

Supporting Information

Phenothiazine and phenothiazine sulfone derivatives: AIE, HTMs for doping free fluorescent and multiple-resonance TADF OLEDs

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General methods

Chemicals were used as received unless otherwise indicated. All oxygen or moisture sensitive reactions were performed under a nitrogen/argon atmosphere. ^1H NMR (500 MHz) and ^{13}C NMR (126 MHz) spectra were recorded on a Bruker 500 MHz FT-NMR spectrometer at room temperature. ^1H NMR chemical shifts are reported in parts per million (ppm) relative to the solvent residual peak (CDCl_3 , 7.26 ppm). Multiplicities are given as s (singlet), d (doublet), t (triplet) and m (multiplet) and the coupling constants, J , are given in hertz. ^{13}C NMR chemical shifts are reported relative to the solvent residual peak (CDCl_3 , 77.0 ppm). UV-visible absorption spectra were recorded on a PerkinElmer Lambda 35 instrument. Emission spectra were taken in a Fluoromax-4C, S/n.1579D-1417- FM Fluorescence software Ver 3.8.0.60. The relative quantum yields in solution were calculated using Quinine sulphate in 0.5 M H_2SO_4 as standard for $(\text{PTZ})_3$, $\text{PTZO}_2\text{-}(\text{PTZ})_2$, $\text{PTZ}\text{-}(\text{TPE})_2$ and $\text{PTZO}_2\text{-}(\text{TPE})_2$. All the measurements were carried out at 25 °C. HRMS was recorded with a Bruker-Daltonics micrOTOF-Q II mass spectrometer.

Optical characterization

Ultraviolet-visible (UV-Vis) absorbance spectra of the films of the compounds were recorded Perkin Elmer Lambda 35 spectrometer. Toluene solutions (1 mg/ml) were drop casted on the quartz substrates for photophysical measurements (Figure S\$). The photoluminescence spectra of the solid state of the materials and doped in ZEONEX were measured in air and vacuum at room temperature by Edinburgh Instruments' FLS980 Fluorescence Spectrometer with a excitation wavelength of 340 nm. Photoluminescence quantum yields (PLQY) of the compounds in solid state (spin coated on quartz) and in toluene were obtained using the integrated sphere. PL decay curves were measured by utilizing the PicoQuant LDH-D-C-375 laser (wavelength 374 nm) as the excitation source. Since the PL decay was fitted with two τ

components the average lifetime was calculated using the intensity average lifetime Equation (1):⁵²

$$\langle \tau \rangle_f = \frac{\int_0^x t \sum a_i \exp(-t/\tau_i) dt}{\int_0^x \sum a_i \exp(-t/\tau_i) dt} = \frac{\sum a_i \tau_i^2}{\sum a_i \tau_i} \quad (1)$$

Thermal characterization

Thermal properties of the synthesized compounds were analysed using differential scanning calorimetry (DSC) technique performed with Perkin Elmer DSC 8500 equipment at heating and cooling rates of 10°C/min under nitrogen atmosphere. An empty aluminum pan was used as a reference. Thermogravimetric analysis (TGA) was performed using Perkin Elmer TGA 4000 apparatus at a heating rate of 20°C/min under nitrogen atmosphere. Melting points were measured by MEL-TEMP (Electrothermal) melting point apparatus.

Ionization potential

Ionization potential (IP^{PE}) of the solid samples were estimated by electron photoemission spectrometry in air. The samples were vacuum-deposited (2×10^{-6} mbar) onto fluorine doped tin oxide (FTO) coated glass substrates, exploiting setup consisting of ASBN-D130-CM deep UV deuterium light source, CM110 1/8m monochromator and 6517B Keithley electrometer. Photoelectron emission spectra was recorded exciting the samples from low energy to high energy with the step of 1 nm and recording electron photoemission current at different excitation energies.

For charge mobilities, time of flight technique (TOF) was used. The samples were vacuum-deposited (2×10^{-6} mbar) onto indium doped tin oxide (ITO) coated glass substrates. The TOF

setup contains a laser (EKSPLA NL300) used for excitation at 355 nm wavelength. For electron and hole transport in deposited layers at different electric fields, a variety of positive external voltages (U) were applied. TDS 3032C oscilloscope (Tektronix) was used to record the photocurrent transients of holes or electrons.

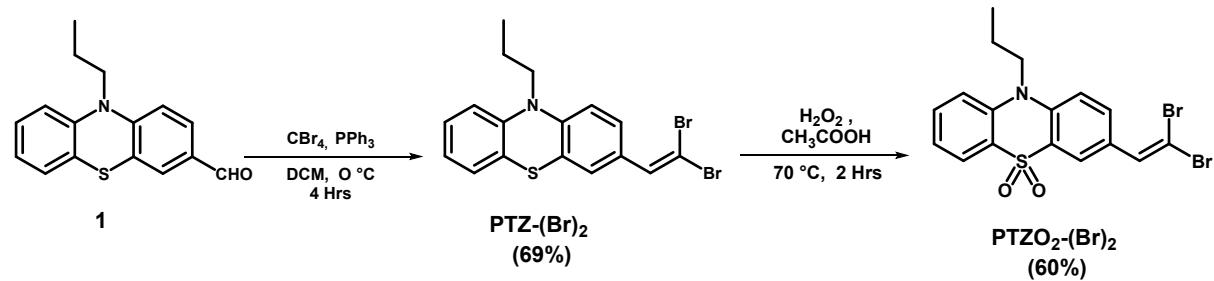
Electrochemical properties

Electrochemical properties of the compounds in solution were studied using cyclic voltammetry (CV) technique. Micro-Autolab III (Metrohm Autolab) potentiostat-galvanostat equipped with a standard three-electrode configuration consisting of a glassy carbon working electrode, an Ag/Ag^+ reference electrode, and a platinum wire counter electrode. Compounds and ferrocene were dissolved in a solution containing 0.1 M tetrabutylammonium hexafluorophosphate (TBAF_6) as the supporting electrolyte and dry dichloromethane solvent under nitrogen atmosphere at room temperature. CV measurements were performed with a scan rate of 0.1 V/s. The measurements were calibrated using ferrocene/ferrocenium (Fc) system, as an internal standard.

OLEDs

Electroluminescent devices were fabricated using vacuum deposition (2×10^{-6} mbar) of organic and metal layers onto pre-cleaned ITO coated glass substrates. Density-voltage and luminance-voltage characteristics were recorded using photodiode PH100-Si-HA-D0 together with the PC-based power and energy Monitor 11S-LINK and Keithley 2400C source meter. Electroluminescence (EL) spectra were recorded by Avantes AvaSpec-2048XL spectrometer.

Synthetic Scheme



Scheme S1. Synthesis of intermediates **PTZ-(Br)₂** and **PTZO₂-(Br)₂**.

