

## Supplementary Information

### **DFT Study on a Fluorine-Functionalized Nitrogen- and Boron-doped Triangulene as an Electrocatalyst for the Oxygen Reduction Reaction**

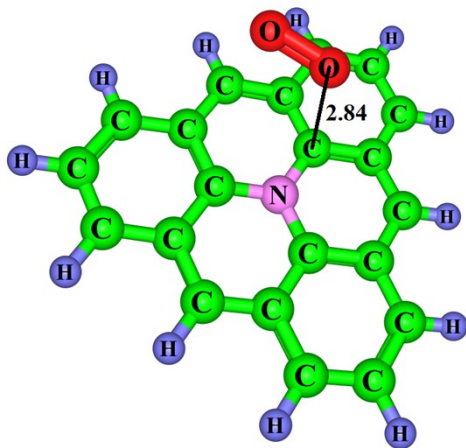
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Department of Physics, Bharathiar University, Coimbatore-641 046, Tamil Nadu, India

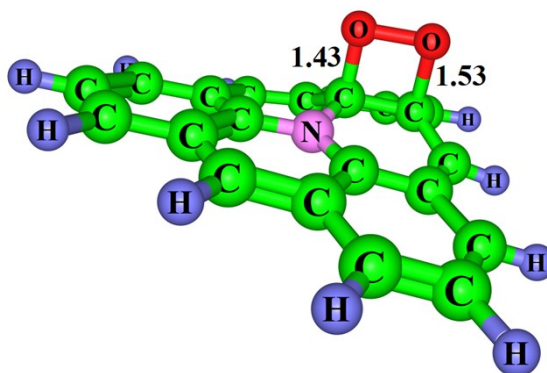
*[\*] Corresponding author: Senthilkumar Lakshmiopathi Email: [lsenthilkumar@buc.edu.in](mailto:lsenthilkumar@buc.edu.in)*

**Table S1.** Thermodynamic properties of the N- and B-doped triangulene (N-tri and B-tri) of ORR at calculated B3LYP-D3(BJ) level of theory/eV.

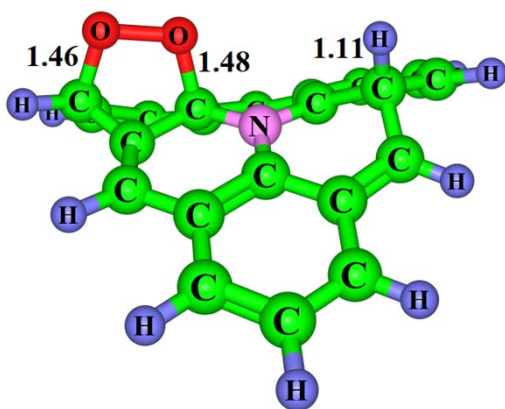
STRUCTURE	N-tri			B-tri		
	$\Delta E$	$\Delta H$	$\Delta G$	$\Delta E$	$\Delta H$	$\Delta G$
<b>R1</b>	0	0	0	0	0	0
<b>I1</b>	1.51	1.46	1.63	0.41	0.38	0.46
<b>R2</b>	0	0	0	0	0	0
<b>I2</b>	-2.35	-2.33	-2.37	0.61	0.64	0.59
<b>R3</b>	0	0	0	0	0	0
<b>I3</b>	-4.58	-4.54	-4.64	-1.38	-1.36	-1.42
<b>R4</b>	0	0	0	0	0	0
<b>I4</b>	-3.09	-3.12	-3.06	-0.09	-0.07	-0.12
<b>R5</b>	0	0	0	0	0	0
<b>P1</b>	-2.82	-2.78	-2.91	-2.90	-2.85	-3.02



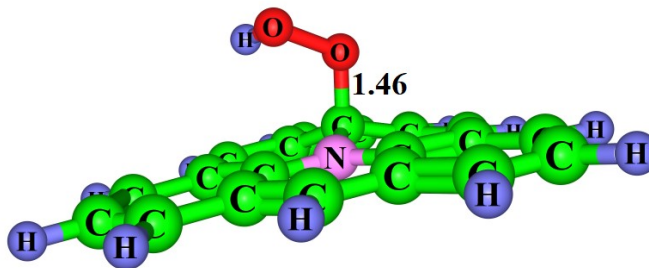
R1



I1

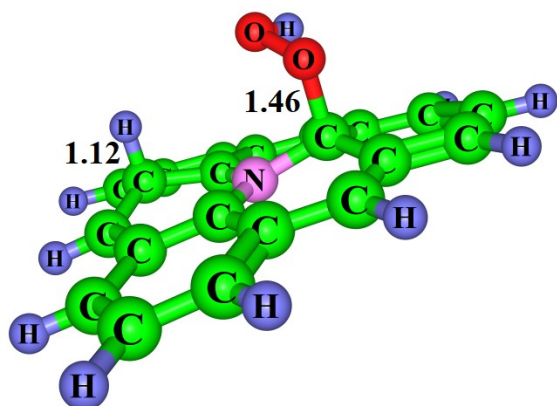


R2

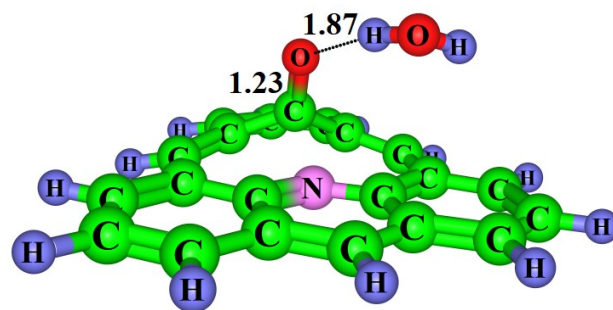


I2

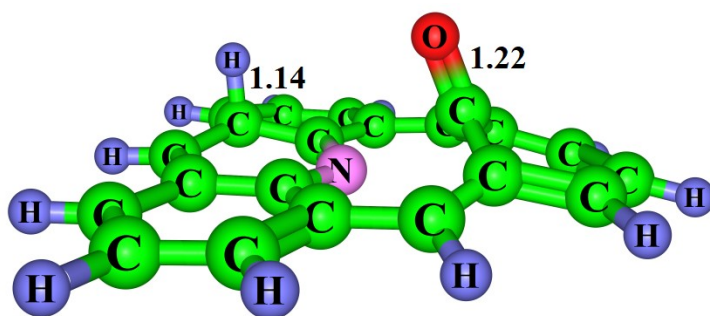
**Fig S1.** The optimized geometry of N-doped triangulene (N-tri) with  $^*OO$ ,  $^*OOH$ ,  $^*O+H_2O$ ,  $^*OH$  and  $H_2O$  molecules in reactants (R), intermediates (I) and the final product (P).



