Electronic Supplementary Material (ESI) for Sustainable Energy & Fuels. This journal is © The Royal Society of Chemistry 2023

## Supporting Information

# High Voltage Ionic Liquid Based Flexible Solid Polymer Electrolyte for high-Performance Li-Ion battery

Rushikesh B. Kale, Sahebrao S. More, Nageshwar D. Khupse, Ramchandra S. Kalubarme, Milind V. Kulkarni, Sunit B. Rane, Bharat B. Kale\*

<sup>†</sup>Centre for Materials for Electronic Technology (C-MET), Ministry of Electronics and

Information Technology (MeitY), Govt. of India, Panchawati, Off Pashan Road, Pune 411008,

India.

E-mail: <u>bbkale@cmet.gov.in</u>, <u>bbkale1@gmail.com</u>

#### **Experimental Section:**

The details of loading of cathode with C-rates are given below

**Table S1:** The details of loading of cathode with C-rates

| Cell Configuration | LFP loading | <b>0.1C</b> | <b>0.2</b> C | <b>0.5</b> C | 1C     |
|--------------------|-------------|-------------|--------------|--------------|--------|
|                    | (mg)        | (mA)        | (mA)         | (mA)         | (mA)   |
| Li/SPE-IL-II/LFP   | 3.008       | 0.05113     | 0.1022       | 0.2556       | 0.5113 |

| <b>S.</b> | Composition | Ionic                 | Temp        | Reference                                   |
|-----------|-------------|-----------------------|-------------|---|
| No        |             | Conductivity          |             |   |
|           |             | (S/cm)                |             |   |
| 1         | PEO+PTFE    | $6.62 \times 10^{-8}$ | Room        | Jokhakar, Deep A., et al. "All-solid-state  |
|           |             |                       | Temperature | Li-metal batteries: role of blending PTFE   |
|           |             |                       |             | with PEO and LiTFSI salt as a composite     |
|           |             |                       |             | electrolyte with enhanced thermal           |
|           |             |                       |             | stability." Sustainable Energy Fuels, 2020, |
|           |             |                       |             | 4, 2229-2235                                |
| 2         | PEO+HEMC    | 1.30×10-4             | Room        | Wu, Hailong, et al. "A branched cellulose-  |
|           |             |                       | Temperature | reinforced composite polymer electrolyte    |
|           |             |                       |             | with upgraded ionic conductivity for anode  |
|           |             |                       |             | stabilized solid-state Li metal             |
|           |             |                       |             | batteries." Sustainable Energy Fuels, 2019, |
|           |             |                       |             | 3, 2642-2656.                               |
| 3         | VBIm+PEG    | $1.4 \times 10^{-4}$  | Room        | Zhang, Fengrui, et al. "Highly conductive   |
|           | DA          |                       | Temperature | polymeric ionic liquid electrolytes for     |
|           |             |                       |             | ambient-temperature solid-state lithium     |
|           |             |                       |             | batteries." ACS Appl. Mater.                |
|           |             |                       |             | Interfaces 2020, 12, 23774–23780            |
| 4         | CTA+PEGM    | $1.24 \times 10^{-3}$ | Room        |   |
|           | A+EMIM      |                       | Temperature | Our System                                  |
|           | TFSI        |                       |             |   |
|           |             |                       |             |   |

 Table S2: The comparative literature data on the reported system

<sup>1</sup>H NMR:



**Figure S1**: <sup>1</sup>H NMR for [EMIM][TFSI] 1-Ethyl-3-methyl-imidazolium Bis(trifluromethanesulfonyl) imide ionic liquids

## **Nyquist Plot of SPE**



Figure S2: Impedance plot of SPE membranes

### **Flame Retardant Test**



Figure S3: Flame Test of SPE-IL-II



Figure S4: FESEM study after charge – discharge



Figure S5: Impedance plot of SPE IL-II at different temperatures



Figure S6: Strip-plate analysis at 0.1 mA cm<sup>-2</sup>



Figure S7: Cyclic voltagram of Li/SPE-IL-II/SS



**Figure S8:** a) Galvanastatic charge-discharge of half-cell Li/1M LiPF<sub>6</sub> EC:DMC(1:1v)/LFP at different current densities, b) Discharge capacity and coulombic efficiency at 1C of half-cell Li/1M LiPF<sub>6</sub> EC:DMC(1:1v)/LFP