The intermediates **PTZO₂-(Br)₂** and **PTZ-(Br)₂** were synthesized according to the literature procedure. (Ref. S1 and Ref. S2). The boronate ester derivatives of phenothiazine and tetraphenylethylene (**PTZ-Bpin** and **TPE-Bpin**) were synthesized according to the literature procedure. (Ref. S3 and Ref. S4).

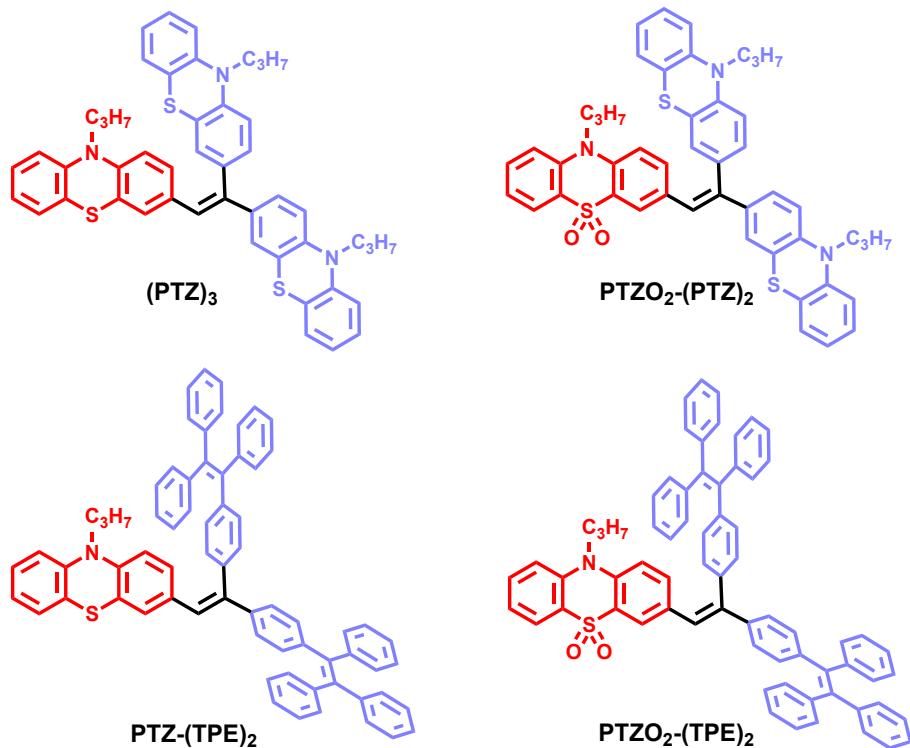


Chart S1. Chemical structures of phenothiazine and tetraphenylethylene based luminogens **(PTZ)₃**, **PTZO₂-(PTZ)₂**, **(PTZ)₂**, **PTZ-(TPE)₂**, and **PTZO₂-(TPE)₂**.

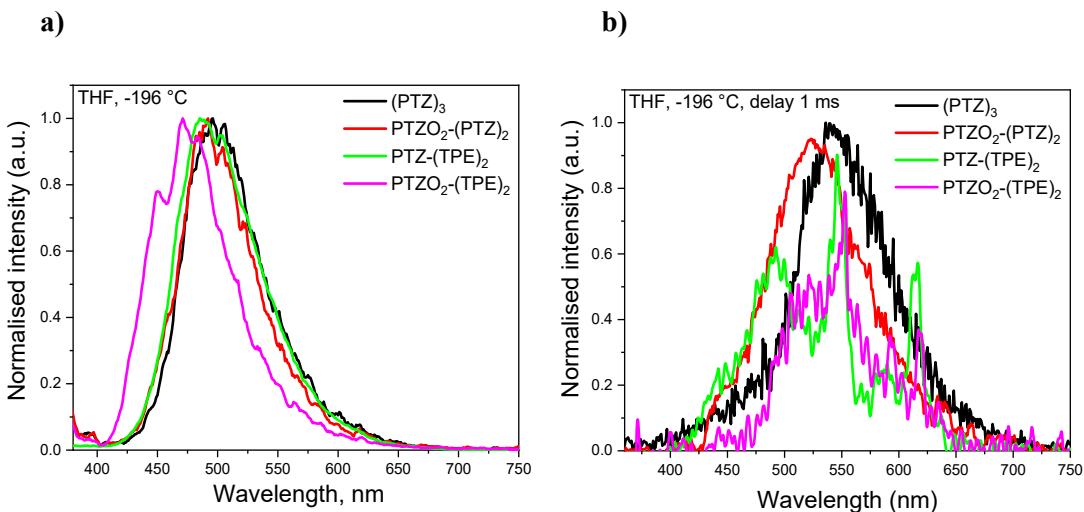


Fig. S1. Normalized PL (a) and phosphorescence (b) spectra of the frozen THF solutions of **(PTZ)₃**, **PTZO₂-(PTZ)₂**, **PTZ-(TPE)₂**, and **PTZO₂-(TPE)₂** recorded at -196 °C ($\lambda_{\text{ex}} = 340$ nm).

