

**Supporting information**

**A versatile core-shell hetero nanostructure catalyzed chemo-selective synthesis of  $\beta$ -enamino carbonyl compounds**

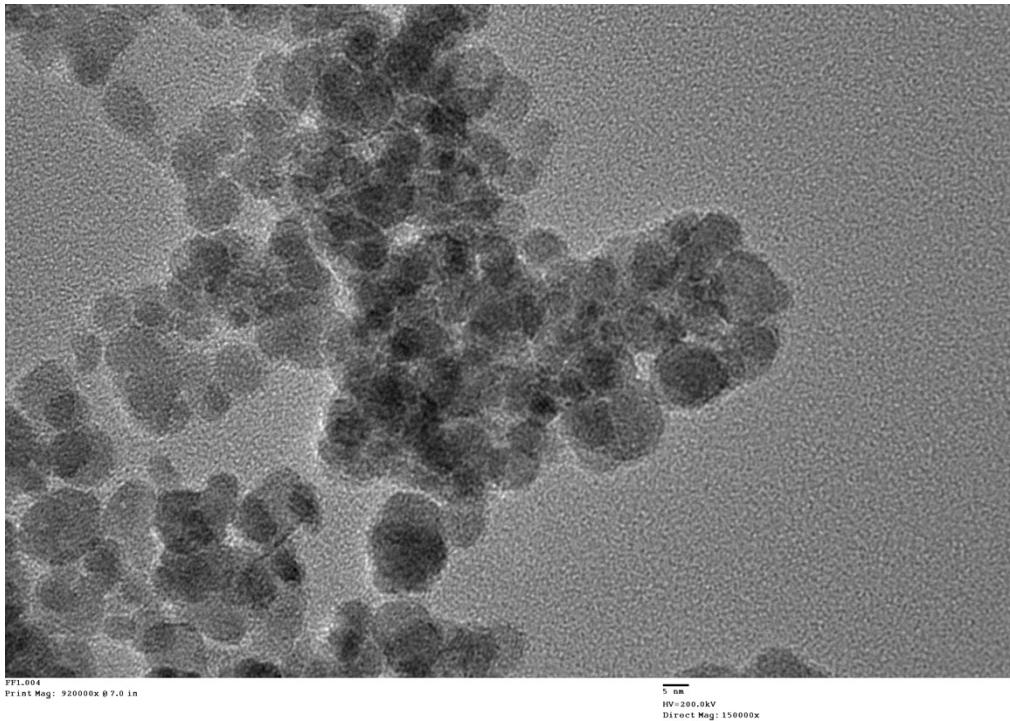
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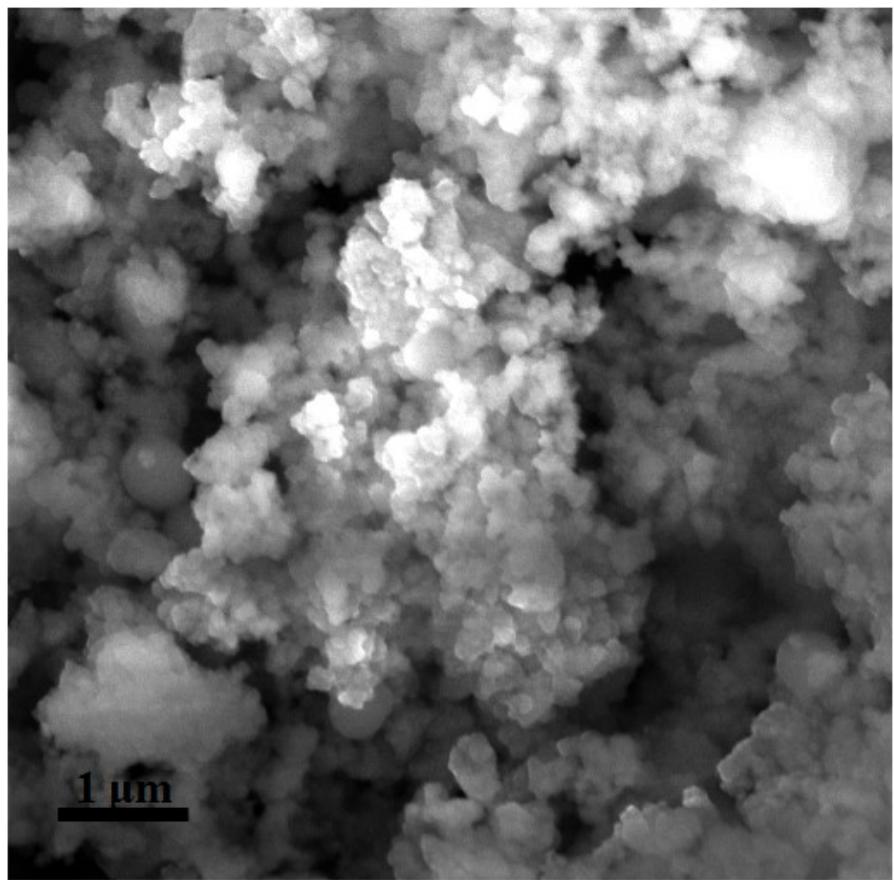
<sup>b</sup>*Hindu College, University of Delhi, Delhi- 110007, India*

