

Supplementary Material

Computational Discovery of Diverse Functionalities in Two-Dimensional Square Disulfide Monolayers: Auxetic Behavior, High Curie Temperature Ferromagnets, Electrocatalysts, and Photocatalysts

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Table S1. The structural parameters and cohesive energies of 68 S-XS₂ monolayers (the italics are the data of the corresponding hexagonal structure, and the bolds in E_{coh} are the values of square structure larger than the corresponding hexagonal structure), where ‘*’ indicates the cases where the optimized structures are not square.

S-XS₂	a = b (Å)	h (Å)	θ (°)	r_{X-S} (Å)	E_{coh} (eV/atom)
S-HS ₂	2.50	2.98	79.82	1.94	2.29
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<i>S</i> -LiS ₂	2.60	4.00	65.91	2.38	2.90
	<i>2.57/2.86</i>	<i>3.92</i>	<i>102.16</i>	<i>2.52</i>	<i>2.74</i>
<i>S</i> -NaS ₂	2.70	4.66	60.03	2.69	2.39
	<i>2.88</i>	<i>4.50</i>	<i>107.17</i>	<i>2.80</i>	<i>2.24</i>
<i>S</i> -KS ₂	2.81	5.28	56.01	2.99	2.04
	<i>2.51/4.77</i>	<i>3.69</i>	<i>72.65</i>	<i>3.12</i>	<i>2.23</i>
<i>S</i> -RbS ₂	2.94	5.54	55.97	3.13	1.83
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<i>S</i> -CsS ₂	5.00	4.19	100.12	3.26	1.22
	<i>5.37</i>	<i>2.02</i>	<i>36.05</i>	<i>3.26</i>	<i>2.60</i>
<i>S</i> -BeS ₂	2.81	3.07	84.94	2.08	3.58
	<i>2.97</i>	<i>2.94</i>	<i>81.10</i>	<i>2.25</i>	<i>3.22</i>
<i>S</i> -MgS ₂	3.05	3.78	77.67	2.43	2.84
	<i>3.14</i>	<i>3.64</i>	<i>90.19</i>	<i>2.57</i>	<i>2.80</i>
<i>S</i> -CaS ₂	3.32	4.21	76.54	2.68	2.96
	<i>3.44</i>	<i>3.97</i>	<i>89.96</i>	<i>2.81</i>	<i>3.10</i>
<i>S</i> -SrS ₂	3.59	4.39	78.52	2.84	2.69
	<i>4.76</i>	<i>2.15</i>	<i>42.79</i>	<i>2.95</i>	<i>3.91</i>
<i>S</i> -BaS ₂	4.13	4.19	89.19	2.94	2.65
	<i>4.99</i>	<i>2.15</i>	<i>40.90</i>	<i>3.07</i>	<i>3.94</i>
<i>S</i> -BS ₂	2.96	2.42	101.52	1.91	4.59
	<i>4.17/2.96</i>	<i>3.20</i>	<i>64.41</i>	<i>1.91/3.52</i>	<i>4.59</i>
<i>S</i> -AlS ₂	3.40	2.97	97.71	2.26	3.94
	<i>3.45</i>	<i>2.76</i>	<i>69.41</i>	<i>2.43</i>	<i>3.74</i>
<i>S</i> -GaS ₂	3.43	3.00	97.50	2.28	3.35
	<i>3.46</i>	<i>2.91</i>	<i>72.03</i>	<i>2.47</i>	<i>3.04</i>
<i>S</i> -InS ₂	3.66	3.31	95.70	2.47	3.01
	<i>3.81</i>	<i>2.94</i>	<i>67.48</i>	<i>2.64</i>	<i>2.93</i>
<i>S</i> -TlS ₂	3.76	3.40	95.72	2.54	2.50
	<i>3.91</i>	<i>3.04</i>	<i>67.94</i>	<i>2.72</i>	<i>2.36</i>
<i>S</i> -CS ₂	2.97	2.30	104.53	1.88	4.03
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<i>S</i> -SiS ₂	3.38	2.70	102.67	2.16	4.27
	<i>3.24</i>	<i>2.91</i>	<i>75.80</i>	<i>2.37</i>	<i>3.77</i>
<i>S</i> -GeS ₂	3.51	2.85	101.90	2.26	3.68
	<i>3.38</i>	<i>3.07</i>	<i>76.38</i>	<i>2.48</i>	<i>3.24</i>
<i>S</i> -SnS ₂	3.78	3.08	101.77	2.44	3.51
	<i>3.61</i>	<i>3.26</i>	<i>76.12</i>	<i>2.64</i>	<i>2.71</i>
<i>S</i> -PbS ₂	3.94	3.20	101.77	2.54	3.05
	<i>3.76</i>	<i>3.36</i>	<i>75.47</i>	<i>2.75</i>	<i>2.87</i>

<i>S</i> -NS ₂	2.73	2.82	88.06	1.96	2.62
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*S-PS ₂	3.57/3.63	2.39	111.28	2.16	3.23
	<i>3.16</i>	<i>3.12</i>	<i>81.06</i>	<i>2.40</i>	<i>2.94</i>
<i>S</i> -AsS ₂	3.82	2.60	111.57	2.31	2.97
	<i>3.33</i>	<i>3.33</i>	<i>81.66</i>	<i>2.54</i>	<i>2.80</i>
<i>S</i> -SbS ₂	4.11	2.80	111.60	2.49	2.97
	<i>3.61</i>	<i>3.41</i>	<i>78.63</i>	<i>2.69</i>	<i>2.91</i>
<i>S</i> -BiS ₂	4.20	3.04	108.26	2.59	2.83
	<i>3.77</i>	<i>3.50</i>	<i>77.53</i>	<i>2.79</i>	<i>2.81</i>
*S-OS ₂	--	--	--	--	--
<i>S</i> -SeS ₂	3.87	2.78	108.58	2.38	2.46
	<i>3.45</i>	<i>3.28</i>	<i>78.98</i>	<i>2.58</i>	<i>2.48</i>
<i>S</i> -TeS ₂	4.37	2.45	121.50	2.50	2.63
	<i>3.71</i>	<i>3.39</i>	<i>76.79</i>	<i>2.73</i>	<i>2.65</i>
<i>S</i> -PoS ₂	4.12	3.29	102.85	2.64	2.68
	<i>3.83</i>	<i>3.43</i>	<i>75.60</i>	<i>2.80</i>	<i>2.78</i>
*S-FS ₂	--	--	--	--	--
<i>S</i> -ClS ₂	4.50	1.03	154.17	2.31	1.49
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<i>S</i> -BrS ₂	4.53	1.70	138.77	2.42	1.51
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<i>S</i> -IS ₂	4.68	1.98	134.05	2.54	1.74
	<i>3.70</i>	<i>3.60</i>	<i>80.13</i>	<i>2.80</i>	<i>1.52</i>
*S-HeS ₂	2.31/2.46	7.64	35.67	4.01	0.61
*S-NeS ₂	2.49/2.33	7.77	32.81	4.13	1.78
*S-ArS ₂	2.69/2.55	8.32	34.06	4.35	1.27
*S-KrS ₂	2.69/2.86	8.14	38.78	4.30	0.90
*S-XeS ₂	--	--	--	--	--
<i>S</i> -ScS ₂	3.68	3.22	97.62	2.44	4.44
	<i>3.79</i>	<i>2.72</i>	<i>63.68</i>	<i>2.58</i>	<i>4.69</i>
<i>S</i> -TiS ₂	3.68	2.71	107.15	2.29	5.00
	<i>3.35</i>	<i>3.02</i>	<i>75.97</i>	<i>2.45</i>	<i>5.10</i>
<i>S</i> -VS ₂	3.53	2.64	107.72	2.21	4.53
	<i>3.17</i>	<i>2.97</i>	<i>78.07</i>	<i>2.36</i>	<i>4.82</i>
<i>S</i> -CrS ₂	3.50	2.48	109.38	2.15	3.78
	<i>3.04</i>	<i>2.94</i>	<i>79.82</i>	<i>2.29</i>	<i>4.17</i>
<i>S</i> -MnS ₂	3.44	2.44	109.44	2.11	3.65
	<i>3.11</i>	<i>2.77</i>	<i>75.37</i>	<i>2.27</i>	<i>3.84</i>
<i>S</i> -FeS ₂	3.46	2.33	112.15	2.08	3.89
	<i>3.15</i>	<i>2.66</i>	<i>72.35</i>	<i>2.25</i>	<i>3.99</i>
<i>S</i> -CoS ₂	3.52	2.23	115.29	2.08	3.86
	<i>3.23</i>	<i>2.51</i>	<i>67.73</i>	<i>2.25</i>	<i>3.98</i>
<i>S</i> -NiS ₂	3.67	2.05	121.57	2.10	3.71
	<i>3.55</i>	<i>2.11</i>	<i>54.42</i>	<i>2.30</i>	<i>3.93</i>
<i>S</i> -CuS ₂	2.66	3.82	69.79	2.33	3.08
	<i>3.72</i>	<i>2.14</i>	<i>52.95</i>	<i>2.40</i>	<i>3.13</i>
*S-ZnS ₂	2.55/3.72	3.44	105.19	2.34	2.55