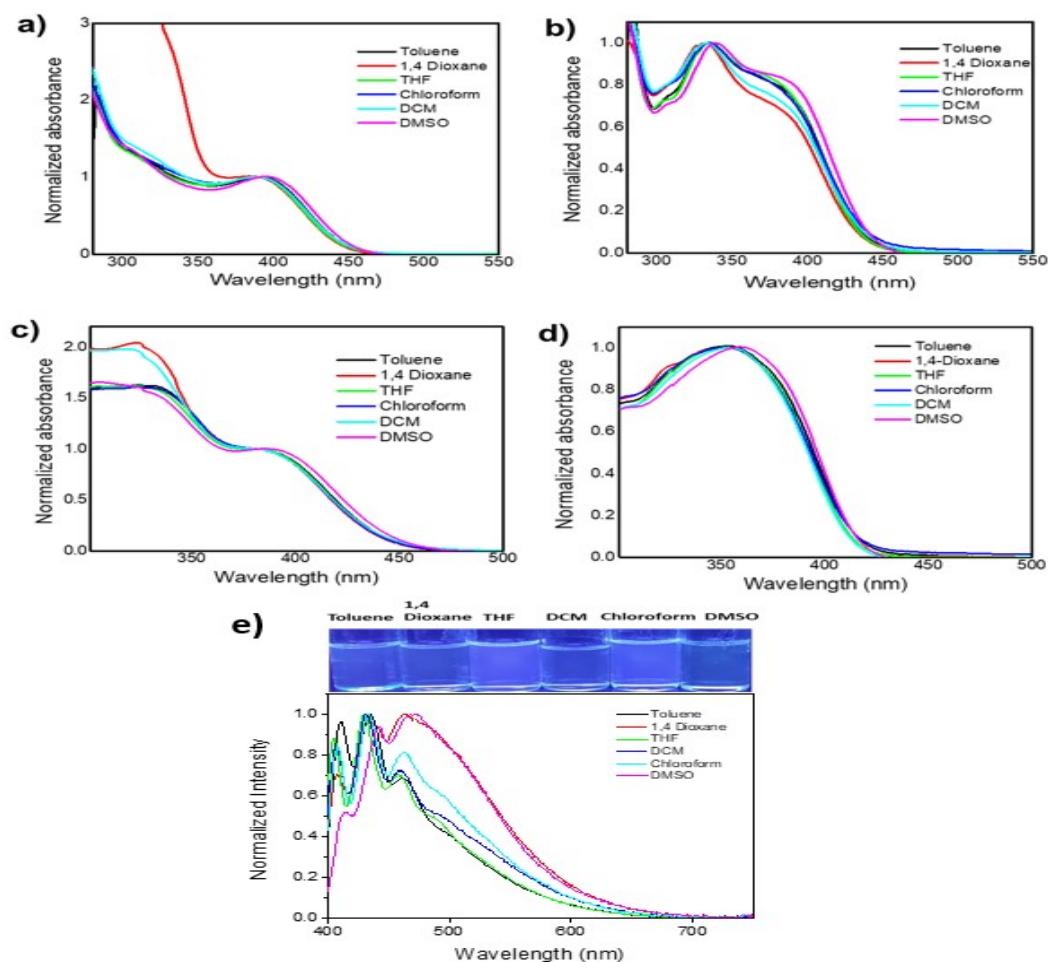


Fig. S2 Normalized absorbance spectra of (a) $(\text{PTZ})_3$, (b) $\text{PTZO}_2\text{-}(\text{PTZ})_2$, (c) $\text{PTZ}\text{-}(\text{TPE})_2$ and (d) $\text{PTZO}_2\text{-}(\text{TPE})_2$ in different polarity solvents, (e) normalized emission spectra of $\text{PTZO}_2\text{-}(\text{TPE})_2$ in different polarity solvents.

Table S1 Photophysical properties of compounds $(\text{PTZ})_3$, $\text{PTZO}_2\text{-}(\text{PTZ})_2$, $\text{PTZ}\text{-}(\text{TPE})_2$, and $\text{PTZO}_2\text{-}(\text{TPE})_2$.

Compounds	Solvent	$\lambda_{\text{ab}}\text{(nm)}^{\text{a}}$	$\lambda_{\text{em}}\text{(nm)}^{\text{a}}$	Stokes shift (cm ⁻¹)	ϕ_f^{b}
$(\text{PTZ})_3$	Toluene	391	530	6708	0.38
	1,4-Dioxane	387	533	7078	0.50
	THF	387	536	7183	0.37
	Chloroform	391	536	6919	0.34
	DCM	387	546	7524	0.36
	DMSO	394	563	7619	0.49
$\text{PTZO}_2\text{-}(\text{PTZ})_2$	Toluene	334	512	10409	0.20
	1,4-Dioxane	332	516	10741	0.18
	THF	333	532	11234	0.15
	Chloroform	334	532	11144	0.15
	DCM	333	545	11682	0.20
	DMSO	338	577	3060	0.24
$\text{PTZ}\text{-}(\text{TPE})_2$	Toluene	383	536	7453	0.39
	1,4-Dioxane	382	541	7694	0.31
	THF	382	559	8289	0.27
	Chloroform	385	547	7693	0.35
	DCM	385	563	8213	0.27
	DMSO	385	600	9308	0.19
$\text{PTZO}_2\text{-}(\text{TPE})_2$	Toluene	354	-	-	-
	1,4-Dioxane	352	-	-	-
	THF	354	-	-	-
	Chloroform	353	-	-	-
	DCM	354	-	-	-

	DMSO	360	-	-	-
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^a absorption and emission maxima were recorded in different polarity solvents, ^b the fluorescence quantum yields were measured using Quinine sulphate in 0.5 M H₂SO₄ as standard for **(PTZ)₃**, **PTZO₂-(PTZ)₂**, **PTZ-(TPE)₂** and **PTZO₂-(TPE)₂**.

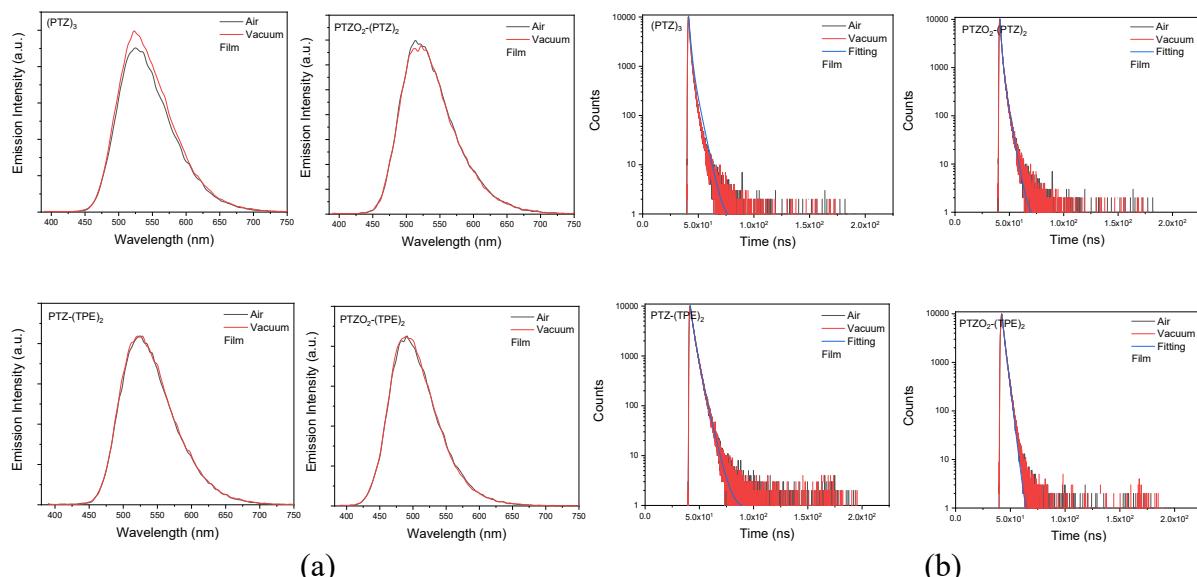
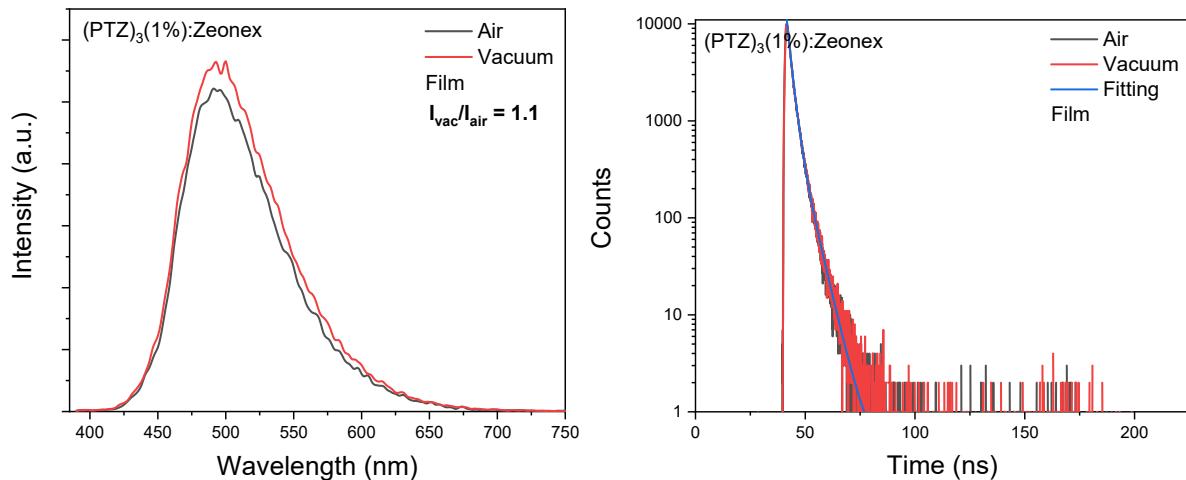