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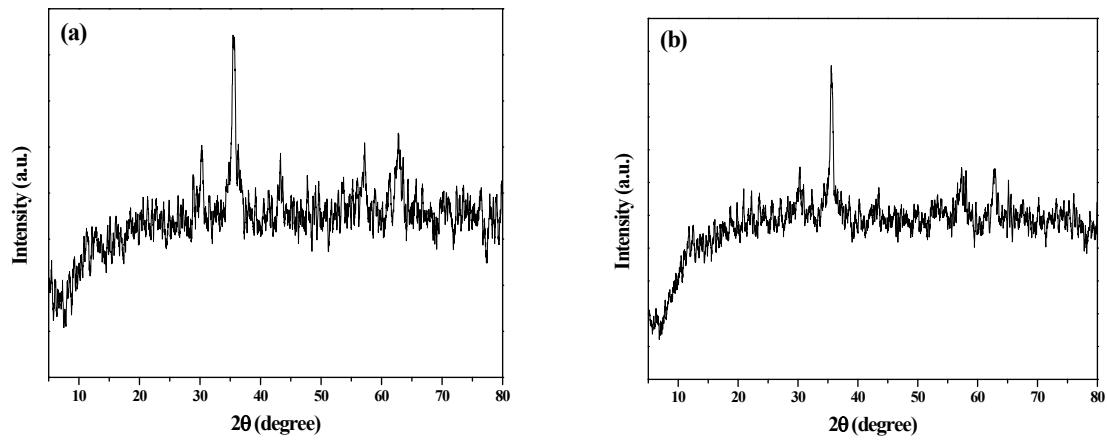
<sup>d</sup>*Department of Chemistry, Ramjas College, University of Delhi, Delhi- 110007, India*



