

## Supporting Information

### **New approaches to three-dimensional positive electrodes enabling scalable high areal capacity**

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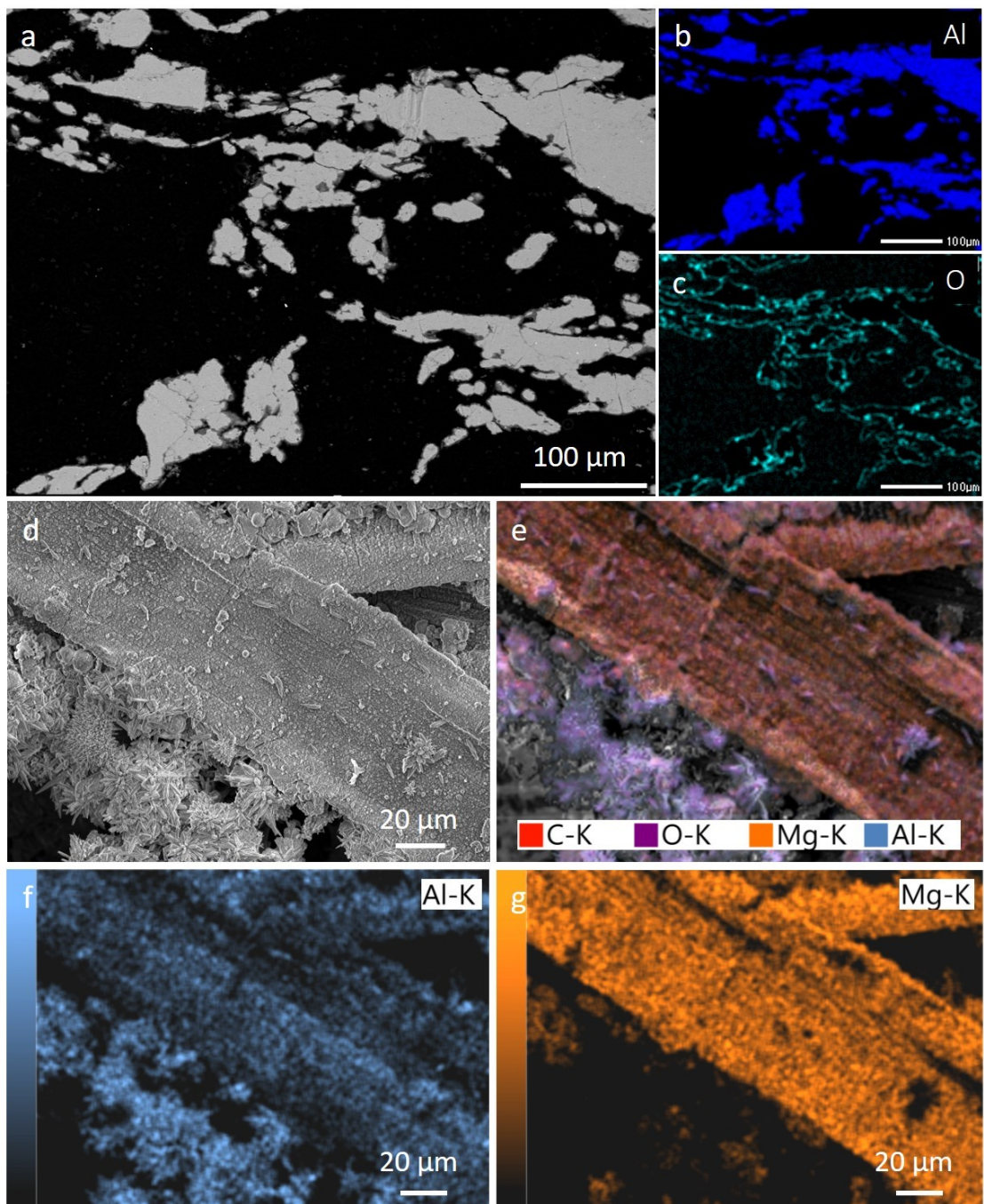
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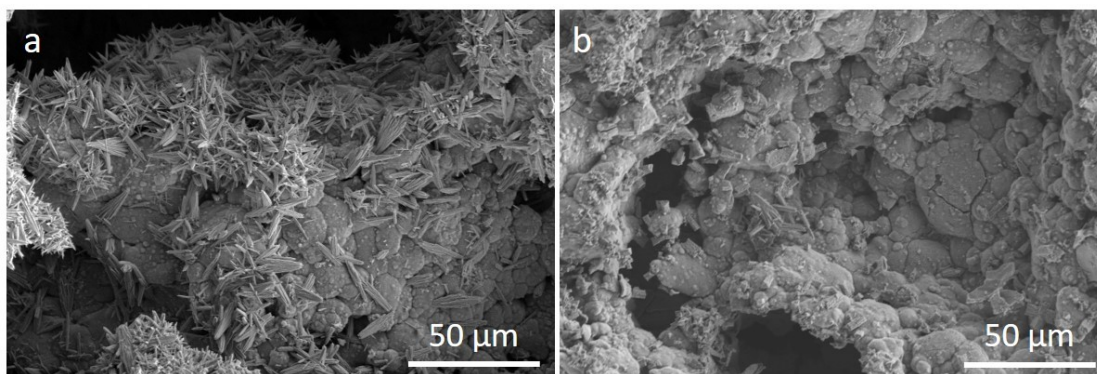
