

1 **Supporting Information for**
2 **Synergy of strong/weak interface adhesion forces and Li₂S**
3 **additive enabling high performance full anode-free lithium-**
4 **metal batteries**

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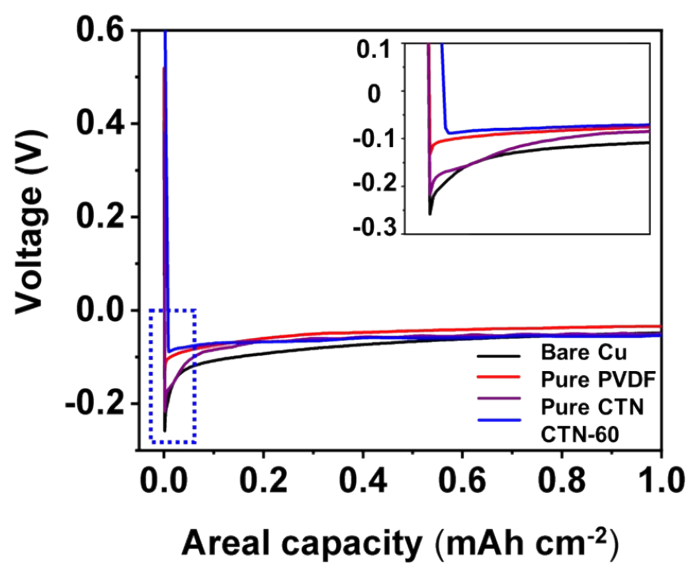
15 Shukai Ding, E-mail: dingshukai@sust.edu.cn

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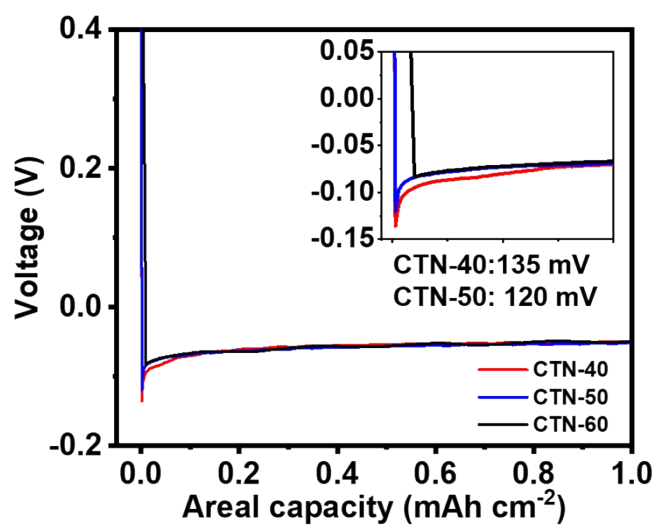
17 Guoquan Suo, E-mail: suoguoquan@sust.edu.cn

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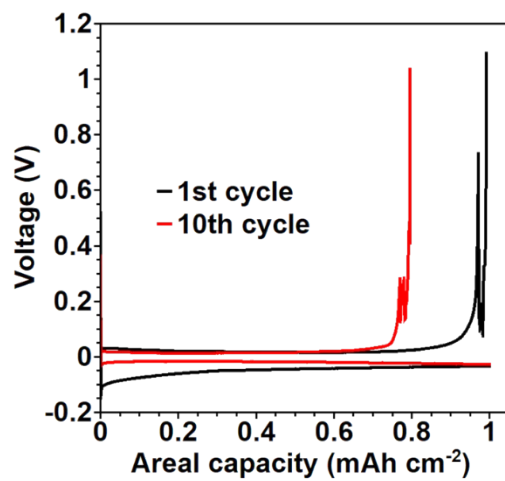
21 Figure S1 Initial voltage versus areal capacity profiles in half-cell of bare Cu, pure PVDF, pure CTN,
22 and CTN-60 at the current density of 0.5 mA cm⁻².