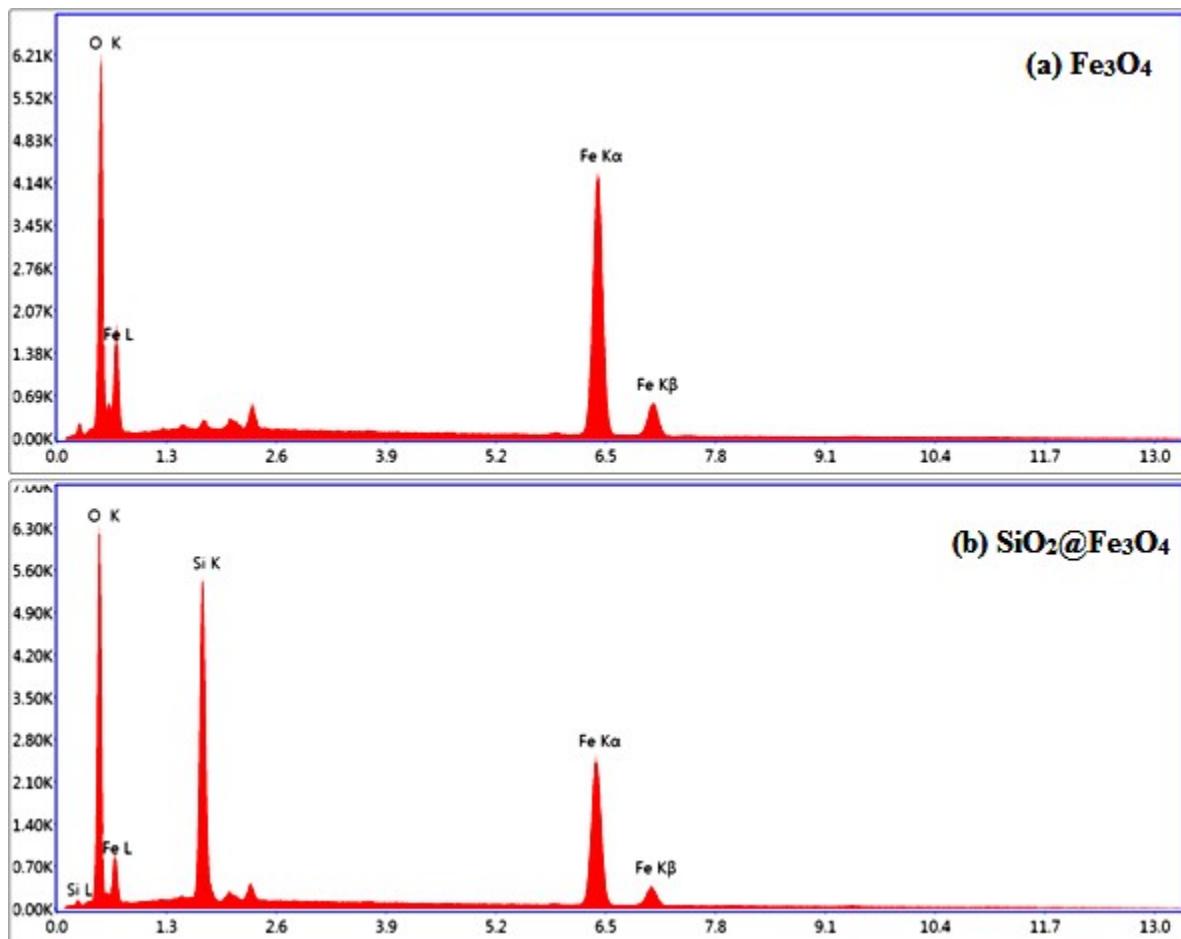
**Figure S1:** TEM image of silica coated magnetite nanoparticles



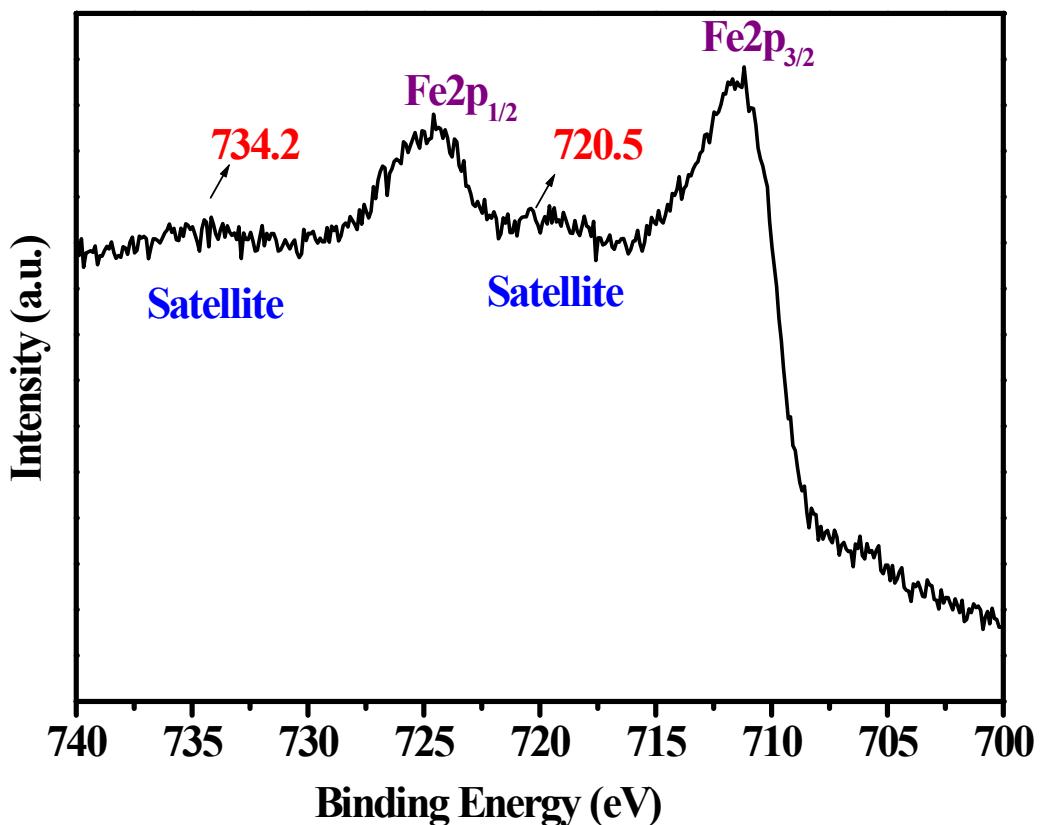
**Figure S2:** SEM Image of recovered catalyst.



