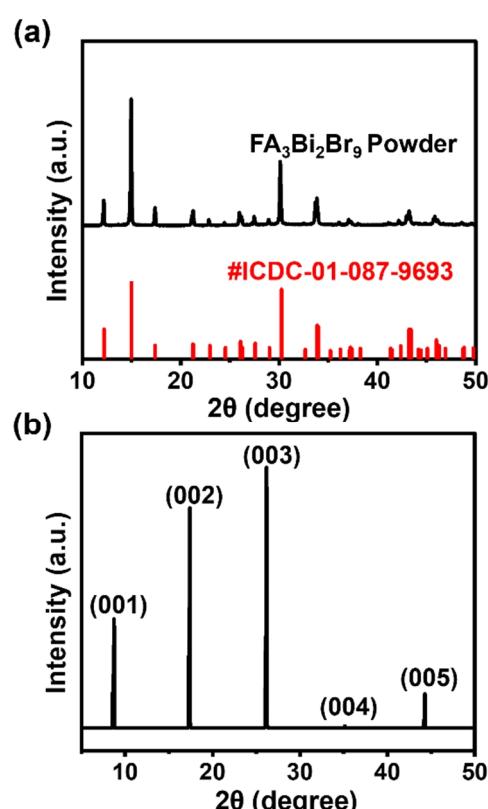


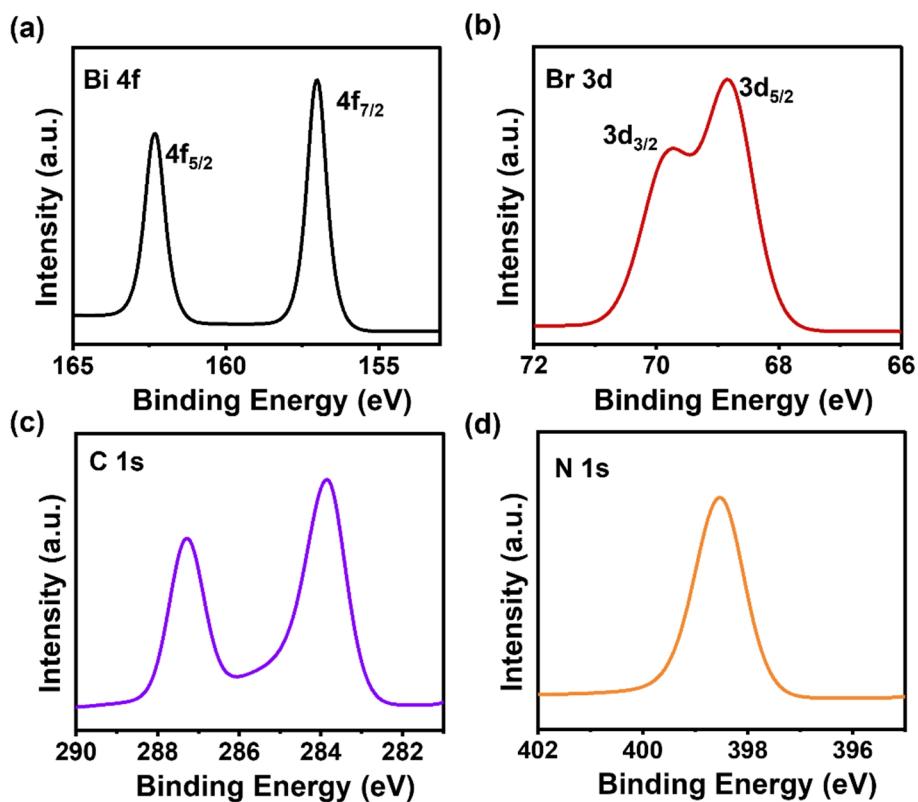
## Electronic Supplementary Information

### Centimeter-scaled lead-free two-dimensional $\text{FA}_3\text{Bi}_2\text{Br}_9$ perovskite single crystal for X-ray detection

