

Supplementary Information

Synergistic carrier and phonon transport advance Ag dynamically-doped *n*-type PbTe thermoelectrics *via* Mn alloying

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1. Computational details

1.1 Single parabolic band model

The Pisarenko curves and the effective mass of n -type PbTe can be modeled by the single parabolic band (SPB) model, and the SPB model is based on the following equations:¹

Seebeck coefficient:

$$S = \pm \frac{k_B}{e} \left(\frac{(r + 5/2)F_{r+2/3}(\eta)}{(r + 3/2)F_{r+1/2}(\eta)} - \eta \right) \quad (\text{eq. S1})$$

Hall carrier concentration:

$$n_H = 4\pi \left[\frac{2m^* k_B T}{\hbar^2} \right]^{3/2} F_{1/2} \quad (\text{eq. S2})$$

Lorenz number:

$$L = \frac{k_B^2 3F_0 F_2 - 4F_1^2}{e^2 F_0^2} \quad (\text{eq. S3})$$

where k_B is the Boltzmann constant, \hbar is the reduce Planck constant, η is the reduced Fermi level, and r denotes the scattering factor and equals -1/2 here assuming that the acoustic scattering mechanism dominates. The Fermi integral is given by:

$$F_n(\eta) = \int_0^\infty \frac{x^n}{1 + e^{x-\eta}} dx \quad (\text{eq. S4})$$

1.2 Single Kane band model

The thermoelectric-transport properties of n -type PbTe can be modeled by adopting the single Kane band (SKB) model. The transport coefficients are determined by the equations as follows:^{2,3}

Hall carrier concentration:

$$n_H = \frac{1}{eR_H} = A^{-1} \frac{N_V (2m_b^* k_B T)^{3/2}}{3\pi^2 \hbar^3} {}^0F_0^{3/2} \quad (\text{eq. S5})$$

Hall factor:

$$A = \frac{3K(K+2) {}^0F_{-4}^{1/20} {}^0F_0^{3/2}}{(2K+1)^2 ({}^0F_{-2}^1)^2} \quad (\text{eq. S6})$$

Hall Carrier mobility:

$$\mu_H = A \frac{2\pi\hbar^4 e C_l}{m_I^* (2m_b^* k_B T)^{3/2} E_{def}^2} \frac{3^0 F_{-2}^1}{0^0 F_{-2}^{3/2}} \quad (\text{eq. S7})$$

Power factor:

$$PF = \frac{2N_V \hbar k_B^2 C_l}{\pi E_{def}^2} \frac{1}{m_I^*} \left(\frac{1^1 F_{-2}^1}{0^0 F_{-2}^1} - \xi \right) 2^0 F_{-2}^1 \quad (\text{eq. S8})$$

nF_k^m has a similar form as the Fermi integral:

$$nF_k^m = \int_0^\infty \left(-\frac{\partial f}{\partial \varepsilon} \right) \varepsilon^n (\varepsilon + \alpha \varepsilon^2)^m [(1 + 2\alpha\varepsilon)^2 + 2]^{k/2} d\varepsilon \quad (\text{eq. S9})$$

Where ξ is the reduced chemical potential, N_V is the band degeneracy ($N_V=4$ for n -type PbTe),⁴ K is the band anisotropy ($K=3.6$ for n -type PbTe),⁵ k_B is the Boltzmann constant, \hbar is the reduced Planck constant, m_I^* is the inertial mass ($m_I^* = 3m_d^*/(N_V^{2/3}(2K^{1/3} + K^{-2/3}))$),⁶ C_l is the longitudinal elastic ($C_l=7.1\times 10^{10}$ Pa for n -type PbTe),^{7, 8} ε is the reduced energy of electronic state, α is the reciprocal reduced band gap ($\alpha = k_B T/E_g$), E_g is the band gap, $E_g = 0.18eV + 0.0004 eV/K\times T$, E_{def} is the deformation potential coefficient.