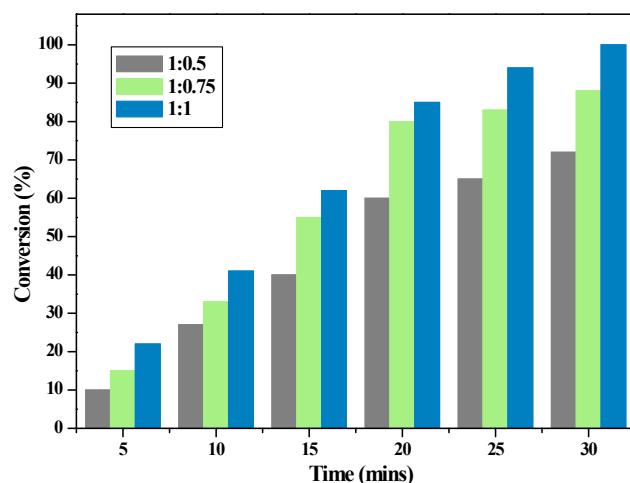
**Figure S3:** XRD Spectrum of (a) Fresh and (b) Recovered catalyst



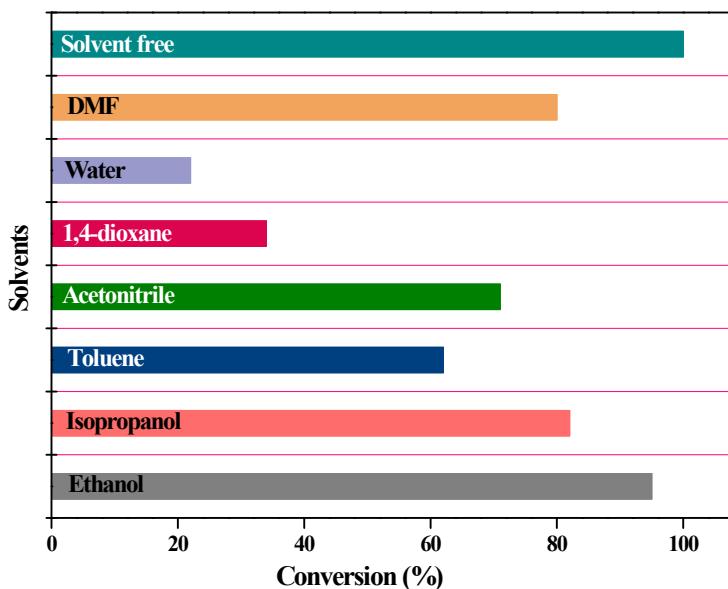
**Figure S4:** EDX spectra of a)  $\text{Fe}_3\text{O}_4$  and b) $\text{SiO}_2@\text{Fe}_3\text{O}_4$  nanoparticles.