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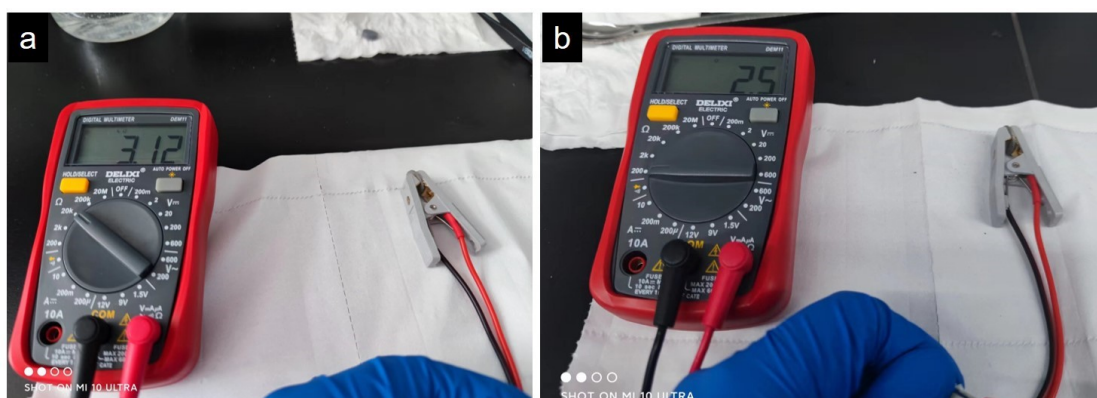
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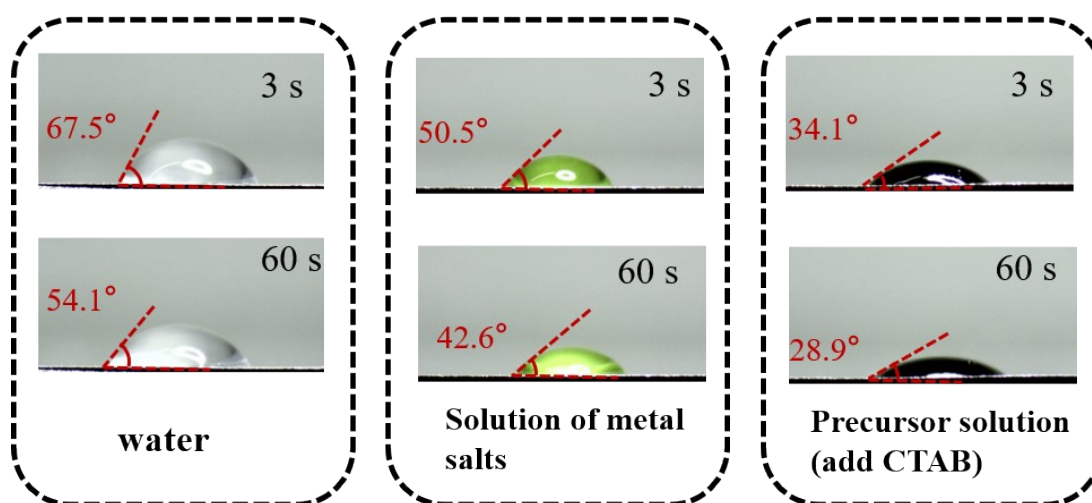
**Figure S1.** (a,d) Cross-section SEM image and (b,c,e,f,g) EDX elemental mapping images of the initial pAlN substrate before HCl treatment.



