

Electronic Supplementary Information

Efficient Emission in Copper-doped Cs_3ZnX_5 (X = Cl, I) for white LED and X- ray Scintillator

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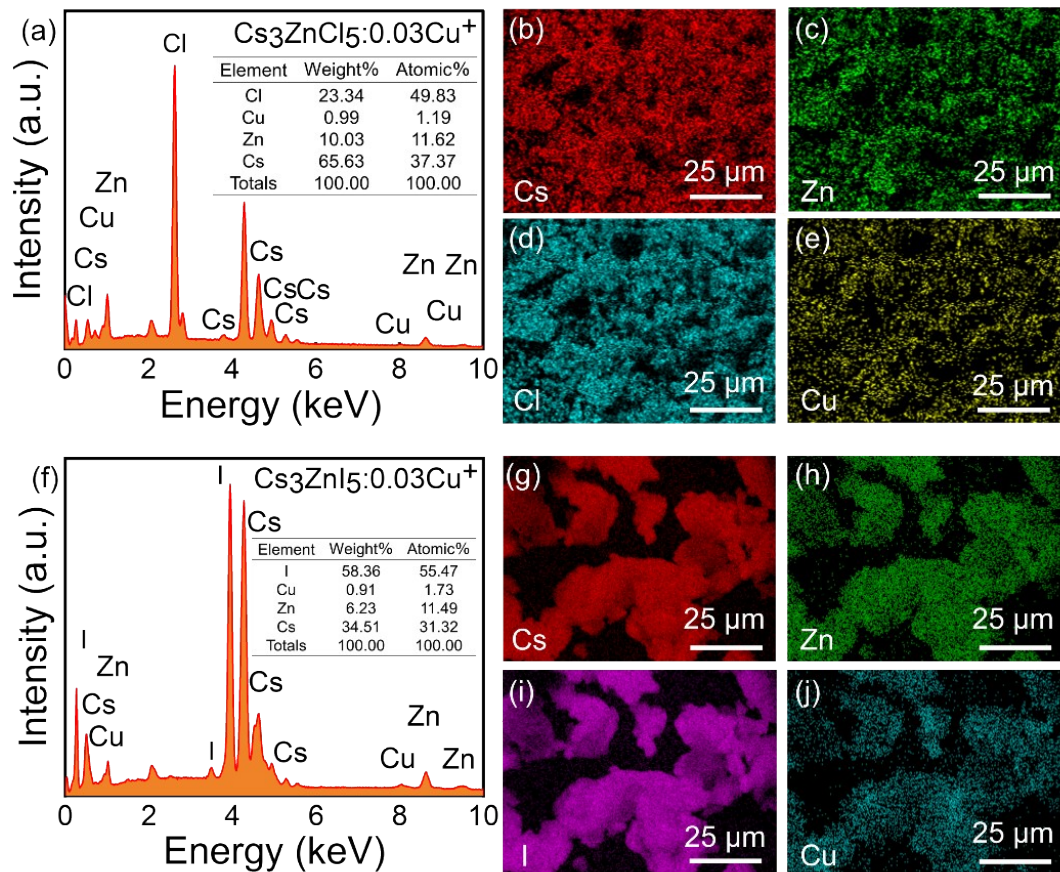


Fig. S1 (a) EDS spectra of $\text{Cs}_3\text{Zn}_{0.97}\text{Cu}_{0.03}\text{Cl}_{4.97}$. (b-e) Elemental mapping of one representative particle of sample $\text{Cs}_3\text{Zn}_{0.97}\text{Cu}_{0.03}\text{Cl}_{4.97}$. (f) EDS spectra of $\text{Cs}_3\text{Zn}_{0.97}\text{Cu}_{0.03}\text{I}_{4.97}$. (g-j) Elemental mapping of one representative particle of sample $\text{Cs}_3\text{Zn}_{0.97}\text{Cu}_{0.03}\text{I}_{4.97}$.

