

## Supporting Information

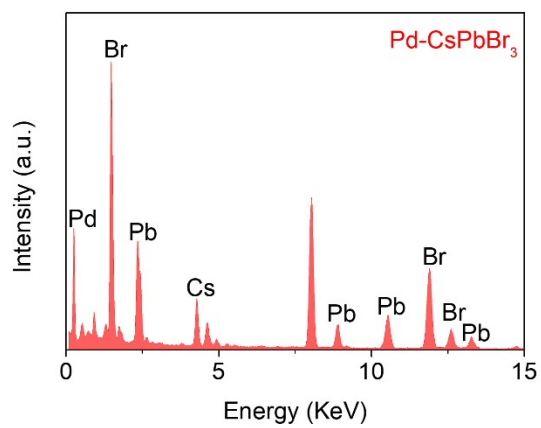
### **Enhanced charge transport from Pd-doping in CsPbBr<sub>3</sub> quantum dots for efficient photoelectrocatalytic water splitting**

Wenxiao Gong,<sup>a</sup> Yulan Li,<sup>\*a</sup> Yang Yang,<sup>a</sup> Heng Guo,<sup>\*b</sup> Xiaobin Niu<sup>\*a</sup>

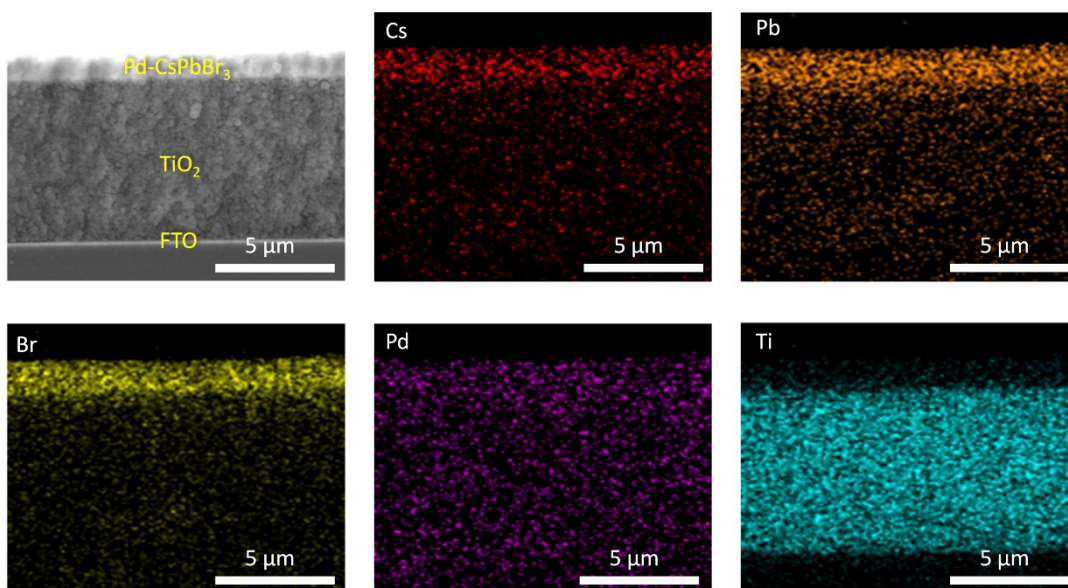
<sup>a</sup> *School of Materials and Energy, University of Electronic Science and Technology of China, Chengdu 610054, P. R. China.*

<sup>b</sup> *School of New Energy and Materials, Southwest Petroleum University, Chengdu, 610050, China.*

