

### Supplementary information

#### Effect of dendrimer generation and surface groups on the optoelectronic properties of green emitting bis-tridentate iridium(III) complexes designed for OLEDs

Vaidehi Pandit, Junhyuk Jang, Manikandan Koodalingam, Chandana Sampath Kumara Ranasinghe, Mile Gao, Paul L. Burn\*, Emma V. Puttock

Centre for Organic Photonics & Electronics (COPE), School of Chemistry & Molecular Biosciences, University of Queensland, St. Lucia, QLD, 4072, Australia

\*p.burn2@uq.edu.au

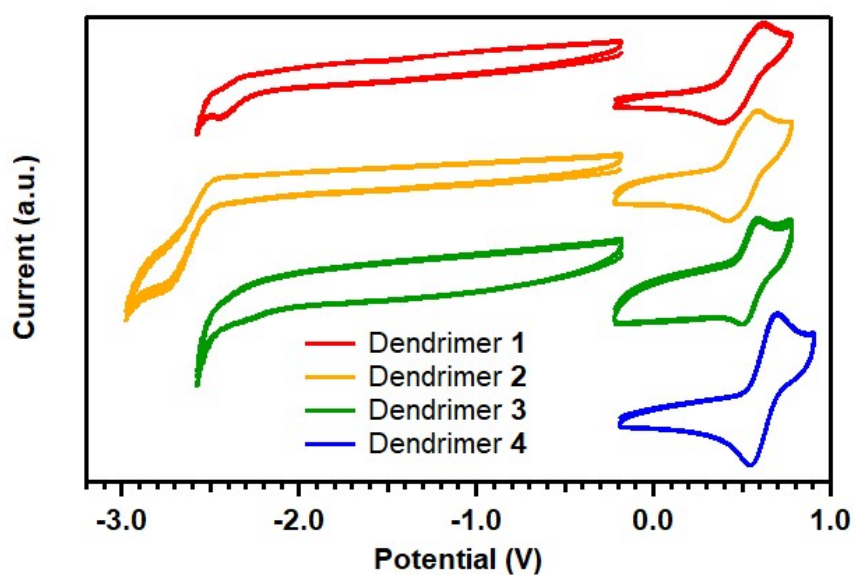


Fig. S1. Cyclic Voltammograms of dendrimers 1, 2, 3 and 4 in dichloromethane (oxidations) and tetrahydrofuran (reductions) in the presence of 0.1 M tetra-*n*-butylammonium perchlorate at room temperature.

