## **Electronic Supplementary Information**

## Boosting electrocatalytic hydrogen evolution over Mo<sub>2</sub>C-W<sub>2</sub>C heterostructure by interface-induced electron modulation

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products	$(NH_4)_6Mo_7O_{24}\cdot 4H_2O^a(g)$	$(NH4)_{6}H_{2}W_{12}O_{40}{\cdot}xH_{2}O^{b}\left(g\right)$	Mo/W <sup>c</sup>	Mo/(Mo+W) <sup>d</sup>
Mo <sub>2</sub> C-W <sub>2</sub> C/RGO-0.24	0.18	0.74	0.33	0.24
Mo <sub>2</sub> C-W <sub>2</sub> C/RGO-0.56	0.36	0.50	0.5	0.56
Mo <sub>2</sub> C-W <sub>2</sub> C/RGO-0.75	0.53	0.25	3.0	0.75

 Table S1. Composition of samples obtained at different Mo/W feeding ratios.

<sup>a</sup> Data was the molar amount of  $(NH_4)_6Mo_7O_{24}$ ;  $4H_2O$  used in synthesis. <sup>b</sup> Data was the molar amount of  $(NH_4)_6H_2W_{12}O_{40}$ ;  $xH_2O$  used in synthesis. <sup>c</sup> Data was Mo/W feeding molar ratio. <sup>d</sup> Data was calculated from the result of EDS.



 $W_2C/RGO-0.24$ , and (d)  $W_2C/RGO$ .



Fig. S2 EDS spectra of (a)  $Mo_2C-W_2C/RGO-0.24$ , (b)  $Mo_2C-W_2C/RGO-0.56$ , and (c)  $Mo_2C-W_2C/RGO-0.75$ .

![](_page_4_Figure_2.jpeg)

Fig. S3 Raman spectra of the GO and RGO.

![](_page_5_Figure_0.jpeg)

Fig. S4 (a) Survey XPS spectra. High-resolution XPS profiles of (b) C 1s; (c) N 1s.

![](_page_6_Figure_0.jpeg)

Fig. S5 (a) Polarization curves and (b) Tafel plots of Mo<sub>2</sub>C/RGO, Mo<sub>2</sub>C-W<sub>2</sub>C/RGO-0.24, Mo<sub>2</sub>C-W<sub>2</sub>C/RGO-0.56, Mo<sub>2</sub>C-W<sub>2</sub>C/RGO-0.75, and W<sub>2</sub>C/RGO in 0.5 M H<sub>2</sub>SO<sub>4</sub>. (c) The  $C_{dl}$  of different materials obtained at 0.15 V versus RHE. (d) Nyquist plots (at  $\eta = 200$  mV).

H<sub>2</sub>SO<sub>4</sub>. **Onset potential** Tafel slope **R**<sub>ct</sub>  $j_0$  $C_{\rm dl}$  $\eta_{10}$ Catalysts (mV vs RHE) (mV vs RHE) (mA cm<sup>-2</sup>) (mV dec<sup>-1</sup>) (mF dec<sup>-1</sup>) (Ω) 110 Mo<sub>2</sub>C/RGO 43 0.384 73 27.3 11.3  $Mo_2C-W_2C/RGO-0.75$ 0.407 37.1 9.8 41 103 68  $Mo_2C\text{-}W_2C/RGO\text{-}0.56$ 30 81 0.428 38.5 5.9 56 Mo<sub>2</sub>C-W<sub>2</sub>C/RGO-0.24 137 25.3 10.34 66 0.251 75 W<sub>2</sub>C/NRGO 67 151 0.244 94 20.0 33.1

Table S2. Comparison of catalytic parameters of  $Mo_2C/RGO$ ,  $Mo_2C-W_2C/RGO-0.75$ ,  $Mo_2C-W_2C/RGO-0.56$ ,  $Mo_2C-W_2C/RGO-0.24$ , and  $W_2C/NRGO$  catalysts in 0.5 M  $H_2SO_4$ .

