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## Supplementary Information

### In Situ Precipitation of Cs<sub>3</sub>Cu<sub>2</sub>I<sub>5</sub> Nanocrystals in Inorganic Glass with Long-Term Water Stability for X-ray Imaging

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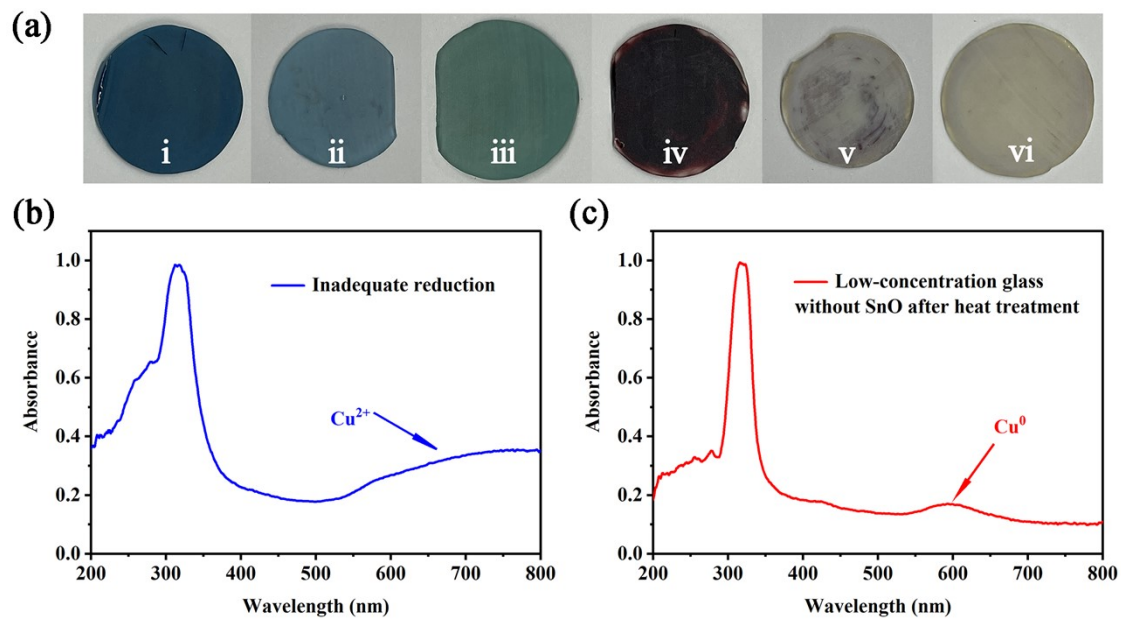
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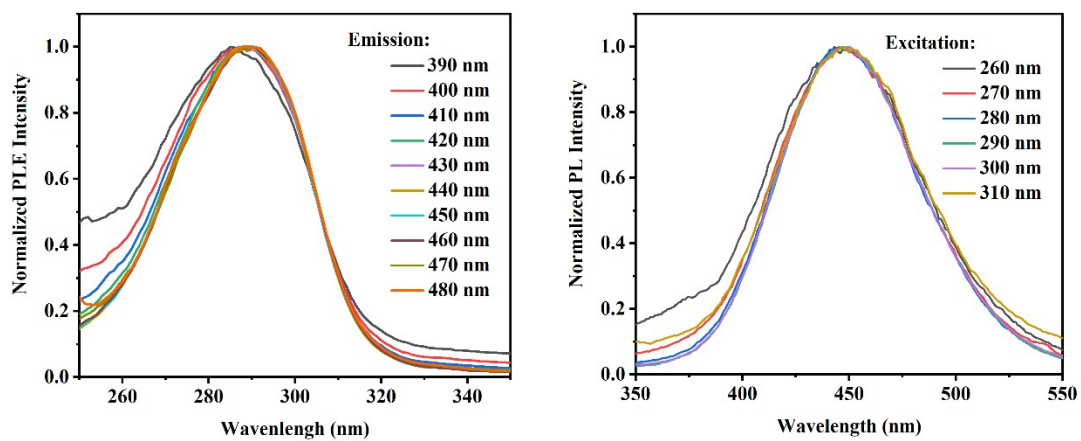
**Table S1.** Composition of Cs<sub>3</sub>Cu<sub>2</sub>I<sub>5</sub> NCs glass (mol.%) with varying SnO content.