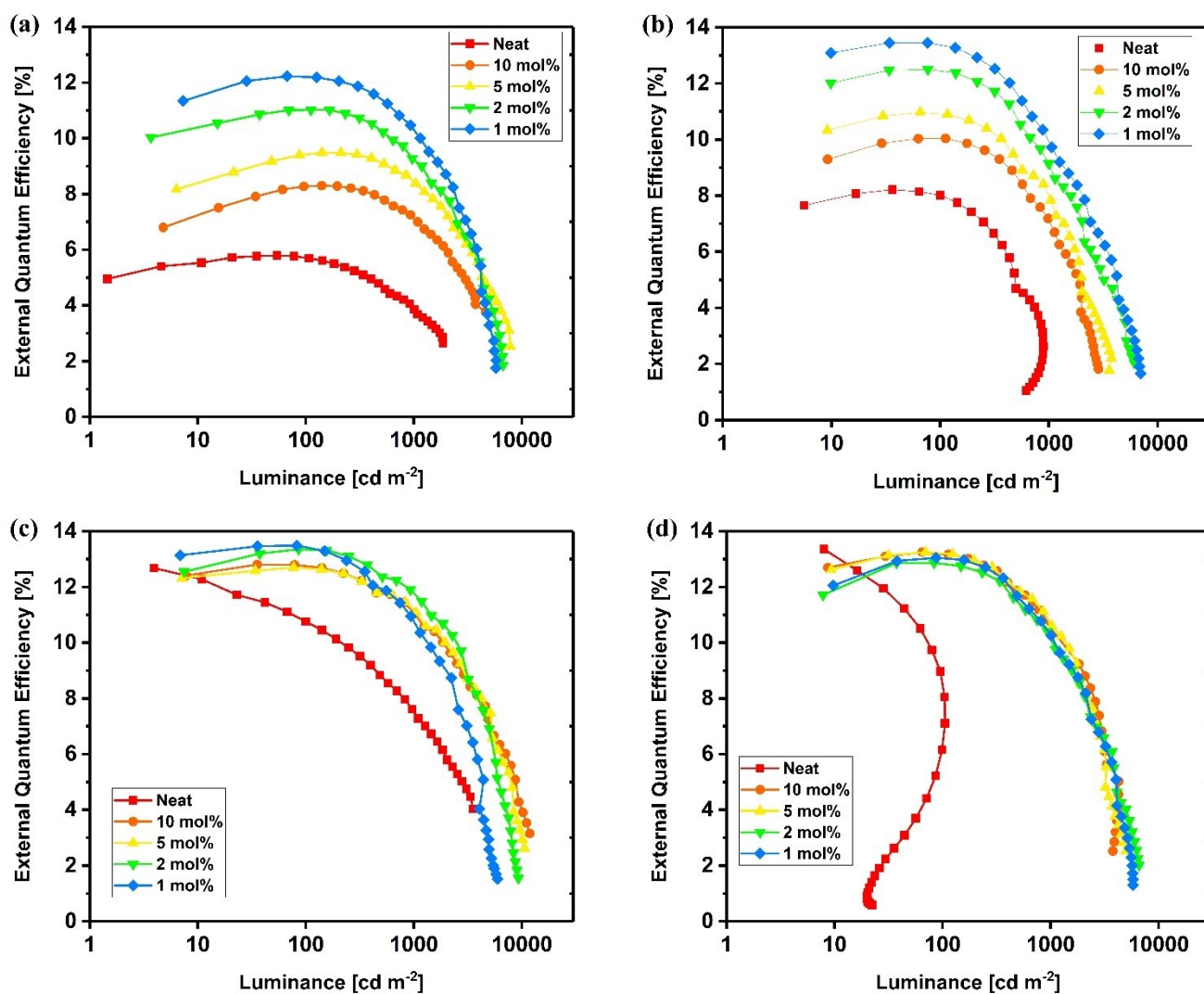


Fig. S2. EQE versus Luminance with various blend concentrations for devices incorporating dendrimer 1 (a), 2 (b), 3 (c) and 4 (d).

Table S1. Photophysical properties of dendrimers 1, 2, 3 and 4 films (2 mol% in TCTA) at  $298 \pm 3$  K

Sample	TCTA PL (%)	TCTA PLQY	Emitter PL (%)	Emitter PLQY
Dendrimer 1	8	4.6	92	58.0
Dendrimer 2	8	4.6	92	54.2
Dendrimer 3	26	11.9	74	33.4
Dendrimer 4	10	5.5	90	49.3

