

Supporting information

Precise control of TiO₂ overlayer on hematite nanorod arrays by ALD for the photoelectrochemical water splitting

Jiao Wang, Letizia Liccardo, Heydar Habibimarkani, Ewa Wierzbicka, Thorsten Schultz, Norbert Koch, Elisa Moretti, and Nicola Pinna**

Jiao Wang, Nicola Pinna

Department of Chemistry and the Center for the Science of Materials Berlin, Brook-Taylor-Str. 2, 12489 Berlin, Germany

E-mail: nicola.pinna@hu-berlin.de

Letizia Liccardo, Heydar Habibimarkani, Elisa Moretti

Department of Molecular Sciences and Nanosystems, Ca' Foscari University of Venice, Via Torino 155, 30172 Venezia Mestre, Italy

Email: elisa.moretti@unive.it

Ewa Wierzbicka

Department of Functional Materials and Hydrogen Technology, Faculty of Advanced Technologies and Chemistry, Military University of Technology, 2 Kaliskiego Street, Warsaw 00908, Poland

Thorsten Schultz, Norbert Koch

Institut für Physik and IRIS Adlershof, Humboldt-Universität zu Berlin, Brook-Taylor-Str. 6, 12489 Berlin, Germany

Thorsten Schultz, Norbert Koch

Helmholtz-Zentrum Berlin für Materialien und Energie GmbH, Hahn-Meitner-Platz 1, 14109 Berlin, Germany

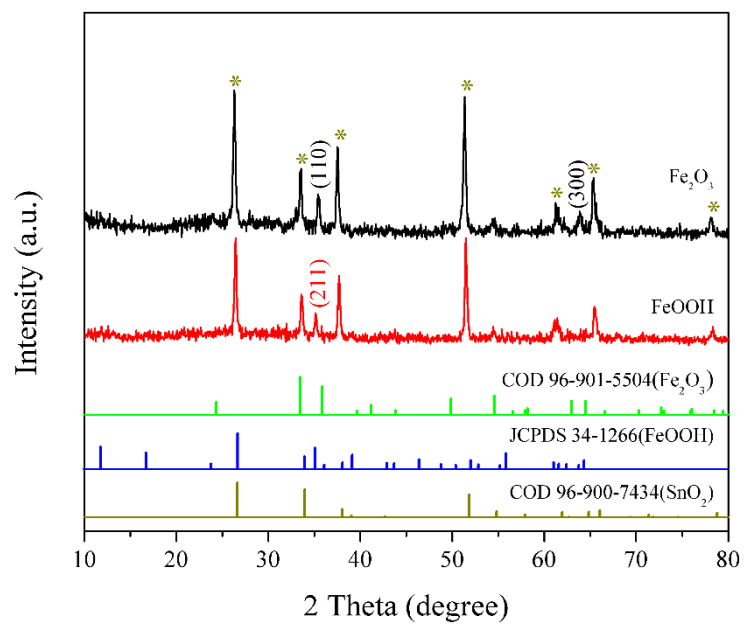


Figure S1. XRD patterns of pristine β -FeOOH and α -Fe₂O₃.

