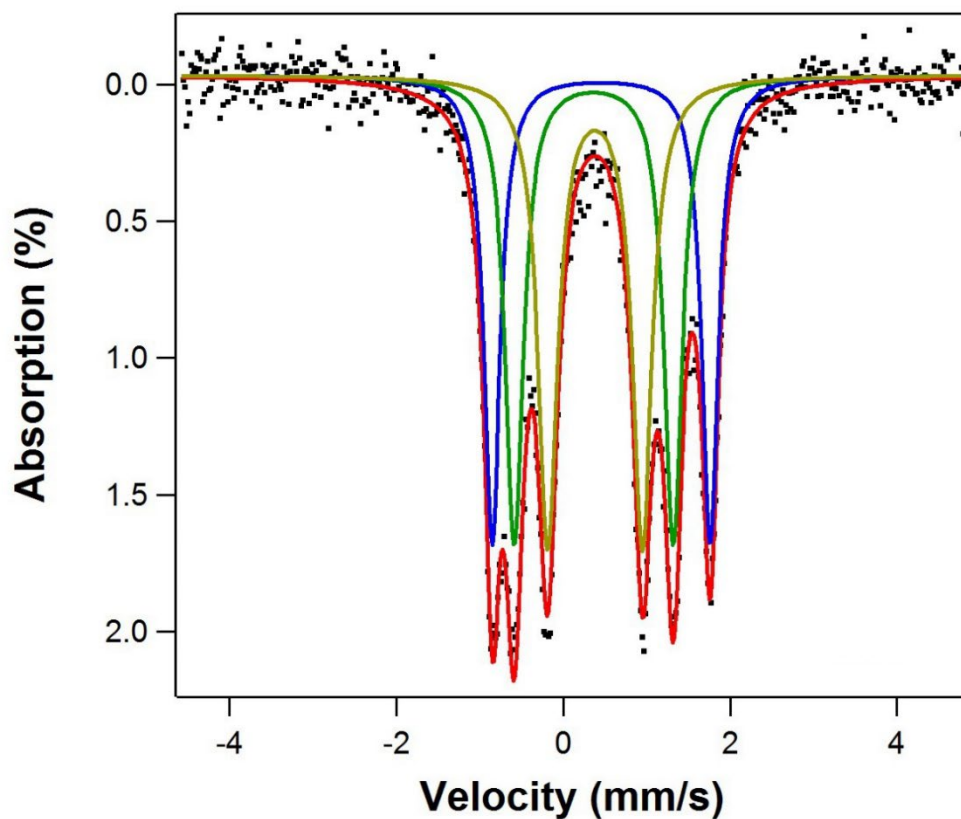


Figure S9. Zero-field ^{57}Fe Mössbauer Spectrum of $(\text{tbsL})\text{Fe}_3(\mu^3\text{-N}(3,5\text{-(CF}_3)_2\text{C}_6\text{H}_3))$ (**13**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) component 1 (blue, 33%) 0.45, 2.61 ($\Gamma = 0.12 \text{ mm/s}$); component 2 (green, 33%) 0.37, 1.80 ($\Gamma = 0.13 \text{ mm/s}$); component 3 (yellow, 34%) 0.38, 1.13 ($\Gamma = 0.14 \text{ mm/s}$).

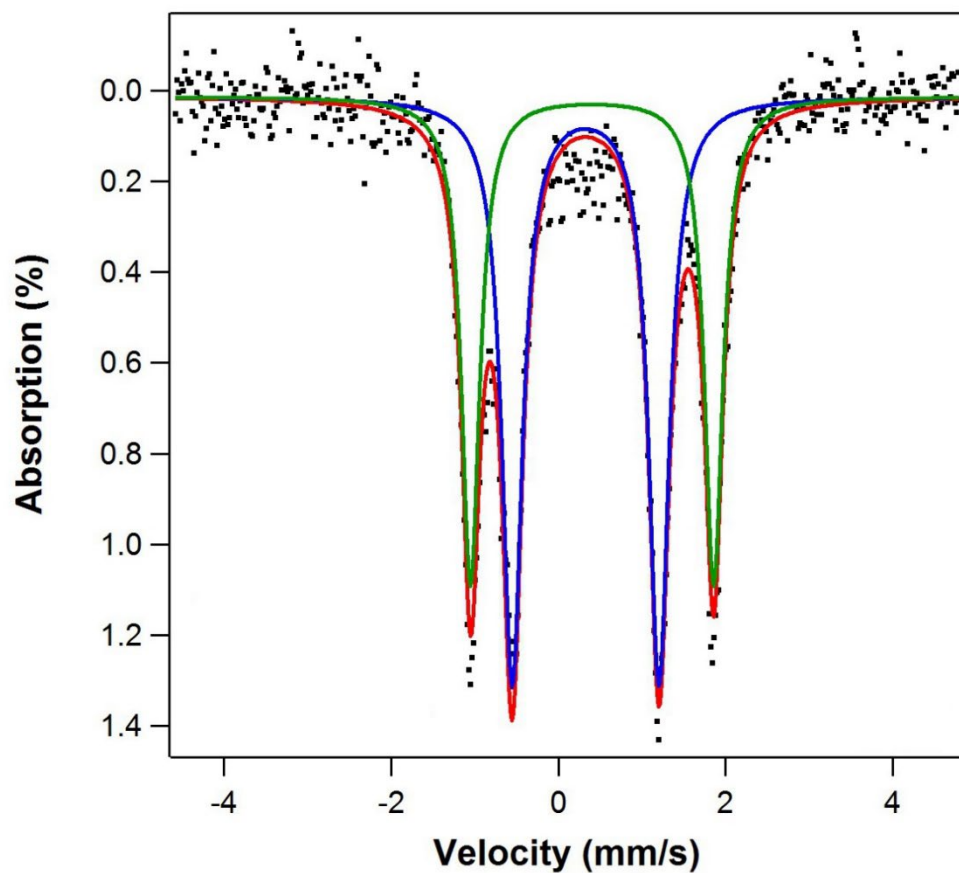


Figure S10. Zero-field ^{57}Fe Mössbauer Spectrum of $(^{\text{tbsL}})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))(\text{py})$ (**14**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) component 1 (green, 46%) 0.33, 1.76 ($\Gamma = 0.15 \text{ mm/s}$); component 2 (blue, 54%) 0.40, 2.92 ($\Gamma = 0.13 \text{ mm/s}$).

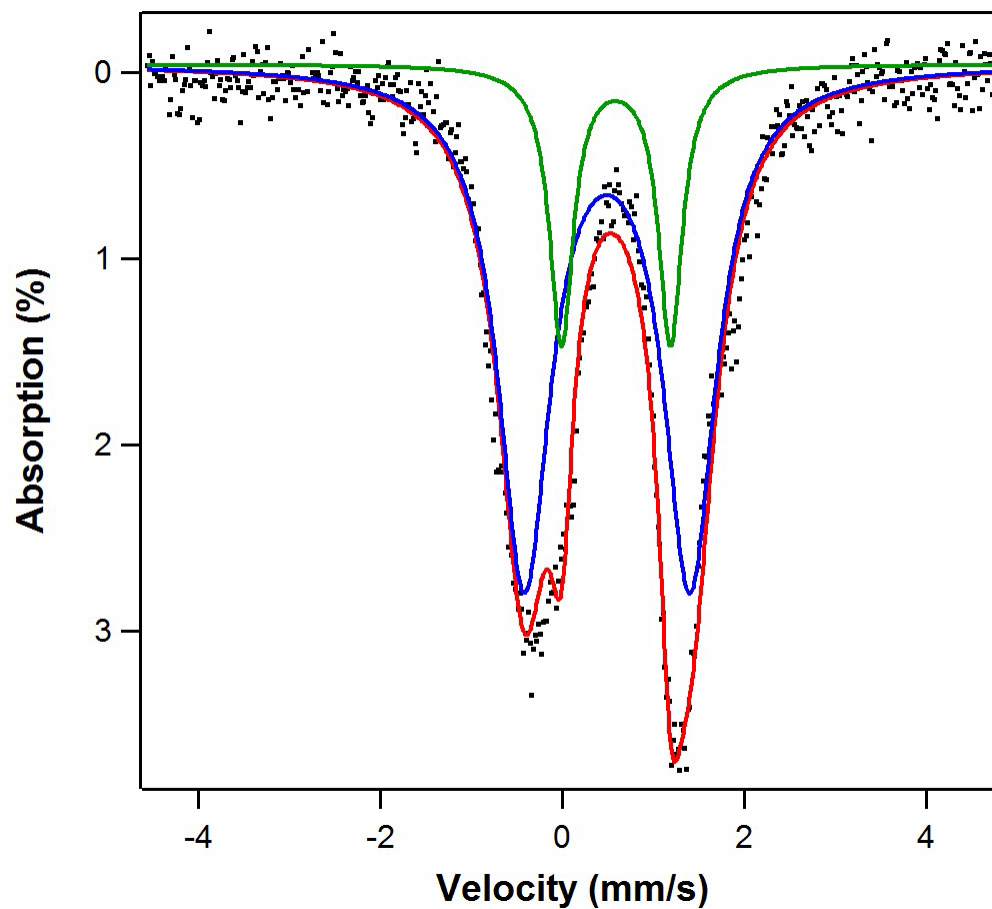


Figure S11. Zero-field ^{57}Fe Mössbauer Spectrum of $[\text{CoCp}^*_2][(\text{tbsL})\text{Fe}_3(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))]$ (**15**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) component 1 (blue, 67%) 0.49, 1.82 ($\Gamma = 0.35 \text{ mm/s}$); component 2 (green, 33%) 0.59, 1.20 ($\Gamma = 0.16 \text{ mm/s}$).

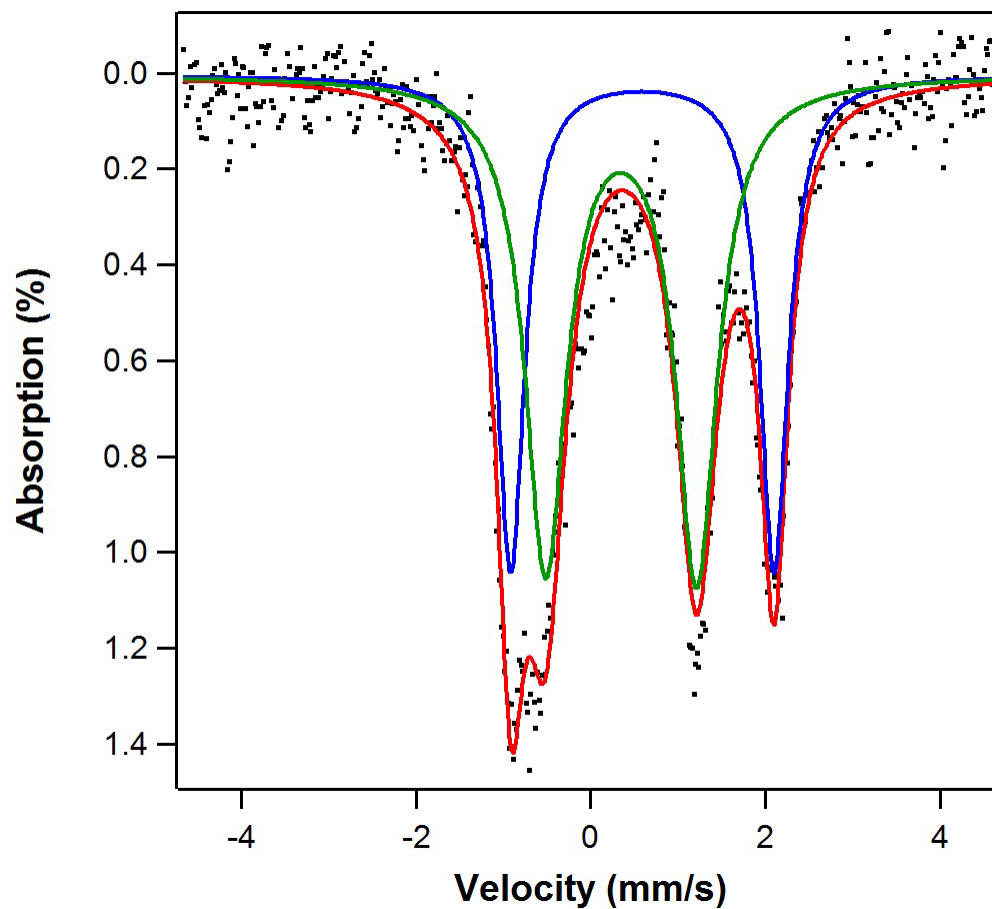


Figure S12. Zero-field ^{57}Fe Mössbauer Spectrum of $[2,2,2\text{-cryptand}(\text{K})][(\text{tbsL})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))]$ (**16**) collected at 90 K. Simulation yields the following parameters: δ , $|\Delta E_Q|$ (mm/s) component 1 (green, 50%) 0.35, 1.73 ($\Gamma = 0.29 \text{ mm/s}$); component 2 (blue, 50%) 0.59, 3.01 ($\Gamma = 0.20 \text{ mm/s}$).

X-Ray Fluorescence Spectroscopy

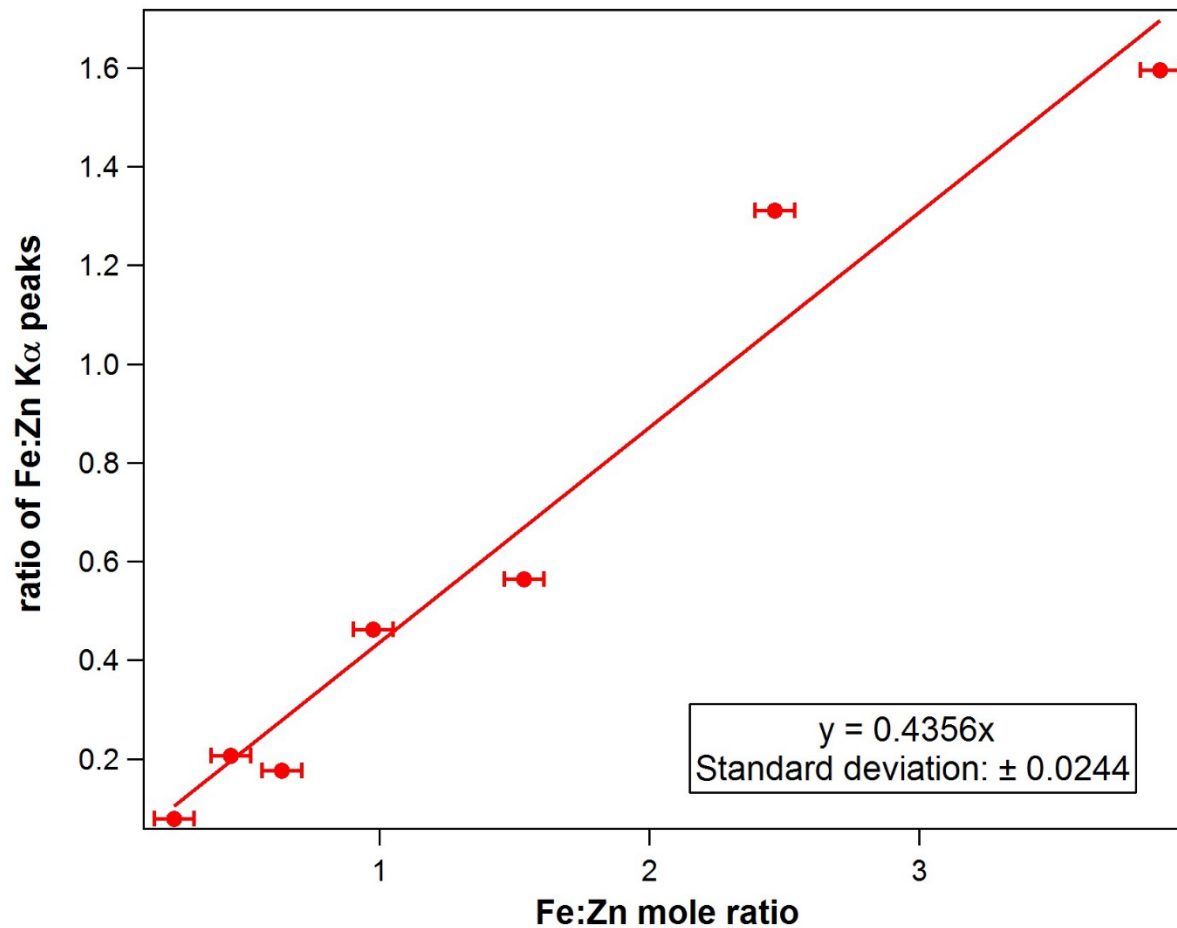


Figure S13. Solid-state X-ray fluorescence calibration curve obtained from homogeneous powder samples of ferrocene and bis[4,4,4-trifluoro-1-(2-thienyl-1,3-butanedionato)]zinc·(TMEDA).

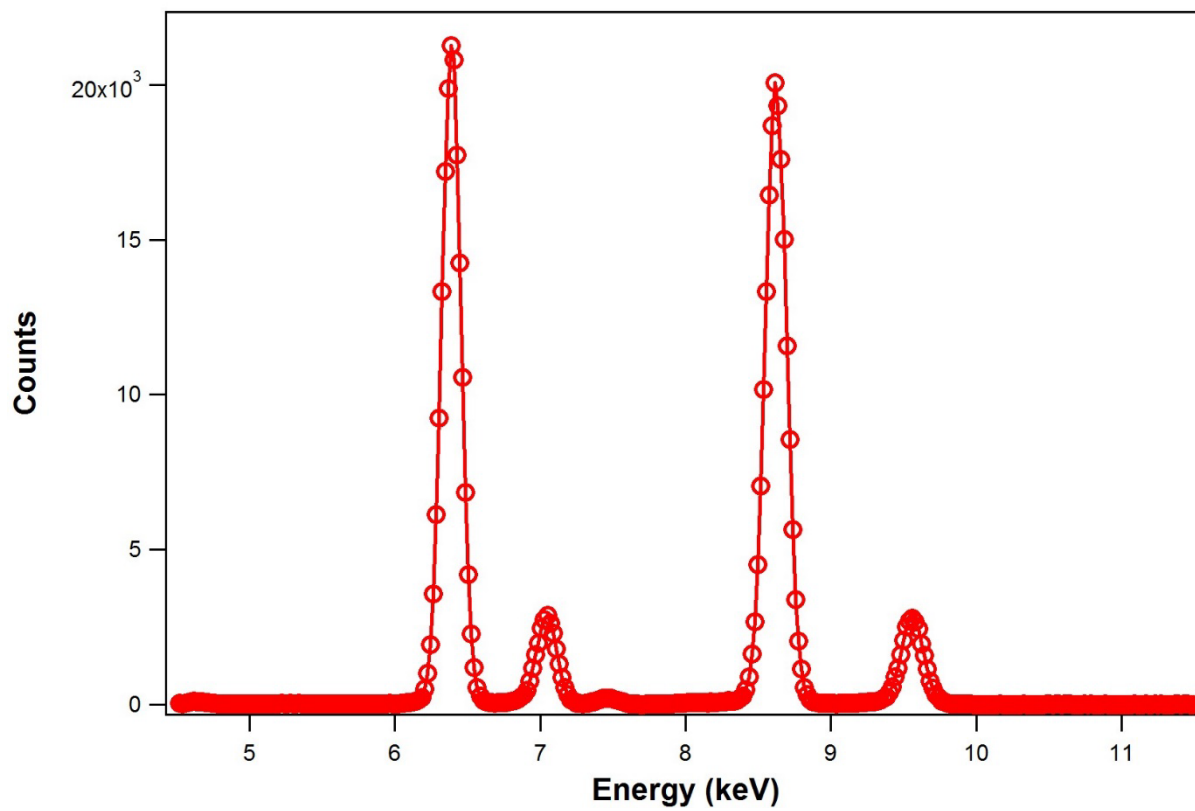


Figure S14. X-ray fluorescence spectrum of $(^{tb^sL})Fe_2Zn(thf)$ (**5**).

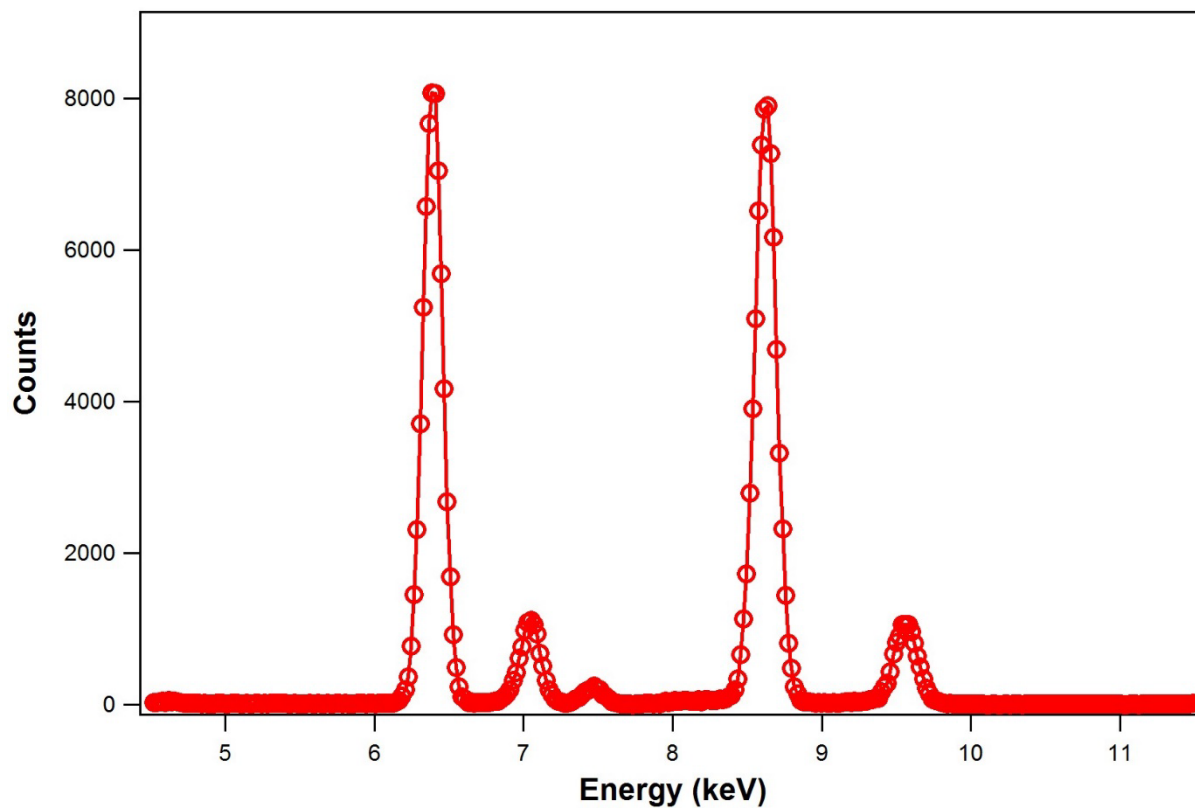


Figure S15. X-ray fluorescence spectrum of $(^{tb^sL})Fe_2Zn(py)$ (**6**).

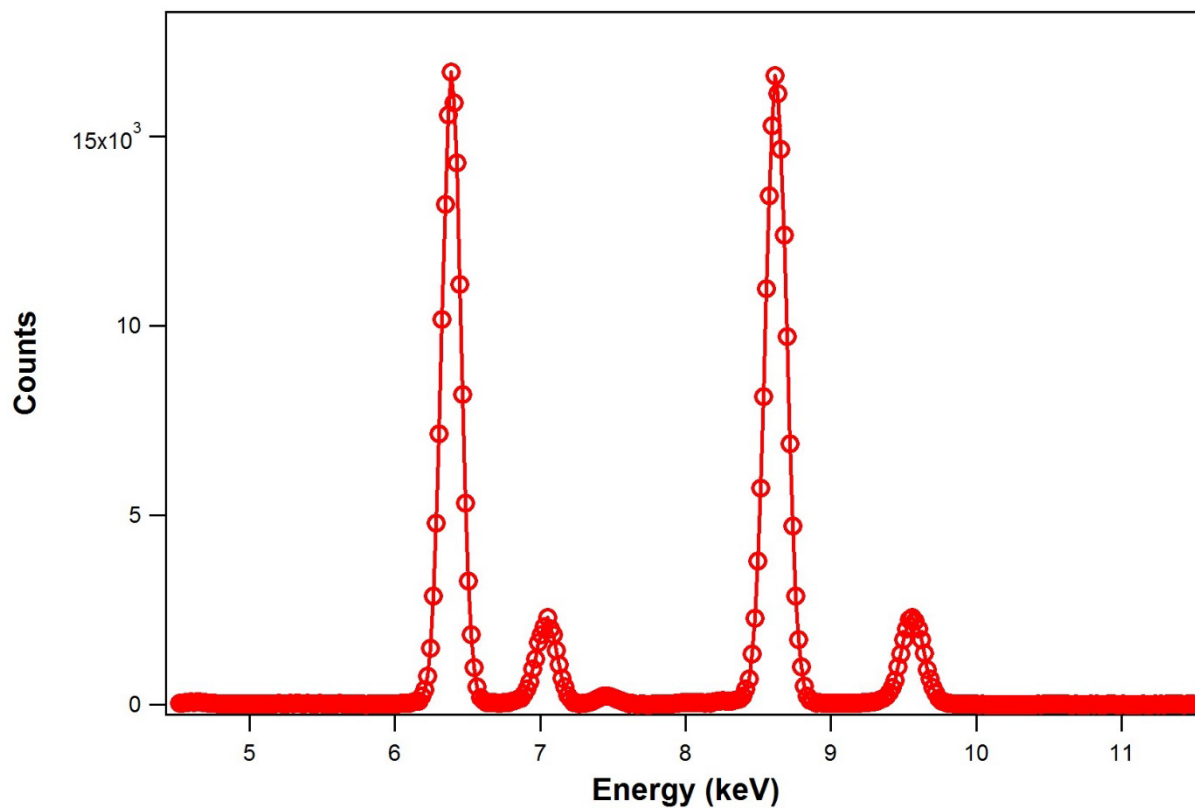


Figure S16. X-ray fluorescence spectrum of $[\text{NBu}_4][((\text{tbsL})\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl}))]$ (**9**).

