Supplementary Information

Emerging disorder in Gd₂(Ti_{1-x}Zr_x)₂O₇ pyrochlores matrices for radioactive waste disposal: symmetry lowering versus defect clustering

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Size and strain analysis

The experimental patterns of x_{zr} =0.75 (Panel A) and x_{zr} =0.85 (Panel B) samples (already shown in figure 1 of the main text,) are displayed in Figure S1 as black circles for sake of clearness. To reveal the compounds microstructure, the experimental peaks up to 20=20° ($Q_{max} \approx 6 \text{ Å}^{-1}$) were fitted using Pseudo-Voigt functions and after the deconvolution of the instrumental resolution function, as described in the instrumental section, their integral breaths were plotted according to the W.-H. equation, to calculate the crystallite size (D_v) and strain (ϵ) contribution to peak broadening.

In the insets of Figure SI1 the W-H plots for the x_{Zr} =0.75 (Panel A) and x_{Zr} =0.85 (Panel B) samples are reported. The experimental $\beta cos\theta$ data for the *fluorite* peaks are represented by full circles, while the *pyrochlore* (*superstructure*) ones are depicted as empty circles.

Up to x=0.75, all the experimental points lie on the same straight line. For x≥0.85, the empty circles are systematically shift toward higher values pointing to a reduced crystallographic coherence of pyrochlore ordering in respect to the crystallite size. so that two linear regression lines have to be used (see inset of Fig.SI1 panel B). It is worth noting that the two interpolating lines are almost parallel to each other, so affecting the D_V value rather than the ε one. The calculated ε and λ / D_V values are shown in Figure SI panel C and D respectively. Strain ε is almost 10⁻⁴ for x=0, as expected for a well grown crystalline phase; then it suddenly grows up of an order of magnitude in the x=0.15 sample and smoothly decreases on raising xZr, reaching $\approx 8x10^{-4}$ in the Ti-free sample. This behavior suggests that all intermediate compositions in the Gd₂(Ti_{1-x}Zr_x)₂O₇ solid solution and the Gd₂Zr₂O₇ endmember are much more disordered than the Gd₂Ti₂O₇ endmember. λ /D_V values are very close to zero, testifying the micrometric size of the crystallites of all the samples. We prefer to plot this parameter instead of the particle dimension D_V because the inverse proportionality between them would generate enormous sigma(D_V) values and confuse the reader. Also, in Figure SI panel D the empty circles refer to the fluorite peaks could indicate a slightly lower structure coherence: some lowering of the pyrochlore ordering in respect to the whole crystallites.



Figure SI1. HR-XRPD patterns of $Gd_2(Ti_{1-x}Zr_x)_2O_7$ for xZr = 0.75 (**A**) and 0.85 (**B**) with corresponding W.H. plots in the insets. The calculated strain (epsilon) values and λ/D_V values are shown in panels (**C**) and (**D**), respectively. Full and empty circles are the refined values for the *fluorite* and *pyrochlore* peaks, respectively.



Figure SI2. Close-up view of $(440)_P$ reflection in selected HR-XRPD patterns of $Gd_2(Ti_{1-x}Zr_x)_2O_7$ solid solution. Cross symbols represent the observed pattern; the dark grey solid line is the total calculated profile by Rietveld refinement, while solid red and blue lines are the calculated profile for Phase 1 and Phase 2 respectively.

Pyrochlore model and results from Rietveld refinements

Atom	Wyckoff letter	х	у	z
Α	16d	$\frac{1}{2}$	$\frac{1}{2}$	$\frac{1}{2}$
В	16c	0	0	0
01	48f	X a	$\frac{1}{8}$	$\frac{1}{8}$
02	8b	$\frac{3}{8}$	$\frac{3}{8}$	$\frac{3}{8}$
03	8a	$\frac{1}{8}$	$\frac{1}{8}$	$\frac{1}{8}$

Table SI1 Pyrochlore structural model employed in Rietveld refinements of HR-XRPD patterns (s.g. $Fd\bar{3}m$, n°227, origin choice 2, origin at the B site)

^a xO1 ranges from 0.3260 in $Gd_2Ti_2O_7$ to 0.3470 in $Gd_2Zr_2O_7$

xZr		0	0.15	0.25	0.50	0.75	0.85	1
	Weight							
Phase 1	Fraction	94.3	99.9	98.0	79.2	57.9	70.8	89.6
	%							
	a / Å	10.178215(8)	10.23104(3)	10.26928(3)	10.35152(2)	10.45829(3)	10.48699(3)	10.52304(2)
	Weight							
Phase 2	Fraction	5.7	0.01	2.0	20.7	42.1	29.2	10.4
	%							
	a / Å	10.232(3)	10.42(2)	10.3372(8)	10.3659(2)	10.46544(6)	10.4690(1)	10.4790(6)
<i>x</i> O1		0.3260(2)	0.3279(2)	0.3294(2)	0.3337 (2)	0.3415(2)	0.3438(2)	0.34698(4)
o.f.(O3)		0	0.06(1)	0.08(1)	0.20(1)	0.27(1)	0.338(9)	0.40(1)
o.f.(Zr _A)		0	0	0	0	0.026(1)	0.045(1)	0.048(2)
Uiso / Ų	А	0.00447(3)	0.00449(4)	0.00536(5)	0.00918 (6)	0.01198(6)	0.01203(6)	0.01135(8)
	В	0.00372(7)	0.0040(1)	0.0055(1)	0.0090(1)	0.0144(1)	0.0165(1)	0.0182(2)
	0	0.0030(3)	0.0034(4)	0.0052(5)	0.0098(6)	0.0233(7)	0.0236(6)	0.0233(9)
wRp(%)		11.5	9.5	9.2	8.4	7.2	6.8	7.6
Goodness-of-		7.0	12	45	2.0	4.5	2.0	5 5
Fit		7.5	4.5	4.5	5.9	4.5	5.0	5.5