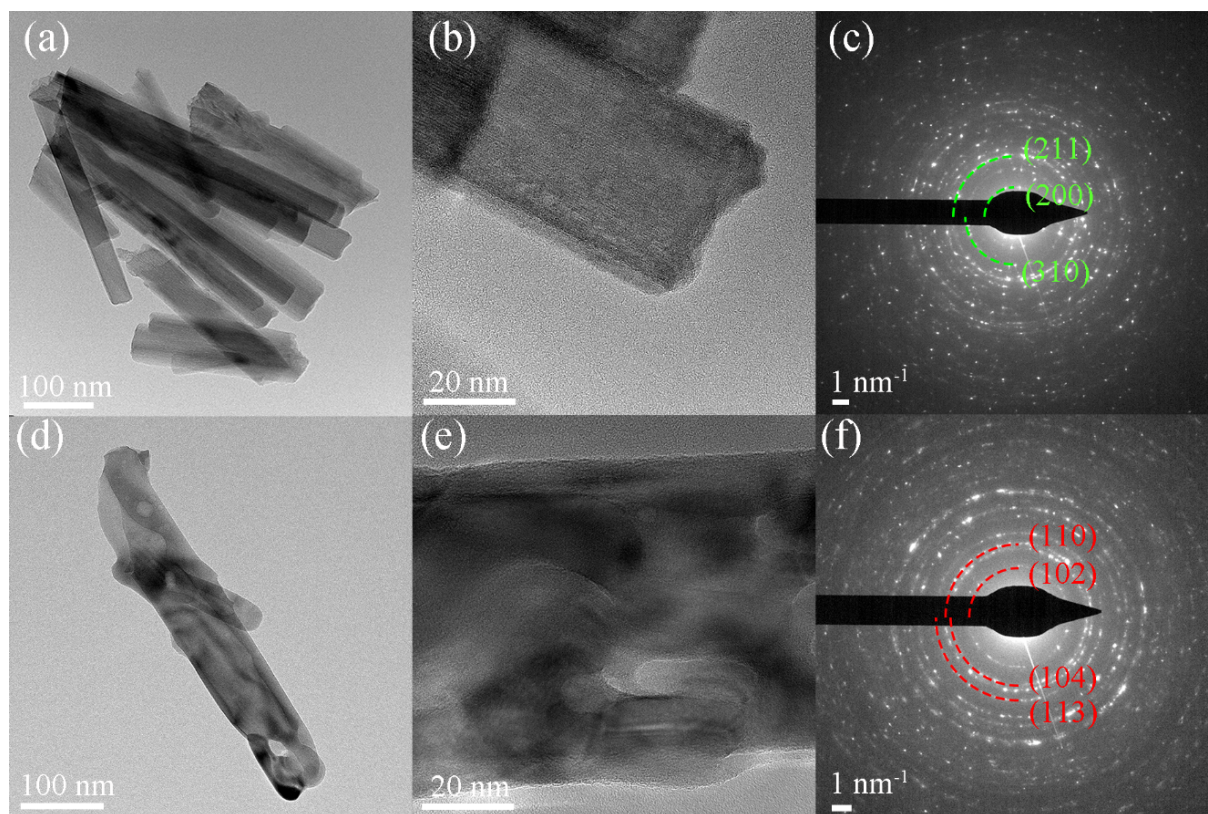


Figure S2. (a,b) TEM images of β -FeOOH nanorods. (c) Corresponding SAED pattern of β -FeOOH nanorods. (d,e) HRTEM images of α -Fe₂O₃ nanorods. (f) Corresponding SAED pattern of α -Fe₂O₃ nanorods.