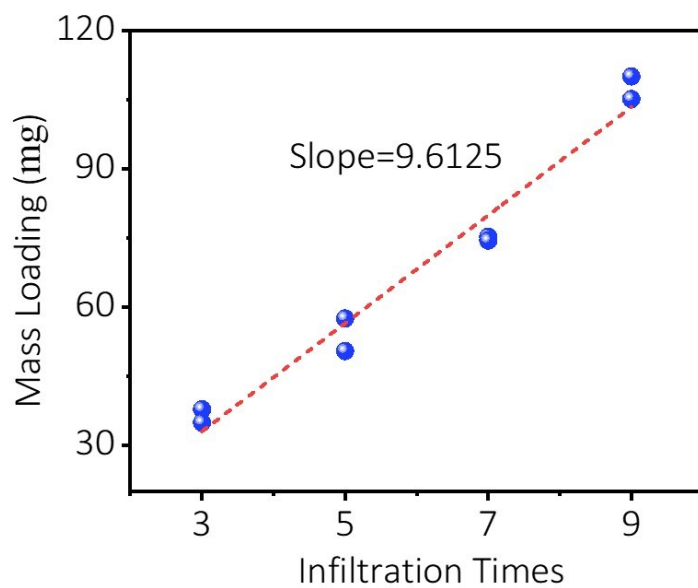
**Figure S2.** SEM images of the pAlN substrate (a) before and (b) after HCl treatment.



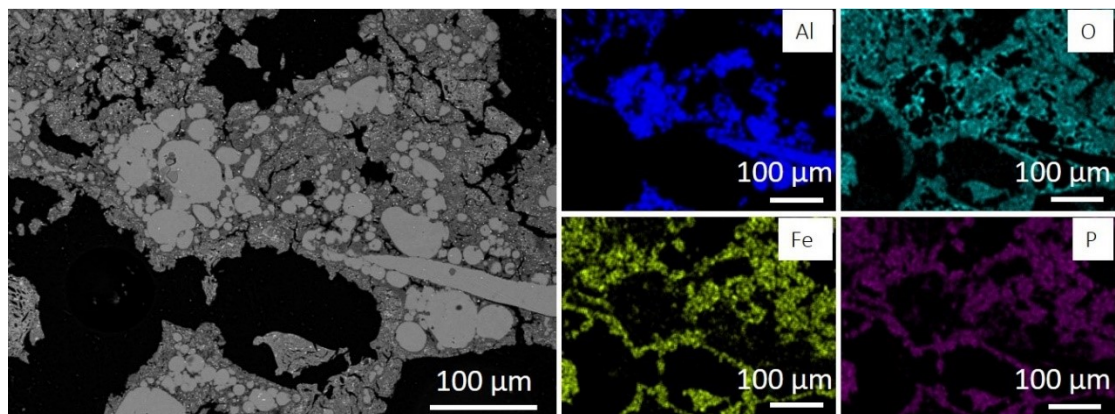
**Figure S3.** Resistance value of the pAlN substrate (a) before and (b) after HCl treatment.



