

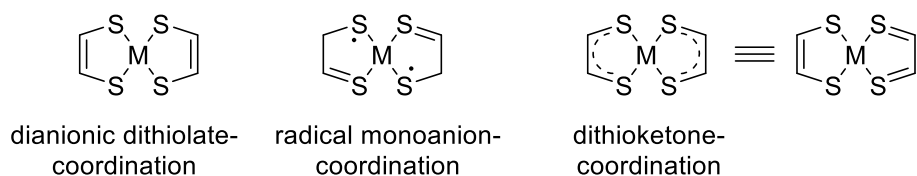
*Electronic Supporting Information*

# Neutral $d^8$ Metal Complexes with Intervalence Charge-Transfer Transition Triggers Effective NIR-II Photothermal Conversion for Solar-Driven Desalination

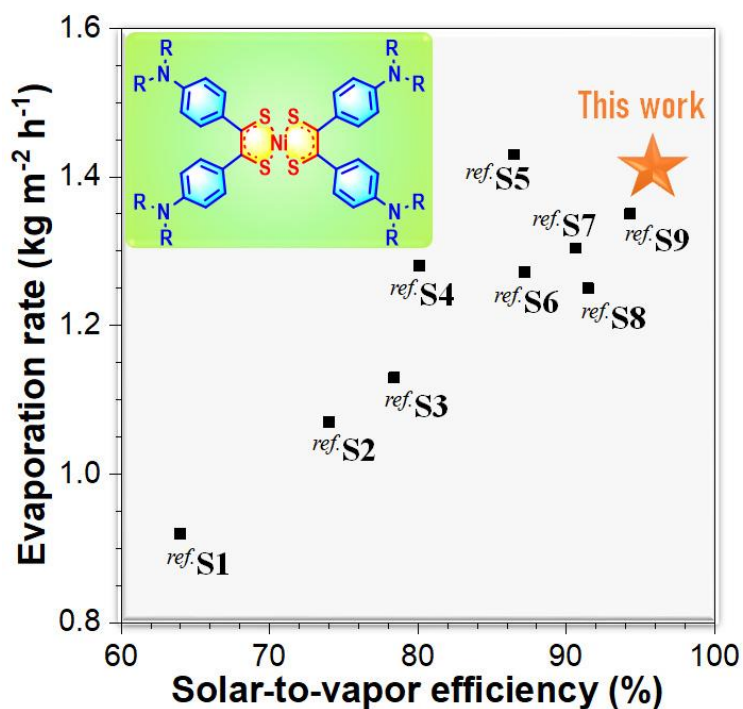
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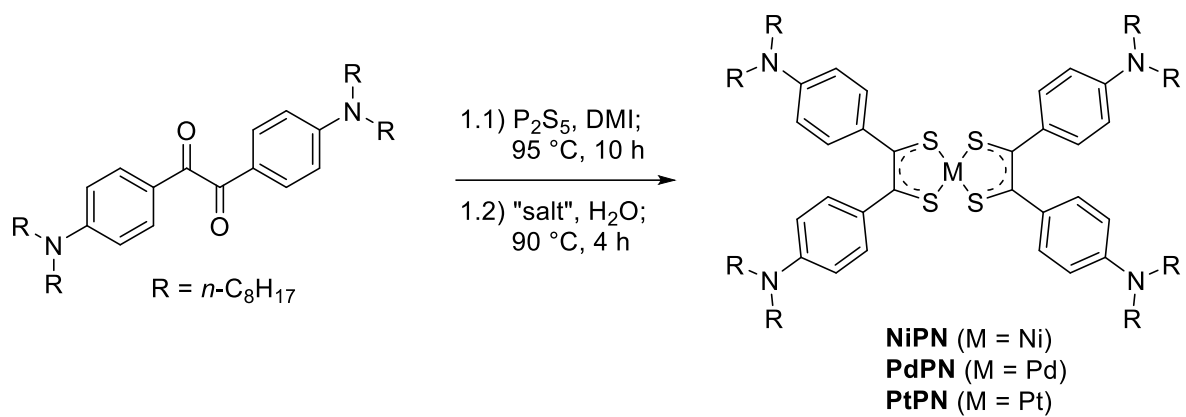
<sup>#</sup> *These authors contributed equally to this work*



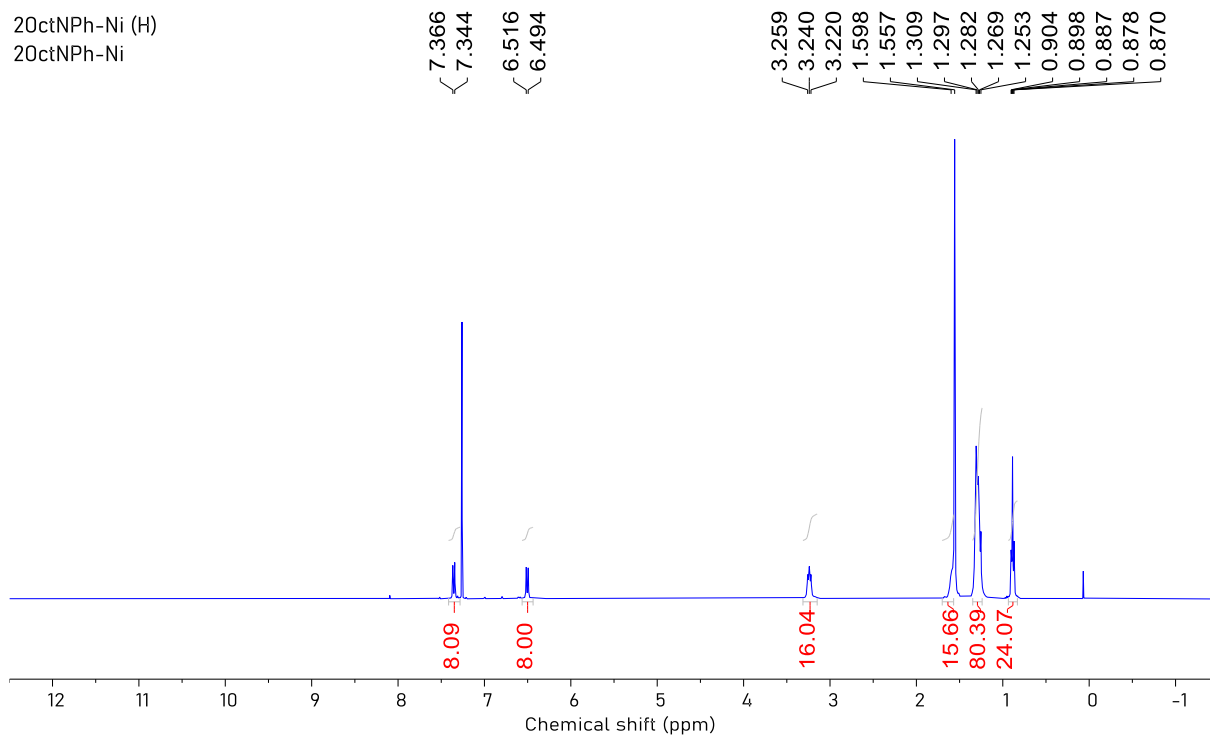
**Scheme S1.** The three coordination forms of neutral  $d^8$  metal bis-dithiolene complexes. (M = Ni, Pd, Pt)