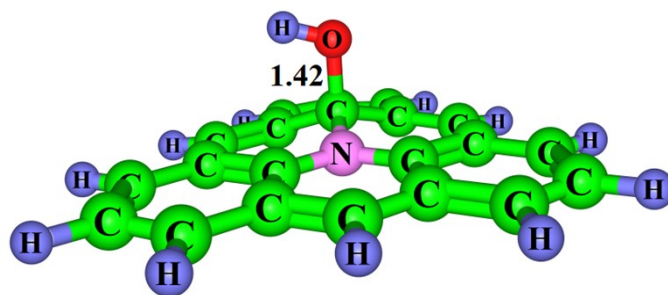
R3



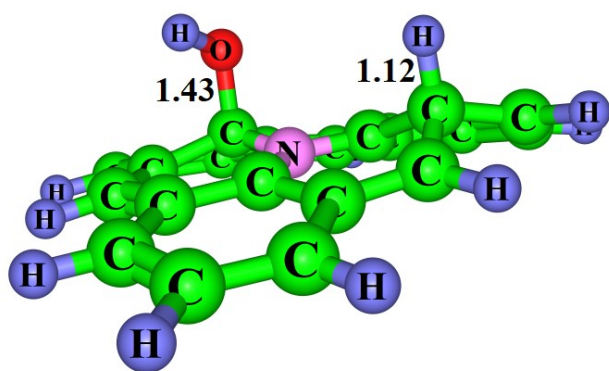
I3



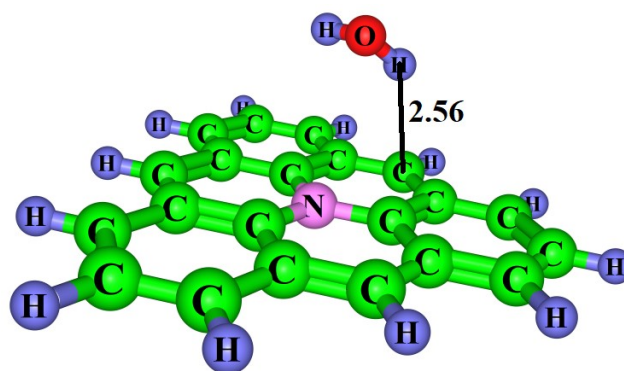
R4



I4

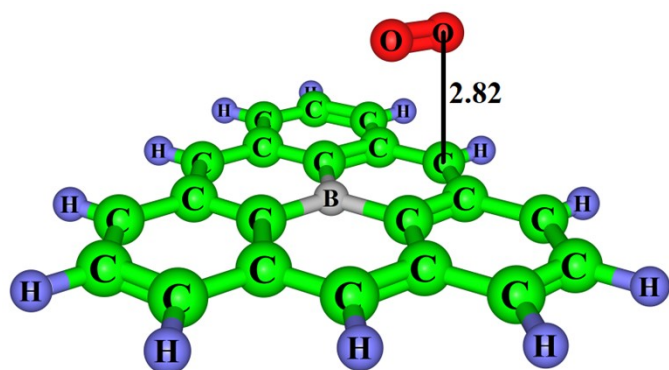


R5

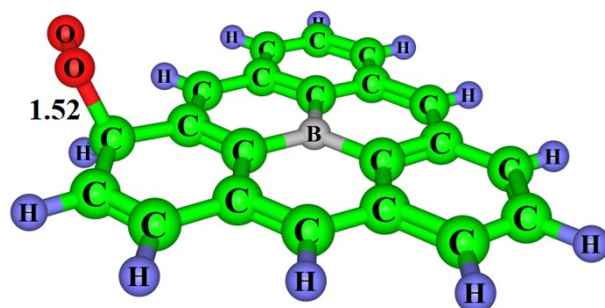


P1

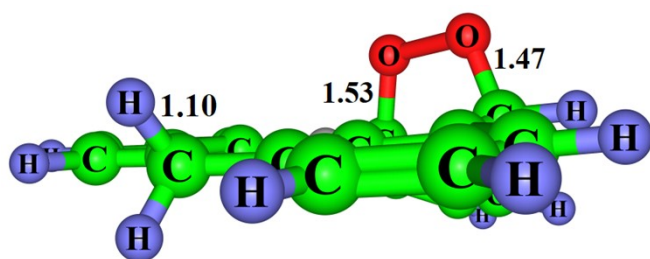
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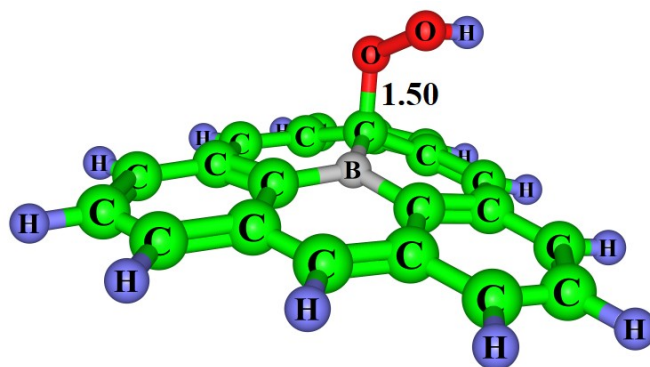
R1



