Supplementary Information (SI)

Regulating the Kinetic Behaviors of Polysulfides by Designing an Au-COF Interface in Lithium-Sulfur Batteries

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Fig. S1 Synthetic scheme of the NUS-2.



Fig. S2 SEM images of NUS-2 powder at different magnifications.



Fig. S3 TEM images of NUS-2 powder at different magnifications.



Fig. S4 TEM and corresponding elemental mapping images of NUS-2 powder.



Fig. S5 TEM image of Au NPs.



Fig. S6 TEM images of Au-COF at different magnification.



Fig. S7 Zeta Potential of various materials.



Fig. S8 SEM and corresponding elemental mapping images of Au-COF/rGO powder.



Fig. S9 TEM and corresponding elemental mapping images of Au-COF/rGO powder.



Fig. S10 Photographs of static liquid electrolyte contact angles of different separators at different rest time. 1 M LiTFSI with 2 wt% $LiNO_3$ in DOL/DME (v/v=1:1) was used as a liquid electrolyte.



Fig. S11 Li ion conductivity of symmetric cells with various separators.



Fig. S12 Permeation experiments. Digital photographs of H-type devices with original PP separator and Au-COF/rGO coated PP separator. The inside of left chamber is Li_2S_6 in DOL/DME (v/v=1:1) solution, while the inside of right chamber is the pure DOL/DME (v/v=1:1) solvent.



Fig. S13 CV curves at different scan rates and the corresponding linear fits of the peak current of the cells with the (a, b) rGO and (c, d) COF/rGO separators.



Fig. S14 EIS curves of Au-COF/rGO, COF/rGO and rGO symmetry cells.



Fig. S15 First five charge-discharge curves of the Li-S batteries with various separators at 0.2 C.



Fig. S16 Side-view SEM images of the Au-COF/rGO layer with different thickness, which was constructed by adjusting the total loading amount of the three components based on the mass ratio of Au: COF: rGO=1: 1: 1. The loading amount of Au: COF: rGO: (a) 1.0 mg: 1.0 mg: (b) 1.5 mg: 1.5 mg: 1.5 mg; (c) 2.0 mg: 2.0 mg: 2.0 mg; (d) 2.5 mg: 2.5 mg: 2.5 mg.



Fig. S17 Charge-discharge voltage profiles of Li-S cells with different interface after 1000 cycles.



Fig. S18 SEM images of Li metal anode after cycling in the Li-S cells with different interface: (a) PP; (b) Au-COF/rGO. The inset presents a corresponding digital photo of Li metal anode.



Fig. S19 SEM images of Au-COF/rGO a before and b after cycling.



Fig. S20 Charge-discharge voltage profiles at 0.1 C of Au-COF/rGO-based Li-S cell with a high sulfur loading of 3.7 mg cm^{-2} .

| Separator | D _{Li+} at peak A [cm² s ⁻¹] | D _{Li+} at peak B [cm ² s ⁻¹] | D _{Li+} at peak C [cm² s ⁻¹] | |
|---------------|--|--|--|--|
| РР | 6.3×10 ⁻⁸ | 1.9×10 ⁻⁸ | 9.5×10-9 | |
| rGO/PP | 1.0×10 ⁻⁷ | 4.7×10 ⁻⁸ | 2.4×10-9 | |
| COF/rGO/PP | 1.6×10-7 | 2.9×10 ⁻⁸ | 3.2×10 ⁻⁸ | |
| Au-COF/rGO/PP | 2.3×10-7 | 3.3×10 ⁻⁸ | 2.9×10 ⁻⁸ | |

Table S1 Li^+ diffusion coefficients of various separators.