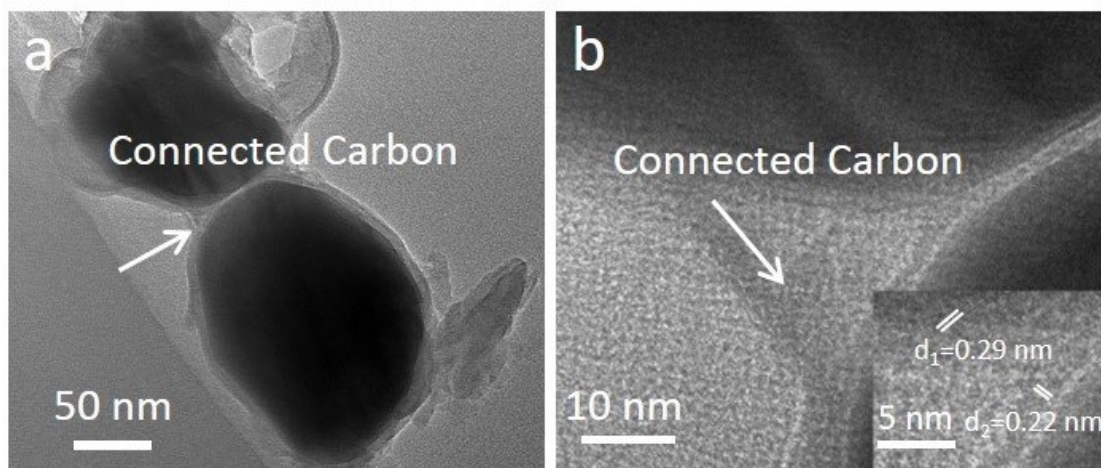
**Figure S4.** The wetting angles on Al foil using water, metal salt solution (mixture of  $\text{LiNO}_3$ ,  $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$  and  $\text{NH}_4\text{H}_2\text{PO}_4$ ) and precursor solution (mixture of  $\text{LiNO}_3$ ,  $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ ,  $\text{NH}_4\text{H}_2\text{PO}_4$  and CTAB).



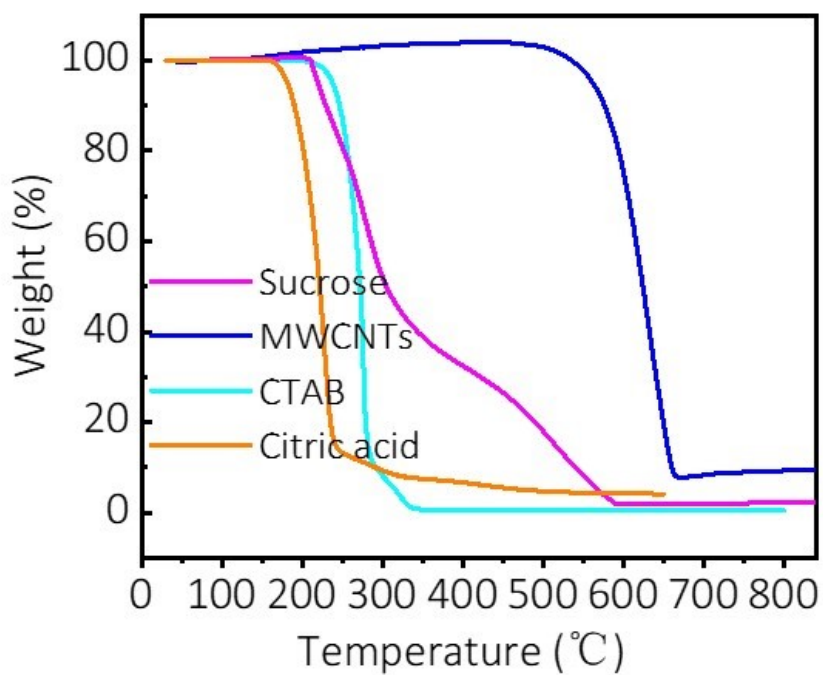
**Figure S5.** The relationship between the mass loading and the infiltration times.