Fig. S3 (a) PL spectra and (b) PL decay of $(\text{PTZ})_3$, $\text{PTZO}_2\text{-}(\text{PTZ})_2$, $\text{PTZ}\text{-}(\text{TPE})_2$, and $\text{PTZO}_2\text{-}(\text{TPE})_2$ on glass substrate in air and vacuum.



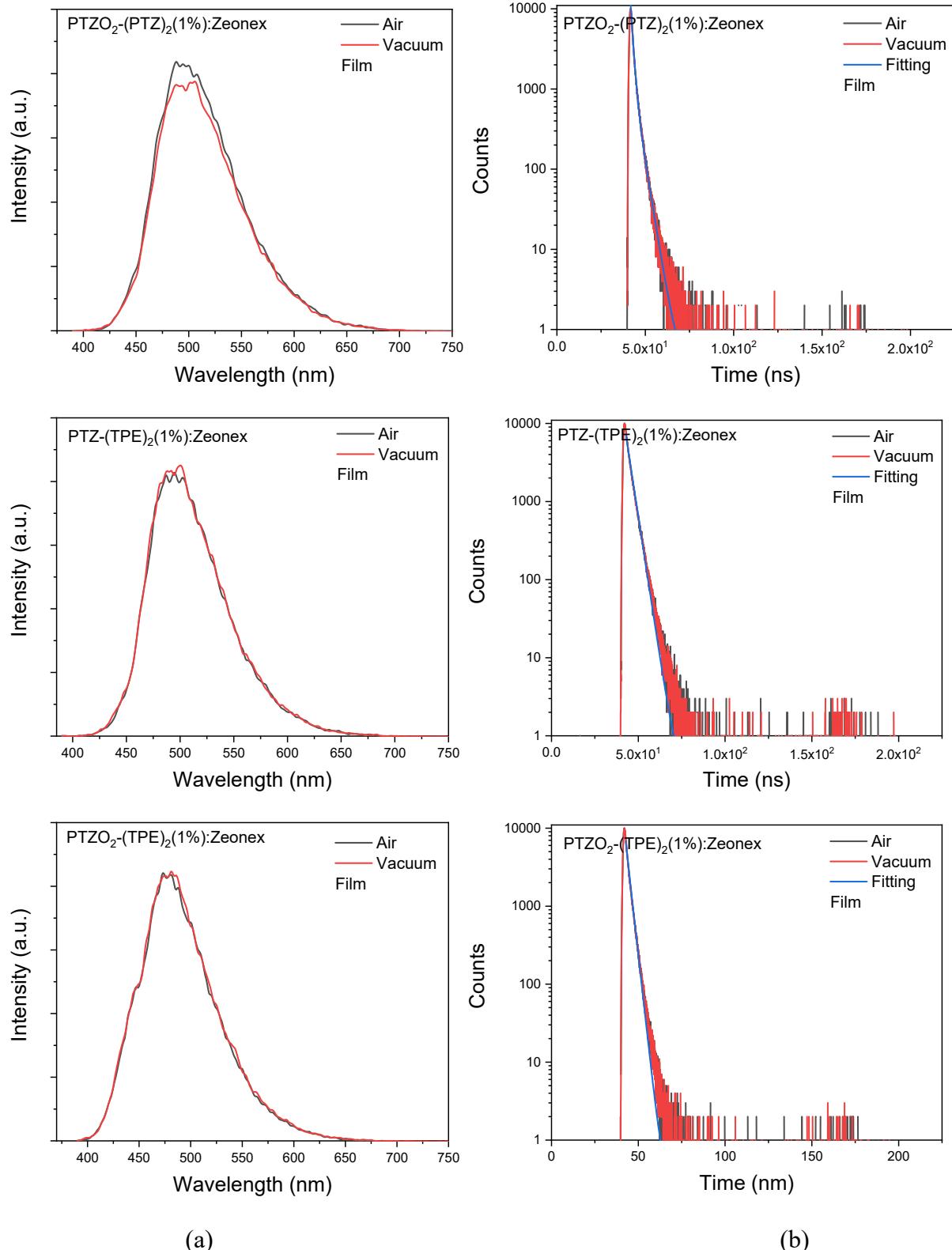
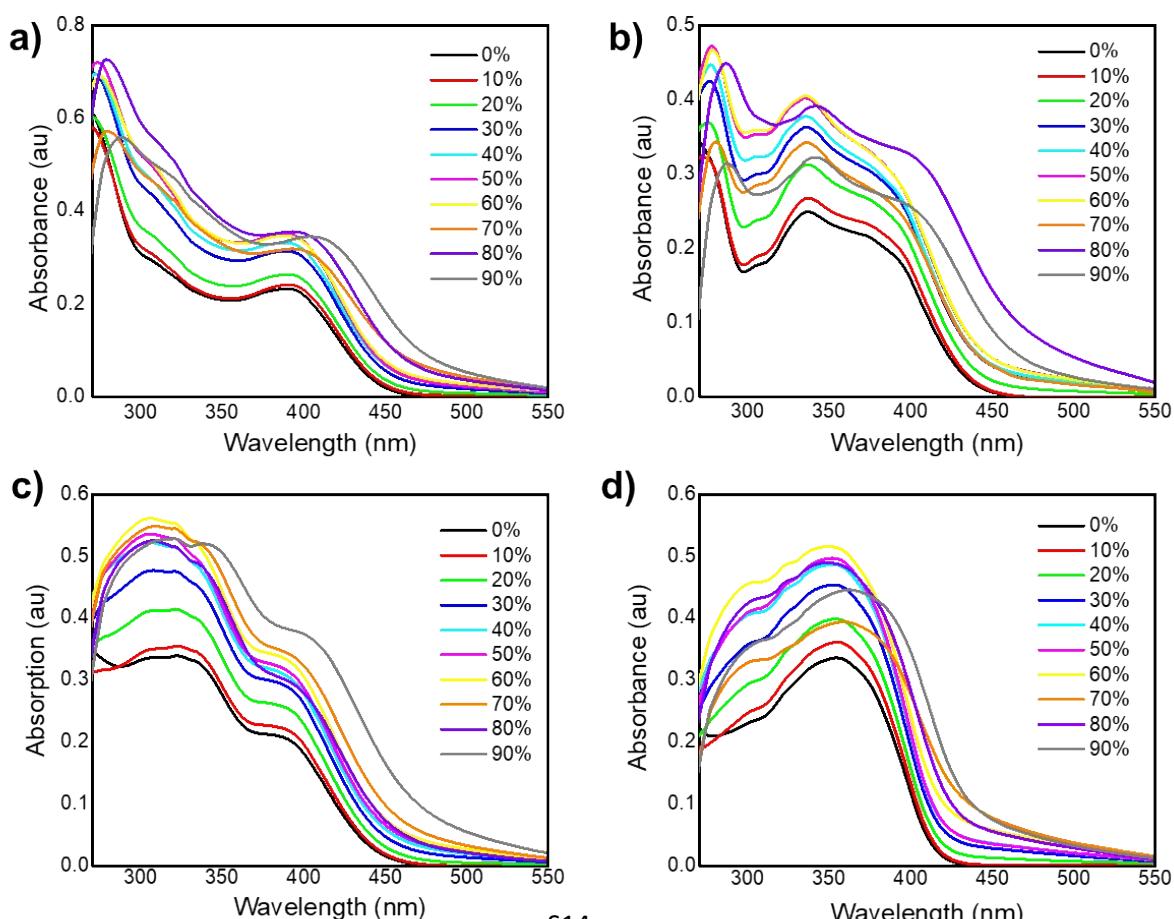