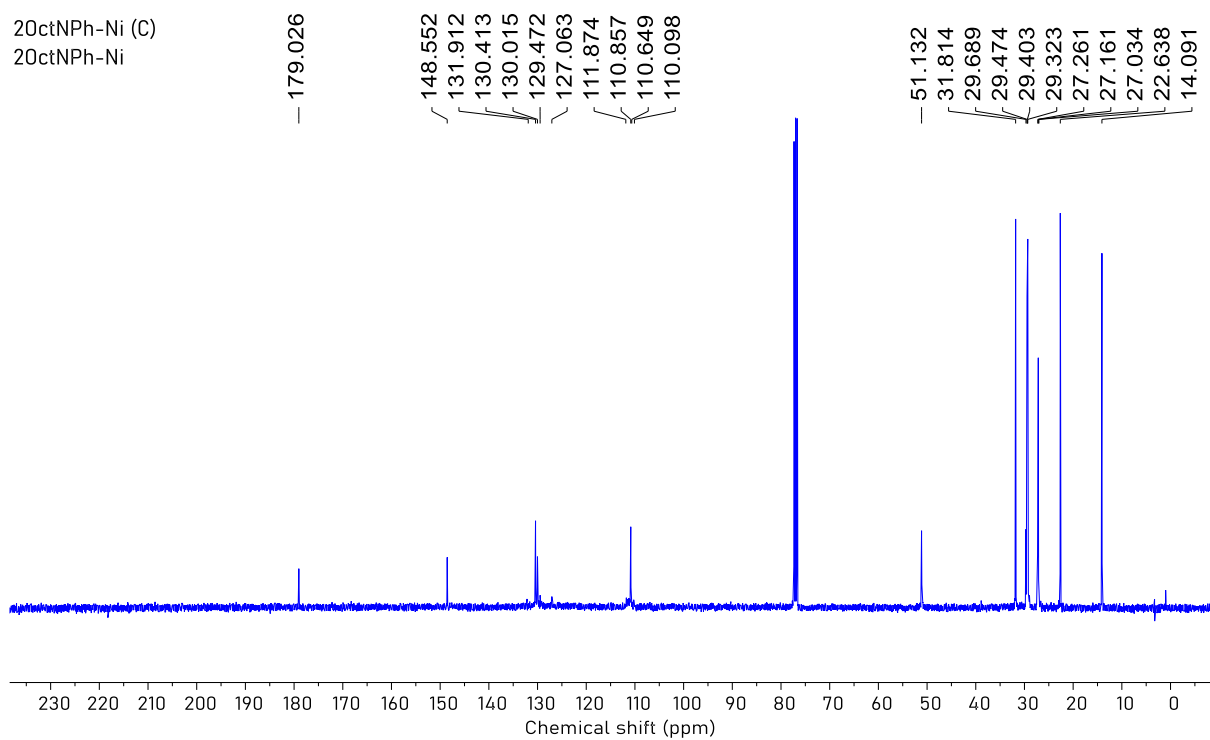
**Scheme S2.** The solar energy-to-vapor efficiency and water-mass evaporation rate of solar-thermal conversion materials based on small molecules.<sup>[S1-S9]</sup>



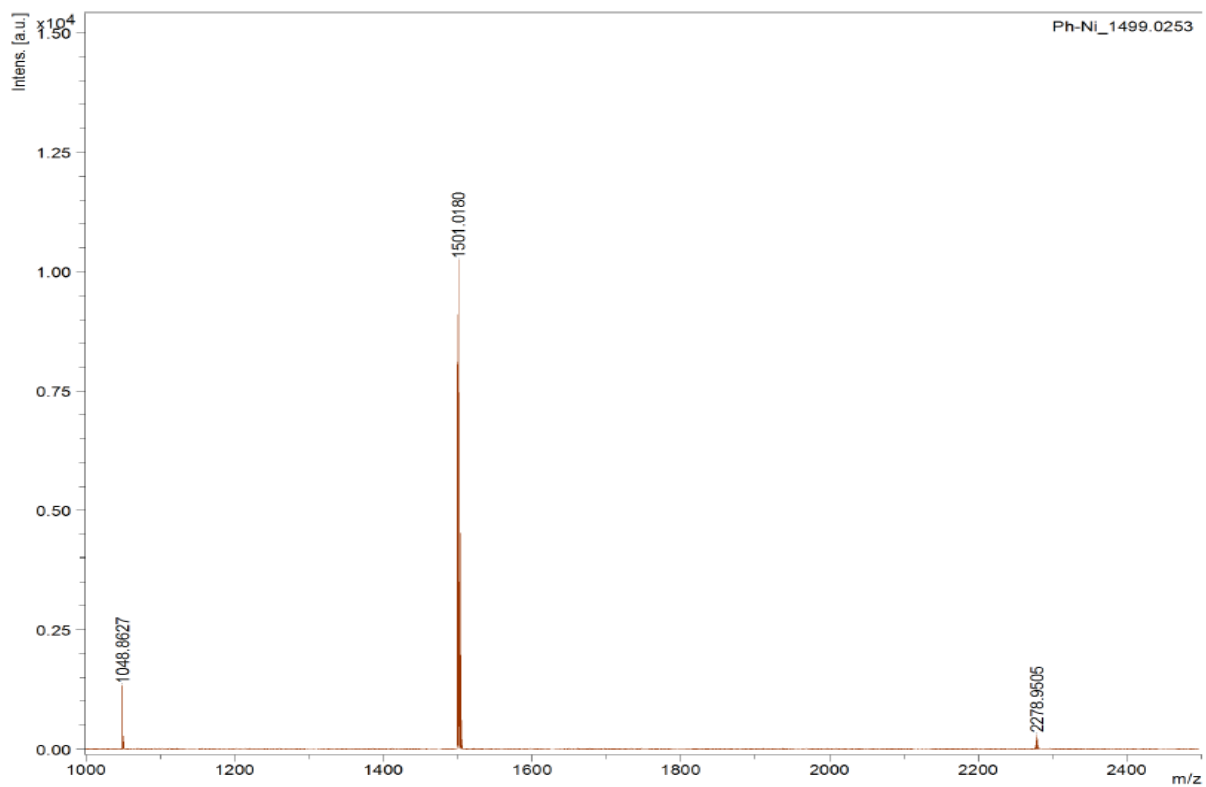
**Scheme S3.** Synthetic routes of neutral  $d^8$  transition-metal bis-dithiolene complexes. The "salt" is  $NiCl_2 \cdot 6H_2O$  for NiPN,  $Na_2PdCl_4$  for PdPN, and  $K_2PtCl_4$  for PtPN, respectively.



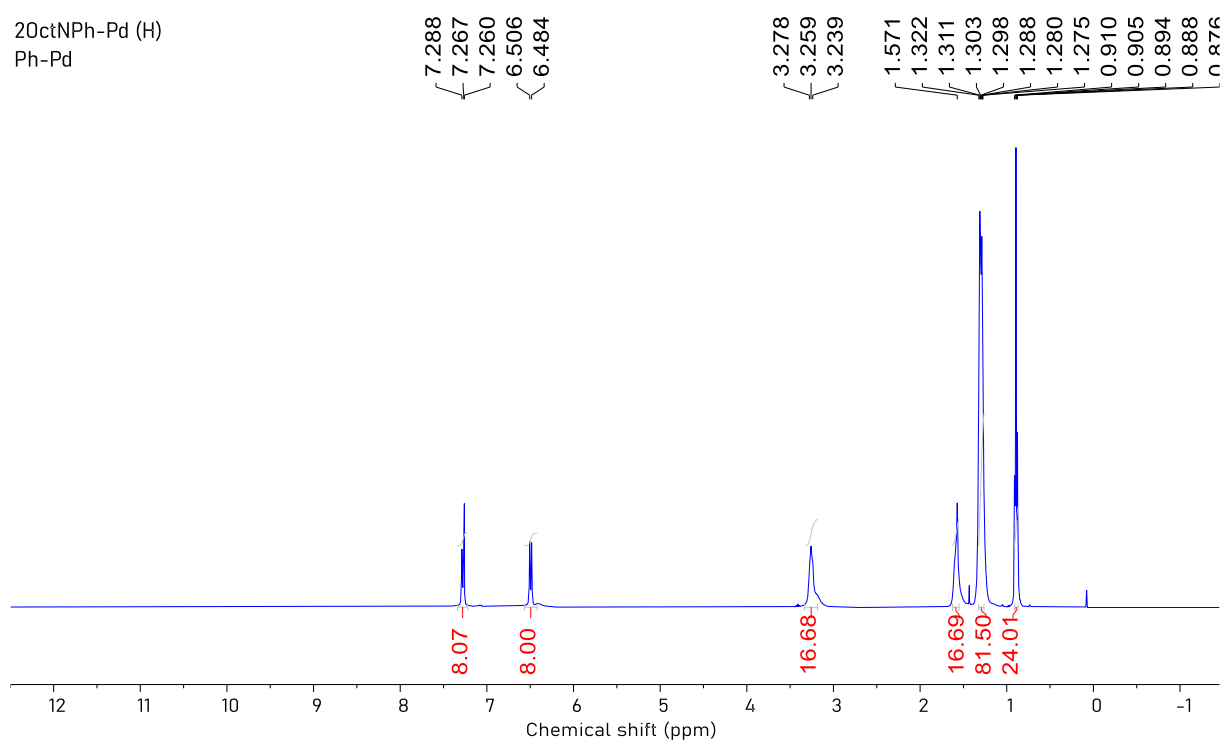
**Figure S1.** The  $^1\text{H}$ -NMR spectrum of NiPN in  $\text{CDCl}_3$ .