Catalyst	Electrolyte	Tafel slope (mV dec <sup>-1</sup> )	η <sub>10</sub> (mV vs. RHE)	Ref.	
M. C.W.C.BCO.A.S.	0.5 M H <sub>2</sub> SO <sub>4</sub>	56	81	This work	
M0 <sub>2</sub> C-W <sub>2</sub> C/RGO-0.56	1.0 M KOH	59	87		
Trainer d WCN	1.0 M KOH	-	138	1	
I winned WCN	0.5 M H <sub>2</sub> SO <sub>4</sub>	65	128	1	
WOWDONG	1.0 M KOH	59.07	196.2	2	
W <sub>2</sub> C/WP@NC	0.5 M H <sub>2</sub> SO <sub>4</sub>	77.4	116.37	2	
W.COONT OF	1.0 M KOH	56.2	148		
W <sub>2</sub> C@CN1-S8	0.5 M H <sub>2</sub> SO <sub>4</sub>	57.4	176	3	
	1.0 M KOH	57.4	145		
Mo <sub>2</sub> C@BNC	0.5 M H <sub>2</sub> SO <sub>4</sub>	68.3	184	4	
	1.0 M KOH	94.7	124	-	
WC@C/NF	0.5 M H <sub>2</sub> SO <sub>4</sub>	-	208	5	
CoO/Mo <sub>2</sub> C	1.0 M KOH	80	107	6	
Co/Mo <sub>2</sub> C@C	1.0 M KOH	68	98	7	
	1.0 M KOH	68	154	8	
Mo <sub>2</sub> N-Mo <sub>2</sub> C/HGr	0.5 M H <sub>2</sub> SO <sub>4</sub>	55	157		
Mo <sub>2</sub> C/MoS <sub>2</sub> -rGO	1.0 M KOH	52	112	9	
	1.0 M KOH	72.2	116	10	
WC-W <sub>2</sub> C/HCDs	$0.5 \mathrm{~M~H_2SO_4}$	52.5	96		
Mo <sub>2</sub> C-CoO@N-CNFs-8	1.0 M KOH	76	115	11	
Ni-W <sub>2</sub> C	1.0 M KOH	73.8	88	12	
	1.0 M KOH	72	205	13	
WS <sub>2</sub> /W <sub>2</sub> C@NSPC	0.5 M H <sub>2</sub> SO <sub>4</sub>	68	126		
	1.0 M KOH	51	147		
W-W <sub>2</sub> C/CNT-6	0.5 M H <sub>2</sub> SO <sub>4</sub>	56	155	14	

**Table S3.** Comparison of HER performance for Mo<sub>2</sub>C-W<sub>2</sub>C/RGO with other reported carbon-based electrocatalysts.

Catalyst	Onset potential	$\eta_{10}$	<i>j</i> 0	Tafel slope		
	(mV vs RHE)	(mV vs RHE)	(mA cm <sup>-2</sup> )	(mV dec <sup>-1</sup> )	$R_{\rm ct}$ (S2)	
Pt/C	0	21	0.945	21	/	
Mo <sub>2</sub> C-W <sub>2</sub> C/RGO-0.56	30	81	0.428	56	5.9	
Mo <sub>2</sub> C/RGO	43	110	0.384	73	11.3	
W <sub>2</sub> C/RGO	67	151	0.278	94	33.1	
RGO	278	380	0.005	104	4040	

Table S4. Comparison of catalytic parameters of different HER catalysts in 0.5 M  $H_2SO_4$ .

![](_page_10_Figure_0.jpeg)

Fig. S6 The  $C_{dl}$  of different materials obtained at 0.15 V versus RHE in 0.5 M H<sub>2</sub>SO<sub>4</sub>.

