

Supplementary Information

In-situ Recrystallization of Zero-Dimensional Hybrid Metal Halide Glass-Ceramics Toward Improved Scintillation Performance

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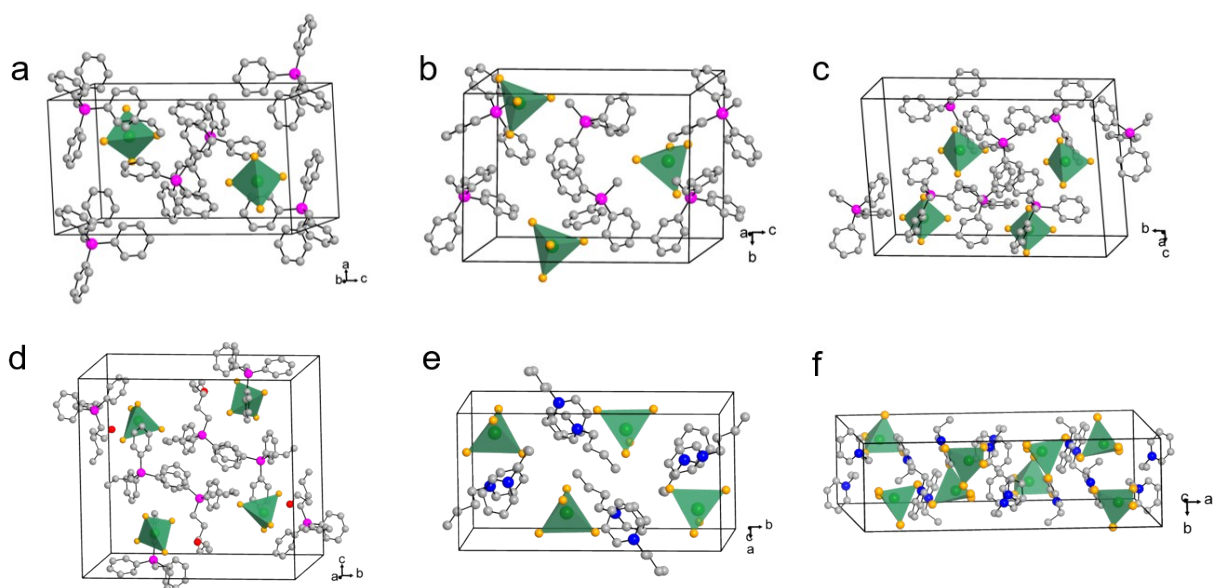
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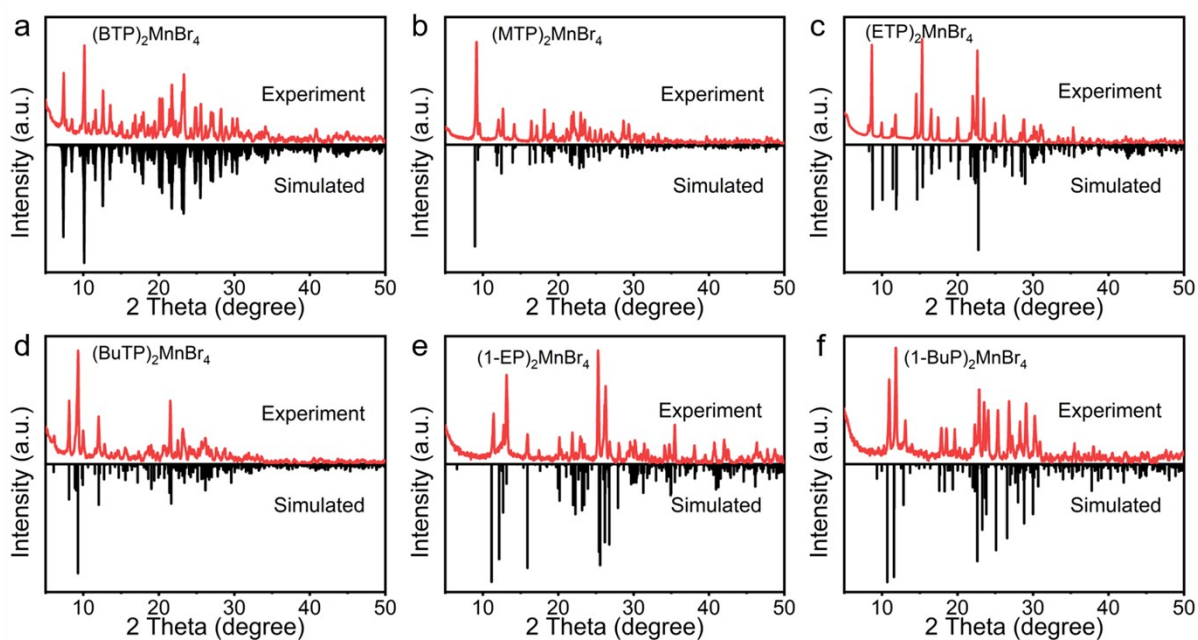
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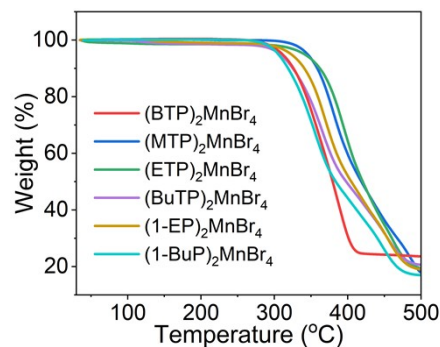
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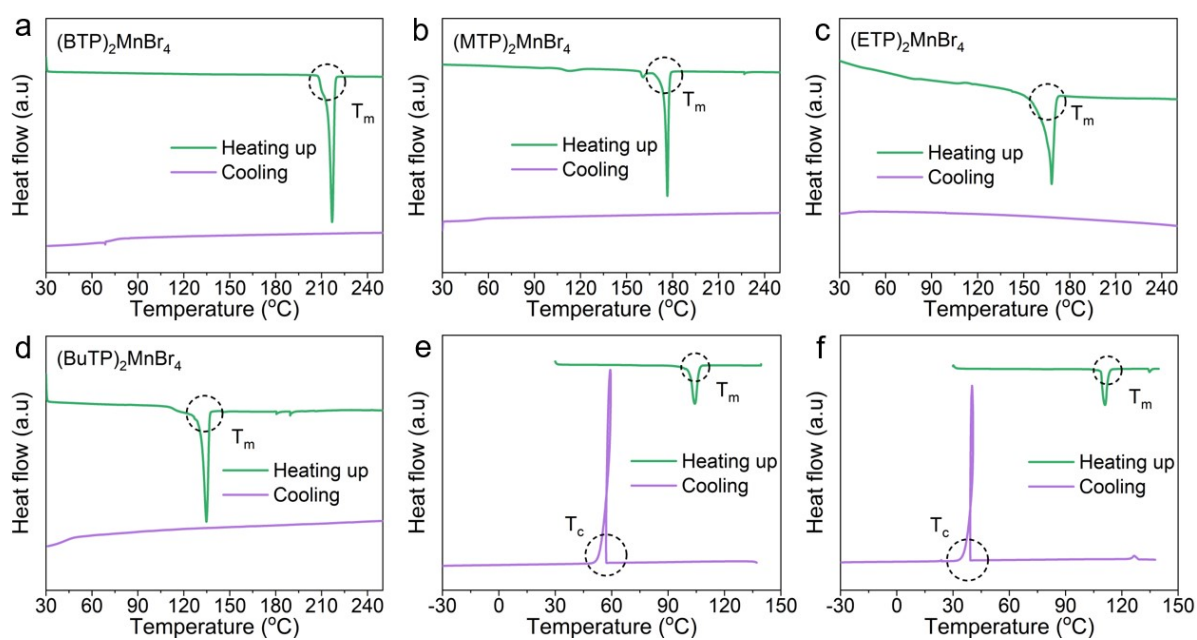
Supplementary Figure 1 Crystal structure of $(\text{BTP})_2\text{MnBr}_4$ (a), $(\text{MTP})_2\text{MnBr}_4$ (b), $(\text{ETP})_2\text{MnBr}_4$ (c), $(\text{BuTP})_2\text{MnBr}_4$ (d), $(1\text{-EP})_2\text{MnBr}_4$ (e), and $(1\text{-BuP})_2\text{MnBr}_4$ (f).