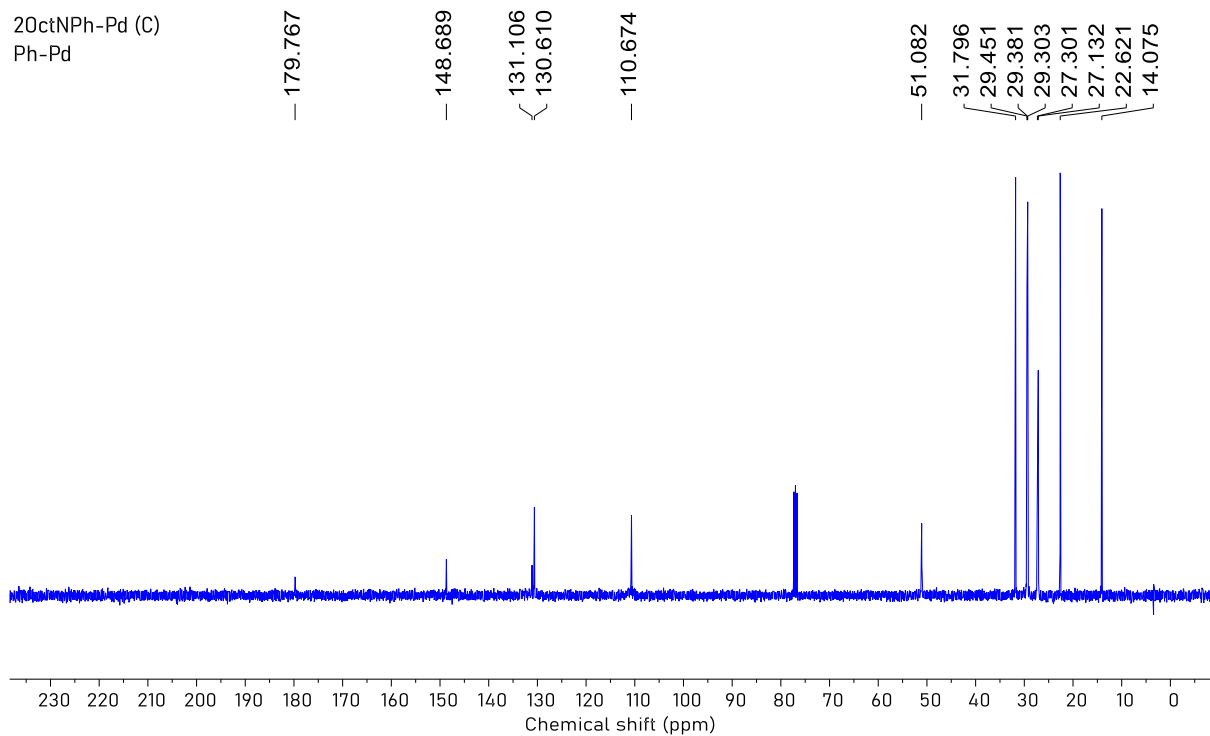
**Figure S2.** The  $^{13}\text{C}$ -NMR spectrum of NiPN in  $\text{CDCl}_3$ .



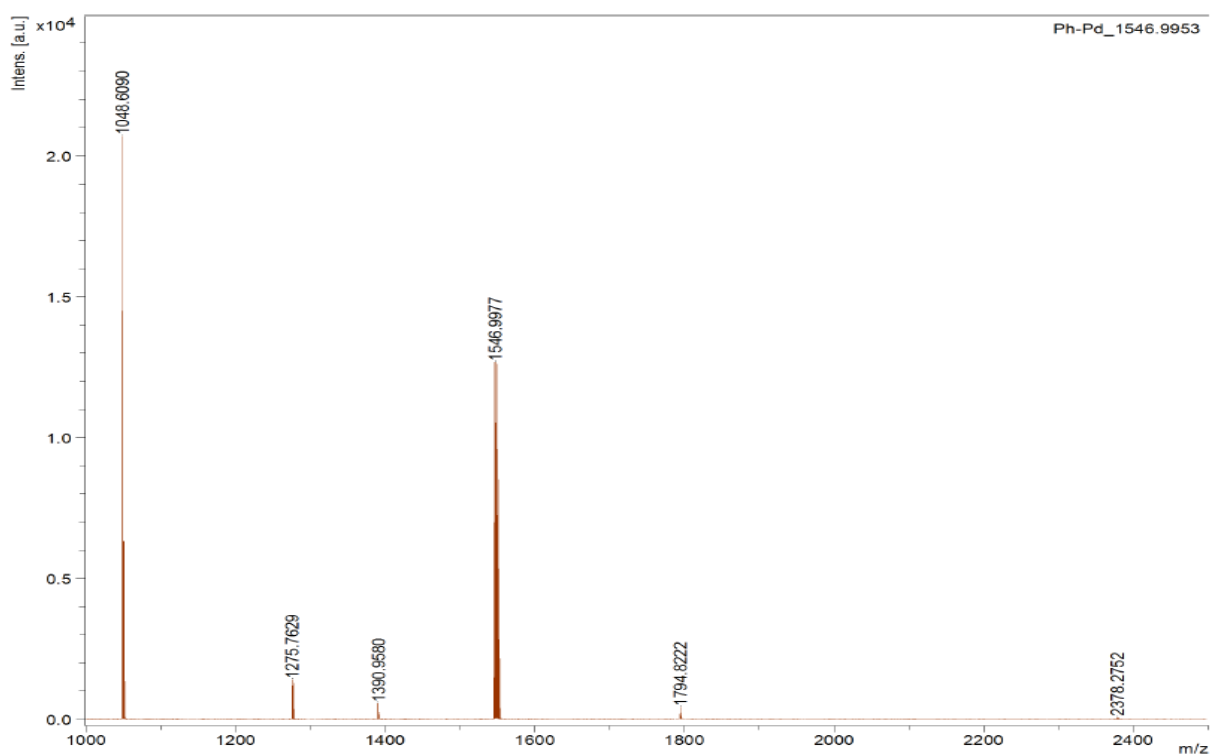
**Figure S3.** High-resolution mass spectrum of NiPN.



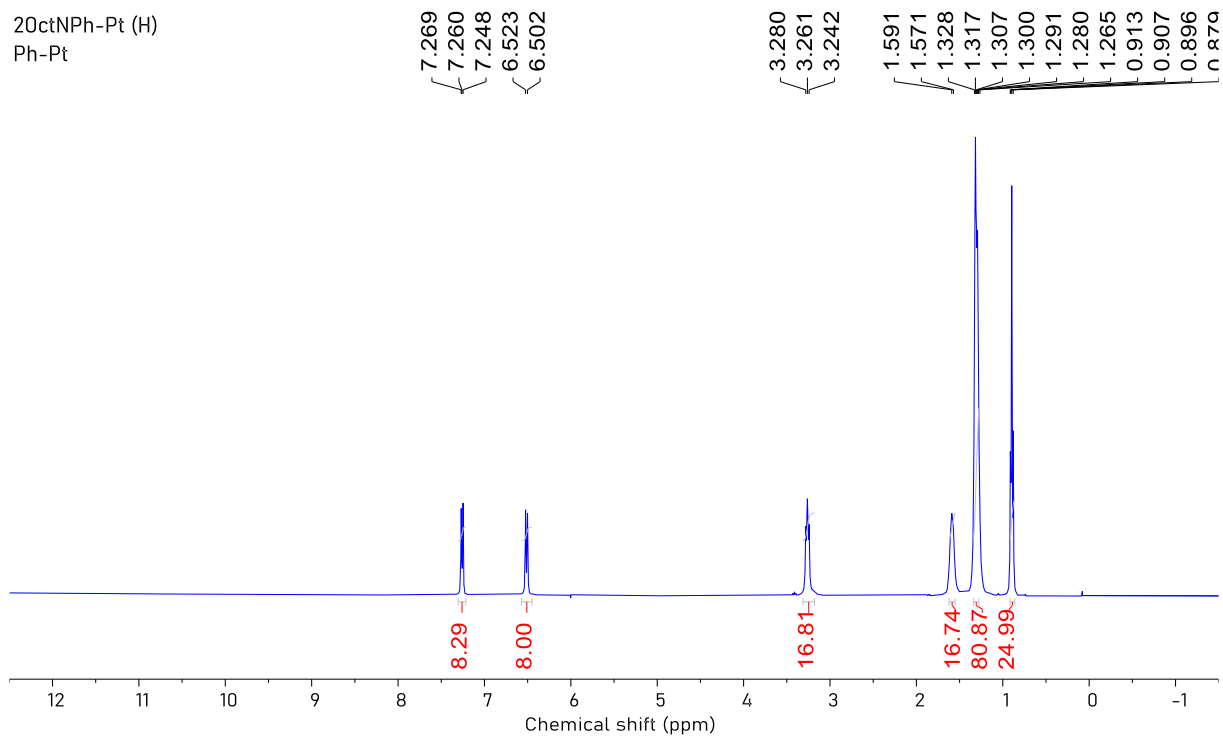
**Figure S4.** The  $^1\text{H}$ -NMR spectrum of PdPN in  $\text{CDCl}_3$ .