<i>S</i> -YS ₂	3.94	3.44	98.55	2.60	4.44
	4.07	2.82	61.93	2.74	4.82
<i>S</i>-ZrS₂	3.93	2.87	107.67	2.43	5.45
	3.58	3.15	74.59	2.60	5.57
<i>S</i> -NbS ₂	3.72	2.81	105.87	2.33	5.39
	3.35	3.12	77.88	2.48	5.80
<i>S</i>-MoS₂	3.71	2.61	109.73	2.27	4.51
	3.18	3.13	80.78	2.41	5.41
<i>S</i> -TcS ₂	3.58	2.65	106.98	2.23	4.65
	3.29	2.90	74.66	2.39	4.90
<i>S</i> -RuS ₂	3.63	2.51	110.63	2.21	4.57
	3.34	2.80	71.98	2.38	4.65
<i>S</i> -RhS ₂	3.67	2.46	112.27	2.21	4.08
	3.34	2.86	72.82	2.32	4.16
<i>S</i> -PdS ₂	2.70	4.16	65.91	2.48	3.03
	3.92	2.07	49.07	2.48	3.32
<i>S</i> -AgS ₂	2.71	4.53	61.73	2.64	2.53
	3.41	3.22	78.70	2.54	2.31
<i>S</i> -CdS ₂	3.09	4.06	74.60	2.55	2.09
	5.12/4.29	2.16	39.74	3.37/2.47	2.49
<i>S</i> -HfS ₂	3.82	2.91	105.41	2.40	5.52
	3.54	3.13	74.93	2.57	5.66
<i>S</i> -TaS ₂	3.69	2.83	105.05	2.33	5.46
	3.34	3.13	78.04	2.48	5.91
<i>S</i> -WS ₂	3.53	2.88	101.51	2.28	5.20
	3.19	3.14	80.93	2.42	5.91
<i>S</i>-ReS₂	3.63	2.61	108.63	2.24	4.76
	3.32	2.88	73.77	2.40	4.85
<i>S</i>-OsS₂	3.67	2.47	112.03	2.21	4.84
	3.38	2.77	70.88	3.39	5.01
<i>S</i> -IrS ₂	3.77	2.34	116.41	2.22	4.53
	3.36	2.88	72.83	2.32	4.43
<i>S</i> -PtS ₂	3.54	2.84	102.52	2.27	3.72
	3.48	2.79	69.64	2.45	3.75
<i>S</i> -AuS ₂	3.38	3.34	90.73	2.37	2.46
	4.70/3.41	3.23	46.72	4.08/4.07	2.86
* <i>S</i> -HgS ₂	4.05/2.71	3.91	61.11	2.67	1.68

Table S2. The elastic constants (in Nm⁻¹), Young's modulus (Y , in Nm⁻¹) and Poisson's ratio (ν) of the 37 thermodynamically and dynamically stable $S\text{-}X\text{S}_2$ structures, including 27 mechanically stable monolayers and 10 mechanically unstable monolayers. Negative Poisson's ratios are indicated in bold within the data, while relevant data for two SiS_2 structures obtained by the CALYPSO code are represented in italics.

	C_{11}	C_{22}	C_{12}	C_{66}	Y_x	Y_y	ν_x	ν_y
$S\text{-BS}_2$	102.62	102.62	27.83	20.47	95.08	95.08	0.271	0.271
$S\text{-AlS}_2$	47.97	47.97	2.91	4.77	47.79	47.79	0.061	0.061
$S\text{-GaS}_2$	45.58	45.58	4.24	6.69	45.19	45.19	0.093	0.093
$S\text{-InS}_2$	33.21	33.21	3.47	2.70	32.85	32.85	0.104	0.104
$S\text{-TlS}_2$	24.02	24.02	2.93	3.34	23.85	23.85	0.121	0.121
$S\text{-SiS}_2$	96.37	96.37	-5.18	2.13	96.09	96.09	-0.054	-0.054
$SiS_2\text{-}2$	<i>97.11</i>	<i>91.33</i>	-2.76	<i>7.04</i>	<i>81.41</i>	<i>76.57</i>	-0.030	-0.028
$SiS_2\text{-}3$	<i>115.12</i>	<i>115.12</i>	24.97	<i>45.07</i>	<i>109.70</i>	<i>109.70</i>	0.217	0.217
$S\text{-GeS}_2$	75.55	75.55	-4.73	1.72	75.26	75.26	-0.063	-0.063
$S\text{-SnS}_2$	51.39	51.39	-4.27	0.98	51.04	51.04	-0.083	-0.083
$S\text{-PbS}_2$	39.49	39.49	-2.45	0.09	39.34	39.34	-0.062	-0.062
$S\text{-TiS}_2$	31.38	31.38	0.51	5.05	31.37	31.37	0.016	0.016
$S\text{-VS}_2$	28.55	28.55	3.38	7.49	28.15	28.15	0.118	0.118
$S\text{-CrS}_2$	40.30	40.30	10.91	11.31	37.35	37.35	0.271	0.271
$S\text{-MnS}_2$	42.33	42.33	10.85	10.85	39.55	39.55	0.256	0.256
$S\text{-FeS}_2$	12.65	12.65	-7.20	6.52	8.55	8.55	-0.569	-0.569
$S\text{-CoS}_2$	20.67	20.67	11.32	3.69	14.47	14.47	0.548	0.548
$S\text{-ZrS}_2$	29.70	29.70	0.44	1.78	29.69	29.69	0.015	0.015
$S\text{-NbS}_2$	22.19	22.19	15.42	7.03	11.48	11.48	0.695	0.695
$S\text{-MoS}_2$	44.10	44.10	12.71	8.54	40.44	40.44	0.288	0.288
$S\text{-RuS}_2$	39.25	39.25	15.13	2.45	33.41	33.41	0.385	0.385
$S\text{-RhS}_2$	13.98	13.98	-6.20	3.90	11.23	11.23	-0.443	-0.443
$S\text{-AgS}_2$	145.48	145.48	-11.17	0.31	144.62	144.62	-0.077	-0.077
$S\text{-HfS}_2$	37.67	37.67	3.75	1.57	37.30	37.30	0.100	0.100
$S\text{-TaS}_2$	27.43	27.43	20.19	8.59	12.57	12.57	0.736	0.736
$S\text{-WS}_2$	22.66	22.66	19.85	13.86	5.27	5.27	0.876	0.876
$S\text{-ReS}_2$	29.96	29.96	11.82	9.71	25.30	25.30	0.395	0.395
$S\text{-OsS}_2$	37.47	37.47	15.20	2.66	31.30	31.30	0.406	0.406
$S\text{-IrS}_2$	34.27	34.27	6.90	3.57	32.88	32.88	0.201	0.201
Mechanically unstable								
$S\text{-BeS}_2$	52.22	52.22	53.49	8.99	--	--	--	--
$S\text{-AsS}_2$	-131.99	-131.99	189.29	2.51	--	--	--	--
$S\text{-SbS}_2$	-43.02	-43.02	82.43	1.36	--	--	--	--
$S\text{-BiS}_2$	8.07	8.07	25.83	0.63	--	--	--	--
$S\text{-NiS}_2$	-19.25	-19.25	-45.40	6.65	--	--	--	--
$S\text{-CuS}_2$	-29.58	-29.58	-3.50	4.63	--	--	--	--
$S\text{-TcS}_2$	15.42	15.42	15.83	9.42	--	--	--	--
$S\text{-CdS}_2$	-4.37	-4.37	53.28	4.26	--	--	--	--
$S\text{-PtS}_2$	-34.02	-34.02	-25.91	-3.46	--	--	--	--
$S\text{-AuS}_2$	22.16	22.16	12.80	-1.54	--	--	--	--

$C_{11} = C_{22}$.

