Asymmetrical Organic D- π -A Conjugate with 'V'-Shape Crystal Packing: Quest to Transcend the Limits of Photophysical Properties and Applications

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Scheme S1. Synthetic pathway to prepare yl)styryl)anthracen-9-yl)-1H-indole (*IAPY*)¹

(E)-1-hexyl-3-(10-(4-(pyridin-2-

Synthetic procedure and characterization of IAPY¹

Diethyl ((10-(1-hexyl-1H-indol-3yl)anthracen-9-yl)methyl)phosphonate (0.37 mmol, 1 equiv., 0.2 g), 4-(pyridin-2-yl)benzaldehyde (0.82 mmol, 2.2 equiv., 0.15 g) and K^tOBu base (0.93 mmol, 2.5 equiv., 0.1 g) were taken in a two-neck round bottom flask and allowed for vacuum drying for 5 min. Later, N₂ was purged throughout the flask and kept sealed with a silicon septum. An N₂-containing balloon was placed at one end, and at the other neck, dry THF (15 mL) was injected into the flask to stir for 5 h. The progress of the reaction was observed by eluting TLC into hexane/ EtOAc (12% EtOAc into hexane (v/v)) medium to see an intense green fluorescent product spot formed at $R_f = 0.29$. The reaction mixture was then worked up with 30 mL of an EtOAc and brine water mixture of 1:2 EtOAc:H₂O. The organic layer was separated, passed through a Na₂SO₄ layer, finally to prepare a slurry with silica gel (60–120 mesh) for performing column chromatography (10% EtOAc into hexane (v/v)), which yielded (71%, 0.15 g) product which is a solid yellowish compound under room light but a bright yellowish-green fluorescent compound under a UV-365 nm bulb. NMR spectroscopy: ¹H NMR (400 MHz, CDCl₃, 25 °C, δ): 8.74 (s, 1H), 8.46 (d, *J* = 8.76 Hz, 2H), 8.13–7.93 (m, 5H), 7.82–7.75 (m, 4H), 7.53–7.43 (m, 3H), 7.32–7.23 (m, 5H), 7.14–7.01 (m, 3H), 4.30 (t, J = 7.04 Hz, 2H), 1.99 (quint, J = 7.32 Hz, 2H), 1.46–1.23 (m, 6H), 0.90 (t, J = 7.04 Hz, 3H). ¹³C NMR: (100 MHz, CDCl₃,25°C, δ): 156.9, 149.8, 138.8, 138.1, 136.9, 136.1, 132.4, 131.6, 130.2, 129.8, 129.7, 128.9, 127.9, 127.3, 127.2, 127.0, 126.2, 126.1, 125.3, 124.9, 122.2, 121.8, 120.6, 120.5, 119.5, 111.9, 109.6, 46.6, 31.5, 30.3, 26.8, 22.6, 14.0. IR (KBr, cm⁻¹): 3445, 3048, 2944, 2849, 1580, 1467, 1431, 1345, 1322, 1233, 1147, 1017, 958. M.p.: 154–156^oC. Mass spectrometry: HRMS (ESI) m/z: calcd for C₄₁H₃₆N₂ 556.2878, found: 557.2674 [M + H]⁺.

Materials and Methods

Steady-state absorption and fluorescence measurements

Solid-state absorption spectra were recorded u sing a JASCO-500 spectrophotometer, and the solution-state absorption spectra were recorded using a UV-vis-NIR spectrophotometer (Hitachi F7000, Japan). Solution-state emission spectra were obtained using an FP-6300 spectrometer (JASCO), and by using a 10 mm path-length quartz cuvette, while the solid-state emission spectra were received using a fluorimeter (Fluorolog, HORIBA). All the emission spectra were obtained at the corresponding absorption wavelengths.

Absolute/relative quantum yield and time-resolved lifetime decay measurement

The absolute quantum yield (Φ_f) values for solid samples were measured using the calibrated integrating sphere method with a fluorimeter (Fluorolog, HORIBA), and absolute errors within $\sim \pm 2\%$ have been included while for the relative quantum yield (relative Φ_f) in the solution state, the error is $\pm 5\%$.

Time-resolved fluorescence measurements were measured using a time-correlated singlephoton counting (TCSPC) unit (Horiba Deltaflex). The laser used for all the samples was of 510 nm. All the measurements were undergone at room temperature. The decay fitting was done by keeping the χ^2 value close to 1.

The average lifetime was obtained by fitting the decay profiles to a tri- or bi-exponential function eqn-1. Fit = A1.exp (-t/ τ 1) + A2.exp (-t/ τ 2) + A3.exp (-t/ τ 3)(eq-1) α 1, α 2 are the weighted components and τ 1, τ 2, τ 3 are the individual lifetime components of the decay. The qualities of the fit were understood by judging the chi square (χ 2) values. The rate constants are calculated by using: $k_r = [\Phi_f / \tau_{avg}] s^{-1}$; $k_{nr} = [1-\Phi_f / \tau_{avg}] s^{-1}$

Powder X-ray diffraction and IR spectra

The PXRD spectra of the pristine, ground, and recovered forms of the compound **IAPY** were recorded by using a Rigaku Ultima IV X-ray diffractometer, and keeping the parameters constant for all samples, with a step width of 0.21 and a scan rate of 21 min1 from 5–501 (Cu K α radiation, I = 1.54 Å). And an FT-IR spectrometer (FT/IR-4200, Jasco) was utilized to record the IR spectra of the samples. Solid samples were mixed with KBr to form pellets to record the spectra.