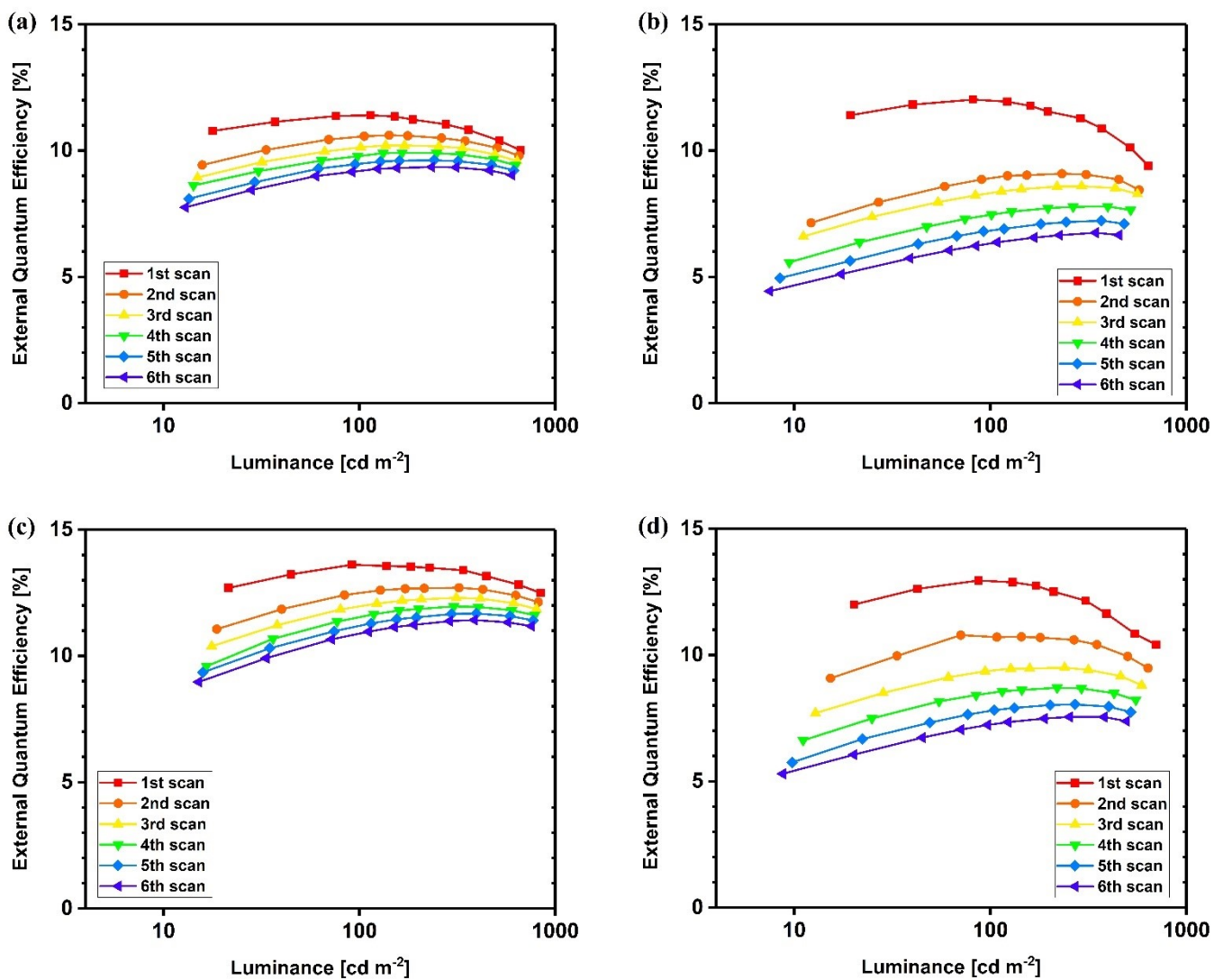


Fig. S3. Changing EQE vs. Luminance curves as a function of electrical stress of (a) Dendrimer 1 (b) Dendrimer 2 (c) Dendrimer 3 (d) Dendrimer 4 blended in TCTA 2 mol%.

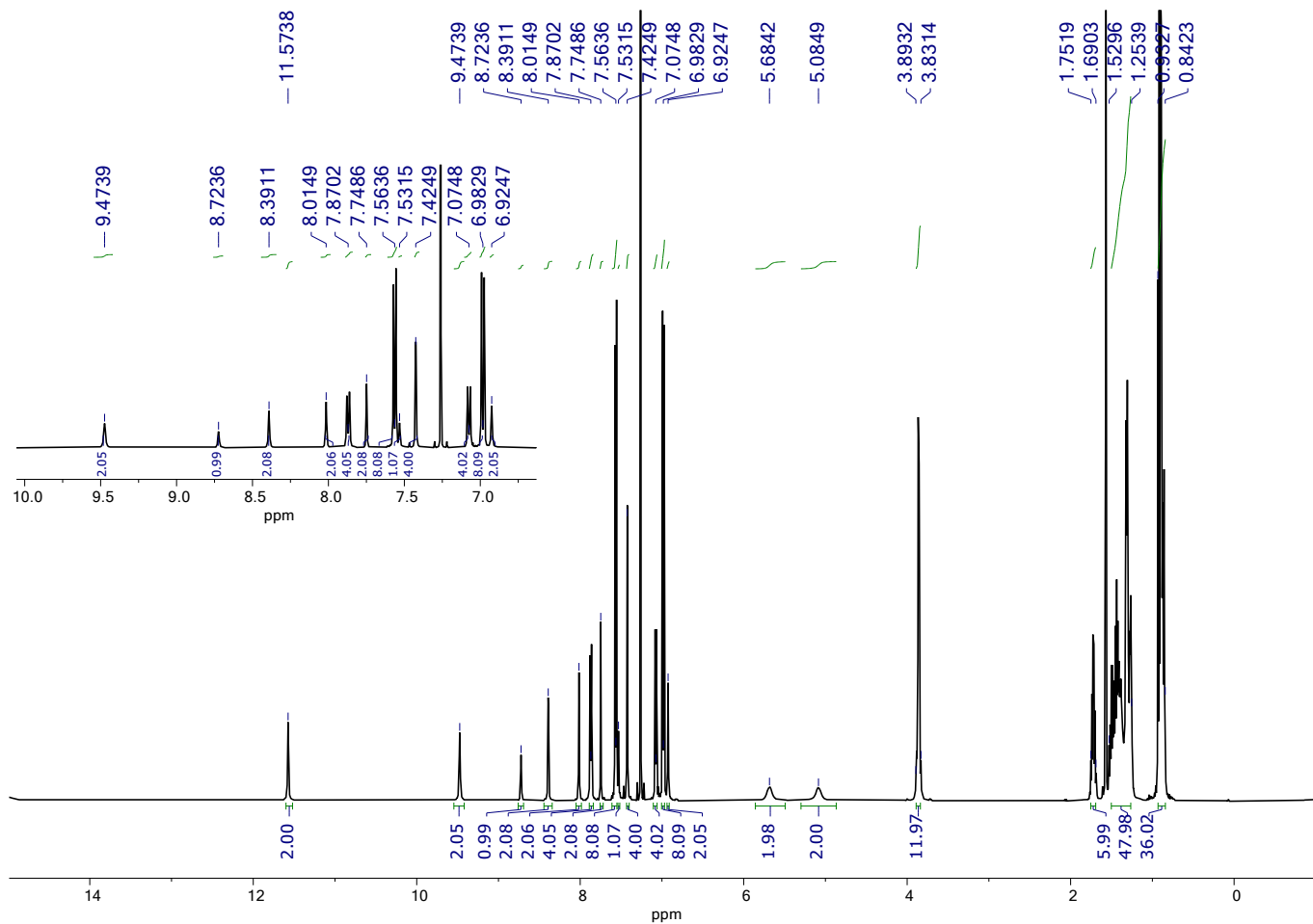


Fig S4.  $^1\text{H}$  NMR of Compound 4 in  $\text{CDCl}_3$ .