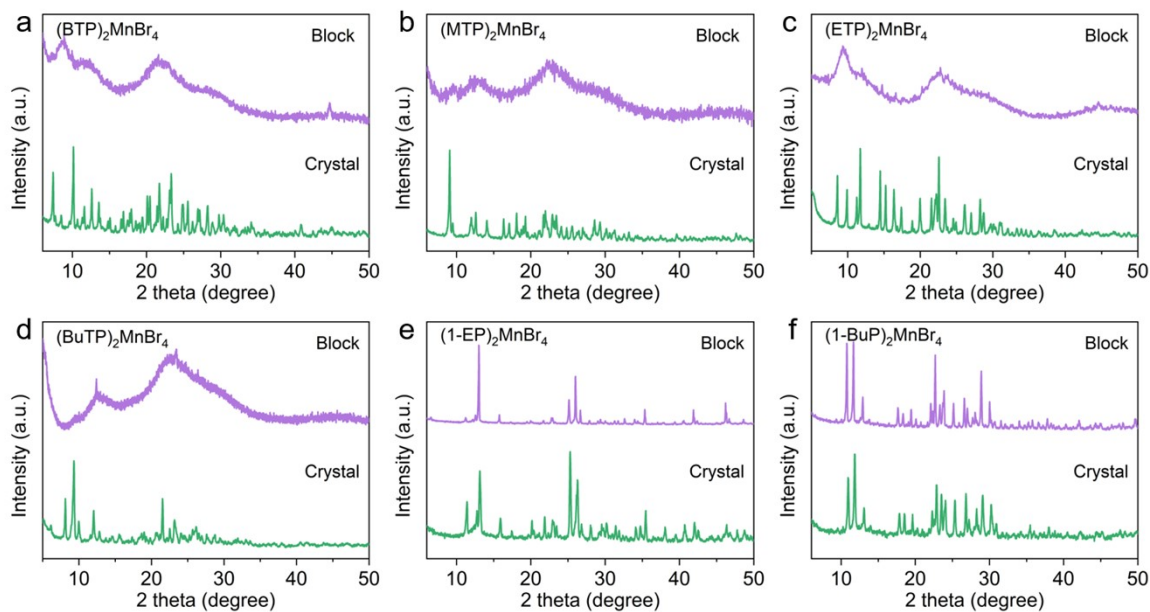
Supplementary Figure 2 Simulated and experimental PXRD patterns for $(\text{BTP})_2\text{MnBr}_4$ (a), $(\text{MTP})_2\text{MnBr}_4$ (b), $(\text{ETP})_2\text{MnBr}_4$ (c) $(\text{BuTP})_2\text{MnBr}_4$ (d), $(1\text{-EP})_2\text{MnBr}_4$ (e), and $(1\text{-BuP})_2\text{MnBr}_4$ (f) crystal.