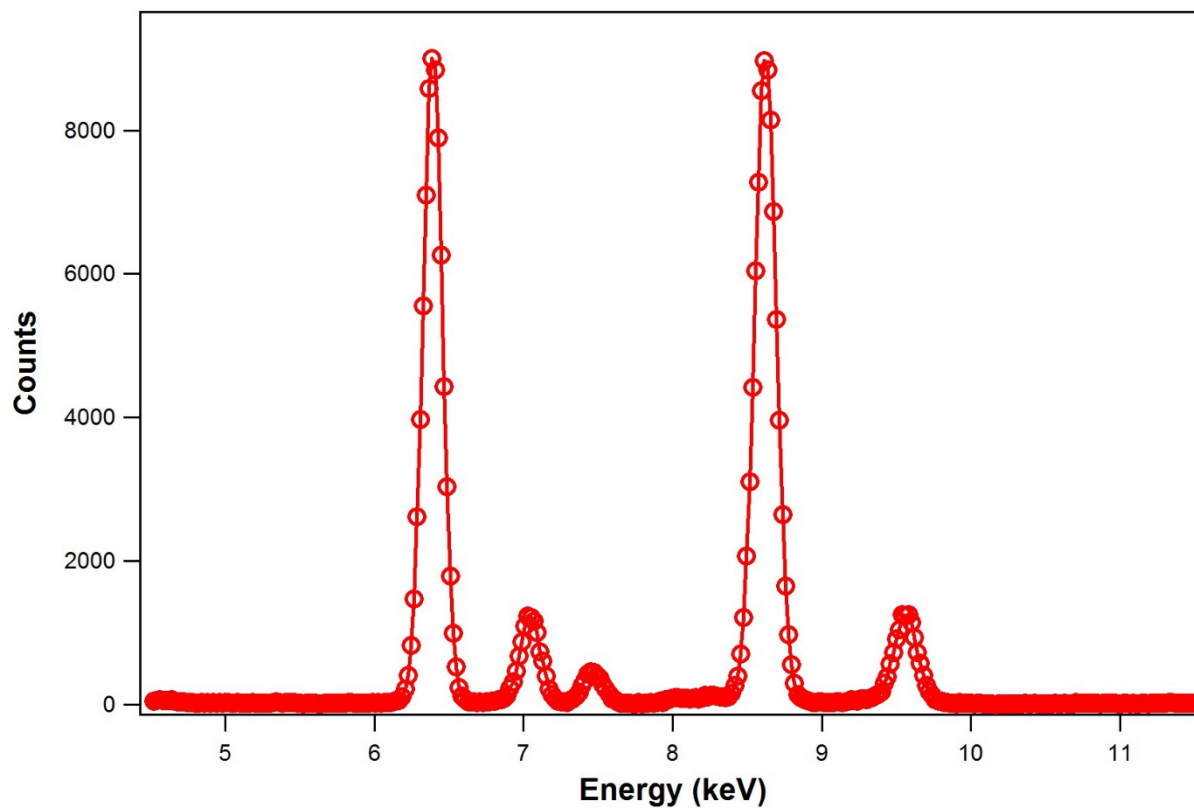


Figure S17. X-ray fluorescence spectrum of $(^{tb^sL})Fe_2Zn(\mu^3-NPh)(py)$ (**12**).

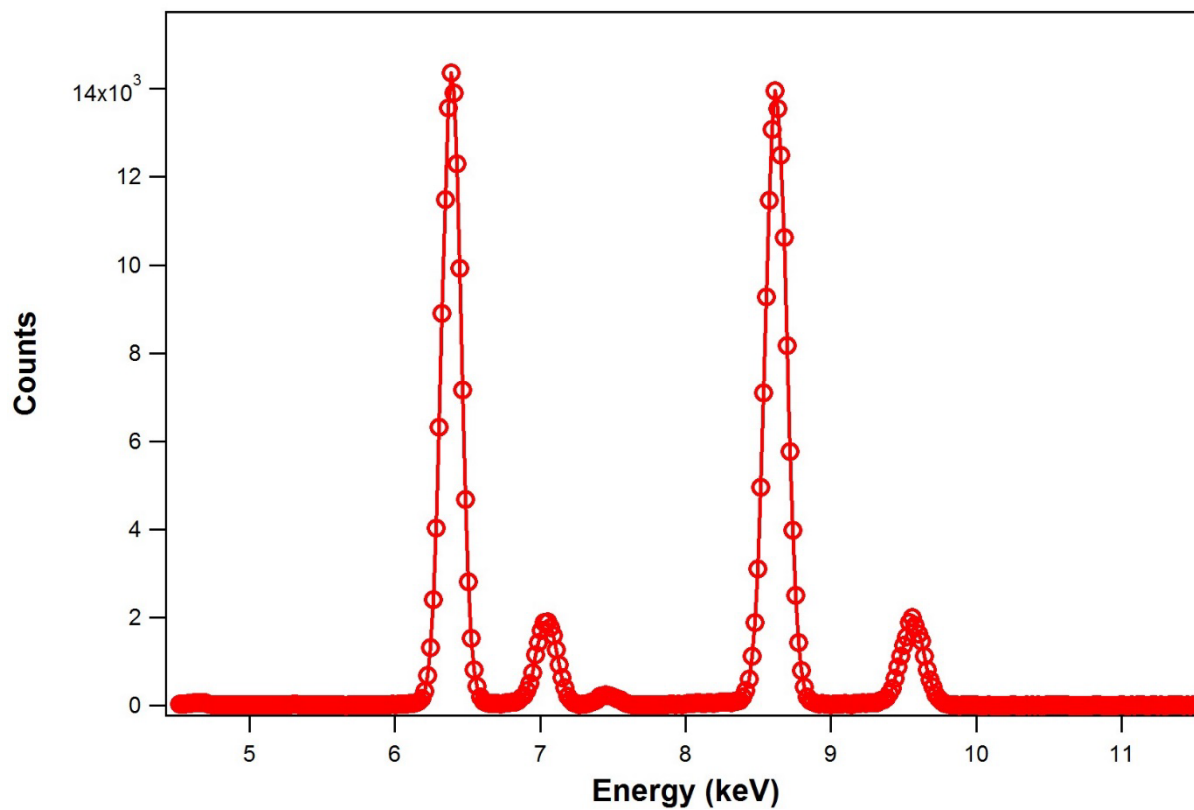


Figure S18. X-ray fluorescence spectrum of $(^{tbsL})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-(CF}_3)_2\text{C}_6\text{H}_3))(\text{py})$ (**14**).

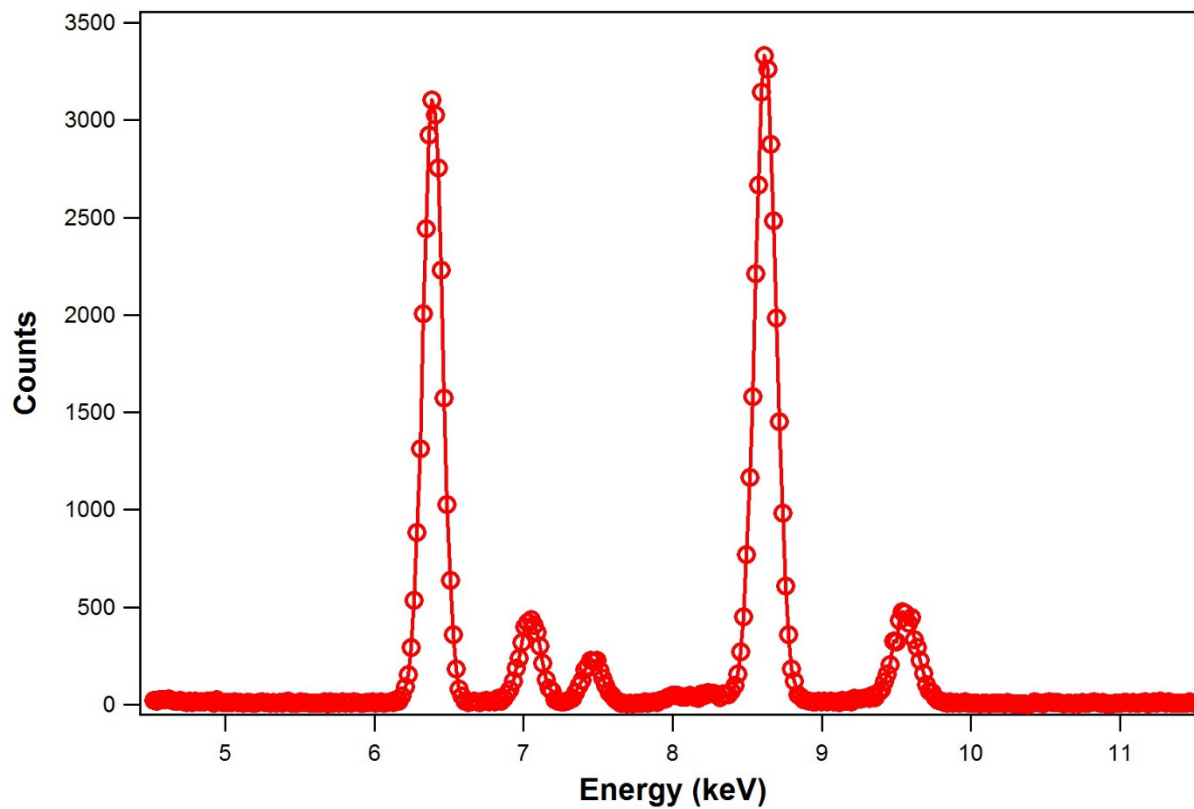


Figure S19. X-ray fluorescence spectrum of [2,2,2-cryptand(K)][(^{tbSL})Fe₂Zn(μ^3 -N(3,5-(CF₃)₂C₆H₃))] (**16**).

NMR Spectroscopy

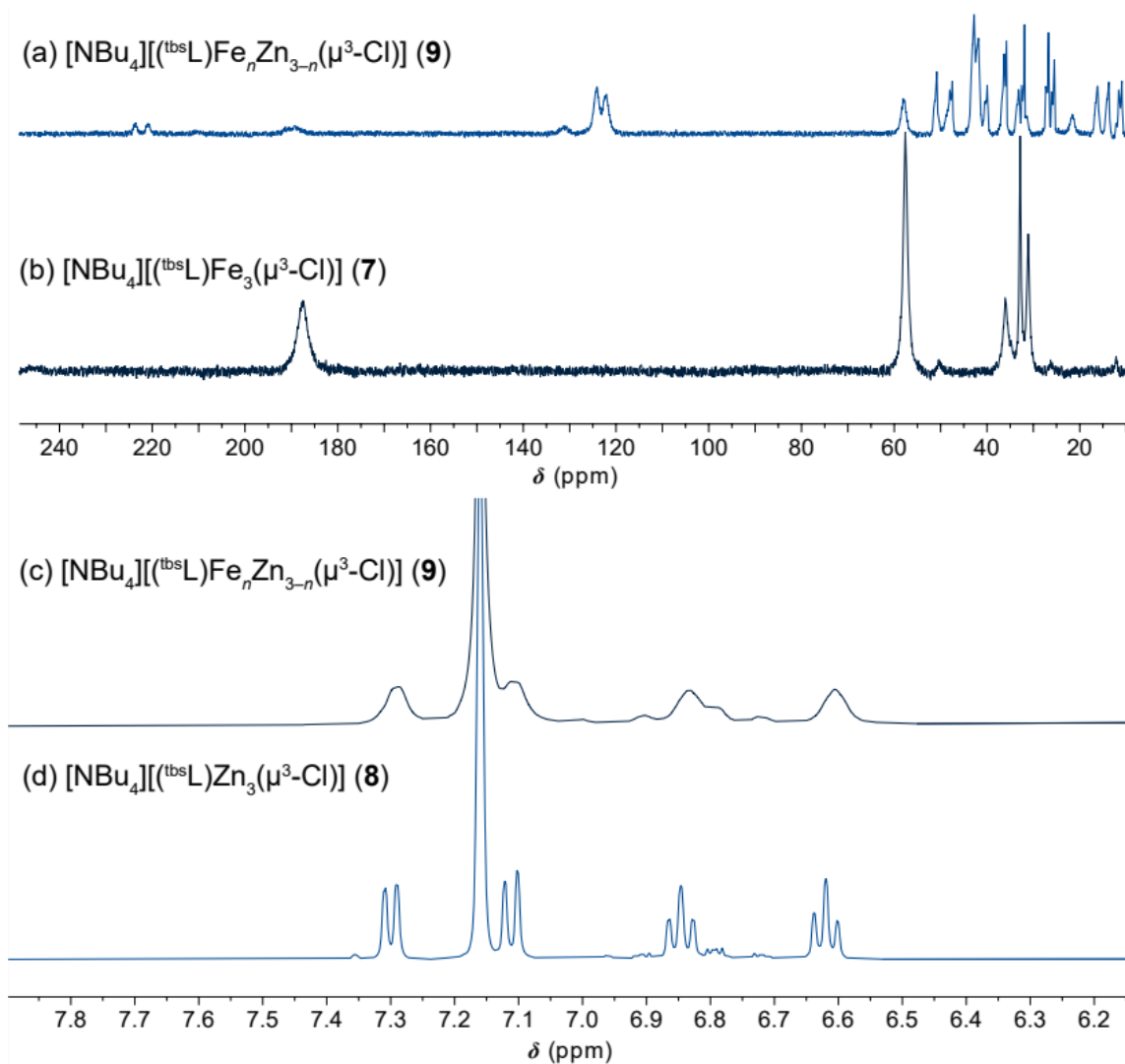


Figure S20. Comparison of ^1H NMR spectra of $[\text{NBu}_4][(\text{tbsL})\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl})]$ (9) with that of the analogous homometallic species $[\text{NBu}_4][(\text{tbsL})\text{Fe}_3(\mu^3\text{-Cl})]$ (7) and $[\text{NBu}_4][(\text{tbsL})\text{Zn}_3(\mu^3\text{-Cl})]$ (8). A portion of the paramagnetic window is shown for 9 (a) and 8 (b). The diamagnetic region has been removed from these spectra for clarity. The bottom figure shows an enlargement of the diamagnetic aromatic region of the spectra 9 (c) and 8 (d). The peaks highlighted in black in (a) and (c) correspond to the homometallic products which form upon anation of the hetero-metallic starting materials; the intense resonance at 7.16 ppm corresponds to the benzene solvent residual.

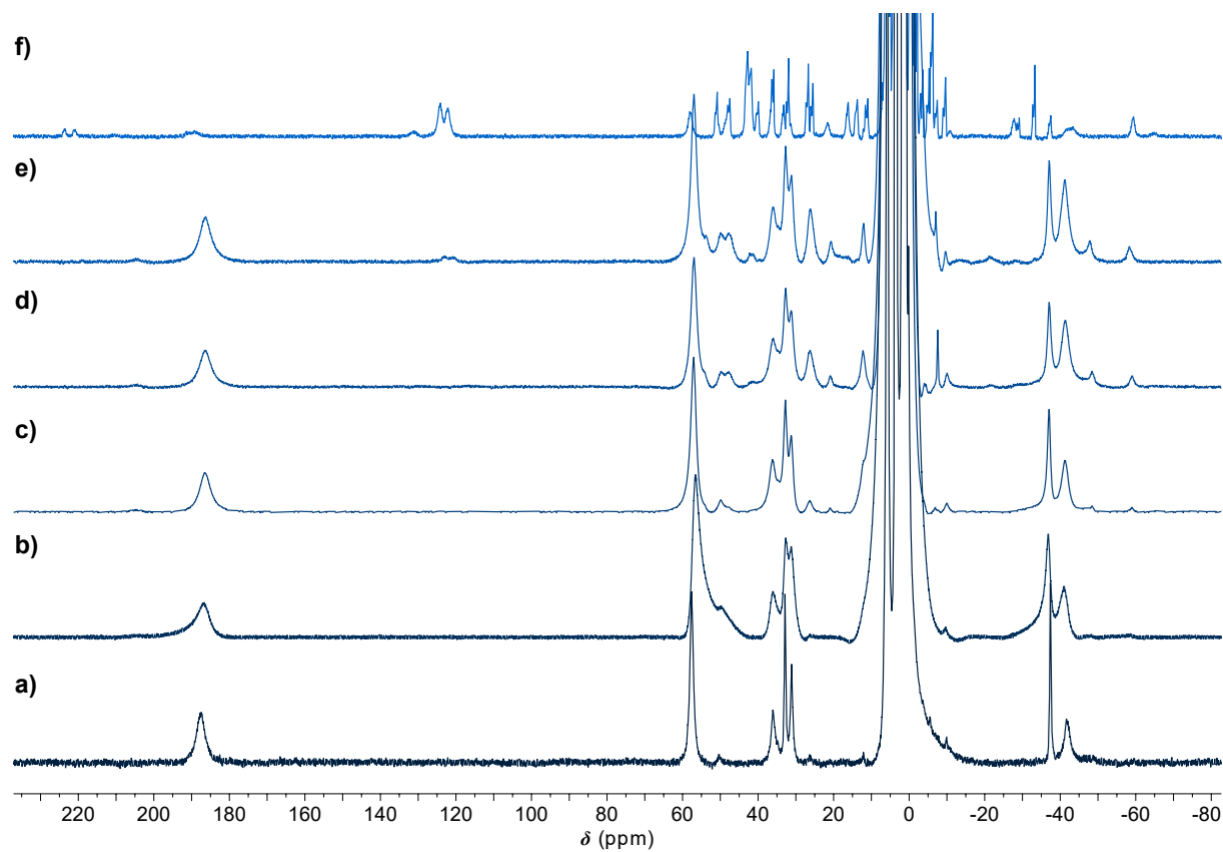


Figure S21. ^1H NMR spectra of a mixture of $[\text{NBu}_4][(\text{tbsL})\text{Fe}_3(\mu^3\text{-Cl})]$ (**7**) and $[\text{NBu}_4][(\text{tbsL})\text{Zn}_3(\mu^3\text{-Cl})]$ (**8**) subjected to heat in THF: (a) authentic **7**, (b) 24 h at room temperature, (c) 24 h at $40\text{ }^\circ\text{C}$, (d) 24 h at $60\text{ }^\circ\text{C}$, (e) 24 h at $80\text{ }^\circ\text{C}$, (f) authentic $[\text{NBu}_4][(\text{tbsL})\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl})]$ (**9**).

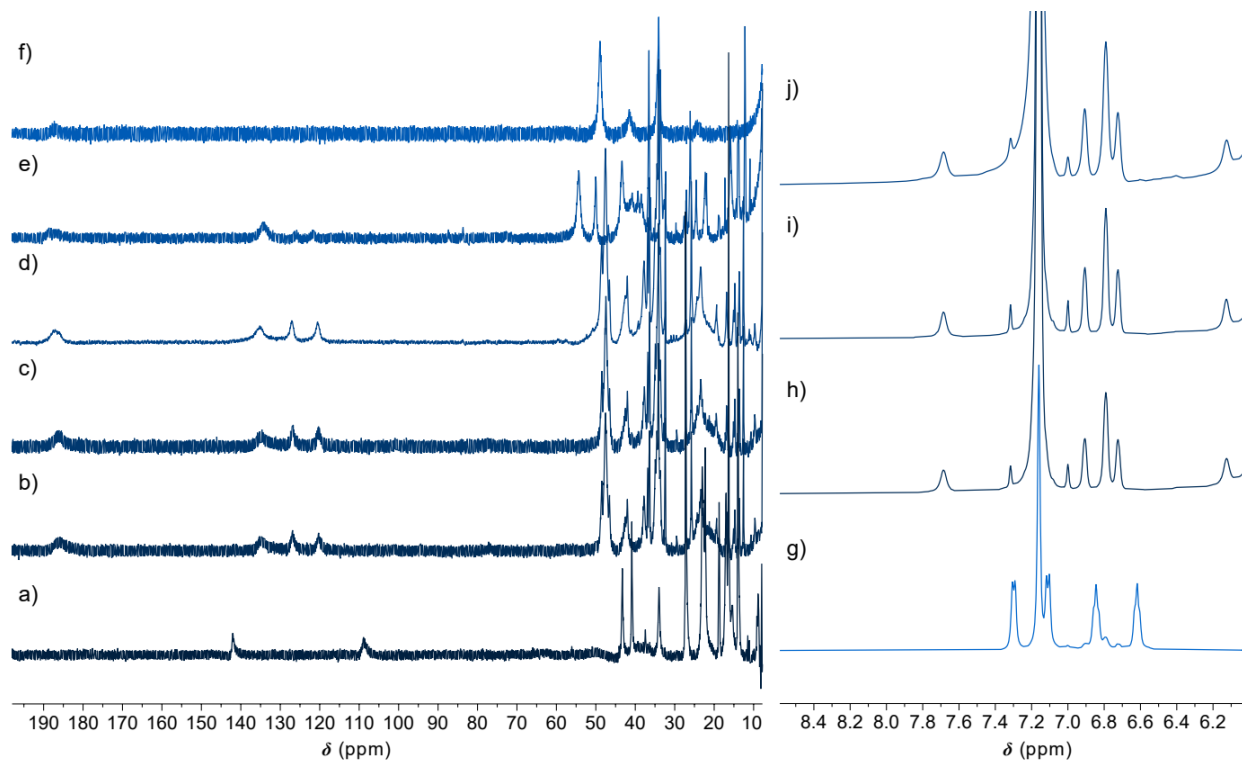


Figure S22 Left: ¹H NMR spectra of a) (^tbsL)Fe₂Zn(thf) (**5**), b) the reaction mixture formed from **5** and [NBu₄][(^tbsL)Fe₃(μ³-Cl)] (**7**) after 4 h, c) after 24 h, d) after 24 h at 60 °C, e) [NBu₄][(^tbsL)Fe_nZn_{3-n}(μ³-Cl)] (**9**), and f) **7**, recorded in THF/C₆D₆ (25% v/v). Right: Magnification of aromatic regions of the same spectra for g) [NBu₄][(^tbsL)Zn₃(μ³-Cl)] (**8**), h) the reaction mixture formed from **5** and **7** after 4 h, i) after 24 h, and j) after 24 h at 60 °C.

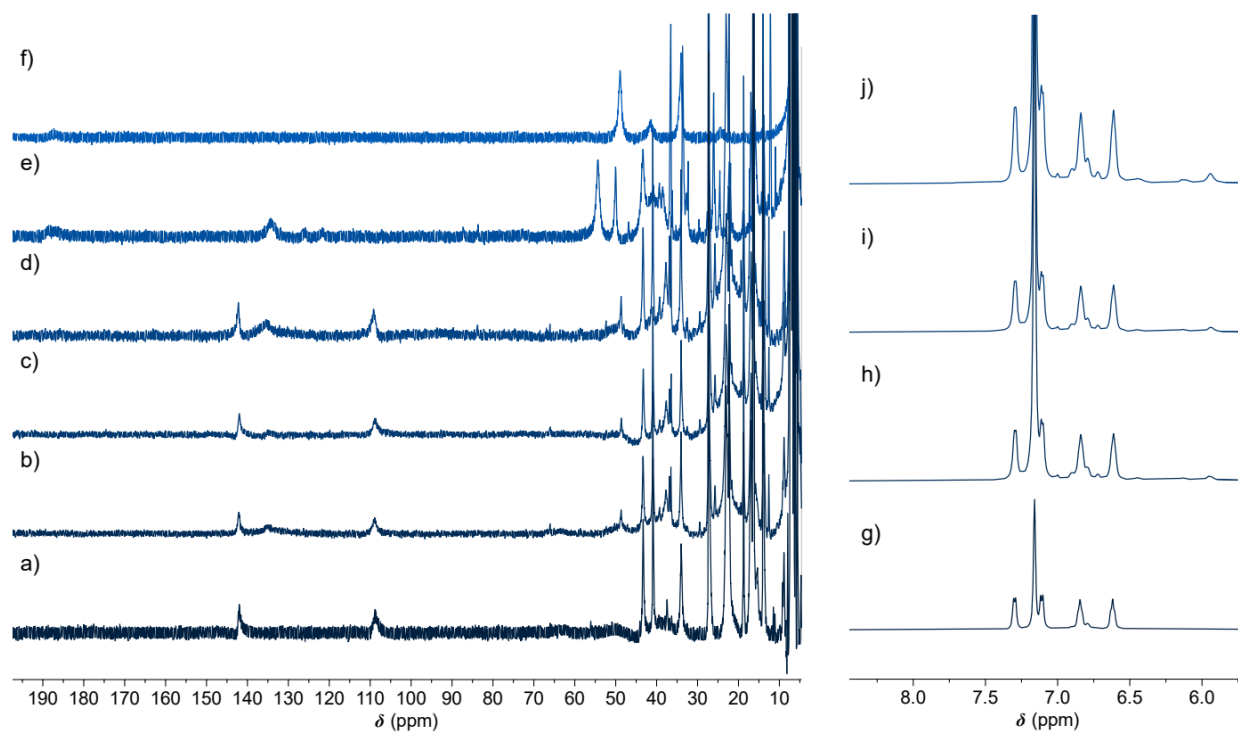


Figure S23. Left: ¹H NMR spectra of a) (^tbsL)Fe₂Zn(thf) (**5**), b) the reaction mixture formed from **5** and [NBu₄][(^tbsL)Zn₃(μ³-Cl)] (**8**) after 4 h, c) after 24 h, d) after 24 h at 60 °C, e) [NBu₄][(^tbsL)Fe_nZn_{3-n}(μ³-Cl)] (**9**), and f) [NBu₄][(^tbsL)Fe₃(μ³-Cl)] (**7**), recorded in THF/C₆D₆ (25% v/v). Right: Magnification of aromatic regions of the same spectra for g) **8**, h) the reaction mixture formed from **5** and **8** after 4 h, i) after 24 h, and j) after 24 h at 60 °C.