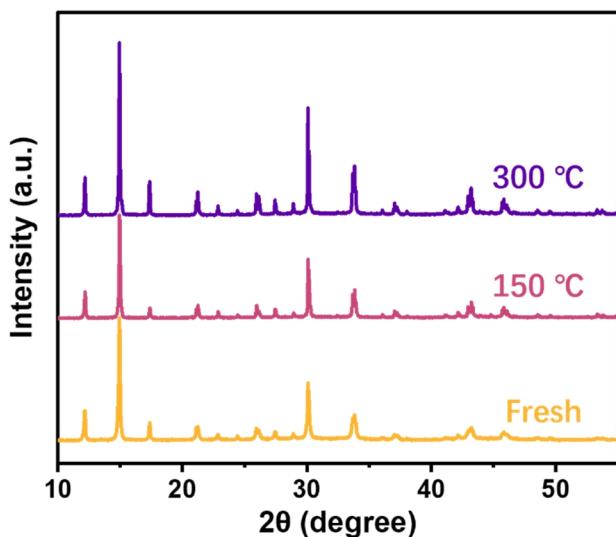
*Xiuting Luo, Lei Gao, Kezheng Tao, Qiang Li and Qingfeng Yan\**



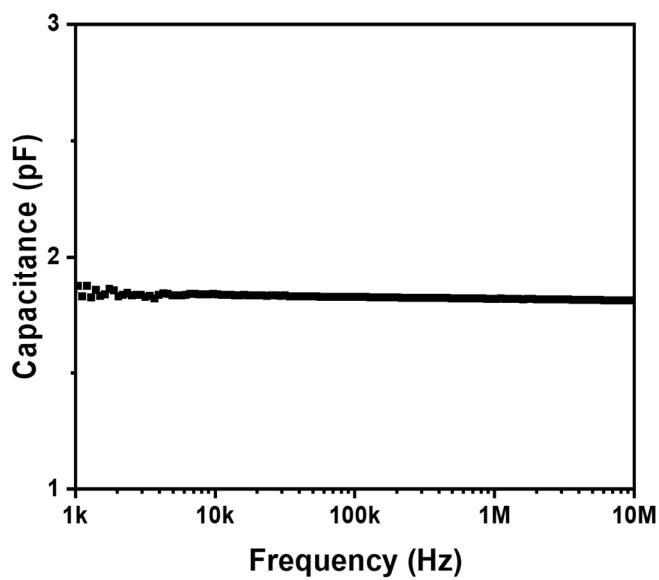
**Fig. S1** XRD pattern of (a)  $\text{FA}_3\text{Bi}_2\text{Br}_9$  powder. (b)  $\text{FA}_3\text{Bi}_2\text{Br}_9$  single crystal.



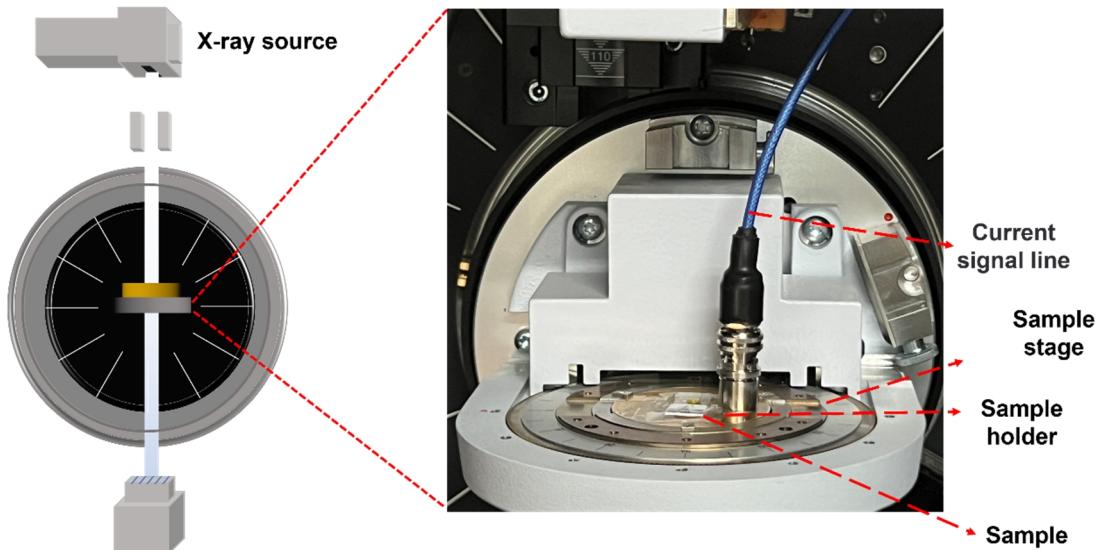
**Fig. S2** XPS high resolution fine spectrum of (a) Bi 4f and (b) Br 3d, (c) C 1s and (d) N 1s.