Fig. S4 (a) PL spectra and (b) PL decay of the compounds (1wt%) in ZEONEX on glass substrate in air and vacuum.

Table S2. Intensity averaged lifetime.

Parameters	Compounds			
	(PTZ) ₃	PTZO ₂ (PTZ) ₂	PTZ(TPE) ₂	PTZO ₂ (TPE) ₂
A ₁	9017.233	9110.058	7823.938	9950.755
τ ₁ (ns)	1.3525	1.1106	2.4278	2.3012
A ₂	1985.008	1737.818	3000.473	-
τ ₂ (ns)	3.9742	3.375	4.8067	-
Σ A _i τ _i ²	47846.57	31031.47	115440	-
Σ A _i τ _i	20084.63	15982.77	33417.33	-
τ _{avg} (ns)	2.382249	1.941558	3.454494	-
X ²	1.05	1.01	1.02	1.00
Compounds(1%) in ZEONEX				
A ₁	9839.454	9992.021	7246.076	6920.84
τ ₁ (ns)	1.8284	1.1459	2.9773	2.1785
A ₂	1524.606	1636.245	-	-
τ ₂ (ns)	4.5609	3.2363	-	-
Σ A _i τ _i ²	64608.32	30257.83	-	-
Σ A _i τ _i	24944.03	16745.24	-	-
τ _{avg} (ns)	2.590131	1.806951	-	-
X ²	1.08	1.00	1.19	1.00



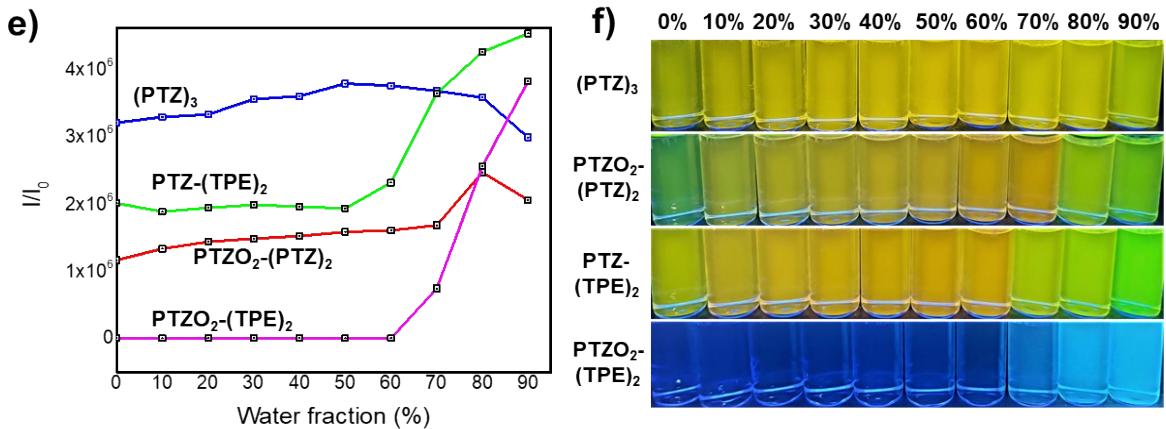


Fig. S5. Absorption spectra of (a) $(\text{PTZ})_3$ (b) $\text{PTZO}_2-(\text{PTZ})_2$ (c) $\text{PTZ}-(\text{TPE})_2$ and (d) $\text{PTZO}_2-(\text{TPE})_2$ in THF–water mixtures with different water fractions; (e) Plot of fluorescence intensity *vs.* % of water fraction for compounds $(\text{PTZ})_3$, $\text{PTZO}_2-(\text{PTZ})_2$, $\text{PTZ}-(\text{TPE})_2$, and $\text{PTZO}_2-(\text{TPE})_2$; (f) Fluorescence images of $(\text{PTZ})_3$, $\text{PTZO}_2-(\text{PTZ})_2$, $\text{PTZ}-(\text{TPE})_2$, and $\text{PTZO}_2-(\text{TPE})_2$ in different THF–water binary mixtures.

Table S3. Fluorescence quantum yields of $(\text{PTZ})_3$, $\text{PTZO}_2-(\text{PTZ})_2$, $\text{PTZ}-(\text{TPE})_2$, and $\text{PTZO}_2-(\text{TPE})_2$ in different THF–water mixtures with increasing water percentage.

Water Vol %	ϕ_f^a			
	$(\text{PTZ})_3$	$\text{PTZO}-(\text{PTZ})_2$	$\text{PTZ}-(\text{TPE})_2$	$\text{PTZO}-(\text{TPE})_2$
0 %	0.31	0.15	0.20	-
10 %	0.30	0.15	0.18	-
20 %	0.28	0.13	0.17	-
30 %	0.25	0.12	0.15	-
40 %	0.24	0.12	0.13	-
50 %	0.24	0.12	0.13	-
60 %	0.23	0.11	0.12	-
70 %	0.23	0.12	0.19	0.04
80 %	0.20	0.12	0.26	0.14
90 %	0.16	0.12	0.22	0.21

(^a Fluorescence quantum yields recorded using quinine sulphate as a standard in 0.5 M H₂SO₄ solution.)