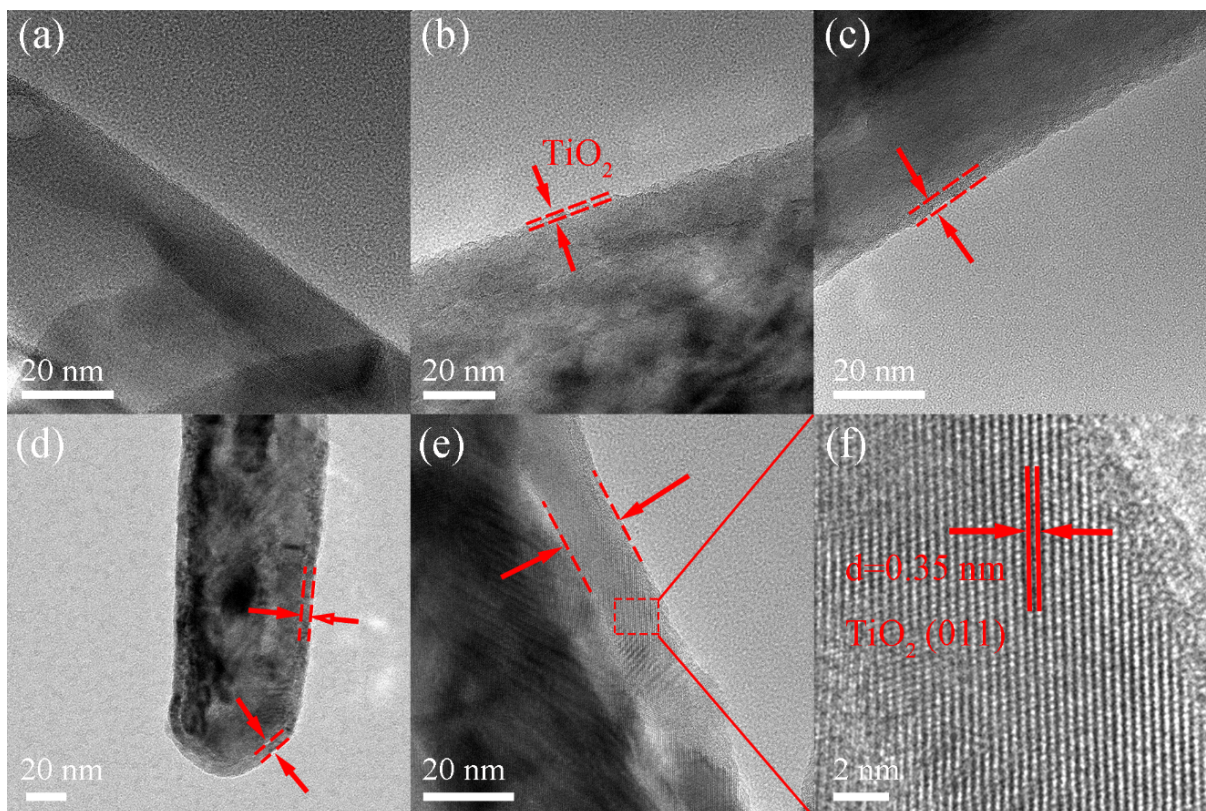


Figure S3. HRTEM images of TiO₂ deposited on pristine hematite with different ALD cycles: (a) TiO₂-10/Fe₂O₃, (b) TiO₂-20/Fe₂O₃, (c) TiO₂-40/Fe₂O₃, (d) TiO₂-80/Fe₂O₃, (e) TiO₂-150/Fe₂O₃, (f) The magnified view of the lattice fringes of TiO₂-150/Fe₂O₃.

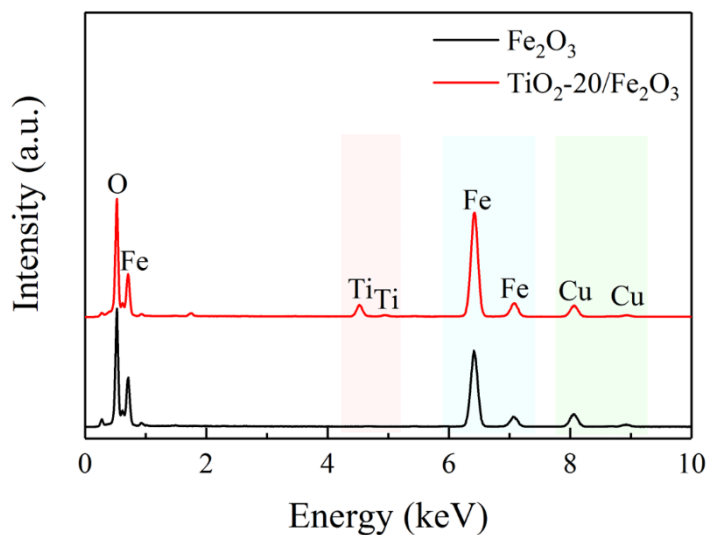


Figure S4. EDX spectra for pristine Fe₂O₃ and TiO₂-20/Fe₂O₃ analyzed on a copper TEM grid.