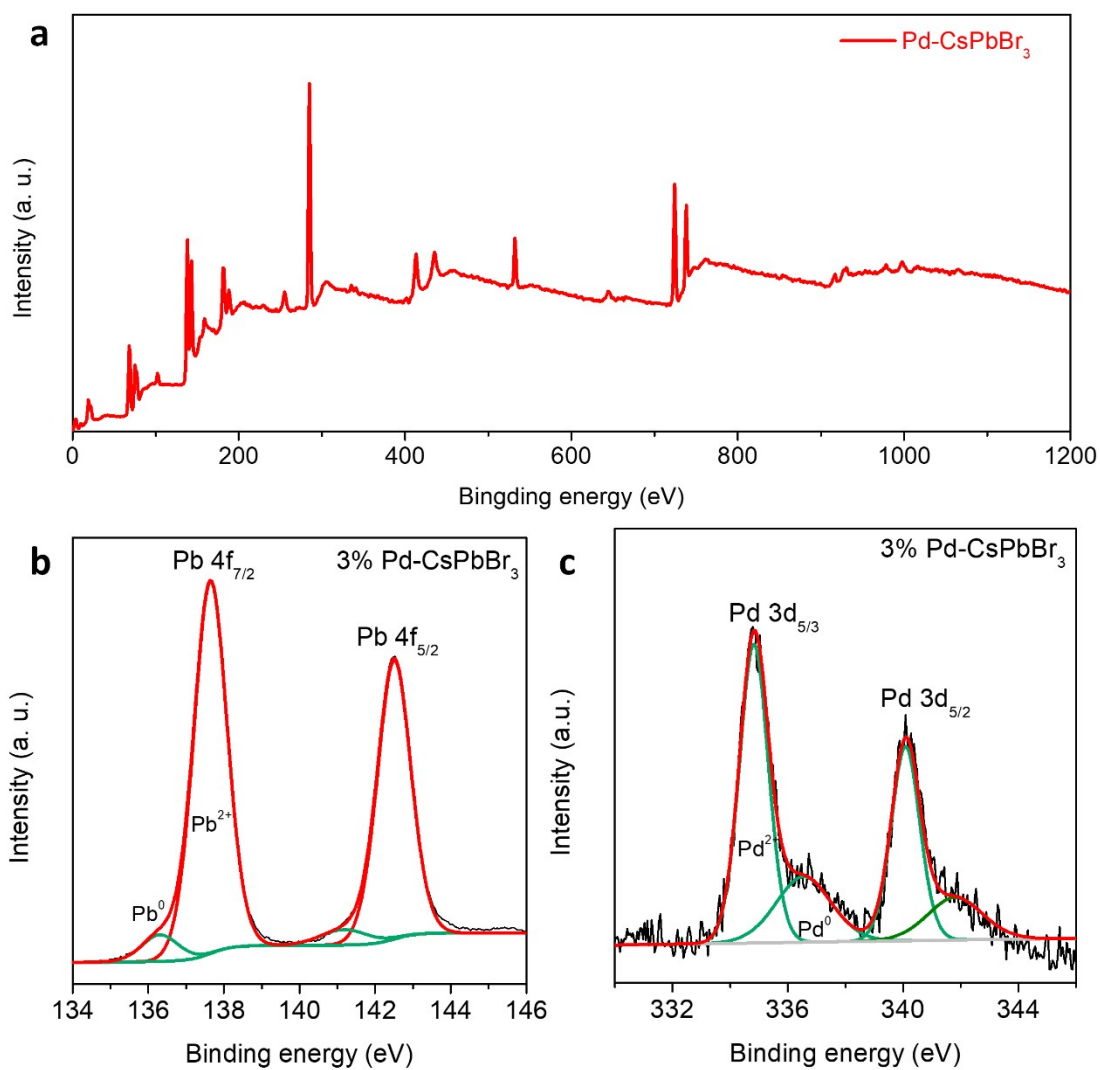
\*E-mail address: [heng.guo@swpu.edu.cn](mailto:heng.guo@swpu.edu.cn); [liy103@uestc.edu.cn](mailto:liy103@uestc.edu.cn); [xbniu@uestc.edu.cn](mailto:xbniu@uestc.edu.cn);



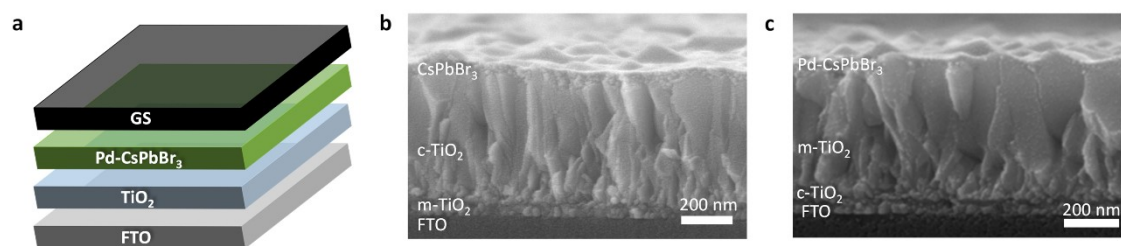
**Figure S1** EDS of Pd-CsPbBr<sub>3</sub> QDs