Calculation of crystallinity from PXRD

Crystallinity is considered as the ratio between the area of crystalline peaks to the area of all peaks. Here the crystallinity is revealed as a percentage just by multiplying the ratio with 100. The decovolution of the PXRD raw data was completed using oringin lab software (Origin Pro 8.5, student version) having a good increment of peaks. Then by selecting the plot the 'Analysis' menu should be clicked followed by the 'Peaks and Baseline' and 'Peak Analyzer' to reach to the 'Open Dialogue'. Here, the 'Peak Analyzer Window' will be opened to perform 'Integrate Peaks.' Followed by the 'Next' option, the 'Constant Y' should be selected to well set up the baseline. Again, 'Next' and 'Find' options should be clicked and 'Untick' to 'Enable Auto Find' should be done. Next 'Add' option should be clicked and all the relevant peaks should be selected accordingly. Next, the 'Done' option should be clicked. In the 'Peak Finding Setting' the 'Direction' should be selected as 'Positive' and 'Next' button should be clicked again. Here 'Adjust ad Preview Graph' should be done. Next the 'Finish' button has to be clicked. Now, from the grown window, 'Select Data' then 'Select Integral Results' should be copied and pasted in a new excel sheet. It was named as 'Area of crystalline peaks'. Here sum it in excel. The area will be obtained. Therefore, the same process should be undergone again but this time not specific peaks but the 'Total area' should have to be integrated and the total area thus can be calculated using the excel sheet. Finally, the area under relevant crystalline peaks should be divided by the total area under the PXRD signal and multiplied by 100 to get the percentage of crystallinity.²

Differential scanning calorimetry analysis

DSC thermograms were recorded by using a Themys One⁺ (Setaram) instrument keeping the temperature in the range of 30–300 °C with a ramp rate of 2 °C min⁻¹ under a N₂ atmosphere.

Density functional theory studies

All the density functional theoretical (DFT) calculations were undergone using the ORCA Version 5.0.3 quantum chemical software package.^{3,4} Our DFT calculations were done using different DFT functionals (viz., CAM-B3LYP, PBE0, B3LYP, WB97X, BHandHLYP and M06-2X) and 6-31G* basis set to compare with experimental results.⁵ The excited stats (S₁) geometry optimization was done using time-dependent DFT (TDDFT). Root mean square deviation (RMSD) calculations were performed using the Kabsch algorithm.⁶ All structural and MOS were made visualized using Avogadro software.⁷



Fig. S1 Different planes of IAPY



Fig. S2 single-crystal with the dihedral angles in three different twisted sites



Fig. S3 'V'-shaped herringbone packing of IAPY



Fig. S4 Label atom for IAPY

Table S1 Short-contacts

No	Atom1	Atom2	Length (A)
1	C13	H27	2.60
2	C13	H17	2.55
3	C12	C27	3.34
4	C12	H27	2.76
5	N1A	H35A	2.48
6	C14	C17	3.22
7	C14	H17	2.64
8	C14	C4	3.33
9	C8	H6A	2.89
10	C24	H29E	2.89
11	H24	C29A	2.46
12	C20	C30A	3.22
13	H20	C30A	2.63
14	H20	C29A	2.72
15	C38A	H33A	2.57
16	H38A	C33A	2.55
17	H38A	H33A	1.97
18	H3B	H29C	2.36
19	H30A	H32A	2.37
20	C36A	H29E	2.86
21	H36A	C29A	2.85
22	H36A	H29E	2.39



Fig. S5 Rotational and screw-axes of IAPY



Fig. S6 Glide plains in packing of IAPY



Fig. S7 Contact surface void of IAPY



Fig. S8a Solid-state (a) absorbances (b) emission and MIEE properties of IAPY



Fig. S8b Photostability of IAPY pristine



Fig. S9 Software generated dendrogram for IAPY pristine

Table S2 Peak list for IAPY pristine PXRD	
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No.	2-theta(deg)	d(ang.)	Height(cps)	FWHM(deg)	Int. I(cps deg)	Int. W(deg)	Asym. factor
1	11.182(5)	7.907(3)	573(24)	0.166(10)	133(11)	0.23(3)	2.2(2)
2	11.584(13)	7.633(8)	453(21)	0.64(2)	404(11)	0.89(7)	2.2(2)
3	13.366(10)	6.619(5)	135(12)	0.26(3)	41(4)	0.30(6)	0.58(6)
4	14.016(2)	6.3136(11)	1044(32)	0.342(7)	420(8)	0.40(2)	0.58(6)
5	16.309(6)	5.430(2)	899(30)	0.177(11)	217(13)	0.24(2)	0.53(11)
6	16.606(8)	5.334(3)	727(27)	0.291(15)	288(14)	0.40(3)	0.53(11)
7	18.881(5)	4.6962(11)	349(19)	0.254(18)	134(5)	0.38(3)	0.6(2)
8	19.836(3)	4.4723(7)	678(26)	0.255(9)	221(6)	0.33(2)	1.1(2)
9	21.62(2)	4.106(4)	120(11)	0.20(5)	26(4)	0.22(6)	0.7(4)
10	22.03(3)	4.031(6)	199(14)	0.26(3)	57(7)	0.29(6)	0.7(4)
11	23.028(7)	3.8591(12)	821(29)	0.26(2)	285(19)	0.35(3)	0.61(9)
12	23.309(12)	3.8132(19)	372(19)	0.16(2)	79(13)	0.21(5)	0.61(9)
13	23.665(12)	3.7565(18)	624(25)	0.382(17)	323(13)	0.52(4)	0.61(9)
14	25.567(8)	3.4813(11)	827(29)	0.325(7)	306(7)	0.37(2)	0.87(9)
15	26.267(14)	3.3901(17)	89(9)	0.17(4)	17(3)	0.19(5)	0.87(9)
16	27.186(8)	3.2775(10)	170(13)	0.28(2)	63(4)	0.37(5)	0.8(4)
17	28.22(3)	3.160(3)	140(12)	0.29(3)	51(4)	0.36(6)	0.7(3)
18	31.12(2)	2.872(2)	72(8)	0.44(6)	36(6)	0.50(14)	0.5(4)
19	45.21(9)	2.004(4)	52(7)	0.31(6)	17(4)	0.33(12)	0.7(8)



Fig. S10 Software generated dendrogram for IAPY ground

No.	2-theta(deg)	d(ang.)	Height(cps)	FWHM(deg)	Int. I(cps deg)	Int. W(deg)	Asym. factor
1	10.540(14)	8.387(11)	165(13)	0.52(3)	104(5)	0.63(8)	1.48(15)
2	11.94(2)	7.405(13)	252(16)	0.84(3)	260(7)	1.03(9)	1.48(15)
3	13.06 <u>2(12)</u>	6.773(6)	278(17)	0.488(19)	167(5)	0.60(5)	1.48(15)
4	14.057(17)	6.295(8)	193(14)	0.62(4)	148(7)	0.77(9)	1.48(15)
5	16.309(7)	5.431(2)	253(16)	0.546(18)	152(6)	0.60(6)	0.96(15)
6	20.145(13)	4.404(3)	244(16)	0.43(2)	111(8)	0.46(6)	0.78(9)
7	21.12(3)	4.204(6)	142(12)	1.12(9)	170(10)	1.19(17)	0.78(9)
8	22.81(3)	3.896(5)	238(15)	1.15(5)	291(11)	1.22(12)	0.78(9)
9	24.113(13)	3.688(2)	294(17)	0.57(2)	180(8)	0.61(6)	0.78(9)
10	25.40(3)	3.504(4)	202(14)	1.21(6)	261(9)	1.29(14)	0.78(9)
11	27.79(3)	3.208(4)	69(8)	0.67(8)	50(5)	0.72(16)	0.78(9)
12	44.45(13)	2.037(6)	16(4)	1.8(4)	30(9)	1.9(10)	0.7(7)