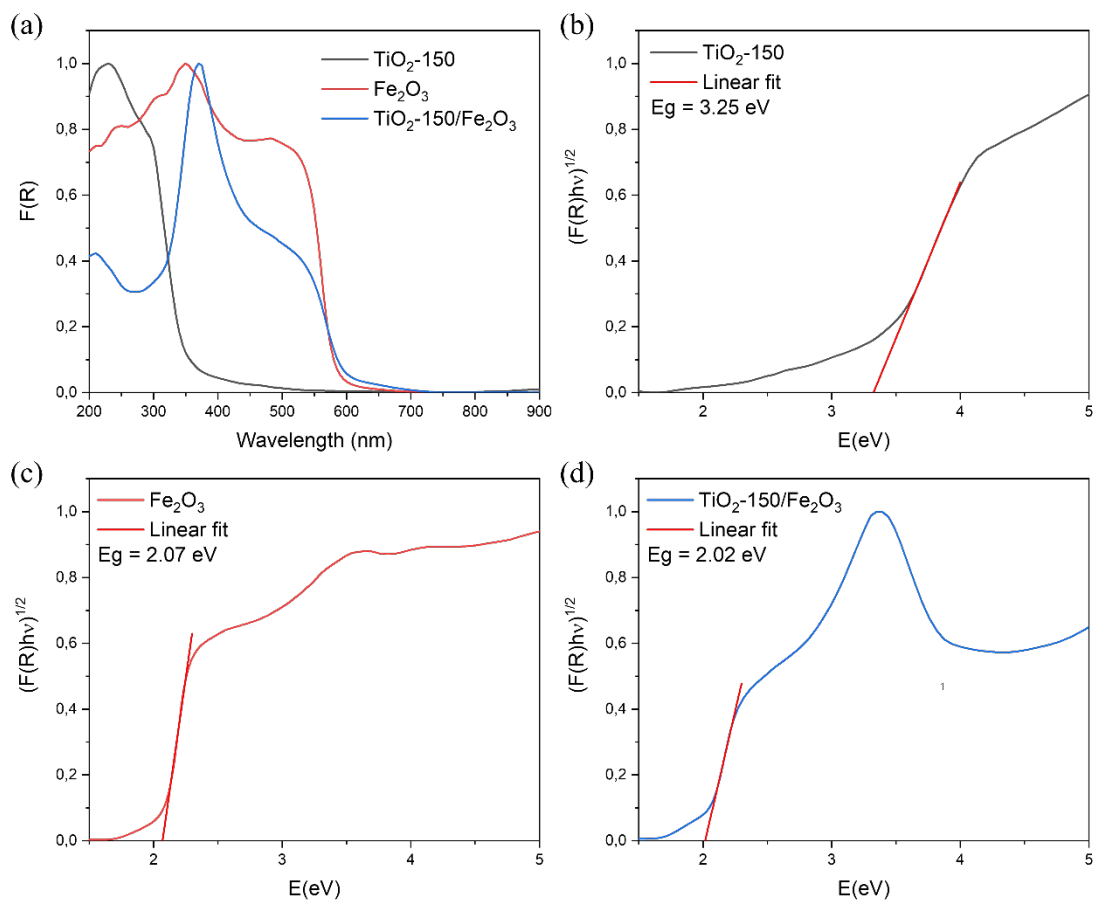


Figure S5. (a) Plots of $F(R)$ versus wavelength of the pristine Fe₂O₃, pristine TiO₂ and TiO₂/Fe₂O₃ nanocomposites from the UV-vis diffuse reflectance spectra. Plots of $(F(R)h\nu)^{1/2}$ versus photon energy for the (b) pristine TiO₂ (c) pristine Fe₂O₃ and (d) TiO₂/Fe₂O₃ heterojunction.