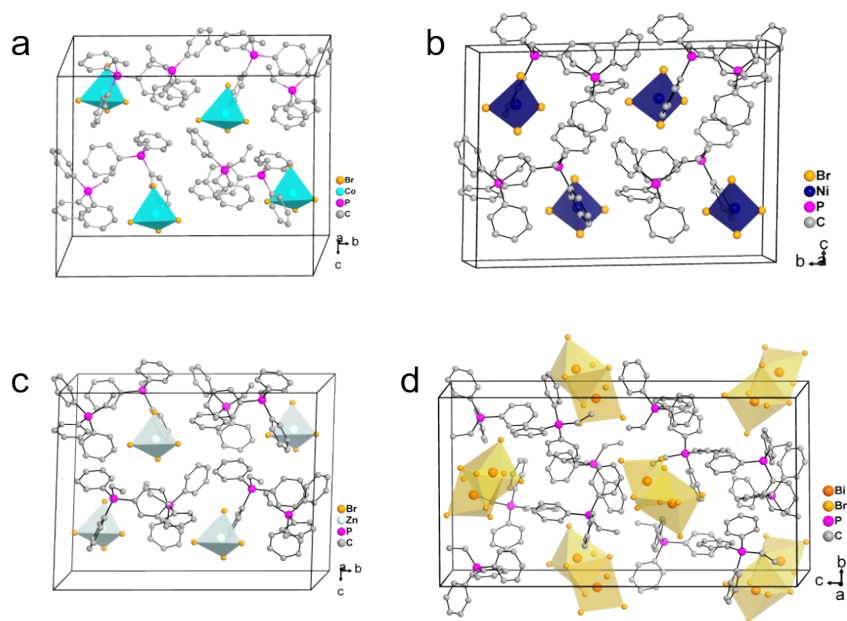
Supplementary Figure 3 TGA curve of $(\text{BTP})_2\text{MnBr}_4$, $(\text{MTP})_2\text{MnBr}_4$, $(\text{ETP})_2\text{MnBr}_4$, $(\text{BuTP})_2\text{MnBr}_4$, $(1\text{-EP})_2\text{MnBr}_4$, $(1\text{-BuP})_2\text{MnBr}_4$ crystal.



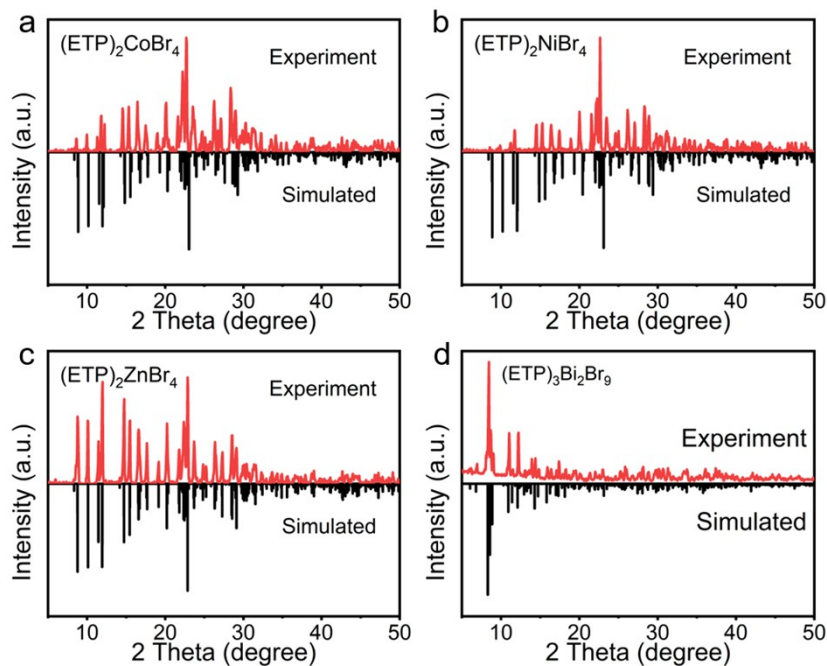
Supplementary Figure 4 Heating up and cooling scans for $(\text{BTP})_2\text{MnBr}_4$ (a), $(\text{MTP})_2\text{MnBr}_4$ (b), $(\text{ETP})_2\text{MnBr}_4$ (c), $(\text{BuTP})_2\text{MnBr}_4$ (d), $(1\text{-EP})_2\text{MnBr}_4$ (e), $(1\text{-BuP})_2\text{MnBr}_4$ (f) crystals.



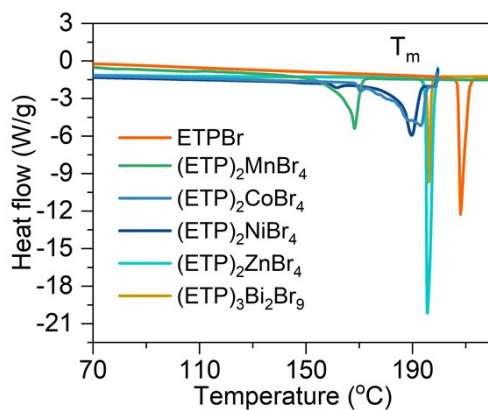
Supplementary Figure 5. PXRD patterns for $(\text{BTP})_2\text{MnBr}_4$ (a), $(\text{MTP})_2\text{MnBr}_4$ (b), $(\text{ETP})_2\text{MnBr}_4$ (c), $(\text{BuTP})_2\text{MnBr}_4$ (d), $(1\text{-EP})_2\text{MnBr}_4$ (e), $(1\text{-BuP})_2\text{MnBr}_4$ (f) crystal and block.