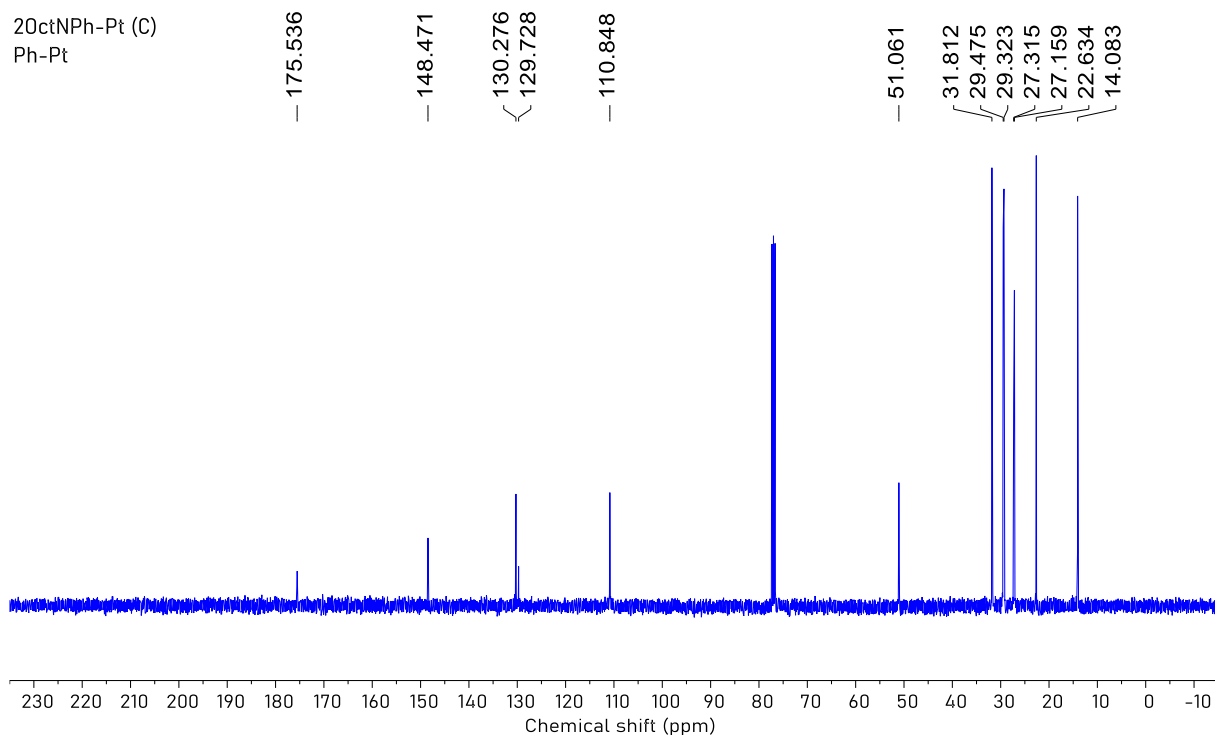
**Figure S5.** The  $^{13}\text{C}$ -NMR spectrum of PdPN in  $\text{CDCl}_3$ .



**Figure S6.** High-resolution mass spectrum of PdPN.



**Figure S7.** The  $^1\text{H}$ -NMR spectrum of PtPN in  $\text{CDCl}_3$ .



**Figure S8.** The  $^{13}\text{C}$ -NMR spectrum of PtPN in  $\text{CDCl}_3$ .

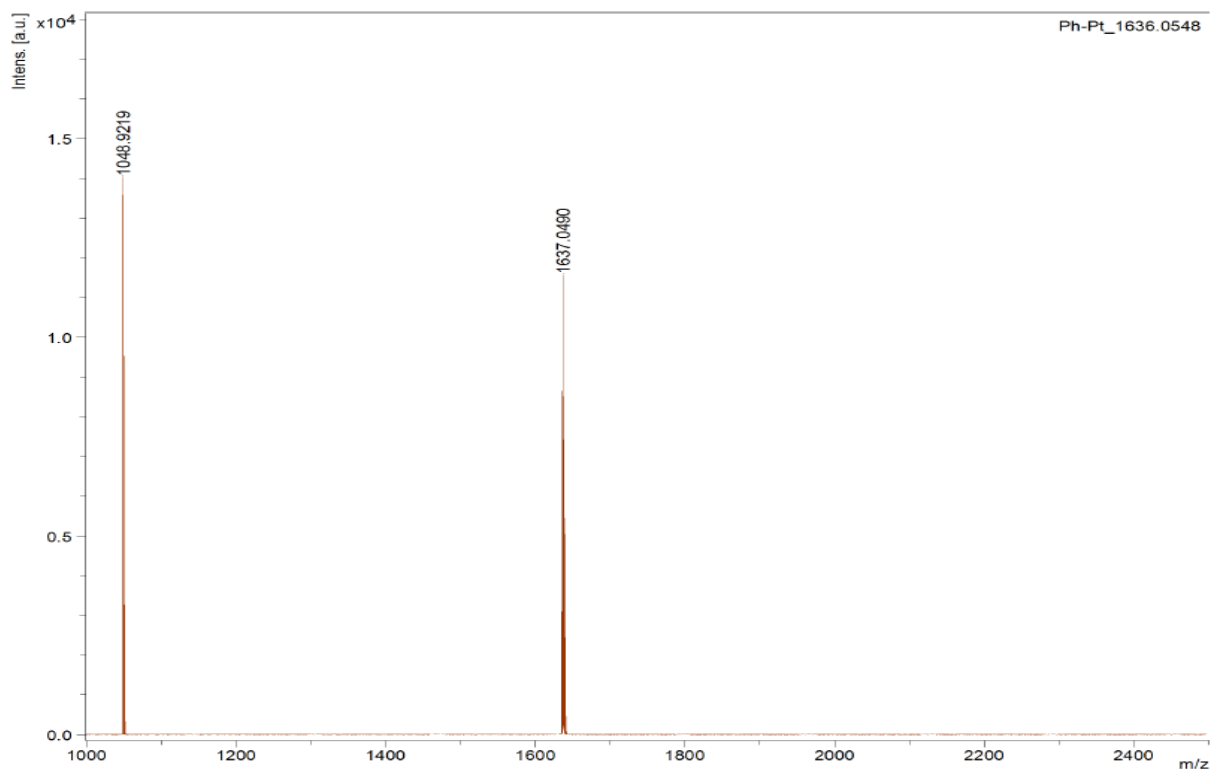


Figure S9. High-resolution mass spectrum of PtPN.