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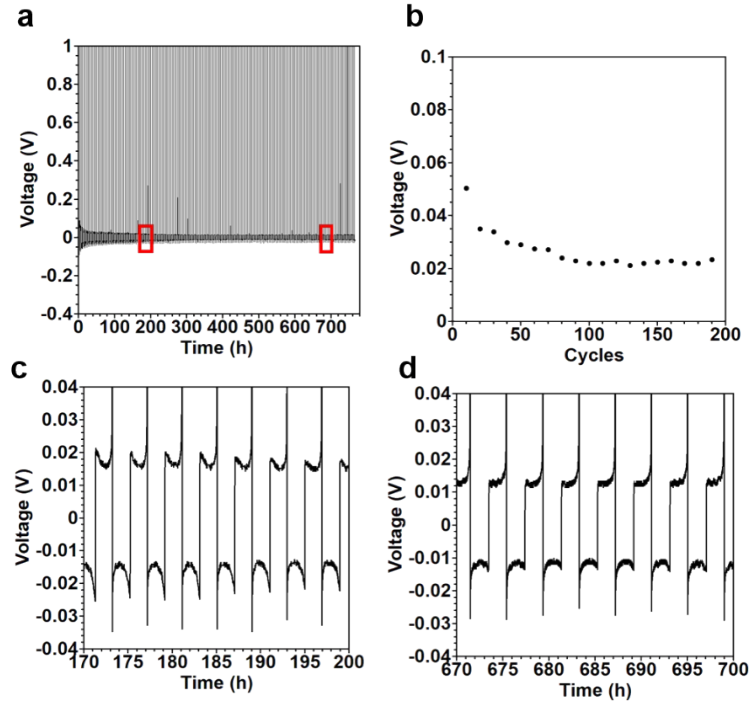
25 Figure S2 Initial voltage versus areal capacity profiles in half-cell of CTN-40, CTN-50, and CTN-60
26 at the current density of 0.5 mA cm⁻².

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Figure S3 Voltage profiles in the half cell of pure PVDF.



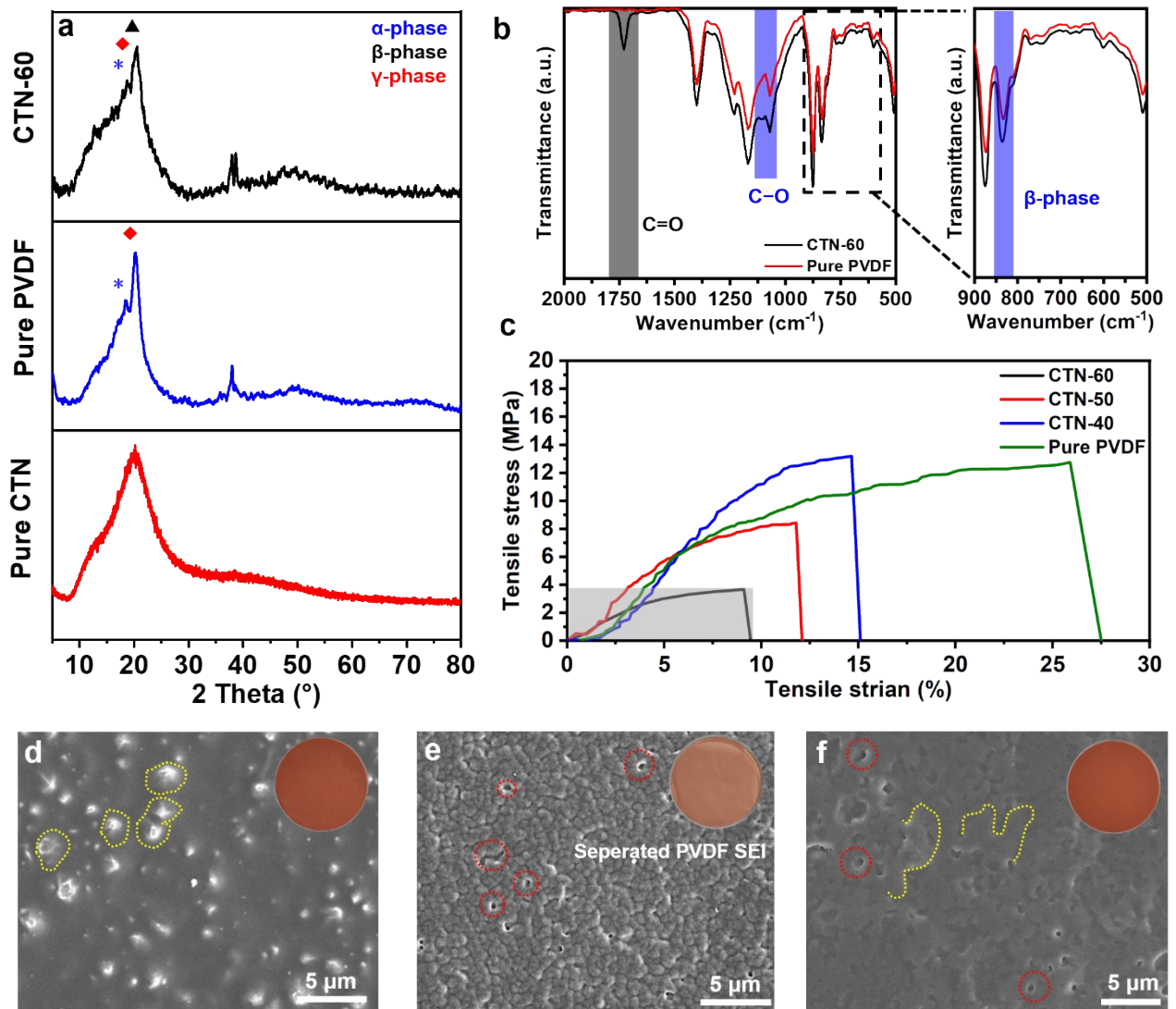
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32 Figure S4 (a) Voltage versus cycle number curve of CTN-60 at 0.5 mA cm^{-2} with an areal capacity of
 33 1 mAh cm^{-2} , (b) voltage hysteresis versus cycle number, and (c, d) voltage profiles of selected cycles.