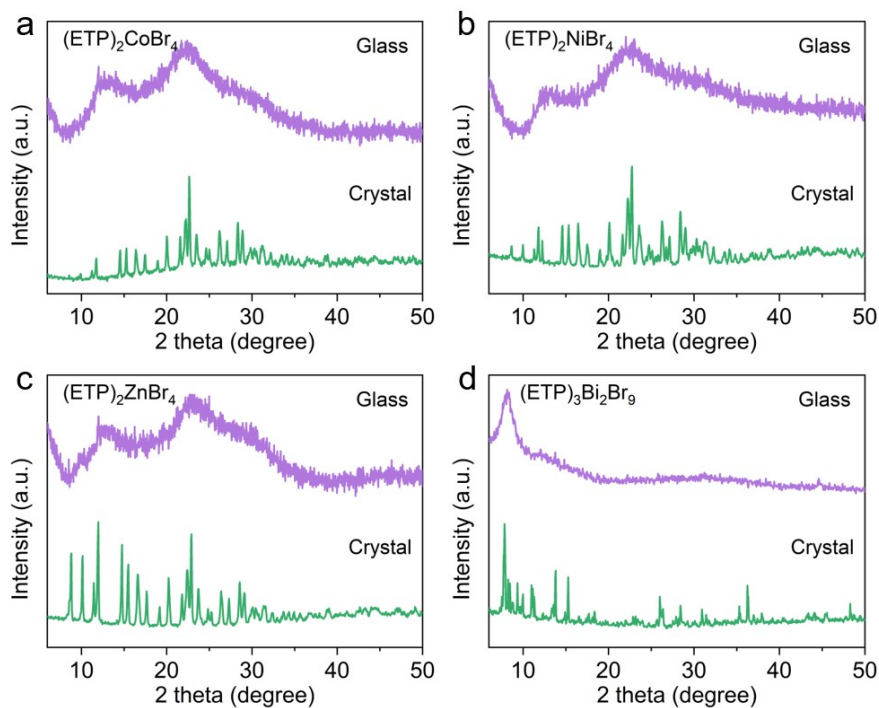
Supplementary Figure 6 Crystal structure of $(\text{ETP})_2\text{CoBr}_4$ (a), $(\text{ETP})_2\text{NiBr}_4$ (b), $(\text{ETP})_2\text{ZnBr}_4$ (c) and $(\text{ETP})_3\text{Bi}_2\text{Br}_9$ (d).



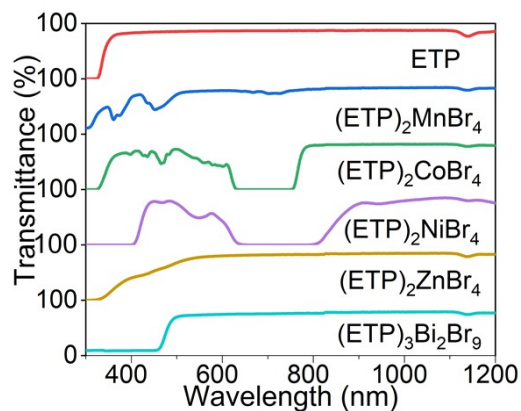
Supplementary Figure 7 Simulated and experimental PXRD patterns for $(\text{ETP})_2\text{CoBr}_4$ (a), $(\text{ETP})_2\text{NiBr}_4$ (b), $(\text{ETP})_2\text{ZnBr}_4$ (c) and $(\text{ETP})_3\text{Bi}_2\text{Br}_9$ (d) crystal.



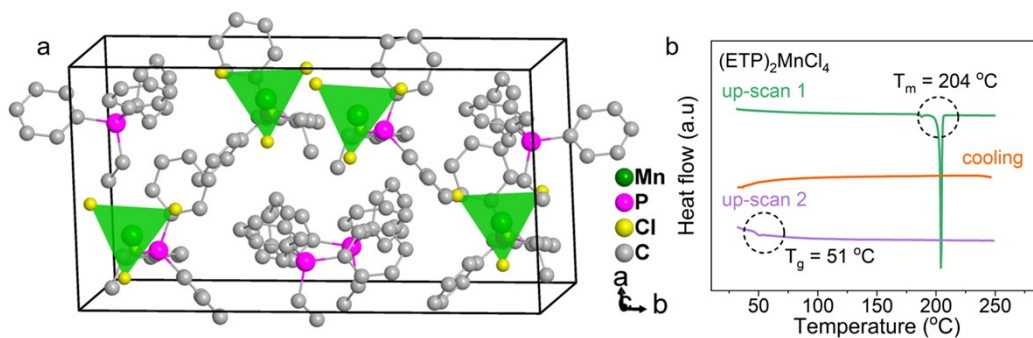
Supplementary Figure 8 DSC up-scans for ETPBr, $(\text{ETP})_2\text{MnBr}_4$, $(\text{ETP})_2\text{CoBr}_4$, $(\text{ETP})_2\text{NiBr}_4$, $(\text{ETP})_2\text{ZnBr}_4$, $(\text{ETP})_3\text{Bi}_2\text{Br}_9$ crystals.



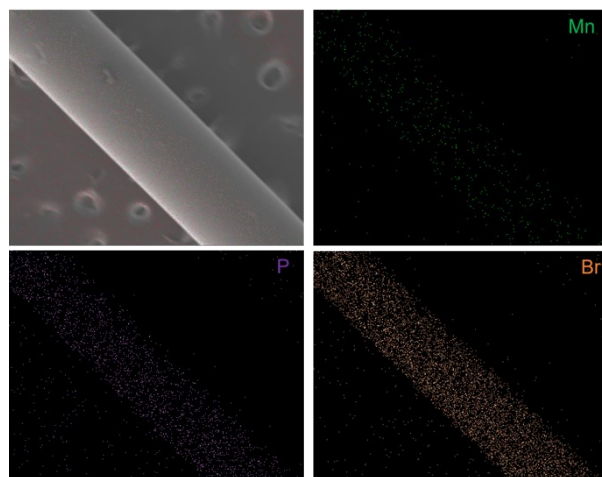
Supplementary Figure 9 PXRD patterns for $(\text{ETP})_2\text{CoBr}_4$ (a), $(\text{ETP})_2\text{NiBr}_4$ (b), $(\text{ETP})_2\text{ZnBr}_4$ (c), $(\text{ETP})_3\text{Bi}_2\text{Br}_9$ (d) crystal and glass.