I1

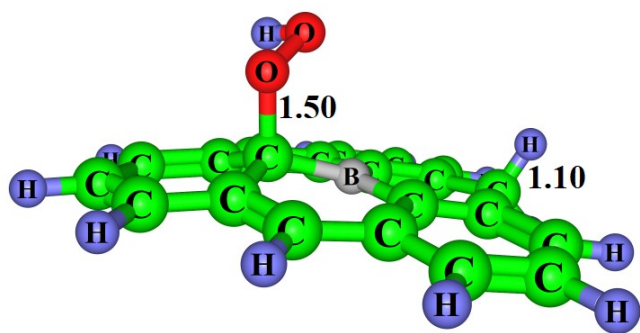


R2

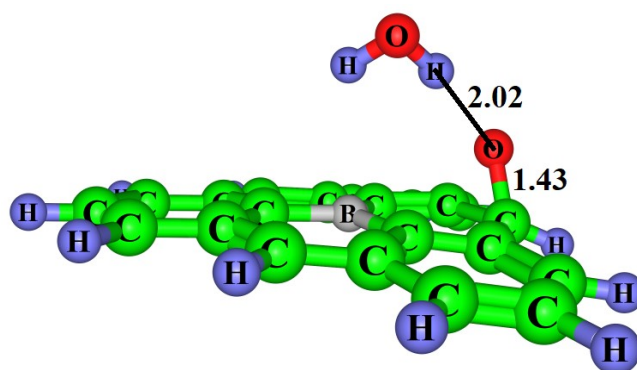


I2

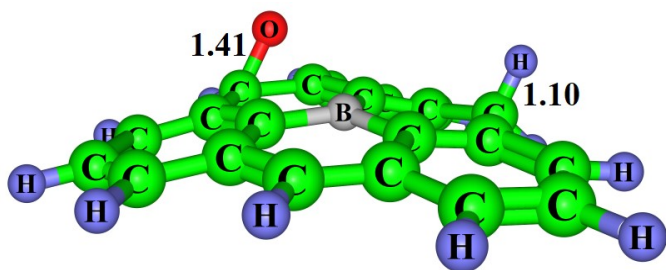
**Fig S2.** The optimized geometry of B-doped triangulene (B-tri) with \*OO, \*OOH, \*O+H<sub>2</sub>O, \*OH and H<sub>2</sub>O molecules in reactants (R), intermediates (I) and the final product (P).



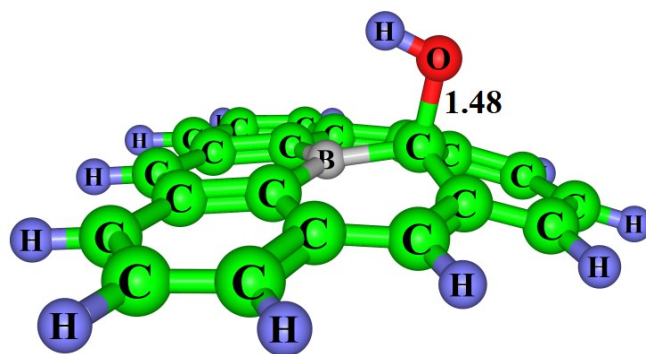
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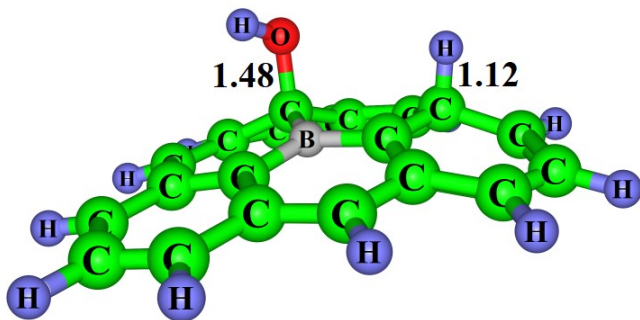
I3



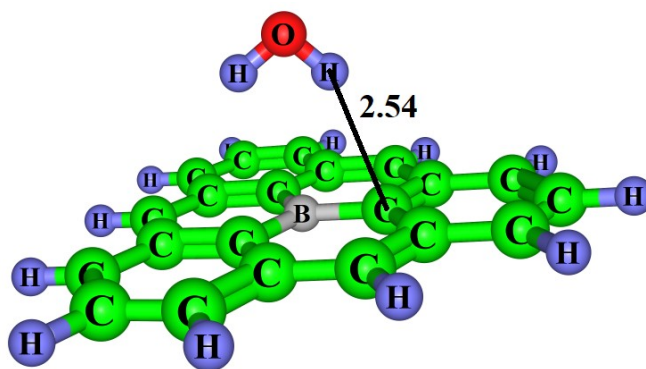
R4



I4

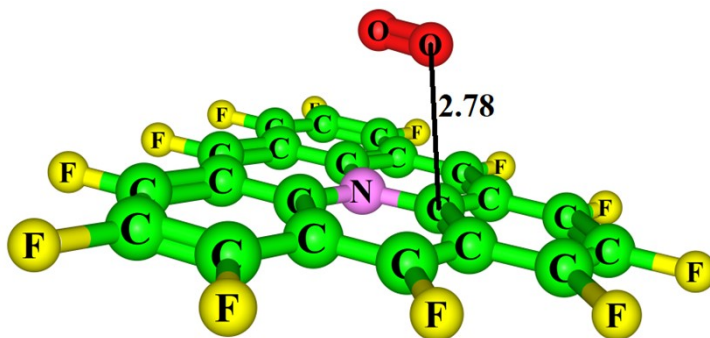


R5

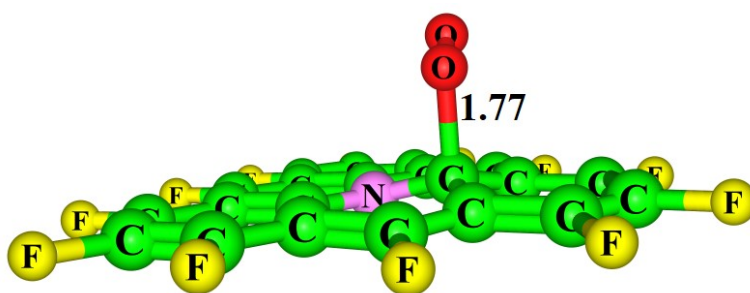


P1

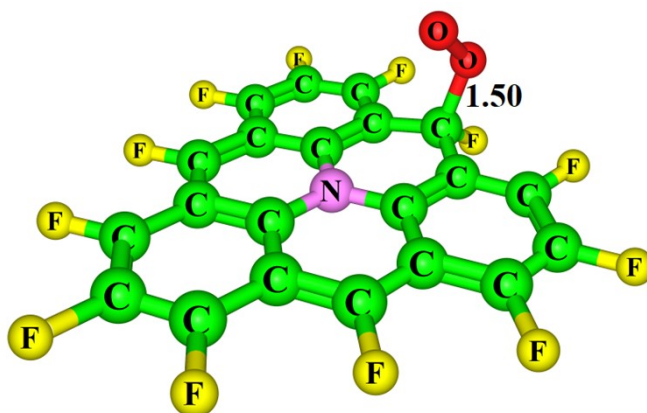
Fig S2 contd...