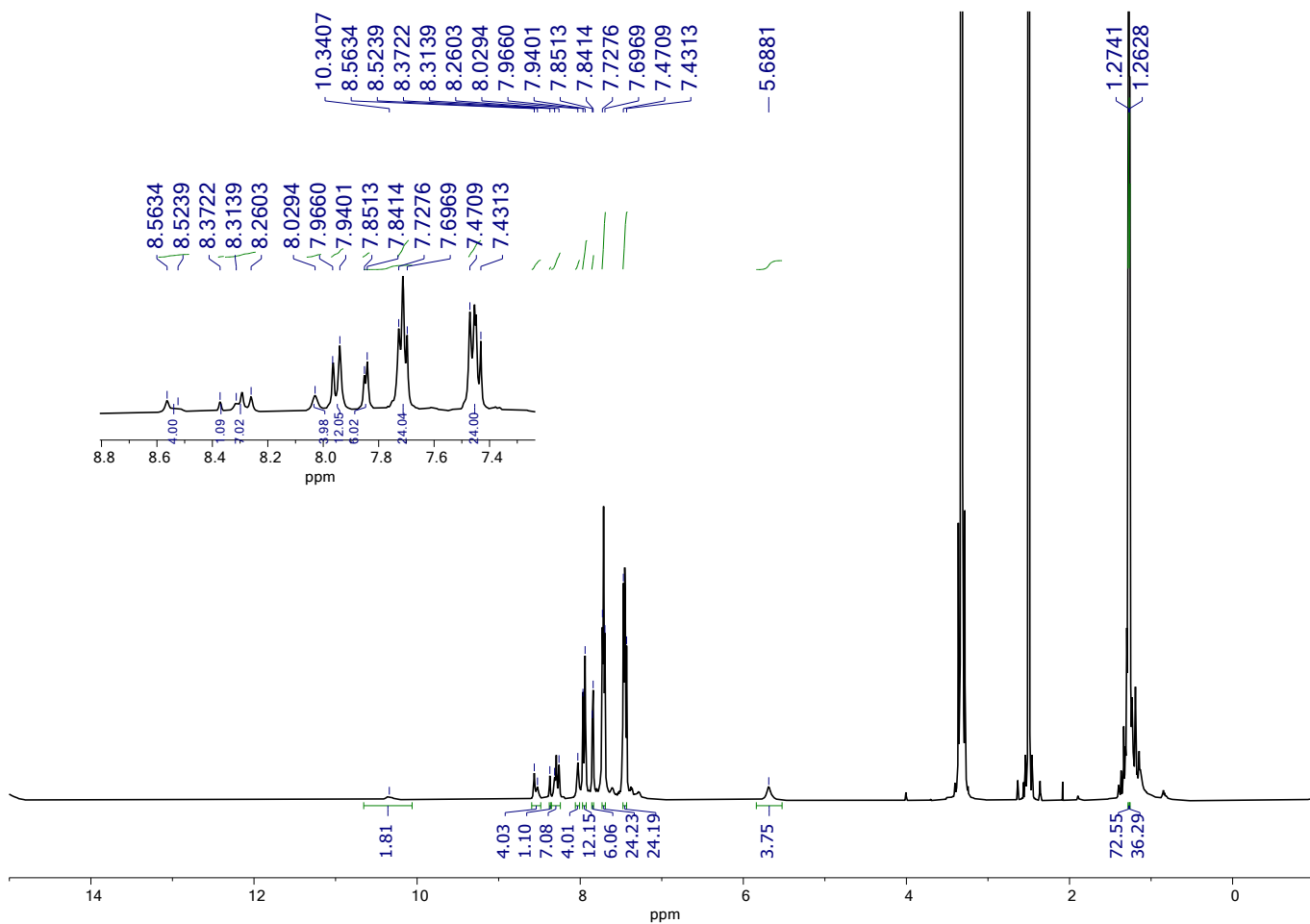


Fig. S5.  $^1\text{H}$  NMR of Compound 15 in  $\text{DMSO}-d_6$ .