Sample	SiO <sub>2</sub>	B <sub>2</sub> O <sub>3</sub>	ZnO	CaO	MgO	Al <sub>2</sub> O <sub>3</sub>	Cs <sub>2</sub> O	CuI	NaI	SnO
i	29	35	5	1	1	3	10	2	12	0
ii	29	35	5	1	1	3	10	2	12	1.94×10 <sup>-4</sup>
iii	29	35	5	1	1	3	10	2	12	2.92×10 <sup>-4</sup>
iv	29	35	5	1	1	3	10	2	12	5.84×10 <sup>-4</sup>
v	29	35	5	1	1	3	10	2	12	4.86×10 <sup>-4</sup>
vi	29	35	5	1	1	3	10	2	12	3.89×10 <sup>-4</sup>

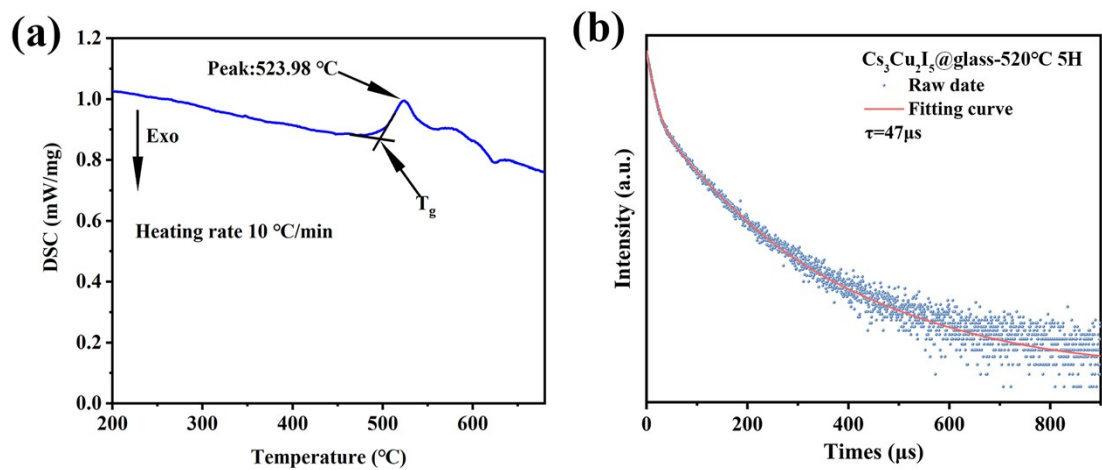
\*To accurately compare all glass compositions, the total molar content of the Cs<sub>3</sub>Cu<sub>2</sub>I<sub>5</sub> NCs glass composition was not set to 100%.



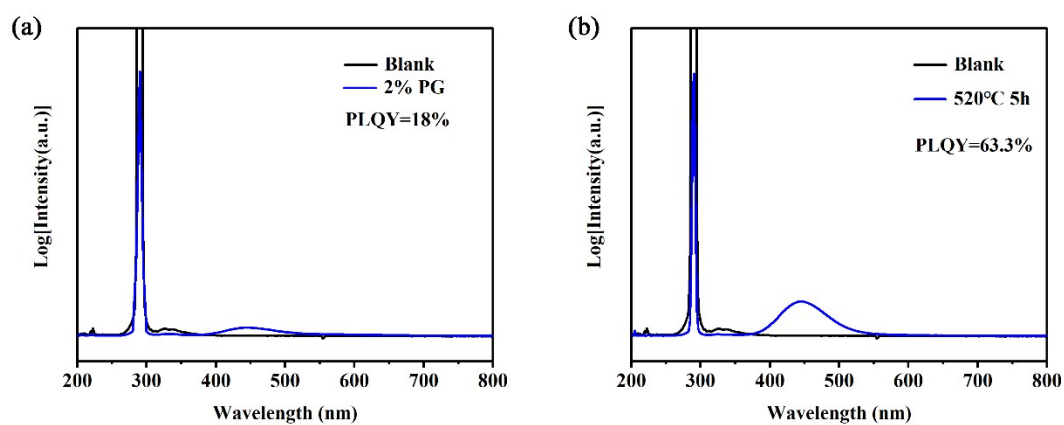
**Figure S1.** (a) Some sample photographs in the experimental optimization process. (b, c) Absorption spectrum of insufficiently reduced glass and low-concentration glass without SnO after heat treatment glass.



