

Supporting Information

Preparation of Freestanding and Ultrastable CsPbX₃ Perovskite Quantum Dots/SEBS Composite Film for Curved and Flexible Surfaces

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1. Calculation of $S_{1/w}$ (CsPbBr₃ PQDs/SEBS/cyclohexane dispersion): According to $S_{1/w} = \gamma_w - \gamma_l - \gamma_{l/w}$ (4), the coefficient S of CsPbBr₃ PQDs/SEBS/cyclohexane dispersion (0.15 g/ml, ambient temperature) can be calculated as $S = \gamma_w - \gamma_l - \gamma_{l/w} = (72.0 - 24.7 - 31.8) \text{ mN/m} = 15.5 \text{ mN/m}$.

2. Supplementary Figures.

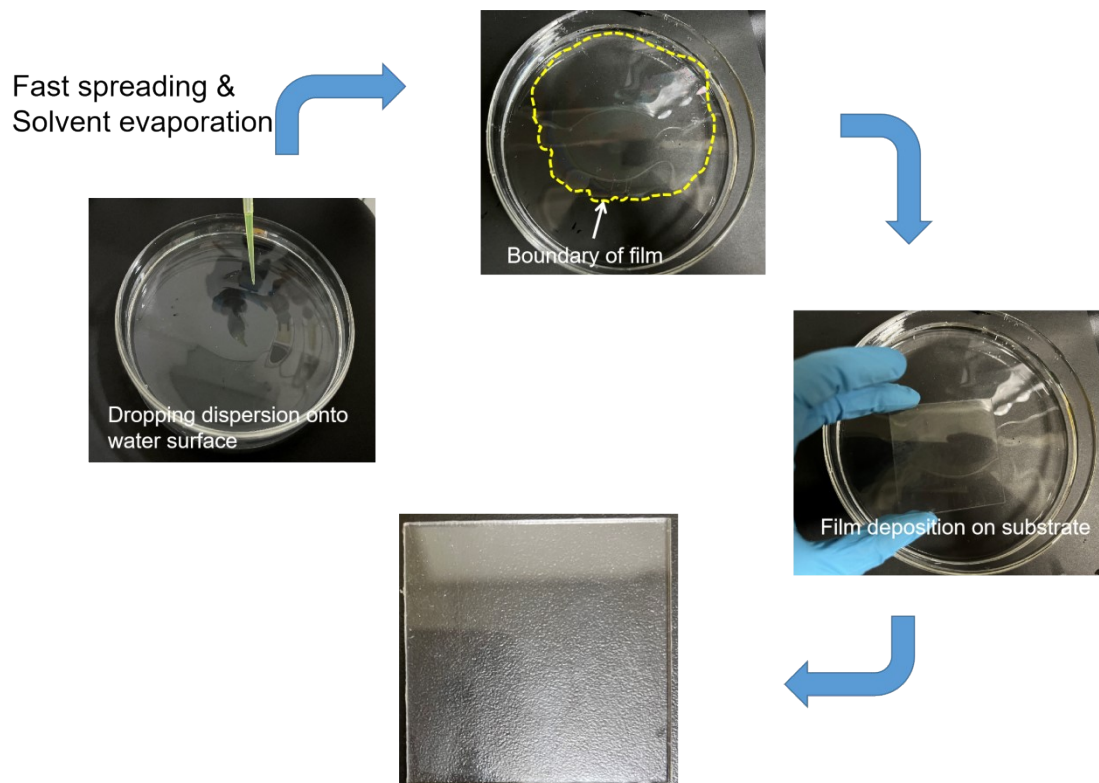


Figure S1. Photographs of the fabrication of CsPbX₃ perovskite quantum dots (PQDs)/polymer composite films.

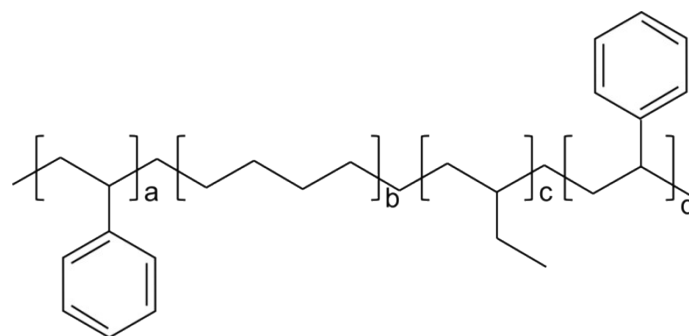


Figure S2. The structure of poly(styrene-ethylene-butylene-styrene) (SEBS) block copolymer.