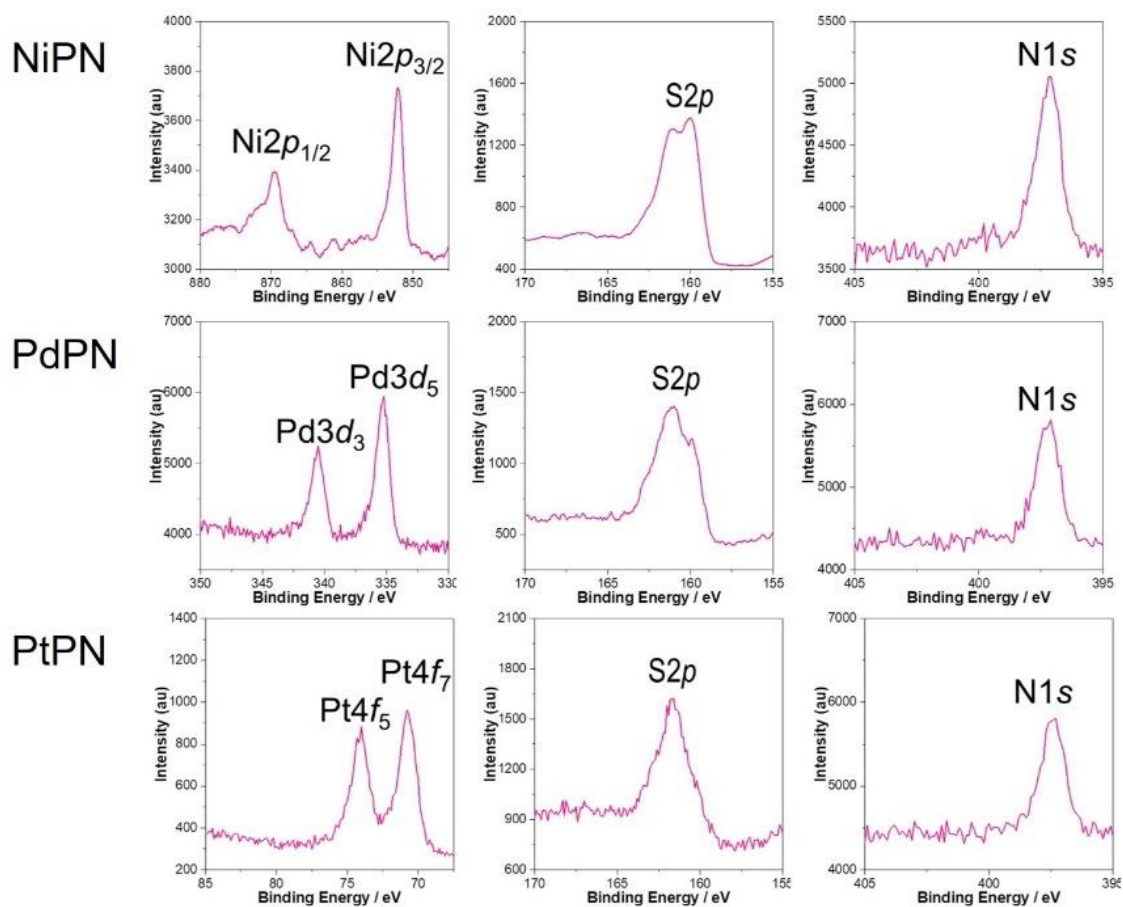
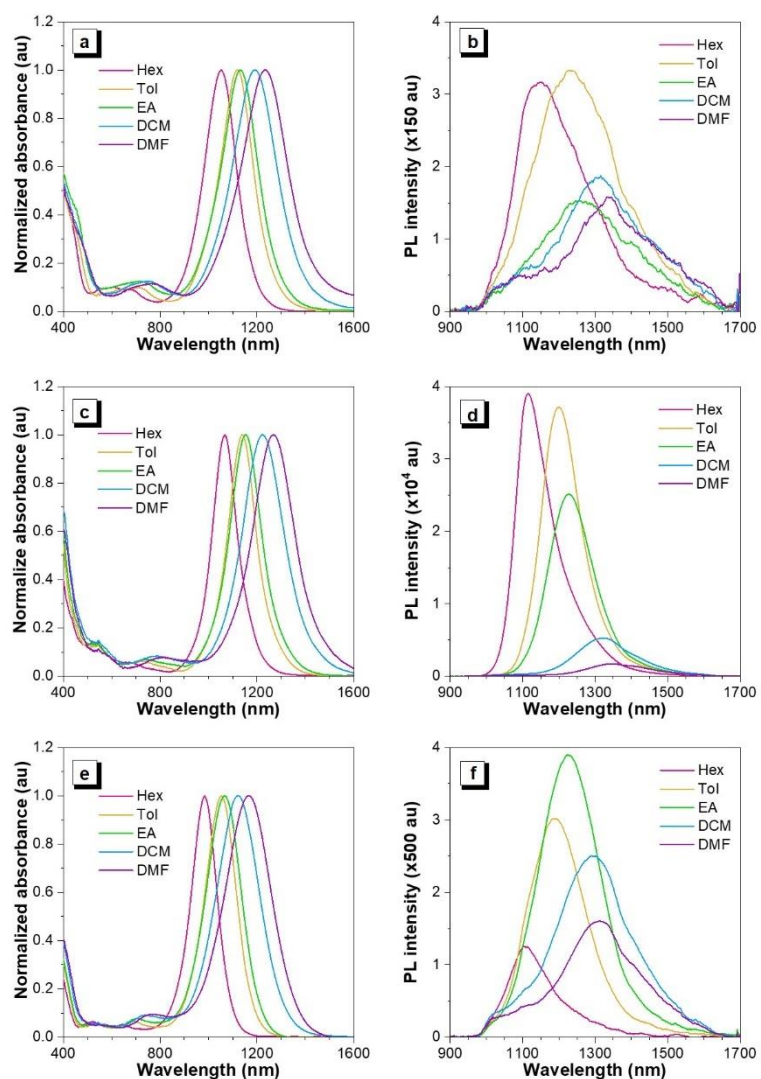
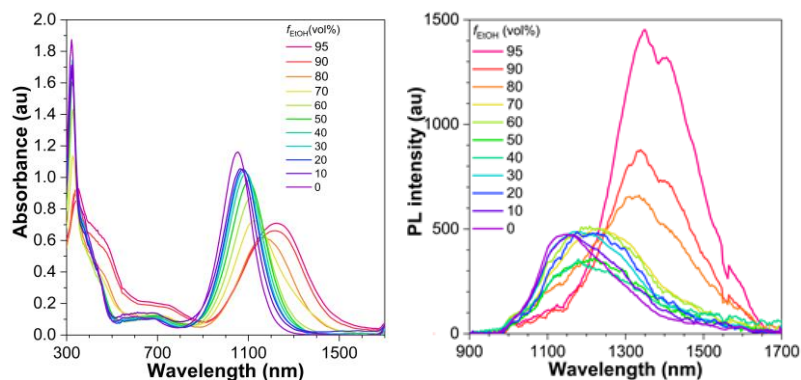


Figure S10. X-ray photoelectron spectroscopy (XPS) spectra of NiPN, PdPN, and PtPN powders.

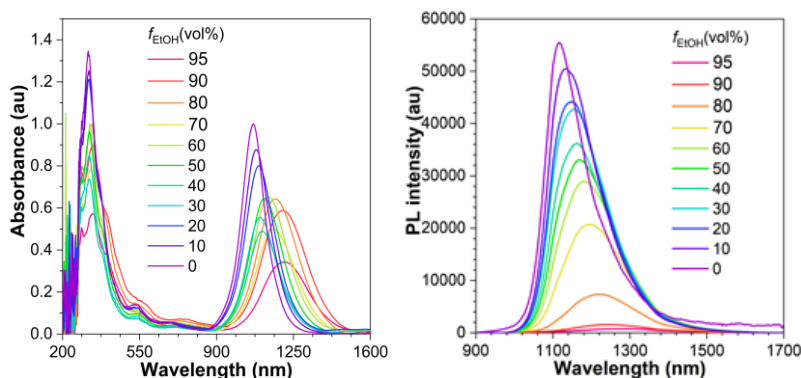




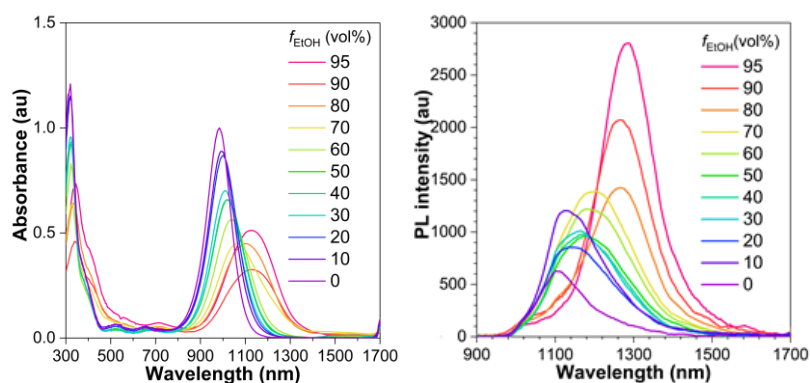
**Figure S11.** (a, c, and e) Normalized UV-vis-NIR absorption and (b, d, and f) PL spectra of NiPN (a, b), PdPN (c, d), and PtPN (e, f), respectively, in the different solvents (10  $\mu$ M). The excitation wavelength is 808 nm. (Hex: hexane; Tol: toluene; EA: ethyl acetate; DCM: dichloromethane; DMF: dimethylformamide)