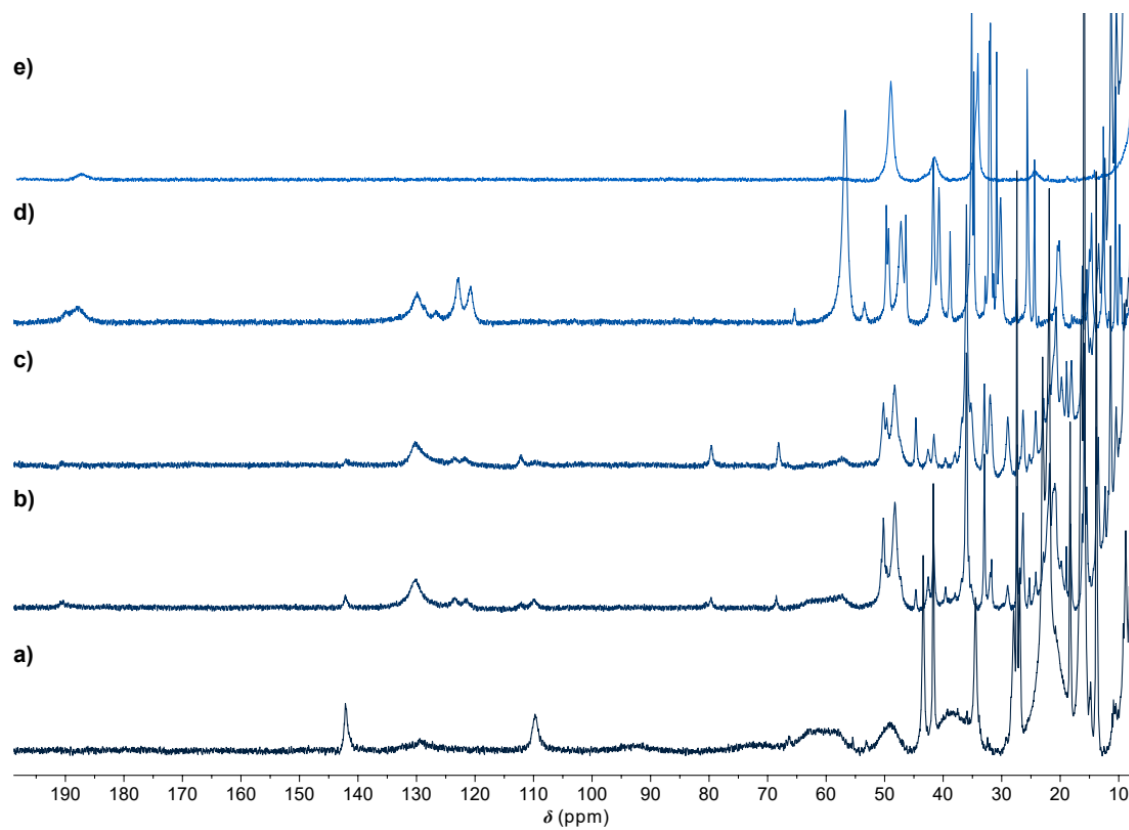


Figure S24. ^1H NMR spectra of a) $(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\text{thf})$ (**5**), b) the reaction mixture formed from **5** and FeCl_2 after 4 h, c) after 24 h, d) $[\text{NBu}_4][(^{\text{tbs}}\text{L})\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl})]$ (**9**), and e) $[\text{NBu}_4][(^{\text{tbs}}\text{L})\text{Fe}_3(\mu^3\text{-Cl})]$ (**7**), recorded in THF.

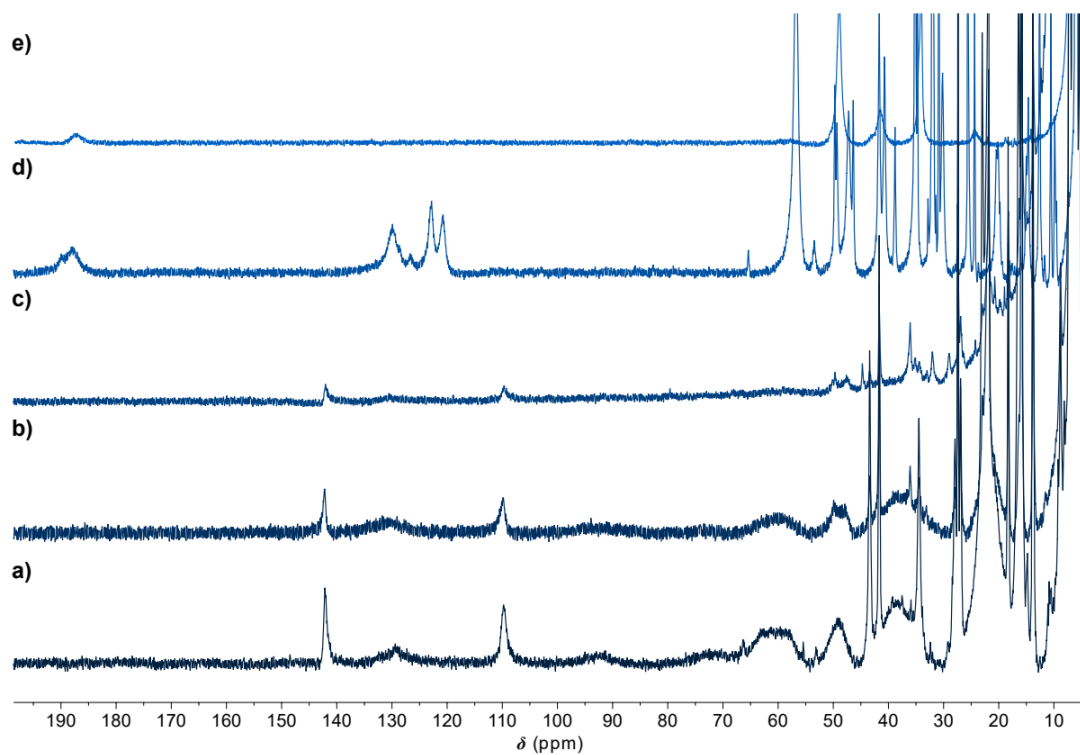


Figure S25. ^1H NMR spectra of a) $(^{\text{tbs}}\text{L})\text{Fe}_2\text{Zn}(\text{thf})$ (**5**), b) the reaction mixture formed from **5** and ZnCl_2 after 4 h, c) after 24 h, d) $[\text{NBu}_4][(^{\text{tbs}}\text{L})\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl})]$ (**9**), and e) $[\text{NBu}_4][(^{\text{tbs}}\text{L})\text{Fe}_3(\mu^3\text{-Cl})]$ (**7**), recorded in THF.

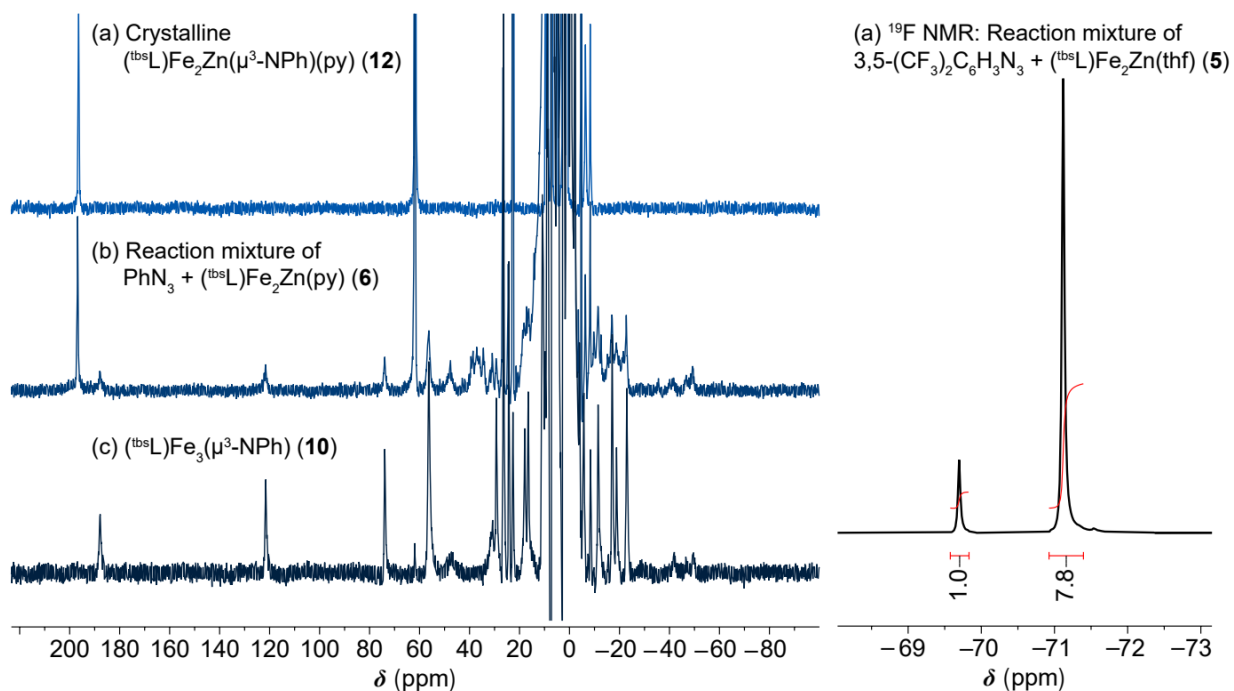


Figure S26. Comparison of ^1H NMR spectrum of $(\text{tbsL})\text{Fe}_3(\mu^3\text{-NPh})$ (**10**) (c) with that of the crystalline mixed-metal product $(\text{tbsL})\text{Fe}_2\text{Zn}(\mu^3\text{-NPh})(\text{py})$ (**12**) (a) and the reaction mixture formed by treatment of $(\text{tbsL})\text{Fe}_2\text{Zn}(\text{py})$ (**6**) with phenyl azide (b). The right-hand figure (d) shows the integrated ^{19}F NMR spectrum of the products obtained via reaction of $(\text{tbsL})\text{Fe}_2\text{Zn}(\text{thf})$ (**5**) with 3,5- $(\text{CF}_3)_2\text{C}_6\text{H}_3\text{N}_3$. The smaller of the two peaks (-69.7 ppm) represents the triiron product $(\text{tbsL})\text{Fe}_3(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))$, while the larger peak (-71.1 ppm) represents the mixed-metal species $(\text{tbsL})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))$, indicating a 7.8:1 ratio of products, with the diiron species dominating the mixture.

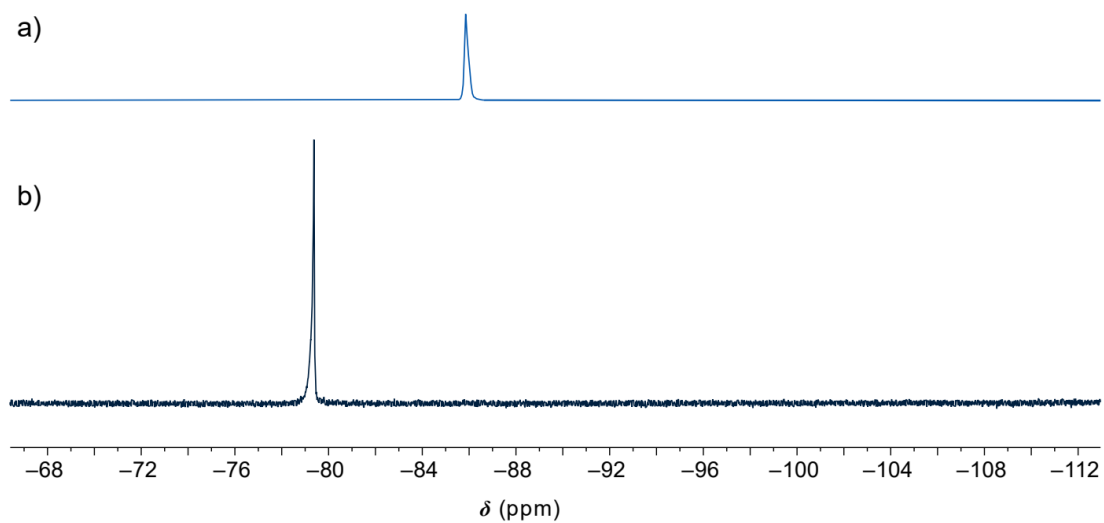


Figure S27. ^{19}F NMR spectra for $[\text{CoCp}^*_2][(\text{t}^{\text{bs}}\text{L})\text{Fe}_3(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))]$ (**15**) (a) and $[\text{2,2,2-cryptand}(\text{K})][(\text{t}^{\text{bs}}\text{L})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))]$ (**16**) (b).

SQUID Magnetometry

Dc susceptibility and magnetization data for **1** were simultaneously modeled according to a spin-Hamiltonian for three exchange-coupled spins in an equilateral arrangement ($J_1 = J_2 = J_3$): $\hat{H} = -2(J_1\hat{S}_1\hat{S}_2 + J_2\hat{S}_1\hat{S}_3 + J_3\hat{S}_2\hat{S}_3) + D\hat{S}_z^2 + E(\hat{S}_x^2 + \hat{S}_y^2) + g\mu_B\hat{S}B$. The fit parameters are: $J = +40(1) \text{ cm}^{-1}$, $D = 20.0(9) \text{ cm}^{-1}$, $|E/D| = 0.082(7) \text{ cm}^{-1}$, and $g = 1.92(1)$. Attempts to fit the data according to an isosceles ($J_1 = J_2 \neq J_3$) or to a scalene triangle ($J_1 \neq J_2 \neq J_3$) gave physically implausible values of the exchange coupling constants, zero-field splitting parameters, or both. A fit of the data according to a single "giant-spin" system using a spin-Hamiltonian of the form $\hat{H} = D\hat{S}_z^2 + E(\hat{S}_x^2 + \hat{S}_y^2) + g\mu_B\hat{S}B$ provided an inferior fit of the dc susceptibility data that did not reproduce the diminution of χ_{MT} at higher temperatures.

Attempts to fit the dc susceptibility and magnetization data for **5** were unsuccessful when considering either an exchange-coupled two spin system or a single $S = 4$ manifold. Inclusion of temperature independent paramagnetism or intermolecular interaction (zJ) terms did not provide physically reasonable fits. A likely explanation for this is the contamination of **5** with trace quantities of **1**. This possibility seems reasonable in light of the facile metal redistribution observed in complexes of this family; indeed, the presence of even trace quantities of **1**, which possesses a much larger value of χ_{MT} across the surveyed temperature range, will significantly impact attempts to fit the data. Nonetheless, the essential features of the dc susceptibility and magnetization data strongly indicate that **5** is an antiferromagnetically coupled system which populates high-spin states at higher temperatures.

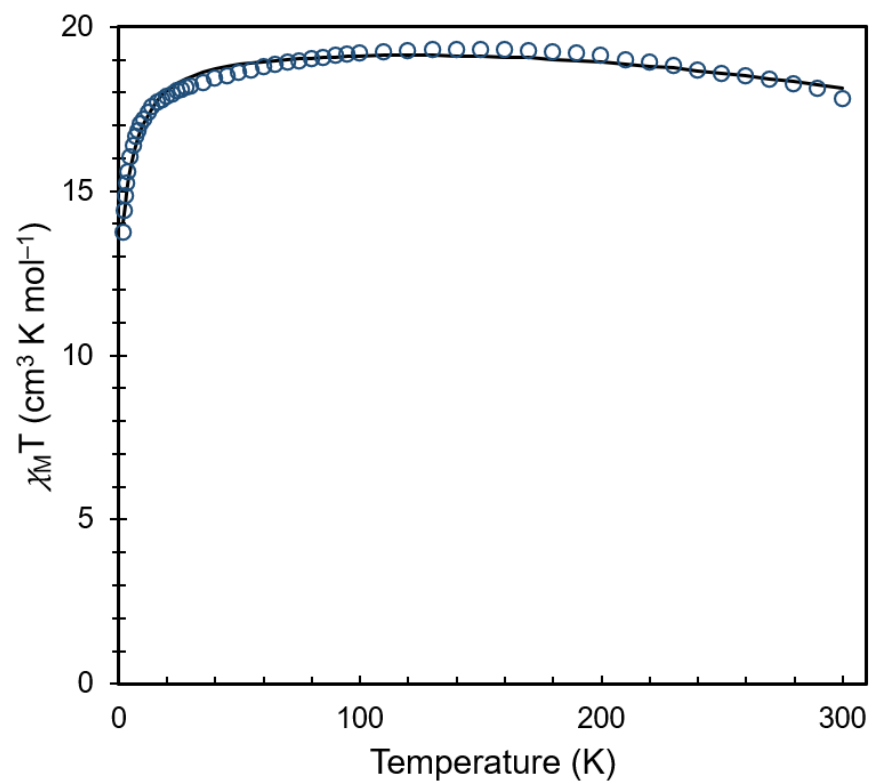


Figure S28. Plot of $\chi_M T$ for $(\text{tbsL})\text{Fe}_3(\text{thf})$ (**1**) over a temperature range of 1.8-300 K. Experimental data are denoted by open circles, and the fit is denoted by a black line.

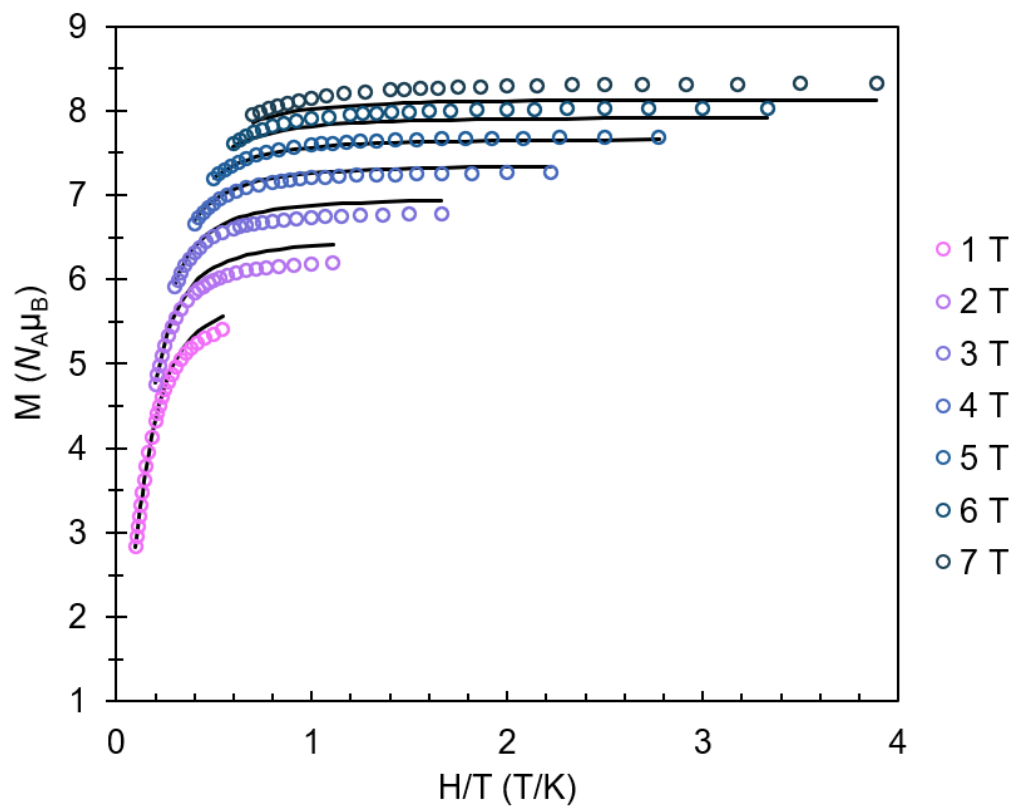


Figure S29. Plot of reduced magnetization for $(t^{bsL})Fe_3(thf)$ (**1**) over a temperature range of 1.8-10 K and an applied field range of 1.0-7.0 T. Experimental data are denoted by open circles, and the fit is denoted by a black line.

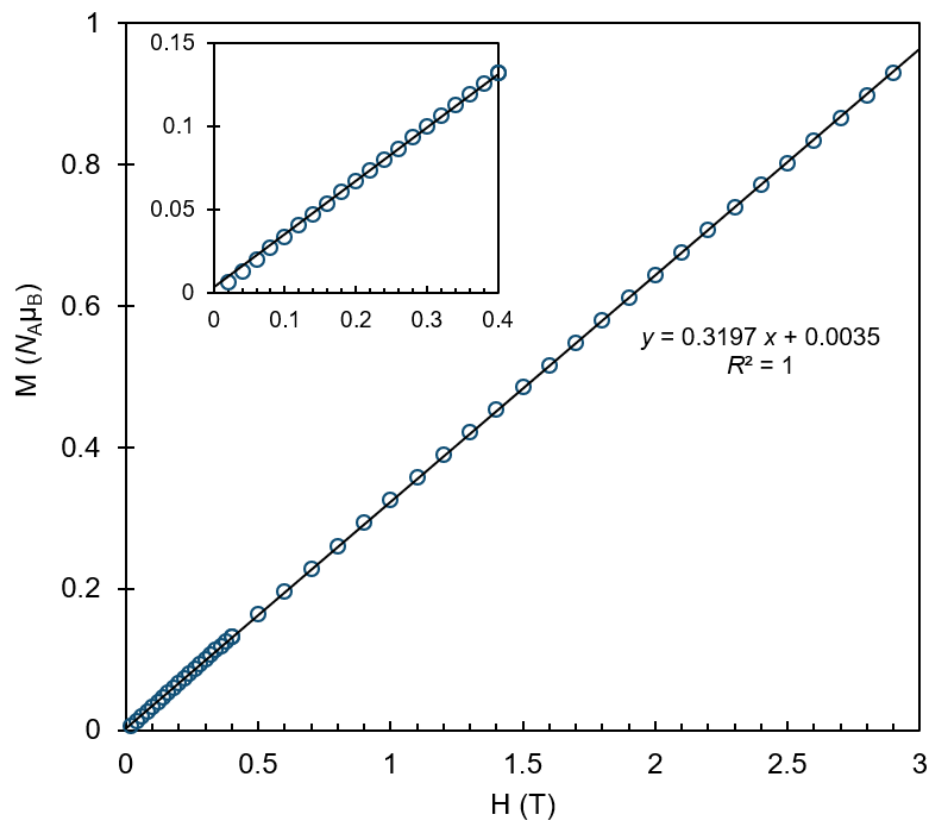


Figure S30. Plot of magnetization for $(tbsL)Fe_3(thf)$ (**1**) obtained at 100 K over an applied field of 0-3.0 T. Inset: expansion of low-field region. Linearity of the plot demonstrates the absence of ferromagnetic impurities.

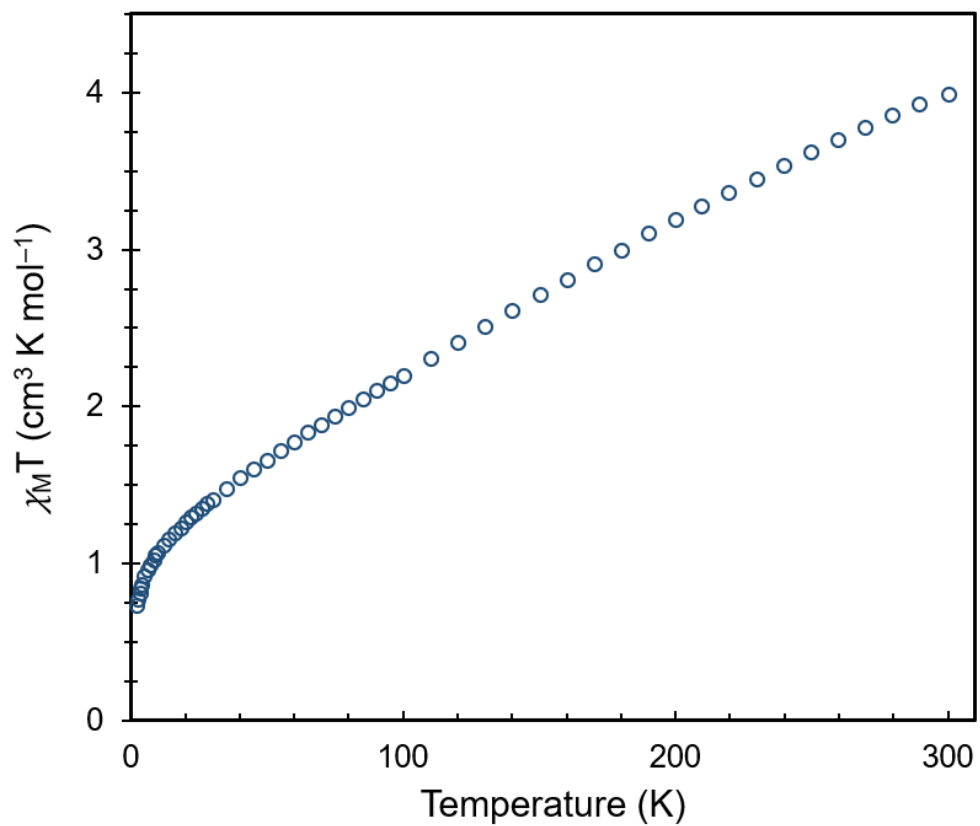


Figure S31. Plot of $\chi_M T$ data for $(^{t^bsL})\text{Fe}_2\text{Zn}(\text{thf})$ (**5**) over a temperature range of 1.8-300 K. Experimental data are denoted by open circles.

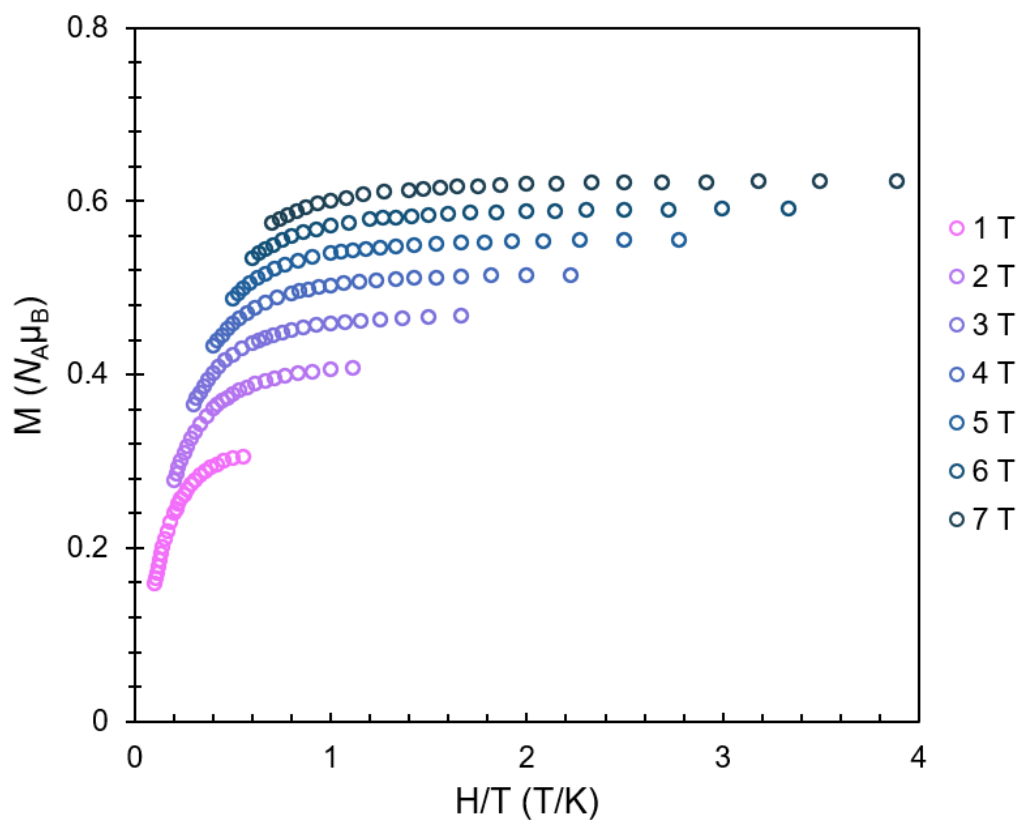


Figure S32. Plot of reduced magnetization for $(^{tb^sL})\text{Fe}_2\text{Zn}(\text{thf})$ (**5**) over a temperature range of 1.8-10 K and an applied field range of 1.0-7.0 T. Experimental data are denoted by open circles.