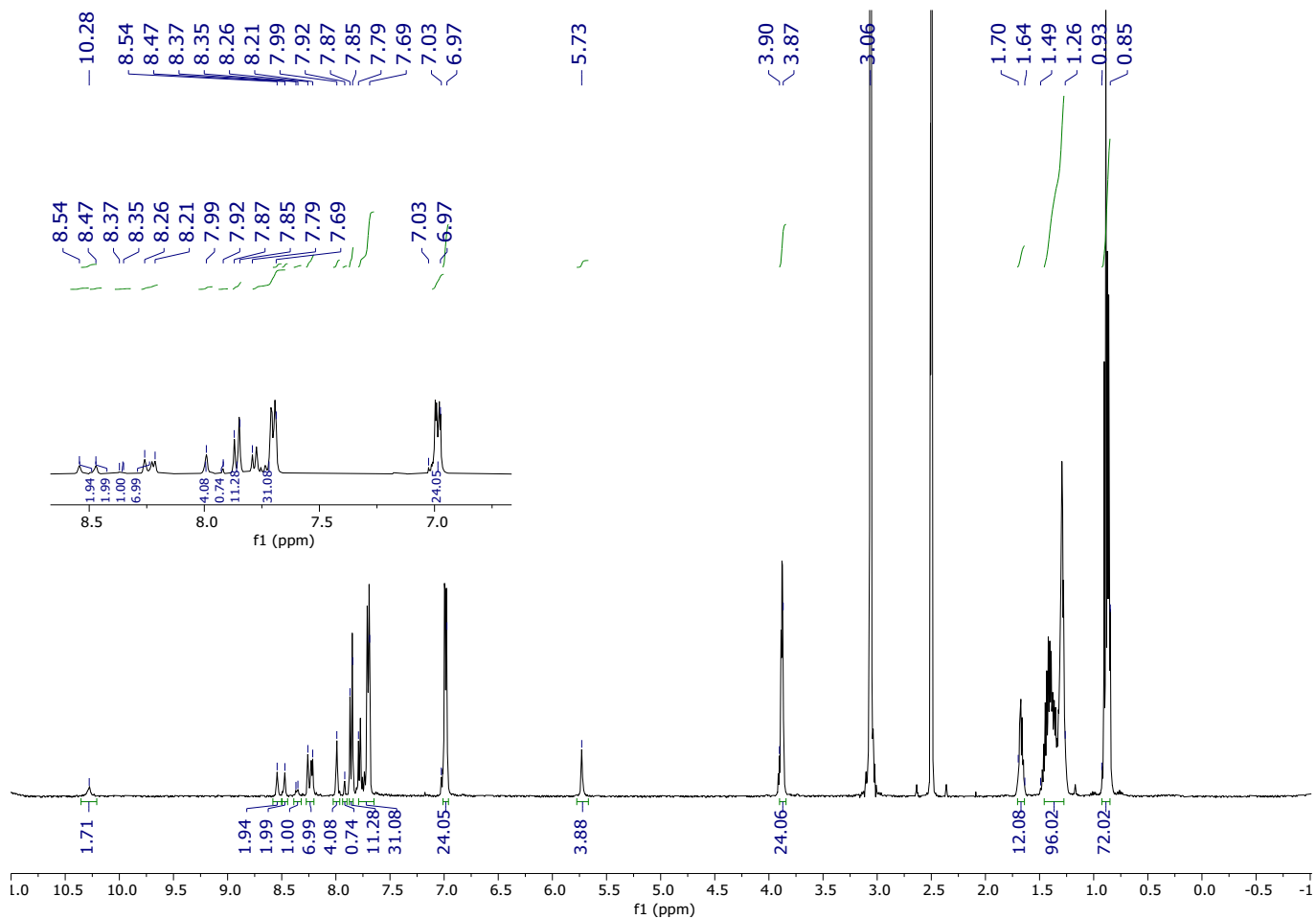


Fig. S6.  $^1\text{H}$  NMR of Compound **21** in  $\text{DMSO-}d_6$ .