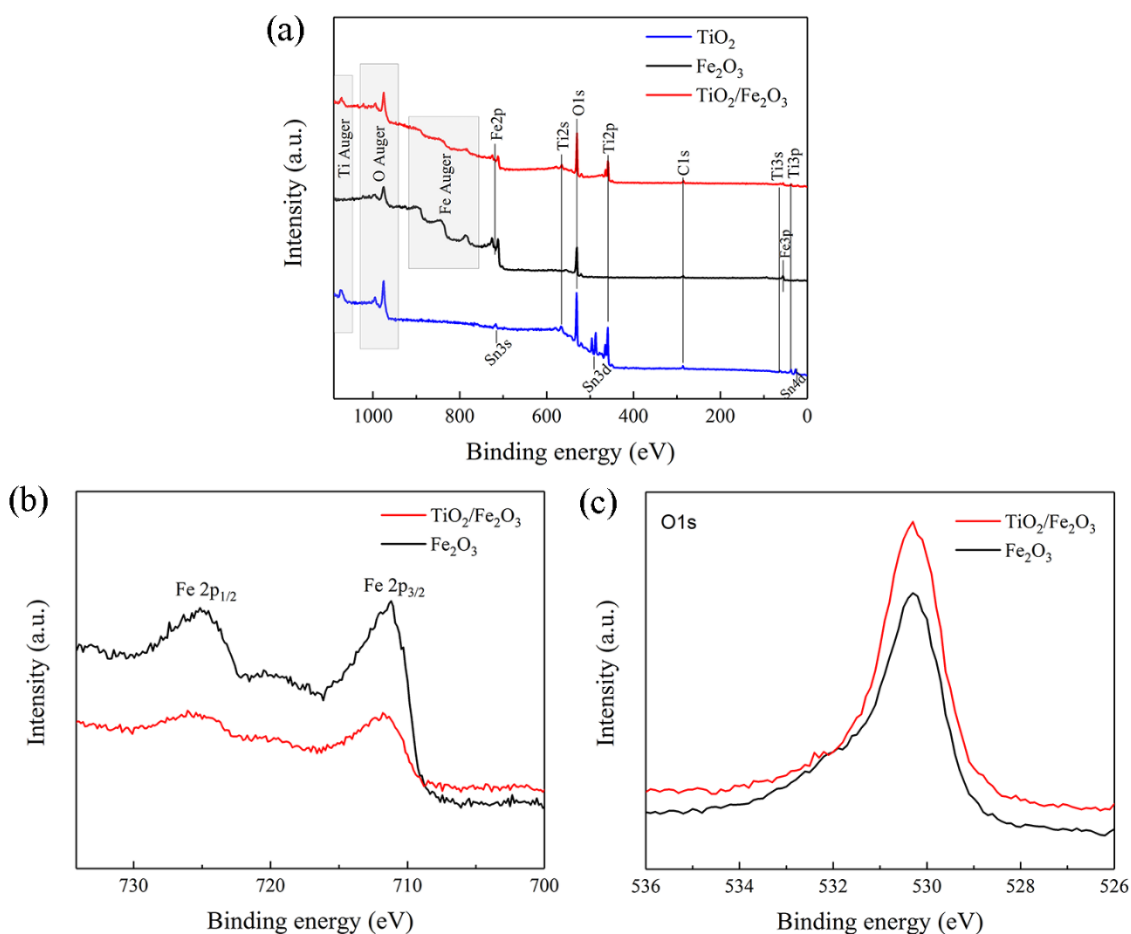


Figure S6. (a) XPS survey spectra of the pristine Fe₂O₃, pristine TiO₂ with 20ALD cycles and Fe₂O₃ coated with 20 ALD cycles of TiO₂ overlayer. For the TiO₂ sample, the tin signal from the FTO substrate is visible due to the small thickness of the TiO₂ layer. (b) High-resolution Fe2p photoelectron spectrum of the pristine Fe₂O₃ and TiO₂-20/Fe₂O₃. (c) High-resolution O1s photoelectron spectrum of pristine Fe₂O₃ and TiO₂-20/Fe₂O₃.

