

## Supporting Information

### p-n Junction formation between CoPi and $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> layers enhances photo-charge separation and catalytic efficiencies for efficient visible light-driven water oxidation

Tomohiro Katsuki,<sup>†</sup> Zaki N. Zahran,<sup>†, ‡\*</sup> Yuta Tsubonouchi,<sup>†</sup> Debraj Chandra,<sup>†</sup> Norihisa Hoshino<sup>†</sup> and Masayuki Yagi <sup>†\*</sup>.

<sup>†</sup>Department of Materials Science and Technology, Faculty of Engineering, Niigata University, 8050 Ikarashi-2, Niigata 9050-2181, Japan.

<sup>‡</sup>Faculty of Science, Tanta University, Tanta 5111, Egypt.

\*Correspondence to: yagi@eng.niigata-u.ac.jp and znzahran@eng.niigata-u.ac.jp

## Contents



**Table S1** Summary of PEC performances of the state-of-the-art  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>/CoPi-based electrodes.

**Figure S1** Nyquist plots and fitting curves for (a)  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> and (b)  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub>/CoPi electrodes.

**Table S2** Summary of Fitting parameter provided by PEIS analysis.

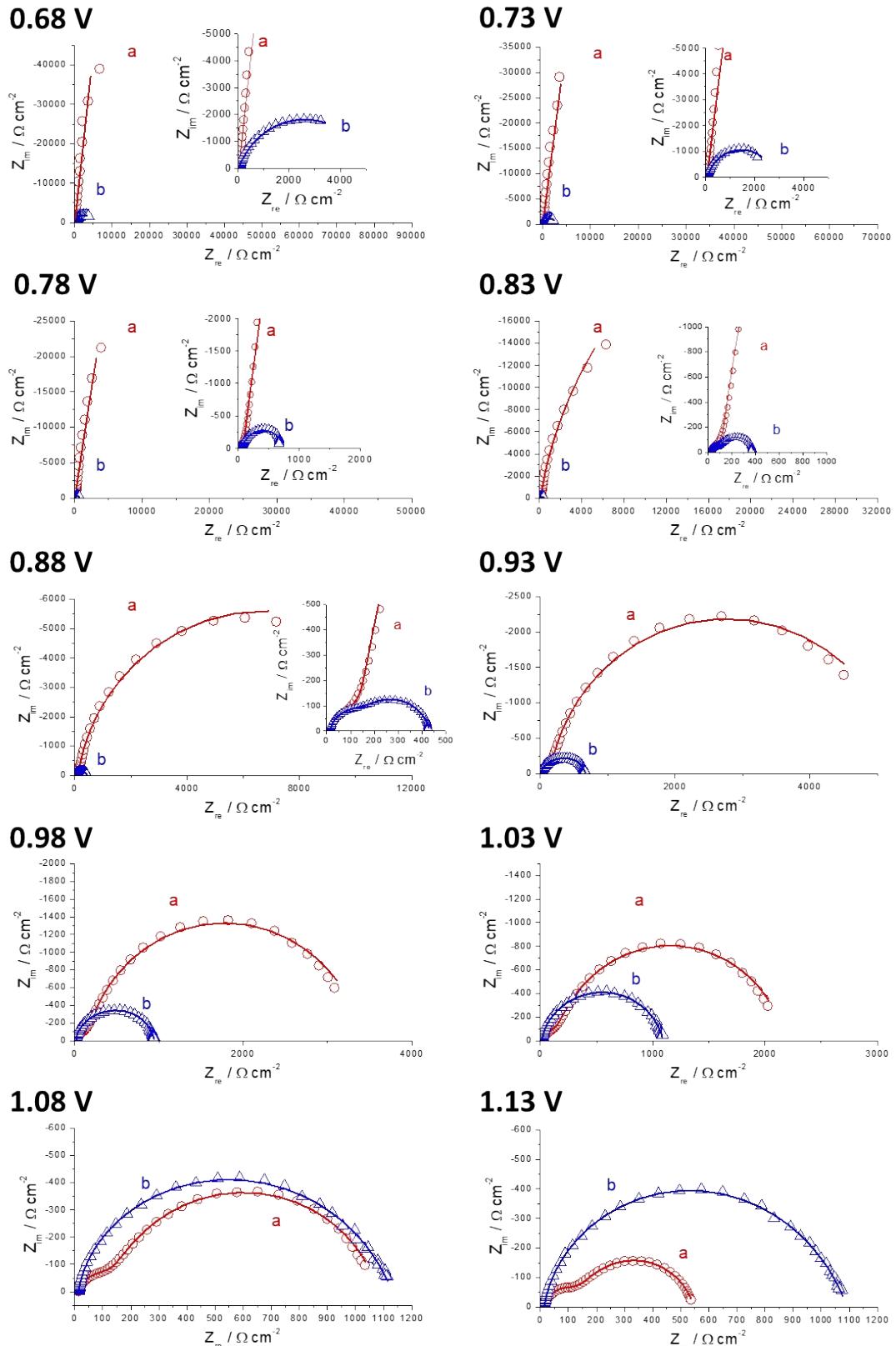
**Table S1** Summary of PEC water oxidation performances of the state-of-the-art  $\alpha\text{-Fe}_2\text{O}_3/\text{CoPi}$ -based electrodes.<sup>a)</sup>