1.3 Calculation of lattice thermal conductivity κ_{lat}

The lattice thermal conductivity (κ_{lat}) of the $\text{Ag}_{0.03}\text{Pb}_{1-x}\text{Mn}_x\text{Te}$ samples was calculated by the modified Debye-Callaway model, which can be expressed by:

$$\kappa_{\text{lat}} = \frac{k_B}{2\pi^2 v} \left(\frac{k_B}{\hbar} \right)^3 \int_0^{\Theta/T} \tau_{\text{tot}}(x) \frac{x^4 e^x}{(e^x - 1)^2} dx \quad (\text{eq. S10})$$

Here, $v = 3^{1/3}(\nu_l^{-3} + 2\nu_t^{-3})^{-1/3}$ is the average speed of phonon, ν_l and ν_t denote the longitudinal and transverse speeds of sound respectively, \hbar is the reduced Planck constant, Θ is the Debye temperature, x is the relation of $\hbar\omega/k_B T$, ω is the phonon frequency, and τ_{tot} is the total phonon scattering relaxation time. The total phonon relaxation time τ_{tot} for the $\text{Ag}_{0.03}\text{Pb}_{1-x}\text{Mn}_x\text{Te}$ samples consists of Umklapp process, normal process, point defects, grain boundary, and nanoprecipitates (U+N+PD+GB+NP) by the following equation:⁹

$$\tau_{\text{tot}}^{-1} = \tau_U^{-1} + \tau_N^{-1} + \tau_{PD}^{-1} + \tau_{GB}^{-1} + \tau_{NP}^{-1} \quad (\text{eq. S11})$$

The phonon scattering relaxation time for respective mechanism can be expressed as follows:
Umklapp phonon scattering:

$$\tau_U^{-1} = A_N * \frac{2}{(6\pi^2)^{\frac{1}{3}}} \frac{k_B \bar{V}^{1/3} \omega^2 \gamma^2 T}{M v^3} \quad (\text{eq. S12})$$

Normal phonon scattering:

$$\tau_N^{-1} = \frac{2}{(6\pi^2)^{\frac{1}{3}}} \frac{K_B V^{1/3} \omega^2 \gamma^2 T}{M v^3} \quad (\text{eq. S13})$$

Point defect scattering:

$$\tau_{PD}^{-1} = \frac{V_0^4}{4\pi^3 v^3} * \sum (1-x_i) \left[\left(\frac{M_i - M}{M} \right)^2 + \varepsilon \left(\frac{a_i - a}{a} \right)^2 \right] \quad (\text{eq. S14})$$

Grain boundary scattering:

$$\tau_{GB}^{-1} = v/Gd \quad (\text{eq. S15})$$

Nanoprecipitates scattering:

$$\tau_{NP}^{-1} = v \left[(2\pi R^2)^{-1} + \left(\frac{4}{9} \pi R^2 (\Delta D/D)^2 \left(\frac{\omega R}{v} \right)^4 \right)^{-1} \right]^{-1} N_p \quad (\text{eq. S16})$$

In above equations, \bar{V} is the average atomic volume, \bar{M} is the average atomic mass, γ is the Grüneisen parameter, A_N is the ratio between normal process and Umklapp phonon scattering.

1.4 Calculation of quality factor B

The quality factor B of the $\text{Ag}_{0.03}\text{Pb}_{1-x}\text{Mn}_x\text{Te}$ samples was calculated by:¹⁰

$$B = \left(\frac{k_B}{e} \right)^2 \frac{8\pi e (2m_e k_B T)^{3/2} \mu_W}{3h^3 \kappa_{lat}} T \quad (\text{eq. S17})$$