Young's moduli $Y_x = Y_y = (C_{11}C_{22} - C_{12}C_{21})/C_{11}$

Poisson's ratio $\nu_x = \nu_y = C_{12}/C_{11}$

Table S3. Possible stable structures predicted by CALYPSO code arranged in descending order based on cohesive energy (E_{coh}).

number	E_{coh} (eV/atom)	space group	top-view	side-view
1	4.27	<i>P</i> -4 <i>m</i> 2		
2	4.26	<i>C</i> 222		
3	4.24	<i>P</i> -3 <i>m</i> 1		
4	4.17	<i>C</i> 2		

Table S4. Bader and Hirshfeld (in italic) charge analysis of $S\text{-}X\text{S}_2$ ($X = \text{Si}, \text{Ge}, \text{Sn}, \text{Pb}, \text{Ti}, \text{V}, \text{Cr}, \text{Mn}, \text{Zr}, \text{Mo}, \text{Re}, \text{Os}$) monolayers (in $|e|$).

$S\text{-}X\text{S}_2$	X	S	$S\text{-}X\text{S}_2$	X	S
<i>S</i> -SiS ₂	+2.31/+0.26	-1.15/-0.13	<i>S</i> -CrS ₂	+1.10/+0.28	-0.55/-0.14
<i>S</i> -GeS ₂	+1.24/+0.28	-0.62/-0.14	<i>S</i> -MnS ₂	+0.94/+0.02	-0.47/-0.01
<i>S</i> -SnS ₂	+1.44/+0.41	-0.72/-0.20	<i>S</i> -ZrS ₂	+1.97/+0.38	-0.99/-0.19
<i>S</i> -PbS ₂	+1.04/+0.46	-0.52/-0.23	<i>S</i> -MoS ₂	+1.14/+0.30	-0.57/-0.15
<i>S</i> -TiS ₂	+1.69/+0.28	-0.84/-0.14	<i>S</i> -ReS ₂	+0.91/+0.06	-0.46/-0.03
<i>S</i> -VS ₂	+1.43/+0.17	-0.72/-0.09	<i>S</i> -OsS ₂	+0.61/+0.04	-0.31/-0.02

Table S5. Relative energies of antiferromagnetic (AFM) and ferromagnetic (FM) states relative to non-magnetic (NM) state in a $2\times 2\times 1$ supercell of $S\text{-}X\text{S}_2$ ($X = \text{Re}, \text{Os}, \text{Si}, \text{Ge}, \text{Sn}, \text{Pb}, \text{Zr}, \text{Ti}, \text{Mo}, \text{V}, \text{Cr}, \text{and Mn}$) monolayers. Bold highlighting indicates monolayers with the lowest energies. The primary cell lattice constants (a) of the magnetic ground state of the $S\text{-}X\text{S}_2$ ($X = \text{Mo}, \text{V}, \text{Cr}, \text{and Mn}$) monolayers.

	$E_{AFM1} - E_{NM}$ (eV)	$E_{AFM2} - E_{NM}$ (eV)	$E_{FM} - E_{NM}$ (eV)	a (Å)
$S\text{-}\text{ReS}_2$	0.00	0.00	0.00	--
$S\text{-}\text{OsS}_2$	0.00	0.00	0.00	--
$S\text{-}\text{SiS}_2$	0.00	0.00	0.00	--
$S\text{-}\text{GeS}_2$	0.00	0.00	0.00	--
$S\text{-}\text{SnS}_2$	0.00	0.00	0.00	--
$S\text{-}\text{PbS}_2$	0.00	0.00	0.00	--
$S\text{-}\text{ZrS}_2$	0.00	0.00	0.00	--
$S\text{-}\text{TiS}_2$	0.00	0.00	0.00	--
$S\text{-}\text{MoS}_2$	-0.91	0.00	-0.04	7.41
$S\text{-}\text{VS}_2$	-1.70	-1.54	-1.96	3.65
$S\text{-}\text{CrS}_2$	-4.44	-3.61	-5.43	3.66
$S\text{-}\text{MnS}_2$	-3.28	-2.88	-3.79	3.75

Table S6. ΔG_{H^*} values (in eV) for H adsorption at different sites on the same side of $S\text{-}XS_2$ ($X = \text{Re, Os, Cr, and Mn}$) monolayers. Certain initial sites transform to more stable geometries after relaxation. Values closest to zero are highlighted in bold.

$S\text{-}XS_2$	S1	S2	S3	S4	S5	S6	S7	S8
$S\text{-}\text{ReS}_2$	0.72	S1	0.78	S5	0.17	S7	0.95	0.95
$S\text{-}\text{OsS}_2$	0.81	S1	0.93	S5	0.46	S7	0.47	0.48
$S\text{-}\text{CrS}_2$	0.15	S1	S1	S8	S8	S8	S8	-0.37
$S\text{-}\text{MnS}_2$	-0.11	S1	S1	S8	S8	S7	-0.12	-0.49

Table S7. ΔG_{H^*} values (in eV) for H adsorption at various sites on the antiferromagnetic (AFM1) $S\text{-}\text{MoS}_2$ monolayer. Certain initial sites transform to more stable geometries after relaxation. Values closest to zero are highlighted in bold.

$S\text{-}XS_2$	S1	S2	S3	S4	S5	S6
$S\text{-}\text{MoS}_2$	S8	S8	0.95	S5	0.36	S8
S7	S8	S9	S10	S11	S12	S13
S8	0.23	S8	0.36	S10	0.95	S1

Table S8. ΔG_{H^*} values (in eV) of the $S\text{-}XS_2$ monolayers at various H coverage (θ) conditions. The data in brackets represent the ΔG_{H^*} values of the corresponding magnetic ground state after H adsorption.

H coverage (θ)	$S\text{-}\text{ReS}_2$	$S\text{-}\text{OsS}_2$	$S\text{-}\text{CrS}_2$	$S\text{-}\text{MnS}_2$	$S\text{-}\text{MoS}_2$
1/4	0.17	0.46	-0.37	-0.49	0.23 (0.37)
1/9	-0.06	0.03	-0.38	-0.50	--
1/16	-0.18	-0.09	-0.41	-0.52	0.09 (0.32)