Electrodes <sup>b)</sup>	Preparation Method	$E_{\text{on}}$ (V vs. RHE)	$E_{\text{on}}$ shift <sup>c)</sup> (mV)	$\eta_{\text{sep}}$ (%) <sup>d)</sup>	$\eta_{\text{cat}}$ (%) <sup>d)</sup>	$\text{IPCE}_{420}$ (%) <sup>d)</sup>	$\text{ABPE}_{\text{max}}$ (%)	Ref.
$\alpha\text{-Fe}_2\text{O}_3/\text{CoPi}$	Casting	0.65 <sup>e)</sup>	200 <sup>e)</sup>	33 <sup>f)</sup>	70 <sup>f)</sup>	20	1.2 <sup>f)</sup>	This work
$\text{Ga}_2\text{O}_3/\alpha\text{-Fe}_2\text{O}_3/\text{CoPi}$	ALD	0.60 <sup>g)</sup>	100 <sup>g)</sup>	na	na	na	na	R1
$\alpha\text{-Fe}_2\text{O}_3\text{-CF}/\text{CoPi}$	APCVD	0.83	170	na	na	40	na	R2
$\alpha\text{-Fe}_2\text{O}_3\text{-CF}/\text{CoPi}$	APCVD	0.80	100	na	na	8	na	R3
$\alpha\text{-Fe}_2\text{O}_3\text{-IO}/\text{CoPi}$	ED	0.88	120	na	na	30 <sup>h)</sup>	na	R4
$\alpha\text{-Fe}_2\text{O}_3\text{-NRs/N-CoPi}$	HT	0.70	50	na	na	7	na	R5
$\alpha\text{-Fe}_2\text{O}_3\text{-CL}/\text{CoPi}$	HT	0.80	110	na	na	na	na	R6
$\alpha\text{-Fe}_2\text{O}_3\text{-NRs/CoPi}$	HT	0.81	110	na	na	12	0.12	R7
$\alpha\text{-Fe}_2\text{O}_3\text{-CL}/\text{CoPi-NaBH}_4$	HT	0.70	100	na	na	na	0.13	R6
$\alpha\text{-Fe}_2\text{O}_3\text{-PNs/CoPi}$	HT	0.68 <sup>i)</sup>	160 <sup>i)</sup>	17 <sup>i)</sup>	90 <sup>i)</sup>	na	na	R8
$\alpha\text{-Fe}_2\text{O}_3\text{-NRs/CoPi}$	HT	0.80	~200	na	80	20	0.21	R9
Co-Pi/CaFe <sub>2</sub> O <sub>3</sub> / $\alpha\text{-Fe}_2\text{O}_3$ BNRs	APCVD	0.64	80	na	na	na	na	R10