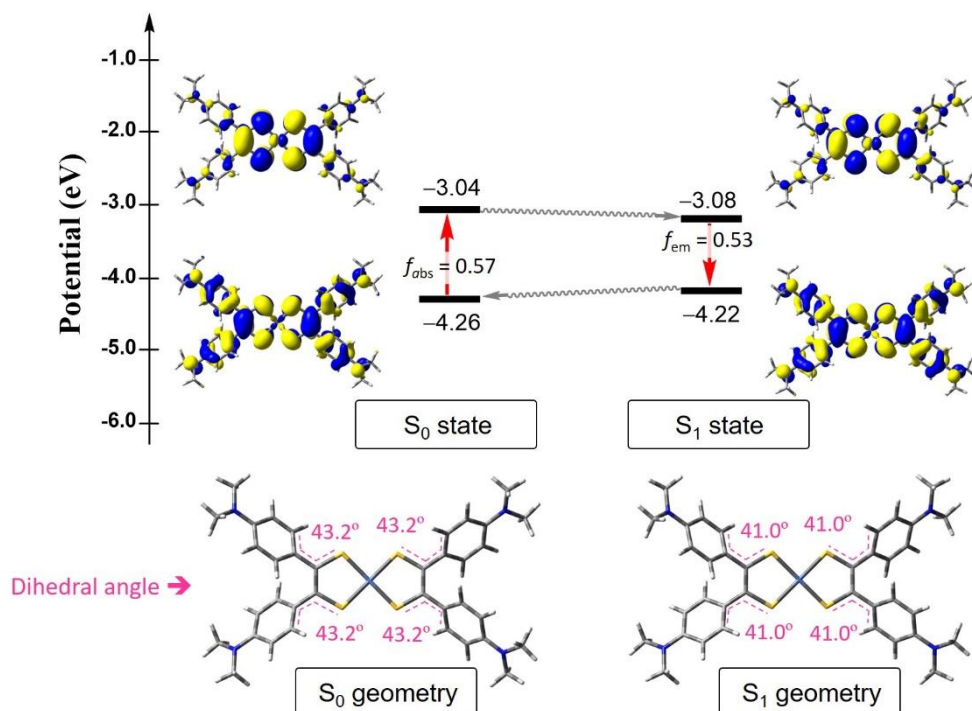
**Figure S12.** UV-vis-NIR absorption and PL spectra of NiPN with the different fractions of EtOH and hexane ( $f_{\text{EtOH}}$ , vol%) at a concentration of 10  $\mu\text{M}$ . The excitation wavelength is 808 nm.



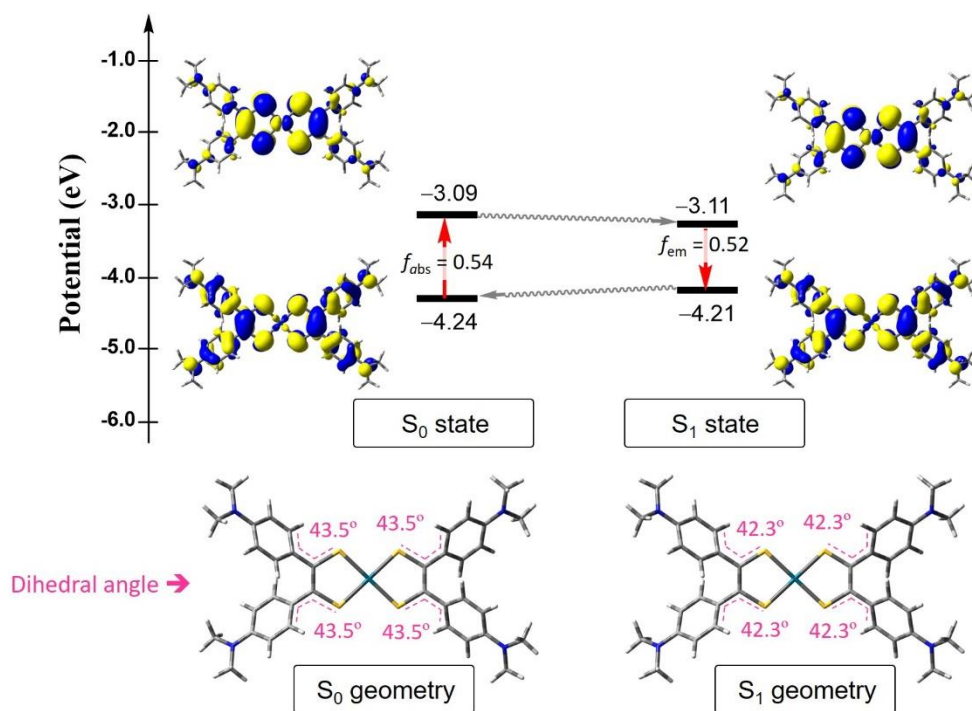
**Figure S13.** UV-vis-NIR absorption and PL spectra of PdPN with the different fractions of EtOH and hexane ( $f_{\text{EtOH}}$ , vol%) at a concentration of 10  $\mu\text{M}$ . The excitation wavelength is 808 nm.



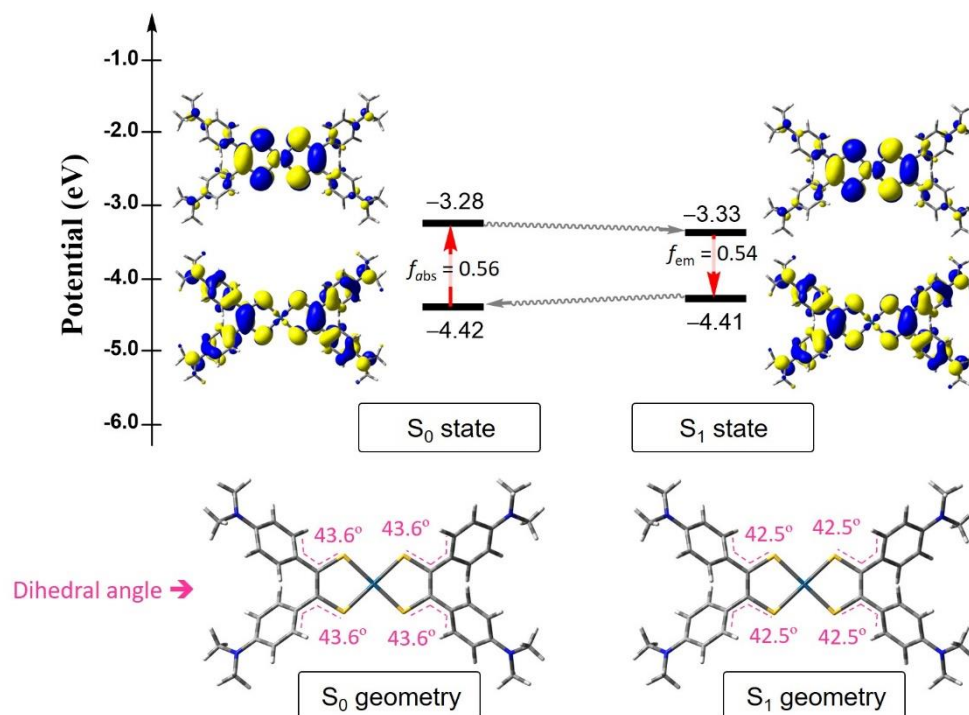
**Figure S14.** UV-vis-NIR absorption and PL spectra of PtPN with the different fractions of EtOH and hexane ( $f_{\text{EtOH}}$ , vol%) at a concentration of 10  $\mu\text{M}$ . The excitation wavelength is 808 nm.



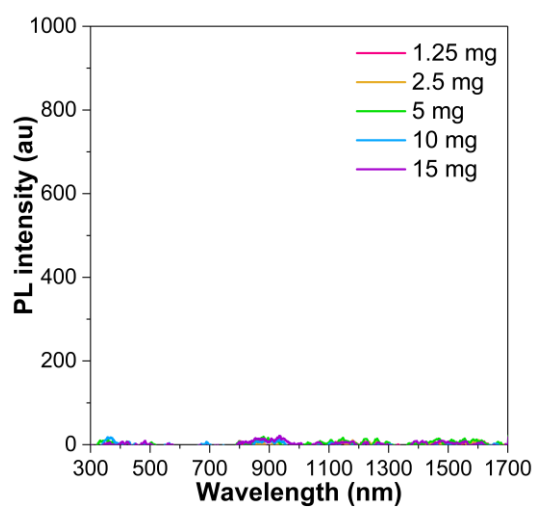
**Figure S15.** Molecular geometries, orbitals with isovalue of 0.02 a.u., energy levels, and dihedral angles of NiPN in the ground ( $S_0$ ) and excited ( $S_1$ ) states, calculated with TD-DFT at the level of B3LYP/6-31G(d)/LANL2DZ.



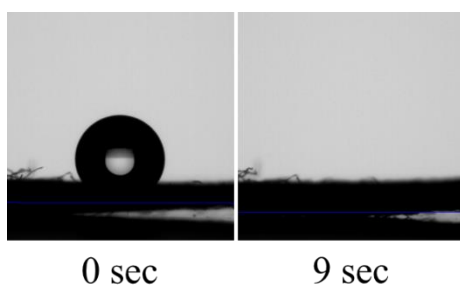
**Figure S16.** Molecular geometries, orbitals with isovalue of 0.02 a.u., energy levels, and dihedral angles of PdPN in the ground ( $S_0$ ) and excited ( $S_1$ ) states, calculated with TD-DFT at the level of B3LYP/6-31G(d)/LANL2DZ.



