

Supporting Information

Promotional effect on multiple active sites in Fe-based oxygen reduction electrocatalysts for zinc-air battery

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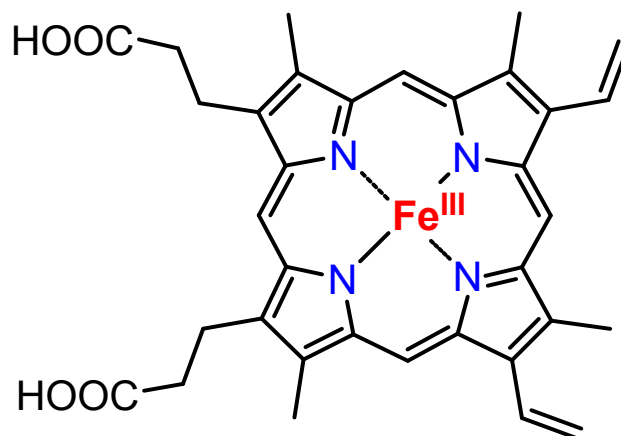
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Scheme S1. The molecular structure of Hemin, which is a Fe-centered natural macrocyclic compound.

II. Supporting figures

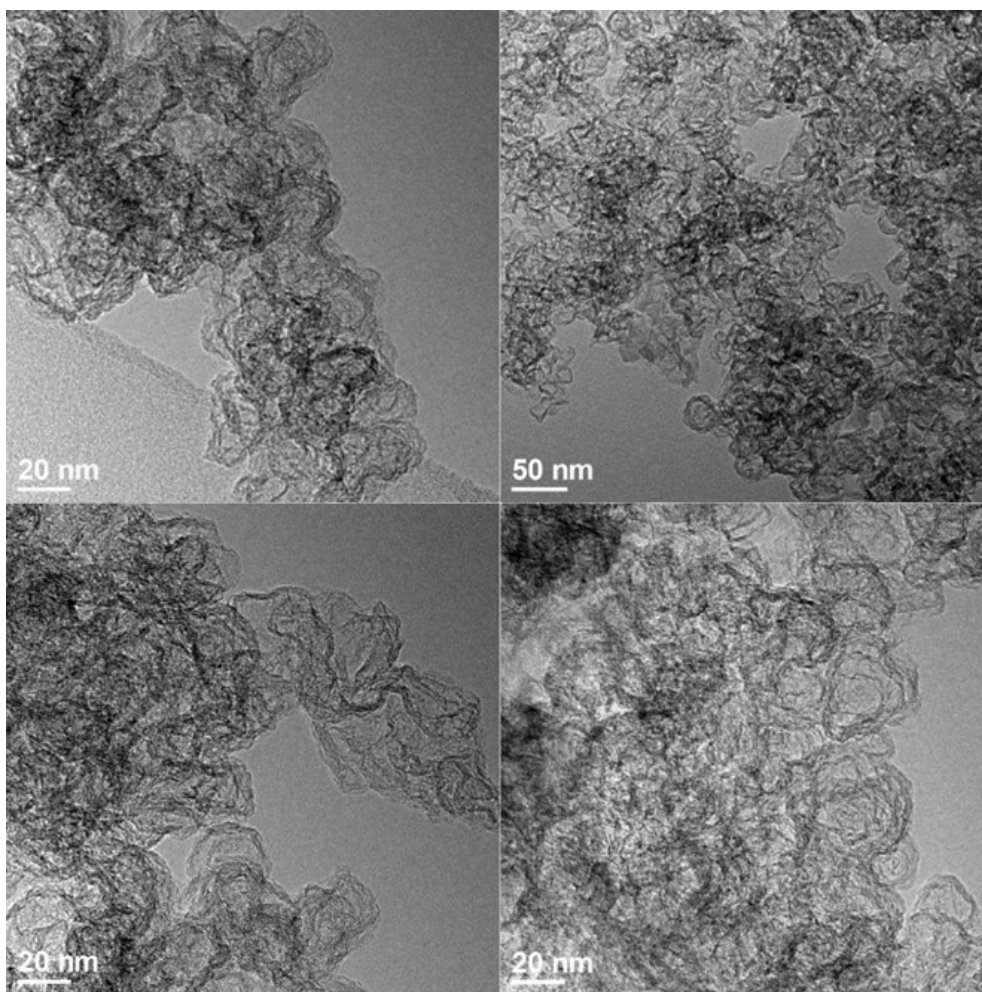


Fig. S1. TEM images of Hm/Cy@C-900.