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Table S1 Performance comparison of half-cell in literatures.

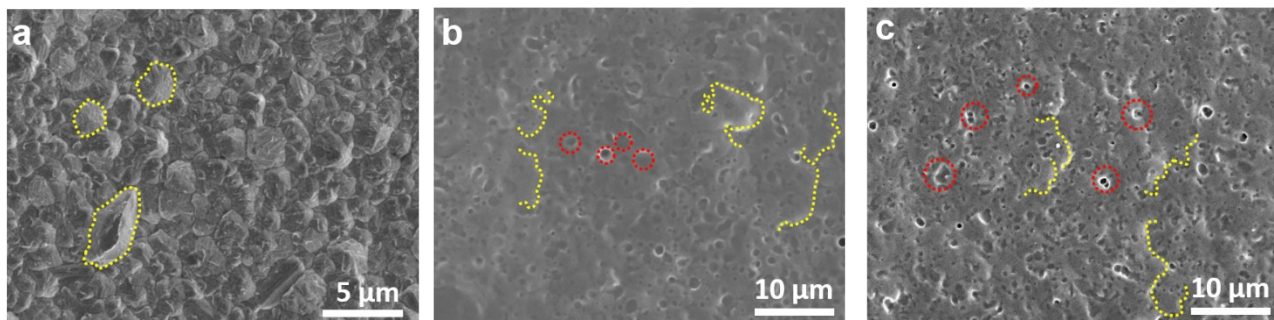
Current density (mA cm⁻²)	Capacity (mAh cm⁻²)	Cycles	Average coulombic efficiency (%)	Reference
0.5	1	200	96.08	This work
5	0.5	30	98.83	1
1	0.5	200	98	2
5	0.5	100	90.2	3
1	1	200	97.7	4
1	1	130	97.3	5
0.2	1.5	40	98.1	6
1	1	100	/	7
0.5	1	200	94.5	8
1	0.5	240	97.6	9
0.5	1	150	87	10
1	1	180	97	11
0.25	0.5	300	98.5	12



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38 Figure S5 (a) XRD pattern for CTN-60, pure PVDF, and pure CTN, (b) FTIR spectra of pure PVDF
 39 and CTN-60, (c) mechanical property of CTN-60, -50, -40, and pure PVDF, morphology (inset:
 40 photographs) of (d) pure CTN, (e) pure PVDF, (f) CTN-60 characterized by SEM (yellow dotted line
 41 representing the convex formed by aggregation of CTN or crystallizing of PVDF molecule, red dotted
 42 circles representing the crack induced by phase segregation).