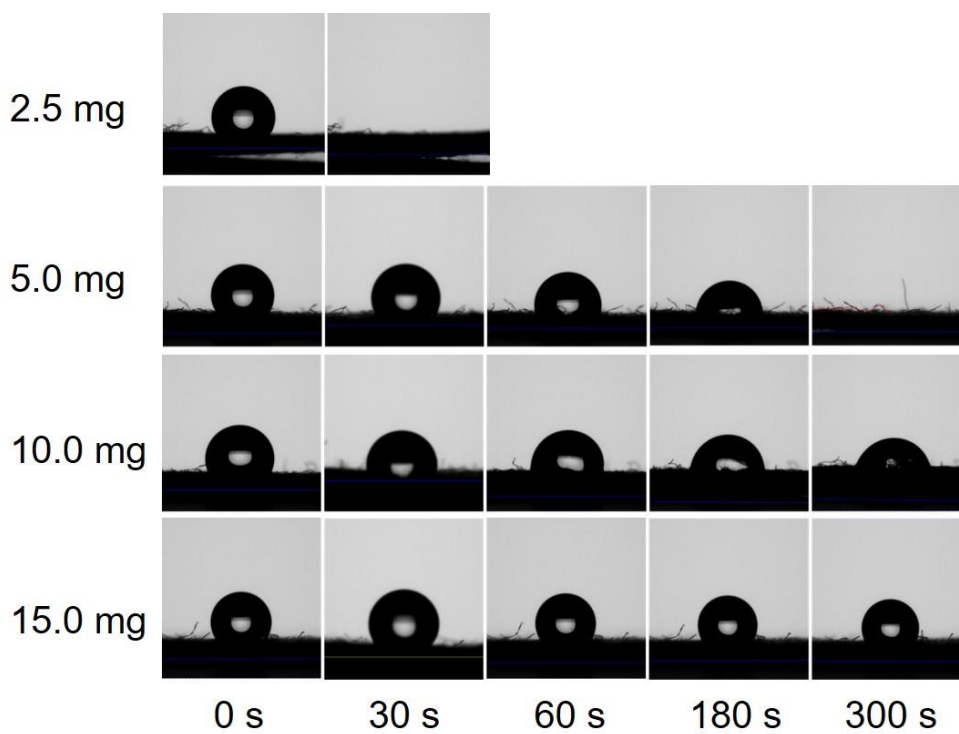
**Figure S17.** Molecular geometries, orbitals with isovalue of 0.02 a.u., energy levels, and dihedral angles of PtPN in the ground ( $S_0$ ) and excited ( $S_1$ ) states, calculated with TD-DFT at the level of B3LYP/6-31G(d)/LANL2DZ.



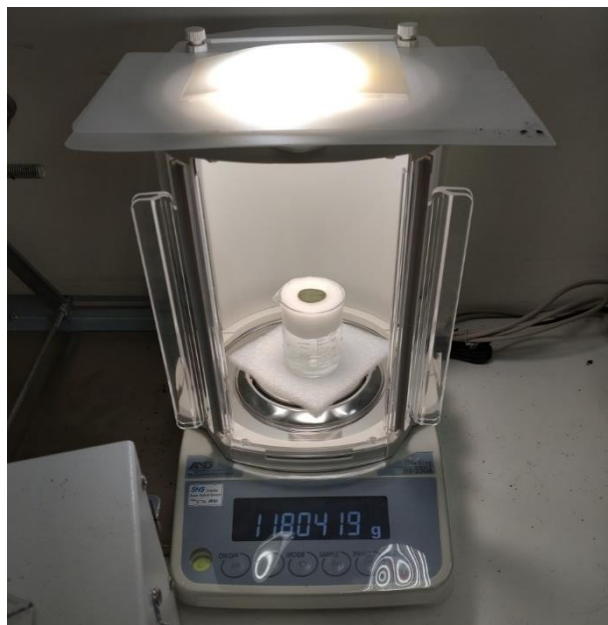
**Figure S18.** PL spectra of NiPN-adsorbed filter papers under the excitation wavelength of 808 nm.



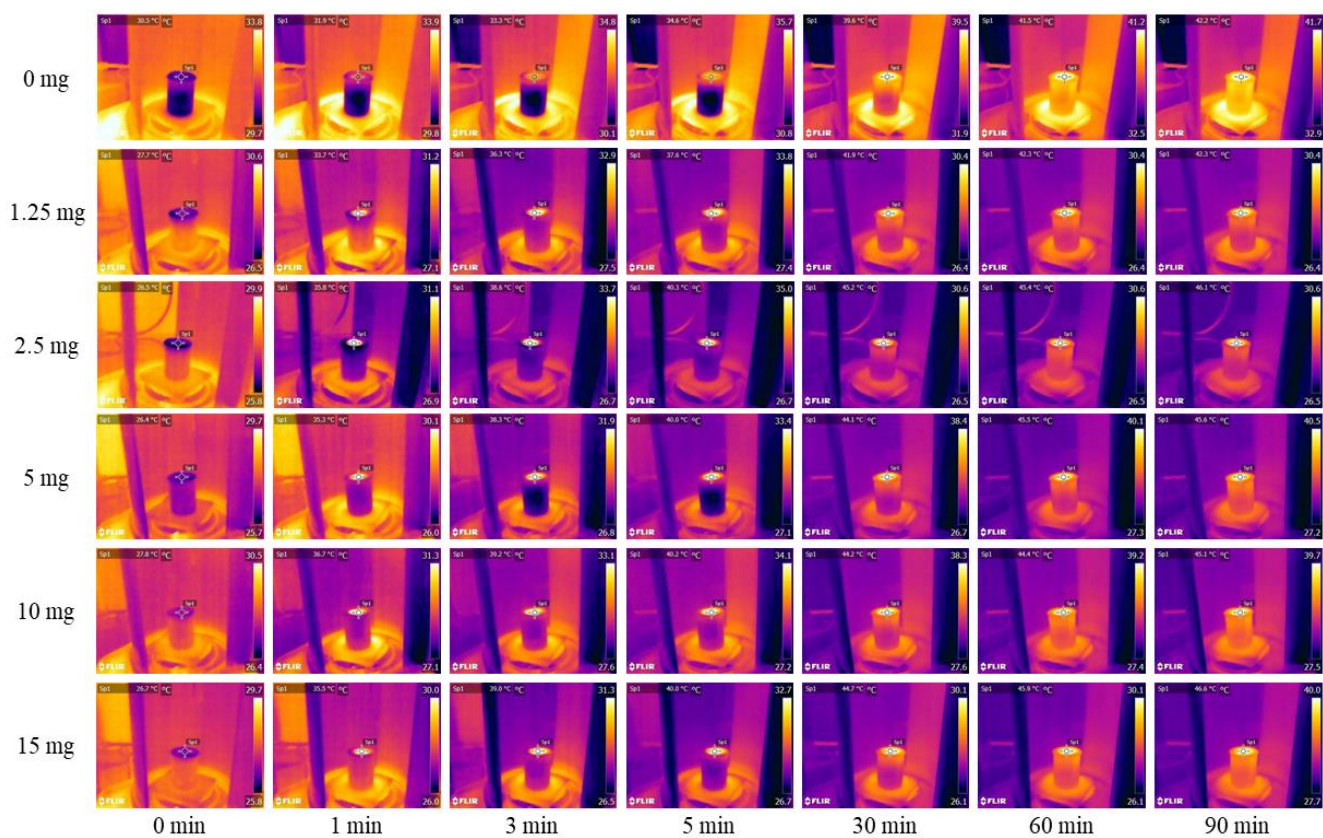
**Figure S19.** The time-dependent water contact angle (WCA) images of 1.25 mg of NiPN-loading filter papers.