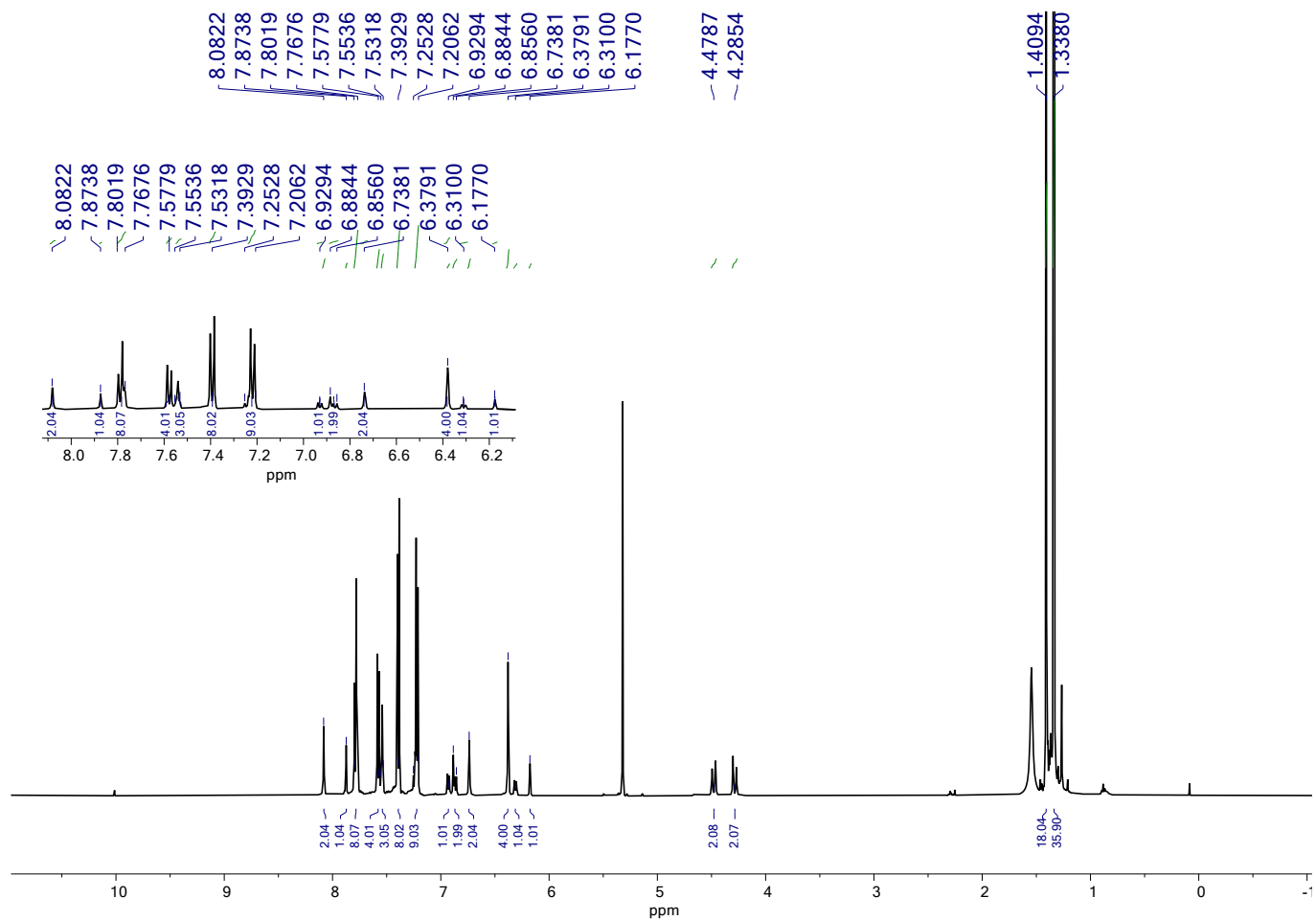


Fig. S7.  $^1\text{H}$  NMR of Dendrimer **1** in  $\text{CD}_2\text{Cl}_2$ .