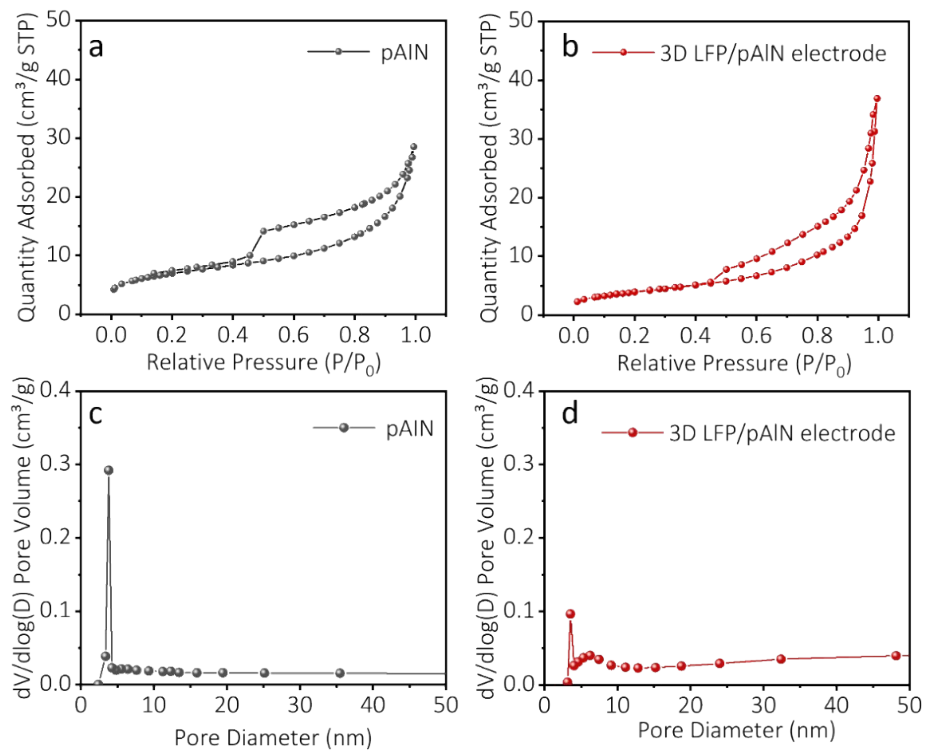
**Figure S6.** SEM image and EDX elemental mapping of the polished cross-section of the 3D LFP/pAlN electrode.



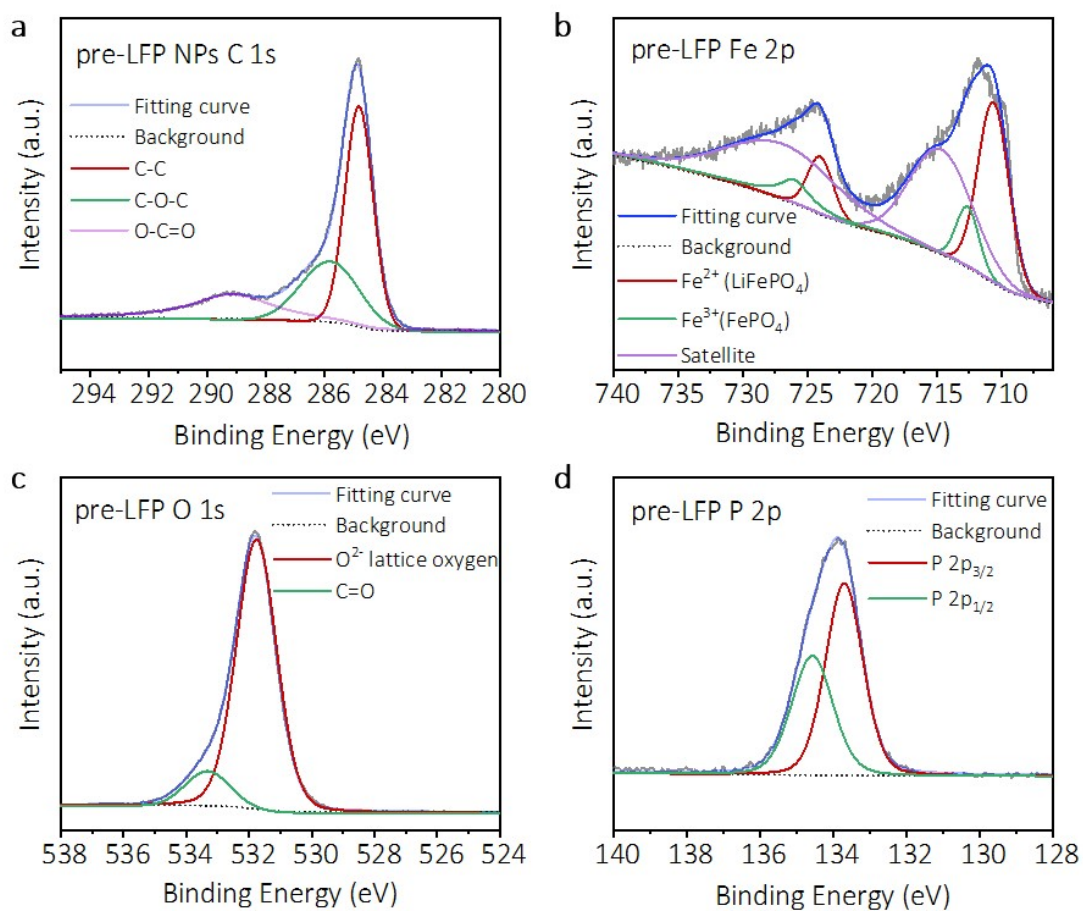
**Figure S7.** (a) TEM image and (b) HRTEM image of the nanoparticles ground from the 3D LFP/pAlN electrode.