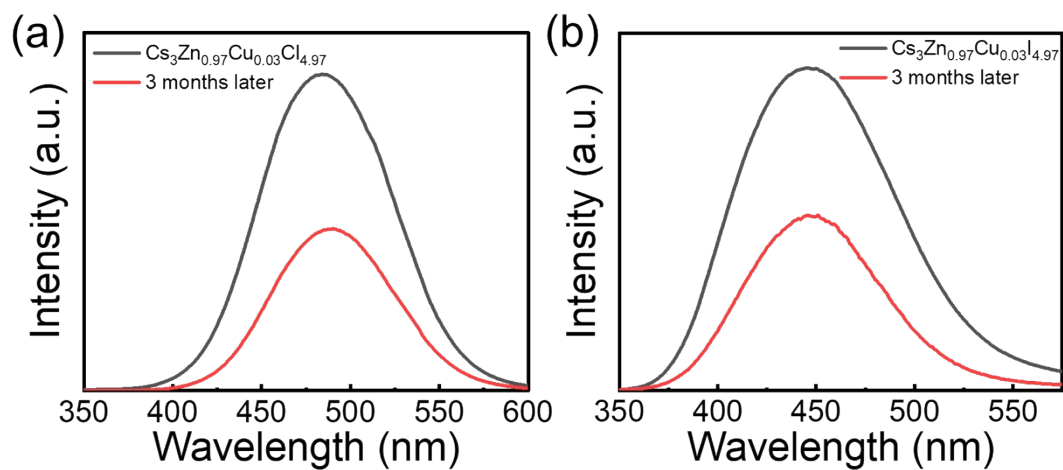


Fig. S2 PL spectra of original crystals (a) $\text{Cs}_3\text{Zn}_{0.97}\text{Cu}_{0.03}\text{Cl}_{4.97}$, (b) $\text{Cs}_3\text{Zn}_{0.97}\text{Cu}_{0.03}\text{I}_{4.97}$, and the corresponding crystals after storage in air for three months.