Table S3 Peak list for IAPY ground PXRD



Fig. S11 Surface accessible void of IAPY







Fig. S13 Hirshfeld surface of IAPY



Fig. S14 Hirshfeld surface and van der Waals and the non-covalent interactions to the packing for **IAPY**

Table S4 Inter-molecular potentials

Calculated inter-molecular potentials: mol1 mol2 distance energy (kJ/mol) 0 1 8.5606 -56.7549 2 3 8.5606 -56.7549 4 5 8.5606 -56.7549 0 6 16.464 -56.1803 2 7 16.464 -56.1803 8 3 16.464 -56.1803 0 2 8.66273 -53.916 9 10 8.66273 -53.916 11 7 8.66273 -53.916 12 13 8.66273 -53.916 7 6 8.66273 -53.916 1 3 8.66273 -53.916 4 14 8.66273 -53.916

Hydrogen normalisation: On

Cluster Energy: Cluster 1 PE = -122.19 kJ/mol 39 interactions PE = -124.06 kJ/mol 119 interactions PE = -124.16 kJ/mol 159 interactions PE = -124.18 kJ/mol 179 interactions PE = -124.19 kJ/mol 189 interactions PE = -124.19 kJ/mol 199 interactions Cluster 2 PE = -310.85 kJ/mol 39 interactions PE = -319.49 kJ/mol 119 interactions PE = -320.21 kJ/mol 159 interactions PE = -320.42 kJ/mol 179 interactions PE = -320.50 kJ/mol 189 interactionsPE = -320.55 kJ/mol 199 interactions

Total packing energy = -444.7 kJ/mol

Potential = $A^{exp}(-Br) - Cr(-6)$

Unified (UNI) pair-potential parameters: atom1 code1 atom2 code2 A B C C15 3 C15 3 226145.2 3.47 2418.0 C15 3 N5 23 491494.0 3.86 2791.0 C15 3 H27 1 120792.1 4.10 472.8 N5 23 N5 23 365263.0 3.65 2891.0 N5 23 H27 1 228279.0 4.52 502.1 H27 1 H27 1 24158.0 4.01 109.2



Fig. S15 Decay profile of IAPY for MIEE property



Fig. S16 SEM images for IAPY (a) pristine (b) ground (c) recovered



Fig. S17 Absorbances for IAPY SSAC



Fig. S18a Changes in PXRD pattern for IAPY due to acidofluorochromism



Fig. S18b SEM images for IAPY (a) acid-fumed (b) base-fumed



Fig. S19 Decay profile for IAPY (a) acid-fumed (IAPYH) (b) base-fumed



Fig. S20 DSC for the '4th day post to base treatment' form of IAPY



Fig. S21 Absorbances of IAPY in different solvents



Fig. S22 TD-DFT optimized structures and calculations for IAPY in Hexane and N,N-DMF using CAM-B3LYP/6-31G* level of theory.

Cyclic voltammetry study

Procedure and calculation: The HOMO/LUMO energy levels were calculated from the oxidation onset/reduction potentials in due course of positive/negative scans. For IAPY, 10⁻⁵ M of DCM (dichloromethane) and 10⁻⁵M of *N*,*N*-DMF (dimethylformamide) solution were used with glassy carbon (GC) electrode as working electrode (WE) along with the Pt-wire and Ag/AgCl were used as a counter electrode (CE) and reference electrode (RE) respectively with a 0.1 M tetrabutylammonium perchlorate (TBAP) in the respective solvents as an electrolyte in a typical three-electrode system at normal condition.



Fig. S23a HOMO-LUMO calculation from CV for IAPY in (a) DCM, and (b) N,N-DMF

 $E_{HOMO}(DCM) = -(E^{OX}_{Onset} - 0.49 + 4.8) eV = -(1.2135 - 0.49 + 4.8) eV = -5.5235 eV$

 $E_{LUMO}(DCM) = -(E^{Red}_{Onset} - 0.49 + 4.8) eV = -(-1.662 - 0.49 + 4.8) eV = -2.648 eV$

Energy Gap (DCM) = E_{LUMO} - E_{HOMO} = (-2.648 + 5.5235) eV = 2.8755 eV

Again,

 $E_{HOMO}(DMF) = -(E^{OX}_{Onset} - 0.49 + 4.8) eV = -(1.1179 - 0.49 + 4.8) eV = -5.4279 eV$

 $E_{LUMO}(DMF) = -(E^{Red}_{Onset} - 0.49 + 4.8) eV = -(-1.6374 - 0.49 + 4.8) eV = -2.6726 eV$

Energy Gap (DMF) = E_{LUMO} - E_{HOMO} = (-2.6726 + 5.4279) eV = 2.7553 eV



Fig. S23b HOMO-LUMO for IAPY in Hexane and DMF using $CAM\mbox{-}B3LYP\mbox{/}6\mbox{-}31G\mbox{*}$



Fig. S23c HOMO-LUMO for IAPY in Hexane and DMF using PBE0/6-31G* level of theory



Fig. S23d NTOs for IAPY in Hexane and DMF using $CAM\mbox{-}B3LYP\mbox{/}6\mbox{-}31G\mbox{*}$



Fig. S23e Hole and electron NTOs in the S_1 state of IAPY in DCM and DMF solvent with PBE0/6-31G* level of theory