Computational analysis

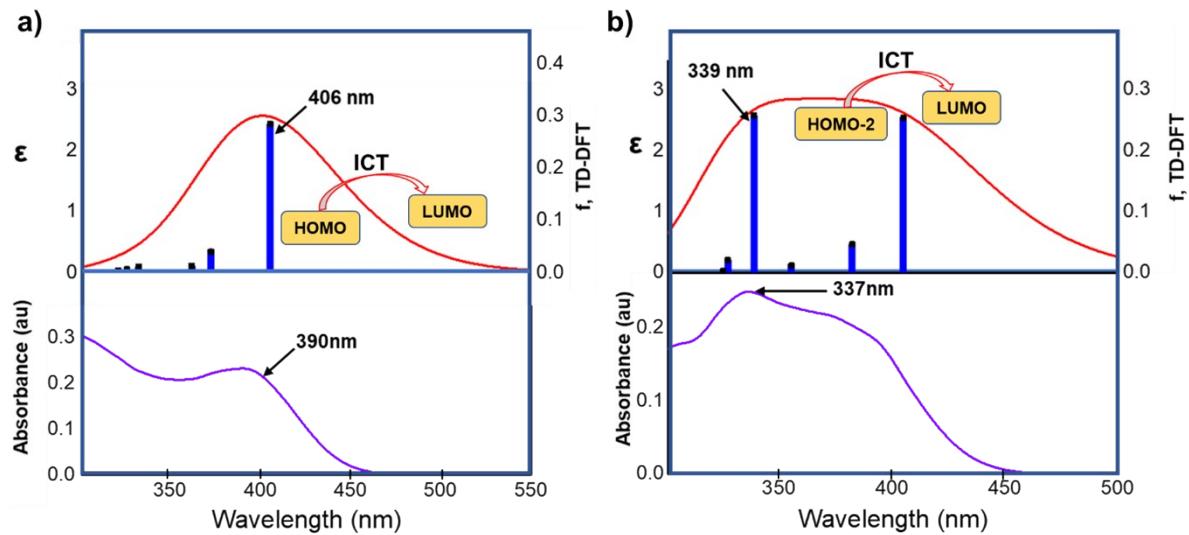


Fig. S6. Comparison of the simulated absorption spectra of (a) $(\text{PTZ})_3$ and (b) $\text{PTZO}_2\text{-}(\text{PTZ})_2$ generated from DFT and TD-DFT calculations (top) with their experimental absorption spectra recorded at room temperature in THF (10^{-5} M) (bottom).

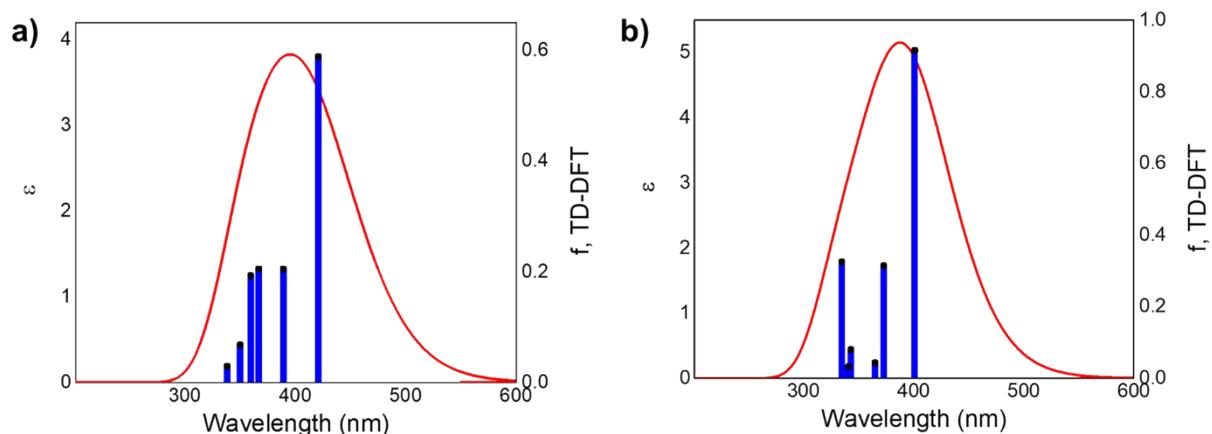


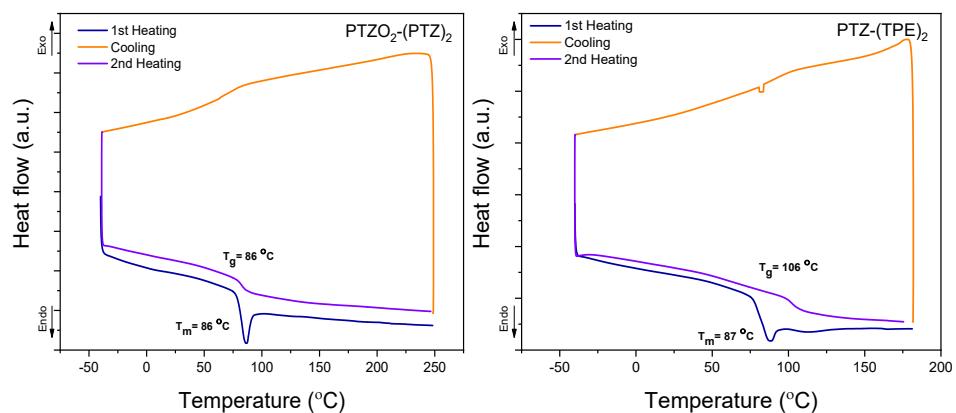
Fig. S7. Simulated absorption spectra of (a) $\text{PTZ}\text{-}(\text{TPE})_2$ and (b) $\text{PTZO}_2\text{-}(\text{TPE})_2$ generated from DFT and TD-DFT calculations.

Table S4. Calculated major electronic transitions for $(\text{PTZ})_3$, $\text{PTZO}_2-(\text{PTZ})_2$, $\text{PTZ}-(\text{TPE})_2$, and $\text{PTZO}_2-(\text{TPE})_2$ in the gas phase.