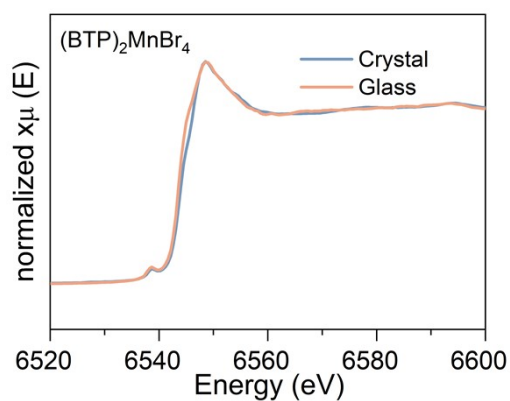
Supplementary Figure 10 Light transmittance of ETP, $(\text{ETP})_2\text{MnBr}_4$, $(\text{ETP})_2\text{CoBr}_4$, $(\text{ETP})_2\text{NiBr}_4$, $(\text{ETP})_2\text{ZnBr}_4$, $(\text{ETP})_3\text{Bi}_2\text{Br}_9$ glass.



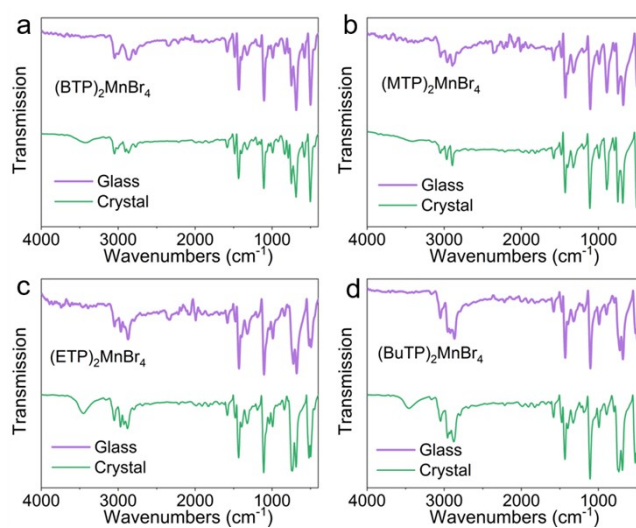
Supplementary Figure 11 Crystal structure (a) and DSC curve (b) for $(\text{ETP})_2\text{MnCl}_4$.



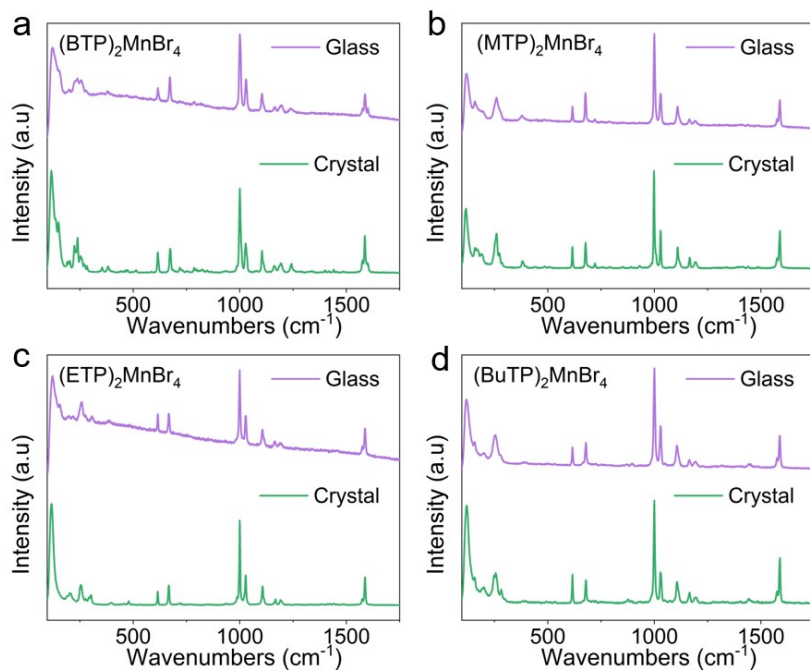
Supplementary Figure 12 SEM image of $(\text{BTP})_2\text{MnBr}_4$ optical fiber and elemental mapping of P, Mn and Br.



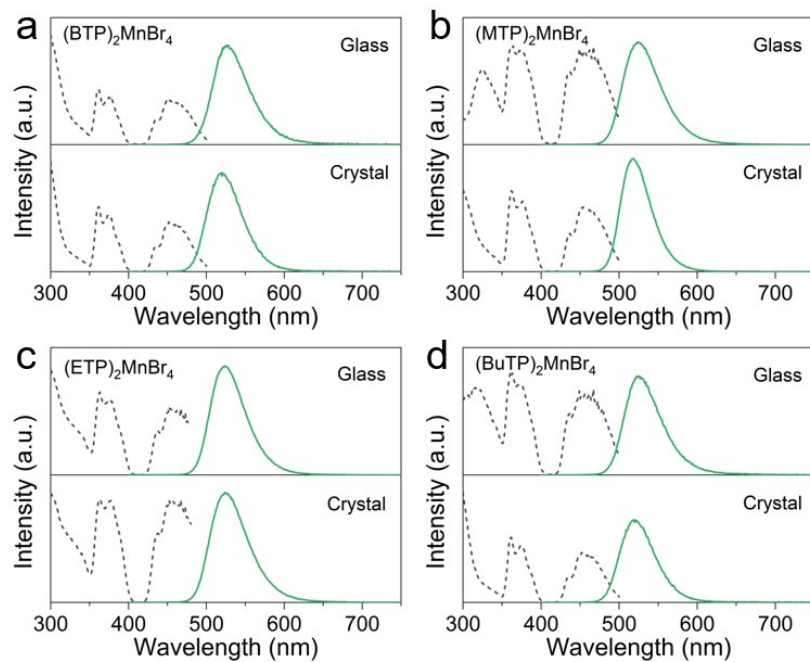
Supplementary Figure 13 Normalized XANES profiles recorded at the Mn K-edges of the $(\text{BTP})_2\text{MnBr}_4$ crystal and glass.