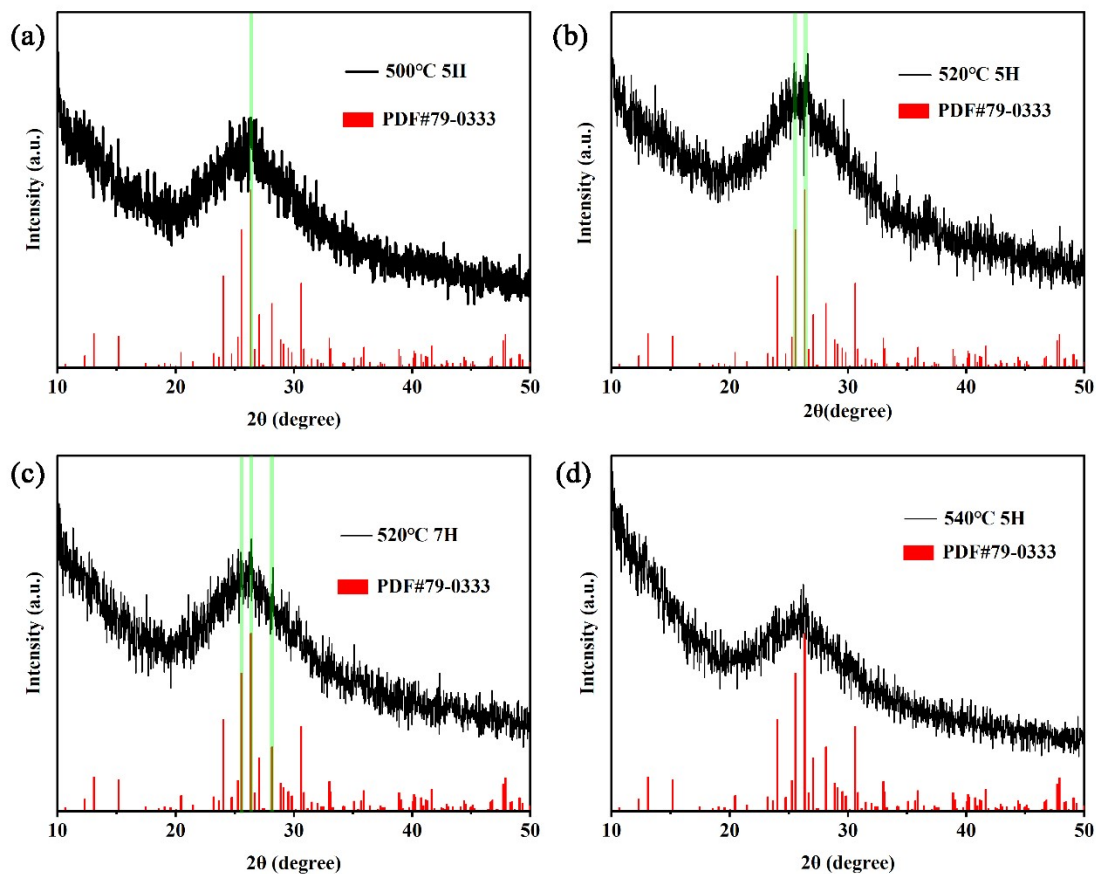
**Figure S2.** (a, b) PLE and PL spectra of  $\text{Cs}_3\text{Cu}_2\text{I}_5$  glass measured at different emission and excitation wavelengths, respectively.



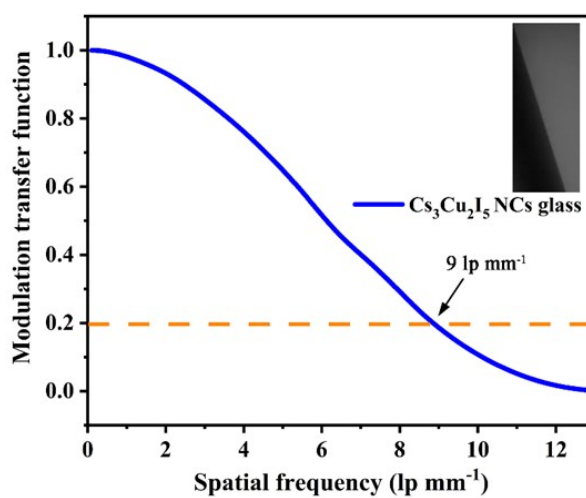
**Figure S3.** (a) DSC curves for Cs<sub>3</sub>Cu<sub>2</sub>I<sub>5</sub> glass. (b) Decay lifetimes of Cs<sub>3</sub>Cu<sub>2</sub>I<sub>5</sub> glass at 520 °C.



**Figure S4.** The PLQY spectrum of Cs<sub>3</sub>Cu<sub>2</sub>I<sub>5</sub> NCs glass.



**Figure S5.** XRD patterns of the glass at different heat treatment temperature.



**Figure S6.** Modulation transfer functions (MTF) of X-ray images obtained from  $\text{Cs}_3\text{Cu}_2\text{I}_5$  NCs glass.

**Table S2.** Compare the recently reported performance of Cs<sub>3</sub>Cu<sub>2</sub>I<sub>5</sub>.

Materials	Emission peak (nm)	PLQY (%)	stability	Ref
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> single crystal	445	91.2	air-stability	(1)
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Nanocrystals	441	67	air stability over 45 days	(2)
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Nanocrystals	442	42	stability in air	(3)
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Nanocrystals	445	29.2	air stability over 45 days	(4)
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Nanocrystals	445	73.7	air stability over 30 days	(5)
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Films	446	50	air stability over 50 days	(6)
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Films	441	58		(7)
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Nanocrystals	445	59	air stability over 90 days	(8)
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Powder	445	49.2		(9)
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Nanocrystals	443	30	air stability over 16 days	(10)
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Films	445	76	air stability over 150 days	(11)
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Powder	445	80	air stability over 30	(12)

			days	
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Nanocrystals	445	87	air stability over 35	(13)
			days	
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Powder	445	89	air stability over 46	(14)
			days	
Cs <sub>3</sub> Cu <sub>2</sub> I <sub>5</sub> Nanocrystal	446	63.3	water stability over 30	This
glass			days	work

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