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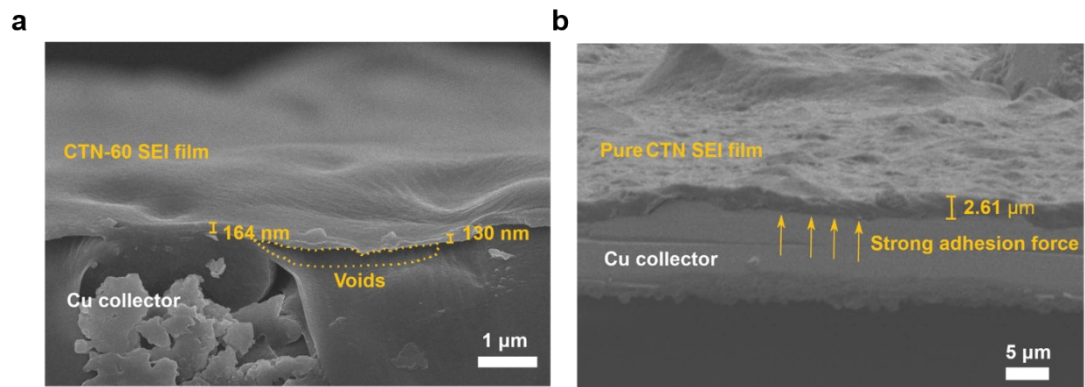
45 Figure S6 Morphology of (a) bare Cu, (b) CTN-50, and (c) CTN-40 characterized by SEM. (yellow
46 dotted line representing the Cu convex in Figure a and imprinted Cu convex by SEI in Figures b and
47 c, red dotted circles representing the crack induced by phase segregation).

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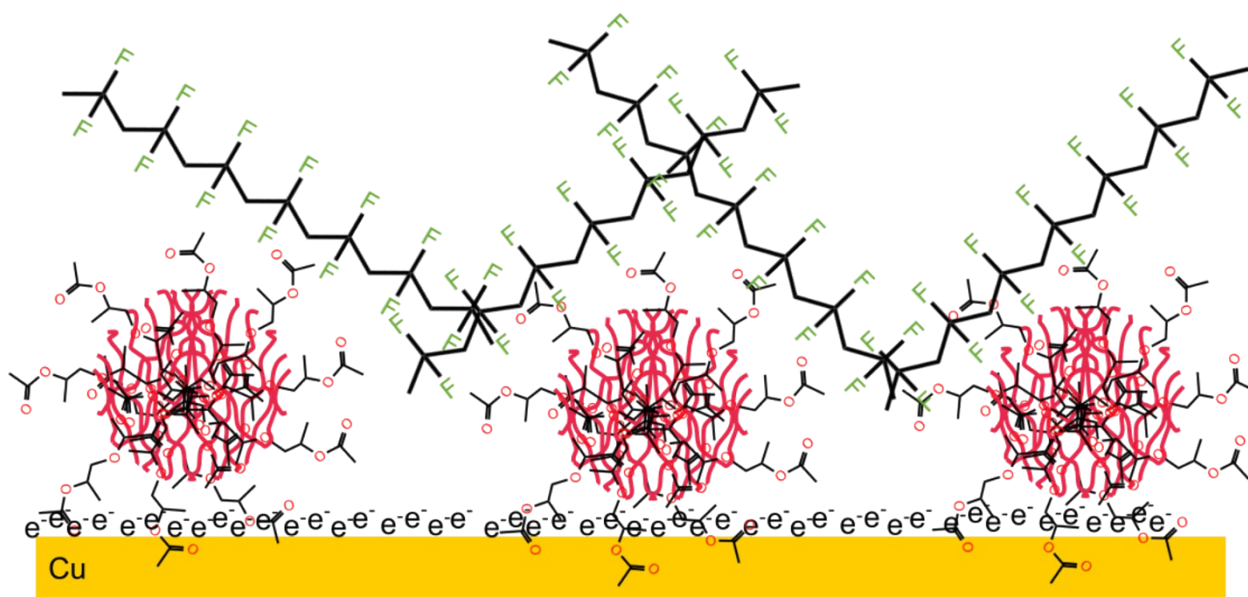


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54 Figure S7 Cross-section of (a) Cu collector with CTN-60 and (b) Cu collector with pure CTN
55 characterized by SEM.