Compounds	Wavelength (nm)	Composition	Assignment	f^a
$(\text{PTZ})_3$	406	HOMO→LUMO (0.70)	ICT	0.57
$\text{PTZO}_2-(\text{PTZ})_2$	339	HOMO-2→LUMO (0.70)	ICT	0.51
$\text{PTZ}-(\text{TPE})_2$	421	HOMO→LUMO (0.70)	ICT	0.59
$\text{PTZO}_2-(\text{TPE})_2$	401	HOMO→LUMO (0.70)	ICT	0.92

f^a = Oscillatory strength

Thermal properties



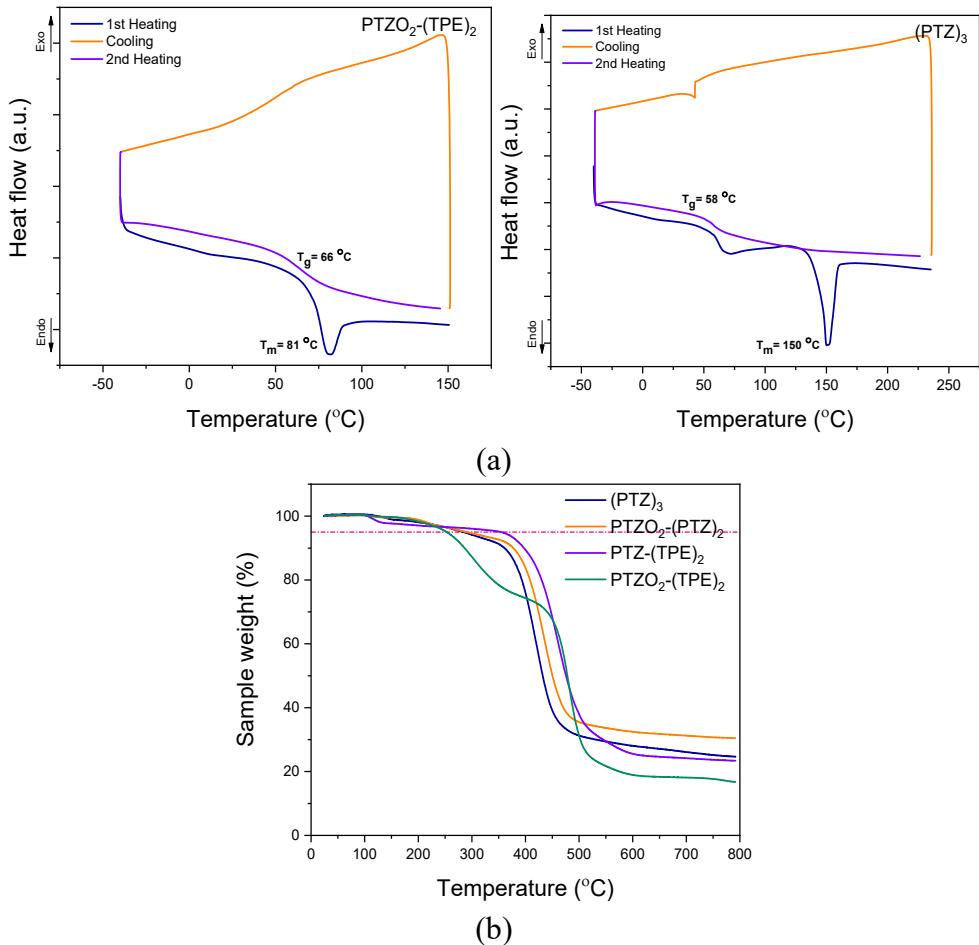


Fig. S8. DSC (a) and TGA (b) thermograms of **(PTZ)₃**, **PTZO₂-(PTZ)₂**, **PTZ-(TPE)₂**, and **PTZO₂-(TPE)₂** recorded at $10\text{ }^{\circ}\text{C min}^{-1}$.

Cyclic Voltammetry (CV)

The potentials of reduction and oxidation processes were obtained relative to an internal ferrocene reference (Fc/Fc^+) (Fig. S10). The IP_{CV} vs Fc values were calculated using equation (1) where the values are 4.98, 5.0, 5.05, and 5.47 eV (vs Fc/Fc^+), respectively (Table 3). The values of **(PTZ)₃** and **PTZO₂-(PTZ)₂** are almost equal because of the PTZ donor moiety.

$$\text{IP}_{\text{CV}} = -e(E_{\text{ox}} + 4.8) \quad (1)$$

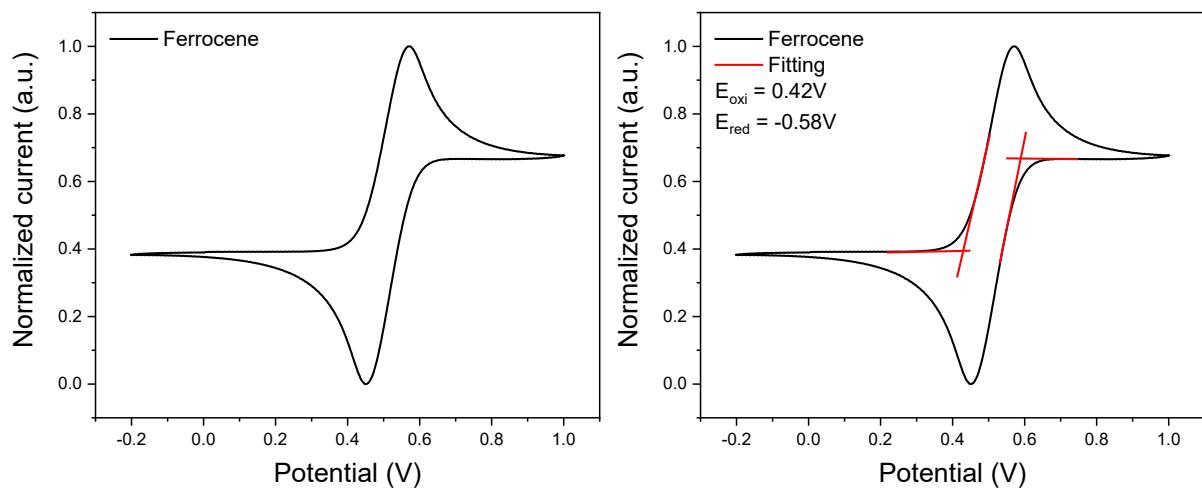
The CV spectra of the compounds indicates that only **PTZO₂-(PTZ)₂** and **PTZO₂-(TPE)₂** have reduction potential (E_{red}) which is due to the D-A-D structure containing the electron