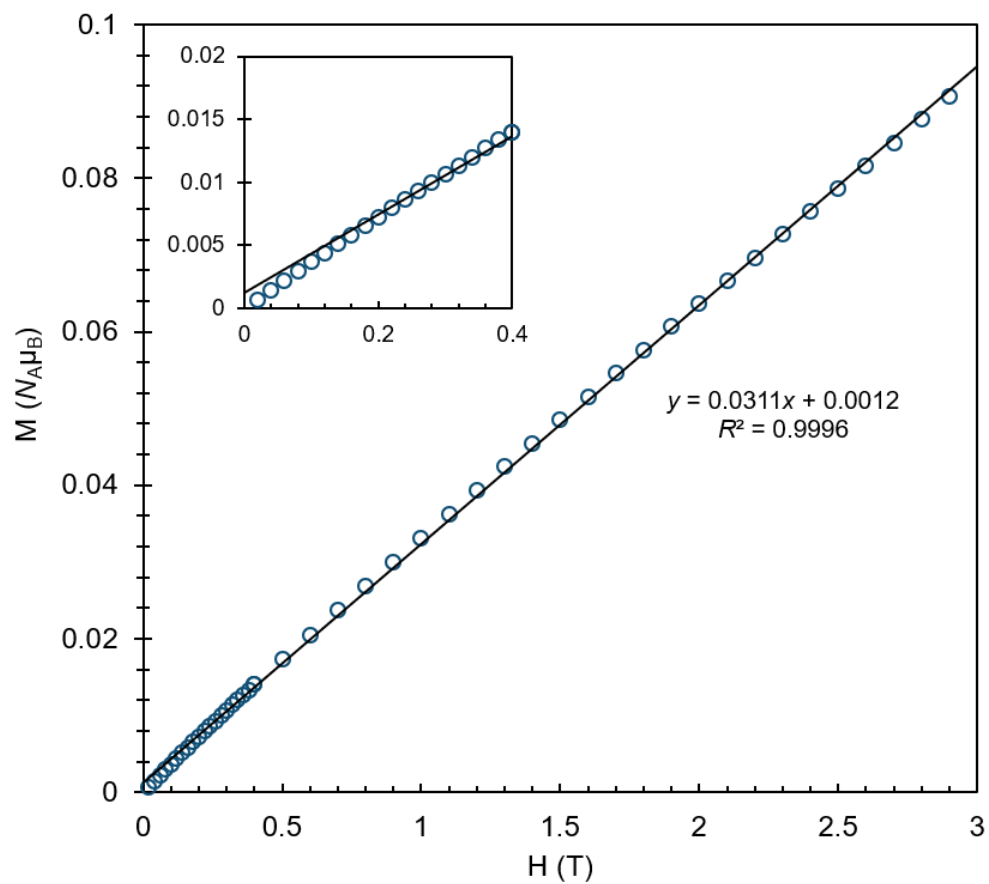


Figure S33. Plot of magnetization for $(^{tb^sL})\text{Fe}_2\text{Zn}(\text{thf})$ (**5**) obtained at 100 K over an applied field of 0-3.0 T. Inset: expansion of low-field region. Linearity of the plot demonstrates the absence of ferromagnetic impurities.

X-Ray Diffraction Techniques

All structures were collected on a Bruker three-circle platform goniometer equipped with an Apex II CCD area detector and an Oxford cryostream cooling device. Radiation was from a graphite fine focus sealed Mo K α (0.71073 Å) source or synchrotron radiation as specified. Crystals were mounted on a cryoloop or glass fiber pin using Paratone-N oil. Structures were collected at 100 K. Data was collected as a series of φ and/or ω scans. Data was integrated using SAINT⁸ and scaled with either a numerical or multi-scan absorption correction using SADABS⁸. The structures were solved by intrinsic phasing or direct methods using SHELXS-97⁹ and refined against F^2 on all data by full matrix least squares with SHELXL-97⁹. All non-hydrogen atoms were refined anisotropically. Hydrogen atoms were placed at idealized positions and refined using a riding model. The isotropic displacement parameters of all hydrogen atoms were fixed to 1.2 times the atoms they are linked to (1.5 times for methyl groups). Further details on particular structures are noted below.

(^{tbs}L)Fe₃(py) (2). The structure was solved in the monoclinic space group $P2_1/c$ with four molecules per unit cell. The asymmetric unit was found to contain one molecule of **2** and approximately two molecules of disordered ether. Data was collected at APS using 0.41328 Å radiation. Seven reflections were omitted due to detector overflow.

[Na(thf)(OEt₂)][(^{tbs}L)Fe₂Na(OEt₂)] (4). The structure was solved in the triclinic space group $P\bar{1}$ with two molecules per unit cell. The asymmetric unit was found to contain one molecule of **4**. Data was collected using sealed-tube Mo radiation. The sodium ion that is contained with the ligand pocket is bound to one molecule of solvent (diethyl ether), which is disordered over two positions. The second sodium ion, which is coordinated to the ligand via arene interactions is bound to two solvent molecules. One of the solvents is disordered between diethyl ether and tetrahydrofuran. Two of the t-butyl dimethylsilyl groups are modeled with positional disorder. Note: crystallization conditions result in difference in solvation from powder material.

(^{tbs}L)Fe₂Zn(thf) (5). The structure was solved in the monoclinic space group $P2_1/n$ with four molecules per unit cell. The asymmetric unit was found to contain one molecule of (^{tbs}L)Fe₂Zn(thf) and one molecule of hexane. Data was collected at APS using 0.40659 Å radiation. Eight reflections were omitted due to detector overflow. Residual electron density from disordered solvent channels was removed using the SQUEEZE function in Platon.

(^{tbs}L)Fe₂Zn(py) (6). The structure was solved using the intrinsic phasing method in the triclinic space group $P\bar{1}$ with two molecules per unit cell. The asymmetric unit was found to contain one molecule of (^{tbs}L)Fe₂Zn(py) and half a molecule of hexane (located on a special position). Data was collected at APS using 0.40651 Å radiation. Eleven reflections were omitted due to detector overflow.

[NBu₄][(^{tbs}L)Fe_nZn_{3-n}(μ^3 -Cl)] (9). The structure was solved in the orthorhombic space group $Pbca$ with eight molecules per unit cell. The asymmetric unit was found to contain one molecule of **9**. Data was collected using sealed-tube Mo radiation. Residual electron density from disordered solvent channels was removed using SQUEEZE function in Platon. Disorder was modeled in butyl group of the tetrabutylammonium counter ion. Occupancy of each metal site was restricted to 67% Fe and 33% Zn.

[NBu₄][(^{tbs}L)Fe₃(μ^3 -Cl)] (7). The structure was solved in the orthorhombic space group $Pbca$ with eight molecules per unit cell. The asymmetric unit was found to contain one molecule of **7**. Data was collected with sealed-tube Mo radiation. Residual electron density from disordered solvent channels was removed using the SQUEEZE function in Platon. Disorder was modeled in butyl group of the tetrabutylammonium counter ion.

[NBu₄][^(tbs)LZn₃(μ³-Cl)] (8). The structure was solved in the orthorhombic space group *Pbca* with 8 molecules per unit cell. The asymmetric unit was found to contain one molecule of **8**. Data was collected using sealed-tube Mo radiation. Residual electron density from disordered solvent channels was removed using the SQUEEZE function in Platon. Disorder was modeled in butyl group of the tetrabutylammonium counter ion.

(^(tbs)L)Fe₂Zn(μ³-NPh) (11). The structure was solved in the monoclinic space group *C2/c* with eight molecules in the unit cell. The asymmetric unit was found to contain one molecule of **11**. Data was collected using sealed-tube Mo radiation. Residual electron density from disordered solvent channels was removed using the SQUEEZE function in Platon. Disorder was modeled in the orientation of one of the *tert*-butyldimethylsilyl groups of the ligand and in the orientation of the imido moiety. Occupancy of each metal site was restricted to 67% Fe and 33% Zn.

(^(tbs)L)Fe₂Zn((μ³-NPh)(py) (12). The structure was solved in the triclinic space group *P-1* with 2 molecules per unit cell. The asymmetric unit was found to contain one molecule of **12**. Data was collected at APS using 0.40651 Å radiation. Residual electron density from disordered solvent channels was removed using the SQUEEZE function in Platon. Bad reflections were omitted due to detector overflow.

(^(tbs)L)Fe₃(μ³-N(3,5-(CF₃)₂C₆H₃)) (13). The structure was solved in the monoclinic space group *P2₁/n* with four molecules per unit cell. The asymmetric unit was found to contain one molecule of **13**. Data was collected using sealed-tube Mo radiation. There appears to be whole-molecule disorder, where the molecule is in two slightly different orientations. However, the occupancy of this second part of is so small that only metal atoms can be modeled. Additionally, disorder was modeled on one *tert*-butyldimethylsilyl group.

(^(tbs)L)Fe₂Zn(μ³-N(3,5-(CF₃)₂C₆H₃))(py) (14). The structure was solved in the triclinic space group *P-1* with two molecules per unit cell. The asymmetric unit was found to contain one molecule of **14**. Data was collected using sealed-tube Mo radiation. Residual electron density from disordered solvent channels was removed using the SQUEEZE function in Platon. One of the CF₃ groups was modeled with rotational disorder.

[CoCp*₂][(^(tbs)L)Fe₃(μ³-N(3,5-(CF₃)₂C₆H₃))] (15). The structure was solved in the monoclinic space group *P2₁/n* with four molecules per unit cell. The asymmetric unit was found to contain one molecule of **15** and one molecule of diethyl ether. Data was collected using sealed-tube Mo radiation. Residual electron density from disordered solvent channels was removed using the SQUEEZE function in Platon. Both CF₃ groups were modeled with rotational disorder.

[2,2,2-cryptand(K)][(^(tbs)L)Fe₂Zn(μ³-N(3,5-(CF₃)₂C₆H₃))] (16). The structure was solved in the triclinic space group *P-1* with four molecules per unit cell. The asymmetric unit was found to contain two molecules of **16** and several disordered diethyl ether solvent molecules. Data was collected at APS using 0.41328 Å radiation. Diethyl ether was modeled to fill the solvent voids in the structures, but has large ADPs due to large amounts of disorder. Occupancy of the metal sites was allowed to freely refine.

Table S2 X-ray diffraction experimental details

	(^{tbs}L)Fe₃(py) (2)	[Na(thf)(OEt₂)] [(^{tbs}L)Fe₂Na(OEt₂)] (4·Et₂O)	(^{tbs}L)Fe₂Zn(thf) (5)	(^{tbs}L)Fe₂Zn(py) (6)
Chemical formula	C ₅₄ H _{88.50} Fe ₃ N ₇ O _{1.75} Si ₃	C ₅₄ H _{94.97} Fe ₂ N ₆ Na ₂ O ₃ Si ₃	C ₅₂ H ₈₈ Fe ₂ N ₆ OSi ₃ Zn	C ₅₀ H ₇₈ Fe ₂ N ₇ Si ₃ Zn
FW	1115.63	1118.27	1074.62	1038.53
Crystal System	Monoclinic	Triclinic	Monoclinic	Triclinic
Space Group	<i>P</i> 2 ₁ / <i>c</i>	<i>P</i> -1	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> -1
Temperature (K)	100	100	100	100
a (Å)	10.4451 (12)	11.040 (2)	10.3835 (18)	10.6298 (8)
b (Å)	27.261 (3)	14.175 (3)	18.015 (2)	11.1655 (9)
c (Å)	21.186 (3)	21.290 (5)	30.663 (7)	23.6682 (18)
α (°)	90	106.479 (4)	90	98.248 (2)
β (°)	102.166 (4)	93.058 (4)	91.703 (6)	101.941 (2)
γ (°)	90	106.770 (4)	90	101.506 (2)
V (Å³)	5897.2 (12)	3025.6 (11)	5733.3 (17)	2642.6 (4)
Z	4	2	4	2
Radiation type	Synchrotron λ = 0.41328 Å	Mo Kα	Synchrotron λ = 0.40659 Å	Synchrotron λ = 0.40651 Å
μ (mm⁻¹)	0.17	0.60	0.51	0.21
Crystal size (mm)	0.19 × 0.11 × 0.04	0.3 × 0.2 × 0.1	0.14 × 0.08 × 0.02	0.17 × 0.06 × 0.02
R1^a	0.042	0.040	0.059	0.044
wR2^b	0.107	0.093	0.148	0.111
S (GOOF)	1.04	1.01	1.02	1.09
Reflections	12080	11243	17045	8937

$$^a R1 = [\sum w(F_o - F_c)^2 / \sum w F_o^2]^{1/2}$$

$$^b wR2 = [\sum [w(F_o^2 - F_c^2)^2] / \sum w(F_o^2)^2]^{1/2} \quad w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP], \text{ where } P = [\max(F_o^2, 0) + 2((F_c^2))] / 3$$

Table S2. Continuation ...

	[NBu ₄][(tbsL)Fe ₃ (Cl)] (7)	[NBu ₄][(tbsL)Zn ₃ (Cl)] (8)	[NBu ₄][(tbsL)Fe _{3-n} Zn _n (Cl)] (9)	(tbsL)Fe _{3-n} Zn _n (NPh) (11)
Chemical formula	C ₅₇ H ₁₀₀ ClFe ₃ N ₇ Si ₃	C ₅₈ H ₁₀₂ ClN ₇ Si ₃ Zn ₃	C ₅₈ H ₁₀₂ ClFe ₂ N ₇ Si ₃ Zn	C ₄₈ H ₇₁ Fe ₂ N ₇ Si ₃ Zn
FW	1170.70	1213.29	1194.25	1007.45
Crystal System	Orthorhombic	Orthorhombic	Orthorhombic	Monoclinic
Space Group	<i>Pbca</i>	<i>Pbca</i>	<i>Pbca</i>	<i>C2/c</i>
Temperature (K)	100	100	100	100
a (Å)	26.839 (3)	27.1744 (9)	27.053 (4)	22.150 (4)
b (Å)	19.358 (2)	19.1482 (7)	19.227 (3)	20.383 (3)
c (Å)	27.601 (4)	27.5979 (10)	27.643 (4)	26.428 (4)
α (°)	90	90	90	90
β (°)	90	90	90	101.975 (3)
γ (°)	90	90	90	90
V (Å³)	14340 (3)	14360.3 (9)	14378 (4)	11672 (3)
Z	8	8	8	8
Radiation type	Mo K _α	Mo K _α	Mo K _α	Mo K _α
μ (mm⁻¹)	0.72	1.12	0.85	0.99
Crystal size (mm)	0.40 × 0.30 × 0.20	0.40 × 0.39 × 0.38	0.33 × 0.26 × 0.16	0.30 × 0.20 × 0.10
R1^a	0.043	0.033	0.056	0.068
wR2^b	0.126	0.079	0.131	0.143
S (GOOF)	1.04	1.03	1.02	1.09
Reflections	13621	12718	12755	10413

$$^a R1 = [\sum w(F_o - F_c)^2 / \sum w F_o^2]^{1/2}$$

$$^b wR2 = [\sum [w(F_o^2 - F_c^2)^2] / \sum w (F_o^2)^2]^{1/2} \quad w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP], \text{ where } P = [\max(F_o^2, 0) + 2((F_c^2)]/3$$

Table S2. Continuation ...

	^(tbsL) Fe ₂ Zn(NPh)(py) (12)	^(tbsL) Fe ₃ (N(3,5-(CF ₃) ₂ C ₆ H ₃)) (13)	^(tbsL) Fe ₂ Zn(N(3,5-(CF ₃) ₂ C ₆ H ₃))(py) (14)
Chemical formula	C ₅₃ H ₇₆ Fe ₂ N ₈ Si ₃ Zn	C ₅₀ H ₆₉ F ₆ Fe ₃ N ₇ Si ₃	C ₅₅ H ₇₄ F ₆ Fe ₂ N ₈ Si ₃ Zn
FW	1086.55	1133.94	1222.56
Crystal System	Triclinic	Monoclinic	Triclinic
Space Group	<i>P</i> -1	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> -1
Temperature (K)	100	100	100
a (Å)	12.4239 (10)	15.1571 (7) 21.4890 (11) 16.3741 (8)	12.3372 (6)
b (Å)	14.2358 (12)		14.5673 (7)
c (Å)	19.0442 (16)		19.6915 (10)
α (°)	109.6426 (17)	90	76.058 (1)
β (°)	95.5642 (18)	93.577 (1)	73.437 (1)
γ (°)	102.0493 (18)	90	79.142 (1)
V (Å³)	3050.8 (4)	5322.8 (4)	3264.8 (3)
Z	2	4	2
Radiation type	Synchrotron, λ = 0.40651 Å	Mo Kα	Mo Kα
μ (mm⁻¹)	0.19	0.94	0.91
Crystal size (mm)	0.30 × 0.15 × 0.10	0.14 × 0.14 × 0.06	0.45 × 0.26 × 0.22
R1^a	0.043	0.066	0.033
wR2^b	0.13	0.194	0.079
S (GOOF)	1.04	1.02	1.03
Reflections	10463	9399	11581

$$^aR1 = [\sum w(F_o - F_c)^2 / \sum w F_o^2]^{1/2}$$

$$^b wR2 = [\sum [w(F_o^2 - F_c^2)^2] / \sum w (F_o^2)^2]^{1/2} \quad w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP], \text{ where } P = [\max(F_o^2, 0) + 2((F_c^2))] / 3$$

Table S2. Continuation ...

	[CoCp*₂][^(tbs)L)Fe₃(N(3,5-(CF₃)₂C₆H₃))] (15)	[2,2,2-crypt(K)][^(tbs)L)Fe₂Zn(N(3,5-(CF₃)₂C₆H₃))] (16)
Chemical formula	C ₇₄ H ₁₀₉ CoF ₆ Fe ₃ N ₇ OSi ₃	C ₅₀ H ₆₉ F ₆ Fe ₂ N ₇ Si ₃ Zn·C ₁₈ H ₃₆ KN ₂ O ₆ ·2.63(C ₄ H ₁₀ O)
FW	1533.57	1753.01
Crystal System	Monoclinic	Triclinic
Space Group	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> -1
Temperature (K)	100	100
a (Å)	11.9733 (5)	15.4746 (7)
b (Å)	26.2570 (9)	23.8238 (11)
c (Å)	26.2043 (10)	27.1431 (13)
α (°)	90	78.401 (1)
β (°)	100.148 (1)	74.315 (1)
γ (°)	90	83.137 (1)
V (Å³)	8109.3 (5)	9414.8 (8)
Z	4	4
Radiation type	Mo Kα	Synchrotron, λ = 0.41328 Å
μ (mm⁻¹)	0.83	0.51
Crystal size (mm)	0.20 × 0.18 × 0.08	0.17 × 0.15 × 0.09
R1^a	0.070	0.055
wR2^b	0.183	0.154
S (GOOF)	1.02	1.03
Reflections	14337	37723

$$^a R1 = [\sum w(F_o - F_c)^2 / \sum w F_o^2]^{1/2}$$

$$^b wR2 = [\sum [w(F_o^2 - F_c^2)^2] / \sum w (F_o^2)^2]^{1/2} \quad w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP], \text{ where } P = [\max(F_o^2, 0) + 2((F_c^2))] / 3$$

Table S3. Selected bond metrics (Å) of trinuclear complexes

Bond	1	2	5	6
M1–M2	2.6128(6)	2.6502(7)	2.7354(9)	2.8141(7)
M1–M3	2.6119(5)	2.6306(7)	3.1629(9)	2.9792(6)
M2–M3	2.5061(5)	2.5422(5)	2.9399(6)	2.8939(6)
M–M _{avg}	2.5769(9)	2.608(1)	2.946(1)	2.896(1)

Table S4. Selected bond metrics of reaction products

Compound		M–M		M–Ligand	
[NBu ₄][^(tbsL) Fe ₃ (μ ³ -Cl)]	(7)	Fe1–Fe2	2.8061(8)	Fe1–Cl	2.5579(8)
		Fe1–Fe3	2.6813(6)	Fe2–Cl	2.5177(8)
		Fe2–Fe3	2.8281(8)	Fe3–Cl	2.4785(7)
		Fe–Fe _{avg}	2.772(1)	Fe–Cl _{avg}	2.518(1)
[NBu ₄][^(tbsL) Zn ₃ (μ ³ -Cl)]	(8)	Zn1–Zn2	3.0176(8)	Zn1–Cl	2.6173(8)
		Zn1–Zn3	2.9444(8)	Zn2–Cl	2.5016(7)
		Zn2–Zn3	2.9942(8)	Zn3–Cl	2.6209(8)
		Zn–Zn _{avg}	2.985(1)	Zn–Cl _{avg}	2.579(1)
[NBu ₄][^(tbsL) Fe _n Zn _{3-n} (μ ³ -Cl)]	(9)	M1–M2	2.8674(9)	M1–Cl	2.641(1)
		M1–M3	2.9830(8)	M2–Cl	2.568(1)
		M2–M3	2.9860(8)	M3–Cl	2.501(1)
		M–M _{avg}	2.945(1)	M–Cl _{avg}	2.570(1)
^(tbsL) Fe ₃ (μ ³ -NPh) ^{1b}	(10)	Fe1–Fe2	2.549(1)	Fe1–N _{im}	1.936(5)
		Fe1–Fe3	2.502(1)	Fe2–N _{im}	1.944(5)
		Fe2–Fe3	2.539(1)	Fe3–N _{im}	1.944(5)
		Fe–Fe _{avg}	2.530(1)	Fe–N _{im avg}	1.941(9)
^(tbsL) Fe ₂ Zn(μ ³ -NPh)	(11)	M1–M2	2.725(1)	M1–N _{im}	2.004(4)
		M1–M3	2.678(1)	M2–N _{im}	2.060(4)
		M2–M3	2.7010(9)	M3–N _{im}	1.968(4)
		M–M _{avg}	2.701(2)	M–N _{im avg}	2.011(7)
^(tbsL) Fe ₂ Zn(μ ³ -NPh)(py)	(12)	Fe1–Fe2	2.4880(6)	Fe1–N _{im}	1.917(2)
		Fe1–Zn	3.2059(7)	Fe2–N _{im}	1.961(2)
		Fe2–Zn	2.7599(6)	Zn–N _{im}	2.077(3)
^(tbsL) Fe ₃ (μ ³ -N(3,5-(CF ₃) ₂ C ₆ H ₃))	(13) ^a	Fe1–Fe2	2.457(2)	Fe1–N _{im}	1.977(5)
		Fe1–Fe3	2.668(2)	Fe2–N _{im}	1.931(5)
		Fe2–Fe3	2.520(2)	Fe3–N _{im}	1.972(5)
		Fe–Fe _{avg}	2.548(3)	Fe–N _{im avg}	1.960(9)
^(tbsL) Fe ₂ Zn(μ ³ -N(3,5-(CF ₃) ₂ C ₆ H ₃))(py)	(14)	Fe1–Fe2	2.5008(6)	Fe1–N _{im}	1.976(2)
		Fe1–Zn	2.7806(4)	Fe2–N _{im}	1.932(2)
		Fe2–Zn	3.1982(6)	Zn–N _{im}	2.084(2)
[CoCp* ₂][^(tbsL) Fe ₃ (μ ³ -N(3,5-(CF ₃) ₂ C ₆ H ₃))]	(15)	Fe1–Fe2	2.532(1)	Fe1–N _{im}	1.953(5)
		Fe1–Fe3	2.632(1)	Fe2–N _{im}	1.978(4)
		Fe2–Fe3	2.596(1)	Fe3–N _{im}	1.996(4)
		Fe–Fe _{avg}	2.587(1)	Fe–N _{im avg}	1.976(7)
[2,2,2-crypt(K)][^(tbsL) Fe ₂ Zn(μ ³ -N(3,5-(CF ₃) ₂ C ₆ H ₃))]	(16) ^b	M1–M2	2.7085(7)	M1–N _{im}	2.032(3)
		M1–M3	2.6751(8)	M2–N _{im}	2.035(3)
		M2–M3	2.7843(6)	M3–N _{im}	2.010(2)
		M–M _{avg}	2.723(1)	M–N _{im avg}	2.026(5)

^aBond metrics reported for one part of the disorder model in **13**, which represents >90% occupancy.

^b Bond metrics listed for only one of the two clusters in the asymmetric unit (see Table S17 for full bond metric details).

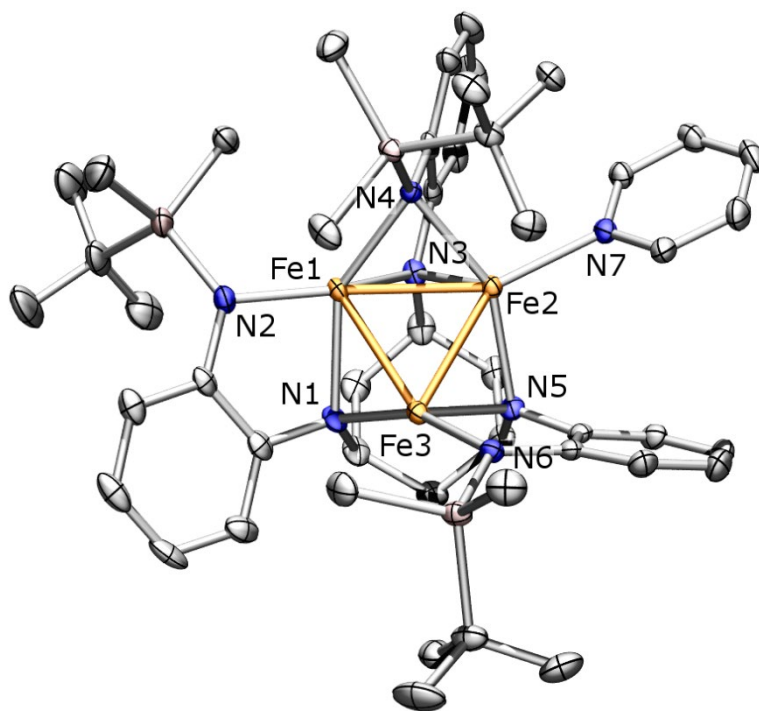


Figure S34. Solid state structure of $(^{tbsL})\text{Fe}_3(\text{py})$ (**2**). Hydrogen atoms and ether solvent molecules have been omitted for clarity.