<sup>a)</sup> Measured in a 1.0 M NaOH (pH 14.0) electrolyte solution, under 1 sun AM 1.5G, ALD: atomic layer deposition, na: not available, CF: cauliflower, APCVD: atmospheric pressure chemical vapor deposition, IO: inverse opal, ED: electrochemical deposition, NRs: nanorods, N-CoPi: nitrogen doped CoPi, HT: hydrothermal, CL: coralline, PNs: porous nanorods, BNRs: branched nanorods, <sup>b)</sup> coated on basal FTO electrodes, <sup>c)</sup> defined as the  $E_{\text{on}}$  difference between the  $\alpha\text{-Fe}_2\text{O}_3$  and  $\alpha\text{-Fe}_2\text{O}_3/\text{CoPi}$  electrodes, <sup>d)</sup> at 1.23 V vs. RHE, <sup>e)</sup> Xenon lamp ( $> 420$  nm, 100 mW cm<sup>-2</sup>), <sup>f)</sup> monoclinic light (420 nm, 7.06 mW cm<sup>-2</sup>) from LED, <sup>g)</sup> 1.0 M KOH with HCl as supporting, <sup>h)</sup> at 0.5 V vs. Ag/AgCl (1.53 V vs RHE), <sup>i)</sup> 0.1 M KOH (pH 12.6)