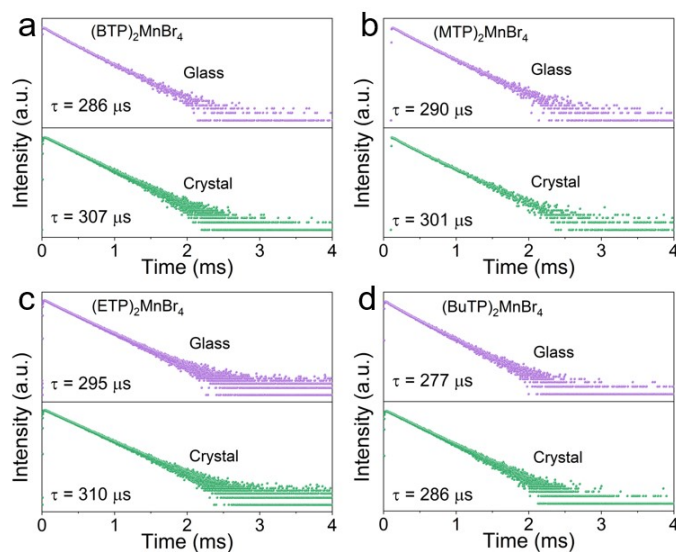
Supplementary Figure 14 FTIR spectra of $(\text{BTP})_2\text{MnBr}_4$ (a), $(\text{MTP})_2\text{MnBr}_4$ (b), $(\text{ETP})_2\text{MnBr}_4$ (c), $(\text{BuTP})_2\text{MnBr}_4$ (d) crystal and glass.



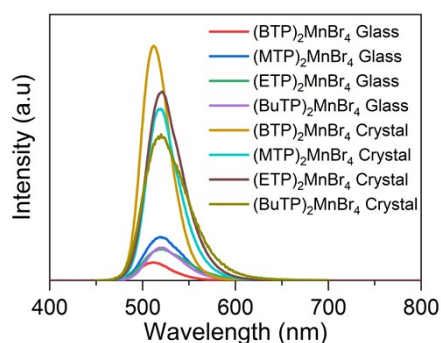
Supplementary Figure 15 Raman spectra of $(\text{BTP})_2\text{MnBr}_4$ (a), $(\text{MTP})_2\text{MnBr}_4$ (b), $(\text{ETP})_2\text{MnBr}_4$ (c), $(\text{BuTP})_2\text{MnBr}_4$ (d) crystal and glass.



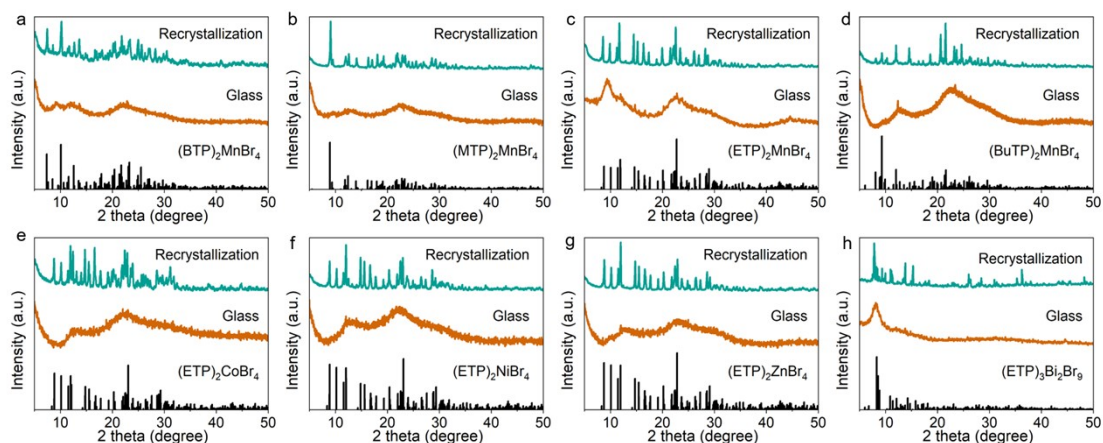
Supplementary Figure 16 PLE and PL spectra of $(\text{BTP})_2\text{MnBr}_4$ (a), $(\text{MTP})_2\text{MnBr}_4$ (b), $(\text{ETP})_2\text{MnBr}_4$ (c), $(\text{BuTP})_2\text{MnBr}_4$ (d) crystal and glass.



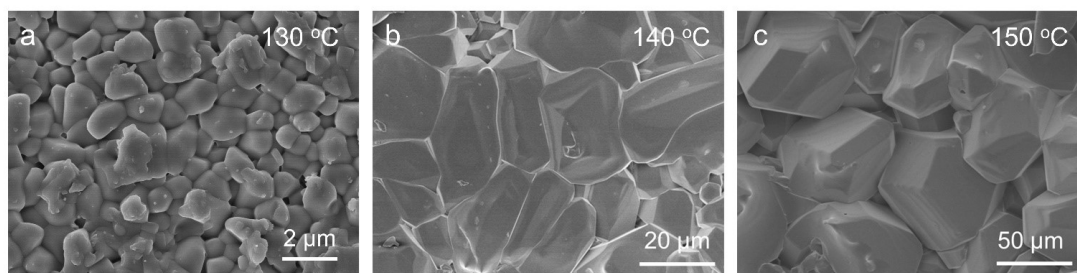
Supplementary Figure 17 Luminescence decay curve of $(\text{BTP})_2\text{MnBr}_4$ (a), $(\text{MTP})_2\text{MnBr}_4$ (b), $(\text{ETP})_2\text{MnBr}_4$ (c), $(\text{BuTP})_2\text{MnBr}_4$ (d) crystal and glass.