2. Supplementary Figures

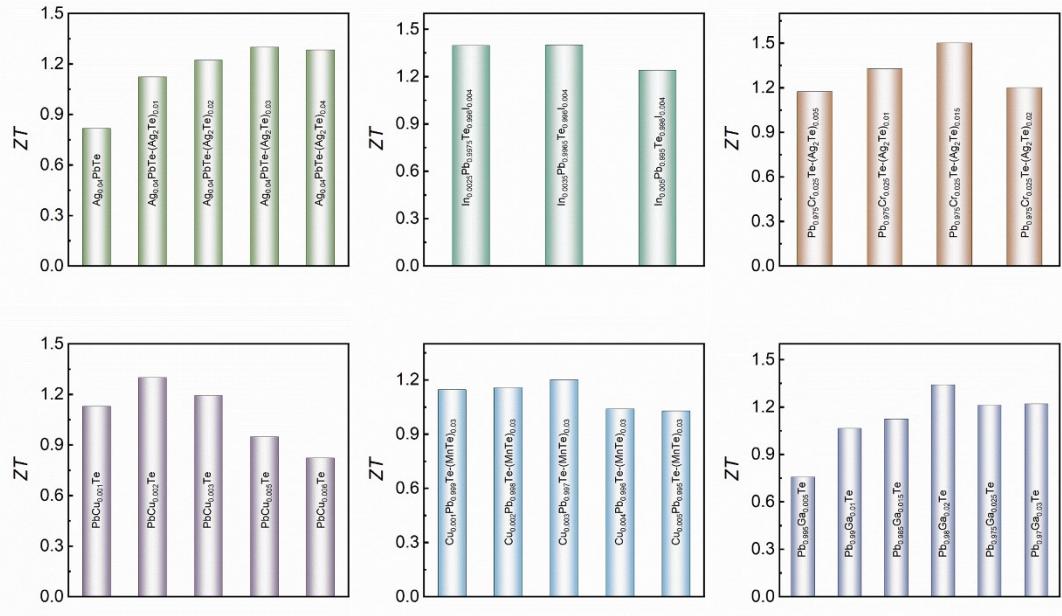


Figure S1. ZT of the main components of dynamic doped PbTe systems.¹¹⁻¹⁶

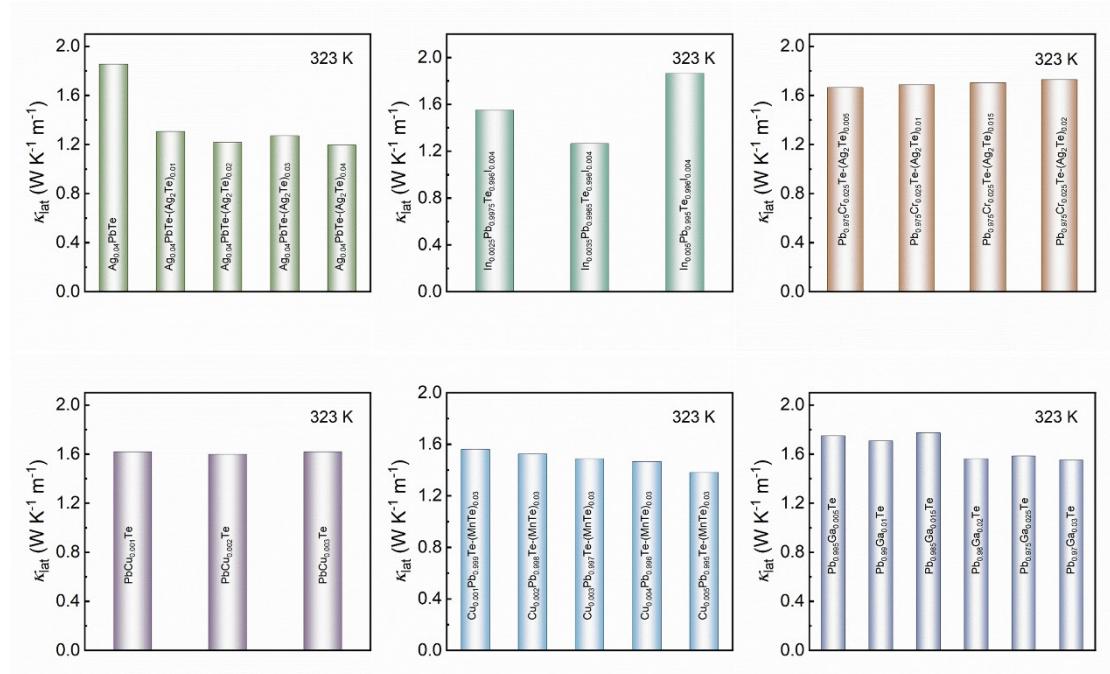


Figure S2. κ_{lat} at 323 K of the main components of dynamic doped PbTe systems.¹¹⁻¹⁶ the Lorenz number calculation methods for representative samples of each system: single parabolic band (SPB) model: PbTe:Cr/Ag₂Te, PbTe:Ag, PbTe:Cu/MnTe, PbTe:Ga; single Kane band (SKB) model: PbTe:In/I, PbTe:Cu.