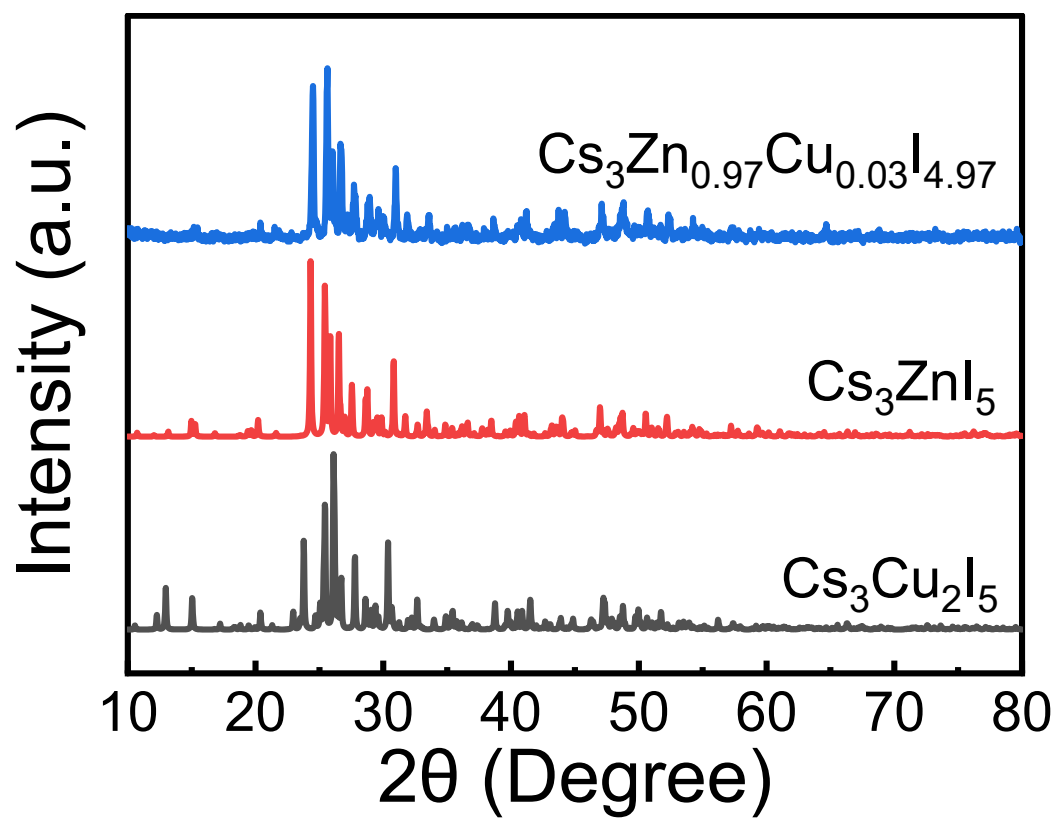


Fig. S3 XRD pattern of $\text{Cs}_3\text{Zn}_{0.97}\text{Cu}_{0.03}\text{I}_{4.97}$ compared with $\text{Cs}_3\text{Cu}_2\text{I}_5$ and Cs_3ZnI_5 .

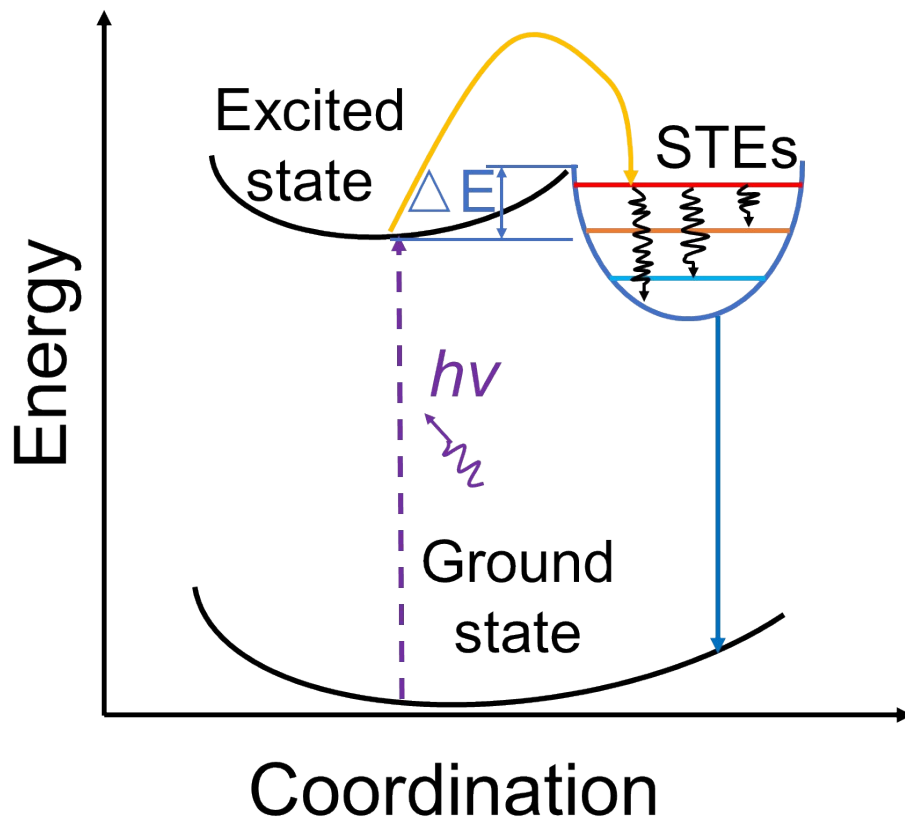


Fig. S4 Schematic diagram of luminescence mechanism of copper-doped Cs_3ZnX_5 ($X = \text{Cl, I}$) series ($\text{Cs}_3\text{Zn}_{1-x}\text{Cu}_x\text{X}_{5-x}$ ($x = 0 - 0.15$))