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Weak adhesion force of PVDF's inert C-F bond



Strong adhesion force from interaction of CTN's carbonyl group with Cu's surface electron

Figure S8 Scheme of the mixed adhesion force of CTN-60 with Cu collector.

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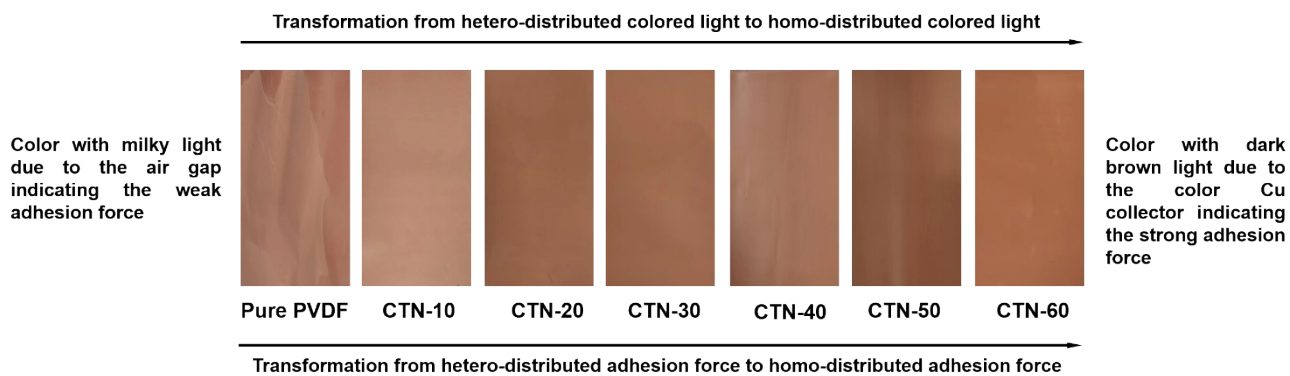


Figure S9 Transformation of visual state characterized by different SEIs photos

Transformation from separated state to bond state between Cu collector and CTN-based SEI