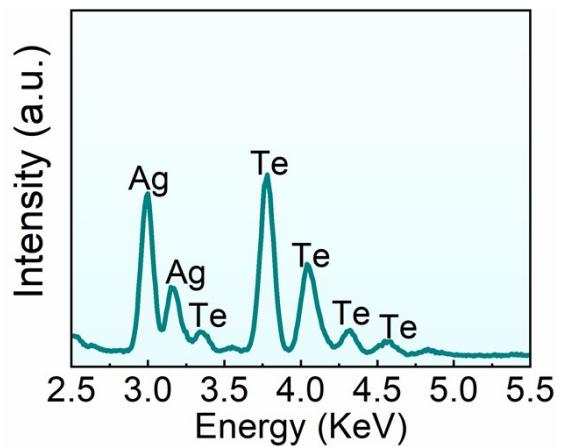


Figure S3. EDS spectra obtained from Ag_2Te secondary phase.

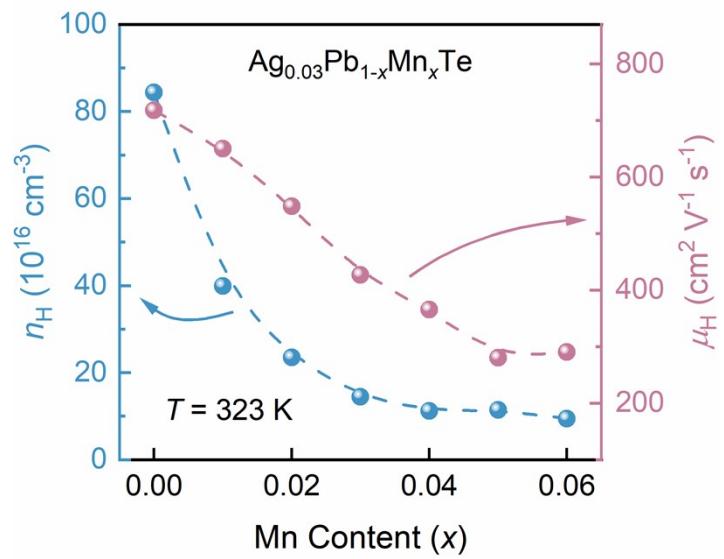


Figure S4. The Hall carrier concentration n_{H} and the Hall carrier mobility μ_{H} as a function of x in $\text{Ag}_{0.03}\text{Pb}_{1-x}\text{Mn}_x\text{Te}$.