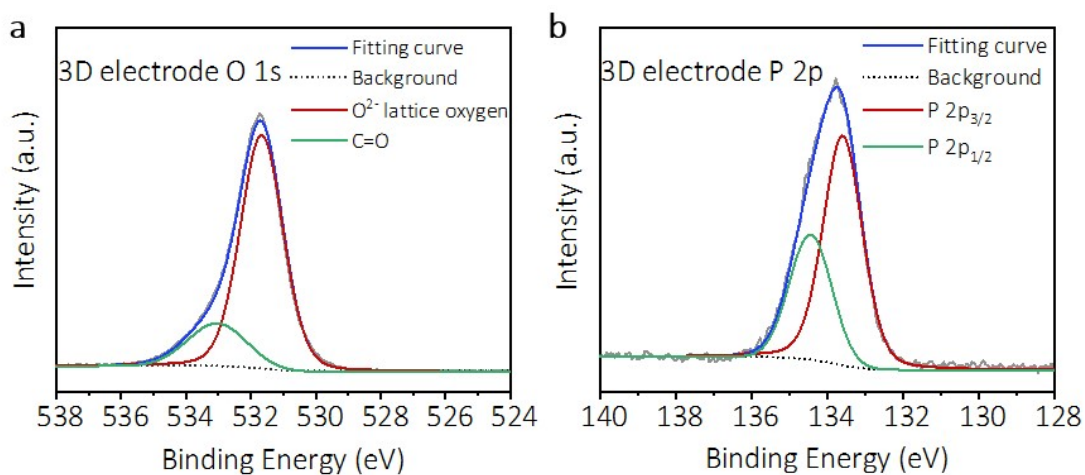
**Figure S8.** TGA curves of sucrose, MWCNTs, CTAB and citric acid measured under air atmosphere.



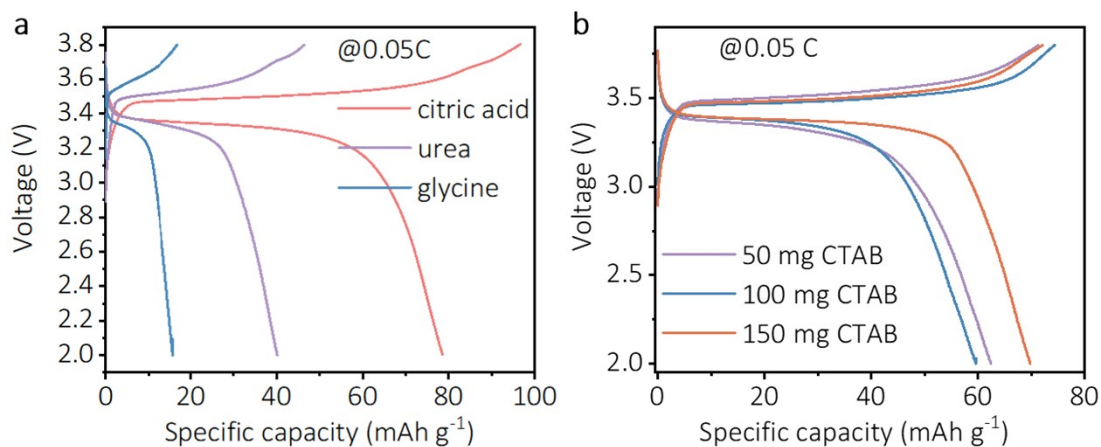
**Figure S9.** N<sub>2</sub> adsorption isotherms of (a) pAlN and (b) 3D LFP/pAlN electrode. Pore size distribution curves of (c) pAlN and (d) 3D LFP/pAlN electrode.