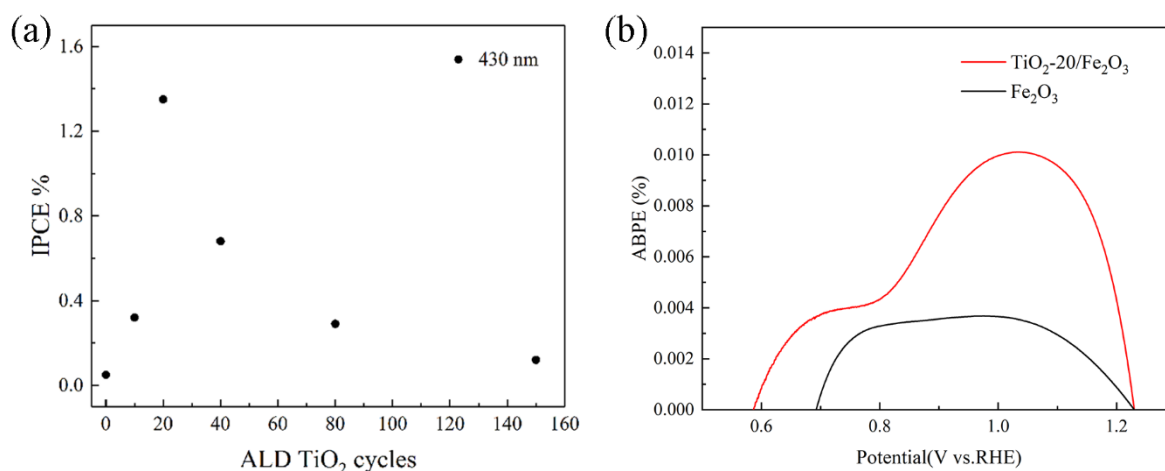


Figure S7. (a) IPCE spectra of pristine Fe₂O₃ and TiO₂/Fe₂O₃ photoanodes measured in 1M KOH at 0.5 V bias vs. Hg/HgO under 430 nm light irradiation. (b) ABPE of pristine Fe₂O₃ and TiO₂-20/Fe₂O₃ photoanodes.

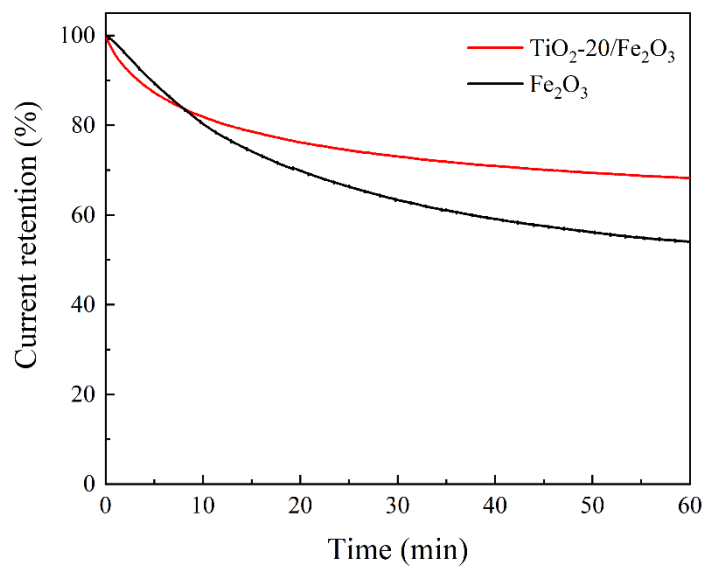


Figure S8. Stability test of pristine Fe₂O₃ and TiO₂-20/Fe₂O₃ photoanodes.