Properties/	B3LY	CAM-	MACON	BHnad	DDEA	WD07V
Functionals	Р	B3LYP	MUOZA	HLYP	PBEU	W B9/A
HOMO	4 0 4 8 2	6 2572	6 1002	5 9096	5 2752	7 1 9
(DMF)	-4.9462	-0.2372	-0.1992	-3.8980	-3.2732	-7.10
LUMO	1 9692	0.770	1 2040	0.9649	1 0227	0.010
(DMF)	-1.0005	-0.779	-1.2049	-0.0040	-1.9327	0.019
Energy						
gap	3.080	5.478	4.994	5.034	3.343	7.199
(DMF)						
HOMO	1 0274	6 7353	6 2675	5 8744	5 2505	7 1550
(DCM)	-4.9274	-0.2333	-0.2075	-3.0744	-3.2303	-7.1339
LUMO	1 8/61	0 7561	1 1 6 0 6	0.8300	1 0057	0.0456
(DCM)	-1.0401	-0.7501	-1.1090	-0.0399	-1.9037	0.0450
Energy						
gap	3.081	5.479	5.098	5.035	3.345	7.202
(DCM)						

 Table S5 Calculation of Electronic Properties in DFT and DCM Solvent

To bring clarity, we have calculated the S_0 state HOMO-LUMO energy gap using the different functions enlisted above.

Table S6 Optimized Ground State (S ₀) Geometry	y in DCM solvent
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CAM-B3I VP/6-31G*	PBF0/6-31G*
	C 1.096/5809495630 3.940/6/589/3123 0.291/4089804622
C 2.495500/0384042 3.83698551155463 0.4/306915401305	C 2.51291162068535 3.88593439739794 0.34858260806975
C 3.14904229923654 2.6/54009630/094 0.19//508415622/	C 3.16306268548559 2.70543882965744 0.11611257177854
C 2.44190210324303 1.481112/8348003 -0.15053914231180	C 2.45048908250955 1.50850062508112 -0.20244795149290
C 1.008/2589136294 1.52924306062491 -0.20981696836379	C 1.01294/000088/9 1.50331464105161 -0.26148099643988
C 0.301209234208/4 2.70829400329378 0.10308007903833	C 0.3/322/840804/8 2.81322943410105 0.0080/82995/909
C 3.11892637761039 0.27478190422099 -0.42326876480818	C 3.12348640985442 0.28589168281/92 -0.4193/2503/1/91
C = 2.57885599020110 = -0.87000009049140 = -0.75547808449077	C 2.5/528950815091 -0.8/41//91292050 -0./1582295025850
C 0.94501459881026 -0.82905571249970 -0.82218521425591	C 0.95588869441062 -0.82140/94405248 -0.79094590099058
0.20555250451005 0.5/5/1880/50948 -0.5515005555502/	C = 0.25955990555077 = 0.39045448982517 = 0.55718972055550
C 3.05575395390005 -2.11452750185089 -1.05508572077117	C 5.02/3646110/405 -2.11/2/053904031 -0.96036649536119
C 2.353392/8003498 -5.22490384291232 -1.413/00/9090430	C 2.32149505800959 -5.2582910814118/ -1.52524222885219
C = 0.91764694906505 - 5.17064747905652 - 1.51454651107074 C = 0.25222207126728 - 2.01605981072710 - 1.22268018726198	C = 0.90838394909029 = -5.1/90403/94390/ = -1.434895354929189
C 0.2522229/150/56 -2.010956810/2/10 -1.25206016/50166	C 0.24429/41500029 -2.010455854995/5 -1.18194650491111
C = 1.21000952855554 = 0.47171008454059 = 0.58200508400040 C = 2.05668882067227 = 0.20400572456424 = 0.10621256585420	C = 1.20/151051011/5 = 0.49/55405195551 = 0.5/6445/91/6990
C -2.03008883007357 -0.30400373430424 0.10021330383420	C -2.000/4815/24540 -0.50/52509565019 0.06558645151609
C -5.52404122//500/ -0.228/0990208048 0.08812/09452580	C -3.52090584551502 -0.22901920490526 0.07189911520705
C = 4.59792950498018 = 0.20020702182805 = -0.54205848759182	C = 4.39003903709347 = 0.21093042790740 = -0.32717872330000
C = 5.38130707433010 = 0.33941370307822 = 0.70313833317034	C = 5.57207571407578 = 0.43870017702575 = 0.09072501504202
C = 0.74115795990797 = 0.22601549509156 = 0.55506019534544	C 0.75647794515954 -0.24408095219780 0.50098759591202
(-5,40099705674945,-0,60079006252592,-1,24759261062616,-2616)	(0.75170555582540 - 0.48555775707054 - 0.775757577120847
C = 5.49988793074843 = 0.09978000333382 = 1.24738301903010 C = 5.06164170426112 = 1.06275005647391 = 1.90910479096590	C = 5.0957606609022 = 0.75201905995750 = 1.20246474519021 C = 5.047440000775500 = 1.15500871200912 = 1.95026072064600
C = 6.08758004001065 = 1.52427578250152 = 2.60827220751746	C = 6.07242925420214 = 1.65212692770107 = 2.64925241025495
C = 7.42500166040420 = 1.25288470727124 = 2.2262857327731740	C = 7.42321160015771 = 1.4020266407882 = 2.20224570610782
C = 7.78147440171555 = 0.70122086291787 = 1.16165058200050	C = 7.77587616850262 = 0.74261766042562 = 1.16022071262772
C = 7.00282077677805 = 0.00125080581787 = 1.10105756279557	C = 7.00020440008951 = 0.74301700743303 = 1.10023771302773
$C = \frac{1.55582077077855}{0.00578110545547} = \frac{1.50055740500085}{0.0057405000085}$	C = 1.77729440908851 = 0.87878117508254 = -1.48584004820007
C = 5.63054093320171 = 0.60924922710488 = 0.77601471357566	C = 5.63454034517160 = 0.64050020651605 = 0.74608737358000
C = 6.354673527171 = 0.00724722710488 = 0.77071471557500	C = 6.35781714367768 = 0.15184600602216 = 0.15155407310008
C = 5.63879217698574 = 0.95651392452219 = 1.03953039938692	C = 5.6357877876866443 = 1.00275941632865 = 0.99844995976041
C = 4.25307205272181 = 0.99401992468059 = 1.00712762174553	C -4 24962978409465 -1 03848995906042 0 95880153535980
H 0 55757777715496 4 81087715830113 0 65676500171717	H 0 58366395313587 4 87795479868118 0 49038223669266
H 3 05237932146487 4 72816777770488 0 74149178042555	H 3 07886622221516 4 78116691991673 0 59203617372346
H 4 23024860396985 2 63482246186236 0 24791024987191	H 4 24505016294826 2 65738329478869 0 17885216921260
H -0.72154710021132 2.81018573083889 0.09737207270248	H -0.70865005523285 2.86484110902271 0.00394959033906
H 4 11702774519867 -2 14758943568299 -1 00052547666944	H 4 11034631564260 -2 15163238688717 -0 93110228086562
H 2.85675363994718 -4.14974926617911 -1.64059576884985	H 2.84216756650519 -4.16977142610403 -1.52885923772714
H 0 36709636237017 -4 05182614877441 -1 82937348768184	H 0 35315861597625 -4 06226272675012 -1 73945754486917
H -0.82434116780475 -1.98135239052495 -1.33402156883678	H -0.83222811064481 -1.97121758176665 -1.30071191003321
H -1.61043306743673 1.26791180304675 -1.20712953660168	H -1.61096327510409 1.32242919535697 -1.16448604802990
H -1.64643317594969 -1.06355880770309 0.76934192808829	H -1.64887112717733 -1.08926931735189 0.72121062637664
H 5.32050061190783 1.22347262113492 -2.17722306532673	H 5.33725706543588 1.31541831140455 -2.11105921112200
H 4.02368090191730 -1.20578449215382 2.18493998298803	H 4.00748886737632 -1.31659146796556 2.13219207366437
H 5.85373480682168 -2.05122754827525 3.62411633186790	Н 5.83497019265312 -2.21046669544770 3.55040549600823
H 8.21878959213242 -1.73434987880844 2.98156361640066	H 8.20498887338734 -1.85174892748332 2.94171975360326
H 8.82080530786655 -0.56248636689111 0.88128421424003	H 8.81780299367327 -0.58564973737287 0.89528720982183
H 8.62572440923082 1.46128901322413 -0.95419108735056	H 8.62254601130210 1.50695421462112 -0.83872257212265
Н 7.71520990905045 1.33279491947581 -2.47359959244135	H 7.72835691957044 1.44800762534838 -2.37459898074541
Н -6.18349768812007 -1.55637140368816 1.75906703838410	H -6.17861513463664 -1.63629549488360 1.69250808130007