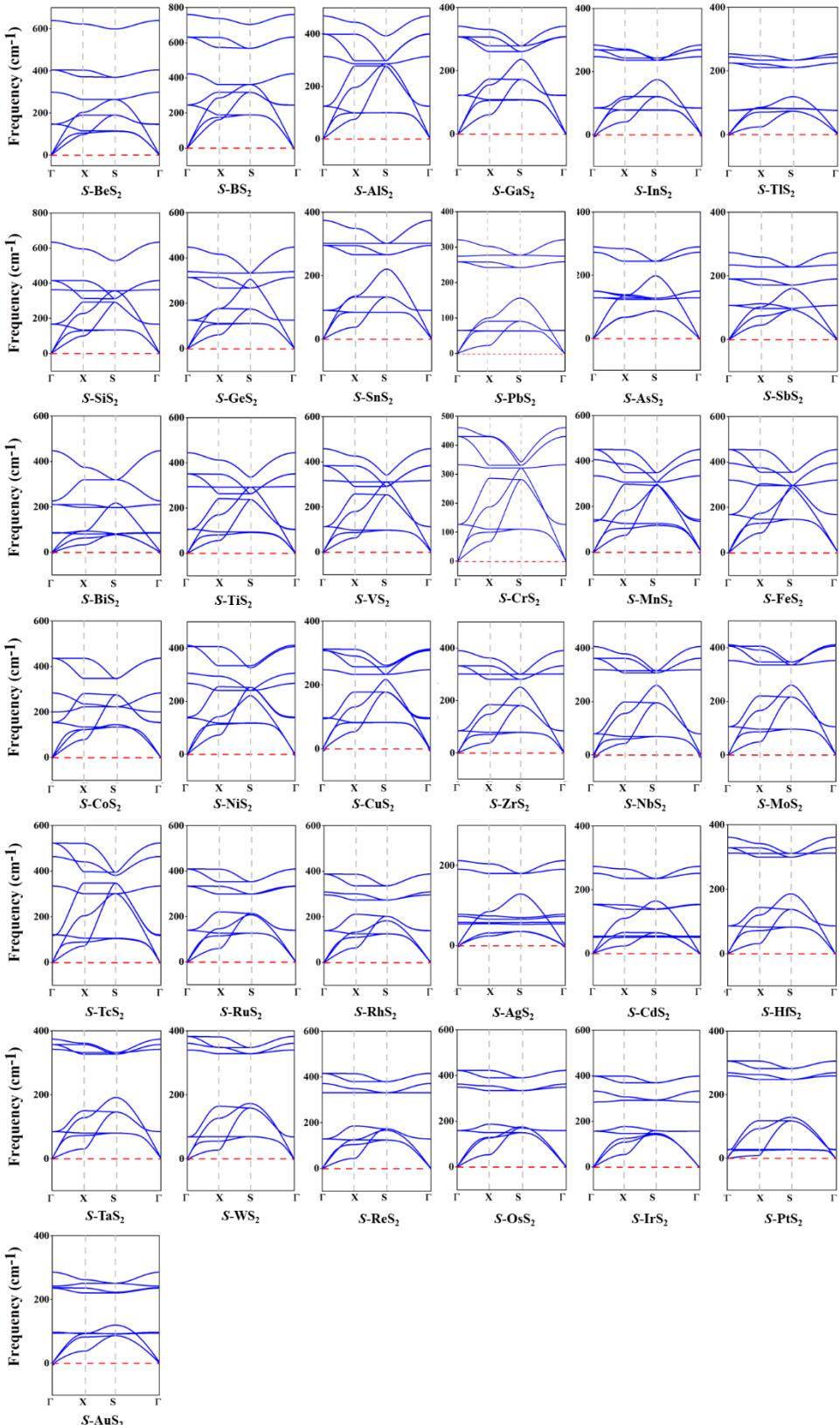


Fig. S1 Calculated phonon dispersion curves of the 37 dynamically stable $S\text{-}X\text{S}_2$ monolayers.

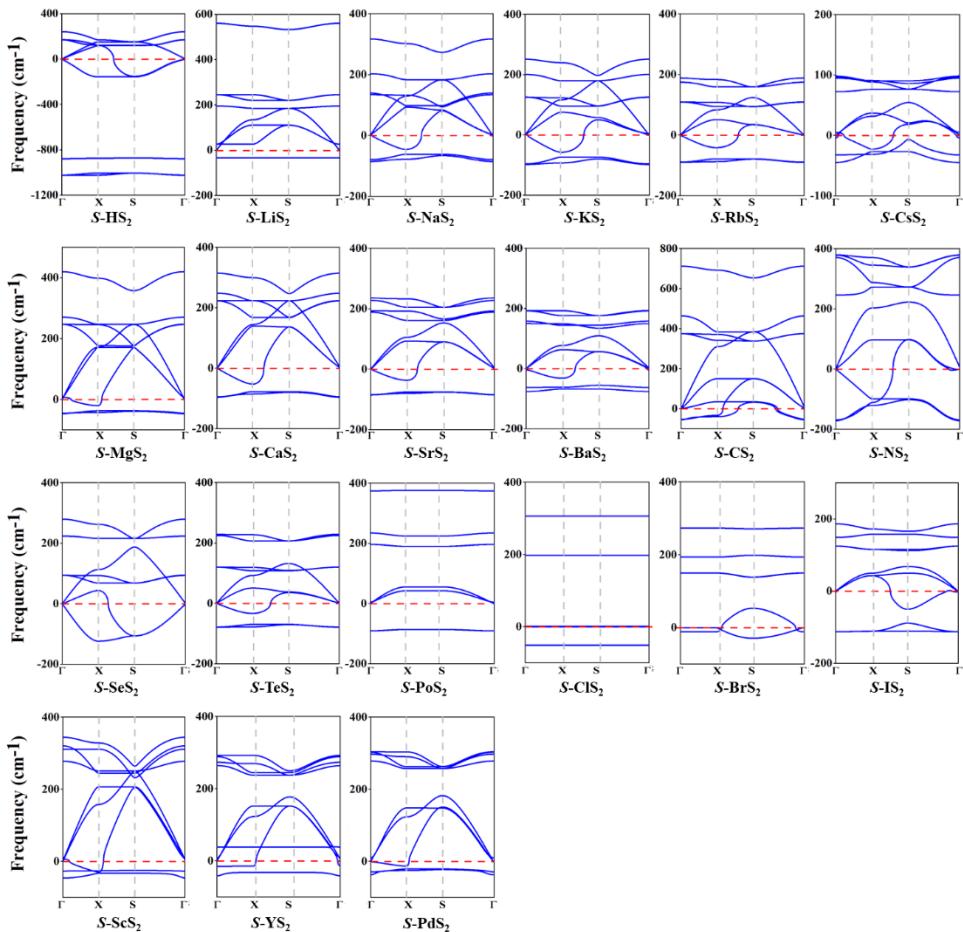


Fig. S2 Calculated phonon dispersion curves of 21 $S\text{-}XS_2$ monolayers with significant imaginary frequencies.