Table S5. Selected bond metrics for **2** (Å).

Fe1—Fe2	2.6502(7)	Fe2—N5	2.031(2)
Fe1—Fe3	2.6306(7)	Fe2—N4	2.120(2)
Fe2—Fe3	2.5422(5)	Fe2—N7	2.154(2)
Fe1—N2	1.966(2)	Fe2—N3	2.166(2)
Fe1—N3	2.030(2)	Fe3—N6	1.937(2)
Fe1—N1	2.115(2)	Fe3—N1	1.959(2)
Fe1—N4	2.134(2)	Fe3—N5	2.053(2)

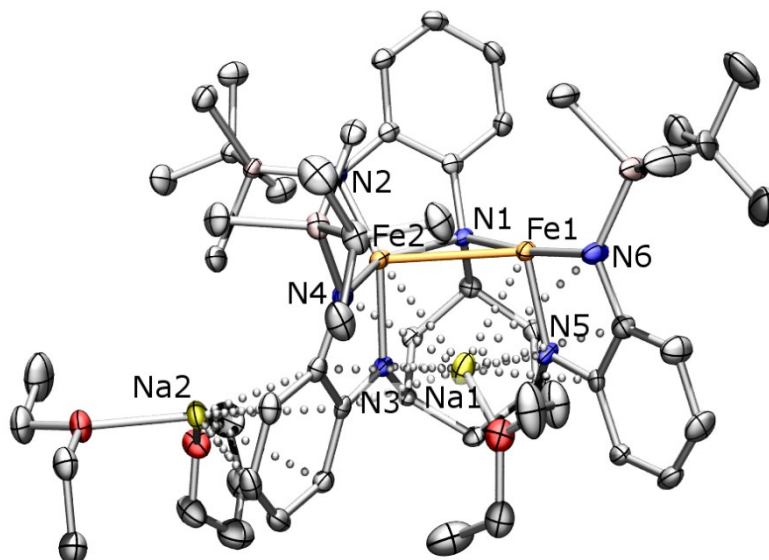


Figure S35. Solid state structure of $[\text{Na}(\text{thf})(\text{OEt}_2)][(\text{tbsL})\text{Fe}_2\text{Na}(\text{OEt}_2)]$ (**4**). Hydrogen atoms have been omitted for clarity. Only one part of the disorder is depicted for clarity.

Table S6. Selected bond metrics for **4** (Å).

Fe1—Fe2	2.8586(8)	Fe1—N5	1.939(2)
Fe1—Na1	2.997(1)	Fe2—N1	2.144(2)
Fe2—Na1	3.182(1)	Fe2—N4	2.058(2)
Fe1—N1	1.955(2)	Fe2—N2	1.991(2)
Fe1—N6	1.945(2)	Fe2—N3	1.999(2)

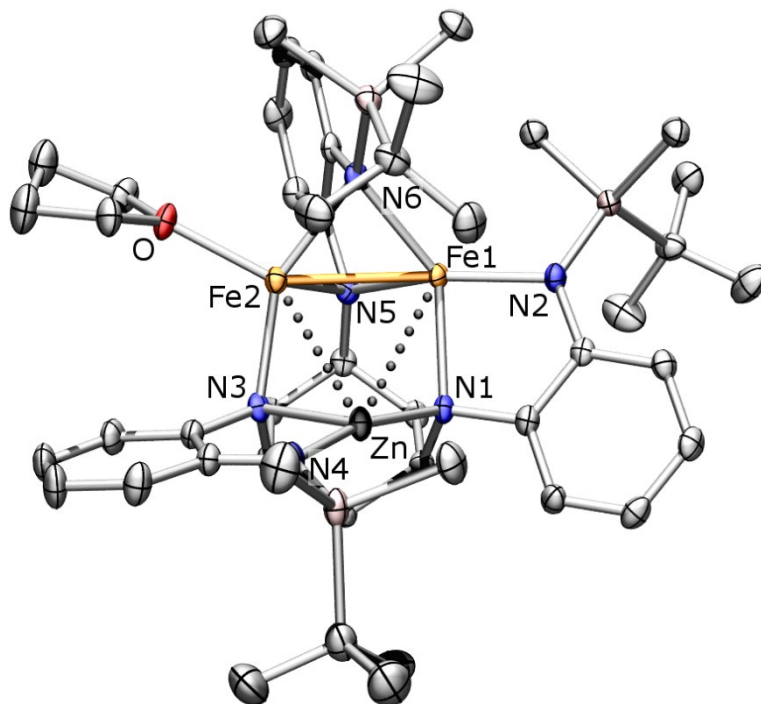


Figure S36. Solid state structure of $(^{tbs}L)Fe_2Zn(thf)$ (**5**). Hydrogen atoms and hexane solvent molecule omitted for clarity.

Table S7. Selected bond metrics for **5** (Å).

Fe1—Fe2	2.7354(9)	Fe2—N3	2.007(2)
Zn—Fe1	3.1629(9)	Fe2—N5	2.09(3)
Zn—Fe2	2.9399(6)	Fe2—N6	2.098(2)
Fe1—N2	1.950(2)	Fe2—O	2.051(2)
Fe1—N5	2.030(2)	Zn—N4	1.885(3)
Fe1—N1	2.132(2)	Zn—N1	1.916(3)
Fe1—N6	2.142(2)	Zn—N3	2.049(2)

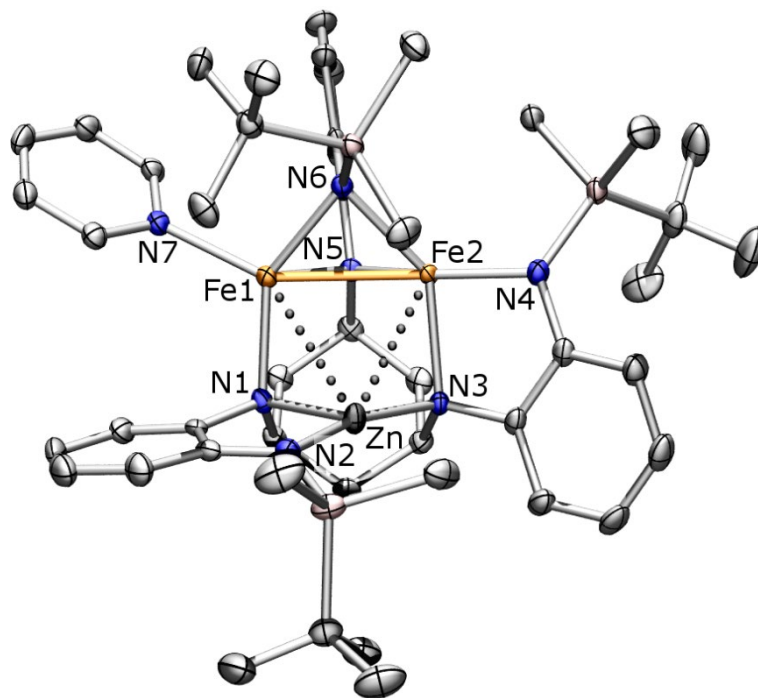


Figure S37. Solid state structure of $(^{tbsL})\text{Fe}_2\text{Zn}(\text{py})$ (**6**). Hydrogen atoms and hexane solvent molecule omitted for clarity.

Table S8. Selected bond metrics for **6** (Å).

Fe1—Fe2	2.8141(7)	Fe2—N3	2.109(2)
Zn—Fe1	2.9792(6)	Fe2—N4	1.938(4)
Zn—Fe2	2.8939(6)	Fe2—N5	2.030(2)
Fe1—N1	2.021(2)	Fe2—N6	2.176(3)
Fe1—N6	2.079(2)	Zn—N2	1.881(2)
Fe1—N5	2.106(3)	Zn—N3	1.911(3)
Fe1—N7	2.113(3)	Zn—N1	2.024(3)

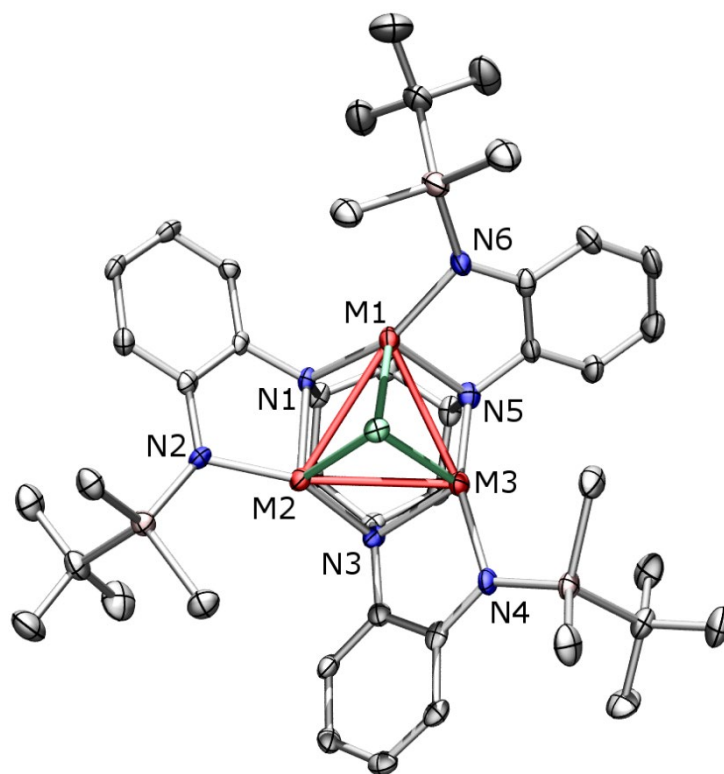


Figure S38. Solid state structure of $[\text{NBu}_4][[(\text{tbsL})\text{Fe}_n\text{Zn}_{3-n}(\mu^3\text{-Cl})]$ (**9**). Hydrogen atoms and tetrabutylammonium counter ion have been omitted for clarity. Because each metal site is a mixture of Fe and Zn, the metal ions are depicted in red for generic metal M.

Table S9. Selected bond metrics for **9** (Å).

M1—M2	2.8674(9)	M2—N2	1.943(3)
M1—M3	2.9830(8)	M2—N3	1.973(3)
M2—M3	2.9860(9)	M2—N1	2.107(3)
M1—Cl1	2.641(1)	M3—N4	1.935(3)
M2—Cl1	2.568(1)	M3—N5	1.982(3)
M3—Cl1	2.501(1)	M3—N3	2.095(3)
M1—N6	1.921(3)	M—M _{avg}	2.945(1)
M1—N1	1.966(3)	M—Cl _{avg}	2.570(1)
M1—N5	2.076(3)		

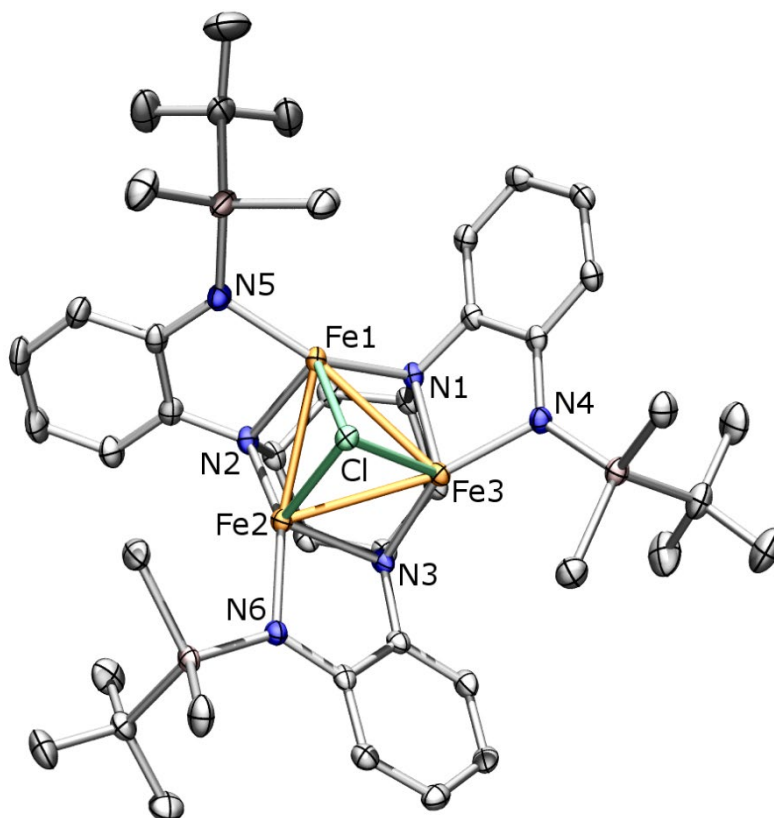


Figure S39. Solid state structure of $[\text{NBu}_4][[(\text{tbsL})\text{Fe}_3(\mu^3\text{-Cl})]$ (**7**). Hydrogen atoms and tetrabutylammonium counter ion have been omitted for clarity.

Table S10. Selected bond metrics for **7** (Å).

Fe1—Fe2	2.8061(8)	Fe2—N6	1.959(2)
Fe1—Fe3	2.6813(8)	Fe2—N2	1.989(3)
Fe2—Fe3	2.8281(8)	Fe2—N3	2.108(2)
Fe1—Cl1	2.5579(8)	Fe3—N4	1.961(3)
Fe2—Cl1	2.4785(7)	Fe3—N3	1.986(2)
Fe3—Cl1	2.5177(8)	Fe3—N1	2.112(2)
Fe1—N5	1.951(3)	Fe—Fe _{avg}	2.772(1)
Fe1—N1	1.988(2)	Fe—Cl _{avg}	2.518(1)
Fe1—N2	2.100(2)		

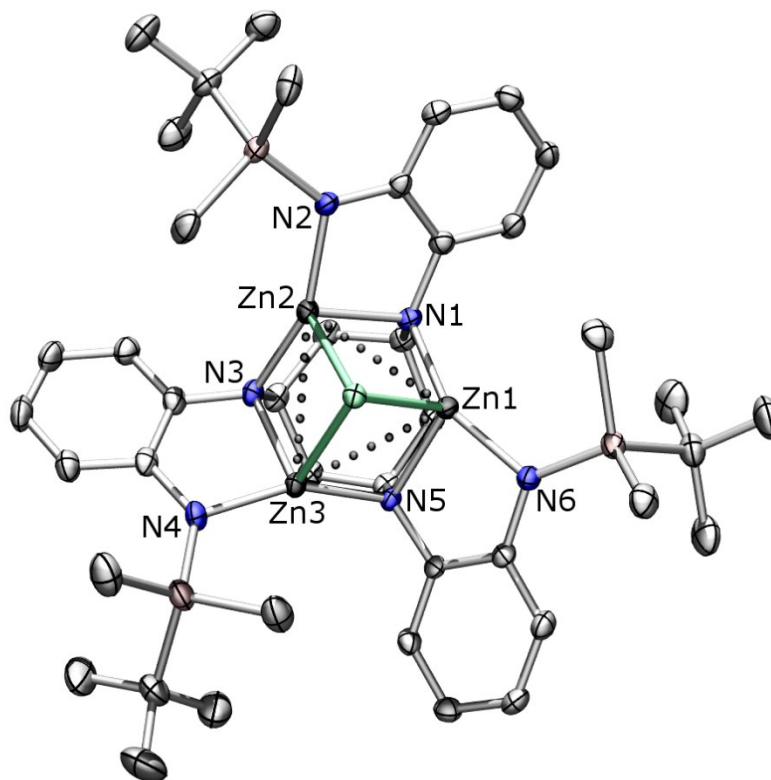


Figure S40. Solid state structure of $[\text{NBu}_4][((\text{tbs})\text{L})\text{Zn}_3(\mu^3\text{-Cl})]$ (**8**). Hydrogen atoms and tetrabutylammonium chloride counter ion omitted for clarity.

Table S11. Selected bond metrics for **8** (Å).

Zn1—Zn2	3.0176(8)	Zn2—N1	2.058(2)
Zn1—Zn3	2.9444(8)	Zn2—N2	1.916(2)
Zn2—Zn3	2.9942(8)	Zn2—N3	1.962(2)
Zn1—Cl1	2.6173(8)	Zn3—N4	1.906(2)
Zn2—Cl1	2.5016(7)	Zn3—N5	1.949(2)
Zn3—Cl1	2.6209(8)	Zn3—N3	2.046(2)
Zn1—N6	1.910(2)	Zn—Zn _{avg}	2.985(1)
Zn1—N1	1.949(2)	Zn—Cl _{avg}	2.579(1)
Zn1—N5	2.062(2)		

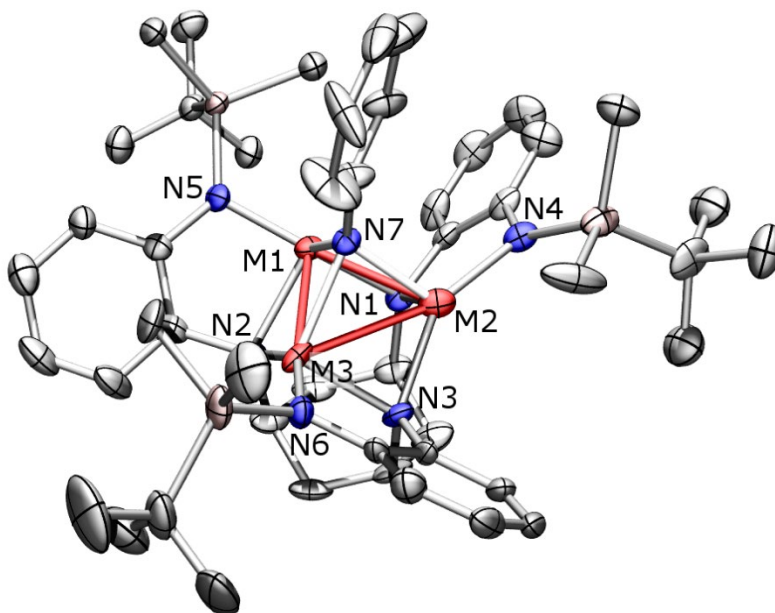


Figure S41. Solid state structure of $(^{tbs}L)Fe_2Zn(\mu^3-NPh)$ (**11**). Hydrogen atoms have been omitted for clarity. Only one part of the disorder is depicted for clarity. Because each metal site is a mixture of Fe and Zn, the metal ions are depicted in red for generic metal M.

Table S12. Selected bond metrics for **11** (Å).

M1—M2	2.725(1)	M2—N1	2.056(4)
M1—M3	2.678(1)	M2—N3	1.995(4)
M2—M3	2.7010(9)	M2—N4	1.888(4)
M1—N7	2.004(4)	M3—N2	2.004(4)
M2—N7	2.060(4)	M3—N3	2.056(4)
M3—N7	1.968(4)	M3—N6	1.902(4)
M1—N5	1.893(5)	M—M _{avg}	2.701(2)
M1—N1	1.975(5)	M—N7 _{avg}	2.011(7)
M1—N2	2.042(4)		

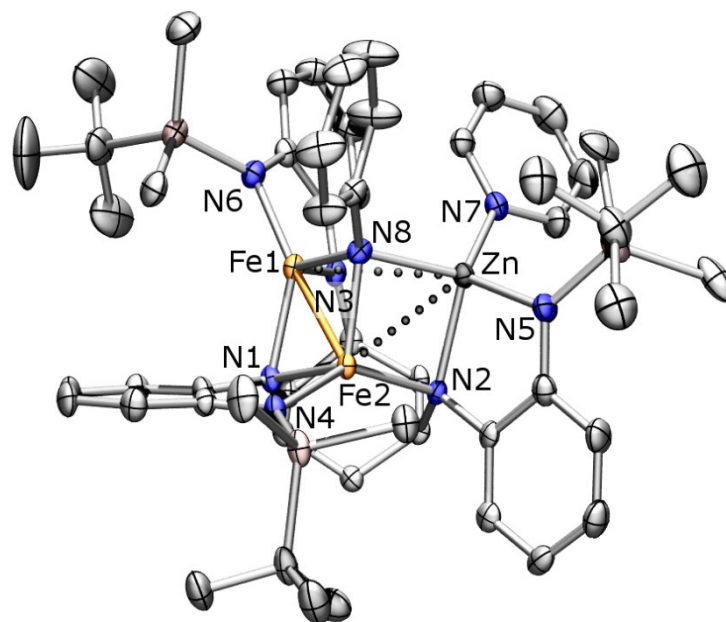


Figure S42. Solid state structure of $(^{tbs}L)Fe_2Zn((\mu^3-NPh)(py))$ (**12**). Hydrogen atoms have been omitted for clarity. Only one part of the disorder is depicted for clarity.

Table S13. Selected bond metrics for **12** (Å).

Fe1—Fe2	2.4880(6)	Fe1—N6	1.917(3)
Fe1—Zn	3.2059(7)	Fe2—N1	2.014(2)
Fe2—Zn	2.7599(6)	Fe2—N2	2.025(3)
Fe1—N8	1.917(2)	Fe2—N4	1.934(3)
Fe2—N8	1.961(2)	Zn—N2	2.083(2)
Zn—N8	2.077(3)	Zn—N5	1.957(3)
Fe1—N1	1.992(2)	Zn—N7	2.039(3)
Fe1—N3	1.924(3)		

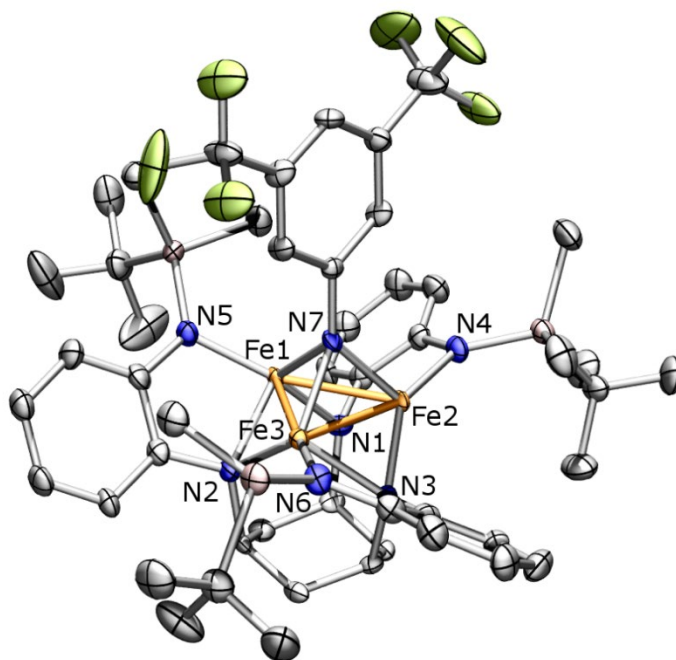


Figure S43. Solid state structure of $(^{tbs}L)Fe_3(\mu^3-N(3,5-(CF_3)_2C_6H_3))$ (**13**). Hydrogen atoms have been omitted for clarity. Only one part of the disorder is depicted for clarity.

Table S14. Selected bond metrics for **13** (Å).^a

Fe1—Fe2	2.457(2)	Fe2—N1	2.026(5)
Fe1—Fe3	2.668(2)	Fe2—N3	1.923(5)
Fe2—Fe3	2.520(2)	Fe2—N4	1.887(5)
Fe1—N7	1.977(5)	Fe3—N2	1.994(5)
Fe2—N7	1.931(5)	Fe3—N3	2.051(5)
Fe3—N7	1.972(5)	Fe3—N6	1.919(5)
Fe1—N1	2.020(5)	Fe—Fe _{avg}	2.548(3)
Fe1—N2	2.081(5)	Fe—N7 _{avg}	1.960(9)
Fe1—N5	1.929(5)		

^a Listed bond metrics are for the Fe centers with >90% occupancy

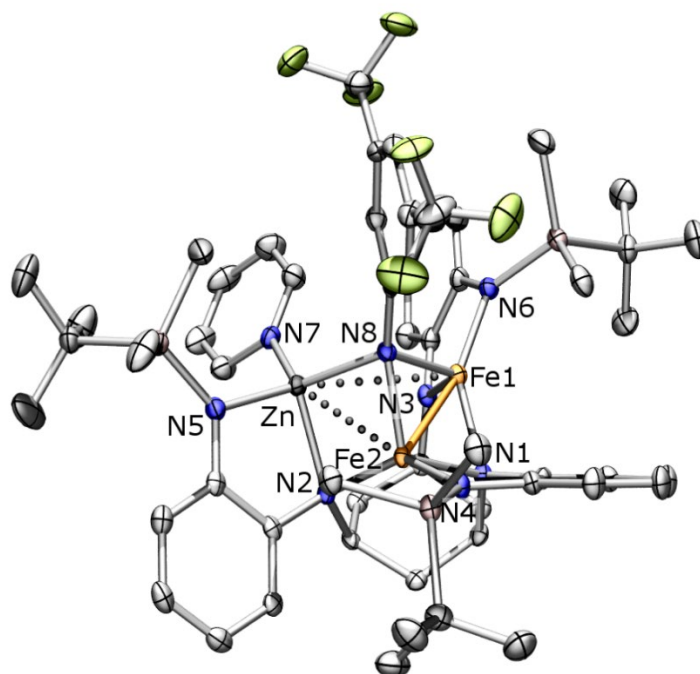


Figure S44. Solid state structure of $(^{tbs}L)Fe_2Zn(\mu^3-N(3,5-(CF_3)_2C_6H_3))(py)$ (**14**). Hydrogen atoms have been omitted for clarity. Only one part of the disorder is depicted for clarity.

Table S15. Selected bond metrics for **14** (Å).

Fe1—Fe2	2.5008(6)	Fe1—N6	1.914(2)
Fe1—Zn	3.1982(6)	Fe2—N1	2.021(2)
Fe2—Zn	2.7806(4)	Fe2—N2	2.030(2)
Fe1—N8	1.932(2)	Fe2—N4	1.925(2)
Fe2—N8	1.976(2)	Zn—N2	2.089(2)
Zn—N8	2.084(2)	Zn—N5	1.959(2)
Fe1—N1	1.973(2)	Zn—N7	2.060(2)
Fe1—N3	1.922(2)		

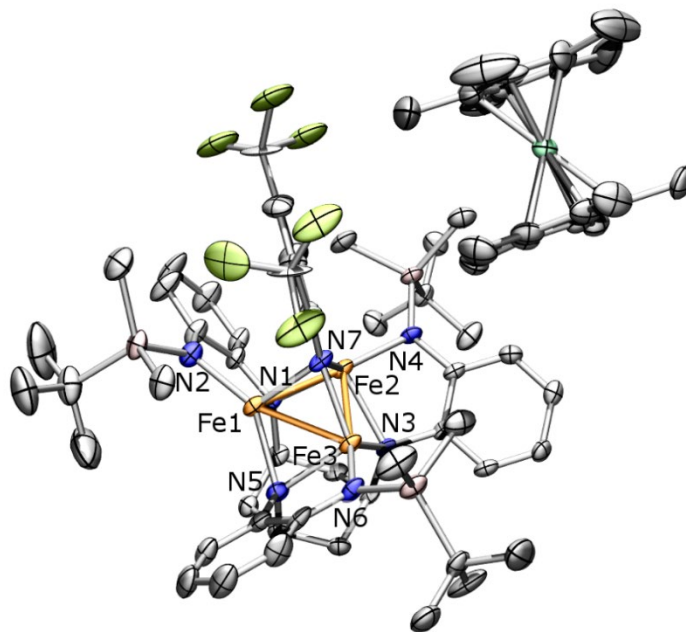


Figure S45. Solid state structure of $[\text{CoCp}^*_2][(\text{tbsL})\text{Fe}_3(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))]$ (**15**). Hydrogen atoms have been omitted for clarity. Only one part of the disorder is depicted for clarity.