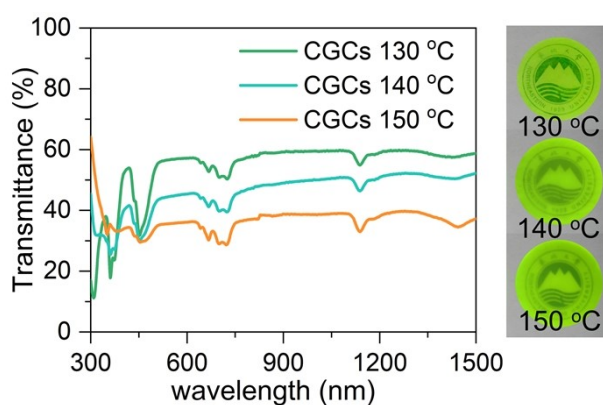
Supplementary Figure 18 Radioluminescence spectra of $(\text{BTP})_2\text{MnBr}_4$, $(\text{MTP})_2\text{MnBr}_4$, $(\text{ETP})_2\text{MnBr}_4$, $(\text{BuTP})_2\text{MnBr}_4$ crystal and glass.



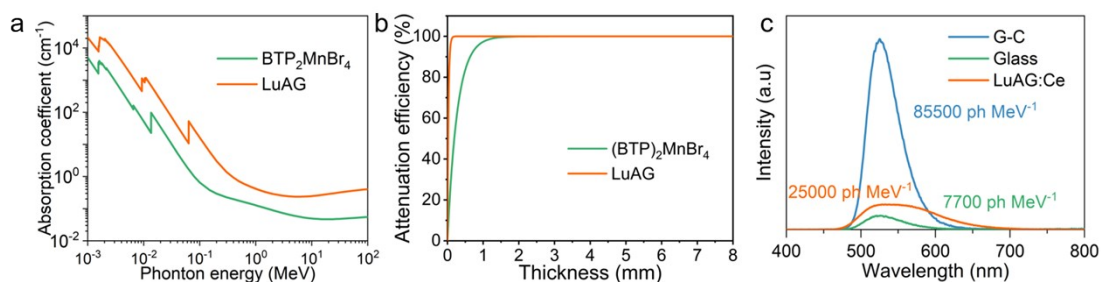
Supplementary Figure 19 XRD pattern of $(\text{BTP})_2\text{MnBr}_4$ (a), $(\text{MTP})_2\text{MnBr}_4$ (b), $(\text{ETP})_2\text{MnBr}_4$ (c), $(\text{BuTP})_2\text{MnBr}_4$ (d), $(\text{ETP})_2\text{CoBr}_4$ (e), $(\text{ETP})_2\text{NiBr}_4$ (f), $(\text{ETP})_2\text{ZnBr}_4$ (g) and $(\text{ETP})_3\text{Bi}_2\text{Br}_9$ (h) comparison of glass recrystallized with glass.



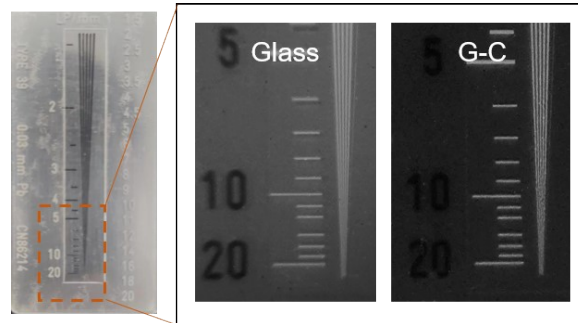
Supplementary Figure 20 SEM images of (BTP)₂MnBr₄ G-Cs crystallized at 130 °C (a), 140 °C (b), 150 °C (c).



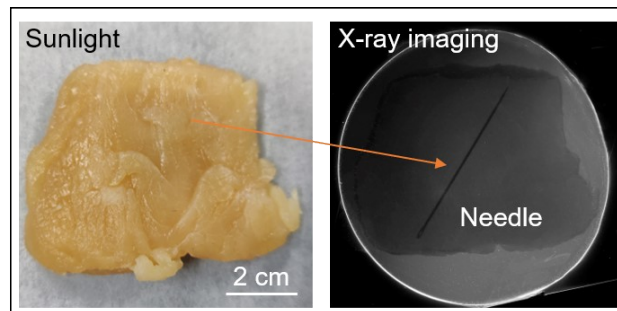
Supplementary Figure 21 Light transmittance of (BTP)₂MnBr₄ G-Cs crystallized at 130 °C, 140 °C, 150 °C.



Supplementary Figure 22 (a) Absorption coefficients of LuAG:Ce and (BTP)₂MnBr₄ as a function of photon energy from 1 keV to 100 MeV. (b) Attenuation efficiency of LuAG:Ce and (BTP)₂MnBr₄ as a function of photon energy 17.5 KeV. (c) X-ray radio luminescence (RL) spectra of (BTP)₂MnBr₄ glass, (BTP)₂MnBr₄ G-C and LAG:Ce under X-ray irradiation.



Supplementary Figure 23 X-ray image of the standard X-ray resolution pattern plate.



Supplementary Figure 24 Daylight photograph and X-ray image of chicken with needle inserted.

Supplementary Table 1. The crystal structure parameters of $(C_{25}H_{22}P)_2MnBr_4$, $(C_{19}H_{18}P)_2MnBr_4$, $(C_{20}H_{20}P)_2MnBr_4$ and $(C_{22}H_{24}P)_2MnBr_4 \cdot C_2H_5OH$.