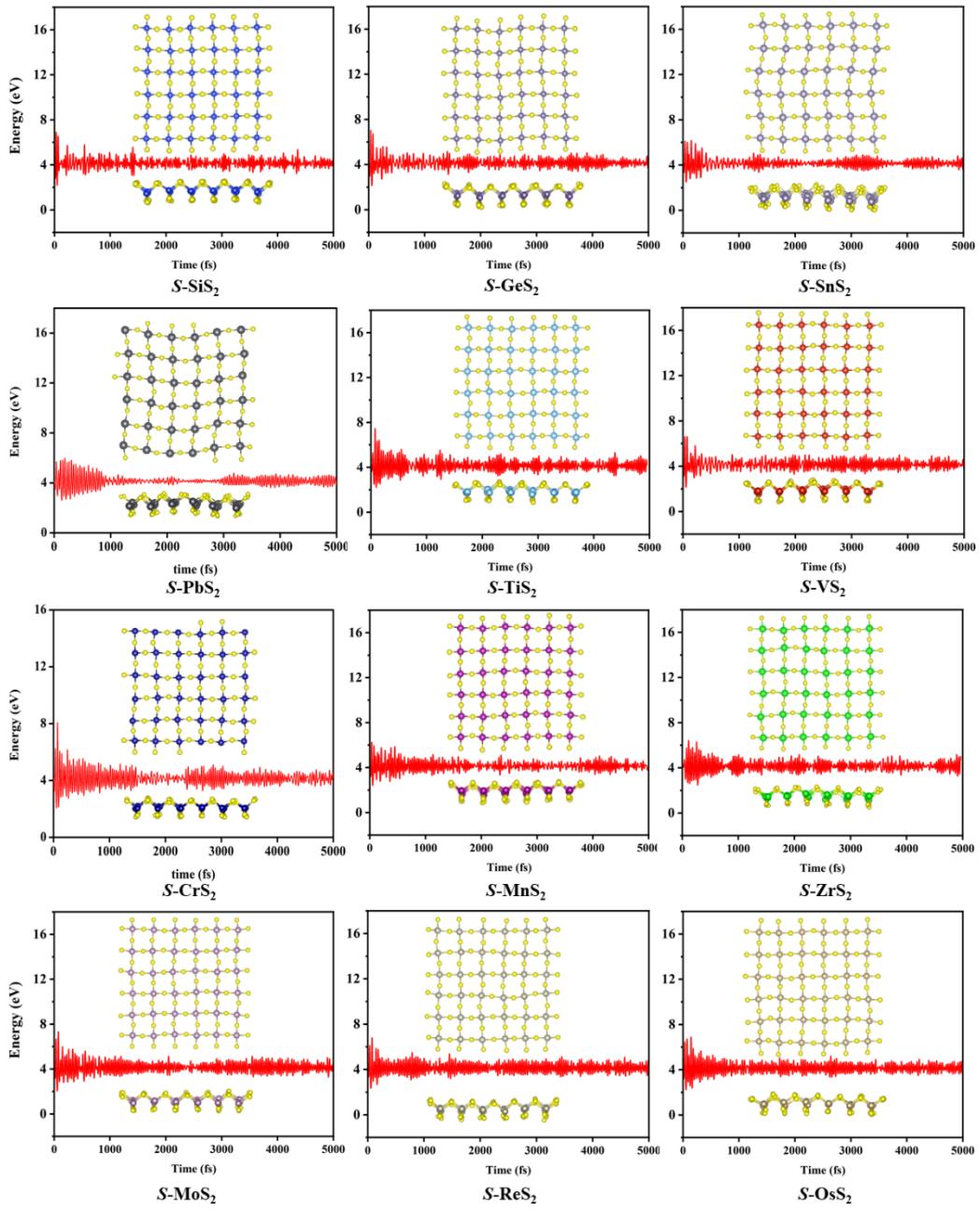


Fig. S3 Final structure and energy fluctuation of 12 thermally stable $S\text{-}XS_2$ monolayers upon 5 ps FPMD simulations at 300 K.

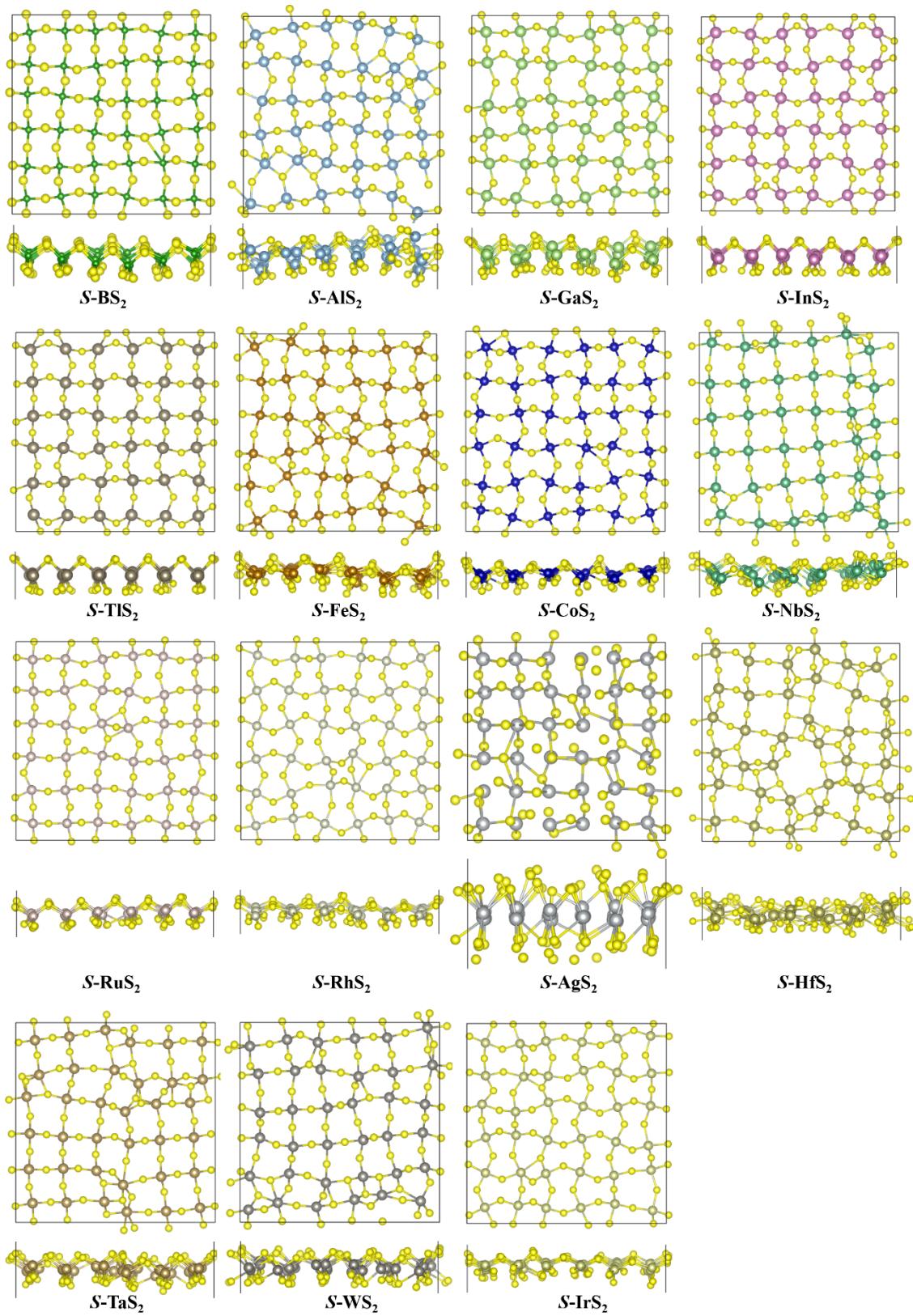


Fig. S4 Final structures of 15 thermally unstable $S\text{-}XS_2$ monolayers after 5 ps FPMD simulations at 300 K.

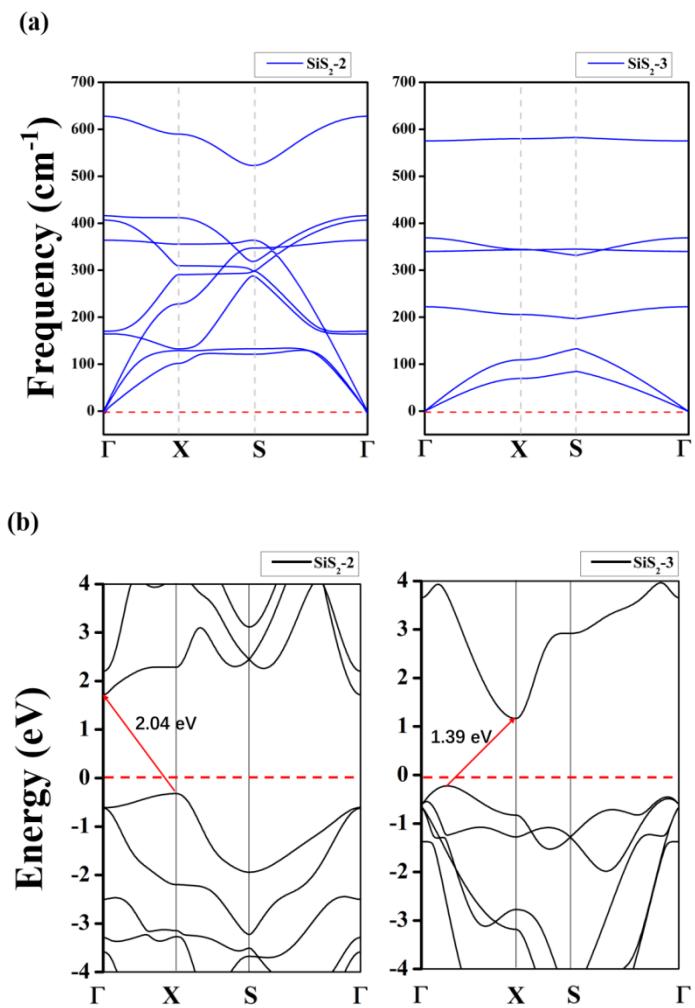


Fig. S5 (a) Phonon spectra and (b) band structures of $\text{SiS}_2\text{-}2$ and $\text{SiS}_2\text{-}3$.