R1

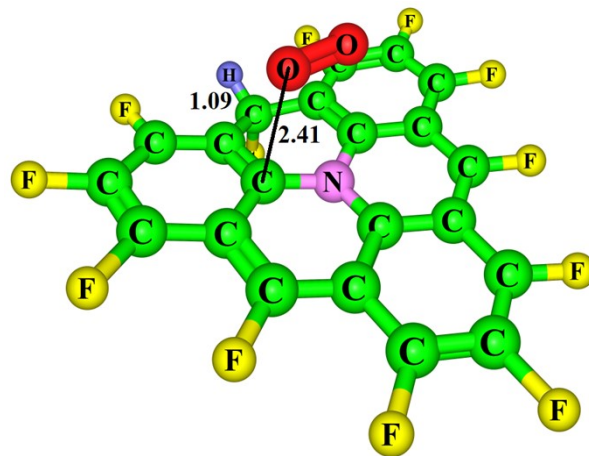


TS1

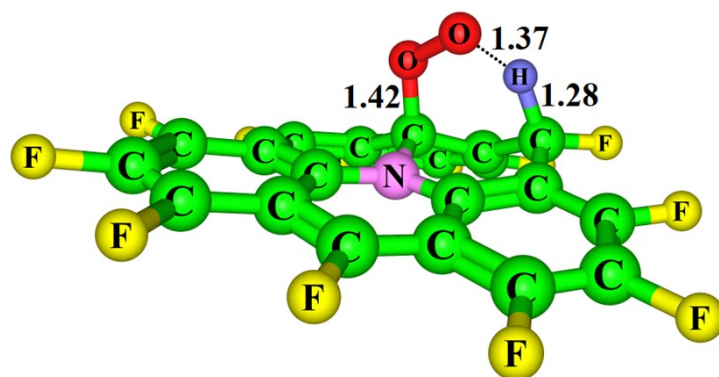


II

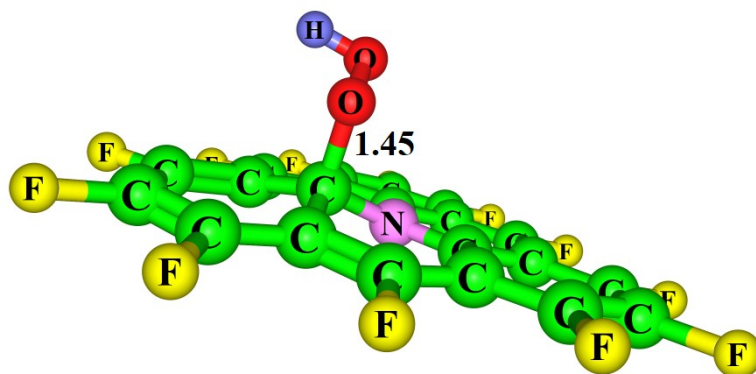
**Fig S3.** The optimized structures of reactants (R), transition states (TS), intermediates (I) and the final product (P) of ORR with F-N-tri.



R2



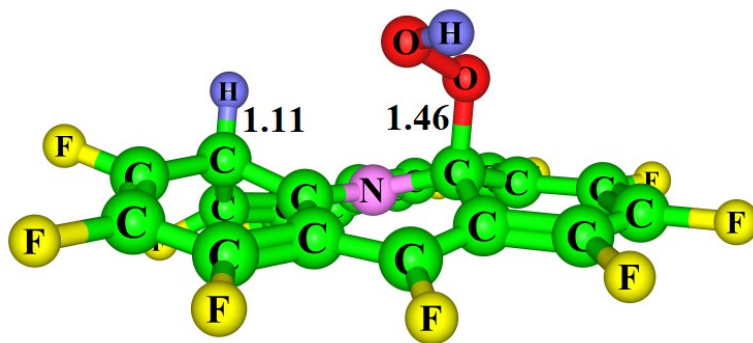
TS2



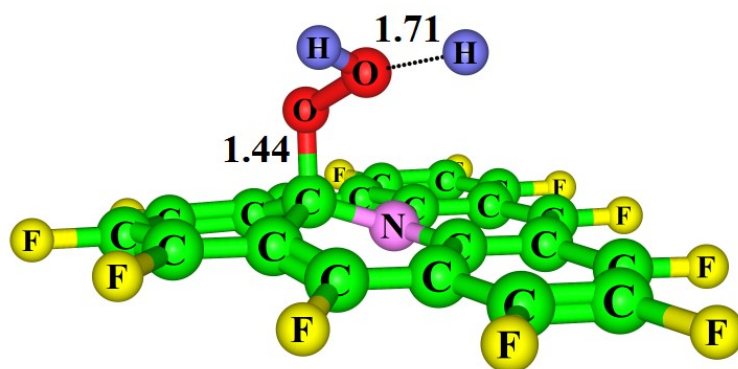
I2

Fig S3 contd...