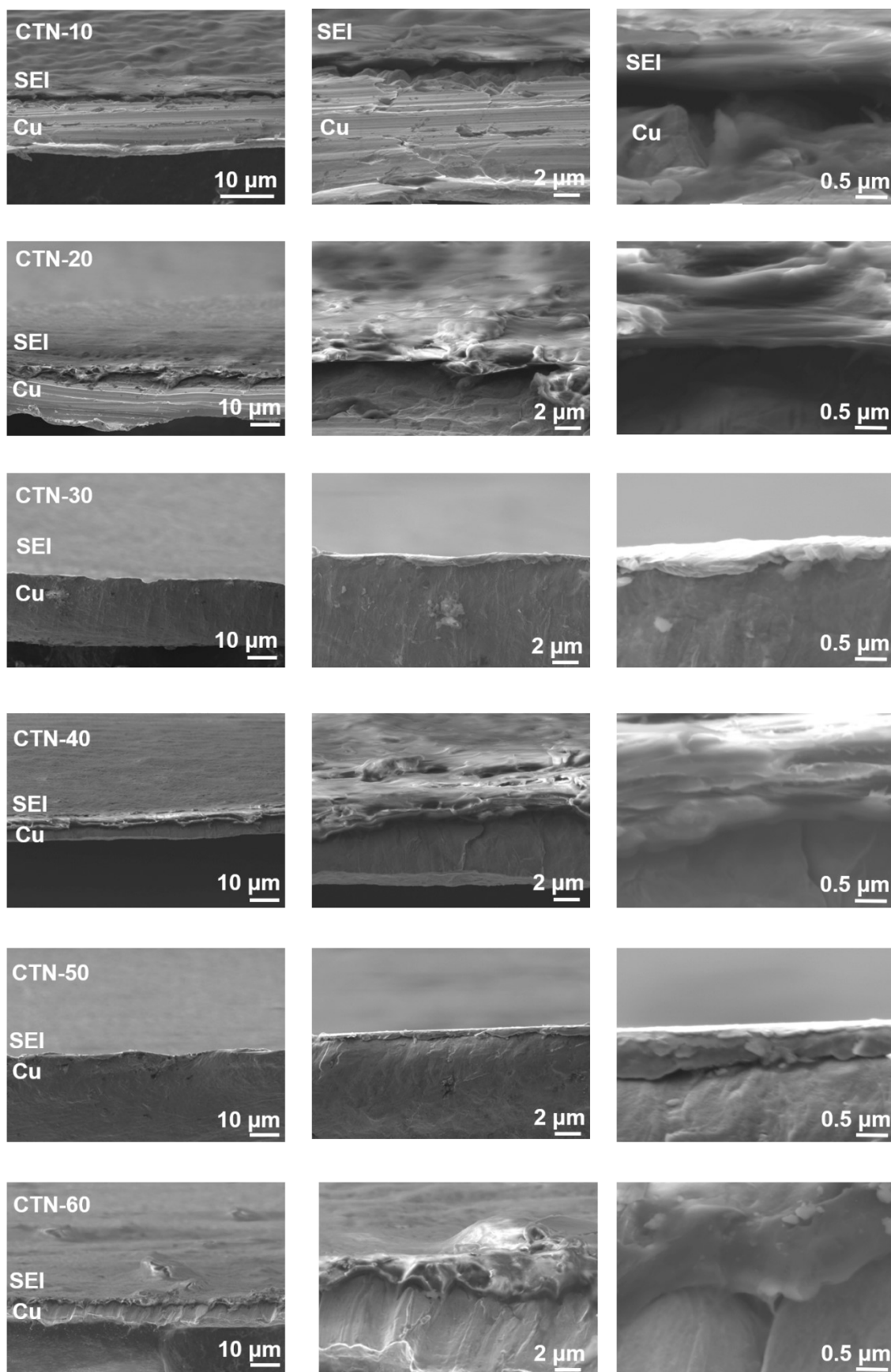
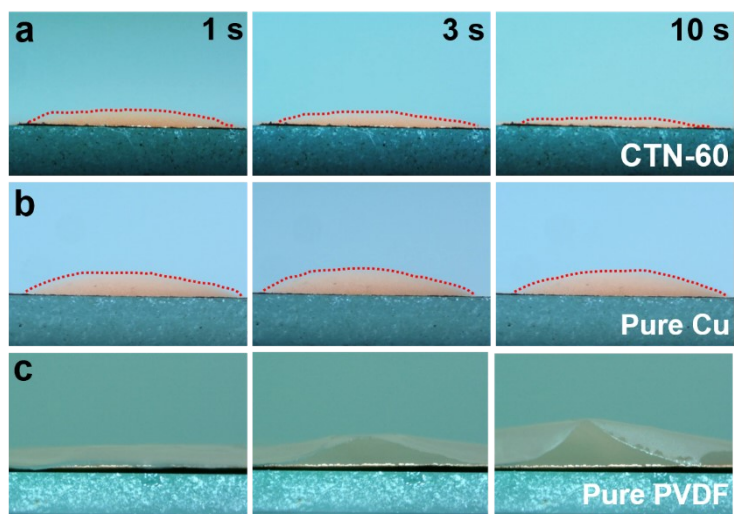


Figure S10 Transformation of different SEIs interfacial state characterized by SEM.