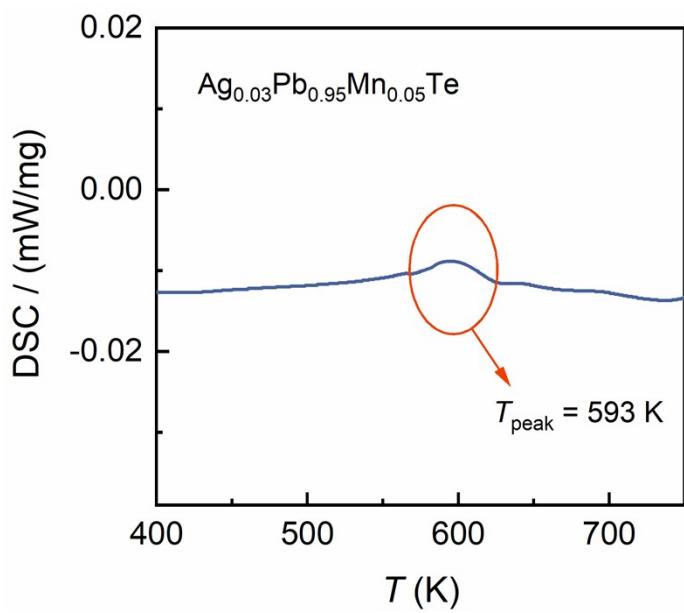
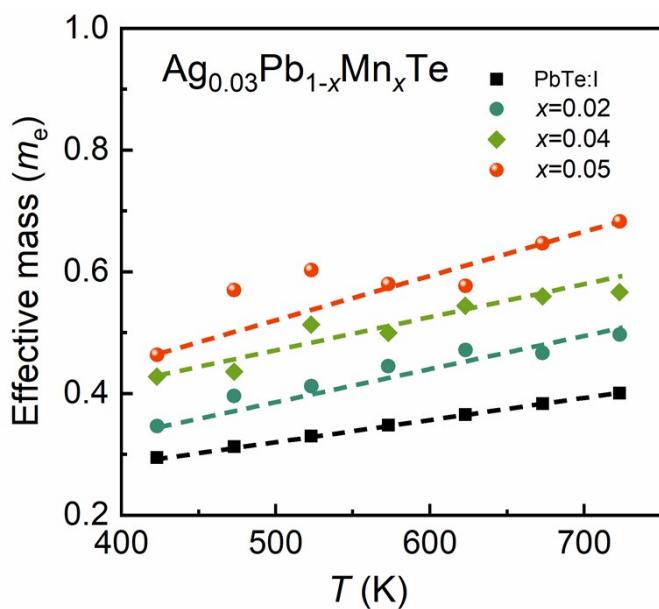


Figure S5. DSC curve of $\text{Ag}_{0.03}\text{Pb}_{0.95}\text{Mn}_{0.05}\text{Te}$, and the peak occurs at 593 K



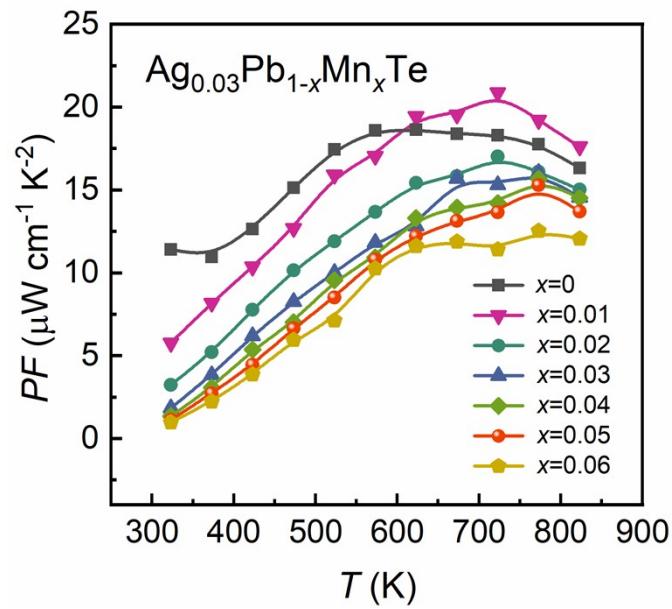


Figure S7. Temperature-dependent power factor PF for n -type $\text{Ag}_{0.03}\text{Pb}_{1-x}\text{Mn}_x\text{Te}$ samples.

