### **Supporting Information**

# Peptide-Polyurea Hybrids: A Platform for Tunable, Thermally-stable, and Injectable Hydrogels

Jessica Thomas<sup>1</sup>, Zachary R. Hinton<sup>1,2</sup>, LaShanda T.J. Korley<sup>1,2</sup>

<sup>1</sup>Materials Science and Engineering, University of Delaware, Newark, DE 19716 <sup>2</sup>Chemical and Biomolecular Engineering, University of Delaware, Newark, DE 19716

#### **Table of Contents**

1.	<sup>1</sup> H Nuclear magnetic resonance (NMR) Spectra			
2.	Vial inversion test of A5-104			
3.	Calculation of peptide weight fractions			
4.	Gel permeation chromatography (GPC) data			
5.	Rheology of non-peptide control7			
6.	Frequency and amplitude sweeps for peptide-polyurea (PPU) hydrogels8			
7.	Injection simulation plots			
8.	Recovery times for PPU hydrogels post injection simulation11			
9.	Temperature sweeps of PPU hydrogels at varying concentrations and measurement frequencies			
10	Quasi gel-point curves of PPU hydrogels15			
11. Circular Dichroism (CD) melting and cooling curves of α-helices				
12	12. Abbreviations			



Figure S1: a. <sup>1</sup>H NMR of ZLY-NCA in CDCl<sub>3:</sub>  $\delta$ = 7.26 ppm (s, CDCl<sub>3</sub>),  $\delta$ = 7.48 ppm (m, 5H, Ph),  $\delta$ = 7.02 ppm (NH)  $\delta$ = 5.12 ppm (s, 2H, PhCH<sub>2</sub>O),  $\delta$ = 4.92 ppm (NH),  $\delta$ = 4.28 ppm(t, 1H, CH),  $\delta$ = 3.22 ppm (q, 2H, CH<sub>2</sub>CH<sub>2</sub>NH),  $\delta$ = 1.90 ppm (m, 2H, CH<sub>2</sub>CH<sub>2</sub>CH),  $\delta$ = 1.51 ppm (m, 4H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>) b. <sup>1</sup>H NMR of BLA-NCA in D<sub>6</sub>DMSO:  $\delta$ = 9.0 ppm (NH),  $\delta$ = 7.37 ppm (m, 5H, Ph),  $\delta$ = 5.13 ppm (s, 2H, PhCH<sub>2</sub>O)  $\delta$ = 4.70 ppm (t, 1H, CH),  $\delta$ = 3.08-2.90 ppm (m, 2H, CH<sub>2</sub>),  $\delta$ = 3.33 ppm (s, H<sub>2</sub>O),  $\delta$ = 2.5 ppm (s, D<sub>6</sub>DMSO)



a.





Figure S2: <sup>1</sup>H NMR of triblock copolymers; a. Z5-10 b. Z20-10, c. Z40-10, and d. A5-10. Peptide repeat length was confirmed via end group analysis of the carbobenzyloxy group,

specifically the benzyl-adjacent protons (peak at ~5.0 ppm) compared to the PEG backbone signal (3.64 ppm).

a. Z5-10, b. Z20-10, and c. Z40-10:  $\delta$ = 7.26 ppm (CDCl<sub>3</sub>),  $\delta$ = 7.34 ppm (m, Ph),  $\delta$ = 5.5 ppm (NH),  $\delta$ = 5.04 ppm (broad s, PhCH<sub>2</sub>O),  $\delta$ = 3.64 ppm (CH<sub>2</sub>CH<sub>2</sub>O),  $\delta$ = 3.12 ppm (m, CH<sub>2</sub>CH<sub>2</sub>CH),  $\delta$ = 2.15 ppm (m, CH<sub>2</sub>CH<sub>2</sub>CH),  $\delta$ = 2.0 (DMAc),  $\delta$ = 1.48 ppm (m, CH<sub>2</sub>CH<sub>2</sub>CH)

d. A5-10  $\delta$ = 7.23 ppm (m, Ph),  $\delta$ = 7.2 ppm (CDCl<sub>3</sub>),  $\delta$ = 5.03 ppm (broad s, PhCH<sub>2</sub>O),  $\delta$ = 2.9 ppm (m, CH),  $\delta$ = 2.0 (s, DMAc)



Figure S3: A5-10 10 wt% in PBS buffer. Solution fails the vial inversion test (*i.e.*, flows under gravity).

$$wt\%(peptide) = 100 \left(\frac{xM_{PZLY}}{xM_{PZLY}+yM_{PEG}+zM_{HDI}}\right)$$

Equation S1: calculated weight fraction of peptide in peptide polyurea hybrids (PPUs), where x, y and z are the molar quantities of the PZLY triblock, PEG and HDI, respectively, and  $M_{ZLY}$ ,  $M_{PEG}$  and  $M_{HDI}$  are the molecular weights of PZLY, PEG and HDI, respectively.