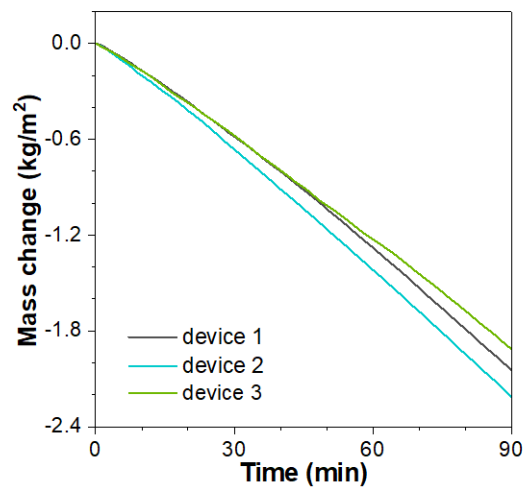
**Figure S20.** The time-dependent WCA images of NiPN-loading filter papers with different amounts, 2.5, 5.0, 10.0, and 15.0 mg, respectively.



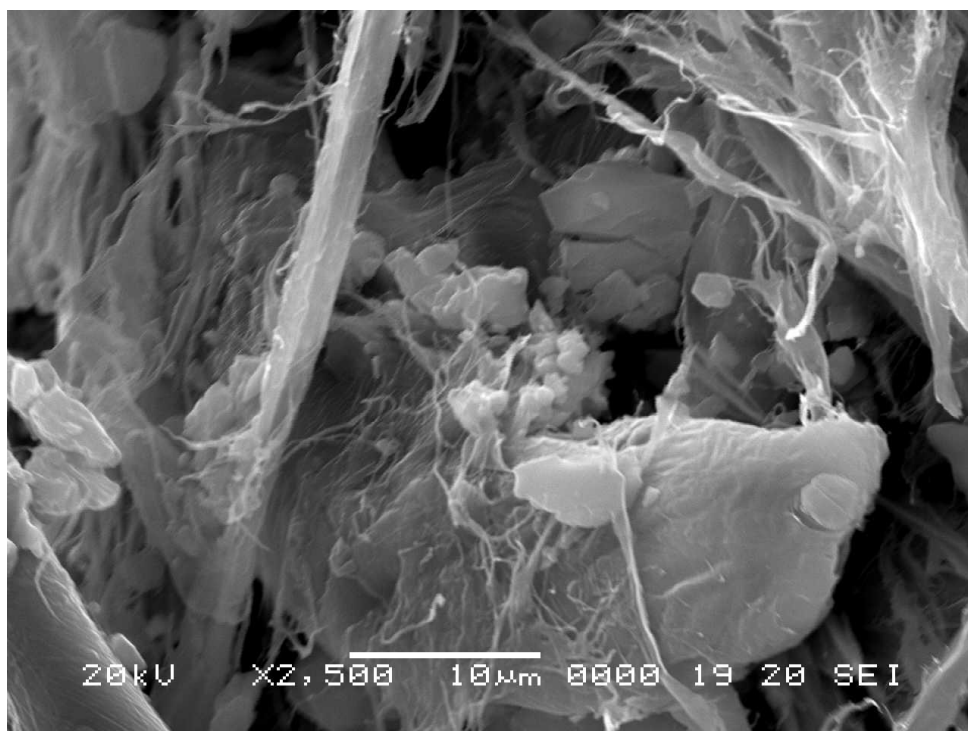
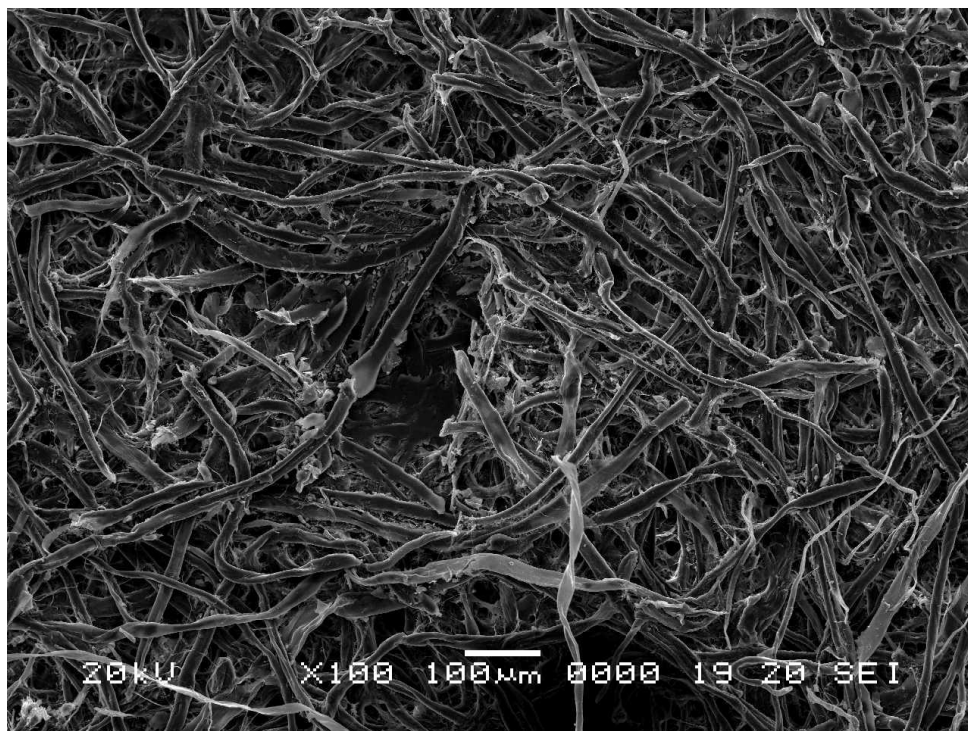
**Figure S21.** Camera photos of solar-driven evaporation device.



**Figure S22.** Infrared images of the solar-thermal interfacial-heating evaporation layers adsorbed with the different amounts of NiPN under 1 sun irradiation during time.



**Figure S23.** Time-dependent water-mass change curves using 2.5 mg of activated carbon-loading device under 1 sun irradiation.



**Figure S24.** Scanning electron microscope images of NiPN loading filter paper, i.e., NiNP-adsorbed evaporators.



**Table S1.** Calculation data of organometallic complexes in  $S_0$  state.

Molecules	States	Configurations	$E$ (eV)	$\lambda$ (nm)	$f_{os}$
NiNP	$S_{01}$	H $\rightarrow$ L (96%)	1.18	1050.05	0.565
	$S_{02}$	H-7 $\rightarrow$ L+1 (3%), H-5 $\rightarrow$ L (12%), H-2 $\rightarrow$ L (83%)	1.52	818.26	0.002
	$S_{03}$	H-6 $\rightarrow$ L (3%), H-1 $\rightarrow$ L (91%), H $\rightarrow$ L+1 (6%)	1.56	796.62	0.000
PdNP	$S_{01}$	H $\rightarrow$ L (96%)	1.13	1093.18	0.536
	$S_{02}$	H-6 $\rightarrow$ L (4%), H-1 $\rightarrow$ L (92%), H $\rightarrow$ L+1 (4%)	1.52	818.24	0.000
	$S_{03}$	H-5 $\rightarrow$ L (2%), H-2 $\rightarrow$ L (97%)	1.55	798.16	0.004
PtNP	$S_{01}$	H $\rightarrow$ L (97%)	1.09	1136.01	0.563
	$S_{02}$	H-5 $\rightarrow$ L (6%), H-2 $\rightarrow$ L (93%)	1.33	931.58	0.002
	$S_{03}$	H-1 $\rightarrow$ L (98%)	1.42	875.25	0.000

\* Calculated with TD-DFT at the level of B3LYP/6-31G(d)/LANL2DZ.  $S_{01}$ ,  $S_{02}$ , and  $S_{03}$  denoted the first, second, and third vertical transition from the  $S_0$  state to the  $S_1$ ,  $S_2$ , and  $S_3$ , respectively, and  $f_{os}$  denoted oscillator strength between the ground and excited states.

**Table S2.** Calculation data of organometallic complexes in  $S_0$  state.

Molecules	States	Configurations	$E$ (eV)	$\lambda$ (nm)	$f_{os}$
NiNP	$S_{10}$	H $\rightarrow$ L (96%)	1.12	1111.46	0.534
PdNP	$S_{10}$	H $\rightarrow$ L (96%), H-6 $\rightarrow$ L+1 (2%)	1.09	1140.88	0.520
PtNP	$S_{10}$	H $\rightarrow$ L (98%)	1.04	1187.31	0.536

Calculated with TD-DFT at the level of B3LYP/6-31G(d)/LANL2DZ.  $S_{10}$  denoted the first vertical transition from the  $S_1$  to  $S_0$  states, and  $f_{os}$  denoted oscillator strength between the ground and excited states.

## Supporting References

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- S9 Zhang, R.; Jin, N.; Jia, T.; Wang, L.; Liu, J.; Nan, M.; Qi, S.; Liu, S.; Pan, Y., *J. Mater. Chem. A* **2023**, *11*, 15380-15388.