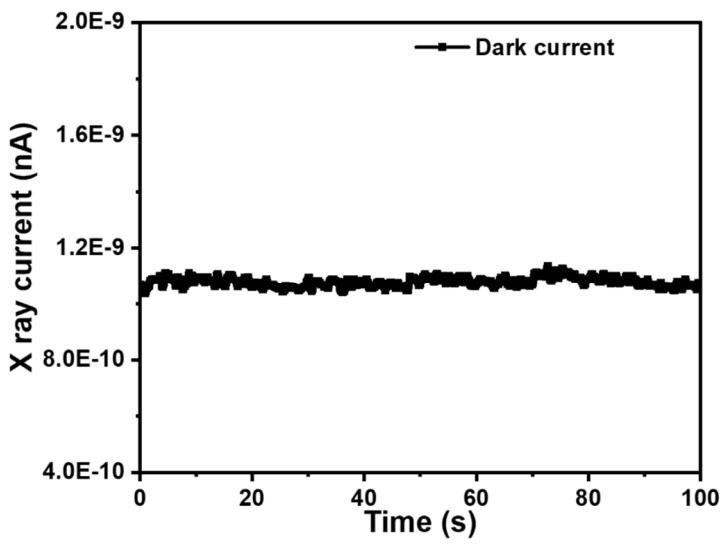
**Fig. S3** XRD patterns of fresh  $\text{FA}_3\text{Bi}_2\text{Br}_9$  powders and after heating at 150 °C and 300 °C for 2 h.



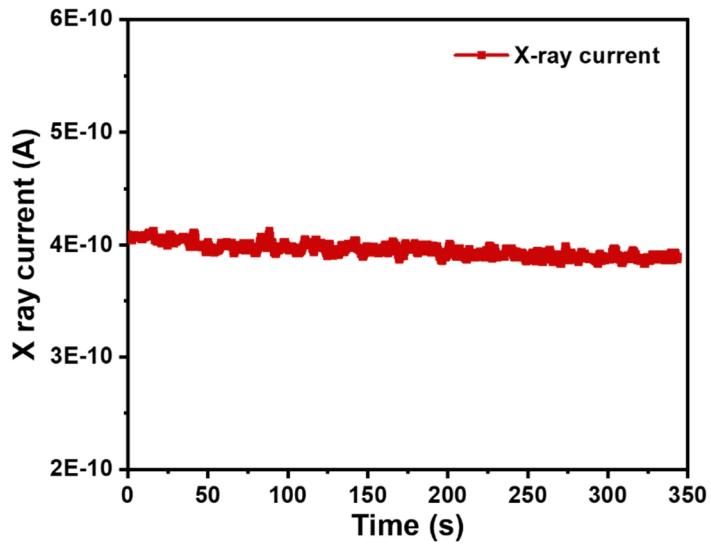
**Fig. S4** Frequency-capacitance curve of  $\text{FA}_3\text{Bi}_2\text{Br}_9$  SC.