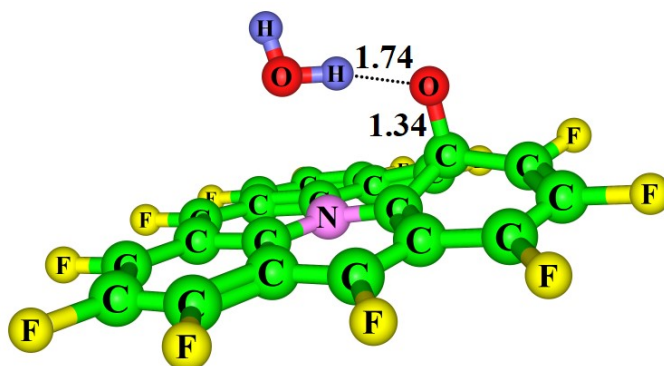




R3

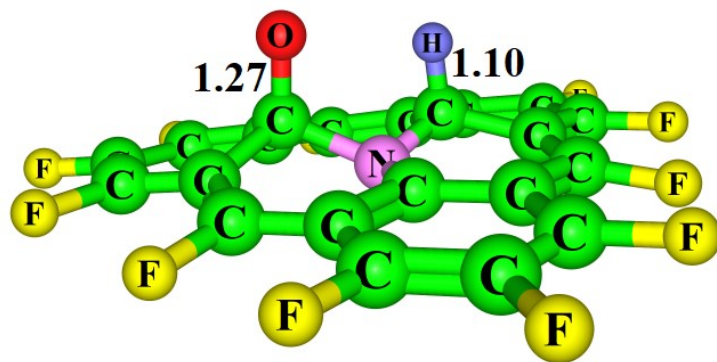


TS3

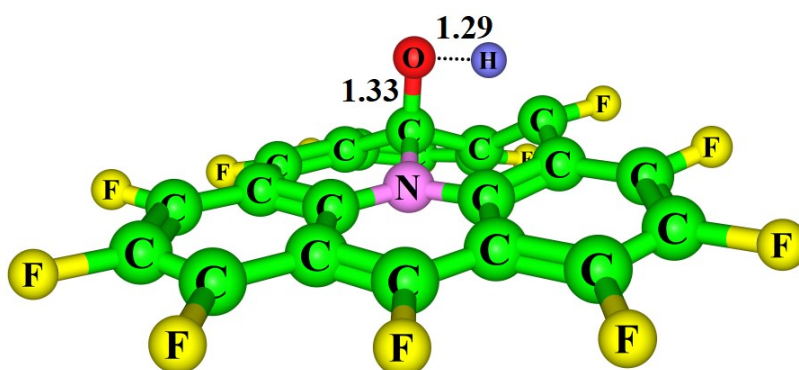


I3

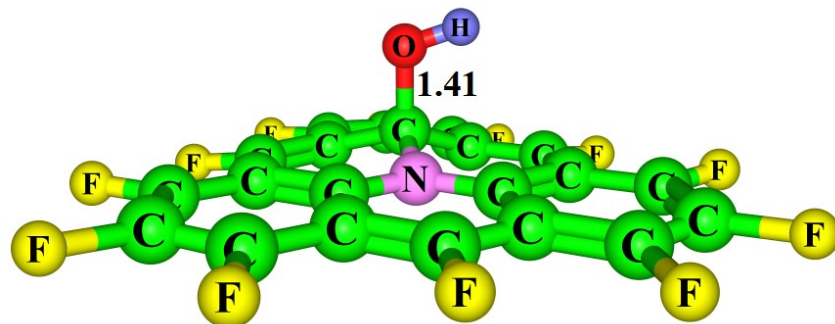
Fig S3 contd...



R4

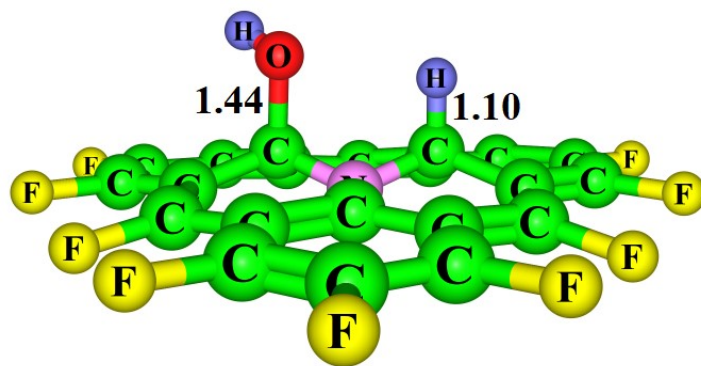


TS4

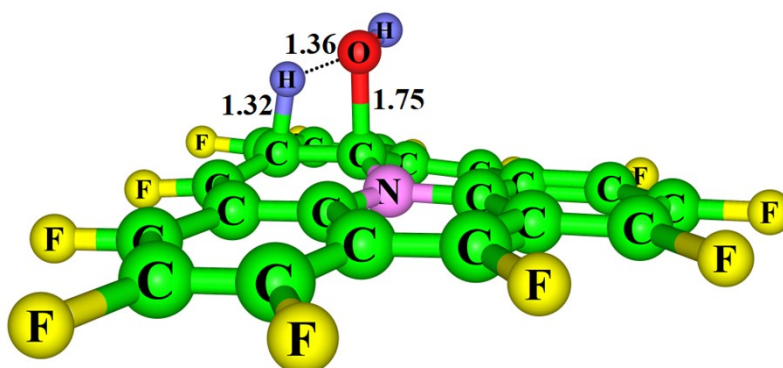


I4

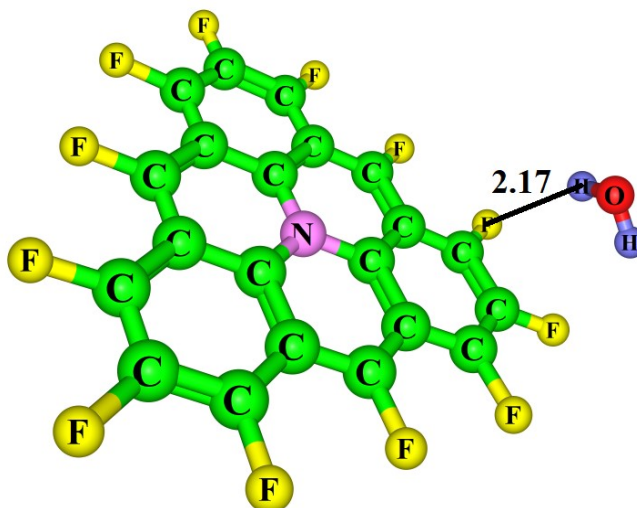
Fig S3 contd...