Table SI2 Structural parameters extracted from Rietveld refinements of HR-XRPD patterns

Raman features

Besides the fundamental bands of the pyrochlore structure discussed in the main text, other modes are observed as well. The band at ≈115 cm⁻¹ (M1) in the spectrum of the Ti-endmember has been already reported in previous investigations on $Gd_2Ti_2O_7^{1,2}$ and could be ascribed to a lowering of the local symmetry, which activates formally forbidden IR modes³. This evidence suggests that although the Gd₂Ti₂O₇ approaches the ideal pyrochlore structure more than any composition in this study, some degree of distortion is retained. The progressive substitution of Zr in the structure enhances the distortion and deviation from the ideal pyrochlore structure and leads to a relaxation of the selection rules. Moving on to the line at \approx 215 cm⁻¹(M2), this has been assigned to one of the four T_{2a} modes by several authors^{1,4–6} and related to the vibration of O2 at the center of OGd₄ tetrahedra¹. The decrease in intensity with Zr-substitution may be ascribed to a change in the position of O2 and/or distortion of the corresponding OGd₄ tetrahedron. A similar trend was observed in the case of pressure induced distortion in Gd₂Ti₂O₇⁵. Kumar and coworkers showed by means of DFT calculations that no Raman active modes should be allowed below 220 cm⁻¹ and it is likely that the **M2** line is another forbidden IR mode activated by displacive disorder of Gd and O2 ions^{7,8}. The band at \approx 258 cm⁻¹, which is hardly visible in the spectrum of Gd₂Ti₂O₇ and completely disappears in the other spectra of the solid solution, has been assigned to a T_{2q} mode by Lummen and coworkers¹. The other weak feature in the spectrum of $Gd_2Ti_2O_7$ is the band ≈ 450 cm⁻¹(M4), which could be assigned again to one of the remaining T_{2g} modes^{4,6,9}. Kumar and coworkers predicted this mode with contributions of Gd-O2 stretching and O1-Gd-O2 bending force constants in addition to the dominant O1-Ti-O1 bending force constant from TiO₆ octahedra⁷. The band was already observed in Gd₂Ti₂O₇¹⁰ and other lanthanide titanates $Ln_2Ti_2O_7$ with Ln = Dy, Lu^{11} and Tb^{12} . Oueslati and coworkers assigned the **M4** band to B-O vibration and the red-shift with increasing x_{2r} was regarded as a clear indication of lighter Ti⁴⁺ being substituted by heavier Zr⁴⁺ in the structure¹³. A wavenumber of \approx 400 cm⁻¹ is reached in Gd₂Zr₂O₇ for **M4** and similar values were recorded for the same composition¹⁴ and other related zirconate pyrochlores like $Nd_{2-y}Y_yZr_2O7^{15}$, $Nd_{1-x}Dy_xZr_2O7^{16}$ and $Sm_2Zr_2O_7^{17}$.



Figure SI3 Selected fits for the Raman spectra of $Gd_2(Ti_{1-x}Zr_x)_2O_7$ series



Figure SI4 Frequency shift of Raman modes in $Gd_2(Ti_{1-x}Zr_x)_2O_7$ as obtained by fitting the spectra with Gaussian functions.

PDF analysis results



Figure SI5. Experimental G(r) functions in the 1.8-8 Å (left side) and 9-22 Å (right side) intervals are displayed as light blue circles. Labels in the panels refer to the sample composition according to $Gd_2(Ti_{1-x}Zr_x)_2O_7$ formula. The fits using the pyrochlore model are reported as red curves. The green curves represent the differences between experimental data and the model-



Figure SI6 Fit of the G(r) function pertinent to $Gd_2Zr_2O_7$ sample in the 1.8-11 Å extending the *Pmma* model fitted in the 1.8-8 range a) without changing the parameters and b) after refinement; panel c) reports the fit using the pyrochlore model.



Figure SI7. Residuals of G(r) fits as a function of composition in the 1.8-8 Å range. Black circles refer to pyrochlore structural model, while Red triangles refer to the *Pmma* weberite-type model. See main text for details.

Structures from DFT calculations



Figure SI8A: Figure_7C enlarged



Figure SI8B: Figure_7D enlarged

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