![](_page_10_Figure_2.jpeg)

Fig. S7 SEM image of  $Mo_2C$ - $W_2C/RGO$ -0.56 after long-term durability test in 0.5 M

 $H_2SO_4.$ 

Catalyst	Onset potential (mV vs RHE)	η <sub>10</sub> (mV vs RHE)	<i>j</i> <sub>0</sub> (mA cm <sup>-2</sup> )	Tafel slope (mV dec <sup>-1</sup> )	$R_{\rm ct}(\Omega)$
Pt/C	8	27	0.933	41	/
Mo <sub>2</sub> C-W <sub>2</sub> C/RGO-0.56	33	87	0.398	59	4.4
Mo <sub>2</sub> C/RGO	66	134	0.120	64	6.5
W <sub>2</sub> C/RGO	76	159	0.102	93	27.9
RGO	296	491	0.0316	189	279.7

**Table S5.** Comparison of catalytic parameters of different HER catalysts in 1.0 MKOH.

![](_page_12_Figure_0.jpeg)

Fig. S8 (a) Polarization curves and (b) Tafel plots of Mo<sub>2</sub>C/RGO, Mo<sub>2</sub>C-W<sub>2</sub>C/RGO-0.24, Mo<sub>2</sub>C-W<sub>2</sub>C/RGO-0.56, Mo<sub>2</sub>C-W<sub>2</sub>C/RGO-0.75, and W<sub>2</sub>C/RGO in 1.0 M KOH. (c) The  $C_{dl}$  of different materials obtained at 0.15 V versus RHE. (d) Nyquist plots (at  $\eta = 200$  mV).

KOH. **Onset potential** Tafel slope  $j_0$  $C_{dl}$ **R**<sub>ct</sub>  $\eta_{10}$ Catalysts (mV vs RHE) (mV vs RHE) (mV dec<sup>-1</sup>) (mF dec<sup>-1</sup>) (mA cm<sup>-2</sup>) **(Ω)** Mo<sub>2</sub>C/RGO 0.112 66 134 64 17.2 6.5 Mo<sub>2</sub>C-W<sub>2</sub>C/RGO-0.75 56 114 0.184 60 22.3 5.5 Mo<sub>2</sub>C-W<sub>2</sub>C/RGO-0.56 33 87 0.398 59 26.3 4.4 Mo<sub>2</sub>C-W<sub>2</sub>C/RGO-0.24 69 135 0.132 64 23.8 8.2 W<sub>2</sub>C/NRGO 159 0.102 93 27.9 76 12.7

**Table S6.** Comparison of catalytic parameters of  $Mo_2C/RGO$ ,  $Mo_2C-W_2C/RGO-0.75$ ,  $Mo_2C-W_2C/RGO-0.56$ ,  $Mo_2C-W_2C/RGO-0.24$ , and  $W_2C/NRGO$  catalysts in 1.0 M KOH.

![](_page_13_Figure_2.jpeg)

Fig. S9 The  $C_{dl}$  of different materials obtained at 0.15 V versus RHE in 1.0 M KOH.

![](_page_14_Picture_0.jpeg)

Fig. S10 SEM image of  $Mo_2C-W_2C/RGO-0.56$  after long-term durability test in 1.0 M KOH.

![](_page_14_Figure_2.jpeg)

**Fig. S11** (a) XRD pattern and (b) SEM image of W<sub>2</sub>C. (c) XRD pattern and (d) SEM image of Mo<sub>2</sub>C. (e) XRD pattern and (f) SEM image of Mo<sub>2</sub>C-W<sub>2</sub>C.

![](_page_15_Figure_0.jpeg)

Fig. S12 (a) Polarization curves of RGO,  $W_2C$ ,  $W_2C/RGO$ ,  $Mo_2C$ ,  $Mo_2C/RGO$  $Mo_2C-W_2C$ , and  $Mo_2C-W_2C/RGO-0.56$  in 0.5 M H<sub>2</sub>SO<sub>4</sub>. (b) Polarization curves of RGO,  $W_2C$ ,  $W_2C/RGO$ ,  $Mo_2C$ ,  $Mo_2C/RGO$   $Mo_2C-W_2C$ , and  $Mo_2C-W_2C/RGO-0.56$ in 1.0 M KOH.

![](_page_16_Figure_0.jpeg)

Fig. S13 The density of states for  $Mo_2C$ ,  $W_2C$ ,  $Mo_2C-W_2C$ . The dashed line denotes the position of the Fermi level.

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