R5

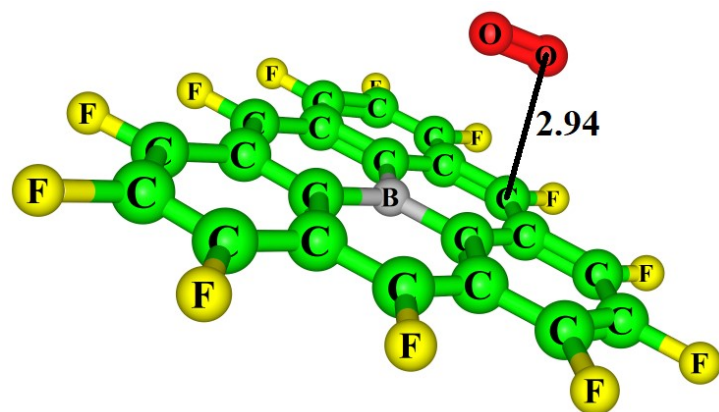


TS5

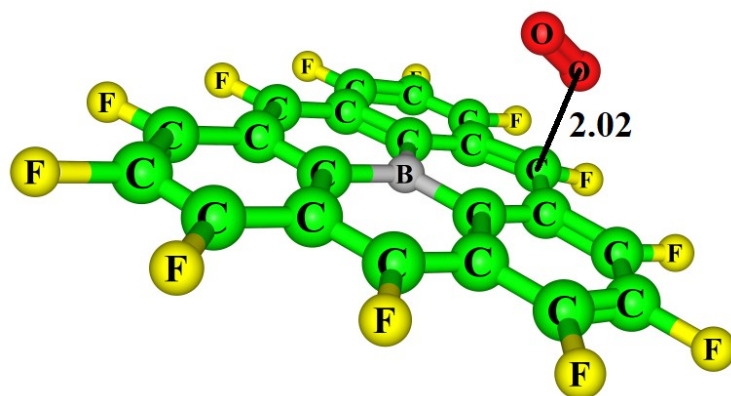


P1

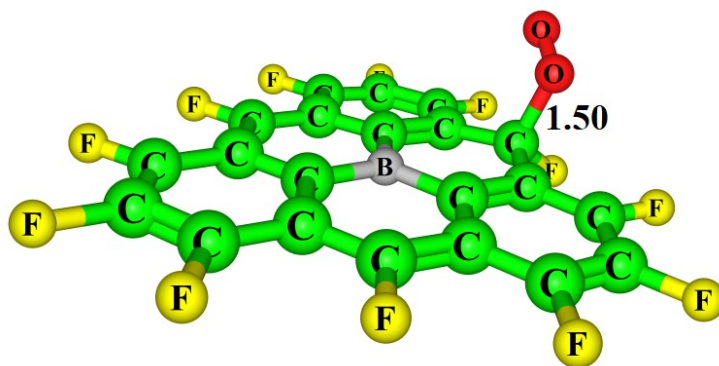
Fig S3 contd...



R1

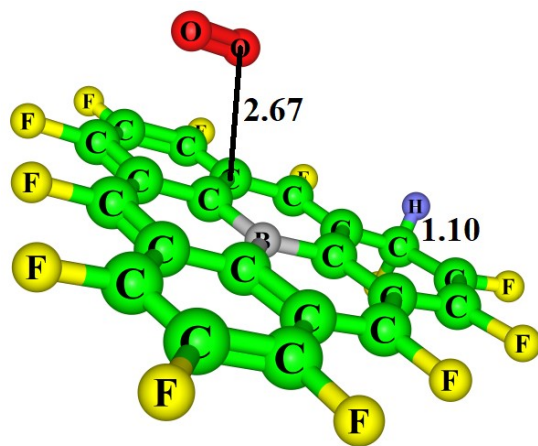


TS1

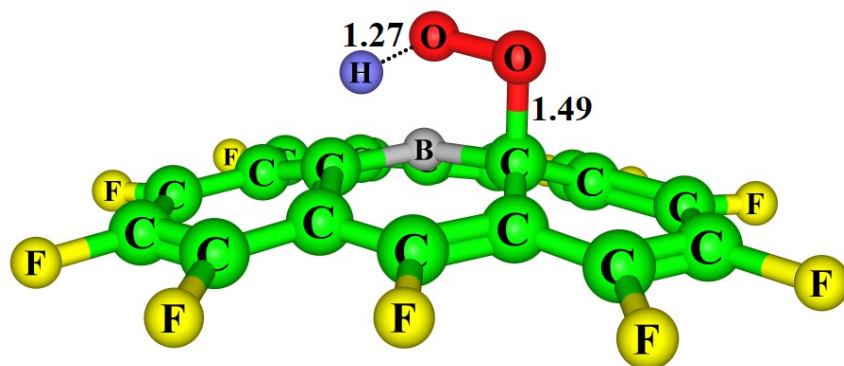


II

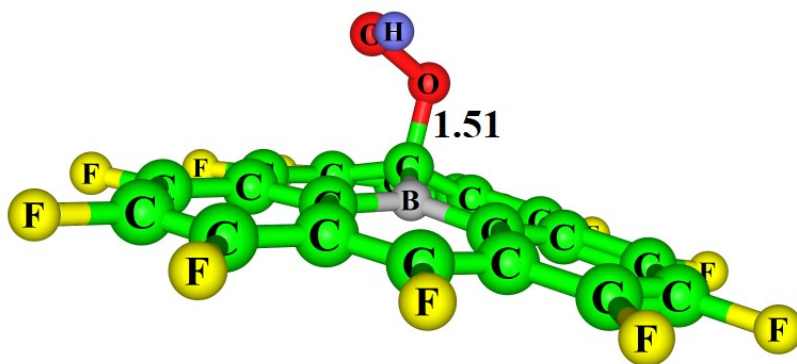
**Fig S4.** The optimized structures of reactants (R), transition states (TS), intermediates (I) and the final product (P) of ORR with F-B-tri.