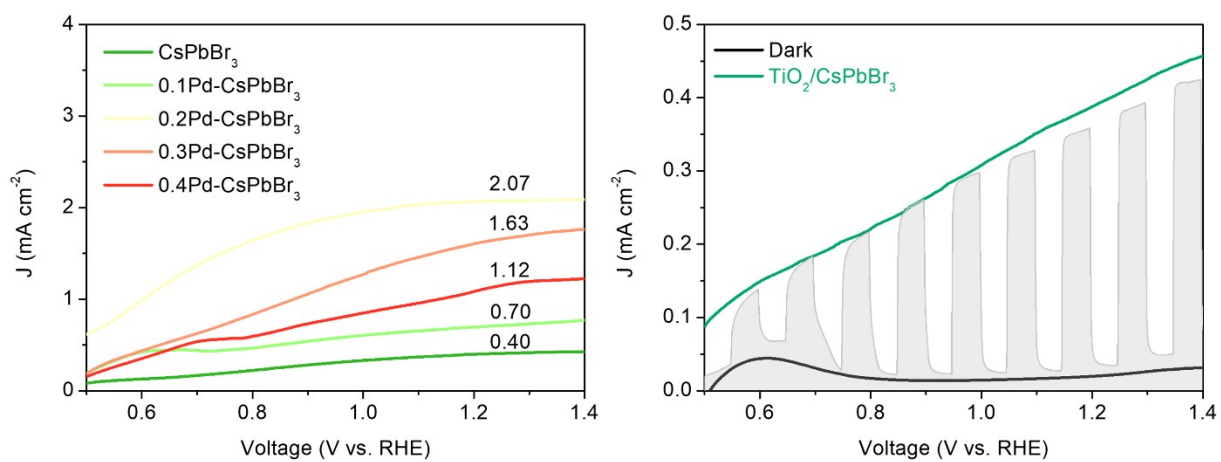
**Figure S2.** The cross-sectional SEM and EDS images of Pd-CsPbBr<sub>3</sub> QDs based device.



**Figure S3.** (a) The survey XPS spectrum of Pd-CsPbBr<sub>3</sub> QDs. The XPS (b) Pb 4f and Pd 3d spectrum of 3% Pd-CsPbBr<sub>3</sub> QDs.



**Figure S4.** (a) The schematic of Pd-CsPbBr<sub>3</sub> QDs based device layer by layer. The cross-sectional EDS of the (b) CsPbBr<sub>3</sub> and (c) Pd-CsPbBr<sub>3</sub> QDs.



**Figure S5.** LSV of the Pd-CsPbBr<sub>3</sub> QDs based photoanode with different concentrations of

Pd doping. The chopped plot of the CsPbBr<sub>3</sub> QDs based device.

**Table S1.** The cross-sectional element EDS mapping of Pd-CsPbBr<sub>3</sub> QDs based device.

Element	Weight percentage	Wt % Sigma
O	38.55	0.43
Ti	36.95	0.35
Pb	4.59	0.32
Br	12.89	0.20
Pd	2.60	0.22
Cs	4.42	0.35

**Table S2.** The Pd/Pb ratio versus dopant concentration of Pd-CsPbBr<sub>3</sub> QDs determined by ICP-OES.

Sample	Cx (ug/kg)	W (%)
Pd	19568723.0	0.2%
Cs	190178909.6	19.02%
Pb	285336133.0	28.53%
Br	84114215.4	8.41%

**Table S3.** High-resolution XPS peak positions of Cs, Br and Pb elements in CsPbBr<sub>3</sub> and Pd-CsPbBr<sub>3</sub> QDs.

Sample	Cs (eV)		Br (eV)		Pb (eV)	
	Cs	Cs	Br	Br	Pb	Pb
	3d <sub>5/3</sub>	3d <sub>3/2</sub>	3d <sub>5/3</sub>	3d <sub>3/2</sub>	4f <sub>7/2</sub>	4f <sub>5/2</sub>
CsPbBr <sub>3</sub>	722.0	735.9	66.2	67.3	136.4	141.3
Pd-CsPbBr <sub>3</sub>	722.0	735.9	66.2	67.3	136.6	141.5

**Table S4.** The corresponding band energy parameters of Pd-CsPbBr<sub>3</sub> and Pd-CsPbBr<sub>3</sub> QDs.

Sample	E <sub>GB</sub> (eV)	E <sub>F</sub> (eV)	E <sub>VB</sub> (eV)	E <sub>CB</sub> (eV)
CsPbBr <sub>3</sub>	2.35	-3.95	-5.95	-3.6
Pd-CsPbBr <sub>3</sub>	2.30	-4.02	-5.79	-3.49

**Table S5.** Fitted parameters of TRPL decay curves in perovskite films with using Pd-CsPbBr<sub>3</sub> and Pd-CsPbBr<sub>3</sub> QDs.

ETLs	A <sub>1</sub>	τ <sub>1</sub> [ns]	A <sub>2</sub>	τ <sub>2</sub> [ns]	τ [ns]
CsPbBr <sub>3</sub>	486.90	5.26	1304.30	0.91	3.87
Pd-CsPbBr <sub>3</sub>	899.99	2.28	396.35	13.38	10.28

The TRPL decay was fitted by a bi-exponential decay function with below equation:

$$PL_{intensity} = A_1 e^{\frac{-t}{\tau_1}} + A_2 e^{\frac{-t}{\tau_2}}$$

where  $A_1$  and  $A_2$  are time-independent coefficients of amplitude fraction for each decay component,  $\tau_1$  and  $\tau_2$  are decay time of a fast and slow component, respectively.