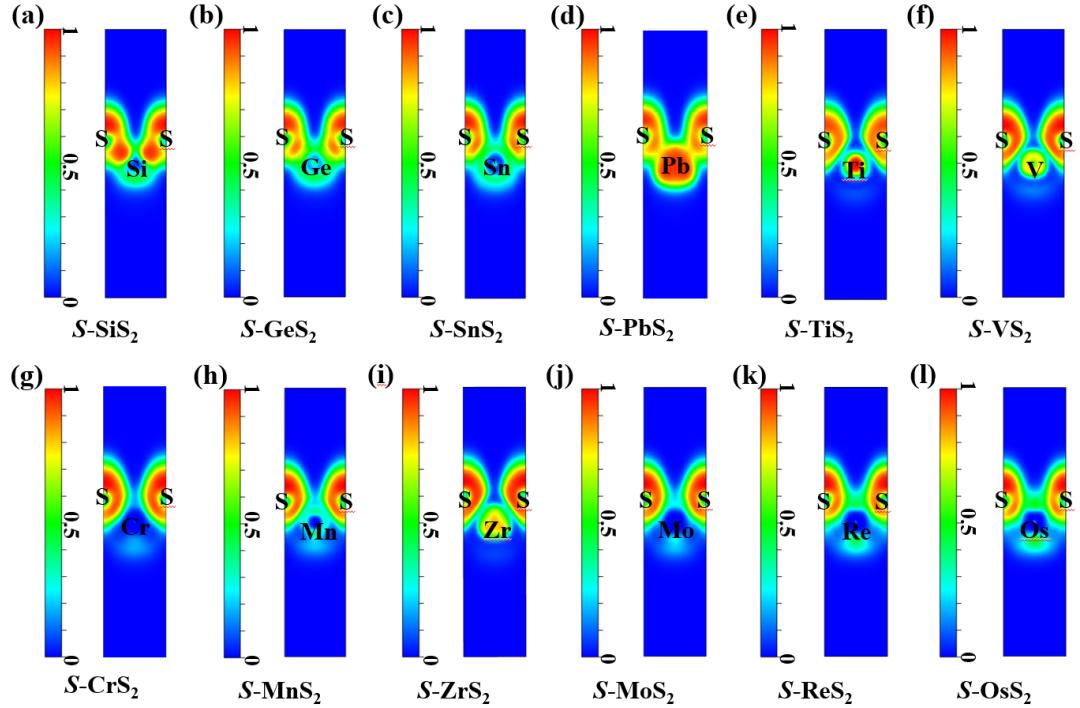


Fig. S6 ELF maps of (010) slice for the square monolayer. ELF = 1 (red) and 0 (blue) indicate the accumulated and vanished electron densities, respectively. (a) $S\text{-SiS}_2$, (b) $S\text{-GeS}_2$, (c) $S\text{-SnS}_2$, (d) $S\text{-PbS}_2$, (e) $S\text{-TiS}_2$, (f) $S\text{-VS}_2$, (g) $S\text{-CrS}_2$, (h) $S\text{-MnS}_2$, (i) $S\text{-ZrS}_2$, (j) $S\text{-MoS}_2$, (k) $S\text{-ReS}_2$, and (l) $S\text{-OsS}_2$.

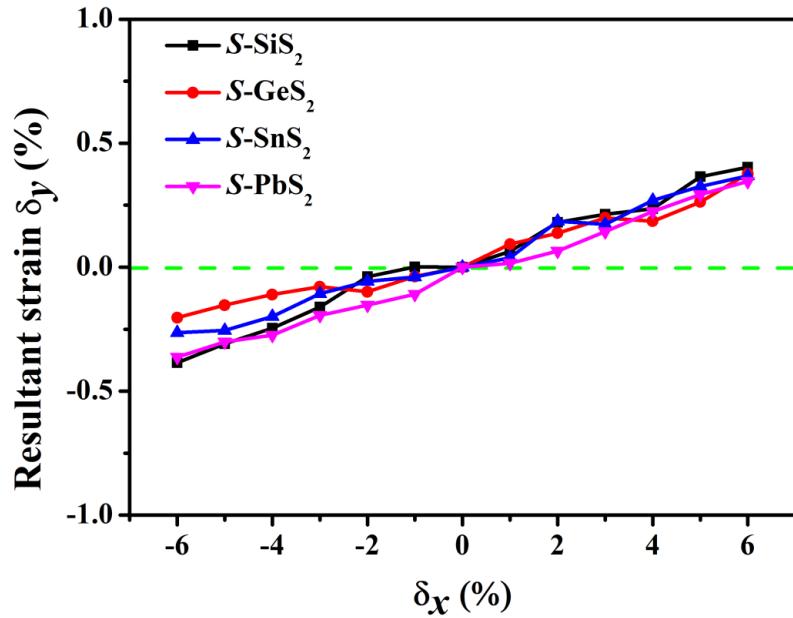


Fig. S7 Mechanical response of the $S\text{-XS}_2$ (X = Si, Ge, Sn, and Pb) monolayers under the uniaxial (ranging from -6% to 6%) strain along the x -direction.

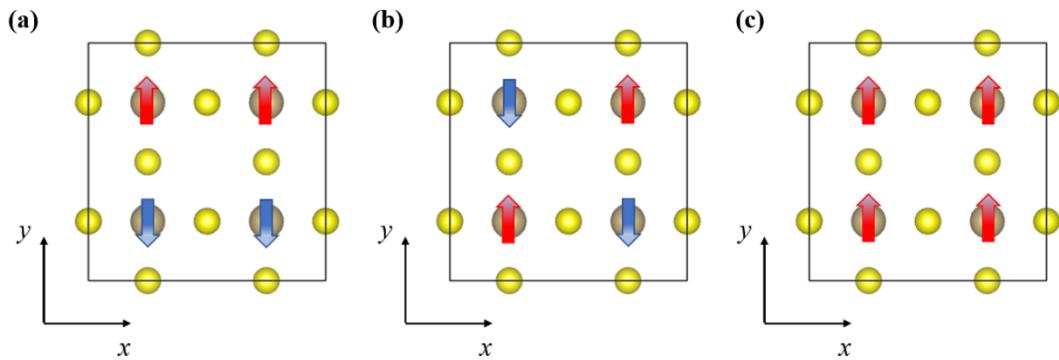


Fig. S8 Schematic illustration of three different magnetic orders of $S\text{-XS}_2$ monolayer.

(a) AFM1. (b) AFM2. (c) FM.

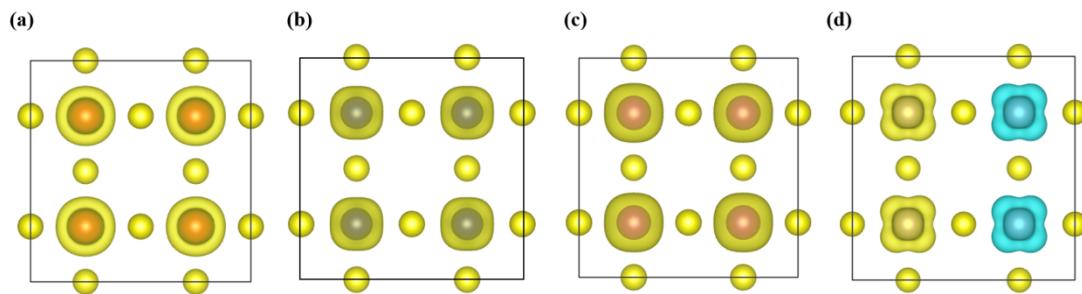


Fig. S9 Magnetism distribution of metal atoms in $S\text{-VS}_2$, $S\text{-MnS}_2$ and $S\text{-MoS}_2$ monolayers. The isosurface values were set as $0.02 \text{ e } \text{\AA}^{-3}$. (a) V atom in $S\text{-VS}_2$ ($1.39 \mu_B$) monolayer. (b) Cr atom in $S\text{-VS}_2$ ($2.91 \mu_B$) monolayer. (c) Mn atom in $S\text{-MnS}_2$ ($3.28 \mu_B$) monolayer. (d) Mo atom in $S\text{-MoS}_2$ ($\pm 1.46 \mu_B$) monolayer.