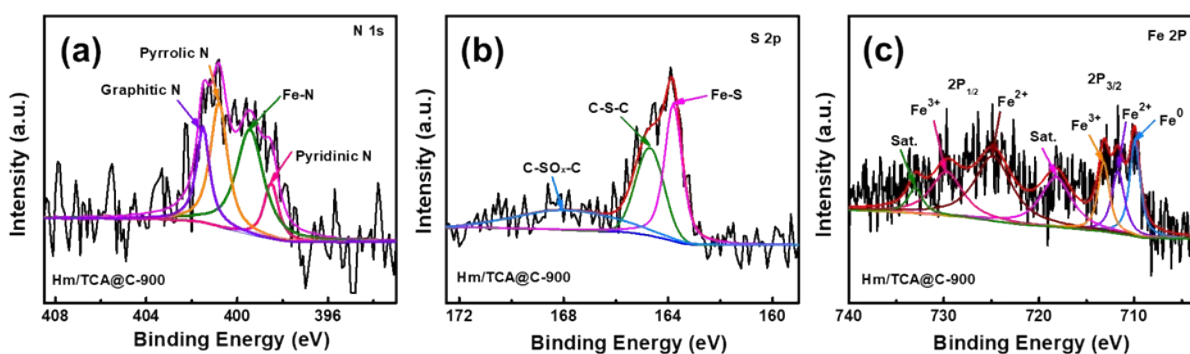


Fig. S2. XPS spectra of Hm/TCA@C-900 sample: (a) N 1s, (b) S 2p, and (c) Fe 2p.

In view of the Fe 2p XPS spectra in Fig. S2c, the Fe²⁺ and Fe³⁺ species and metallic Fe⁰ particles were found, meaning the presence of metallic Fe, Fe₃C, FeN_x, FeS_x, or Fe₃O₄ species in the electrocatalysts.

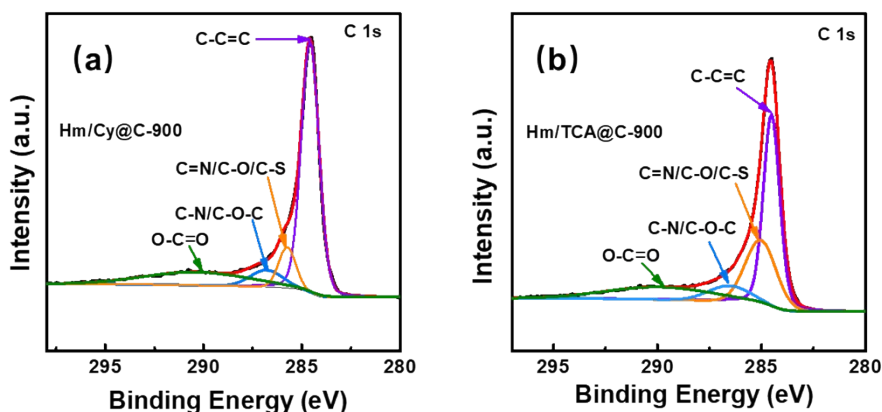


Fig. S3. C 1s XPS spectra of (a) Hm/Cy@C-900 and (b) Hm/TCA@C-900 samples.

C 1s XPS was deconvoluted into four peaks, locating at 284.6, 285.7, 286.8, and 290.3 eV, matching with the C-C=C, C=N/C-O/C-S, C-N/C-O-C, and O-C=O species, respectively.^[1,2] The prominent peak at 284.6 eV, attributed to sp^2 -hybridized C, suggests a high degree of graphitization.

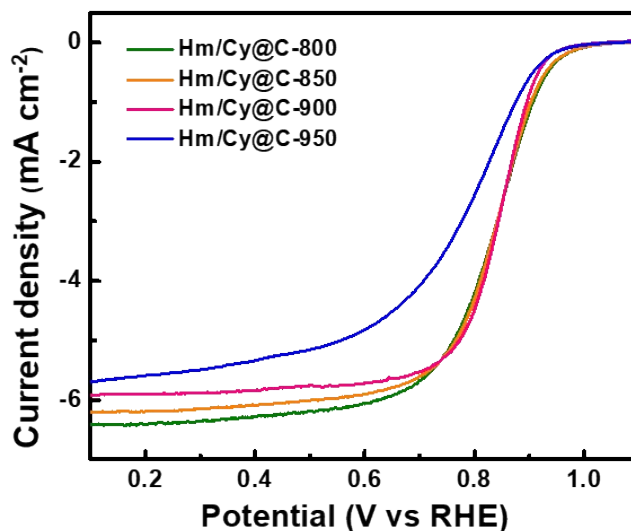


Fig. S4. ORR polarization curves of Hm/Cy@C-T (T: 800, 850, 900 and 950 °C) at 5 mV s^{-1} in O_2 -saturated 0.1 M KOH solution. Thus, Hm/Cy@C-900 exhibited the best ORR activity, based on the value of $E_{1/2}$.