Formula	$C_{50}H_{44}Br_4MnP_2$ (CCDC 2285876)	$C_{38}H_{36}Br_4MnP_2$ (CCDC 2285882)	$C_{40}H_{40}Br_4MnP_2$ (CCDC 2169124)	$C_{46}H_{54}Br_4MnOP_2$ (CCDC 2285877)
Dimension (mm)	0.2 × 0.18 × 0.16	0.2 × 0.18 × 0.16	0.2 × 0.18 × 0.16	0.2 × 0.18 × 0.16
Molecular weight	1081.37	929.19	957.24	1059.41
Temperature (K)	301	273	150	293
Crystal system	Triclinic	monoclinic	monoclinic	monoclinic
Space group	<i>P</i> -1	<i>P</i> 2 ₁	<i>C</i> 1 <i>c</i> 1	<i>P</i> 2 ₁
<i>Z</i>	2	2	4	4
<i>a</i> (Å)	10.4567 (3)	9.7924 (2)	12.2698 (7)	10.8592 (2)
<i>b</i> (Å)	12.5204 (3)	12.5306 (2)	21.2817 (11)	21.7967 (5)
<i>c</i> (Å)	18.4226 (5)	16.6305 (3)	16.3853 (11)	19.6633 (5)
α (°)	105.994 (2)	90	90	90
β (°)	92.776 (2)	105.085 (2)	110.717 (2)	96.743 (1)
γ (°)	92.767 (2)	90	90	90
<i>V</i> (Å ³)	2311.07 (11)	1970.32 (6)	4001.9 (4)	4622.01 (18)
ρ_{calc} (g/cm ³)	1.554	1.566	1.589	1.522
μ (mm ⁻¹)	7.277	8.425	4.430	3.846
Reflections measured	23657	12550	34259	44228
Reflections independent	8929	5843	8146	10611
<i>R</i> _{int}	0.0219	0.0198	0.0624	0.0483
<i>R</i> ₁ [<i>I</i> > 2σ(<i>I</i>)] ^a	0.0300	0.0249	0.0485	0.0366
<i>wR</i> (<i>F</i> ²)·[<i>I</i> > 2σ(<i>I</i>)] ^b	0.0746	0.0655	0.1151	0.0727
<i>Goof</i>	1.051	1.058	1.071	1.017

$$[a] R_1 = \sum \| F_o | - | F_c \| / \sum | F_o |, [b] wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$$

Supplementary Table 2. The crystal structure parameters of $(C_{20}H_{20}P)_2CoBr_4$, $(C_{20}H_{20}P)_2NiBr_4$, $(C_{20}H_{20}P)_2ZnBr_4$ and $(C_{20}H_{20}P)_3Bi_2Br_9$.

Formula moiety	$C_{40}H_{40}Br_4CoP_2$ (CCDC-2285878)	$C_{40}H_{40}Br_4NiP_2$ (CCDC 2285879)	$C_{40}H_{40}Br_4ZnP_2$ (CCDC 2285880)	$C_{60}H_{60}Bi_2Br_9P_3$ (CCDC 2285881)
Dimension (mm)	0.2 × 0.18 × 0.16	0.2 × 0.18 × 0.16	0.2 × 0.18 × 0.16	0.2 × 0.18 × 0.16
Molecular weight	961.23	961.01	967.67	2011.14
Temperature (K)	293	293	300	300
Crystal system	Monoclinic	Monoclinic	Monoclinic	Orthorhombic
Space group	<i>C1c1</i>	<i>C1c1</i>	<i>Cc</i>	<i>P2₁2₁2₁</i>
<i>Z</i>	4	4	4	4
<i>a</i> (Å)	12.2214 (5)	12.1800 (5)	12.3305 (2)	14.5071 (10)
<i>b</i> (Å)	21.1041 (5)	21.0104 (5)	21.1916 (3)	15.5088 (10)
<i>c</i> (Å)	16.4141 (7)	16.4238 (7)	16.6153 (3)	29.2071 (3)
α (°)	90	90	90	90
β (°)	111.1290 (10)	111.4920 (10)	111.612 (2)	90
γ (°)	90	90	90	90
<i>V</i> (Å ³)	3948.9 (3)	3910.7 (3)	4036.41 (12)	6571.24 (9)
ρ_{calc} (g/cm ³)	1.617	1.632	1.592	2.033
μ (mm ⁻¹)	4.590	4.692	6.429	17.767
Reflections measured	19693	18331	18709	30168
Reflections independent	7800	7454	5958	12478
<i>R</i> _{int}	0.0384	0.0337	0.0177	0.0247
<i>R</i> ₁ [<i>I</i> > 2σ(<i>I</i>)] ^a	0.0573	0.0421	0.0375	0.0337
<i>wR</i> (<i>F</i> ²)·[<i>I</i> > 2σ(<i>I</i>)] ^b	0.1521	0.0968	0.0982	0.0885
<i>Goof</i>	1.030	1.027	1.094	1.069

$$[a] R_1 = \sum \| F_o | - | F_c \| / \sum | F_o |, [b] wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$$

Supplementary Table 3. The crystal structure parameters of (ETP)₂MnCl₄.

Formula	C ₄₀ H ₄₀ Cl ₄ MnP ₂ (CCDC 2285898)
Dimension (mm)	0.2 × 0.2 × 0.2
Molecular weight	779.40
Temperature (K)	293
Crystal system	Monoclinic
Space group	C1c1
Z	4
a (Å)	12.1993 (2)
b (Å)	20.8825 (3)
c (Å)	16.3635 (3)
α (°)	90
β (°)	110.775 (2)
γ (°)	90
V (Å ³)	3897.59 (12)
ρ _{calc} (g/cm ³)	1.328
μ (mm ⁻¹)	6.254
Reflections measured	11188
Reflections independent	4819
R _{int}	0.0219
R ₁ [I > 2σ(I)] ^a	0.0303
wR(F ²)·[I > 2σ(I)] ^b	0.0833
Goof	1.082

^[a] $R_1 = \sum \| |F_o| - |F_c| \| / \sum |F_o|$, ^[b] $wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$

Supplementary Table 4. Summary of EXAFS fitting parameters for $(\text{BTP})_2\text{MnBr}_4$ crystal and glass.

Sample	Bond	Coordination No.	R
Crystal	Mn-Br	3	2.63 Å
	Mn-Br	1	2.56 Å
Glass	Mn-Br	3	2.53 Å
	Mn-Br	1	2.63 Å

Supplementary Table 5. PLQY of (BTP)₂MnBr₄, (MTP)₂MnBr₄, (ETP)₂MnBr₄, (BuTP)₂MnBr₄ crystal and glass.

Sample	(BTP) ₂ MnBr ₄	(MTP) ₂ MnBr ₄	(ETP) ₂ MnBr ₄	(BuTP) ₂ MnBr ₄
Crystal	98%	97%	96%	92%
Glass	48%	57%	57%	50%