Table S16. Selected bond metrics for **15** (Å).

Fe1—Fe2	2.532(1)	Fe2—N1	2.002(5)
Fe1—Fe3	2.632(1)	Fe2—N3	2.065(4)
Fe2—Fe3	2.596(1)	Fe2—N4	1.933(5)
Fe1—N7	1.953(5)	Fe3—N3	2.010(4)
Fe2—N7	1.978(4)	Fe3—N5	2.076(5)
Fe3—N7	1.996(4)	Fe3—N6	1.953(5)
Fe1—N1	2.068(5)	Fe—Fe _{avg}	2.587(1)
Fe1—N2	1.957(4)	Fe—N7 _{avg}	1.976(7)
Fe1—N5	2.013(4)		

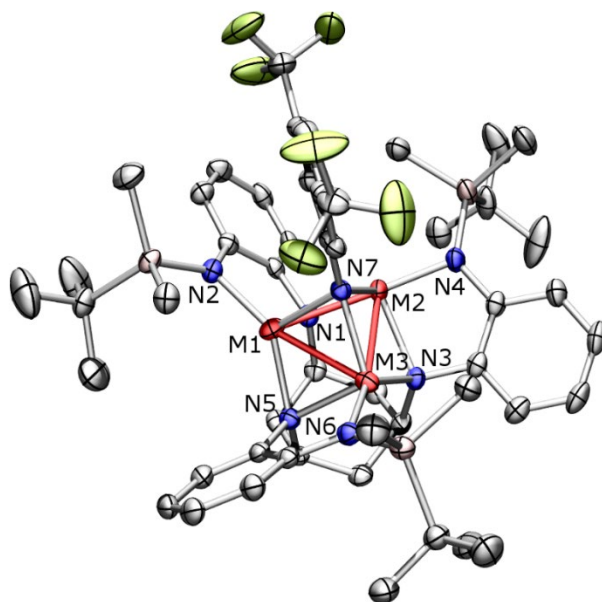


Figure S46. Solid state structure of $[2,2,2\text{-cryptand}(\text{K})][(\text{tbsL})\text{Fe}_2\text{Zn}(\mu^3\text{-N}(3,5\text{-}(\text{CF}_3)_2\text{C}_6\text{H}_3))]$ (**16**). Hydrogen atoms, disordered solvent molecules, and counter ion have been omitted for clarity. Only one cluster from the unit cell is depicted. Because each metal site is a mixture of Fe and Zn, the metal ions are depicted in red for generic metal M.

Table S17. Selected bond metrics for two clusters (A and B) in unit cell of **16** (Å).

M1 _A —M2 _A	2.7085(7)	M1 _B —M2 _B	2.7650(7)
M1 _A —M3 _A	2.6751(8)	M1 _B —M3 _B	2.6660(7)
M2 _A —M3 _A	2.7843(6)	M2 _B —M3 _B	2.6918(9)
M1 _A —N7 _A	2.032(3)	M1 _B —N7 _B	2.014(2)
M2 _A —N7 _A	2.035(3)	M2 _B —N7 _B	2.101(3)
M3 _A —N7 _A	2.010(2)	M3 _B —N7 _B	1.967(3)
M1 _A —N1 _A	2.062(3)	M1 _B —N1 _B	2.006(3)
M1 _A —N5 _A	1.974(2)	M1 _B —N5 _B	2.085(3)
M1 _A —N2 _A	1.923(3)	M1 _B —N2 _B	1.926(3)
M2 _A —N1 _A	2.015(3)	M2 _B —N1 _B	2.090(2)
M2 _A —N3 _A	2.104(2)	M2 _B —N3 _B	1.974(3)
M2 _A —N4 _A	1.939(3)	M2 _B —N4 _B	1.910(4)
M3 _A —N3 _A	2.008(3)	M3 _B —N3 _B	2.096(3)
M3 _A —N5 _A	2.104(3)	M3 _B —N5 _B	2.011(3)
M3 _A —N6 _A	1.935(3)	M3 _B —N6 _B	1.935(3)
M _A —M _A avg	2.723(1)	M _B —M _B avg	2.708(1)
M _A —N7 _A avg	2.026(5)	M _B —N7 _B avg	2.027(5)

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checkCIF/PLATON report

You have not supplied any structure factors. As a result the full set of tests cannot be run.

THIS REPORT IS FOR GUIDANCE ONLY. IF USED AS PART OF A REVIEW PROCEDURE FOR PUBLICATION, IT SHOULD NOT REPLACE THE EXPERTISE OF AN EXPERIENCED CRYSTALLOGRAPHIC REFEREE.

No syntax errors found. CIF dictionary Interpreting this report

Datablock: 2

Bond precision:	C-C = 0.0041 A	Wavelength=0.41328	
Cell:	a=10.4451(12)	b=27.261(3)	c=21.186(3)
	alpha=90	beta=102.166(4)	gamma=90
Temperature:	100 K		
	Calculated	Reported	
Volume	5897.1(13)	5897.2(12)	
Space group	P 21/c	P 1 21/c 1	
Hall group	-P 2ybc	-P 2ybc	
Moiety formula	4(C47 H71 Fe3 N7 Si3), 7(C4 H10 O)	C47 H71 Fe3 N7 Si3, 1.75(C4 H10 O)	
Sum formula	C216 H354 Fe12 N28 O7 Si12	C54 H88.50 Fe3 N7 O1.75 Si3	
Mr	4462.55	1115.63	
Dx, g cm ⁻³	1.257	1.257	
Z	1	4	
Mu (mm ⁻¹)	0.194	0.165	
F000	2382.0	2382.0	
F000'	2384.23		
h, k, lmax		13, 34, 26	
Nref		12080	
Tmin, Tmax	0.978, 0.993	0.673, 0.744	
Tmin'	0.969		

Correction method= # Reported T Limits: Tmin=0.673 Tmax=0.744
AbsCorr = MULTI-SCAN

Data completeness= Theta(max)= 14.997

R(reflections)= 0.0421(9217) wR2(reflections)=
0.1073(12080)

S = 1.042 Npar= 745

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

● **Alert level C**

PLAT042_ALERT_1_C Calc. and Reported MoietyFormula Strings Differ Please Check
Calc.: 4(C47 H71 Fe3 N7 Si3), 7(C4 H10 O)
Rep.: C47 H71 Fe3 N7 Si3, 1.75(C4 H10 O)

● **Alert level G**

ABSMU01_ALERT_1_G Calculation of _exptl_absorpt_correction_mu
not performed for this radiation type.

PLAT002_ALERT_2_G Number of Distance or Angle Restraints on AtSite 15 Note
PLAT003_ALERT_2_G Number of Uiso or Uij Restrained non-H Atoms ... 20 Report
PLAT045_ALERT_1_G Calculated and Reported Z Differ by a Factor ... 0.250 Check
PLAT092_ALERT_4_G Check: Wavelength Given is not Cu,Ga,Mo,Ag,In Ka 0.41328 Ang.
PLAT175_ALERT_4_G The CIF-Embedded .res File Contains SAME Records 3 Report
PLAT178_ALERT_4_G The CIF-Embedded .res File Contains SIMU Records 1 Report
PLAT302_ALERT_4_G Anion/Solvent/Minor-Residue Disorder (Resd 2) 100% Note
PLAT302_ALERT_4_G Anion/Solvent/Minor-Residue Disorder (Resd 3) 100% Note
PLAT302_ALERT_4_G Anion/Solvent/Minor-Residue Disorder (Resd 4) 100% Note
PLAT302_ALERT_4_G Anion/Solvent/Minor-Residue Disorder (Resd 5) 100% Note
PLAT304_ALERT_4_G Non-Integer Number of Atoms in (Resd 2) 7.58 Check
PLAT304_ALERT_4_G Non-Integer Number of Atoms in (Resd 3) 7.21 Check
PLAT304_ALERT_4_G Non-Integer Number of Atoms in (Resd 4) 7.42 Check
PLAT304_ALERT_4_G Non-Integer Number of Atoms in (Resd 5) 4.04 Check
PLAT720_ALERT_4_G Number of Unusual/Non-Standard Labels 23 Note
H1SA H1SB H1SC H2SA H2SB H3SA H3SB H4SA
H4SB H4SC H5SA H5SB H5SC H6SA H6SB H7SA
H7SB H8SA H8SB H8SC H9SA H9SB H9SC

PLAT789_ALERT_4_G Atoms with Negative _atom_site_disorder_group # 60 Check
PLAT790_ALERT_4_G Centre of Gravity not Within Unit Cell: Resd. # 3 Note
C4 H10 O
PLAT790_ALERT_4_G Centre of Gravity not Within Unit Cell: Resd. # 5 Note
C4 H10 O
PLAT822_ALERT_4_G CIF-embedded .res Contains Negative PART Numbers 4 Check
PLAT860_ALERT_3_G Number of Least-Squares Restraints 264 Note
PLAT883_ALERT_1_G No Info/Value for _atom_sites_solution_primary . Please Do !
PLAT933_ALERT_2_G Number of HKL-OMIT Records in Embedded .res File 7 Note
1 0 0, 0 1 1, 0 1 2, 0 2 0, -1 1 1, 1 1 0,
-1 0 2,

0 **ALERT level A** = Most likely a serious problem - resolve or explain
0 **ALERT level B** = A potentially serious problem, consider carefully
1 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
23 **ALERT level G** = General information/check it is not something unexpected

4 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
3 ALERT type 2 Indicator that the structure model may be wrong or deficient
1 ALERT type 3 Indicator that the structure quality may be low
16 ALERT type 4 Improvement, methodology, query or suggestion
0 ALERT type 5 Informative message, check

Datablock: 4

Bond precision: C-C = 0.0037 Å Wavelength=0.71073
Cell: a=11.040(2) b=14.175(3) c=21.290(5)
alpha=106.479(4) beta=93.058(4) gamma=106.770(4)
Temperature: 100 K

	Calculated	Reported
Volume	3025.6(11)	3025.6(11)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	C46 H75.24 Fe2 N6 Na O Si3, C8 H19.72 Na O2	C54 H94.963 Fe2 N6 Na2 O3 Si3
Sum formula	C54 H94.96 Fe2 N6 Na2 O3 Si3	C54 H94.96 Fe2 N6 Na2 O3 Si3
Mr	1118.27	1118.27
Dx, g cm ⁻³	1.227	1.227
Z	2	2
Mu (mm ⁻¹)	0.597	0.597
F000	1201.9	1202.0
F000'	1204.20	
h, k, lmax		13, 17, 25
Nref		11243
Tmin, Tmax	0.867, 0.942	0.696, 0.745
Tmin'	0.836	

Correction method= # Reported T Limits: Tmin=0.696 Tmax=0.745
AbsCorr = MULTI-SCAN

Data completeness= Theta(max)= 25.709

R(reflections)= 0.0396(8448) wR2(reflections)=
0.0926(11243)
S = 1.015 Npar= 728

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.
Click on the hyperlinks for more details of the test.



Alert level C

PLAT042_ALERT_1_C Calc. and Reported MoietyFormula Strings Differ Please Check
Calc: C46 H75.24 Fe2 N6 Na O Si3, C8 H19.72 Na O2
Rep.: C54 H94.963 Fe2 N6 Na2 O3 Si3

PLAT094_ALERT_2_C	Ratio of Maximum / Minimum Residual Density	2.33	Report
PLAT220_ALERT_2_C	NonSolvent Resd 1 C Ueq(max)/Ueq(min) Range	3.9	Ratio
PLAT222_ALERT_3_C	NonSolvent Resd 1 H Uiso(max)/Uiso(min) Range	5.0	Ratio
PLAT230_ALERT_2_C	Hirshfeld Test Diff for Si2 --C25 .	5.3	s.u.

Alert level G

PLAT002_ALERT_2_G	Number of Distance or Angle Restraints on AtSite	8	Note
PLAT154_ALERT_1_G	The s.u.'s on the Cell Angles are Equal ..(Note)	0.004	Degree
PLAT171_ALERT_4_G	The CIF-Embedded .res File Contains EADP Records	21	Report
PLAT175_ALERT_4_G	The CIF-Embedded .res File Contains SAME Records	1	Report
PLAT180_ALERT_4_G	Check Cell Rounding: # of Values Ending with 0 =	3	Note
PLAT230_ALERT_2_G	Hirshfeld Test Diff for C43 --C44 .	7.7	s.u.
PLAT232_ALERT_2_G	Hirshfeld Test Diff (M-X) Na2 --O3A .	6.9	s.u.
PLAT301_ALERT_3_G	Main Residue Disorder(Resd 1)	27%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder (Resd 2)	45%	Note
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in (Resd 1)	134.24	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in (Resd 2)	30.72	Check
PLAT412_ALERT_2_G	Short Intra XH3 .. XHn H13C ..H40E .	1.90	Ang.
	x,y,z =	1_555	Check
PLAT412_ALERT_2_G	Short Intra XH3 .. XHn H14C ..H17A .	1.95	Ang.
	x,y,z =	1_555	Check
PLAT764_ALERT_4_G	Overcomplete CIF Bond List Detected (Rep/Expd) .	1.13	Ratio
PLAT811_ALERT_5_G	No ADDSYM Analysis: Too Many Excluded Atoms	!	Info
PLAT860_ALERT_3_G	Number of Least-Squares Restraints	6	Note
PLAT883_ALERT_1_G	No Info/Value for _atom_sites_solution_primary .		Please Do !
PLAT941_ALERT_3_G	Average HKL Measurement Multiplicity	2.3	Low

0 **ALERT level A** = Most likely a serious problem - resolve or explain
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4 ALERT type 3 Indicator that the structure quality may be low
7 ALERT type 4 Improvement, methodology, query or suggestion
1 ALERT type 5 Informative message, check

Datablock: 5

Bond precision:	C-C = 0.0047 A	Wavelength=0.40659	
Cell:	a=10.3835 (18)	b=18.015 (2)	c=30.663 (7)
	alpha=90	beta=91.703 (6)	gamma=90
Temperature:	100 K		

	Calculated	Reported
Volume	5733.3(18)	5733.3(17)
Space group	P 21/n	P 1 21/n 1
Hall group	-P 2yn	-P 2yn
Moiety formula	C46 H74 Fe2 N6 O Si3 Zn, C6 H14 [+ solvent]	C46 H74 Fe2 N6 O Si3 Zn, C6 H14
Sum formula	C52 H88 Fe2 N6 O Si3 Zn [+ solvent]	C52 H88 Fe2 N6 O Si3 Zn
Mr	1074.64	1074.62
Dx, g cm ⁻³	1.245	1.245
Z	4	4
Mu (mm ⁻¹)	0.226	0.507
F000	2296.0	2296.0
F000'	2298.39	
h, k, lmax		14, 25, 43
Nref		17045
Tmin, Tmax	0.952, 0.990	0.628, 0.719
Tmin'	0.931	

Correction method= # Reported T Limits: Tmin=0.628 Tmax=0.719
AbsCorr = MULTI-SCAN

Data completeness= Theta(max)= 16.741

R(reflections)= 0.0593(10447)

wR2(reflections)=
0.1480(17045)

S = 1.015

Npar= 603

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.
Click on the hyperlinks for more details of the test.

Alert level C

RINTA01_ALERT_3_C The value of Rint is greater than 0.12
Rint given 0.120

Alert level G

ABSMU01_ALERT_1_G Calculation of _exptl_absorpt_correction_mu
not performed for this radiation type.

PLAT020_ALERT_3_G The Value of Rint is Greater Than 0.12	0.120	Report
PLAT051_ALERT_1_G Mu(calc) and Mu(CIF) Ratio Differs from 1.0 by .	55.34	%
PLAT092_ALERT_4_G Check: Wavelength Given is not Cu,Ga,Mo,Ag,In Ka	0.40659	Ang.
PLAT605_ALERT_4_G Largest Solvent Accessible VOID in the Structure	202	A**3
PLAT794_ALERT_5_G Tentative Bond Valency for Zn1 (II) .	1.64	Info
PLAT869_ALERT_4_G ALERTS Related to the Use of SQUEEZE Suppressed		! Info
PLAT883_ALERT_1_G No Info/Value for _atom_sites_solution_primary .		Please Do !

PLAT933_ALERT_2_G Number of HKL-OMIT Records in Embedded .res File 5 Note
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 PLAT941_ALERT_3_G Average HKL Measurement Multiplicity 4.5 Low
 PLAT984_ALERT_1_G The Fe-f' = 0.2886 Deviates from the B&C-Value 0.1938 Check
 PLAT984_ALERT_1_G The Si-f' = 0.0522 Deviates from the B&C-Value 0.0258 Check
 PLAT984_ALERT_1_G The Zn-f' = 0.3242 Deviates from the B&C-Value 0.2556 Check
 PLAT985_ALERT_1_G The Fe-f" = 0.5448 Deviates from the B&C-Value 0.2969 Check
 PLAT985_ALERT_1_G The Si-f" = 0.0431 Deviates from the B&C-Value 0.0221 Check
 PLAT985_ALERT_1_G The Zn-f" = 0.9375 Deviates from the B&C-Value 0.5209 Check

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 16 **ALERT level G** = General information/check it is not something unexpected

9 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
 1 ALERT type 2 Indicator that the structure model may be wrong or deficient
 3 ALERT type 3 Indicator that the structure quality may be low
 3 ALERT type 4 Improvement, methodology, query or suggestion
 1 ALERT type 5 Informative message, check

Datablock: 6

Bond precision: C-C = 0.0056 A

Wavelength=0.40651

Cell: a=10.6298(8) b=11.1655(9) c=23.6682(18)
 alpha=98.248(2) beta=101.941(2) gamma=101.506(2)
 Temperature: 100 K

	Calculated	Reported
Volume	2642.6(4)	2642.6(4)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	C47 H71 Fe2 N7 Si3 Zn, 0.5(C6 H14)	C47 H71 Fe2 N7 Si3 Zn, C3 H7
Sum formula	C50 H78 Fe2 N7 Si3 Zn	C50 H78 Fe2 N7 Si3 Zn
Mr	1038.55	1038.53
Dx, g cm ⁻³	1.305	1.305
Z	2	2
Mu (mm ⁻¹)	0.244	0.213
F000	1102.0	1102.0
F000'	1103.19	
h, k, lmax		12, 13, 28
Nref		8937
Tmin, Tmax	0.985, 0.996	0.633, 0.744
Tmin'	0.964	

Correction method= # Reported T Limits: Tmin=0.633 Tmax=0.744
AbsCorr = MULTI-SCAN

Data completeness= Theta(max)= 14.031

R(reflections)= 0.0441(7925)

wR2(reflections)=
0.1106(8937)


S = 1.094

Npar= 584

The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

 **Alert level A**

PLAT051_ALERT_1_A Mu(calc) and Mu(CIF) Ratio Differs from 1.0 by . 14.58 %

Author Response: Structure collected with synchrotron radiation; difference in Mu(calc) from Shelx program and checkcif due to rounding errors in f', f'', and mu values.

 **Alert level B**

PLAT029_ALERT_3_B _diffrn_measured_fraction_theta_full value Low . 0.950 Why?

Author Response: Structure collected with fixed chi geometry at synchrotron source; hardware limit prevents >95% completeness.

 **Alert level C**

PLAT042_ALERT_1_C Calc. and Reported MoietyFormula Strings Differ Please Check

Calc: C47 H71 Fe2 N7 Si3 Zn, 0.5(C6 H14)

Rep.: C47 H71 Fe2 N7 Si3 Zn, C3 H7

PLAT094_ALERT_2_C Ratio of Maximum / Minimum Residual Density 2.05 Report

PLAT232_ALERT_2_C Hirshfeld Test Diff (M-X) Fe2 --N5 . 5.6 s.u.

 **Alert level G**

ABSMU01_ALERT_1_G Calculation of _exptl_absorpt_correction_mu
not performed for this radiation type.

PLAT083_ALERT_2_G SHELXL Second Parameter in WGHT Unusually Large 7.03 Why ?

PLAT092_ALERT_4_G Check: Wavelength Given is not Cu,Ga,Mo,Ag,In Ka 0.40651 Ang.

PLAT154_ALERT_1_G The s.u.'s on the Cell Angles are Equal ..(Note) 0.002 Degree

PLAT794_ALERT_5_G Tentative Bond Valency for Zn1 (II) . 1.68 Info

PLAT883_ALERT_1_G No Info/Value for _atom_sites_solution_primary . Please Do !

PLAT933_ALERT_2_G Number of HKL-OMIT Records in Embedded .res File 11 Note

0 0 2, 0 2 1, -1 2 2, -1 -1 3, -1 1 5, -1 -3 1,
-1 0 4, -1 -4 3, -1 -3 3, -2 0 7, -1 -4 2,

1 **ALERT level A** = Most likely a serious problem - resolve or explain
1 **ALERT level B** = A potentially serious problem, consider carefully
3 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
7 **ALERT level G** = General information/check it is not something unexpected

5 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
4 ALERT type 2 Indicator that the structure model may be wrong or deficient
1 ALERT type 3 Indicator that the structure quality may be low
1 ALERT type 4 Improvement, methodology, query or suggestion
1 ALERT type 5 Informative message, check

Datablock: 7

Bond precision: C-C = 0.0045 A Wavelength=0.71073

Cell: a=26.839(3) b=19.358(2) c=27.601(4)
alpha=90 beta=90 gamma=90

Temperature: 100 K

	Calculated	Reported
Volume	14340(3)	14340(3)
Space group	P b c a	P b c a
Hall group	-P 2ac 2ab	-P 2ac 2ab
Moiety formula	C42 H66 Cl Fe3 N6 Si3, C15 H34 N [+ solvent]	C42 H66 Cl Fe3 N6 Si3, C15 H34 N
Sum formula	C57 H100 Cl Fe3 N7 Si3 [+ solvent]	C57 H100 Cl Fe3 N7 Si3
Mr	1170.71	1170.70
Dx, g cm ⁻³	1.084	1.085
Z	8	8
Mu (mm ⁻¹)	0.722	0.722
F000	5024.0	5024.0
F000'	5036.37	
h, k, lmax		32, 23, 33
Nref		13621
Tmin, Tmax	0.771, 0.866	0.645, 0.745
Tmin'	0.749	

Correction method= # Reported T Limits: Tmin=0.645 Tmax=0.745
AbsCorr = MULTI-SCAN

Data completeness= Theta(max)= 25.694

R(reflections)= 0.0426(11027)

wR2(reflections)=
0.1258(13621)

S = 1.040

Npar= 661

The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

Alert level B

PLAT094_ALERT_2_B Ratio of Maximum / Minimum Residual Density 6.63 Report

Author Response: Remaining residual density is located next to a disordered butyl group on the tetrabutyl ammonium counterion. The large Q peak is accounted for by a failure of the disorder modeling to successfully fit all of the remaining electron density.

PLAT097_ALERT_2_B Large Reported Max. (Positive) Residual Density 3.04 eA-3

Author Response: Remaining residual density is located next to a disordered butyl group on the tetrabutyl ammonium counterion. The large Q peak is accounted for by a failure of the disorder modeling to successfully fit all of the remaining electron density.

Alert level C

DIFMX02_ALERT_1_C The maximum difference density is > 0.1*ZMAX*0.75

The relevant atom site should be identified.

PLAT220_ALERT_2_C NonSolvent Resd 1 C Ueq(max)/Ueq(min) Range 3.7 Ratio

PLAT222_ALERT_3_C NonSolvent Resd 1 H Uiso(max)/Uiso(min) Range 4.3 Ratio

Alert level G

PLAT083_ALERT_2_G SHELXL Second Parameter in WGHT Unusually Large 31.70 Why ?

PLAT171_ALERT_4_G The CIF-Embedded .res File Contains EADP Records 1 Report

PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Fe1 --C11 . 6.2 s.u.

PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Fe2 --C11 . 6.0 s.u.

PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Fe3 --C11 . 7.2 s.u.

PLAT302_ALERT_4_G Anion/Solvent/Minor-Residue Disorder (Resd 2) 13% Note

PLAT410_ALERT_2_G Short Intra H...H Contact H43B ..H55C . 1.84 Ang.

x,y,z = 1_555 Check

Author Response: H43B and H55C are on the tetrabutyl ammonium counterion. H55C is on the butyl group which has been modeled for disorder. The short interatomic distance is a result of this modeling.

PLAT410_ALERT_2_G Short Intra H...H Contact H44B ..H54A . 2.02 Ang.
x,y,z = 1_555 Check

Author Response: H43B and H55C are on the tetrabutyl ammonium counterion. H55C is on the butyl group which has been modeled for disorder. The short interatomic distance is a result of this modeling.

PLAT410_ALERT_2_G Short Intra H...H Contact H44B ..H54D . 2.09 Ang.
x,y,z = 1_555 Check

Author Response: H43B and H55C are on the tetrabutyl ammonium counterion. H55C is on the butyl group which has been modeled for disorder. The short interatomic distance is a result of this modeling.

PLAT606_ALERT_4_G Solvent Accessible VOID(S) in Structure ! Info
PLAT720_ALERT_4_G Number of Unusual/Non-Standard Labels 3 Note
COAA HOAA HOAB
PLAT869_ALERT_4_G ALERTS Related to the Use of SQUEEZE Suppressed ! Info
PLAT883_ALERT_1_G No Info/Value for _atom_sites_solution_primary . Please Do !