R2

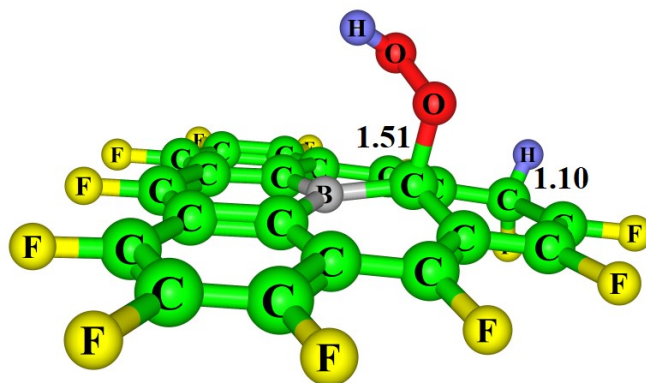


TS2

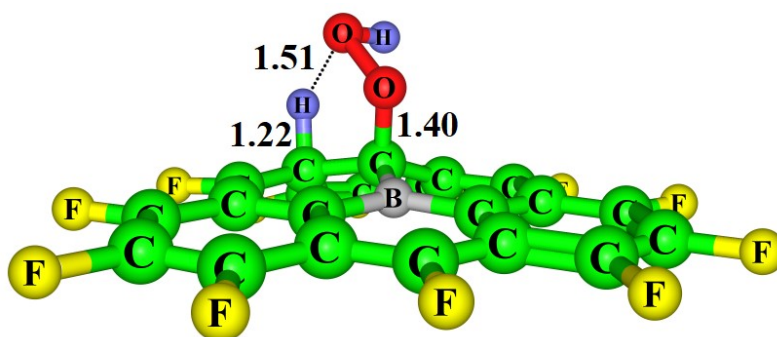


I2

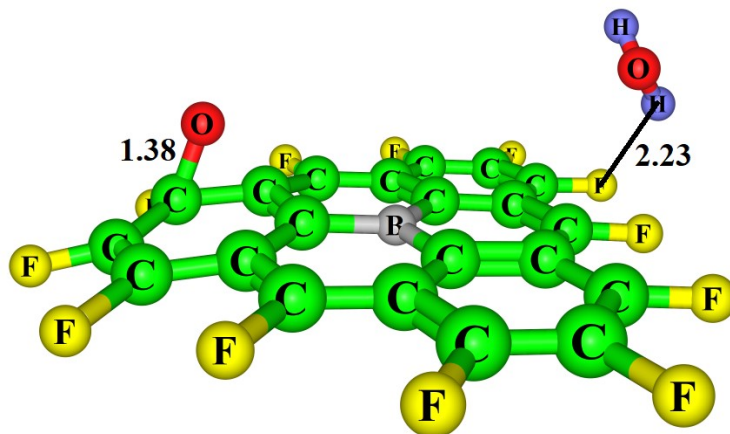
Fig S4 contd...



R3

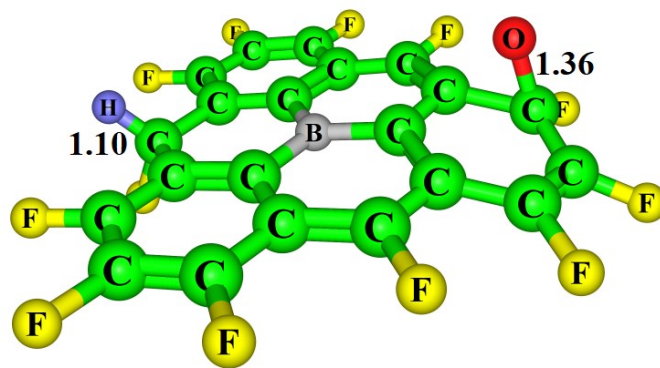


TS3

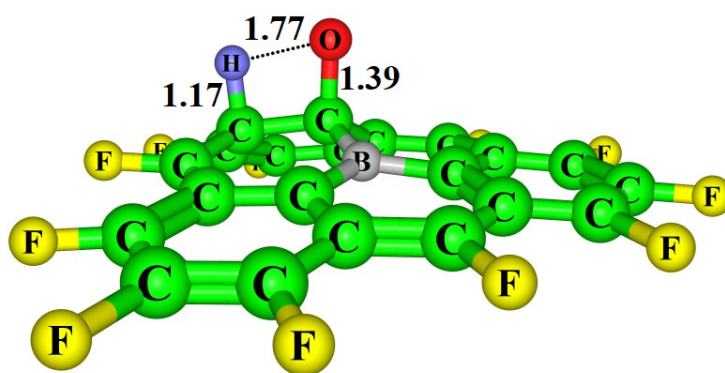


I3

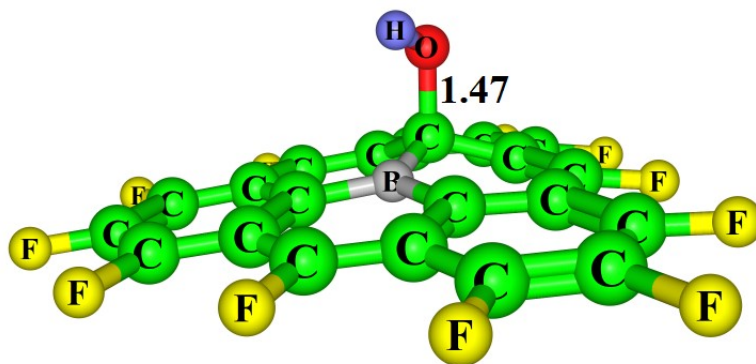
Fig S4 contd...



R4

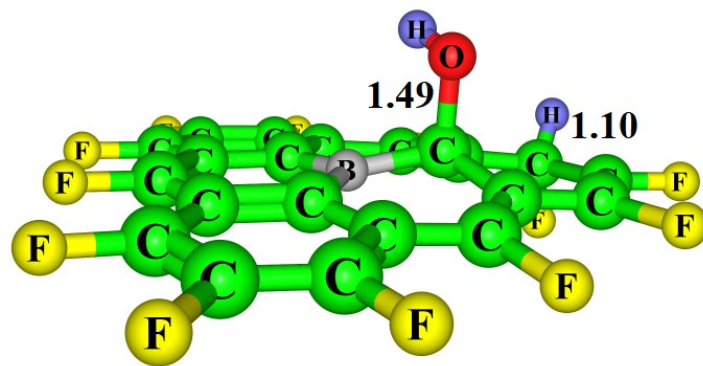


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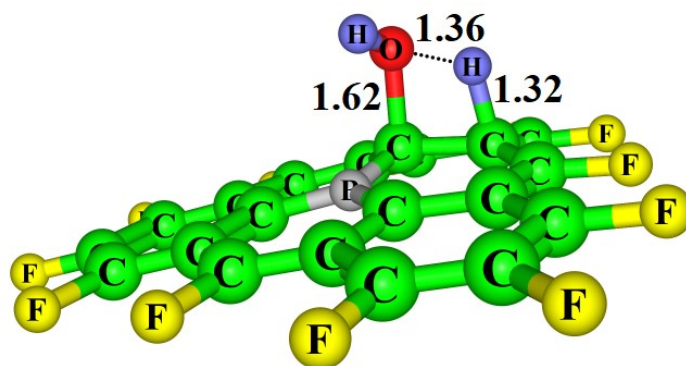


I4

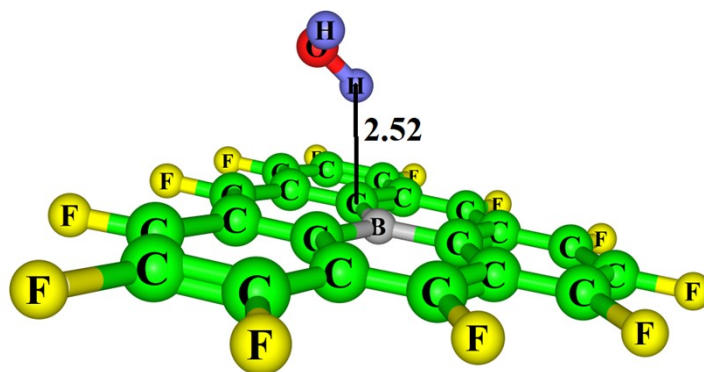
Fig S4 contd...