Н -3.71861243040447 -1.62576632654438 1.71144296259104	H -3.71186094106091 -1.70266823687944 1.63205840557901
H -6.15124953778968 1.22231542137825 -1.50493739969233	H -6.15760427929118 1.29511841661688 -1.44576077808487
H -3.72294791350051 1.15470603810043 -1.55936238190652	H -3.73039028006898 1.23164838499788 -1.51285950298157
C -7.83958339738065 -0.11444296345949 0.20465296938715	C -7.83748454781612 -0.12394152912688 0.21272322823735
C -8.57072137035129 0.95847607804949 -0.31588254656841	C -8.57824148323229 0.95959226715721 -0.28041618418451
C -9.77628884290251 -1.16502940266227 0.86327160256431	C -9.76526864112198 -1.19749193071997 0.86180203004511
C -9.95610854031338 0.93621401491102 -0.23974648994440	C -9.96470484256108 0.92852115361060 -0.19614865620727
H -8.06431827803136 1.80826768144423 -0.75765458287328	H -8.07964561806118 1.82396631069652 -0.70619367732917
C -10.58117872948028 -0.14945380152419 0.36098894529248	C -10.58153003238311 -0.17492394560018 0.38527746193209
H -10.22402461517712 -2.03075469895104 1.34667337464271	H -10.20597670834142 -2.07648011003563 1.33045140941495
H -10.53793750212881 1.76192332064174 -0.63741505080904	H -10.55357473798964 1.76088219223351 -0.57201779009646
H -11.66051244945738 -0.20979943904656 0.44575084537476	H -11.66115058569300 -0.24308233591921 0.47493337312015
N -8.44620113032407 -1.15814521349933 0.79092376490745	N -8.43564643767042 -1.18441898265456 0.78228204989546
H 8.56675056927409 -0.08690250097470 -1.82880688392331	H 8.57821374774403 0.00003714071522 -1.78746101853338