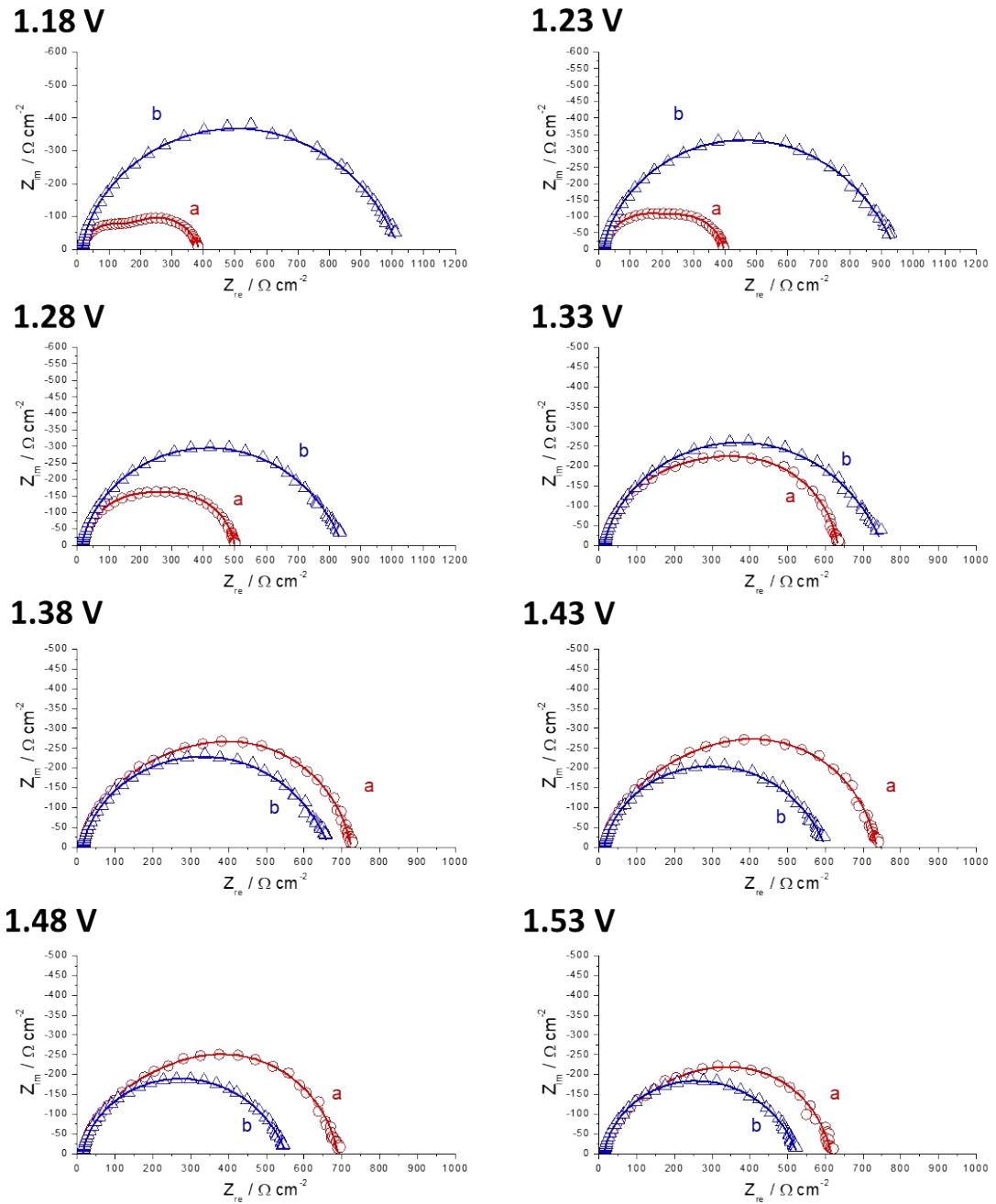
**References:**

- R1 O. Zandi and T. W. Hamann, *Journal of Physical Chemistry Letters*, 2014, **5**, 1522–1526.
- R2 D. K. Zhong, M. Cornuz, K. Sivula, M. Grätzel and D. R. Gamelin, *Energy Environ Sci*, 2011, **4**, 1759–1764.
- R3 D. K. Zhong, J. Sun, H. Inumaru and D. R. Gamelin, *J Am Chem Soc*, 2009, **131**, 6086–6087.
- R4 X. Shi, K. Zhang and J. H. Park, *Int J Hydrogen Energy*, 2013, **38**, 12725–12732.
- R5 H. Lee, K. H. Kim, W. H. Choi, B. C. Moon, H. J. Kong and J. K. Kang, *ACS Appl Mater Interfaces*, 2019, **11**, 44366–44374.
- R6 J. Xiao, L. Fan, F. Zhao, Z. Huang, S. F. Zhou and G. Zhan, *J Catal*, 2020, **381**, 139–149.
- R7 J. Xiao, L. Fan, Z. Huang, J. Zhong, F. Zhao, K. Xu, S. F. Zhou and G. Zhan, *Chinese Journal of Catalysis*, 2020, **41**, 1761–1771.
- R8 G. Liu, N. Li, Y. Zhao, M. Wang, R. Yao, F. Zhao, Y. Wu and J. Li, *ACS Sustain Chem Eng*, 2019, **7**, 11377–11385.
- R9 T. T. Li, Q. Zhou, J. Qian, Y. Hu and Y. Q. Zheng, *Electrochim Acta*, 2019, **307**, 92–99.
- R10 D. Chen, Z. Liu, Z. Guo, M. Ruan and W. Yan, *ChemSusChem*, 2019, **12**, 3286–3295.



**Figure S1.** Nyquist plots and fitting curves using an equivalent circuit (inset of Figure 7A) for (a)  $\alpha\text{-Fe}_2\text{O}_3$  and (b)  $\alpha\text{-Fe}_2\text{O}_3/\text{CoPi}$  electrodes in a 1.0 M NaOH solution (pH 14.0) at frequencies from 100 mHz to 20 kHz at different potentials ( $E$ ) of 0.68–1.53 V vs RHE (-0.35 – 0.50 V vs Ag/AgCl) under visible light irradiation (>420 nm, 100 mW cm<sup>-2</sup>).

Figure S2 continued



**Table S2** Fitting parameters of components of  $R_{\text{sol}}$ ,  $R_{\text{ct}}$  and CPE analyzed in the equivalent circuit (inset of Figure 7A) at different potentials ( $E$ ) for  $\alpha\text{-Fe}_2\text{O}_3$  and  $\alpha\text{-Fe}_2\text{O}_3/\text{CoPi}$  electrodes.