R5



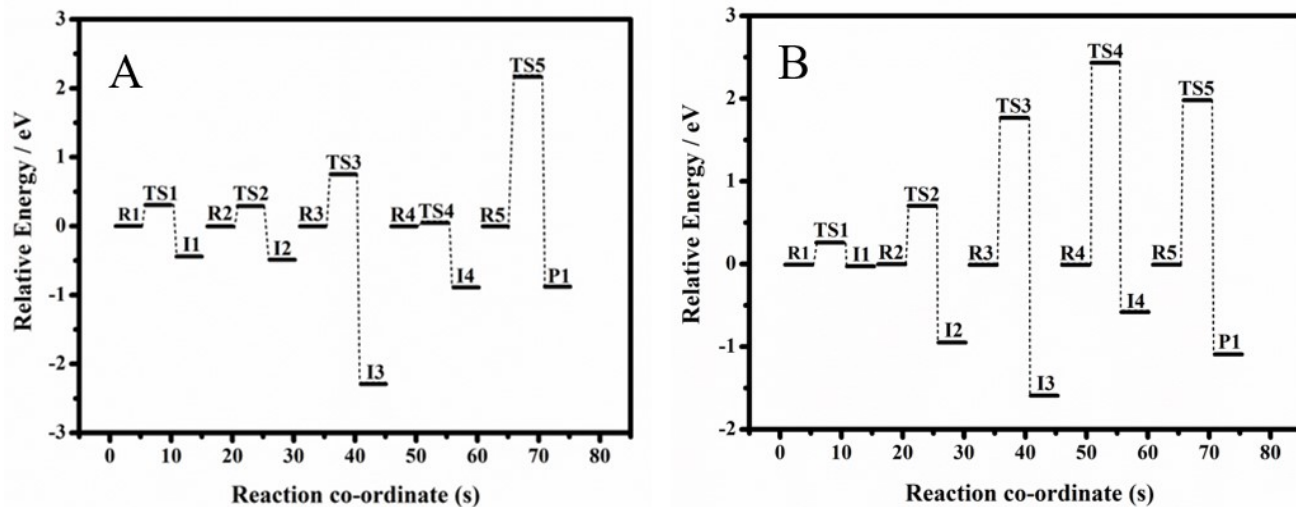
TS5



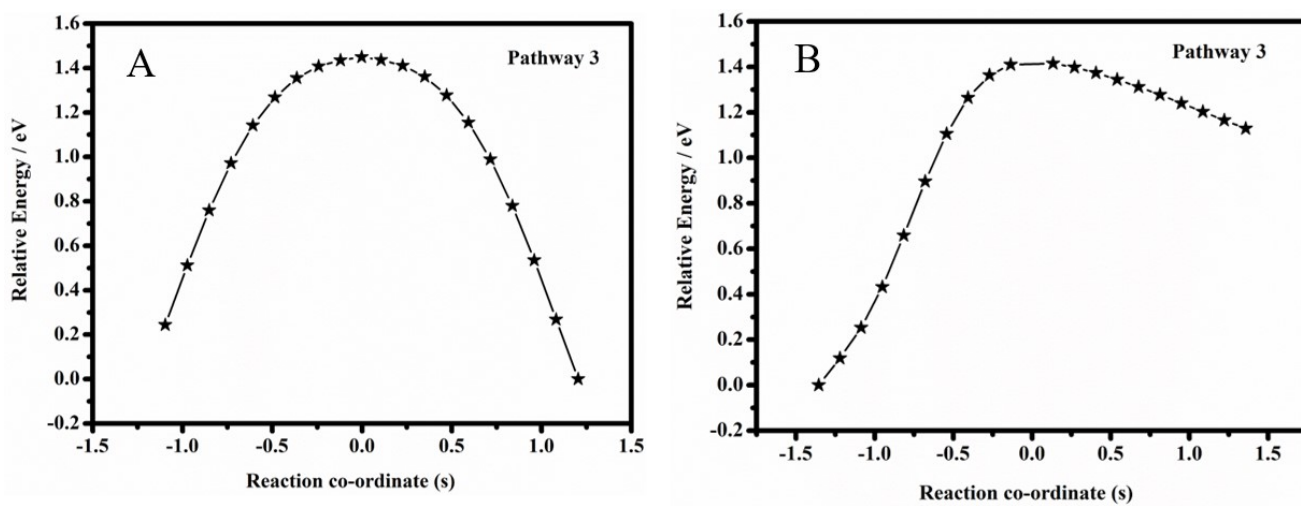
P1

Fig S4 contd...





**Fig S5.** The relative energy profile of  $4e^-$  reduction pathway in the ORR steps are calculated at B3LYP-D3(BJ) level of theory for **A)** F-N-tri **B)** F-B-tri.



**Fig S6.** The intrinsic reaction coordinate (IRC) plots of ORR reaction calculated at B3LYP-D3(BJ) level of theory for **A)** F-N-tri **B)** F-B-tri.

**Table S2.** The rate limiting step ( $\Delta G_1$ ,  $\Delta G_2$ ,  $\Delta G_3$ ,  $\Delta G_4$ ), maximum free energy ( $G^{\text{ORR}}$ ) and overpotential ( $\eta^{\text{ORR}}$ ) are calculated for F-N-tri and F-B-tri molecules.

<b>Molecules</b>	$\Delta G_1$ (eV)	$\Delta G_2$ (eV)	$\Delta G_3$ (eV)	$\Delta G_4$ (eV)	$G^{\text{ORR}}$ (eV)	$\eta^{\text{ORR}}$ (V)
<b>F-N-tri</b>	-0.81	-2.59	-0.28	-1.24	-0.28	1.16
<b>F-B-tri</b>	-0.53	-2.81	-0.06	-1.52	-0.06	1.21