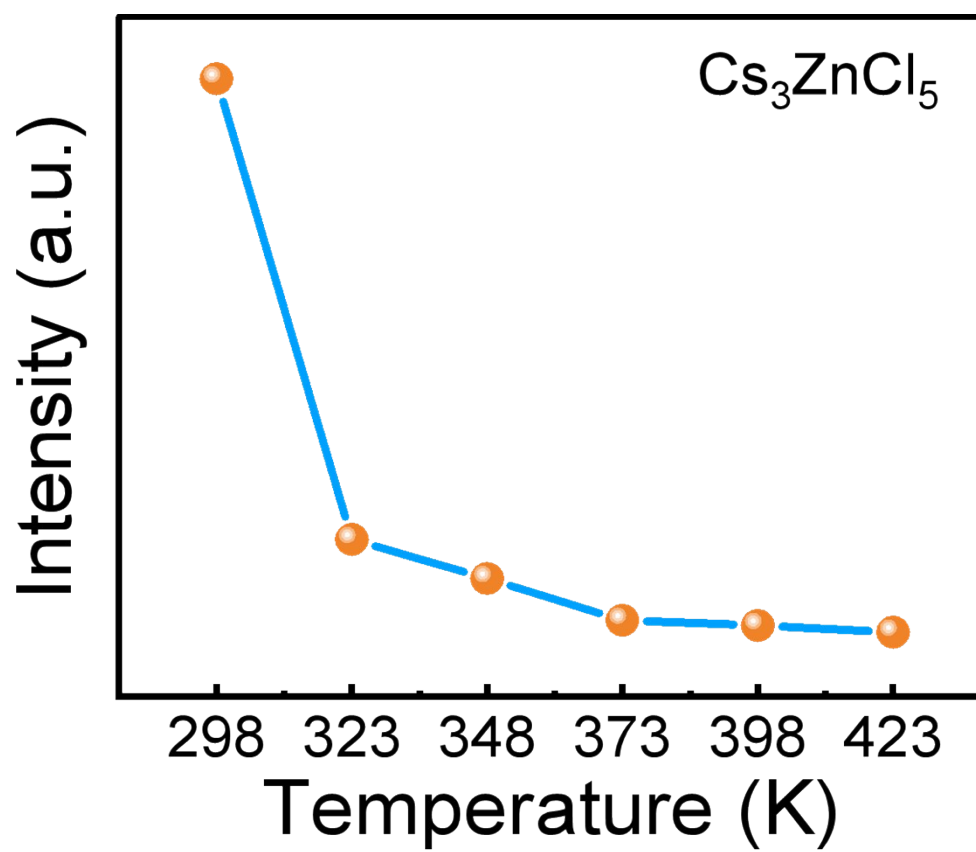


Fig. S5 Variation of the emission intensity for Cs_3ZnCl_5 with temperature.

Table S1 Photoluminescence excitation (PLE), full width at half maximum (FWHM), quantum efficiency (PLQY), and chromaticity coordinates (x, y) of phosphors.

Sample no.	composition	λ_{exc} (nm)	FWHM	PLQY(%)	CIE(x,y)
1	$\text{Cs}_3\text{Zn}_{0.97}\text{Cu}_{0.03}\text{Cl}_{4.97}$	265	120	90.2	(0.1494, 0.2821)
2	$\text{Cs}_3\text{Zn}_{0.97}\text{Cu}_{0.03}\text{I}_{4.97}$	310	70	11.6	(0.1514, 0.1215)
3	CASN:Eu ²⁺	440	60	96.0	(0.6470, 0.3470)
4	White-light LED	Ra = 92.7	CCT = 4236 K		(0.3464, 0.2706)