Table S7 Optimized Ground State (S_0) Geometry in DMF solvent

CAM-B3LYP/6-31G*	PBE0/6-31G*
C 1.07003888627897 3.887060720/2150 0.41052875475638	C 1 00648013438470 3 04344107087403 0 27631151842883
C = 2.49090605431695 = 3.84106280373719 = 0.46170443928103	C = 2.51284677274510 = 3.88881574701157 = 0.33438488400220
C 3 14725037065322 2 67939533127090 0 18840429016711	C 3 16310078860084 2 70767660866717 0 10536503736101
C = 2.44116580656204 = 1.48450555249168 = 0.16630480612701	C = 2.45064254119061 = 1.50962917513585 = 0.21054750577161
C 1.00700120728462 1.53212008830476 -0.22175573258155	C 1.01300537770012 1.56417817130300 -0.27056647518030
C = 0.25050704750785 = 2.77127840612271 = 0.08880200061126	C = 0.27500040025242 = 2.81516714477546 = 0.00402271266822
C = 3.11921194504257 = 0.27812062258142 = 0.43095203531994	C = 3.12412305988216 = 0.28647915523500 = 0.42408423893409
C = 2.27081887575005 = 0.87214674846087 = 0.762777202052557777	C = 2.7506102702601 = 0.87202257686176 = 0.72012504206066
C = 2.57561887575705 = -0.87514074840087 = -0.70577720505225	C 0.03660605101250 -0.82158502580660 -0.72558131480507
C = 0.26517428750075 = 0.27504187865665 = 0.54285220040566	C = 0.750000005171257 = 0.82150502500007 = 0.77550151405507
C = 0.20517458759075 = 0.57594187805005 = 0.54285550049500	C = 0.23503737390913 = 0.35034730247470 = 0.34323507817875 C = 2.02858212261544 = 2.11727602404215 = 0.00020505080416
C = 3.05700045052571 = 2.11007501440226 = 1.00772120542750	C = 3.02858212201544 - 2.11727032404515 - 0.3302053580410 C = 2.22255400061206 - 2.22878274040541 - 1.22662764472208
C = 2.53627500555071 = 5.22045555005151 = 1.42675585772557	C = 2.52525477001570 = -5.25878274047541 = -1.52005704475208
C = 0.254070505128912 = -5.10700072552207 = -1.55018499059201 C = 0.25407402457240 = 2.01401022645581 = 1.24682757004026	C = 0.205768770082107 = -5.18001554117881 = -1.45855205807870 C = 0.24547605708462 = 2.01145252840255 = 1.18580042127720
C = 1.21041740612565 = 0.47100170612207 = 0.50100784001114	C = 1.20714999677004 = 0.40655572950615 = 0.59219040410560
C = 1.21041/49012505 = 0.471991/0015527 = 0.59199784291114 C = 2.05250026227208 = 0.20664792121202 = 0.007020852687222	C = 1.20/1488807/004 = 0.49033375859013 = 0.38218949419300
C = 2.03339030237278 = 0.30004782121203 = 0.097793783308722 C = 2.52160220086020 = 0.22252100168547 = 0.08278027020814	C = 2.03900292493404 = 0.30934221134744 = 0.07982303038283
C = 5.52109220080039 = 0.25532199108347 = 0.08578927950814 C = 4.50770005601042 = 0.20862140152145 = 0.24416470462557	C = 5.51785505455778 = 0.25051546250205 = 0.07105505574255
C = 4.597/9005091942 = 0.20805140155145 = -0.34410470402557	
C 5.5/44494/000482 -0.402839313/3823 0.70330438022299	C 5.508920500008099 -0.44018852170590 0.09800758484550
U 0./303/355/48032 -0.2305401452915/ 0.30490000/04038	C 6./303195100/830 -0.24458050023185 0.3/2/9951/02525
N 0./91809/8481590 0.449/0580525290 -0.829/4005555/441	N 0.79571050555001 0.48420052084089 -0.79040059818990
C 5.505/6/3/804930 0./0620/21899820 -1.24109626644151	C 5.3131490839/104 0.73331211123933 -1.19948232838341
	C 6.0642501//3/83/ -1.65660622919143 2.650//841116130
C /.41910045556080 -1.36504920244941 2.34140386379104	C /.41592666/66383 -1.4518010/330196 2.30941600985412
C /.//188199204358 -0./0/5102084//2/ 1.1/521210/68046	C /.//14939/420941 -0./44/625285522/ 1.169090/1284635
C -4.24630555336725 0.54779934270085 -0.82618000646267	C -4.24945114285674 0.61320797580336 -0.78425294634858
C -5.63106/19219553 0.5890090563/185 -0./8/62199913041	C -5.634/4954/416/2 0.6519/253/63/40 -0./4033896502680
C -6.35082281300431 -0.15221573982564 0.15771725814985	C -6.35684601855955 -0.15308209375534 0.15514083655789
C -5.631/0/0253/6/1 -0.94581404510049 1.05555531611826	C -5.63343681143554 -1.00689809430874 0.99825069237270
C -4.24600176528502 -0.98378201142249 1.01831604186811	C -4.24/20116063883 -1.042832/3691432 0.95645130495451
H 0.55439424136316 4.81474491905235 0.64041776067154	H 0.58318770963310 4.88129907730332 0.47127024470717
H 3.04902980816/42 4./32/65858353/2 0./2999802851460	H 3.0/868492/92648 4./8490305942610 0.5/501234586084
H 4.22842/30/3/112 2.6396456186324/ 0.24000350813036	H 4.24519303780460 2.66034178631688 0.16761425275906
H -0./23190/9584309 2.813/9039962345 0.0805626336264/	H -0./08/512/866802 2.86/96024601234 -0.01139536441943
H 4.11889397173037 -2.14274575148090 -1.01237176749085	H 4.111625306/9082 -2.1511/912866819 -0.93514121194934
H 2.8603/48684/1/8 -4.14441146305238 -1.65/48159162264	H 2.84429343228563 -4.17012508390662 -1.53209529882522
H 0.3/0/5634953338 -4.04//14005/1202 -1.84/44/20/84024	H 0.35485/14161805 -4.063425426420/6 -1./42//2048/1142
H -0.82225095990221 -1.9/929490588//3 -1.54889283142183	H -0.831105028//960 -1.9/331825283001 -1.30463545801095
H -1.6142/402/80504 1.26851/3485656/ -1.213815/803120/	H -1.61250903284307 1.32251968727916 -1.16591014533888
H -1.64009155941/90 -1.065/42143/5891 0./596244/231055	H -1.04/10849253345 -1.09285609655651 0./15429/9500//5
H 5.33336779689950 1.23468239117876 -2.16939964025732	H 5.34483341444845 1.31836/52412211 -2.10/82930086837
H 4.00//84481/5250 -1.21559082001159 2.1/590448502920	H 5.99983958088539 -1.32014490854783 2.12844084580284
H 3.828/330/039/90 -2.00/438/39/2104 3.62026832989937	H 3.8229/130030084 -2.21303430081412 3.55100018964441
H 8.19/98559/1/928 -1./4845455/64894 2.99546/61054415	H 8.195246898853549 -1.85474434794040 2.95088061666195
H 8.81283481932348 -0.30/0/44699/092 0.901492495038/8	H 8.81419340331/20 -0.38003343203//5 0.90/00846336035
П 6.02/30015515/00 1.40839809015/02 -0.92611955820468 П 7.73815470373106 1.24003874507030 2.45254871008500	п 6.02570787542995 1.50970009408456 -0.82432907682836
H /./28154/05/2106 1.540958/450/050 -2.452548/1998500	H /./3533491552830 1.4508205382/1/3 -2.303/80/309405/
H -0.1/252/55825882 -1.55500800400451 1.78800010800551	H -0.1/441184230820 -1.042330/18040/4 1.09192849244032
H -3./0842030005823 -1.6033581269//38 1./3103302622603	H -3./0845846/58035 -1./0942998920325 1.6266129/6248/5
H -0.15015045855402 1.188/035390581/ -1.5250919/810504	H -0.138/923230313/ 1.300830393383/3 -1.43023131341116
H -5./20090/08215/1 1.1199592152825/ -1.58/45444/85515	H -5./5102884590191 1.25/5/5/125/580 -1.50/2//01915/84
(-7.65362257705717 - 0.11719109892970 - 0.21453855895370	C - 7.65070215944198 - 0.12521725552215 0.21752505504087
C 0.77285020128056 1.16280570207707 0.800000707070510	C -0.57004903875200 0.90384501453935 -0.20455500718110
C -9.//203030120930 -1.102803/920//0/ 0.8800000/9/02319	C -9.70373464303327 -1.20370240231087 0.83002827009908
U -7.7310/443003333 U.73/8384202933/ -U.224948489/20/3	U -7.70524404042145 U.75554152080820 -0.1/9/6459485402
П -0.03/40400/94234 1.80433291921099 -0./48/21//300489 С 10.57665430784300 0.14504699540306 0.27991863676997	П -6.0//44475045001 1.831/8009/8/000 -0.0825/05/8/0820 С 10.59005247577475 0.17520101611126 0.20006624181677
U -10.37003420764200 -0.14394066349290 U.57881803076887 U -10.22160220142112 -2.026776444555510 -1.26561270000021	U -10.3007334/3//4/3 -0.1/330101011120 U.390900241810//
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	п -10.2073013023971 -2.08080928780893 1.31030328712376 П 10.55173001071721 1.77000033887762 0.57703115067003
11 -10.35250005100090 1.70451640540558 -0.0217577750055 11 -11 65595962121295 0.20406779769006 0.46650291614115	11 -10.331420010/1/31 1.//000022664402 -0.34/03113904093
$ \begin{array}{c} \Pi & -11.05363602121383 \\ NI & 8.44257450240720 \\ NI & 8.44257450240720 \\ \end{array} $	п -11.00000132613220 -0.243280/3301443 0.48042023005010 N 9.42560596519912 1.10121591529567 0.77620565409920
H 8 57785617337329 _0 07834200609166 _1 80386746266520	H 8 58453677438472 0 00328156461087 -1 77397021915104