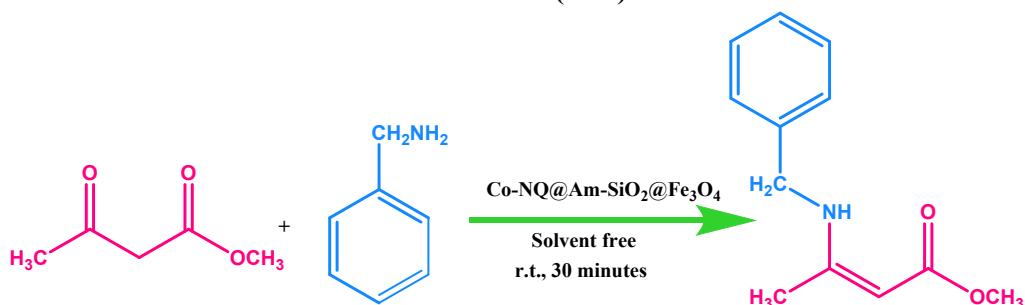
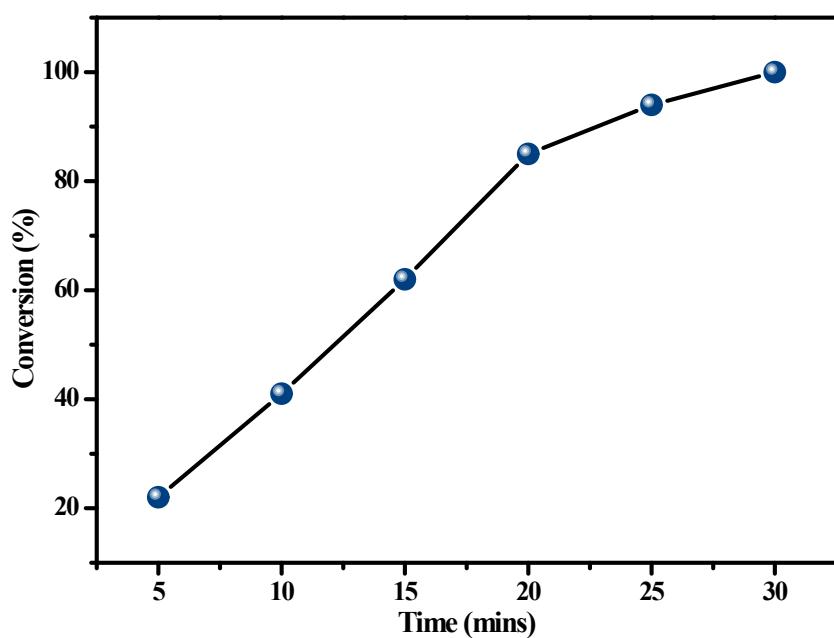
**Figure S5:** XPS Spectrum of Fe2p of catalyst.



**Figure S6:** Effect of substrate molar ratio for the enamination reaction [Reaction conditions: benzylamine (a mmol), methylacetoacetate (b mmol), Co-NQ@Am-SiO<sub>2</sub>@Fe<sub>3</sub>O<sub>4</sub> (20 mg), solvent free conditions, r.t. (25 °C)].



**Figure S7:** Effect of solvent on the enamination reaction [Reaction Conditions: benzylamine (1 mmol), methylacetoacetate (1 mmol), Co-NQ@Am-SiO<sub>2</sub>@Fe<sub>3</sub>O<sub>4</sub> catalyst (20 mg), solvent (2 mL), stirring at r.t. (25 °C)]



**Figure S8:** Effect of time on Co-NQ@Am-SiO<sub>2</sub>@Fe<sub>3</sub>O<sub>4</sub> catalyzed enamination reaction [Reaction conditions: benzylamine (1 mmol), methylacetoacetate (1 mmol), Co-NQ@Am-SiO<sub>2</sub>@Fe<sub>3</sub>O<sub>4</sub> catalyst (20 mg), stirring at r.t. (25 °C)].

**Table S1:** Effect of temperature on the Co-NQ@Am-SiO<sub>2</sub>@Fe<sub>3</sub>O<sub>4</sub> catalyzed enamination reaction.

Entry	Temperature	Conversion (%)
1.	25	100
2.	35	98
3.	45	94
4.	55	92
5.	65	89

[Reaction Conditions: Benzylamine (1 mmol), Methylacetoacetate (1 mmol), Co-NQ@Am-SiO<sub>2</sub>@Fe<sub>3</sub>O<sub>4</sub> catalyst (20 mg), stirring].