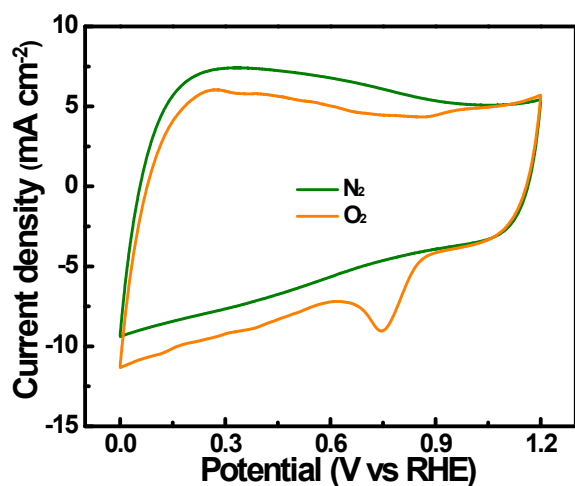


Fig. S5. CV curves in O₂- or N₂-saturated 0.1 M KOH solution of Hm/Cy@C-900 at 100 mV s⁻¹. The reduction peak observed at ~ 0.78 V (vs RHE) suggested that Hm/Cy@C-900 is effective in ORR.

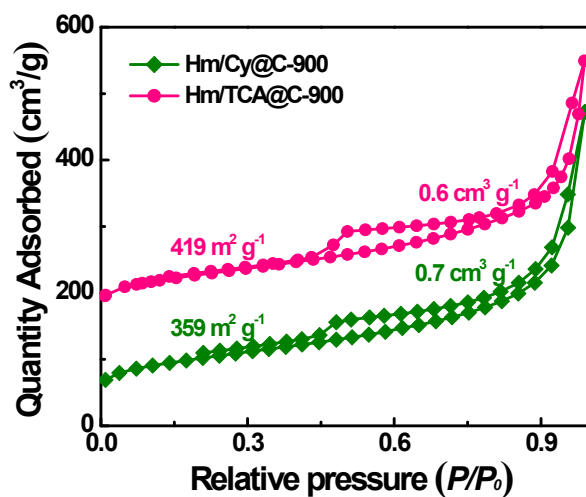


Fig. S6. N₂ adsorption-desorption isotherms of Hm/Cy@C-900 and Hm/TCA@C-900 samples. The isotherm for Hm/TCA@C-900 was offset vertically by 100 cm³ g⁻¹.

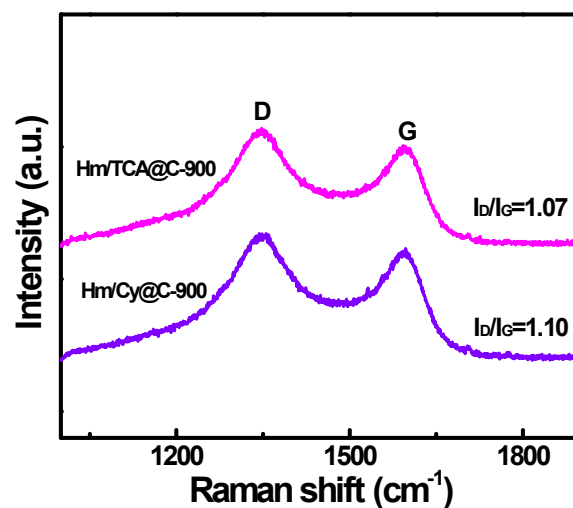


Fig. S7. Raman spectra of Hm/Cy@C-900 and Hm/TCA@C-900 samples.