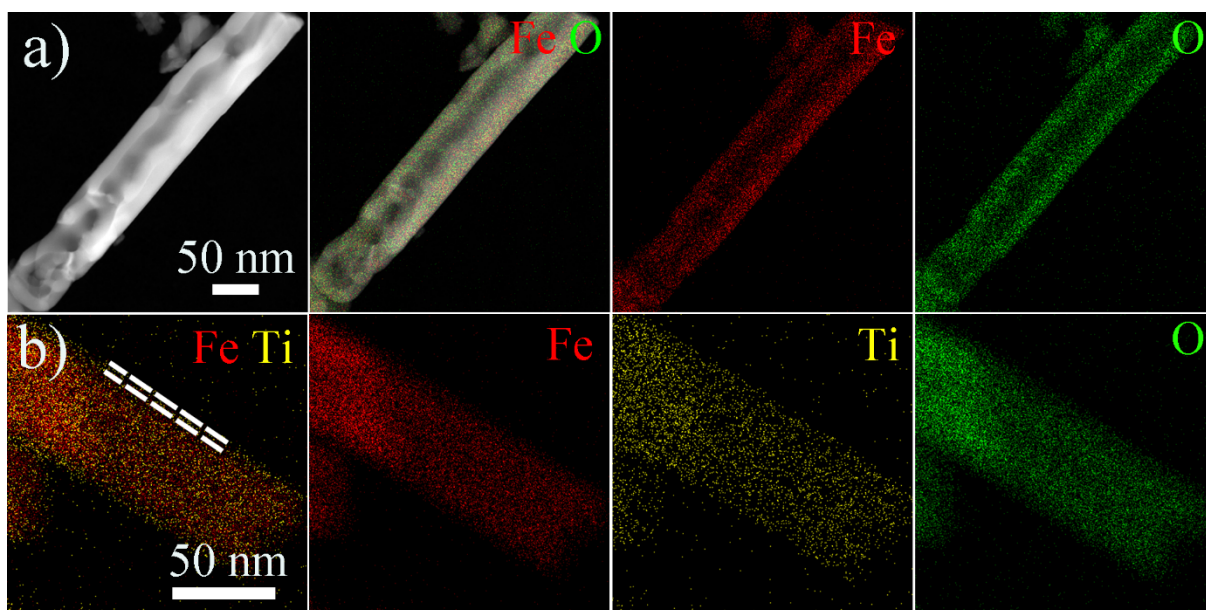


Figure S9. The morphology of the synthesized photocatalyst after stability test: a) pristine Fe₂O₃, b) TiO₂-20/Fe₂O₃.

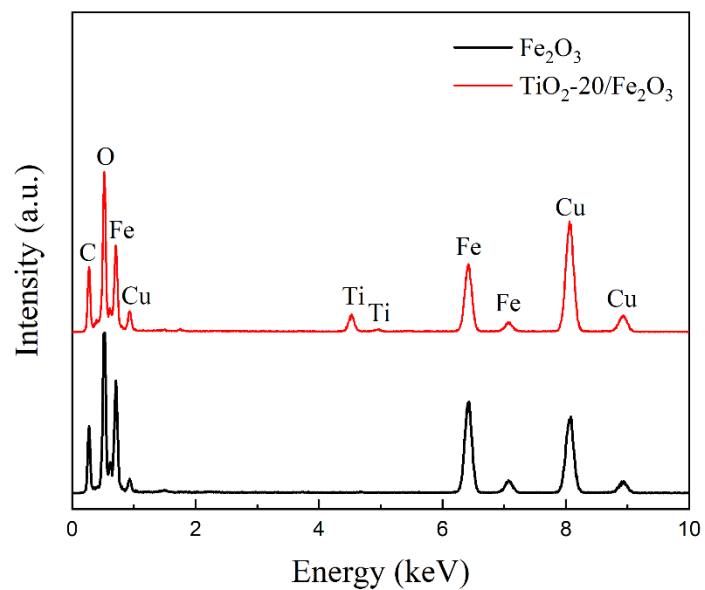


Figure S10. The composition of the pristine Fe₂O₃ and TiO₂-20/Fe₂O₃ after stability test.