Compound	Solvent	ET ₃₀	$\lambda_{abs.max}$ (nm)	λ _{em.max} (nm)	Stokes shift (cm ⁻¹)	$\Phi_{\rm f}$ (%)
IAPY	Hexane	31.0	404	515	90090.09	20.22
with $\lambda_{em.max}$ (Solid) - 517 nm	1,4- dioxane	36.0	409	518	91743.119	17.73
- 317 IIII	EtOAc	38.1	407	522	86956.521	14.54
	CHCl ₃	39.1	411	521	90909.09	13.33
	DCM	40.7	412	514	98039.216	12.79
	N,N- DMF	43.2	413	526	88495.575	9.79

Table S8 DSE properties of IAPY



Fig. S24 Geometry comparison between Ground state (S_0) geometry and S_1 -state optimized geometry IAPY in DMF using CAM-B3LYP/6-31G* level of theory.



Fig. S25 Dihedral scanning for emission oscillator strength at different twisted sites of IAPY



Fig. S26 Decay profiles of IAPY in different solvents



Fig. S27 LE state emissions of IAPY in different solvents



CIE chromaticiy diagram 1931

Fig. S28 CIE-coordinates of whitish emission of IAPY



Fig. S29a Absorbances for whitish emission for IAPY in the binary mixture of 1,4-dioxane and CHCl_{3.}



Fig. S29b Whitish emission for IAPY in the binary mixture of 1,4-dioxane and CHCl₃. (The percentage of CHCl₃ has been increased gradually, and the excitation wavelength was in the range of 409-416 nm). The CIE coordinates for 0% to 99% are as: 0% (0.27, 0.52), 10% (0.27, 0.52), 20% (0.28, 0.58), 30% (0.29, 0.52), 40% (0.30, 0.52), 50% (0.32, 0.51), 60% (0.32, 0.50), 70% (0.34, 0.50), 80% (0.39, 0.46), 90% (0.34, 0.51), 99% (0.31, 0.52).



Fig. S30 Whitish emission for **IAPY** in the binary mixture of EtOAc and $CHCl_3$. (The percentage of $CHCl_3$ has been increased gradually, and the excitation wavelength was in the range of 411-418 nm).

Compound/Sol	χ2	λ _{em}	τ_1	τ_2	τ3	τ	α1	α2	α3	α	τ _{avg.}	$\Phi_{\rm f}$	k _r (s ⁻¹)	k _{nr} (s ⁻¹) x
vent		max				4				4	(ns)	(%)	x 10 ⁻⁶	10-6
		(n												
		m)												
IAPY	1.01	517	0.606	2.960	-	-	0.903	0.096	-	-	0.834	26.6	319.27	879.194
Pristine	2		8	7			2	7			4	4	1	
IAPY	1.06	534	0.498	2.110		-	0.907	0.092		-	0.646	53.3	824.49	721.818
Ground	7		3	0			9	0			7	2	3	
IAPY H	1.00	530	0.600	1.410	3.44	-	0.626	0.318	0.053	-	1.023	1.16	11.331	965.517
	0		4	0	9		9	9	1		7			
IAPY	1.05	539	0.518	2.101			0.909	0.090			0.580	5.89	101.44	1620.909
base	8		6	6			8	8			6		7	
fumed														
IAPY	1.03	515	0.055	0.596			0.490	0.509			0.055	20.2	3636.6	14348.92
Hexane	7		9	2			7	3			6	2	9	
IAPY 1,4-	1.06	518	0.064	0.063			0.484	0.515			0.064	17.7	2770.3	12854.68
dioxane	9		5	4			2	7				3	12	8
IAPY	1.02	522	0.062	0.062			0.494	0.505			0.063	14.5	2307.9	13565.07
EtOAc	3		9	5			9	1				4	36	9
I A DX/	1.07	514	2 1 7 7	0.064			0.001	0.000			0.070	12.2	10(0.2	10745 59
IAPY	1.07	514	3.177	0.064			0.001	0.998			0.068	13.3	1960.2	12/45.58
DCM	1						2	8				3	94	8
IADV	1.00	521	2 0 4 2	0.075			0.001	0.009			0.070	127	1619.0	11030 24
	1.00	521	2.945	0.075			0.001	0.998			0.079	12.7	1010.9	11039.24
CHCl3	0		3	9				Ŏ				9	0/	
IAPV	1.01	52.6	0.071	5.098			0 999	0.000			0.074	9 79	1322.9	12190 54
N.N-DMF	9	520	9	8			5	5			0.074		72	05
			Í										·-	
	1	1	1	1	1	1	1	1	1	1	1	1	1	1

Table S9 Results of times-resolved studies, altogether



Fig. S31 Absorbances for AIE-study of IAPY



Fig. S32 Intensity of emission increment for AIE at 10⁻⁵M concentration of IAPY



Fig. S33 TD-DFT optimized structure of IAPY in water



The rotation barrier is very small ~6 kcal/mol. That's free rotation takes place and excited state energy can decay non radiatively.