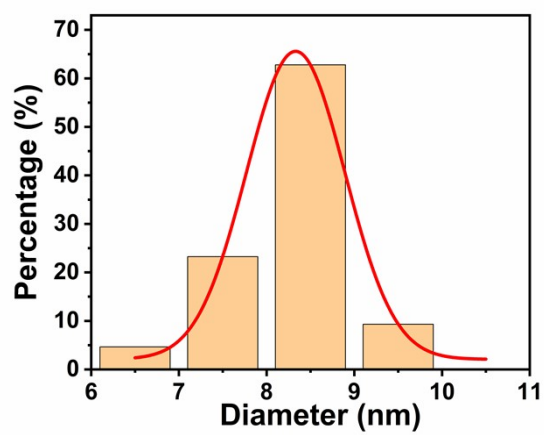


Figure S3. Histogram of particle-size distribution of CsPbBr₃ PQDs.

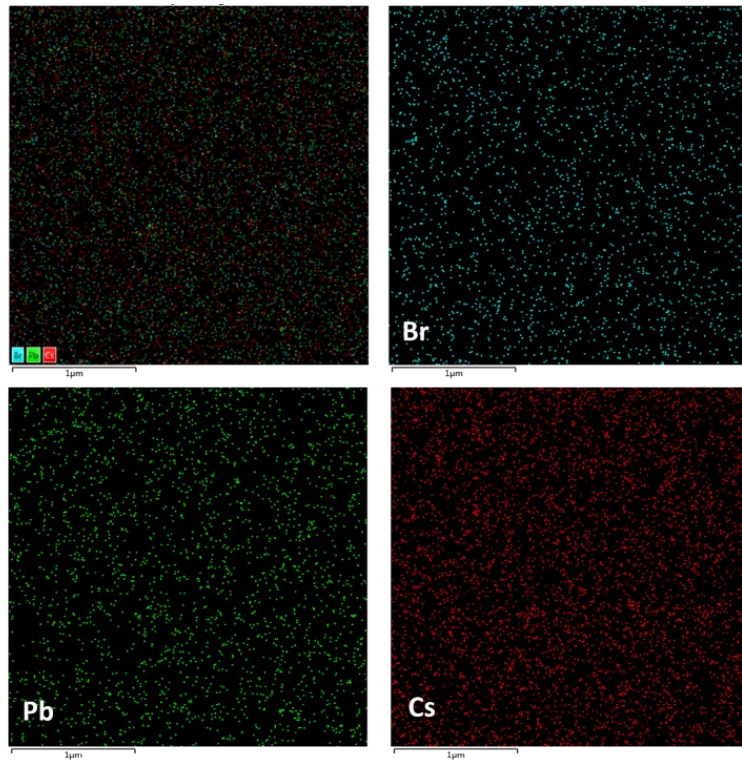


Figure S4. Electronic image and elemental mapping of CsPbBr₃ PQDs/polymer composite films.

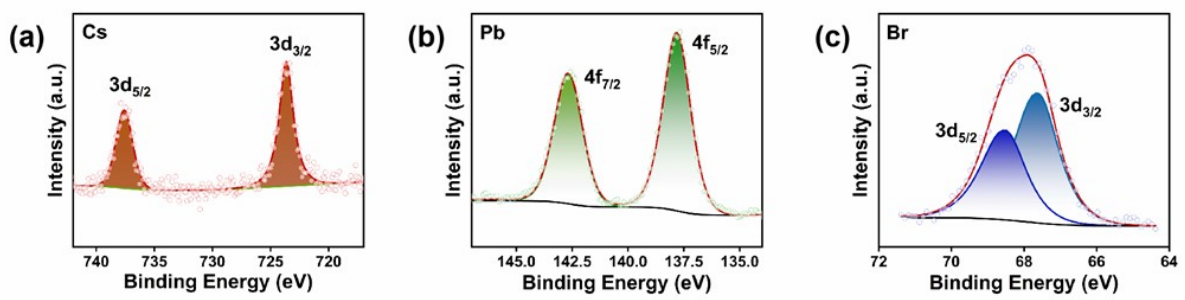


Figure S5. XPS spectra of (a) Cs 3d, (b) Pb 4f, and (c) Br 3d

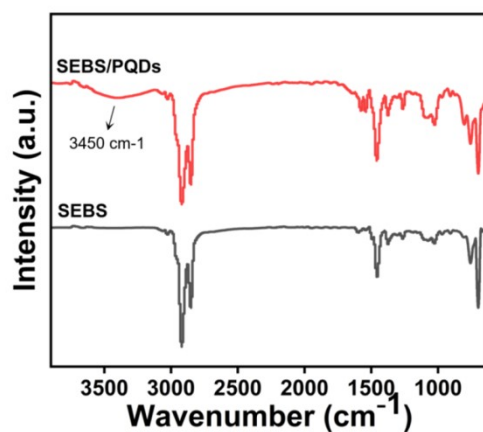


Figure S6. FTIR spectra of the CsPbBr₃ PQD/polymer composite film and the raw material of SEBS.