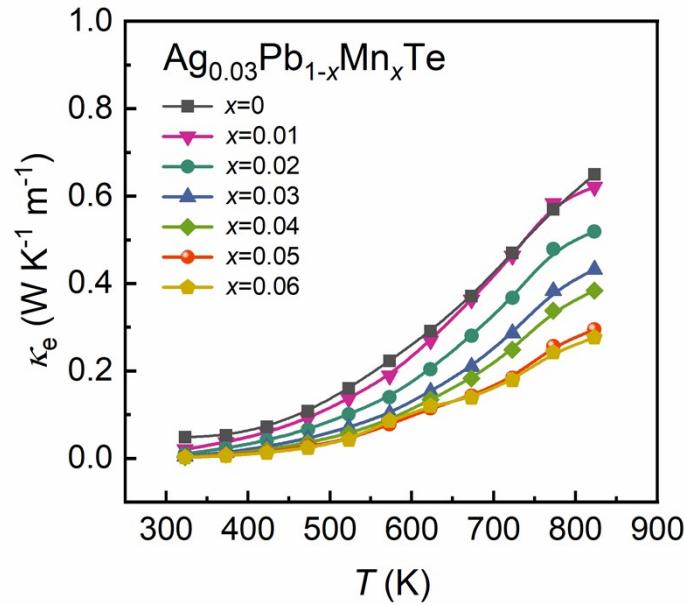


Figure S8. Temperature-dependent electrical thermal conductivity κ_e for n -type $\text{Ag}_{0.03}\text{Pb}_{1-x}\text{Mn}_x\text{Te}$ samples.

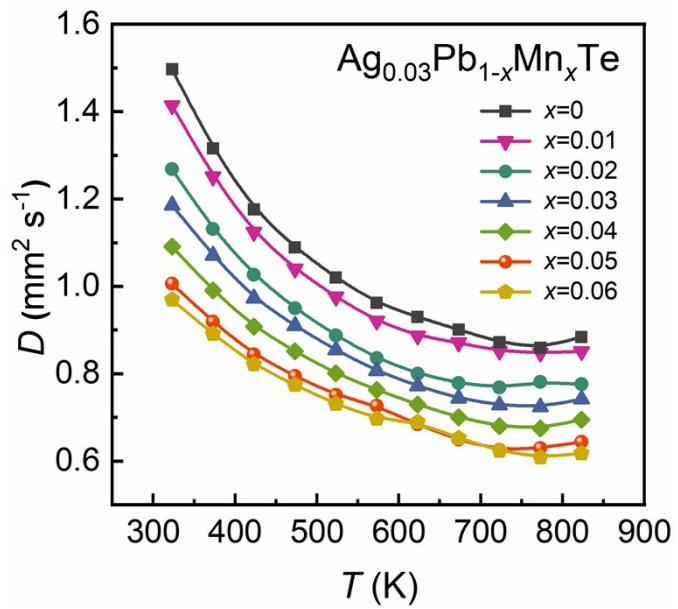


Figure S9. Temperature-dependent thermal diffusivity (D) for n -type $\text{Ag}_{0.03}\text{Pb}_{1-x}\text{Mn}_x\text{Te}$ samples.

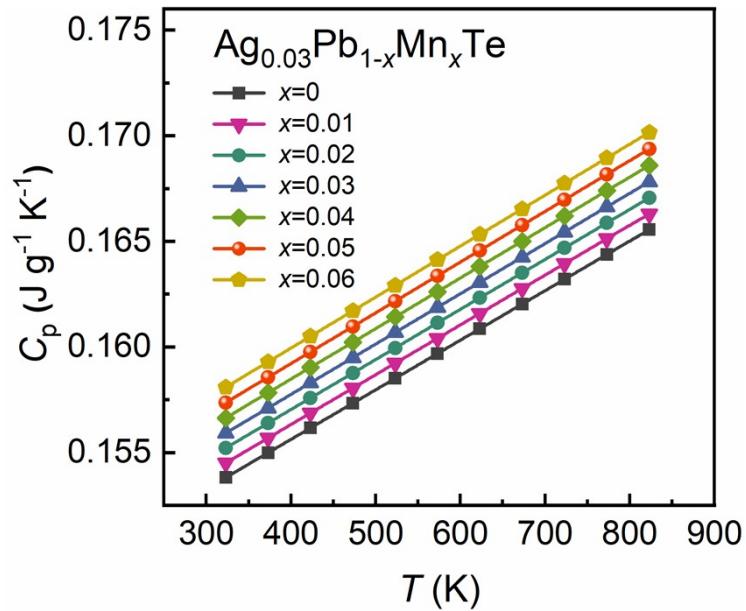


Figure S10. Temperature-dependent heat capacity (C_p) for n -type $\text{Ag}_{0.03}\text{Pb}_{1-x}\text{Mn}_x\text{Te}$ samples.

