

## Supporting Information

# Multi-Functional Perovskite Oxide $\text{Pr}_{0.6}\text{Sr}_{0.4}\text{Mn}_{0.2}\text{Fe}_{0.7}\text{Ni}_{0.1}\text{O}_{3-\delta}$ as an Efficient Quasi-Symmetric Electrode for Solid oxide Fuel/Electrolysis Cells

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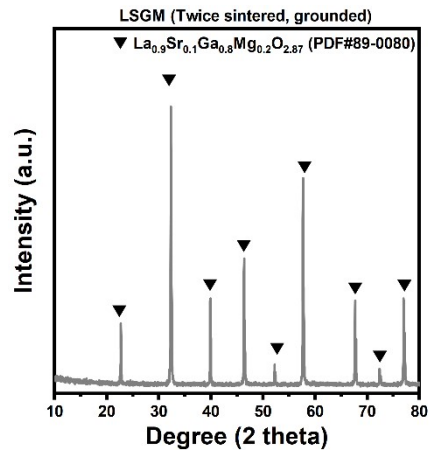
**Table S1.** Peak power density (PPD) comparison of symmetric electrodes in H<sub>2</sub>-operated

Material	Electrolyte	Cell configuration	PPD (W cm <sup>-2</sup> )		Ref
			Gas	Value (Temp.)	
(La <sub>0.8</sub> Sr <sub>0.2</sub> ) <sub>0.9</sub> Sc <sub>0.2</sub> Mn <sub>0.75</sub> Ru <sub>0.05</sub> O <sub>3-δ</sub> (LSCMR)	SSZ (200μm)	LSSMR SSZ LS SMR	wet H <sub>2</sub>	0.318 (800°C)	[1]
Sr <sub>2</sub> Ti <sub>0.8</sub> Co <sub>0.2</sub> FeO <sub>6</sub> (STCF)	LSGM (270μm)	STC02F LSGM  STC02F	dry H <sub>2</sub>	0.555 (800°C)	[2]
La <sub>0.6</sub> Sr <sub>0.4</sub> Fe <sub>0.95</sub> Ru <sub>0.05</sub> O <sub>3-δ</sub> (LSFRu)	LSGM (300μm)	LSGMR05 LSG M LSGMR05	dry H <sub>2</sub>	602 (850°C)	[3]
La <sub>0.5</sub> Sr <sub>0.5</sub> Fe <sub>0.9</sub> W <sub>0.1</sub> O <sub>3-δ</sub> (LSFW)	LSGM (250μm)	LSFW LSGM LS FW	wet H <sub>2</sub>	0.618 (800°C)	[4]
Ce <sub>0.2</sub> Sr <sub>0.8</sub> Fe <sub>0.95</sub> Ru <sub>0.05</sub> O <sub>3</sub> (CeSFR)	LSGM (320μm)	Ce20SFR LSGM  Ce20SFR	dry H <sub>2</sub>	0.846 (800°C)	[5]
Pr <sub>0.6</sub> Sr <sub>0.4</sub> Fe <sub>0.8</sub> Mn <sub>0.2</sub> O <sub>3-δ</sub> - Ce <sub>0.9</sub> Gd <sub>0.1</sub> O <sub>2-δ</sub> (PSMFN-GDC)	LSGM (300μm)	PSMFN- GDC LSGM PS MFN-GDC	dry H <sub>2</sub>	602 (850°C)	This work