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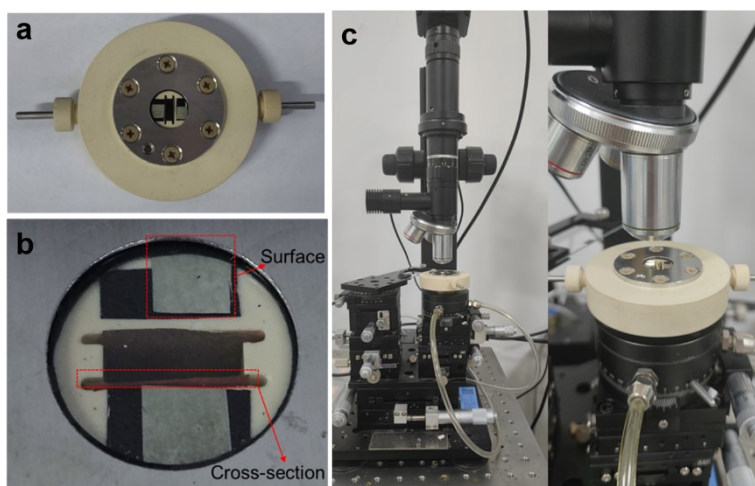
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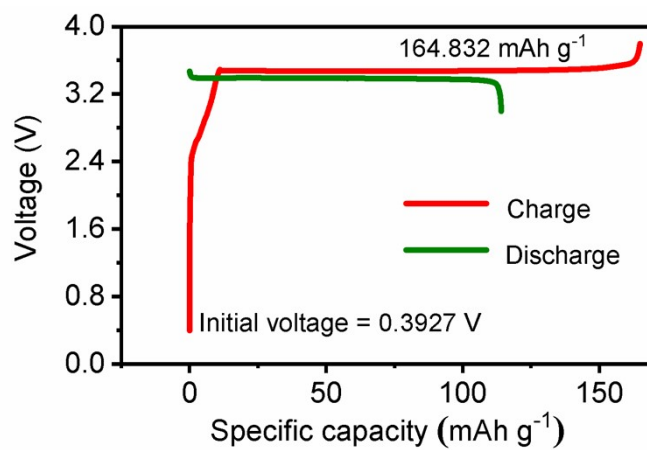
65 Figure S11 (a) Electrolyte wetting testing of (a) CTN-60, (b) bare Cu, and (c) pure PVDF.

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69 Figure S12 (a) Sealed equipment with glass windows, (b) locations of the sample in sealed equipment
70 for observation the morphology of surface and cross-section, and (c) optical microscope for
71 observation.

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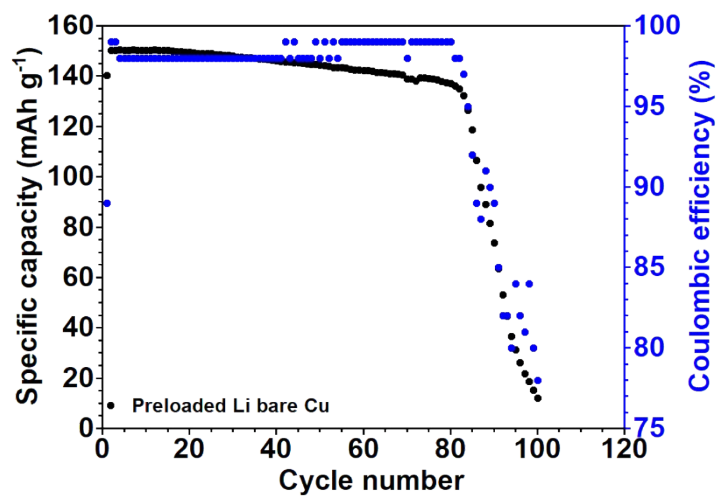


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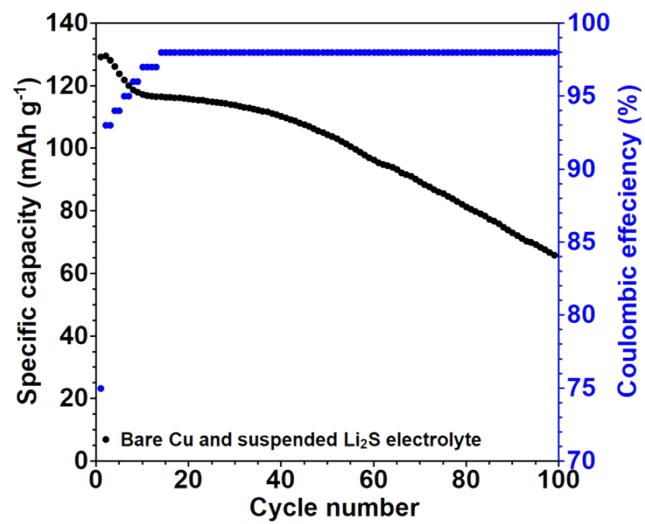
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Figure S13 Initial charge and discharge curves of LiFePO₄ based full cell with CTN-60.

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77 Figure S14 Electrochemical performance of LiFePO₄ full cell with the preloaded Li bare Cu with 2
78 mAh cm⁻².
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80 Figure S15 Electrochemical performance of LiFePO₄ full cell with the bare Cu and the suspended
 81 Li₂S electrolyte.