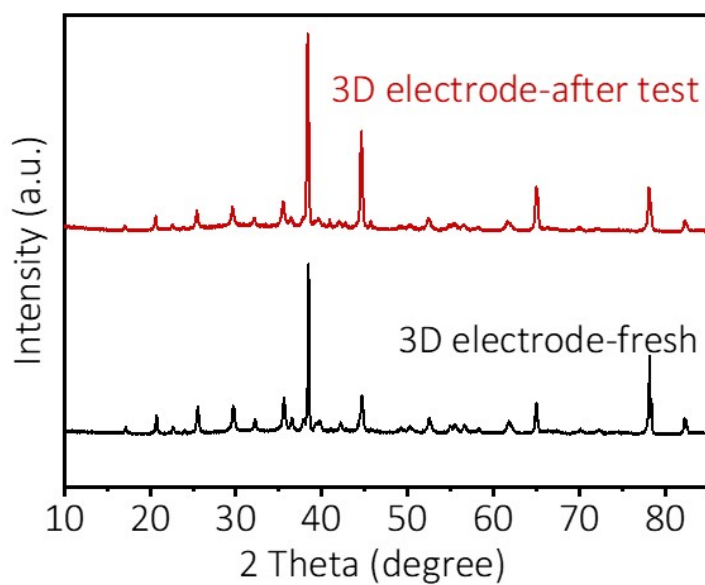
**Figure S10.** High-resolution XPS spectra of (a) C 1s, (b) Fe 2p, (c) O 1s and (d) P 2p for commercial carbon-coating pre-LFP NPs.



**Figure S11.** High-resolution XPS spectra of (a) O 1s and (b) P 2p for 3D LFP/pAlN electrode.

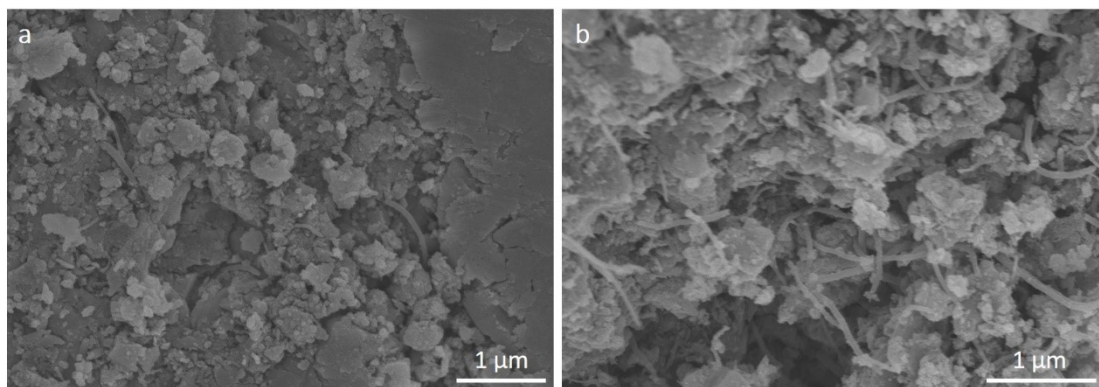


**Figure S12.** (a) Galvanostatic charge/discharge profiles of 3D LFP/pAlN electrode using different organic fuels (citric acid, urea and glycine) and (b) different amount of CTAB (50 mg, 100 mg, 150 mg).



**Figure S13.** XRD curves of the 3D electrode powder before and after galvanostatic charge/discharge cycles.





**Figure S14.** The SEM image of the 3D electrode after galvanostatic charge/discharge cycles.

**Table S1.** The calculation of porosity of Al foam.

Resistance ( $\Omega$ )	m (mg)	thickness (mm)	$\rho_f$ ( $\text{g}\cdot\text{cm}^{-3}$ )	Porosity
7.8	66.1	0.869	0.49	81.7
12	72.3	0.944	0.50	81.6
28	83.1	0.898	0.60	77.7
8	76.1	0.934	0.53	80.4
2.7	80.2	0.958	0.54	79.9
4.6	85.1	0.933	0.59	78.1
3.1	83.3	0.962	0.56	79.2
2.5	79.4	0.949	0.54	79.9
5.8	84.9	0.923	0.60	77.9
4.2	82.3	0.969	0.55	79.6
4	77.2	0.96	0.52	80.7
3.1	70.6	0.953	0.48	82.2

The porosity is calculated by:

$$\text{Porosity} = 1 - \rho_f / \rho_{Al}$$

The average porosity of Al foam is 79.9.