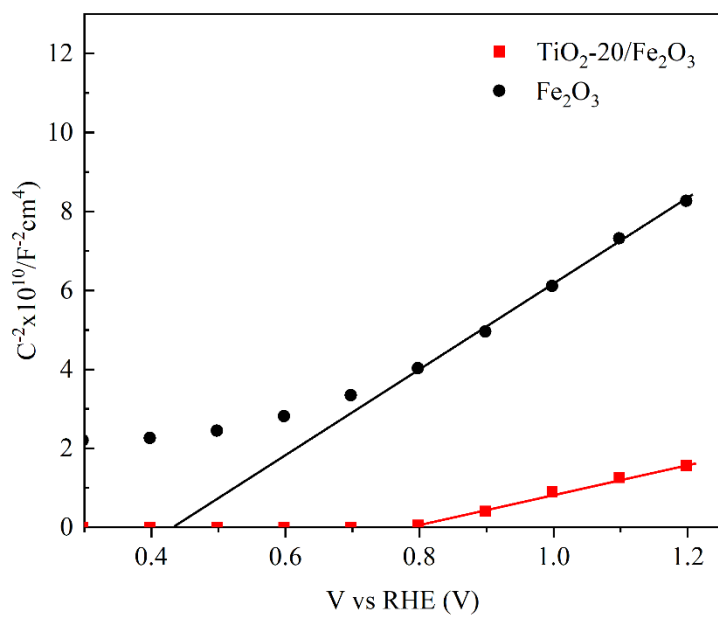


Figure S11. Mott-Schottky plots of pristine Fe₂O₃ and TiO₂-20/Fe₂O₃ heterostructure.

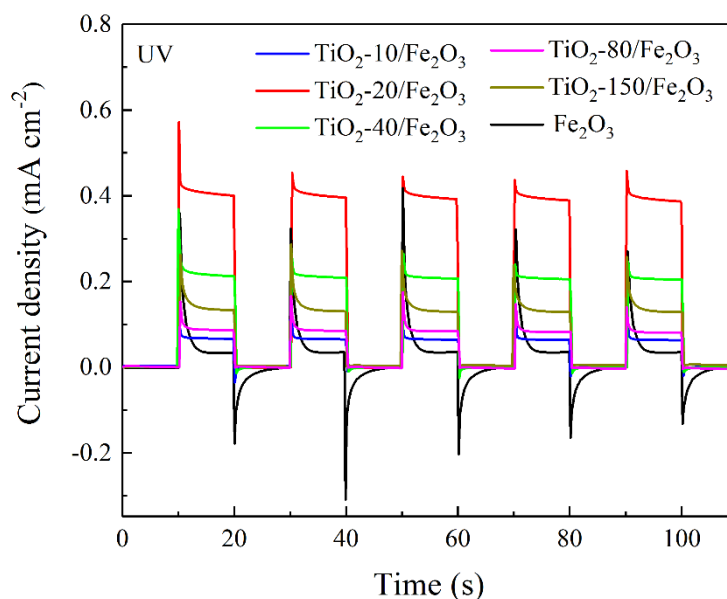


Figure S12. Time-based photocurrent density for pristine and TiO₂-coated hematite photoanodes at 0.5 V bias vs. Hg/HgO under UV light irradiation.

Table S1. Average TiO₂ overlayer thickness and the GPC calculated from TEM micrographs and spectroscopic ellipsometry.

Samples	Transmission electron microscopy		Spectroscopic ellipsometry	
	Thickness (nm)	Slope of line fit (GPC)	Thickness (nm)	Slope of line fit (GPC)
TiO ₂ -10/Fe ₂ O ₃	---		2.2	
TiO ₂ -20/Fe ₂ O ₃	1.7		3.5	
TiO ₂ -40/Fe ₂ O ₃	3.5	0.88 Å	5.0	0.96 Å
TiO ₂ -80/Fe ₂ O ₃	6.8		9.0	
TiO ₂ -150/Fe ₂ O ₃	13.2		15.1	

Table S2. The atomic/weight percentage of all the samples with different ALD cycles.

Samples	Atom fraction (%)		Mass fraction (%)	
	Ti	Fe	Ti	Fe
TiO ₂ -10/Fe ₂ O ₃	1.28	28.2	2.21	57.0
TiO ₂ -20/Fe ₂ O ₃	3.89	39.9	5.62	67.2
TiO ₂ -40/Fe ₂ O ₃	6.59	24.7	11.3	49.3
TiO ₂ -80/Fe ₂ O ₃	9.88	22.1	16.9	44.1
TiO ₂ -150/Fe ₂ O ₃	18.4	22.7	28.5	41.0