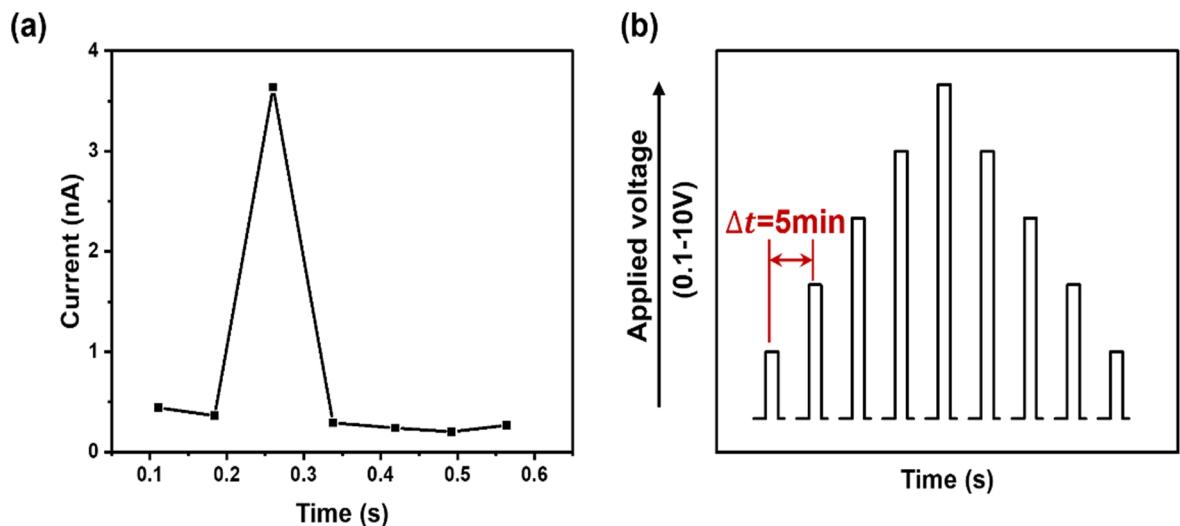
**Fig. S5** Home-made setup for X-ray detector measurement. The X-ray source was provided by a commercial Bruker D8 Advance XRD system equipped with a Cu tube.



**Fig. S6** Dark current of the FA<sub>3</sub>Bi<sub>2</sub>Br<sub>9</sub> SC detector under a bias voltage of 5V.



**Fig. S7** X-ray current of FA<sub>3</sub>Bi<sub>2</sub>Br<sub>9</sub> single crystal device under continuous X-ray irradiation on the basis of 5 V and a dose rate of 241  $\mu\text{Gy s}^{-1}$ .



**Fig. S8** (a) The processing time for single pulsed voltage in the Pulsed-SCLC measurement. (b) In the diagram of pulsed voltage applied, the time interval between different pulse voltage values is 5 min.

**Table S1.** Key parameters comparison of different perovskites-based X-ray detectors.

2D hybrid halide perovskite single crystal	Electrical field (V/cm)	$\mu\tau$ (cm <sup>2</sup> V <sup>-1</sup> )	Sensitivity (μC Gy <sup>-1</sup> cm <sup>-2</sup> )	Lowest detection limit (nGy s <sup>-1</sup> )	Reference
BA <sub>2</sub> EA <sub>2</sub> Pb <sub>3</sub> Br <sub>10</sub>	50	1.0 × 10 <sup>-2</sup>	6800	5500	<sup>1</sup>
(F-PEA)PbI <sub>4</sub>	1333	5.1 × 10 <sup>-4</sup>	3402	23	<sup>2</sup>
(BA) <sub>2</sub> CsAgBiBr <sub>7</sub>	50	1.21 × 10 <sup>-3</sup>	4.2	-	<sup>3</sup>
(CPA <sub>4</sub> )AgBiBr <sub>8</sub>	50	1.0 × 10 <sup>-3</sup>	0.8	-	<sup>4</sup>
Cu-(PMA) <sub>2</sub> PbI <sub>4</sub>	12500	8.05 × 10 <sup>-3</sup>	283	2130	<sup>5</sup>
(BDA)PbI <sub>4</sub>	3100	4.43 × 10 <sup>-4</sup>	242	430	<sup>6</sup>
(DFPIP) <sub>4</sub> AgBiI <sub>8</sub>	~1000	1.10 × 10 <sup>-5</sup>	188	3130	<sup>7</sup>
(PA) <sub>4</sub> AgBiBr <sub>8</sub>	1250	-	6.89	2070000	<sup>8</sup>
FA <sub>3</sub> Bi <sub>2</sub> Br <sub>9</sub>	1000 2000	4.53 × 10 <sup>-4</sup>	410 623	2100	This work

\*BA = n-butylammonium

EA = ethylammonium

CPA = chloropropylammonium

BDA= 1,4-butanediammonium

DFPIP= 4,4-difluoropiperidinium

F-PEA = fluorophenethylammonium

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