Two typical peaks at 1344 (D band) and 1596 cm⁻¹ (G band) stand for the disordered carbon (the defects) and sp²-hybridization carbon species (graphitization), respectively.^[3] The I_D/I_G value of Hm/Cy@C-900 (1.10) is higher than that of Hm/TCA@C-900 (1.07), demonstrating that Hm/Cy@C-900 has more defect carbon and disordered carbon matrix, which is conducive to ORR.^[4]

III. Supporting tables

Table S1. The fitting parameters of high-resolution XPS N1s spectra of Hm/Cy@C-900 and Hm/TCA@C-900 samples.

Catalysts	Binding energy of N species (eV) and content (%)				N relative to C (at%)
	Pyridinic N	Fe-N	Pyrolic N	Graphitic N	
Hm/Cy@C-900	398.4 (37.3)	399.3 (18.1)	400.7 (34.1)	401.7 (10.5)	2.4
Hm/TCA@C-900	398.5 (11.4)	399.4 (39.6)	400.8 (30.1)	401.5 (18.9)	1.9

Table S2. The fitting parameters of high-resolution XPS S 2p spectra of Hm/Cy@C-900 and Hm/TCA@C-900 samples.

Catalysts	Binding energy of S species (eV) and content (%)		
	Fe-S	C-S-C	C-SO _x -C
Hm/Cy@C-900	163.8 (38.1)	164.8 (39.4)	167.5 (22.5)
Hm/TCA@C-900	163.8 (40.4)	164.7 (32.6)	167.8 (27.0)

Table S3. Fitting parameters of ⁵⁷Fe Mössbauer spectrum for Hm/Cy@C-900 catalyst.

Site	δ_{iso} (mms ⁻¹)	ΔE_Q	FWHM (mms ⁻¹)	Area (%)	Assignment
Doublet 1	0.29	0.74	0.58	8.55	Fe ^{III} N ₄ (five fold)
Doublet 2	0.40	2.81	0.58	7.18	Fe ^{III} N ₄ (six fold)
Sextet 1	0.19	0.00	0.38	53.51	Fe ₃ C
Sextet 2	0.77	-0.16	0.33	14.96	FeS _x
Sextet 3	0.10	-0.23	0.42	6.46	α -Fe
Sextet 4	0.42	-0.11	0.58	9.34	FeS _x

Table S4. Fitting parameters of ^{57}Fe Mössbauer spectrum for Hm/TCA@C-900 catalyst.

Site	δ_{iso} (mms $^{-1}$)	ΔE_{Q}	FWHM (mms $^{-1}$)	Area (%)	Assignment
Doublet 1	0.28	-0.07	0.38	6.22	Fe_3O_4 (+3.0)
Doublet 2	0.56	0.10	0.68	12.43	Fe_3O_4 (+2.5)
Sextet 1	-0.02	0.01	0.31	18.84	α -Fe
Sextet 2	0.17	0.01	0.35	41.41	Fe_3C
Sextet 3	0.29	0.83	0.58	6.92	$\text{Fe}^{\text{III}}\text{N}_4$ (five fold)
Sextet 4	0.52	1.65	0.58	4.83	$\text{Fe}^{\text{II}}\text{N}_4$ (four fold)
Sextet 5	0.75	-0.26	0.58	9.35	FeS_x

Table S5. Comparison of the ORR performances of Hm/Cy@C-900 and presentative reported Fe-based NNMEs in a O_2 -saturated 0.1 M KOH solution in recent five years.

Catalyst	Carbon source	E_{onset} (V)	$E_{1/2}$ (V)	J_{L} (mA cm $^{-2}$)	Ref.
Hm/Cy@C-900	carbon ECP	0.93	0.845	5.76	This work
Fe_1 -HNC-500-850	glucose	0.93	0.842	5.80	[5]
Fe/N/S-PCNT	polypyrrole	0.96	0.840	--	[6]
Fe_3C @NCNTs	g- C_3N_4	0.92	0.840	5.80	[7]
Fe-N-C/MXene	C_3N_4	0.92	0.840	--	[8]
Fe/OES	2-methylimidazole	1.00	0.850	--	[9]
Fe-SAC/NC	adenine	0.95	0.840	--	[10]
OM-NCNF- FeN_x	PAN, branched silica nanoaggregates	0.905	0.836	5.6	[11]
Fe/Mn-N-C	dopamine hydrochloride	1.02	0.880	5.7	[12]
Ni-N $_4$ /GHSs/Fe-N $_4$	graphene oxide	0.93	0.83	--	[13]
PtFeNC	2-methylimidazole	1.05	0.895	6.20	[14]

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