## **Electronic supplementary information**

## Understanding phase stability of yttria stabilized zirconia electrolyte

## under solid oxide electrolysis cell operation condition

Seong Kyun Kim <sup>a, b</sup>, Hyeon Jin Lee <sup>a</sup>, Jin Young Moon <sup>c</sup>, Yong-Ryun Jo <sup>c</sup>, Jinsil Lee<sup>a, b</sup>, Ji-Hoon Park<sup>a</sup>, Sun-Dong Kim <sup>d, \*</sup>, Jong Hoon Joo <sup>a, b, \*</sup>

<sup>a</sup>School of Earth Sciences and Environmental Engineering, Gwangju Institute of Science and

Technology, 123 Cheomdangwagi-ro, Buk-gu, Gwangju 61005, Republic of Korea

<sup>b</sup>Research Center for Innovative Energy and Carbon Optimized Synthesis for Chemicals(Inn-

ECOSysChem), Gwangju Institute of Science and Technology, 123 Cheomdan-gwagiro, Buk-gu,

Gwangju 61005, Korea

<sup>c</sup>GCRF (GIST Central Research Facilities), 123 Cheomdangwagi-ro, Buk-gu, Gwangju 61005, Republic of Korea

<sup>d</sup>Hydrogen Convergence Materials Laboratory, Korea Institute of Energy Research, 152 Gajeong-ro, Yuseong-gu, Daejeon, 34129, Republic of Korea

\* Corresponding author.

E-mail addresses: amastra@kier.re.kr (S.-D. Kim) E-mail addresses: jhjoo@gist.ac.kr, jonghoonjoo@gmail.com (J. H. Joo) The derivation of van der Pauw's technique for measuring resistivity is described as follows:  $R_A = R_{21,34} + R_{12,43} + R_{43,12} + R_{34,21}$  (1)  $R_B = R_{32,41} + R_{23,14} + R_{14,23} + R_{41,32}$  (2)  $\frac{-\pi R_A}{R_1 - \pi R_2} = \frac{-\pi R_2}{R_1 - \pi R_2}$ 

$$e^{\overline{R_S}} + e^{\overline{R_S}} = 1 (3)$$

$$\rho = R_S \times t (4)$$

 $R_{ab,cd}$  indicates that a and b represent the I<sup>+</sup> and I<sup>-</sup>, respectively, while c and d represent the V<sup>+</sup> and V<sup>-</sup>. The resistivity of the sample can be calculated using equation (4), where t denotes the sample's thickness. For in-plane measurements, the conductivity was measured using the van der Pauw technique, which is commonly employed for samples with uniform thickness. This method is well-suited for measuring the conductivity of our thick film specimens, which have a consistent thickness.



Fig. S1 SEM images of 8YSZ and 9.5YSZ.



Fig. S2 The configuration of electrodes in the (a) across-plane and (b) van-der Pauw method.



**Fig. S3** (a) Variations in ohmic resistance before and after the 900  $^{\circ}$ C degradation test with 1.7V and (b, c) changes in impedance spectroscopy over time at 750  $^{\circ}$ C with 1.7V.

	Conductivity m	neasured by in-	Conductivity measured by across-		
Temperature	plane method	$(log\sigma(S\cdot cm^{-1}))$	plane method ( $log\sigma(S \cdot cm^{-1})$ )		
(1000/T(K <sup>-1</sup> ))	8YSZ	9.5YSZ	8YSZ	9.5YSZ	
0.85241	-0.968	-1.00	-1.15	-1.12	
0.89035	-1.13	-1.16	-1.28	-1.31	
0.93184	-1.29	-1.33	-1.44	-1.45	
0.97737	-1.49	-1.55	-1.64	-1.72	
1.02759	-1.71	-1.78	-1.87	-1.95	
1.08325	-1.98	-2.06	-2.14	-2.19	
1.14528	-2.29	-2.35	-2.49	-2.44	
1.21485	-2.62	-2.76	-2.79	-2.82	
1.29341	-2.94	-3.13	-3.09	-3.21	

 Table S1 Conductivities measured using the in-plane and across-plane methods as a function of temperature.





Fig. S4 XRD data of pristine 8.5YSZ, 9YSZ, 9.5YSZ and 10YSZ. 5



Fig. S5 Raman spectra of (a) 8YSZ and (b) 9.5YSZ at 750  $^\circ\!C$  under 1.7V for 100 hours.



Fig. S6 Raman spectra indicating the variation in the tetragonal 8YSZ ratio with increasing annealing time.

	8YSZ				9.5YSZ		
	Centroid coordinates (at%)		Zr/Y		Centroid coordinates (at%)		Zr/Y
	Zr	Y			Zr	Y	
Fresh	36.74	7.14	5.15	Fresh	32.53	7.61	4.27
	27.39	8.25	3.32		40.26	6.66	6.05
	46.28	5.84	7.92		24.05	7.48	3.21
Anode side	37.19	5.88	6.32	Anode side	30.00	10.50	2.86
	29.93	6.17	4.85		46.65	7.61	6.13
	47.08	6.72	7.00		38.26	9.08	4.21
Cathode side	31.27	5.13	6.10	Cathode side	39.63	10.15	3.90
	25.66	5.39	4.76		30.19	10.77	2.80
	37.89	4.79	7.90		48.26	8.47	5.70

 Table S2 Centroid coordinates of Zr and Y for samples.



Fig. S7 STEM images of (a) 8YSZ fresh, (b) 8YSZ cathode side, (c) 8YSZ anode side, (d) 9.5YSZ fresh, (e) 9.5YSZ

cathode side and (f) 9.5YSZ anode side.



Fig. S8 Elbow method for optimal k values of (a) 8YSZ and (b) 9.5YSZ fresh samples.



Fig. S9 Y<sub>2</sub>O<sub>3</sub>-ZrO<sub>2</sub> phase diagram.<sup>41</sup> Copyright 2007, American Chemical Society.