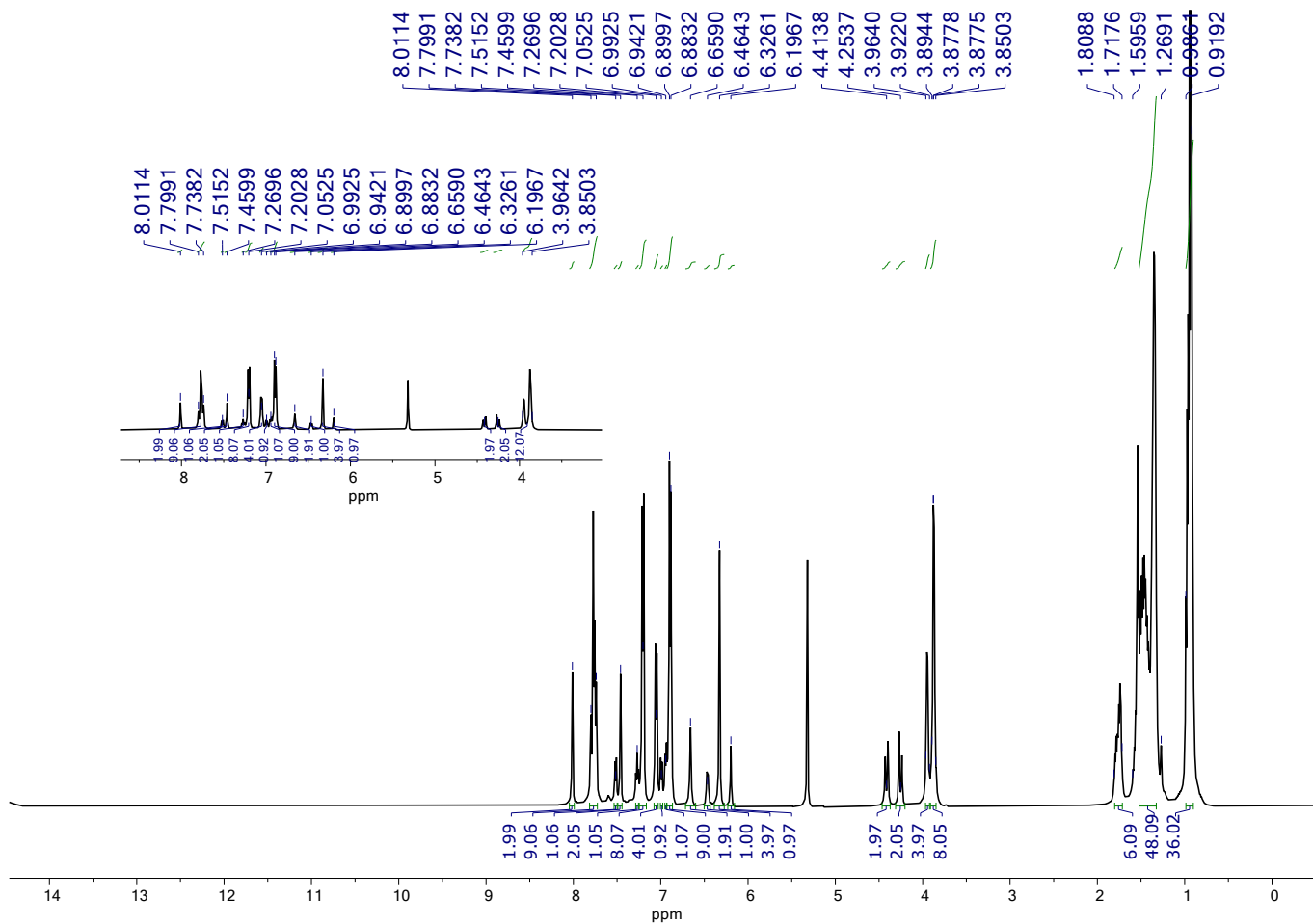


Fig. S8.  $^1\text{H}$  NMR of Dendrimer 2 in  $\text{CD}_2\text{Cl}_2$ .