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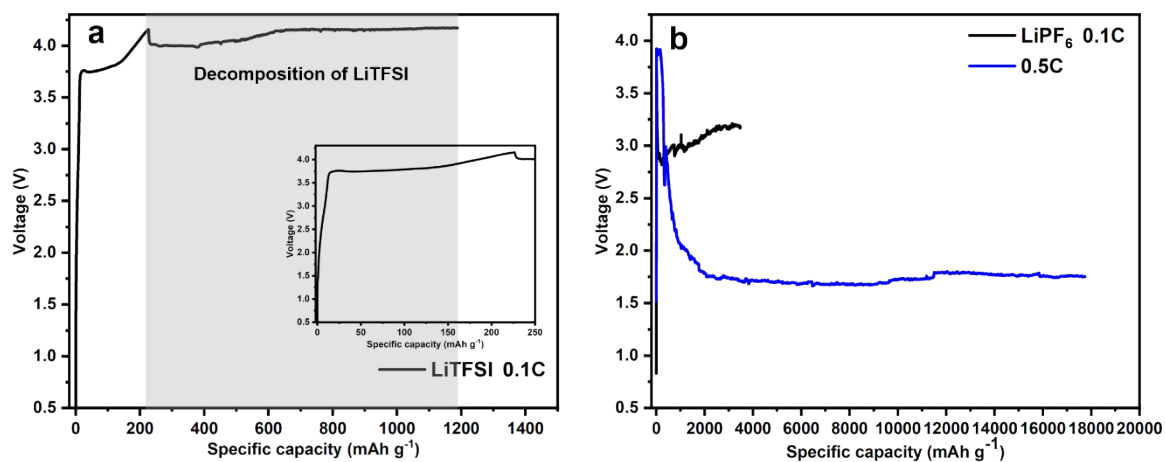


Figure S16 Voltage versus specific capacity curve of NCM 811 full cell with Li_2S (a) 1 M LiTFSI in DOL: DME = 1:1 Vol % with 2 % LiNO_3 at current density of 0.1 C and (b) 1M LiPF_6 in EC:DEC:EMC=1:1:1 Vol% at current density of 0.1 C and 0.5 C, respectively.

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Table S2 Performance comparison of full-cell in literatures.

Collector	Cathode	Electrolyte	Current rate	Cycles	Capacity retention (%)	Ref.
Cu@CTN-60	LFP	1 M LiTFSI + 2%LiNO ₃ in DOL: DME (1:1, v/v) +2%Li ₂ S	2C (1.19 mA cm ⁻²)	100	85.98%	This work
Cu@LiF-PVDF	LFP	1 M LiTFSI + 2%LiNO ₃ in DOL: DME (1:1, v/v)	1C	40	~18.5%	3
Cu Ag@PDA-GO	NCM	1 M LiPF ₆ in EC/DEC (1:1 v/v) with 5% FEC	0.5 mA cm ⁻²	50	65.47%	6
Cu@MLG	LFP	1 M LiTFSI + 2%LiNO ₃ in DOL: DME (1:1, v/v)	0.1C	100	60.92%	7
Cu@PEO	LFP	1 M LiTFSI + 2%LiNO ₃ in DOL: DME (1:1, v/v)	0.2C	100	49.6%	10
Cu	NCM622	2.3 M LiFSI and 4.6 M LiTFSI in DME	C/3	55	~55.62%	13
Cu@GO	NCM	1 M LiPF ₆ in EC/DEC (1:1 v/v) with 5% FEC	0.2 mA cm ⁻²	20	46.32%	14
3DLN	LFP	1 M LiPF ₆ in EC/DM (1:1, v/v)	0.5 mA cm ⁻²	100	49.1%	15
Cu	NCM111	2 M LiPF ₆ in EC/DEC+50% FEC	0.2 mA cm ⁻²	50	40%	16
Cu	LFP	2 M LiFSI + 1 M LiTFSI in DOL/DME (1:1, v/v)	0.2 mA cm ⁻²	100	33.3%	17
Cu	LFP	4 M LiFSI in DME	0.2 mA cm ⁻²	50	52%	18

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