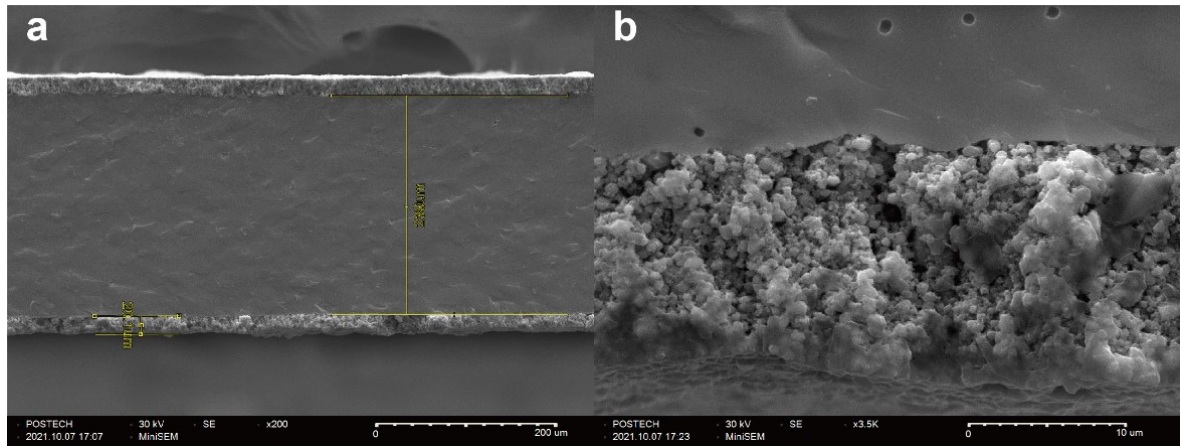
SOFC

Material	Electrolyte	Cell configuration	Current density (A cm <sup>-2</sup> )		Ref
			Gas	Value (Temp., Voltage)	
La <sub>0.4</sub> Sr <sub>0.6</sub> Co <sub>0.2</sub> Fe <sub>0.7</sub> Nb <sub>0.1</sub> O <sub>3-δ</sub> (LSCFNb)	YSZ (200μm)	LSCFN-GDC YSZ LSCFN-GDC	30% CO/CO <sub>2</sub>	0.4 (800°C, 1.5V)	[6]
La <sub>0.6</sub> Sr <sub>0.4</sub> Fe <sub>0.95</sub> Pt <sub>0.05</sub> O <sub>3-δ</sub> (Fe@LSPt)	LSGM (300μm)	LSPt LSGM LSPt	50% CO/CO <sub>2</sub>	0.65 (850°C, 1.5V)	[7]
(La,Sr)Fe <sub>0.9</sub> Ni <sub>0.1</sub> O <sub>4+δ</sub> (RPLSFN0.1)	LSGM (300μm)	RPLSFN0.1 LSGM RPLSFN0.1	50% CO/CO <sub>2</sub>	0.75 (800°C, 1.5V)	[8]
La <sub>0.8</sub> Sr <sub>0.2</sub> Cr <sub>0.5</sub> Fe <sub>0.5</sub> O <sub>3-δ</sub> -Zr <sub>0.84</sub> Y <sub>0.16</sub> O <sub>2-δ</sub> (LSCrF-YSZ)	YSZ (200μm)	LSCrF-YSZ YSZ LSCrF-YSZ	30% CO/CO <sub>2</sub>	0.75 (850°C, 1.5V)	[9]
La <sub>0.65</sub> Bi <sub>0.1</sub> Sr <sub>0.25</sub> Cr <sub>0.5</sub> Fe <sub>0.5</sub> O <sub>3-δ</sub> -Ce <sub>0.8</sub> Sm <sub>0.2</sub> O <sub>1.9</sub> (LBiSCrF-SDC)	LSGM (330μm)	Bi-LSCrF-SDC LSGM Bi-LSCrF-SDC	50% CO/CO <sub>2</sub>	0.79 (800°C, 1.5V)	[10]
Pr <sub>0.6</sub> Sr <sub>0.4</sub> Fe <sub>0.8</sub> Mn <sub>0.2</sub> O <sub>3-δ</sub> -Ce <sub>0.9</sub> Gd <sub>0.1</sub> O <sub>2-δ</sub> (PSMFN-GDC)	LSGM (300μm)	PSMFN-GDC LSGM PSMFN-GDC	30% CO/CO <sub>2</sub>	1.02 (850°C, 1.5V)	This work

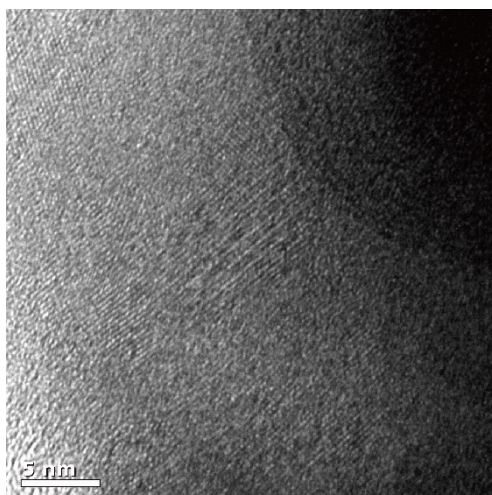
**Table S2.** Current density comparison of symmetric electrodes for CO<sub>2</sub> electrolysis in SOEC.



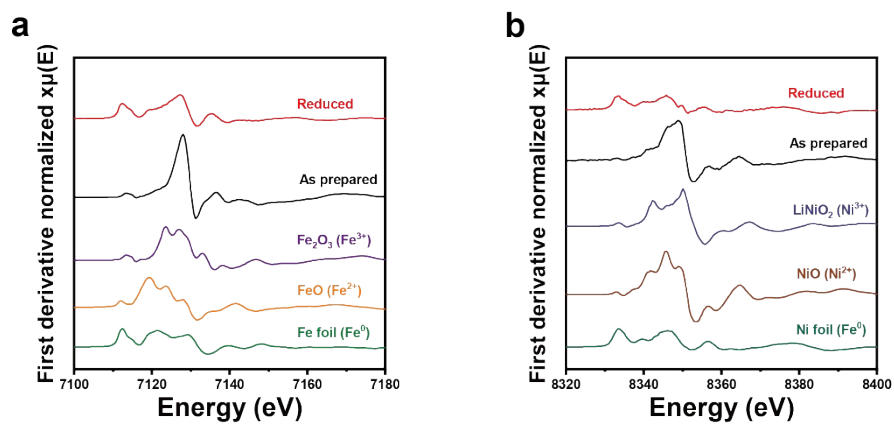
**Fig. S1.** XRD pattern of grounded twice-sintered LSGM pellet.