 Table S2 Discharge specific capacity of the Li-S cells with different mass rations of

 Au decorated COF/rGO interface.

| Separator | Original capacity (mA h g ⁻¹) | Capacity after 500 cycles (mA h g ⁻¹) | Capacity decay rate (%) | |
|---------------|--|--|----------------------------|--|
| 0.5Au-COF/rGO | 1204 | 458 | 0.124 | |
| 1.0Au-COF/rGO | 1221 | 719 | 0.082 | |
| 1.5Au-COF/rGO | 1265 | 515 | 0.119 | |
| 2.0Au-COF/rGO | 1135 | 452 | 0.120 | |

Table S3 Discharge specific capacity of the Li-S cells with Au-COF/rGO interface in

| different thicknesses based on the mass ratio of Au: COF: rGO=1: 1: | 1. |
|---|----|
|---|----|

| Separator | Original capacity (mA h g ⁻¹) | Capacity after 500 cycles (mA h g ⁻¹) | Capacity decay rate(%) | |
|-----------|--|--|---------------------------|--|
| L-1.0 | 1094 | 458 | 0.162 | |
| L-1.5 | 1221 | 719 | 0.082 | |
| L-2.0 | 1137 | 536 | 0.163 | |
| L-2.5 | 883 | 369 | 0.163 | |

Table S4 The comprehensive comparison of the electrochemical performance of Li-Scells with various interface/separators.

| Separator | Max | Rate | | | | |
|---------------------------|-------------------------------------|---|-------|-----------------|-------------------------------|------------|
| | Capacity (mA h g ⁻¹) | Capacity (mA h g ⁻¹) Rat | Rate | Cycle Number | Capacity Fading Rate/Cycle | Reference |
| SCOF-2 | 1219 (0.1 C) | 470 (5 C) | 1 C | 800 | 0.047% | S1 |
| S/D-[4+3]COFs2-3 | 1350 (0.1 C) | 632 (2.5 C) | 0.5 C | 500 | 0.03% | S2 |
| Li-CON@GN/Celgard | 1622 (0.1 C) | 782 (4 C) | 1 C | 600 | 0.057% | \$3 |
| COF-1 NN | 1120 (0.2 C) | 580 (10 C) | 2 C | 300 | 0.053% | S 4 |
| R-COF-BTD@S | 1600 (0.1 C) | 700 (2 C) | 0.5 C | 200 | 0.012% | 85 |
| TPB-DMTP-COF | 880 (0.5 C) | 480 (2.5 C) | 1 C | 800 | 0.05% | S6 |
| TpPa-SO3Li/CNT/Celgard | 1590 (0.1 C) | 610 (4 C) | 4 C | 400 | 0.039% | S7 |
| TpPa-SO ₃ H@PP | 1290 (0.1 C) | 490 (4 C) | 1 C | 500 | 0.05% | S8 |
| CTP-1 | 1500 (0.05 C) | 600 (5 C) | 1 C | 800 | 0.048% | S 9 |
| Au-COF/rGO | 1420 (0.2 C) | 568 (4 C) | 1 C | 1000 | 0.047% | This work |

| | Cycling Performance | | | | | | |
|---------------------------|---------------------|--|-----------------|---|-------------------------------|-----------|--|
| Separator | Rate (C) | Max Capacity (mA h g ⁻¹) | Cycle Number | Sulfur loading (mg cm ⁻²) | Capacity Fading Rate/Cycle | Reference | |
| SCOF-2 | 0.2 | 855 | 100 | 3.2 | 82.5% | S1 | |
| TpPa–SO ₃ H@PP | 0.2 | 800 | 100 | 5.0 | 75.0% | S8 | |
| COF-TPT(OH)@S | 0.1 | 1080 | 10 | 5.0 | 90.1% | S10 | |
| COF-MF@S | 0.2 | 905 | 150 | 3.7 | 80.3% | S11 | |
| COF-PDA/SWCNT | 0.1 | 922 | 60 | 4.5 | 75.5% | S12 | |
| 4F-COF | 0.05 | 844 | 200 | 9.0 | 74.0% | S13 | |
| S/D- [4+3] COFs2-3 | 0.5 | 795 | 50 | 4.0 | 74.0% | S14 | |
| COF66 | 0.2 | 1015 | 100 | 3.6 | 76.5% | S15 | |
| Au-COF/rGO | 0.1 | 1162 | 50 | 3.7 | 82.8% | This work | |

Table S5 The comprehensive comparison of the electrochemical performance of Li-Scells with various functional interface/separators at high sulfur loading.

Supporting References

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