0 **ALERT level A** = Most likely a serious problem - resolve or explain
2 **ALERT level B** = A potentially serious problem, consider carefully
3 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
13 **ALERT level G** = General information/check it is not something unexpected

2 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
10 ALERT type 2 Indicator that the structure model may be wrong or deficient
1 ALERT type 3 Indicator that the structure quality may be low
5 ALERT type 4 Improvement, methodology, query or suggestion
0 ALERT type 5 Informative message, check

Datablock: 8

Bond precision: C-C = 0.0038 A Wavelength=0.71073

Cell: a=27.1744 (9) b=19.1482 (7) c=27.5979 (10)
alpha=90 beta=90 gamma=90

Temperature: 100 K

	Calculated	Reported
Volume	14360.3(9)	14360.3(9)
Space group	P b c a	P b c a
Hall group	-P 2ac 2ab	-P 2ac 2ab
Moiety formula	C42 H66 Cl N6 Si3 Zn3, C16 H36 N [+ solvent]	C42 H66 Cl N6 Si3 Zn3, C16 H36 N
Sum formula	C58 H102 Cl N7 Si3 Zn3 [+ solvent]	C58 H102 Cl N7 Si3 Zn3
Mr	1213.36	1213.29
Dx, g cm ⁻³	1.122	1.122
Z	8	8
Mu (mm ⁻¹)	1.117	1.117
F000	5184.0	5184.0
F000'	5194.86	
h, k, lmax		32, 22, 32
Nref		12718
Tmin, Tmax	0.646, 0.654	0.680, 0.745
Tmin'	0.633	

Correction method= # Reported T Limits: Tmin=0.680 Tmax=0.745

AbsCorr = MULTI-SCAN

Data completeness=

Theta(max)= 25.056

R(reflections)= 0.0332 (9744)

wR2(reflections)=
0.0791 (12718)

S = 1.028

Npar= 676

The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

● Alert level C

PLAT220_ALERT_2_C	NonSolvent	Resd 1	C	Ueq(max)/Ueq(min)	Range	3.9	Ratio
PLAT222_ALERT_3_C	NonSolvent	Resd 1	H	Uiso(max)/Uiso(min)	Range	4.1	Ratio
PLAT223_ALERT_4_C	Solv./Anion	Resd 2	H	Ueq(max)/Ueq(min)	Range	4.1	Ratio
PLAT250_ALERT_2_C	Large U3/U1	Ratio for Average U(i, j)	Tensor		2.2	Note

● Alert level G

PLAT083_ALERT_2_G	SHELXL	Second Parameter in WGHT	Unusually Large			9.88	Why ?
PLAT171_ALERT_4_G	The CIF-Embedded .res File	Contains EADP Records				2	Report
PLAT232_ALERT_2_G	Hirshfeld Test Diff (M-X)	Zn01	--C11	.		12.3	s.u.
PLAT232_ALERT_2_G	Hirshfeld Test Diff (M-X)	Zn02	--C11	.		12.0	s.u.
PLAT232_ALERT_2_G	Hirshfeld Test Diff (M-X)	Zn03	--C11	.		13.3	s.u.
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue	Disorder (Resd 2)				12%	Note

PLAT410_ALERT_2_G Short Intra H...H Contact H47B ..H56C . 1.97 Ang.
 x,y,z = 1_555 Check
 PLAT413_ALERT_2_G Short Inter XH3 .. XHn H40C ..H57D . 1.98 Ang.
 1-x,1-y,1-z = 5_666 Check

Author Response: H57D is located on a disordered butyl group on the tetrabutyl ammonium counterion. Modeling of this disorder results in unusually short interatomic distances in some cases.

PLAT606_ALERT_4_G Solvent Accessible VOID(S) in Structure ! Info
 PLAT720_ALERT_4_G Number of Unusual/Non-Standard Labels 3 Note
 Zn01 Zn02 Zn03
 PLAT794_ALERT_5_G Tentative Bond Valency for Zn01 (II) . 1.73 Info
 PLAT794_ALERT_5_G Tentative Bond Valency for Zn02 (II) . 1.77 Info
 PLAT794_ALERT_5_G Tentative Bond Valency for Zn03 (II) . 1.75 Info
 PLAT869_ALERT_4_G ALERTS Related to the Use of SQUEEZE Suppressed ! Info
 PLAT883_ALERT_1_G No Info/Value for _atom_sites_solution_primary . Please Do !

0 **ALERT level A** = Most likely a serious problem - resolve or explain
 0 **ALERT level B** = A potentially serious problem, consider carefully
 4 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
 15 **ALERT level G** = General information/check it is not something unexpected

1 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
 8 ALERT type 2 Indicator that the structure model may be wrong or deficient
 1 ALERT type 3 Indicator that the structure quality may be low
 6 ALERT type 4 Improvement, methodology, query or suggestion
 3 ALERT type 5 Informative message, check

Datablock: 9

Bond precision: C-C = 0.0062 A Wavelength=0.71073

Cell: a=27.053 (4) b=19.227 (3) c=27.643 (4)
 alpha=90 beta=90 gamma=90

Temperature: 100 K

	Calculated	Reported
Volume	14379(4)	14378(4)
Space group	P b c a	P b c a
Hall group	-P 2ac 2ab	-P 2ac 2ab
Moiety formula	C42 H66 Cl Fe2 N6 Si3 Zn, C16 H36 N [+ solvent]	C42 H66 Cl Fe2 N6 Si3 Zn1, C16 H36 N
Sum formula	C58 H102 Cl Fe2 N7 Si3 Zn [+ solvent]	C58 H102 Cl Fe2 N7 Si3 Zn
Mr	1194.28	1194.25
Dx, g cm ⁻³	1.103	1.103
Z	8	8
Mu (mm ⁻¹)	0.852	0.852
F000	5120.0	5120.0
F000'	5131.87	
h, k, lmax		32, 22, 32
Nref		12755
Tmin, Tmax	0.767, 0.873	0.626, 0.745
Tmin'	0.755	

Correction method= # Reported T Limits: Tmin=0.626 Tmax=0.745
AbsCorr = MULTI-SCAN

Data completeness= Theta(max)= 25.103

R(reflections)= 0.0563(8699) wR2(reflections)=
0.1307(12755)
S = 1.021 Npar= 670

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.
Click on the hyperlinks for more details of the test.

Alert level C

RINTA01_ALERT_3_C The value of Rint is greater than 0.12
Rint given 0.151

PLAT220_ALERT_2_C NonSolvent Resd 1 C Ueq(max)/Ueq(min) Range	4.2 Ratio
PLAT222_ALERT_3_C NonSolvent Resd 1 H Uiso(max)/Uiso(min) Range	4.6 Ratio
PLAT341_ALERT_3_C Low Bond Precision on C-C Bonds	0.00619 Ang.

Alert level G

PLAT020_ALERT_3_G The Value of Rint is Greater Than 0.12 0.151 Report
PLAT042_ALERT_1_G Calc. and Reported MoietyFormula Strings Differ Please Check
Calc: C42 H66 Cl Fe2 N6 Si3 Zn, C16 H36 N
Rep.: C42 H66 Cl Fe2 N6 Si3 Zn1, C16 H36 N

PLAT083_ALERT_2_G SHELXL Second Parameter in WGHT Unusually Large	22.96 Why ?
PLAT168_ALERT_4_G The CIF-Embedded .res File Contains EXYZ Records	3 Report


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PLAT171_ALERT_4_G The CIF-Embedded .res File Contains EADP Records          5 Report
PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Zn1      --Cl1      .      21.0 s.u.
PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Zn2      --Cl1      .      22.5 s.u.
PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Zn3      --Cl1      .      7.5 s.u.
PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Fe1A     --Cl1      .      21.0 s.u.
PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Fe2A     --Cl1      .      22.5 s.u.
PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Fe3A     --Cl1      .      7.5 s.u.
PLAT300_ALERT_4_G Atom Site Occupancy of Zn1         Constrained at 0.3333 Check
PLAT300_ALERT_4_G Atom Site Occupancy of Zn2         Constrained at 0.3333 Check
PLAT300_ALERT_4_G Atom Site Occupancy of Zn3         Constrained at 0.3333 Check
PLAT300_ALERT_4_G Atom Site Occupancy of Fe1A        Constrained at 0.6667 Check
PLAT300_ALERT_4_G Atom Site Occupancy of Fe2A        Constrained at 0.6667 Check
PLAT300_ALERT_4_G Atom Site Occupancy of Fe3A        Constrained at 0.6667 Check
PLAT301_ALERT_3_G Main Residue Disorder .....(Resd 1 )      5% Note
PLAT302_ALERT_4_G Anion/Solvent/Minor-Residue Disorder (Resd 2 ) 12% Note
PLAT410_ALERT_2_G Short Intra H...H Contact H51A     ..H56B     .      1.99 Ang.
                                     x,y,z = 1_555 Check
PLAT606_ALERT_4_G Solvent Accessible VOID(S) in Structure ..... ! Info
PLAT720_ALERT_4_G Number of Unusual/Non-Standard Labels ..... 10 Note
      C1AA  C3AA  H3AA  H3AB  C0AA  H0AA  C2AA  H2AA
      H2AB  H2AC
PLAT869_ALERT_4_G ALERTS Related to the Use of SQUEEZE Suppressed ! Info
PLAT883_ALERT_1_G No Info/Value for _atom_sites_solution_primary . Please Do !

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0 ALERT level A = Most likely a serious problem - resolve or explain
0 ALERT level B = A potentially serious problem, consider carefully
4 ALERT level C = Check. Ensure it is not caused by an omission or oversight
24 ALERT level G = General information/check it is not something unexpected

2 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
9 ALERT type 2 Indicator that the structure model may be wrong or deficient
5 ALERT type 3 Indicator that the structure quality may be low
12 ALERT type 4 Improvement, methodology, query or suggestion
0 ALERT type 5 Informative message, check

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Datablock: 11

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Bond precision:  C-C = 0.0082 A          Wavelength=0.71073

Cell:           a=22.150 (4)           b=20.383 (3)           c=26.428 (4)
                alpha=90              beta=101.975 (3)       gamma=90

Temperature:    100 K

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	Calculated	Reported
Volume	11672 (3)	11672 (3)
Space group	C 2/c	C 1 2/c 1
Hall group	-C 2yc	-C 2yc
Moiety formula	C48 H71 Fe2 N7 Si3 Zn [+ solvent]	C48 H71 Fe2 N7 Si3 Zn, 1[C6H14]
Sum formula	C48 H71 Fe2 N7 Si3 Zn [+ solvent]	C54 H85 Fe2 N7 Si3 Zn
Mr	1007.48	1093.62
Dx, g cm ⁻³	1.147	1.245
Z	8	8
Mu (mm ⁻¹)	0.994	0.999
F000	4256.0	4656.0
F000'	4266.50	
h, k, lmax		26, 24, 31
Nref		10413
Tmin, Tmax	0.787, 0.905	0.647, 0.745
Tmin'	0.741	

Correction method= # Reported T Limits: Tmin=0.647 Tmax=0.745
AbsCorr = MULTI-SCAN

Data completeness= Theta (max)= 25.158

R(reflections)= 0.0681 (7862)

wR2(reflections)=
0.1430 (10413)

S = 1.090

Npar= 568

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.
Click on the hyperlinks for more details of the test.

Alert level C

PLAT213_ALERT_2_C	Atom C45	has ADP max/min Ratio	3.4	prolat
PLAT213_ALERT_2_C	Atom C5A	has ADP max/min Ratio	3.4	prolat
PLAT220_ALERT_2_C	NonSolvent Resd 1 C	Ueq(max)/Ueq(min) Range	4.1	Ratio
PLAT234_ALERT_4_C	Large Hirshfeld Difference C45	--C46	0.21	Ang.
PLAT241_ALERT_2_C	High 'MainMol'	Ueq as Compared to Neighbors of	C14	Check
PLAT242_ALERT_2_C	Low 'MainMol'	Ueq as Compared to Neighbors of	C27	Check
PLAT341_ALERT_3_C	Low Bond Precision on	C-C Bonds	0.0082	Ang.

Alert level G

FORMU01_ALERT_2_G There is a discrepancy between the atom counts in the
_chemical_formula_sum and the formula from the _atom_site* data.
Atom count from _chemical_formula_sum: C54 H85 Fe2 N7 Si3 Zn1
Atom count from the _atom_site data: C48 H71. Fe2. N7 Si3 Zn1.0049

CELLZ01_ALERT_1_G Difference between formula and atom_site contents detected.

CELLZ01_ALERT_1_G ALERT: Large difference may be due to a
symmetry error - see SYMMG tests

From the CIF: _cell_formula_units_Z 8

From the CIF: _chemical_formula_sum C54 H85 Fe2 N7 Si3 Zn

TEST: Compare cell contents of formula and atom_site data

atom	Z*formula	cif sites	diff
C	432.00	384.00	48.00
H	680.00	568.00	112.00
Fe	16.00	16.00	-0.00
N	56.00	56.00	0.00
Si	24.00	24.00	0.00
Zn	8.00	8.00	0.00

PLAT003_ALERT_2_G Number of Uiso or Uij Restrained non-H Atoms ... 2 Report

PLAT041_ALERT_1_G Calc. and Reported SumFormula Strings Differ Please Check

Calc: C48 H71 Fe2 N7 Si3 Zn

Rep.: C54 H85 Fe2 N7 Si3 Zn

PLAT042_ALERT_1_G Calc. and Reported MoietyFormula Strings Differ Please Check

Calc: C48 H71 Fe2 N7 Si3 Zn

Rep.: C48 H71 Fe2 N7 Si3 Zn, 1[C6H14]

PLAT083_ALERT_2_G SHELXL Second Parameter in WGHT Unusually Large 126.69 Why ?

PLAT168_ALERT_4_G The CIF-Embedded .res File Contains EXYZ Records 5 Report

PLAT171_ALERT_4_G The CIF-Embedded .res File Contains EADP Records 14 Report

PLAT186_ALERT_4_G The CIF-Embedded .res File Contains ISOR Records 1 Report

PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Zn1 --N7 . 8.0 s.u.

PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Zn2 --N7 . 6.0 s.u.

PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Fe1 --N7 . 8.0 s.u.

PLAT232_ALERT_2_G Hirshfeld Test Diff (M-X) Fe2 --N7 . 6.0 s.u.

PLAT300_ALERT_4_G Atom Site Occupancy of Zn1 Constrained at 0.3333 Check

PLAT300_ALERT_4_G Atom Site Occupancy of Zn2 Constrained at 0.3333 Check

PLAT300_ALERT_4_G Atom Site Occupancy of Zn3 Constrained at 0.3333 Check

PLAT300_ALERT_4_G Atom Site Occupancy of Fe1 Constrained at 0.6667 Check

PLAT300_ALERT_4_G Atom Site Occupancy of Fe2 Constrained at 0.6667 Check

PLAT300_ALERT_4_G Atom Site Occupancy of Fe3 Constrained at 0.6667 Check

PLAT301_ALERT_3_G Main Residue Disorder(Resd 1) 23% Note

PLAT411_ALERT_2_G Short Inter H...H Contact H22 ..H1A . 2.09 Ang.

1-x,y,1/2-z = 2_655 Check

PLAT412_ALERT_2_G Short Intra XH3 .. XHn H9 ..H13B . 2.13 Ang.

x,y,z = 1_555 Check

PLAT412_ALERT_2_G Short Intra XH3 .. XHn H9 ..H13B . 2.13 Ang.

x,y,z = 1_555 Check

PLAT412_ALERT_2_G Short Intra XH3 .. XHn H38A ..H48 . 2.08 Ang.

x,y,z = 1_555 Check

PLAT606_ALERT_4_G Solvent Accessible VOID(S) in Structure ! Info

PLAT720_ALERT_4_G Number of Unusual/Non-Standard Labels 2 Note

H4AA H6AA

PLAT860_ALERT_3_G Number of Least-Squares Restraints 12 Note

PLAT868_ALERT_4_G ALERTS Due to the Use of _smtbx_masks Suppressed ! Info

PLAT883_ALERT_1_G No Info/Value for _atom_sites_solution_primary . Please Do !

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0 **ALERT level B** = A potentially serious problem, consider carefully

7 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight

30 **ALERT level G** = General information/check it is not something unexpected

5 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
16 ALERT type 2 Indicator that the structure model may be wrong or deficient
3 ALERT type 3 Indicator that the structure quality may be low
13 ALERT type 4 Improvement, methodology, query or suggestion
0 ALERT type 5 Informative message, check

Datablock: 12

Bond precision: C-C = 0.0053 A Wavelength=0.40651

Cell: a=12.4239(10) b=14.2358(12) c=19.0442(16)
alpha=109.6426(17) beta=95.5642(18) gamma=102.0493(18)

Temperature: 100 K

	Calculated	Reported
Volume	3050.8(4)	3050.8(4)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	C53 H76 Fe2 N8 Si3 Zn [+ solvent]	C53 H76 Fe2 N8 Si3 Zn, 0.75[C6H14]
Sum formula	C53 H76 Fe2 N8 Si3 Zn [+ solvent]	C57.50 H86.50 Fe2 N8 Si3 Zn
Mr	1086.58	1151.18
Dx, g cm ⁻³	1.183	1.253
Z	2	2
Mu (mm ⁻¹)	0.213	0.186
F000	1148.0	1223.0
F000'	1149.17	
h, k, lmax		14, 16, 22
Nref		10455
Tmin, Tmax	0.967, 0.982	0.670, 0.744
Tmin'	0.946	

Correction method= # Reported T Limits: Tmin=0.670 Tmax=0.744
AbsCorr = MULTI-SCAN

Data completeness= Theta(max)= 14.015

R(reflections)= 0.0426(8583) wR2(reflections)=
0.1294(10455)

S = 1.035 Npar= 646

The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

● **Alert level C**

PLAT029_ALERT_3_C _diffrn_measured_fraction_theta_full value Low . 0.968 Why?
PLAT220_ALERT_2_C NonSolvent Resd 1 C Ueq(max)/Ueq(min) Range 5.3 Ratio
PLAT222_ALERT_3_C NonSolvent Resd 1 H Uiso(max)/Uiso(min) Range 6.6 Ratio
PLAT242_ALERT_2_C Low 'MainMol' Ueq as Compared to Neighbors of C39 Check

● **Alert level G**

FORMU01_ALERT_2_G There is a discrepancy between the atom counts in the
_chemical_formula_sum and the formula from the _atom_site* data.
Atom count from _chemical_formula_sum: C57.5 H86.5 Fe2 N8 Si3 Zn1
Atom count from the _atom_site data: C53 H76 Fe2 N8 Si3 Zn1

ABSMU01_ALERT_1_G Calculation of _exptl_absorpt_correction_mu
not performed for this radiation type.

CELLZ01_ALERT_1_G Difference between formula and atom_site contents detected.
CELLZ01_ALERT_1_G ALERT: Large difference may be due to a
symmetry error - see SYMMG tests
From the CIF: _cell_formula_units_Z 2
From the CIF: _chemical_formula_sum C57.50 H86.50 Fe2 N8 Si3 Zn
TEST: Compare cell contents of formula and atom_site data

atom	Z*formula	cif sites	diff
C	115.00	106.00	9.00
H	173.00	152.00	21.00
Fe	4.00	4.00	0.00
N	16.00	16.00	0.00
Si	6.00	6.00	0.00
Zn	2.00	2.00	0.00

PLAT002_ALERT_2_G Number of Distance or Angle Restraints on AtSite 4 Note
PLAT041_ALERT_1_G Calc. and Reported SumFormula Strings Differ Please Check
Calc: C53 H76 Fe2 N8 Si3 Zn
Rep.: C57.50 H86.50 Fe2 N8 Si3 Zn

PLAT042_ALERT_1_G Calc. and Reported MoietyFormula Strings Differ Please Check
Calc: C53 H76 Fe2 N8 Si3 Zn
Rep.: C53 H76 Fe2 N8 Si3 Zn, 0.75[C6H14]

PLAT092_ALERT_4_G Check: Wavelength Given is not Cu,Ga,Mo,Ag,In Ka 0.40651 Ang.
PLAT171_ALERT_4_G The CIF-Embedded .res File Contains EADP Records 7 Report
PLAT176_ALERT_4_G The CIF-Embedded .res File Contains SADI Records 2 Report
PLAT191_ALERT_3_G A Non-default SADI Restraint Value has been used 0.0010 Report
PLAT191_ALERT_3_G A Non-default SADI Restraint Value has been used 0.0010 Report
PLAT301_ALERT_3_G Main Residue Disorder(Resd 1) 10% Note
PLAT412_ALERT_2_G Short Intra XH3 .. XHn H21 ..H28F . 2.09 Ang.
x,y,z = 1_555 Check

PLAT606_ALERT_4_G Solvent Accessible VOID(S) in Structure ! Info
PLAT860_ALERT_3_G Number of Least-Squares Restraints 4 Note
PLAT868_ALERT_4_G ALERTS Due to the Use of _smtbx_masks Suppressed ! Info
PLAT883_ALERT_1_G No Info/Value for _atom_sites_solution_primary . Please Do !
PLAT933_ALERT_2_G Number of HKL-OMIT Records in Embedded .res File 28 Note

-2	1	0,	-2	1	1,	-2	2	0,	-2	2	1,	-1	0	2,	-1	0	3,
-1	0	5,	-1	2	0,	0	-2	2,	0	-2	3,	0	-1	3,	0	-1	4,
0	0	3,	0	1	2,	1	-3	1,	1	-2	2,	1	-1	3,	1	0	1,
1	1	0,	1	1	1,	1	2	1,	2	-3	4,	2	-2	1,	2	-2	4,

2 -1 1, 2 -1 2, 2 0 3, 2 1 3,

0 **ALERT level A** = Most likely a serious problem - resolve or explain
0 **ALERT level B** = A potentially serious problem, consider carefully
4 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
19 **ALERT level G** = General information/check it is not something unexpected

6 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
6 ALERT type 2 Indicator that the structure model may be wrong or deficient
6 ALERT type 3 Indicator that the structure quality may be low
5 ALERT type 4 Improvement, methodology, query or suggestion
0 ALERT type 5 Informative message, check

Datablock: 13

Bond precision: C-C = 0.0085 Å Wavelength=0.71073

Cell: a=15.1571(7) b=21.4890(11) c=16.3741(8)
alpha=90 beta=93.577(1) gamma=90

Temperature: 100 K

	Calculated	Reported
Volume	5322.8(5)	5322.8(4)
Space group	P 21/n	P 1 21/n 1
Hall group	-P 2yn	-P 2yn
Moiety formula	C50 H69 F6 Fe3 N7 Si3	C50 H69 F6 Fe2.741 N7 Si3, 0.086(Fe3)
Sum formula	C50 H69 F6 Fe3 N7 Si3	C50 H69 F6 Fe3 N7 Si3
Mr	1133.95	1133.94
Dx, g cm ⁻³	1.415	1.415
Z	4	4
Mu (mm ⁻¹)	0.937	0.937
F000	2368.0	2368.0
F000'	2373.92	
h, k, lmax		18, 25, 19
Nref		9399
Tmin, Tmax	0.873, 0.942	0.635, 0.745
Tmin'	0.873	

Correction method= # Reported T Limits: Tmin=0.635 Tmax=0.745
AbsCorr = MULTI-SCAN

Data completeness= Theta(max)= 25.115

R(reflections)= 0.0664(6425)

wR2(reflections)=
0.1942(9399)

S = 1.018

Npar= 648

The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

Alert level C

PLAT042_ALERT_1_C	Calc. and Reported MoietyFormula Strings Differ	Please Check
	Calc: C50 H69 F6 Fe3 N7 Si3	
	Rep.: C50 H69 F6 Fe2.741 N7 Si3, 0.086(Fe3)	
PLAT213_ALERT_2_C	Atom F5 has ADP max/min Ratio	3.2 prolat
PLAT213_ALERT_2_C	Atom C30 has ADP max/min Ratio	3.3 prolat
PLAT213_ALERT_2_C	Atom C3A has ADP max/min Ratio	3.3 prolat
PLAT220_ALERT_2_C	NonSolvent Resd 1 C Ueq(max)/Ueq(min) Range	4.6 Ratio
PLAT222_ALERT_3_C	NonSolvent Resd 1 H Uiso(max)/Uiso(min) Range	5.8 Ratio
PLAT230_ALERT_2_C	Hirshfeld Test Diff for F1 --C49 .	6.8 s.u.
PLAT230_ALERT_2_C	Hirshfeld Test Diff for F3 --C49 .	5.2 s.u.
PLAT341_ALERT_3_C	Low Bond Precision on C-C Bonds	0.00853 Ang.