Table S1: Weight average molecular weight, number average molecular weight, and dispersity of PPU hybrids and non-peptidic control calculated from gel permeation chromatography (GPC) using 0.5 wt% LiBr in N,N-dimethyl acetamide (DMAc) used at the mobile phase. Distributions were generated using the calibration curve constructed for six poly(methyl methacrylate) standards (Agilent) in the range of 4.76 to 675.5 kg/mol.

Polymer	Weight-Average Molecular Weight, M <sub>w</sub> (kg mol <sup>-1</sup> )	Number-Average Molecular Weight, M <sub>n</sub> (kg mol <sup>-1</sup> )	Dispersity (M <sub>w</sub> /M <sub>n</sub> )
Z5-10	190	81.5	2.3
Z20-10	192	73.9	2.6
Z40-10	428	105	4.1
A5-10	88.4	42.2	2.0





Figure S4: GPC traces of synthesized PPUs and PEG-PU control.

# **Rheological Characterization**



Figure S5: Frequency sweep of PEG-PU control sample at 10 and 25 wt% polymer in PBS buffer confirming liquid-like behavior. Storage moduli are omitted because of the measurements' low signal-to-noise ratio.



Figure S6. Frequency and amplitude sweeps for PPU hydrogels of varying peptide segment lengths and polymer concentration. Strain sweeps and frequency sweeps were performed at 1 rad/s and 0.5% strain, respectively. All measurements were made at 37 °C.



Figure S7. Simulated injection experiments on Z5-10 hydrogels at a. 5 wt% and b. 10 wt% in PBS. All experiments were performed at 37 °C. Filled and open symbols indicate storage (G') and loss (G'') modulus, respectively.



Figure S8. Simulated injection experiments on Z20-10 hydrogels at a. 5 wt%, b. 10 wt%, and c. 25 wt% in PBS. All experiments were performed at 37 °C. Filled and open symbols indicate G' and G'', respectively.



Figure S9. Simulated injection experiments on Z40-10 hydrogels at 5 wt% in PBS. The experiment was performed at 37 °C. Filled and open symbols indicate G' and G'', respectively.



Figure S10. Recovery after simulated injection of a. Z5-10, b. Z20-10, and c. Z40-10 hydrogels at different concentrations in PBS. Experiment was performed at 37 °C. Moduli have been normalized by the value of the complex modulus at equilibrium  $(G^*_{\infty})$ . Filled and open symbols indicate  $G'/G^*_{\infty}$  and  $G''/G^*_{\infty}$ , respectively.



Figure S11. Temperature sweeps of Z5-10 PPU hydrogels of varying concentrations and measurement frequencies. Panels in the top and bottom rows are for heating and cooling cycles, respectively. Measurements were made at a fixed strain amplitude of 0.1%.



Figure S12. Temperature sweeps of Z20-10 hydrogels of varying concentrations and measurement frequencies. Panels in the top and bottom rows are for heating and cooling cycles, respectively. Measurements were made at a fixed strain amplitude of 0.1%. The cooling cycle for 25 wt% was stopped prematurely because of an experimental error.



Figure S13. Temperature sweeps of Z40-10 hydrogels of varying concentrations and measurement frequencies. Panels in the top and bottom rows are for heating and cooling cycles, respectively. Measurements were made at a fixed strain amplitude of 0.1%.



Figure S14. Quasi gel-point curves for 10 wt% a. Z5-10, b. Z20-10, and c. Z40-10 hydrogels undergoing heating (circles) and cooling (crosses) cycles of a temperature ramp depicting the phase angle ( $\delta$ ). Each curve was measured at a single frequency.



Figure S15: CD melting curves of a. Z5-10 and b. Z40-10 PPU hybrids. As in the Z20-10 sample shown in the main text, only slight melting of the secondary structures is observed, with the  $\alpha$ -helices remaining largely stable up to 80 °C.



Figure S16: CD heating and cooling curves of PPUs monitored at 222 nm showing that slight melting of  $\alpha$ -helices in PPUs is reversible, with some hysteresis being present in the cooling cycle for all samples.

## Abbreviations:

BLA-NCA	$\beta$ -Benzyl-L-aspartate N-carboxyanhydride
CD	Circular dichroism
CDCl <sub>3</sub>	Deuterated chloroform
D <sub>6</sub> DMSO	Deuterated dimethyl sulfoxide
GPC	Gel permeation chromatography
HDI	Hexamethylene diisocyanate
NMR	Nuclear magnetic resonance
PBLA	Poly(β-benzyl-L- aspartate)
PEG	Poly(ethylene glycol)
PPU	Peptide-polyurea
PZLY	Poly(ε-carbobenzyloxy-L-lysine)
SEM	Scanning electron microscopy
Cryo-TEM	Cryogenic-transmission electron microscopy
ZLA-NCA	Carbobenzyloxy-L-lysine N-carboxyanhydride

Final PPU samples are labeled with the nomenclature ZN-X, or AN-X where Z or A refer to PZLY or PBLA, respectively, N denotes the peptide repeat length, and X represents peptide weight percent in the polyurea.