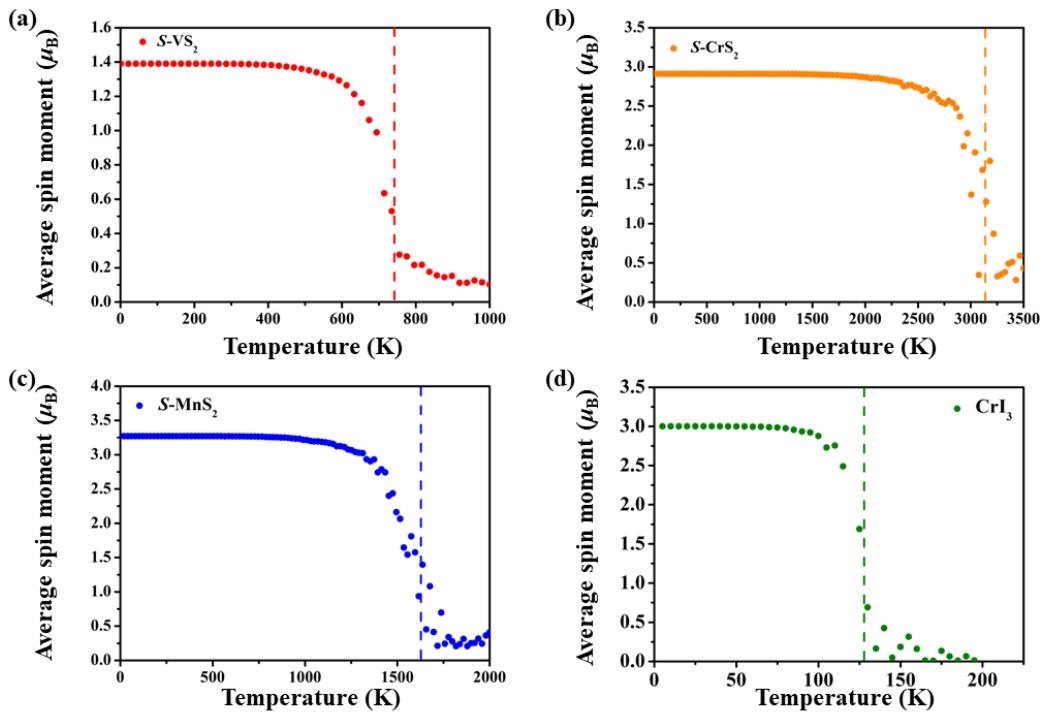


Fig. S10 On-site magnetic moments of transition metal atoms as a function of temperature from MC simulations. (a) $S\text{-VS}_2$. (b) $S\text{-CrS}_2$. (c) $S\text{-MnS}_2$. (d) CrI_3 .

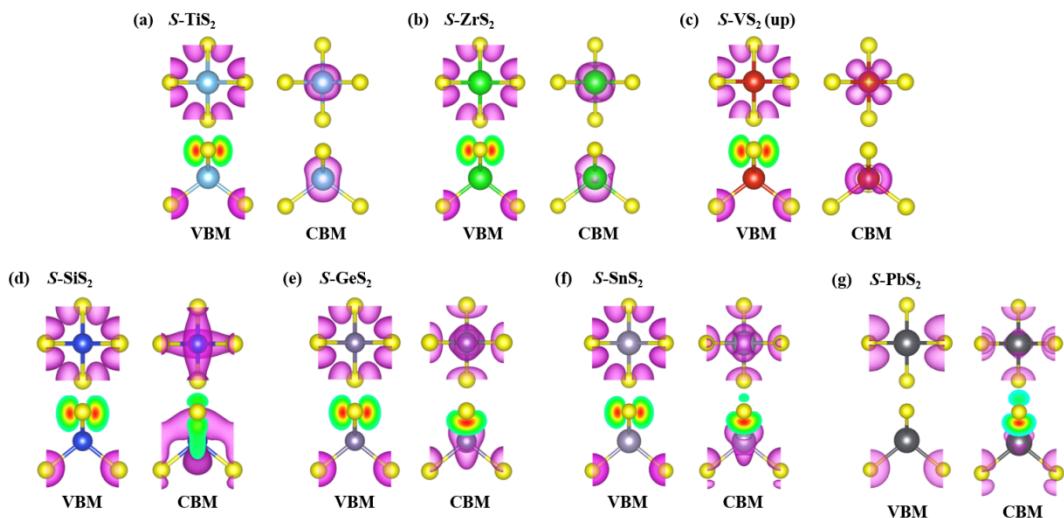


Fig. S11 Partial charge distribution of VBM and CBM of $S\text{-XS}_2$ monolayers. (a) $S\text{-TiS}_2$. (b) $S\text{-ZrS}_2$. (c) $S\text{-VS}_2$ (spin-up channel). (d) $S\text{-SiS}_2$. (e) $S\text{-GeS}_2$. (f) $S\text{-SnS}_2$. (g) $S\text{-PbS}_2$. The isosurface values were set as $0.01 \text{ e } \text{\AA}^{-3}$.