Alert level G

PLAT002_ALERT_2_G	Number of Distance or Angle Restraints on AtSite	8 Note
PLAT003_ALERT_2_G	Number of Uiso or Uij Restrained non-H Atoms ...	8 Report
PLAT083_ALERT_2_G	SHELXL Second Parameter in WGHT Unusually Large	21.24 Why ?
PLAT168_ALERT_4_G	The CIF-Embedded .res File Contains EXYZ Records	1 Report
PLAT171_ALERT_4_G	The CIF-Embedded .res File Contains EADP Records	9 Report
PLAT175_ALERT_4_G	The CIF-Embedded .res File Contains SAME Records	1 Report
PLAT177_ALERT_4_G	The CIF-Embedded .res File Contains DELU Records	2 Report
PLAT178_ALERT_4_G	The CIF-Embedded .res File Contains SIMU Records	2 Report
PLAT232_ALERT_2_G	Hirshfeld Test Diff (M-X) Fe1 --N3 .	5.5 s.u.
PLAT232_ALERT_2_G	Hirshfeld Test Diff (M-X) Fe2 --N3 .	5.5 s.u.
PLAT232_ALERT_2_G	Hirshfeld Test Diff (M-X) Fe2 --N5 .	6.0 s.u.
PLAT232_ALERT_2_G	Hirshfeld Test Diff (M-X) Fe3A --N5 .	6.0 s.u.
PLAT242_ALERT_2_G	Low 'MainMol' Ueq as Compared to Neighbors of	C49 Check
PLAT242_ALERT_2_G	Low 'MainMol' Ueq as Compared to Neighbors of	C50 Check
PLAT301_ALERT_3_G	Main Residue Disorder(Resd 1)	10% Note
PLAT720_ALERT_4_G	Number of Unusual/Non-Standard Labels	9 Note
	H3AA H3AB H3AC H4AA H4AB H4AC H5AA H5AB	
	H5AC	
PLAT860_ALERT_3_G	Number of Least-Squares Restraints	54 Note
PLAT883_ALERT_1_G	No Info/Value for _atom_sites_solution_primary .	Please Do !
PLAT941_ALERT_3_G	Average HKL Measurement Multiplicity	3.1 Low

-
- 0 **ALERT level A** = Most likely a serious problem - resolve or explain
0 **ALERT level B** = A potentially serious problem, consider carefully
9 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
19 **ALERT level G** = General information/check it is not something unexpected
- 2 **ALERT type 1** CIF construction/syntax error, inconsistent or missing data

15 ALERT type 2 Indicator that the structure model may be wrong or deficient
5 ALERT type 3 Indicator that the structure quality may be low
6 ALERT type 4 Improvement, methodology, query or suggestion
0 ALERT type 5 Informative message, check

Datablock: 14

Bond precision: C-C = 0.0036 A Wavelength=0.71073
Cell: a=12.3372(6) b=14.5673(7) c=19.6915(10)
alpha=76.058(1) beta=73.437(1) gamma=79.142(1)
Temperature: 100 K

	Calculated	Reported
Volume	3264.8(3)	3264.8(3)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	C55 H74 F5.93 Fe2 N8 Si3 Zn, 0.065(F) [+ solvent]	C55 H74 F6 Fe2 N8 Si3 Zn, 1.5[C6H6]
Sum formula	C55 H74 F6 Fe2 N8 Si3 Zn [+ solvent]	C64 H83 F6 Fe2 N8 Si3 Zn
Mr	1222.53	1339.72
Dx, g cm ⁻³	1.244	1.363
Z	2	2
Mu (mm ⁻¹)	0.914	0.920
F000	1275.9	1402.0
F000'	1278.80	
h, k, lmax		14, 17, 23
Nref		11581
Tmin, Tmax	0.750, 0.817	0.690, 0.745
Tmin'	0.661	

Correction method= # Reported T Limits: Tmin=0.690 Tmax=0.745
AbsCorr = MULTI-SCAN

Data completeness= Theta(max)= 25.065

R(reflections)= 0.0327(9245) wR2(reflections)=
0.0791(11581)
S = 1.026 Npar= 712

The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

Alert level C

PLAT220_ALERT_2_C	NonSolvent	Resd 1	C	Ueq(max)/Ueq(min)	Range	3.9	Ratio
PLAT222_ALERT_3_C	NonSolvent	Resd 1	H	Uiso(max)/Uiso(min)	Range	4.9	Ratio
PLAT230_ALERT_2_C	Hirshfeld	Test Diff	for	Si3	--C38	.	6.3 s.u.
PLAT242_ALERT_2_C	Low	'MainMol'	Ueq	as Compared to	Neighbors of	C15	Check

Alert level G

FORMU01_ALERT_2_G There is a discrepancy between the atom counts in the
_chemical_formula_sum and the formula from the _atom_site* data.
Atom count from _chemical_formula_sum: C64 H83 F6 Fe2 N8 Si3 Zn1
Atom count from the _atom_site data: C55 H74 F6.001999 Fe2 N8 Si3 Zn1

CELLZ01_ALERT_1_G Difference between formula and atom_site contents detected.
CELLZ01_ALERT_1_G ALERT: Large difference may be due to a
symmetry error - see SYMMG tests
From the CIF: _cell_formula_units_Z 2
From the CIF: _chemical_formula_sum C64 H83 F6 Fe2 N8 Si3 Zn
TEST: Compare cell contents of formula and atom_site data

atom	Z*formula	cif sites	diff
C	128.00	110.00	18.00
H	166.00	148.00	18.00
F	12.00	11.99	0.01
Fe	4.00	4.00	0.00
N	16.00	16.00	0.00
Si	6.00	6.00	0.00
Zn	2.00	2.00	0.00

PLAT041_ALERT_1_G Calc. and Reported SumFormula Strings Differ Please Check
Calc: C55 H74 F6 Fe2 N8 Si3 Zn
Rep.: C64 H83 F6 Fe2 N8 Si3 Zn

PLAT042_ALERT_1_G Calc. and Reported MoietyFormula Strings Differ Please Check
Calc: C55 H74 F5.93 Fe2 N8 Si3 Zn, 0.065(F)
Rep.: C55 H74 F6 Fe2 N8 Si3 Zn, 1.5[C6H6]

PLAT154_ALERT_1_G The s.u.'s on the Cell Angles are Equal ..(Note) 0.001 Degree

PLAT171_ALERT_4_G The CIF-Embedded .res File Contains EADP Records 3 Report

PLAT230_ALERT_2_G Hirshfeld Test Diff for F1 --C54 . 7.7 s.u.

PLAT230_ALERT_2_G Hirshfeld Test Diff for F3 --C54 . 6.3 s.u.

PLAT242_ALERT_2_G Low 'MainMol' Ueq as Compared to Neighbors of C55 Check

PLAT301_ALERT_3_G Main Residue Disorder(Resd 1) 4% Note

PLAT302_ALERT_4_G Anion/Solvent/Minor-Residue Disorder (Resd 2) 100% Note

PLAT304_ALERT_4_G Non-Integer Number of Atoms in (Resd 1) 148.93 Check

PLAT304_ALERT_4_G Non-Integer Number of Atoms in (Resd 2) 0.06 Check

PLAT432_ALERT_2_G Short Inter X...Y Contact F1 ..C34 . 2.95 Ang.
-1+x,y,z = 1_455 Check

PLAT432_ALERT_2_G Short Inter X...Y Contact F3A ..C54 . 1.74 Ang.
x,y,z = 1_555 Check

PLAT432_ALERT_2_G Short Inter X...Y Contact F3A ..C52 . 2.46 Ang.
x,y,z = 1_555 Check

PLAT606_ALERT_4_G Solvent Accessible VOID(S) in Structure ! Info

PLAT720_ALERT_4_G Number of Unusual/Non-Standard Labels 3 Note
F0AA F1AA F2AA

PLAT860_ALERT_3_G Number of Least-Squares Restraints 1 Note

PLAT868_ALERT_4_G ALERTS Due to the Use of _smtbx_masks Suppressed ! Info
 PLAT883_ALERT_1_G No Info/Value for _atom_sites_solution_primary . Please Do !
 PLAT941_ALERT_3_G Average HKL Measurement Multiplicity 3.1 Low

0 **ALERT level A** = Most likely a serious problem - resolve or explain
 0 **ALERT level B** = A potentially serious problem, consider carefully
 4 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
 23 **ALERT level G** = General information/check it is not something unexpected

6 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
 10 ALERT type 2 Indicator that the structure model may be wrong or deficient
 4 ALERT type 3 Indicator that the structure quality may be low
 7 ALERT type 4 Improvement, methodology, query or suggestion
 0 ALERT type 5 Informative message, check

Datablock: 15

Bond precision:	C-C = 0.0089 A	Wavelength=0.71073	
Cell:	a=11.9733 (5)	b=26.2570 (9)	c=26.2043 (10)
	alpha=90	beta=100.148 (1)	gamma=90
Temperature:	100 K		
	Calculated	Reported	
Volume	8109.3 (5)	8109.3 (5)	
Space group	P 21/n	P 1 21/n 1	
Hall group	-P 2yn	-P 2yn	
	C50 H69 F6 Fe3 N7 Si3, C20	C50 H69 F6 Fe3 N7 Si3, C20	
Moiety formula	H30 Co, C4 H10 O [+ solvent]	H30 Co, C4 H10 O, 1[C4H8O]	
Sum formula	C74 H109 Co F6 Fe3 N7 O Si3 [+ solvent]	C78 H117 Co F6 Fe3 N7 O2 Si3	
Mr	1537.44	1609.53	
Dx, g cm ⁻³	1.259	1.318	
Z	4	4	
Mu (mm ⁻¹)	0.829	0.833	
F000	3244.0	3404.0	
F000'	3251.56		
h, k, lmax		14, 31, 31	
Nref		14337	
Tmin, Tmax	0.847, 0.936	0.683, 0.745	
Tmin'	0.847		

Correction method= # Reported T Limits: Tmin=0.683 Tmax=0.745
 AbsCorr = MULTI-SCAN

Data completeness=

Theta(max)= 25.089

R(reflections)= 0.0691(8091)

wR2(reflections)=
0.1765(14337)

S = 1.026

Npar= 878

The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

● **Alert level C**

PLAT220_ALERT_2_C	NonSolvent	Resd 1	C	Ueq(max)/Ueq(min)	Range	5.0	Ratio	
PLAT222_ALERT_3_C	NonSolvent	Resd 1	H	Uiso(max)/Uiso(min)	Range	5.9	Ratio	
PLAT234_ALERT_4_C	Large Hirshfeld	Difference	C54	--C55	.	0.18	Ang.	
PLAT242_ALERT_2_C	Low	'MainMol'	Ueq	as Compared to Neighbors of		C15	Check	
PLAT242_ALERT_2_C	Low	'MainMol'	Ueq	as Compared to Neighbors of		C27	Check	
PLAT250_ALERT_2_C	Large	U3/U1	Ratio	for Average U(i,j)	Tensor	2.1	Note	
PLAT260_ALERT_2_C	Large	Average	Ueq	of Residue Including	01	0.117	Check	
PLAT341_ALERT_3_C	Low	Bond	Precision	on C-C Bonds		0.0089	Ang.	
PLAT360_ALERT_2_C	Short	C(sp3)-C(sp3)	Bond	C71	- C72	.	1.42	Ang.

● **Alert level G**

FORMU01_ALERT_2_G There is a discrepancy between the atom counts in the
_chemical_formula_sum and the formula from the _atom_site* data.
Atom count from _chemical_formula_sum: C78 H117 Co1 F6 Fe3 N7 O2 Si3
Atom count from the _atom_site data: C74 H109 Co1 F6 Fe3 N7 O1 Si3

CELLZ01_ALERT_1_G Difference between formula and atom_site contents detected.

CELLZ01_ALERT_1_G ALERT: Large difference may be due to a
symmetry error - see SYMMG tests

From the CIF: _cell_formula_units_Z 4

From the CIF: _chemical_formula_sum C78 H117 Co F6 Fe3 N7 O2 Si3

TEST: Compare cell contents of formula and atom_site data

atom	Z*formula	cif sites	diff
C	312.00	296.00	16.00
H	468.00	436.00	32.00
Co	4.00	4.00	0.00
F	24.00	24.00	0.00
Fe	12.00	12.00	0.00
N	28.00	28.00	0.00
O	8.00	4.00	4.00
Si	12.00	12.00	0.00

PLAT002_ALERT_2_G Number of Distance or Angle Restraints on AtSite 14 Note

PLAT041_ALERT_1_G Calc. and Reported SumFormula Strings Differ Please Check

Calc: C74 H109 Co F6 Fe3 N7 O Si3

Rep.: C78 H117 Co F6 Fe3 N7 O2 Si3

PLAT042_ALERT_1_G Calc. and Reported MoietyFormula Strings Differ Please Check

Calc: C50 H69 F6 Fe3 N7 Si3, C20 H30 Co, C4 H10 O

Rep.: C50 H69 F6 Fe3 N7 Si3, C20 H30 Co, C4 H10 O, 1[C4H8O]

PLAT083_ALERT_2_G SHELXL Second Parameter in WGHT Unusually Large 14.44 Why ?

PLAT171_ALERT_4_G The CIF-Embedded .res File Contains EADP Records 4 Report

PLAT176_ALERT_4_G The CIF-Embedded .res File Contains SADI Records 4 Report

PLAT187_ALERT_4_G	The CIF-Embedded .res File Contains RIGU Records	2	Report
PLAT190_ALERT_3_G	A Non-default RIGU Restraint Value for First Par	0.0050	Report
PLAT190_ALERT_3_G	A Non-default RIGU Restraint Value for SecondPar	0.0050	Report
PLAT190_ALERT_3_G	A Non-default RIGU Restraint Value for First Par	0.0010	Report
PLAT190_ALERT_3_G	A Non-default RIGU Restraint Value for SecondPar	0.0010	Report
PLAT230_ALERT_2_G	Hirshfeld Test Diff for F6 --C49 .	7.7	s.u.
PLAT230_ALERT_2_G	Hirshfeld Test Diff for F4A --C49 .	6.5	s.u.
PLAT230_ALERT_2_G	Hirshfeld Test Diff for F5A --C49 .	9.0	s.u.
PLAT242_ALERT_2_G	Low 'MainMol' Ueq as Compared to Neighbors of	C49	Check
PLAT301_ALERT_3_G	Main Residue Disorder(Resd 1)	9%	Note
PLAT380_ALERT_4_G	Incorrectly? Oriented X(sp2)-Methyl Moiety	C59	Check
PLAT606_ALERT_4_G	Solvent Accessible VOID(S) in Structure	!	Info
PLAT720_ALERT_4_G	Number of Unusual/Non-Standard Labels	26	Note
	Ha Hb Hc Hd He Hf Hg Hh		
	Hi Hj Hk Hl Hm Hn Ho Hp		
	Hq Hr Hs Ht Hu Hv Hw Hx		
	Hy Hz		
PLAT794_ALERT_5_G	Tentative Bond Valency for Co1 (III) .	3.31	Info
PLAT860_ALERT_3_G	Number of Least-Squares Restraints	117	Note
PLAT868_ALERT_4_G	ALERTS Due to the Use of _smtbx_masks Suppressed	!	Info
PLAT883_ALERT_1_G	No Info/Value for _atom_sites_solution_primary .	Please Do !	
PLAT941_ALERT_3_G	Average HKL Measurement Multiplicity	4.3	Low

0 **ALERT level A** = Most likely a serious problem - resolve or explain
0 **ALERT level B** = A potentially serious problem, consider carefully
9 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
27 **ALERT level G** = General information/check it is not something unexpected

5 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
13 ALERT type 2 Indicator that the structure model may be wrong or deficient
9 ALERT type 3 Indicator that the structure quality may be low
8 ALERT type 4 Improvement, methodology, query or suggestion
1 ALERT type 5 Informative message, check

Datablock: 16

Bond precision: C-C = 0.0062 A

Wavelength=0.41328

Cell: a=15.4746(7) b=23.8238(11) c=27.1431(13)
alpha=78.401(1) beta=74.315(1) gamma=83.137(1)

Temperature: 100 K

	Calculated	Reported
Volume	9414.8(8)	9414.8(8)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	C50 H69 F6 Fe2 N7 Si3 Zn, C18 H36 K N2 O6, 2.625(C4 H10 O)	C50 H69 F6 Fe2.01 N7 Si3 Zn0.99, C18 H36 K N2 O6, 2.625(C4 H10
Sum formula	C78.50 H131.25 F6 Fe2 K N9 O8.63 Si3 Zn	C78.50 H131.25 F6 Fe2.01 K N9 O8.62 Si3 Zn0.99
Mr	1753.54	1753.51
Dx, g cm ⁻³	1.237	1.237
Z	4	4
Mu (mm ⁻¹)	0.168	0.168
F000	3724.8	3725.0
F000'	3727.58	
h, k, lmax		19, 29, 33
Nref		37723
Tmin, Tmax	0.972, 0.985	0.909, 1.014
Tmin'	0.972	

Correction method= # Reported T Limits: Tmin=0.909 Tmax=1.014
AbsCorr = MULTI-SCAN

Data completeness= Theta(max)= 14.987

R(reflections)= 0.0548(27581) wR2(reflections)=
0.1539(37723)
S = 1.026 Npar= 2087

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.
Click on the hyperlinks for more details of the test.

Alert level C

RINTA01_ALERT_3_C The value of Rint is greater than 0.12
Rint given 0.124

PLAT029_ALERT_3_C _diffn_measured_fraction_theta_full value Low . 0.979 Why?

PLAT041_ALERT_1_C Calc. and Reported SumFormula Strings Differ Please Check
Calc: C78.50 H131.25 F6 Fe2 K N9 O8.63 Si3 Zn
Rep.: C78.50 H131.25 F6 Fe2.01 K N9 O8.62 Si3 Zn0.99

PLAT042_ALERT_1_C Calc. and Reported MoietyFormula Strings Differ Please Check
Calc: C50 H69 F6 Fe2 N7 Si3 Zn, C18 H36 K N2 O6, 2.625(C4 H10 O)
Rep.: C50 H69 F6 Fe2.01 N7 Si3 Zn0.99, C18 H36 K N2 O6, 2.625(C4 H

PLAT077_ALERT_4_C Unitcell Contains Non-integer Number of Atoms .. Please Check

PLAT213_ALERT_2_C Atom F4 has ADP max/min Ratio 3.2 prolat

PLAT213_ALERT_2_C Atom F4A has ADP max/min Ratio 3.2 prolat

PLAT223_ALERT_4_C Solv./Anion Resd 2 H Ueq(max)/Ueq(min) Range 4.1 Ratio

PLAT241_ALERT_2_C	High	'MainMol'	Ueq as Compared to Neighbors of	C125	Check
PLAT241_ALERT_2_C	High	'MainMol'	Ueq as Compared to Neighbors of	C126	Check
PLAT242_ALERT_2_C	Low	'MainMol'	Ueq as Compared to Neighbors of	C66	Check
PLAT244_ALERT_4_C	Low	'Solvent'	Ueq as Compared to Neighbors of	O13	Check
PLAT244_ALERT_4_C	Low	'Solvent'	Ueq as Compared to Neighbors of	O14	Check
PLAT260_ALERT_2_C	Large	Average	Ueq of Residue Including	O13	0.150 Check
PLAT260_ALERT_2_C	Large	Average	Ueq of Residue Including	O14	0.112 Check
PLAT260_ALERT_2_C	Large	Average	Ueq of Residue Including	O16	0.200 Check
PLAT260_ALERT_2_C	Large	Average	Ueq of Residue Including	O17	0.185 Check
PLAT260_ALERT_2_C	Large	Average	Ueq of Residue Including	O16A	0.200 Check
PLAT260_ALERT_2_C	Large	Average	Ueq of Residue Including	O18	0.130 Check
PLAT260_ALERT_2_C	Large	Average	Ueq of Residue Including	O18A	0.130 Check
PLAT341_ALERT_3_C	Low	Bond Precision on	C-C Bonds	0.00621	Ang.

Alert level G

ABSMU01_ALERT_1_G	Calculation of _exptl_absorpt_correction_mu not performed for this radiation type.				
PLAT002_ALERT_2_G	Number of Distance or Angle Restraints on AtSite			43	Note
PLAT003_ALERT_2_G	Number of Uiso or Uij Restrained non-H Atoms ...			28	Report
PLAT012_ALERT_1_G	N.O.K. _shelx_res_checksum Found in CIF				Please Check
PLAT020_ALERT_3_G	The Value of Rint is Greater Than 0.12			0.124	Report
PLAT068_ALERT_1_G	Reported F000 Differs from Calcd (or Missing)...				Please Check
PLAT083_ALERT_2_G	SHELXL Second Parameter in WGHT Unusually Large			14.64	Why ?
PLAT092_ALERT_4_G	Check: Wavelength Given is not Cu,Ga,Mo,Ag,In Ka			0.41328	Ang.
PLAT154_ALERT_1_G	The s.u.'s on the Cell Angles are Equal ..(Note)			0.001	Degree
PLAT168_ALERT_4_G	The CIF-Embedded .res File Contains EXYZ Records			7	Report
PLAT171_ALERT_4_G	The CIF-Embedded .res File Contains EADP Records			25	Report
PLAT175_ALERT_4_G	The CIF-Embedded .res File Contains SAME Records			6	Report
PLAT177_ALERT_4_G	The CIF-Embedded .res File Contains DELU Records			4	Report
PLAT178_ALERT_4_G	The CIF-Embedded .res File Contains SIMU Records			4	Report
PLAT187_ALERT_4_G	The CIF-Embedded .res File Contains RIGU Records			1	Report
PLAT188_ALERT_3_G	A Non-default SIMU Restraint Value has been used			0.0100	Report
PLAT188_ALERT_3_G	A Non-default SIMU Restraint Value has been used			0.0100	Report
PLAT188_ALERT_3_G	A Non-default SIMU Restraint Value has been used			0.0100	Report
PLAT188_ALERT_3_G	A Non-default SIMU Restraint Value has been used			0.0100	Report
PLAT192_ALERT_3_G	A Non-default DELU Restraint Value for First Par			0.0050	Report
PLAT192_ALERT_3_G	A Non-default DELU Restraint Value for SecondPar			0.0050	Report
PLAT192_ALERT_3_G	A Non-default DELU Restraint Value for First Par			0.0050	Report
PLAT192_ALERT_3_G	A Non-default DELU Restraint Value for SecondPar			0.0050	Report
PLAT192_ALERT_3_G	A Non-default DELU Restraint Value for First Par			0.0050	Report
PLAT192_ALERT_3_G	A Non-default DELU Restraint Value for SecondPar			0.0050	Report
PLAT192_ALERT_3_G	A Non-default DELU Restraint Value for First Par			0.0050	Report
PLAT192_ALERT_3_G	A Non-default DELU Restraint Value for SecondPar			0.0050	Report
PLAT242_ALERT_2_G	Low	'MainMol'	Ueq as Compared to Neighbors of	C49	Check
PLAT242_ALERT_2_G	Low	'MainMol'	Ueq as Compared to Neighbors of	C100	Check
PLAT242_ALERT_2_G	Low	'MainMol'	Ueq as Compared to Neighbors of	C101	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of Zn1	Constrained at		0.33	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of Zn2	Constrained at		0.33	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of Zn3	Constrained at		0.33	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of Fe1	Constrained at		0.67	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of Fe2	Constrained at		0.67	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of Fe3	Constrained at		0.67	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of Zn4	Constrained at		0.33	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of Zn5	Constrained at		0.33	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of Zn6	Constrained at		0.33	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of Fe4	Constrained at		0.67	Check

PLAT300_ALERT_4_G	Atom Site Occupancy of Fe5	Constrained at	0.67	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of Fe6	Constrained at	0.67	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of O17	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of C155	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of C156	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of C157	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of C158	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of H15G	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of H15H	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of H15I	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of H15J	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of H15K	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of H15L	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of H15M	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of H15N	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of H15O	Constrained at	0.5	Check
PLAT300_ALERT_4_G	Atom Site Occupancy of H15P	Constrained at	0.5	Check
PLAT301_ALERT_3_G	Main Residue Disorder	(Resd 1)	10%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder	(Resd 2)	4%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder	(Resd 7)	100%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder	(Resd 8)	100%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder	(Resd 9)	100%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder	(Resd 10)	100%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder	(Resd 11)	100%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder	(Resd 12)	100%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder	(Resd 13)	100%	Note
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in	(Resd 7)	9.06	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in	(Resd 8)	10.83	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in	(Resd 9)	7.50	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in	(Resd 10)	5.94	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in	(Resd 11)	4.17	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in	(Resd 12)	3.85	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in	(Resd 13)	7.39	Check
PLAT398_ALERT_2_G	Deviating C-O-C Angle From 120 for O13	.	109.3	Degree
PLAT398_ALERT_2_G	Deviating C-O-C Angle From 120 for O17	.	141.3	Degree
PLAT411_ALERT_2_G	Short Inter H...H Contact H2B ..H15V	.	2.10	Ang.
		1-x,1-y,-z =	2_665	Check
PLAT413_ALERT_2_G	Short Inter XH3 .. XHn H13V ..H1IB	.	1.79	Ang.
		1-x,1-y,1-z =	2_666	Check

Author Response: H atoms are located on a disordered solvent molecule. Modeling of this disorder results in unusually short interatomic distances in some cases.

PLAT432_ALERT_2_G	Short Inter X...Y Contact O13 ..C159	.	2.62	Ang.
		1-x,1-y,1-z =	2_666	Check
PLAT432_ALERT_2_G	Short Inter X...Y Contact C139 ..C159	.	3.18	Ang.
		1-x,1-y,1-z =	2_666	Check
PLAT720_ALERT_4_G	Number of Unusual/Non-Standard Labels		32	Note
	Ha H1AA H1AB H1AC H1BA H1BB H1CA H1CB			
	H1DA H1DB H1DC H1EA H1EB H1EC H1FA H1FB			
	H1GA H1GB H1HA H1HB H1HC Hb H1IA H1IB			
	H1IC H1JA H1JB H1KA H1KB H1LA H1LB H1LC			
PLAT789_ALERT_4_G	Atoms with Negative _atom_site_disorder_group #		45	Check
PLAT822_ALERT_4_G	CIF-embedded .res Contains Negative PART Numbers		3	Check

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PLAT860_ALERT_3_G Number of Least-Squares Restraints ..... 365 Note
PLAT933_ALERT_2_G Number of HKL-OMIT Records in Embedded .res File 40 Note
-2 -5 1, -2 0 1, -2 0 2, -2 1 1, -1 -1 2, -1 0 2,
-1 2 1, -1 2 2, -1 3 1, -1 3 2, 0 -3 3, 0 -2 1,
0 -1 4, 0 1 4, 0 3 4, 1 1 4, 1 2 3, 1 2 8,
1 3 0, 2 0 1, 2 3 2, 2 3 4, 2 4 3, 2 5 1,
3 2 0, 3 5 7, 1 0 0, -1 1 0, 0 1 0, 1 1 0,
0 2 0, 0 -1 1, 1 -1 -1, -1 0 1, 0 0 1, 1 0 1,
0 1 1, 1 1 1, 0 0 2, 0 1 2,

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0 ALERT level A = Most likely a serious problem - resolve or explain
0 ALERT level B = A potentially serious problem, consider carefully
21 ALERT level C = Check. Ensure it is not caused by an omission or oversight
84 ALERT level G = General information/check it is not something unexpected

6 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
25 ALERT type 2 Indicator that the structure model may be wrong or deficient
18 ALERT type 3 Indicator that the structure quality may be low
56 ALERT type 4 Improvement, methodology, query or suggestion
0 ALERT type 5 Informative message, check

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It is advisable to attempt to resolve as many as possible of the alerts in all categories. Often the minor alerts point to easily fixed oversights, errors and omissions in your CIF or refinement strategy, so attention to these fine details can be worthwhile. In order to resolve some of the more serious problems it may be necessary to carry out additional measurements or structure refinements. However, the purpose of your study may justify the reported deviations and the more serious of these should normally be commented upon in the discussion or experimental section of a paper or in the "special_details" fields of the CIF. checkCIF was carefully designed to identify outliers and unusual parameters, but every test has its limitations and alerts that are not important in a particular case may appear. Conversely, the absence of alerts does not guarantee there are no aspects of the results needing attention. It is up to the individual to critically assess their own results and, if necessary, seek expert advice.

Publication of your CIF in IUCr journals

A basic structural check has been run on your CIF. These basic checks will be run on all CIFs submitted for publication in IUCr journals (*Acta Crystallographica*, *Journal of Applied Crystallography*, *Journal of Synchrotron Radiation*); however, if you intend to submit to *Acta Crystallographica Section C* or *E* or *IUCrData*, you should make sure that full publication checks are run on the final version of your CIF prior to submission.

Publication of your CIF in other journals

Please refer to the *Notes for Authors* of the relevant journal for any special instructions relating to CIF submission.

PLATON version of 14/11/2023; check.def file version of 14/09/2023

Datablock 2 - ellipsoid plot