$\alpha\text{-Fe}_2\text{O}_3$ electrode	High frequency				Low frequency			
	$E$ vs RHE	$R_{\text{sol}}$ [ $\Omega$ ]	CPE <sup>bulk</sup>		$R_{\text{ct}}^{\text{surf}}$ [ $\Omega$ ]	CPE <sup>surf</sup>		
			$R_{\text{ct}}^{\text{bulk}}$ [ $\Omega$ ]	$T_{\text{CPE}}^{\text{bulk}}$ [ $\mu\text{F s}^{(p-1)}$ ]		$T_{\text{CPE}}^{\text{surf}}$ [ $\mu\text{F s}^{(p-1)}$ ]	$p$	
	0.68	14.5	132	27.7	0.921	$5.17 \times 10^{13}$	13.4	0.942
	0.73	14.4	193	34.7	0.888	$6.07 \times 10^9$	20.3	0.962
	0.78	14.1	201	42.5	0.851	$7.82 \times 10^6$	33.5	0.966
	0.83	13.7	221	51.0	0.817	$7.77 \times 10^4$	48.9	0.965
	0.88	13.5	250	50.2	0.809	$1.33 \times 10^4$	62.0	0.950
	0.93	13.4	298	47.7	0.808	5120	64.9	0.929
	0.98	13.3	302	46.1	0.806	3120	69.8	0.901
	1.03	13.3	223	39.7	0.816	1950	103	0.860
	1.08	13.4	161	29.7	0.843	906	180	0.825
	1.13	13.6	148	23.4	0.864	385	268	0.807
	1.18	13.6	177	21.3	0.872	195	290	0.822
	1.23	13.7	252	21.2	0.870	131	232	0.878
	1.28	13.7	368	21.2	0.867	118	144	0.947
	1.33	13.7	499	21.7	0.862	121	89.2	1.00
	1.38	13.6	564	22.1	0.857	147	58.9	1.00
	1.43	13.6	535	21.5	0.857	188	41.7	0.997
	1.48	13.6	477	21.0	0.857	199	44.8	0.962
	1.53	13.6	411	20.2	0.859	191	51.1	0.930

$\alpha\text{-Fe}_2\text{O}_3/\text{CoPi}$ electrode	High frequency				Low frequency			
	$E$ vs RHE	$R_{\text{sol}}$ [ $\Omega$ ]	CPE <sup>bulk</sup>		CPE <sup>surf</sup>		$p$	
			$R_{\text{ct}}^{\text{bulk}}$ [ $\Omega$ ]	$T_{\text{CPE}}^{\text{bulk}}$ [ $\mu\text{F s}^{(p-1)}$ ]	$R_{\text{ct}}^{\text{surf}}$ [ $\Omega$ ]	$T_{\text{CPE}}^{\text{surf}}$ [ $\mu\text{F s}^{(p-1)}$ ]		
0.68	13.6	96.4		62.3	0.846	5100	92.7	0.735
0.73	13.3	91.9		67.5	0.821	2730	153	0.831
0.78	13.4	76.1		47.7	0.851	654	225	0.863
0.83	13.5	97.8		37.0	0.875	270	279	0.851
0.88	13.6	186		30.4	0.895	232	251	0.838
0.93	13.7	385		26.2	0.909	269	187	0.829
0.98	13.8	631		24.4	0.914	301	154	0.830
1.03	13.8	740		24.0	0.914	340	153	0.760
1.08	13.8	708		24.3	0.911	409	191	0.646
1.13	13.7	676		25.7	0.905	404	193	0.634
1.18	13.7	594		27.0	0.899	417	167	0.628
1.23	13.7	511		28.8	0.893	420	167	0.603
1.28	13.7	451		30.3	0.888	380	162	0.623
1.33	13.7	390		30.8	0.887	349	168	0.629
1.38	13.6	367		31.4	0.883	287	199	0.635
1.43	13.6	356		30.5	0.884	232	248	0.638
1.48	13.6	347		28.0	0.888	192	274	0.658
1.53	13.6	353		25.0	0.896	160	313	0.662