**Table S2.** The BET surface areas of pAlN and 3D LFP/pAlN electrode.

Sample	$S_{\text{BET}}$ ( $\text{m}^2\text{g}^{-1}$ )
pAlN	24.37
3D LFP/pAlN electrode	14.20

**Table S3.** Performance comparison of 3D LFP/pAlN electrode with recently reported novel structure 3D electrodes using non-situ growth methods.

Electrode structure	Mass loading (mg cm <sup>-2</sup> )	Cycles	Areal capacity (mAh cm <sup>-2</sup> )	Ref.
<b>3D LFP/pAlN electrode</b>	<b>65</b>	<b>100</b>	<b>6.56</b>	<b>This work</b>
Magnetically aligned Fe <sub>3</sub> O <sub>4</sub> -modified graphite flakes	9.1	50	0.93	1
LiFePO <sub>4</sub> slurry in wood-templating	60 (LiFePO <sub>4</sub> )	140	3.8	2
Copolymer-aligned structure of NMC811 secondary particle	50	50	8.84	3
LiFe <sub>0.7</sub> Mn <sub>0.3</sub> PO <sub>4</sub> nanoplates on 3D graphene frameworks with aligned vertical channels	21.2	500	1.77	4
Electrophoretic deposition of LiFePO <sub>4</sub> onto 3D carbon fiber cloth	20 (LiFePO <sub>4</sub> )	100	2	5
Micrometer-size bulk-porous Si/carbonized polyacrylonitrile composites	1.6	50	3	6
Thick porous LiFePO <sub>4</sub> /C composite electrodes by spark plasma sintering	≈150	20	21	7
Freestanding Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub> /Super P/PVDF electrode	14	100	1.95	8

**Table S4.** Performance comparison of 3D LFP/pAlN electrode with recently reported porous and conductive 3D electrodes using in situ grown methods.

Electrode structure	Mass loading (mg·cm <sup>-2</sup> )	Cycles	Areal capacity (mAh·cm <sup>-2</sup> )	Ref.
<b>3D LFP/pAlN electrode</b>	<b>65</b>	<b>100</b>	<b>6.56</b>	<b>This work</b>
<b>3-D nanoporous network composed of NiCuMn@oxide core@shell structures supported by an intercalary Cu film</b>	6.9 (NiCuMn oxide)	100	6.1	9
<b>FeOx porous nanoblocks array on Cu foam</b>	2.5 (FeOx)	50	3.33	10
<b>Outer few-layer NiO nanoflowers@medial NiO layer@Ni foam</b>	-	100	1.45	11
<b>Thick mesoporous Co<sub>3</sub>O<sub>4</sub> nanosheet on Ni foam</b>	4.15 (Co <sub>3</sub> O <sub>4</sub> )	30	4.39	12
<b>Carbon-encapsulated Fe<sub>3</sub>O<sub>4</sub> nanospheres anchored on CNT network</b>	0.8	500	0.32	13
<b>LiMn<sub>2</sub>O<sub>4</sub>@C grown on CNT</b>	2	1000	0.11	14
<b>LiMn<sub>2</sub>O<sub>4</sub>@C grown on carbon nanofiber</b>	2	1000	0.18	14
<b>Carbon fiber cloth coated with exfoliated porous N-doped carbon fiber, NiO nanosheets, and carbon quantum dots</b>	10.58	250	2.91	15
<b>MoS<sub>2</sub> nanoflakes grown on the twine carbon fibers of the carbonized cotton cloth</b>	4.4 (MoS <sub>2</sub> )	35	5.2	16
<b>Nb<sub>2</sub>O<sub>5</sub>@3D holey-graphene framework composites</b>	6	10000	1	17

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