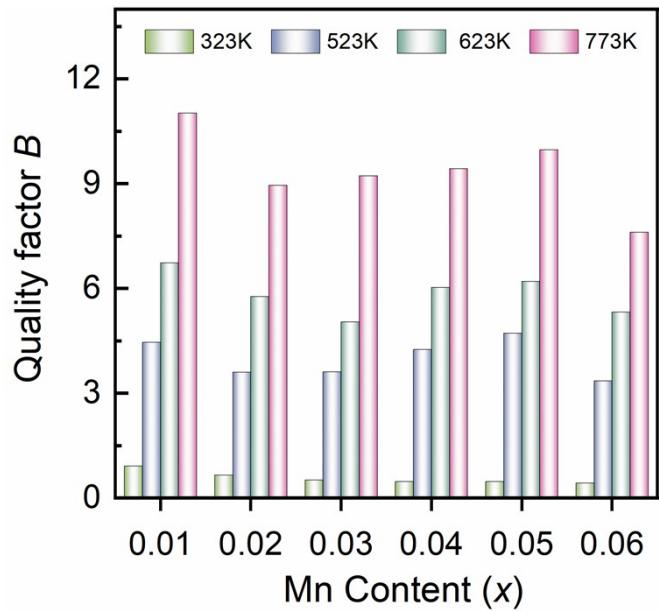


Figure S11. The calculated quality factor B at 323 K, 523 K, 623 K and 773 K for n -type $\text{Ag}_{0.03}\text{Pb}_{1-x}\text{Mn}_x\text{Te}$ samples.