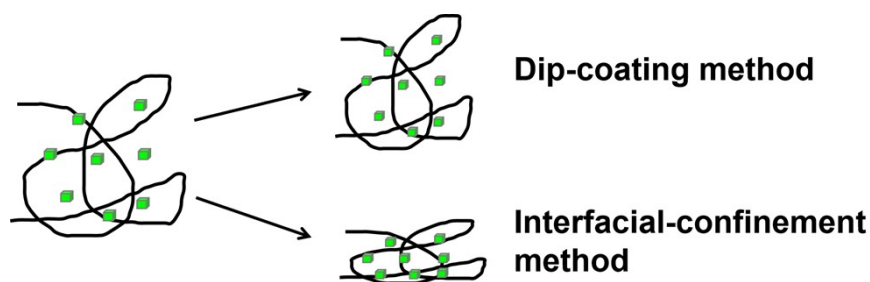


Figure S7. Conformation model of films by dip-coating and interfacial-confinement methods.

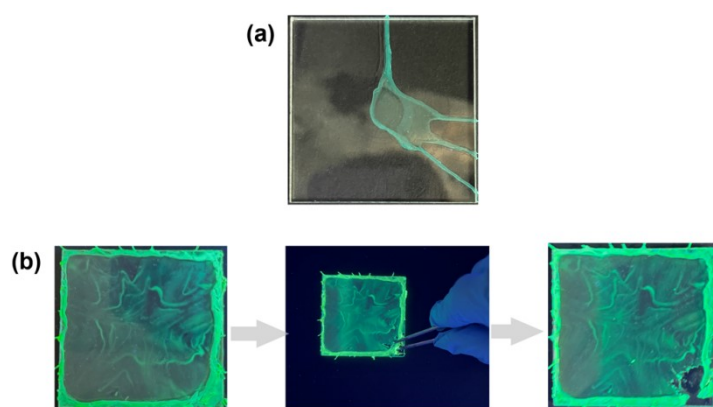


Figure S8. (a) Photographs of film fabricated by spin-coating, adding 40 μL of precursor solution (0.15 g/mL). (b) Tearing process of a film fabricated by spin-coating.

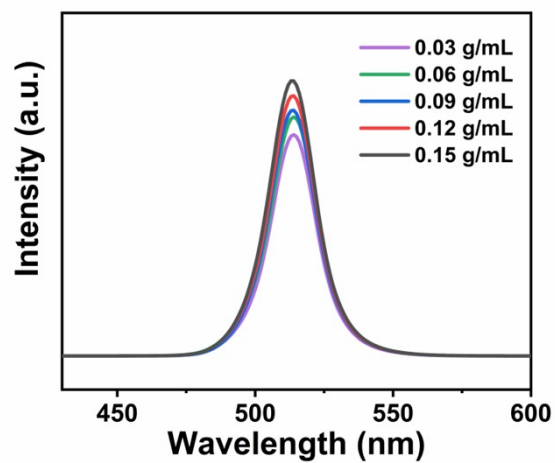


Figure S9. PL spectra of films with different concentrations of SEBS solution.

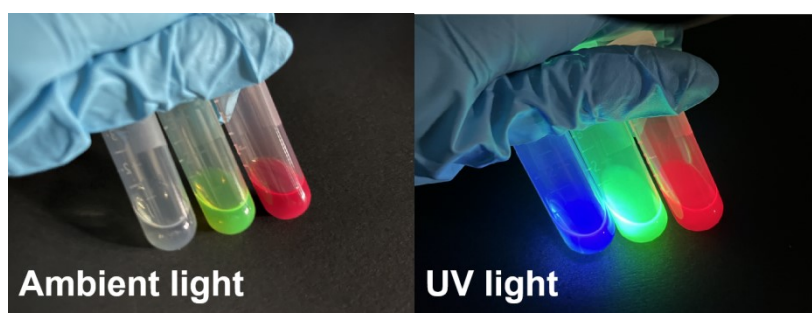


Figure S10. Photographs of CsPbX₃ PQDs from blue to red under ambient light and UV light.

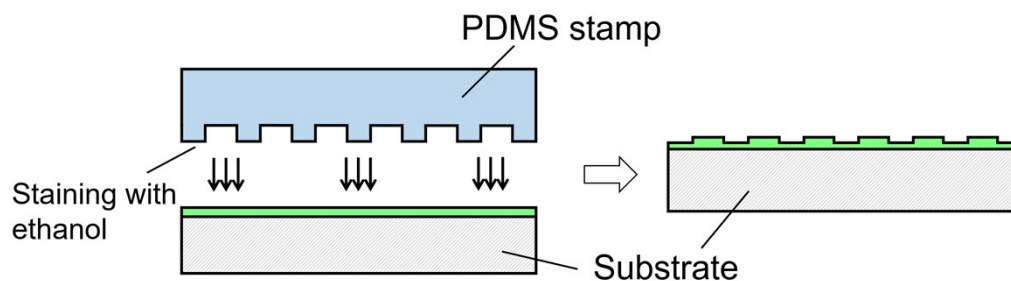


Figure S11. Illustration of patterning on the film using poly(dimethylsiloxane) (PDMS) stamp.