## Electronic Supplementary Information

## Investigation of Enhanced Third-Order Optical Nonlinearity in Novel Coenzyme-A Capped Silver Nanoparticles

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## S1. Estimation of Lattice planes of silver nanoparticles based on TEM-SAED pattern

By using Image J software, we calculated the diameter of the 4 rings as followed in Table as Lengths. These values may vary depending on the taken set scale, but the ratios will come always same.

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|  | Area | Mean | Min | Max | Angle | Length | - |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 1 | 135 | 74.186 | 0.802 | 255.000 | -47.411 | 134.469 |  |
| 2 | 157 | 72.232 | 0.000 | 254.944 | -98.858 | 155.859 |  |
| 3 | 222 | 76.548 | 0.000 | 238.790 | -59.482 | 220.554 |  |
| 4 | 259 | 80.780 | 0.000 | 255.000 | -90.000 | 258.000 | - |
| 4 |  |  |  |  |  |  |  |

From the figure it appears that the first two rings were closer followed by 3rd and 4th were closer indicating a FCC pattern.

For cubic crystals, $\mathrm{d}_{\mathrm{hkl}}=\mathrm{a}_{0} /\left(\mathrm{h}^{2}+\mathrm{k}^{2}+\mathrm{l}^{2}\right)^{1 / 2}$ where $\mathrm{d}_{\mathrm{hkl}}$ is the d spacing, $\mathrm{a}_{0}$ is the lattice constant, $\mathrm{h}, \mathrm{k}$ \& 1 are the Miller indices.

Using the equivalence of D spacing and radium of rings,
$\mathrm{R}_{\mathrm{hkl}}=\left(\lambda \mathrm{L} / \mathrm{a}_{0}\right)\left(\mathrm{h}^{2}+\mathrm{k}^{2}+\mathrm{l}^{2}\right)^{1 / 2}$
As the value of $\left(\lambda \mathrm{L} / \mathrm{a}_{0}\right)$ is constant for a given diffraction pattern, the values of $\mathrm{R}_{\mathrm{hkl}}$ vary according to $\left(\mathrm{h}^{2}+\mathrm{k}^{2}+\mathrm{l}^{2}\right)^{1 / 2}$.

By using the ratios:
$\mathrm{R}_{1} / \mathrm{R}_{2}=\mathrm{d}_{1} / \mathrm{d}_{2}=\left(\mathrm{h}_{1}{ }^{2}+\mathrm{k}_{1}{ }^{2}+\mathrm{l}_{1}{ }^{2}\right)^{1 / 2} /\left(\mathrm{h}_{2}{ }^{2}+\mathrm{k}_{2}{ }^{2}+\mathrm{l}_{2}{ }^{2}\right)^{1 / 2}=134.469 / 155.859=0.862$
$\mathrm{R}_{2} / \mathrm{R}_{3}=\mathrm{d}_{2} / \mathrm{d}_{3}=\left(\mathrm{h}_{2}{ }^{2}+\mathrm{k}_{2}{ }^{2}+\mathrm{l}_{2}{ }^{2}\right)^{1 / 2} /\left(\mathrm{h}_{3}{ }^{2}+\mathrm{k}_{3}{ }^{2}+\mathrm{l}_{3}{ }^{2}\right)^{1 / 2}=155.859 / 220.554=0.706$
$\mathrm{R}_{3} / \mathrm{R}_{4}=\mathrm{d}_{3} / \mathrm{d}_{4}=\left(\mathrm{h}_{3}{ }^{2}+\mathrm{k}_{3}{ }^{2}+\mathrm{l}_{3}{ }^{2}\right)^{1 / 2} /\left(\mathrm{h}_{4}{ }^{2}+\mathrm{k}_{4}{ }^{2}+\mathrm{l}_{4}{ }^{2}\right)^{1 / 2}=220.554 / 258.000=0.854$
However, the above obtained values are only possible if the planes are $\{111\},\{200\},\{220\}$, $\{311\}$ as
$\mathrm{R}_{1} / \mathrm{R}_{2}=0.862 \sim\left(1^{2}+1^{2}+1^{2}\right)^{1 / 2} /\left(2^{2}+0^{2}+0^{2}\right)^{1 / 2}=0.866$
$\mathrm{R}_{2} / \mathrm{R}_{3}=0.706 \sim\left(2^{2}+0^{2}+0^{2}\right)^{1 / 2} /\left(2^{2}+2^{2}+0^{2}\right)^{1 / 2}=0.707$
$\mathrm{R}_{3} / \mathrm{R}_{4}=0.854 \sim\left(2^{2}+2^{2}+0^{2}\right)^{1 / 2} /\left(3^{2}+1^{2}+1^{2}\right)^{1 / 2}=0.852$
Thus, we can conclude that allowed reflections are for an fcc crystal lattice. The order of rings in increasing radius are: $\{111\},\{200\},\{220\},\{311\},\{222\},\{400\},\{331\},\{420\},\{422\} \ldots$

Table S1- Third-order non-linear susceptibility $\chi^{(3)}$ of various nanoparticle systems reported in the literature

| S.No | Material | Reference | $\chi^{(3)}$ material | $\chi^{(3)}$ standard | Reference |
| :--- | :--- | :---: | :--- | :--- | :--- |
| 1 | AgNPs- CoA <br> in water | $\mathrm{CS}_{2}$ | $1.38 \times 10^{-13}$ | $1.75 \mathrm{X} \mathrm{10} 0^{-12}$ | Present work |
| 2 | AgNPs in <br> water | $\mathrm{CS}_{2}$ | $2.95 \times 10^{-14}$ | - | $(45)$ |
| 3 | Ag colloids in <br> acetone | $\mathrm{CCL}_{4}$ | $1.89 \times 10^{-14}$ | $4.40 \times 10^{-14}$ | $(52)$ |
| 4 | Ag colloids in <br> DCM | $\mathrm{CCL}_{4}$ | $3.6 \times 10^{-12}$ | $4.40 \times 10^{-14}$ | $(52)$ |
| 5 | Ag colloids in <br> chloroform | $\mathrm{CCL}_{4}$ | $5.3 \times 10^{-13}$ | $4.40 \times 10^{-14}$ | $(52)$ |
| 6 | AgNPs in <br> water | $\mathrm{CS}_{2}$ | $7.7 \times 10^{-14}$ | $9.32 \times 10^{-12}$ | $(53)$ |
| 7 | AuNPs in <br> water | $\mathrm{CS}_{2}$ | $5.52 \times 10^{-13}$ | $9.32 \times 10^{-12}$ | $(53)$ |
| 8 | AuNPs in <br> water | $\mathrm{CCL}_{4}$ | $1.93 \times 10^{-14}$ | $4.40 \times 10^{-14}$ | $(54)$ |