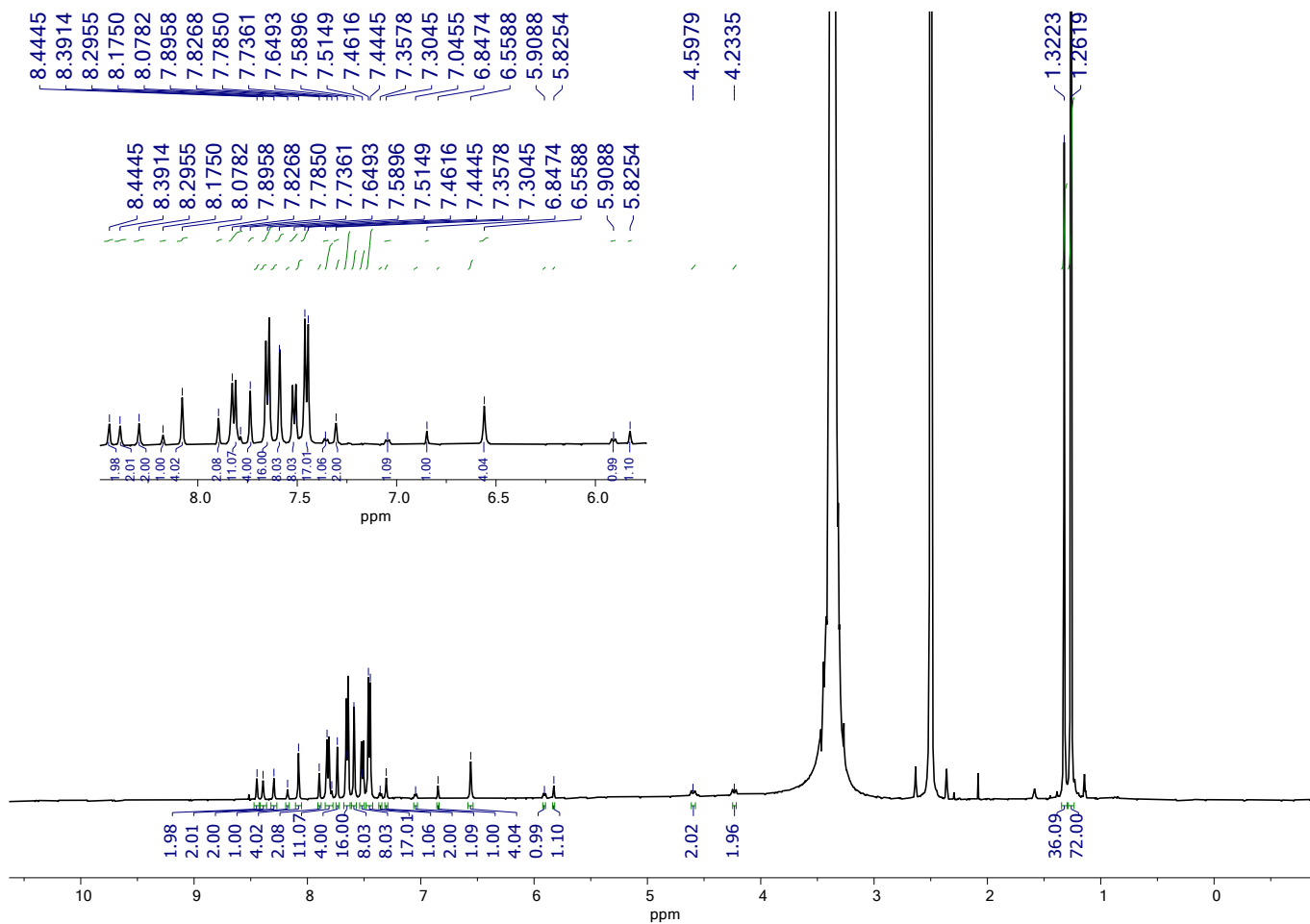


Fig. S9.  $^1\text{H}$  NMR of Dendrimer 3 in  $\text{DMSO}-d_6$ .

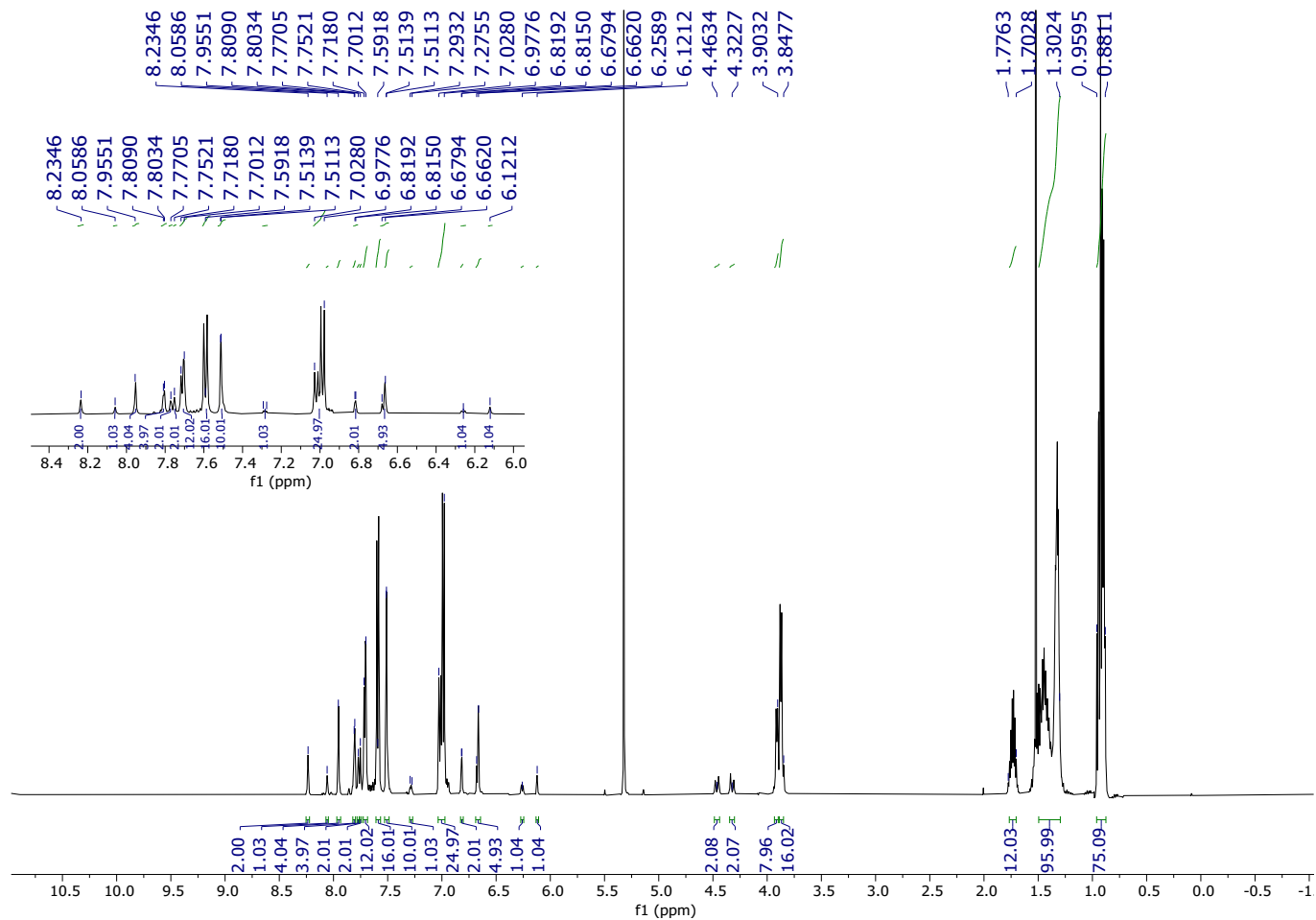


Fig. S10.  $^1\text{H}$  NMR of Dendrimer 4 in  $\text{CD}_2\text{Cl}_2$ .