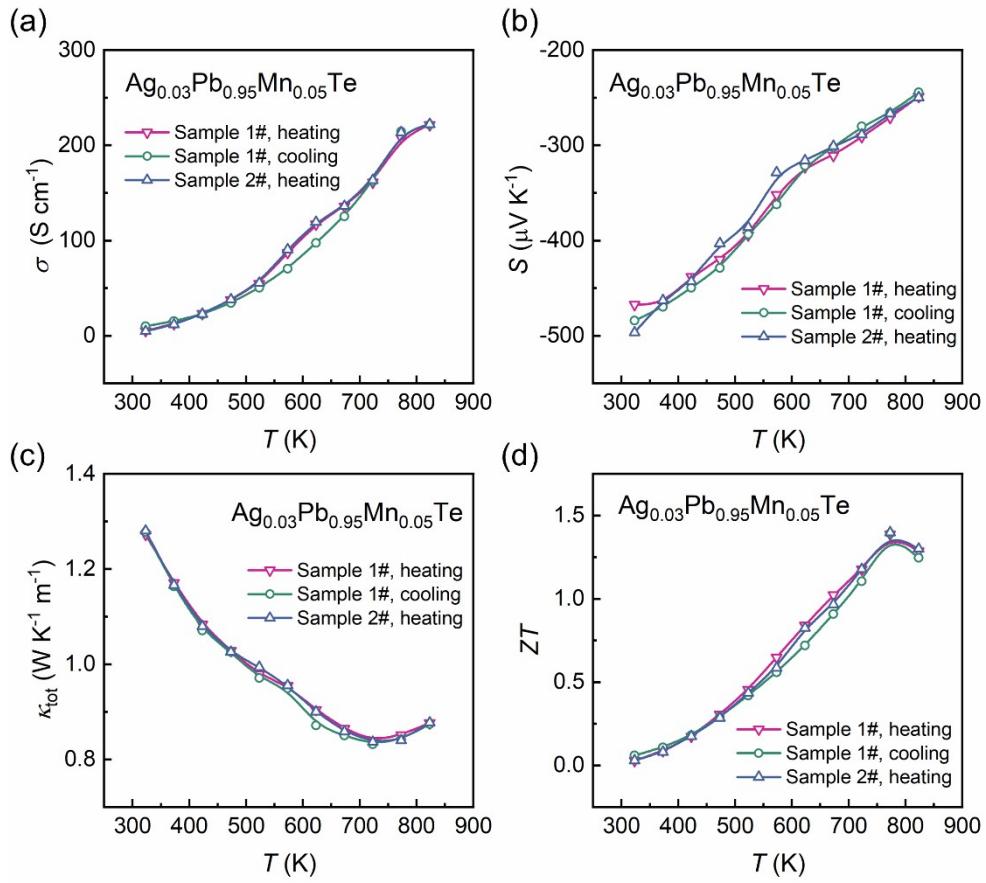


Figure S12. Thermal stability and reproducibility of (a) electrical conductivity σ , (b) Seebeck coefficient S , (c) total thermal conductivity κ_{tot} , and (d) figure of merit ZT for the high-performance n -type $\text{Ag}_{0.03}\text{Pb}_{0.95}\text{Mn}_{0.05}\text{Te}$ sample.

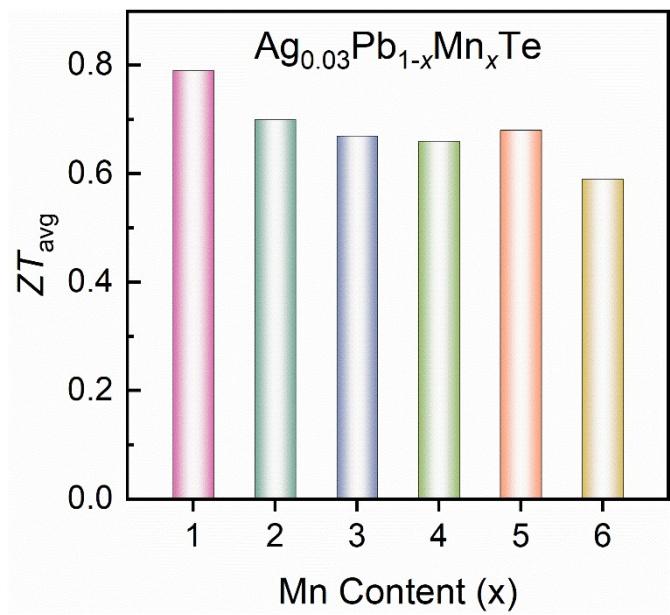


Figure S13. The average ZT (ZT_{avg}) for $\text{Ag}_{0.03}\text{Pb}_{1-x}\text{Mn}_x\text{Te}$ samples.

3. Supplementary tables

Table S1. The sample densities of $\text{Ag}_{0.03}\text{Pb}_{1-x}\text{Mn}_x\text{Te}$ ($x = 0, 0.01, 0.02, 0.03, 0.04, 0.05, 0.06$) measured by Archimedes method.

Sample No.	Measured density (g cm ⁻³)	Relative density (%)
$x=0$	8.17	99.15
$x=0.01$	8.099	98.29
$x=0.02$	8.098	98.28
$x=0.03$	8.003	97.12
$x=0.04$	8.087	98.14
$x=0.05$	8.037	97.53
$x=0.06$	8.167	99.11

Table S2. The input parameters based on the theoretical simulation of lattice thermal conductivity in this work.

Parameters	Symbol	Value	Ref.
Gruneisen parameter	γ	1.65	3
Debye temperature	θ_D (K)	163	17
Average sound velocity	v (m/s)	1720	17
Average atomic mass for PbTe	M (kg)	2.78×10^{-25}	18
Ratio of N- to U- process	A_N	2.9	19
Phenomenological parameter		65	18
Mole mass of Pb	M_{Pb} (g/mol)	207.2	-
Mole mass of Te	M_{Te} (g/mol)	127.6	-
Mole mass of Ag	M_{Ag} (g/mol)	107.8682	-
Mole mass of Mn	M_{Mn} (g/mol)	54.938	-
Radius of Pb atom	r_{Pb} (Å)	1.2	-
Radius of Ag atom	r_{Ag} (Å)	1.26	-
Radius of Mn atom	r_{Mn} (Å)	0.8	-

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