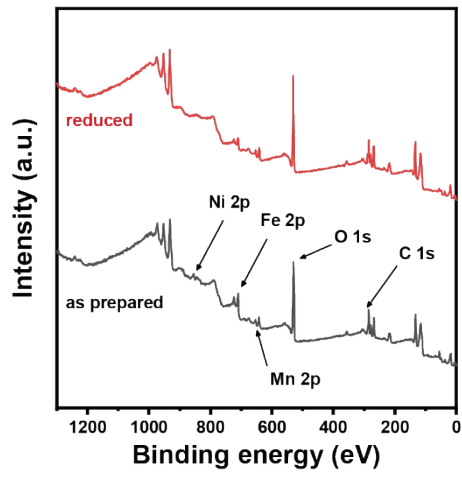
**Fig S2.** The Cross-sectional SEM images of (a) PSMFN-GDC symmetrical cell based on LSGM electrolyte and (b) magnified electrode.



**Fig S3.** HR-TEM image of reduced PSMFN.

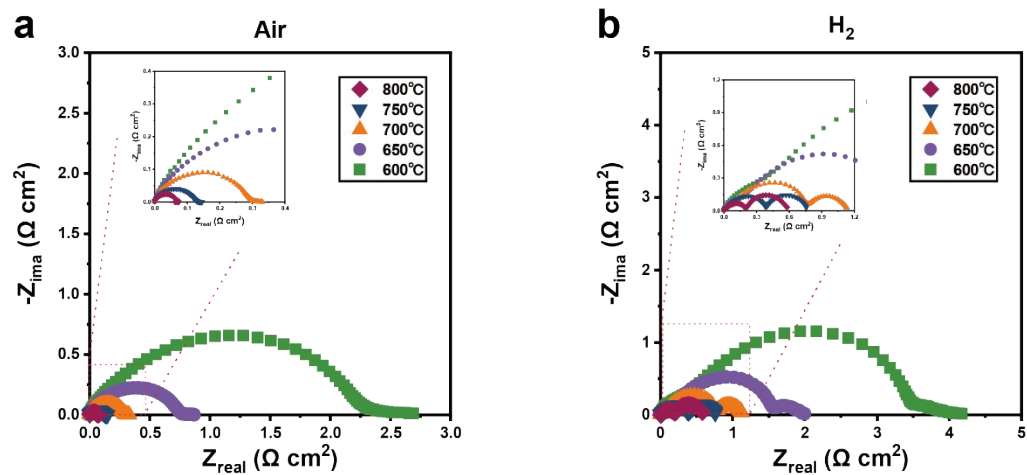


**Figure S4.** The 1<sup>st</sup> derivative curves of XAENS spectra of PSMFN, reduced PSMFN and corresponding reference oxides at (a) Fe-K edge and (b) Ni K-edge

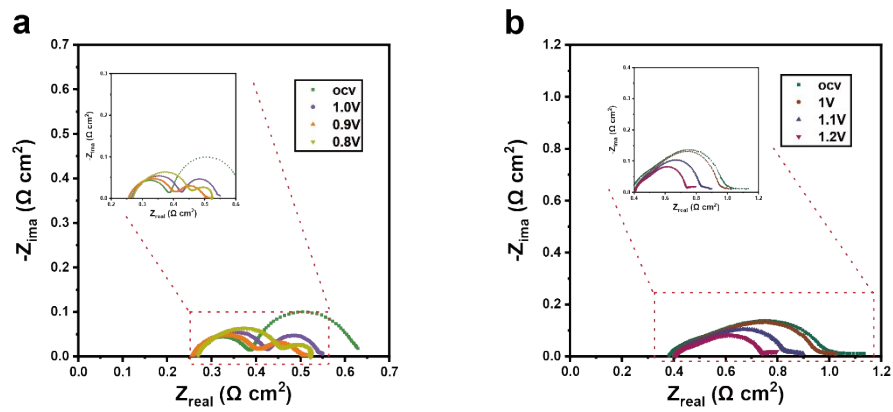


**Figure S5.** XPS surveys of the as prepared and reduced PSMFN





**Figure S6.** Symmetric half-cell tests of PSMFN-GDC under (a) air and (b)  $\text{H}_2$  atmosphere.



**Figure S7.** EIS profiles of PSMFN-GDC symmetric cell under applied voltage conditions in (a) SOFC mode and (b) SOEC mode

## References

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