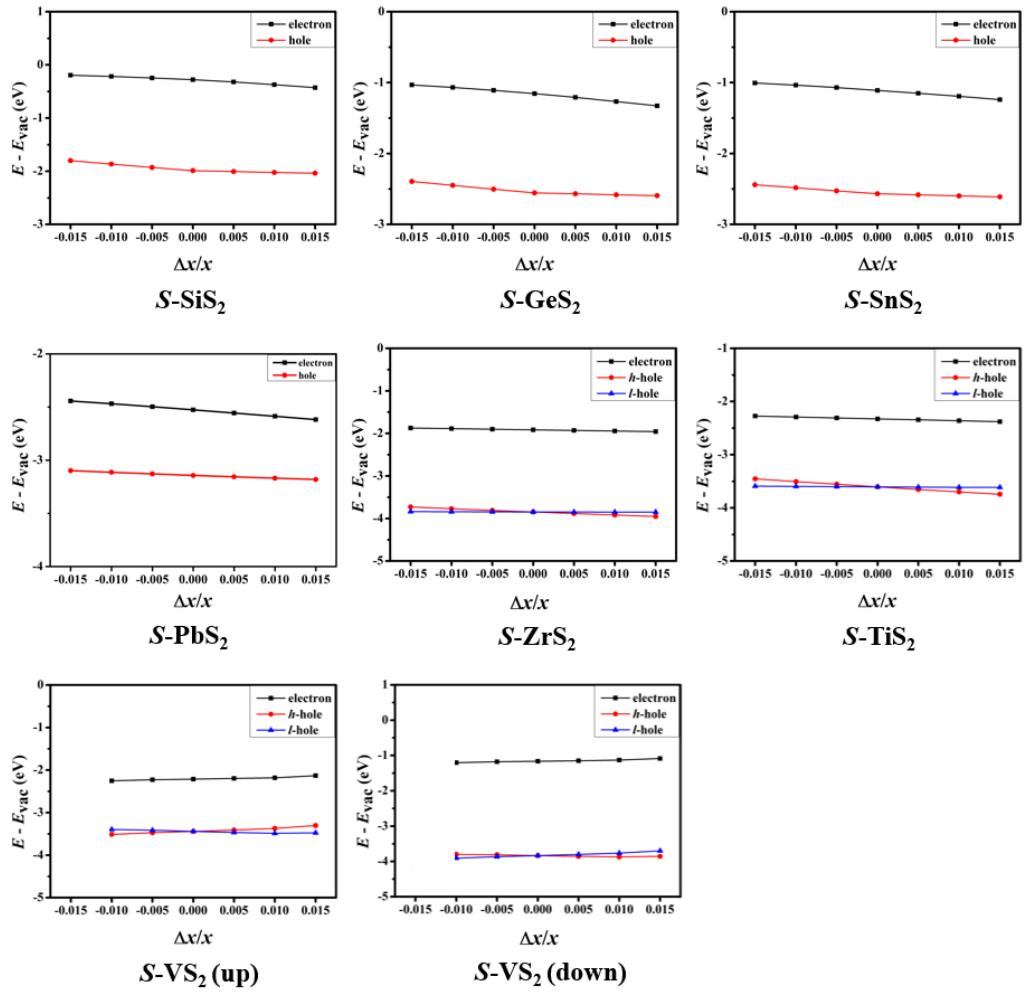


Fig. S12 Fitting curves of deformation potentials of $S\text{-}X\text{S}_2$ ($X = \text{Si}, \text{Ge}, \text{Sn}, \text{Pb}, \text{Zr}, \text{Ti}$, and V (spin-up and spin-down)) monolayers. The strain ($\Delta x/x$) is ranging from -0.015 to 0.015 with the steps of 0.005 .

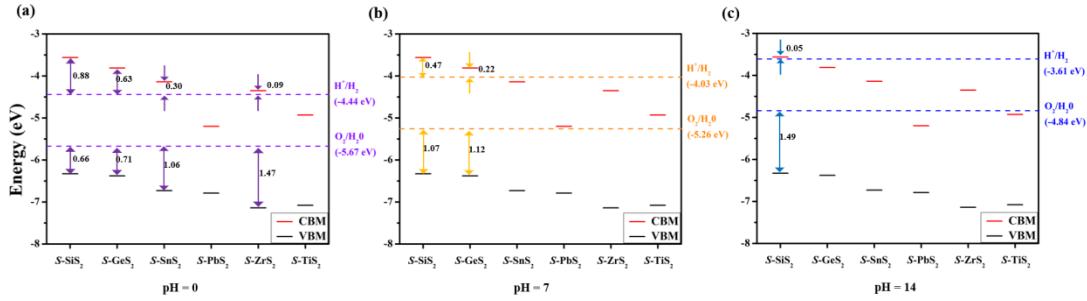


Fig. S13 CBM and VBM energies of S-XS₂ (X = Si, Ge, Sn, Pb, Zr, and Ti) monolayers. Redox potentials of water splitting at (a) pH = 0 (purple dashed lines), (b) pH = 7 (orange dashed lines), and (c) pH = 14 (blue dashed lines).

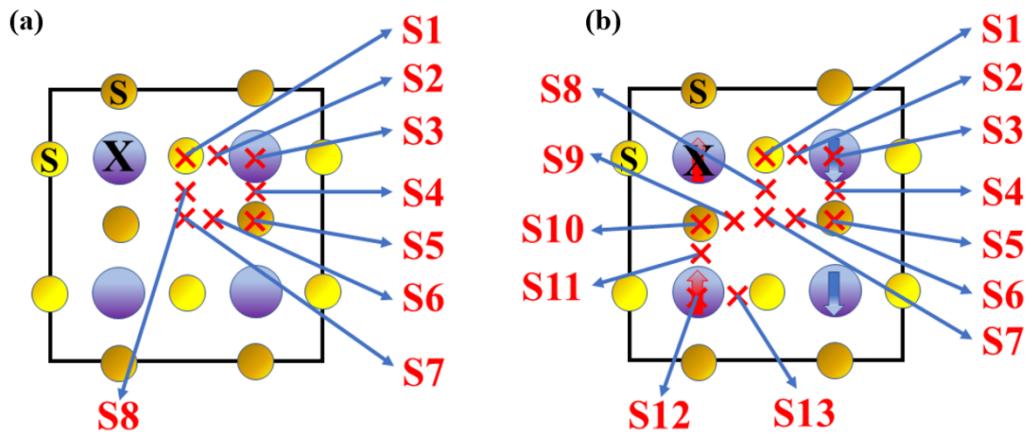
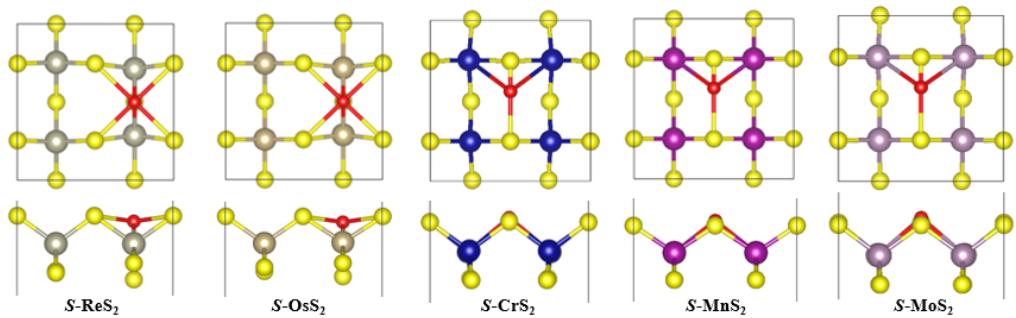
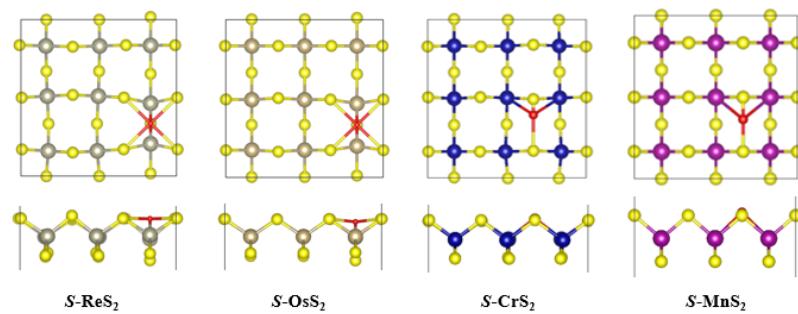


Fig. S14 (a) Possible H adsorption sites on the nonmagnetic (NM) S-ReS₂, S-OsS₂ and ferromagnetic (FM) S-MnS₂ monolayers with a 2×2×1 supercell. (b) Possible H adsorption sites on the antiferromagnetic (AFM1) S-MoS₂ monolayer, the red/blue arrow indicate the spin up/down magnetic configuration of Mo. Purple and dark/light yellow balls represent the X and S atoms of bottom/top layer, respectively.

(a) 1/4 H coverage



(b) 1/9 H coverage



(c) 1/16 H coverage

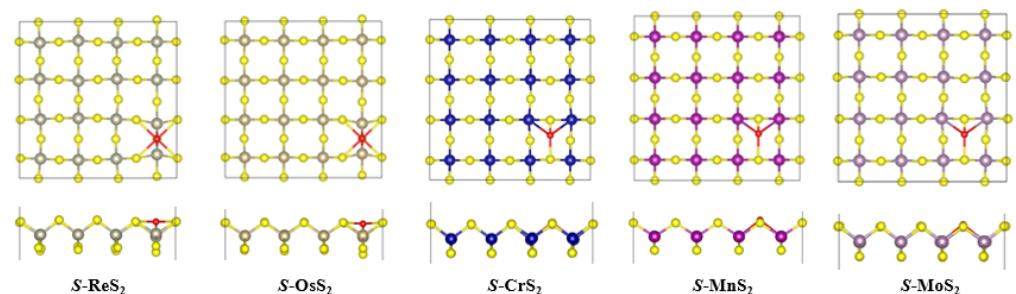


Fig. S15 Optimized structures of the energetically most favorable H adsorption sites on the S-XS₂ (X = Re, Os, Cr, Mo, and Mn) monolayer with H coverage of (a) 1/4, (b) 1/9 and (c) 1/16.