Fig. S34a Rotation barrier of the Pyridine group in S_1 state



Fig. S34b TD-DFT optimized structure of IAPY in MeOH



Fig. S35 Absorbances for viscofluorochromism of IAPY



Fig. S36 Intensity of emission increment for VIE at 10⁻⁵M concentration of IAPY



Fig. S37 Cellular uptake of the probe IAPY through Fluorescence Activated Cell Sorting (FACS)



Fig. S38 Bright field images for wash-free bioimaging of FaDu with IAPY (10 μ M)

Cell Culture and confocal microscopy

The cell line was bought from ATCC and preserved in DMEM supplemented with 10% FBS with 1% (v/v) penicillin-streptomycin antibiotics. Cells were incubated at 37° C in a humidified atmosphere comprising 5% CO₂. All the culture supplies were procured from Thermo Fisher Scientific.

Cell line was seeded on a lysine-coated sterile coverslip placed on a 12-well cell culture plate (1×10^5 cells per well) in DMEM containing 10% FBS with 1% antibiotics. Cells were allowed to adhere completely. After 24 h, cells were treated with formulation using 10µM probe concentration for 4 h. Later, cells were not washed with sterile PBS and mounted on a glass slide. Cell imaging was done using a confocal microscope (Leica LAS X).

FACS

FaDu cells were seeded in a 12-well plate at a density of 0.2 million cells per well. The cells were treated with IAPY at 10 μ M concentration while the unstained control cells were treated with DMSO. After 24 hours of treatment, the medium was removed and cells were washed with PBS and trypsinized to obtain a single cell suspension. The cells in trypsin were centrifuged at 2000 rpm for 10 minutes at 4 degrees and resuspended in PBS for uptake analysis using BD FACS Aria II flow cytometer.

Device fabrication, formulation with dye for screen printing and the procedure for screen printing

A market-available screen-printing liquid (SPL) white non-fluorescent dye of 4000 cP was purchased. Later 30 mg of IAPY dye was dissolved in 10 ml of DMSO and sonicated unless the solution becomes clear. Now this solution was added with 30 ml of the screen-printing liquid white non-fluorescent dye and mixed properly to prepare an **IAPY-SPL** media. Again, by using another dye **SB6** the **SB6-SPL** media was prepared in the same fashion. The QR code was drawn with CAD tool solid works, and on the PVC sheet the QR code was engraved with the universal laser system CO₂ laser with 10-micron wavelength with 10% into 30 watts. The **SB6-SPL** was first painted on a glass surface and dried subsequently. The surface emiited green under 365-UV lamp. Later on, the laser-cut QR code pattern was screen printed with **IAPY-SPL** on the surface of **SB6-SPL**. With naked eyes the QR-code is not clearly visible under ambient light. Even under 365-UV lamp also the entire surface remains green. Once the acid fumigation is done the QR-code will be visible under 365-UV lamp as it glows yellow on a green surface.

Procedure and method of data storage and anticounterfeiting applications with stimuliresponsive hidden stamp:

Three dyes- IAPY, ATh4P, and SB6 were initially chosen as all of them are green emitters ($\lambda_{em. max}$ = 501 nm-517 nm). An ethanolic solution of 10^{-7} M concentration was prepared of each dye and soaked with three different sponges. Now a wooden stamp was pressed on the IAPY solution-soaked sponge and placed on a TLC plate to get an IAPY led stamp. Later on, the wooden stamp was washed with acetone properly, and consecutively, two more **ATh4P** and **SB6** led stamps were prepared on the TLC plates. Now, none of the stamps are visible to the naked eye but under the UV-365 nm lamp, all the stamps glow green. Now how to identify which stamp is real and which ones are counterfeited? Let us say that the IAPY is the real dye; hence, the IAPY led stamp will be the real stamp. IAPY emits orange-red on acid-fumigation but immediately does not revert back to green on base-fumigation as it turn yellow. But this information is secret. The counterfeiters might mimic a dye (SB6) which is a green emitter but not acidofluorochromic or can mimic a dye (ATh4P) which can emit red on acid fuming and again revert back to green on base fuming. Because reversible acidofluorochromism is usually used in most of the cases. So, if an unknown stamp that looks alike the real one, is acid-fumed and still emits green; it means the dye is counterfeited and the stamp is unreal. Next, if an unknown stamp that looks alike again as the real one, is acid-fumed and turns red emitting then that stamp may or may not be real. But again, with base-fuming if it turns green emitting instantly, then the stamp (made of dye ATh4P) is fake with no doubt. But on further base-fuming if it turns yellowish then the stamp is the real stamp.



Fig. S39 Data encryption with IAPY



Fig. S40 Data encryption with IAPY to make different digits from '8'



Fig. S41 Xanthan gum sensing (a) absorbances (b) emission (c) picture taken under UV-365 nm lamp (the mg of Xanthan gum taken per 2ml of water is written on each cuvette picture)



Fig. S42 Na-CMC sensing (a) absorbances (b) emission (c) picture taken under UV-365 nm lamp (the mg of Na-CMC taken per 2ml of water is written on each cuvette picture)



Fig. S43 False-positive taste under UV-365 nm lamp; (a) dried patches of cold drinks (b) almost no fluorescence after staining with **IAPY** (c) tomato sauce (d) feeble green fluorescence after staining with **IAPY**.



Fig. S44b Blood serum-NMR comparison (the maroon color line for **normal serum** and the sea-green color line for (**IAPY+serum**). The peaks for (**IAPY+serum**) are de-shielded from that of **normal serum**.



Fig. S45 Hospital-disposed of biohazards detection by fluorescence; under normal room light (a) blood-smeared cotton (c) blood containing glove; under UV-365 nm lamp (b) blood-smeared cotton (d) blood containing glove



Fig. S46 ¹H NMR of IAPY¹



Fig. S48 ¹³C NMR of IAPY¹







Fig. S49b ¹³C NMR (aromatic part zoomed) of IAPYH

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