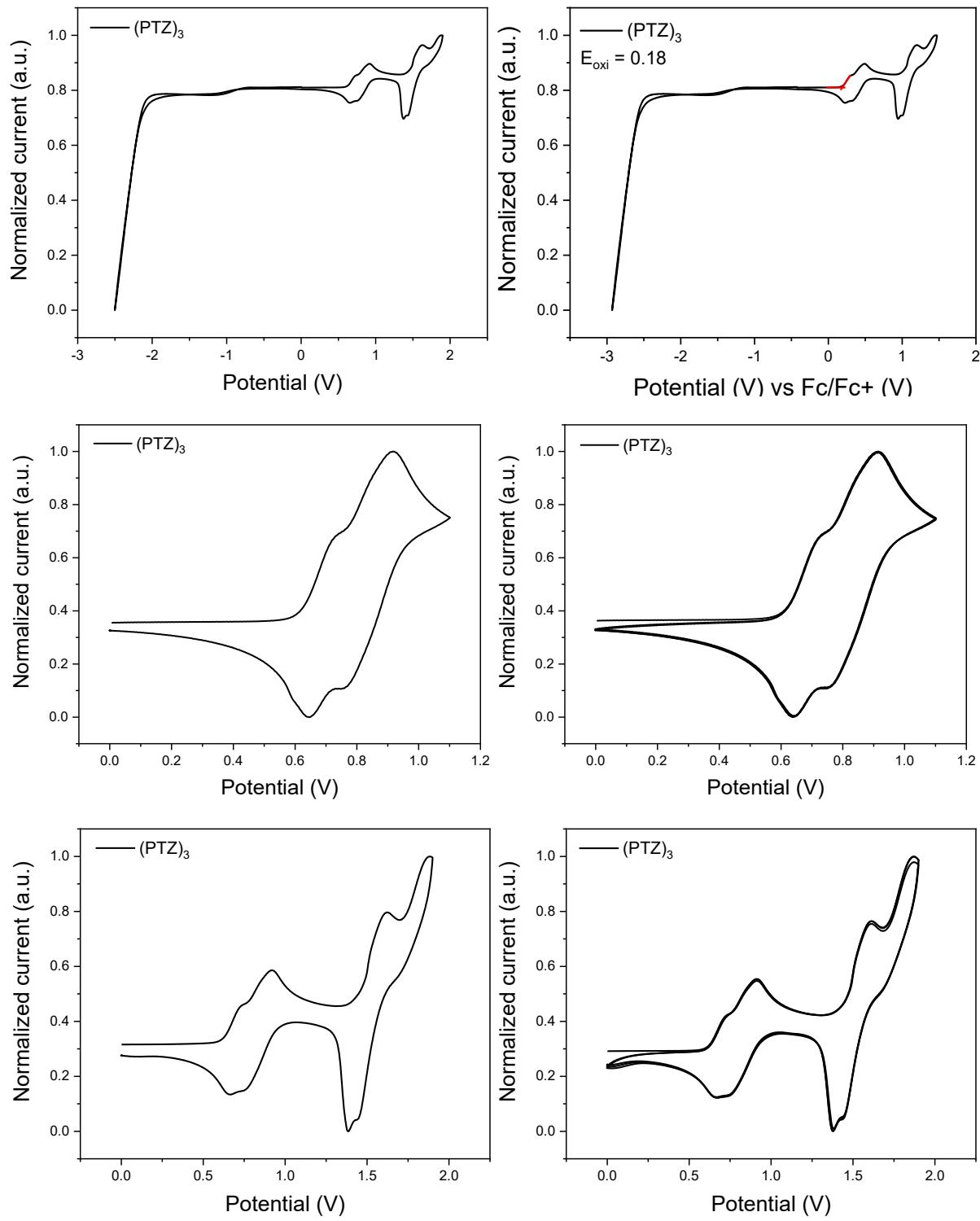
accepting moiety PTZ dioxide. The EA_{CV} values were calculated using Equation (2) and are 3.40eV and 3.52eV, respectively (Table 3).

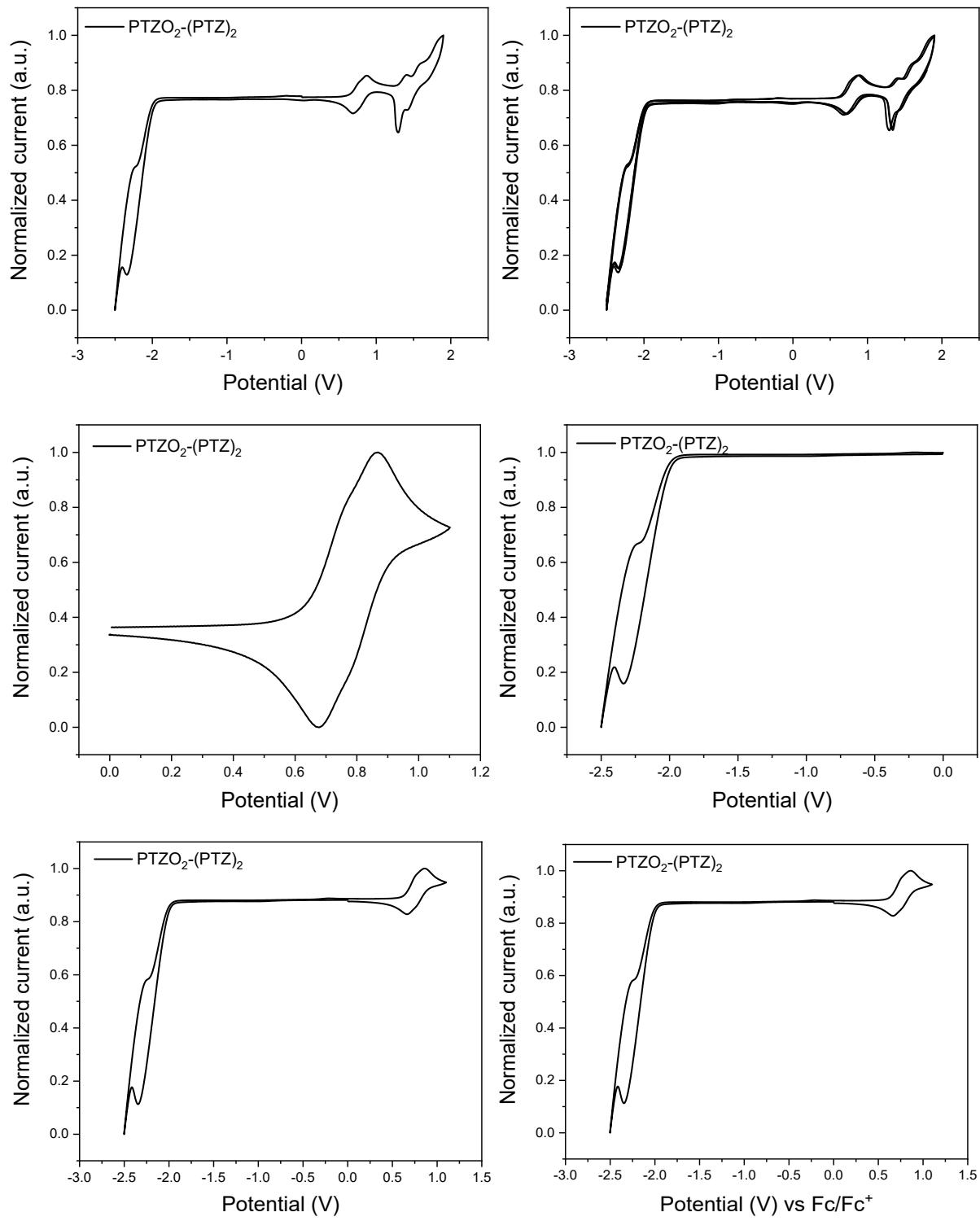
$$EA_{CV} = -e(E_{red} + 4.8) \quad (2)$$

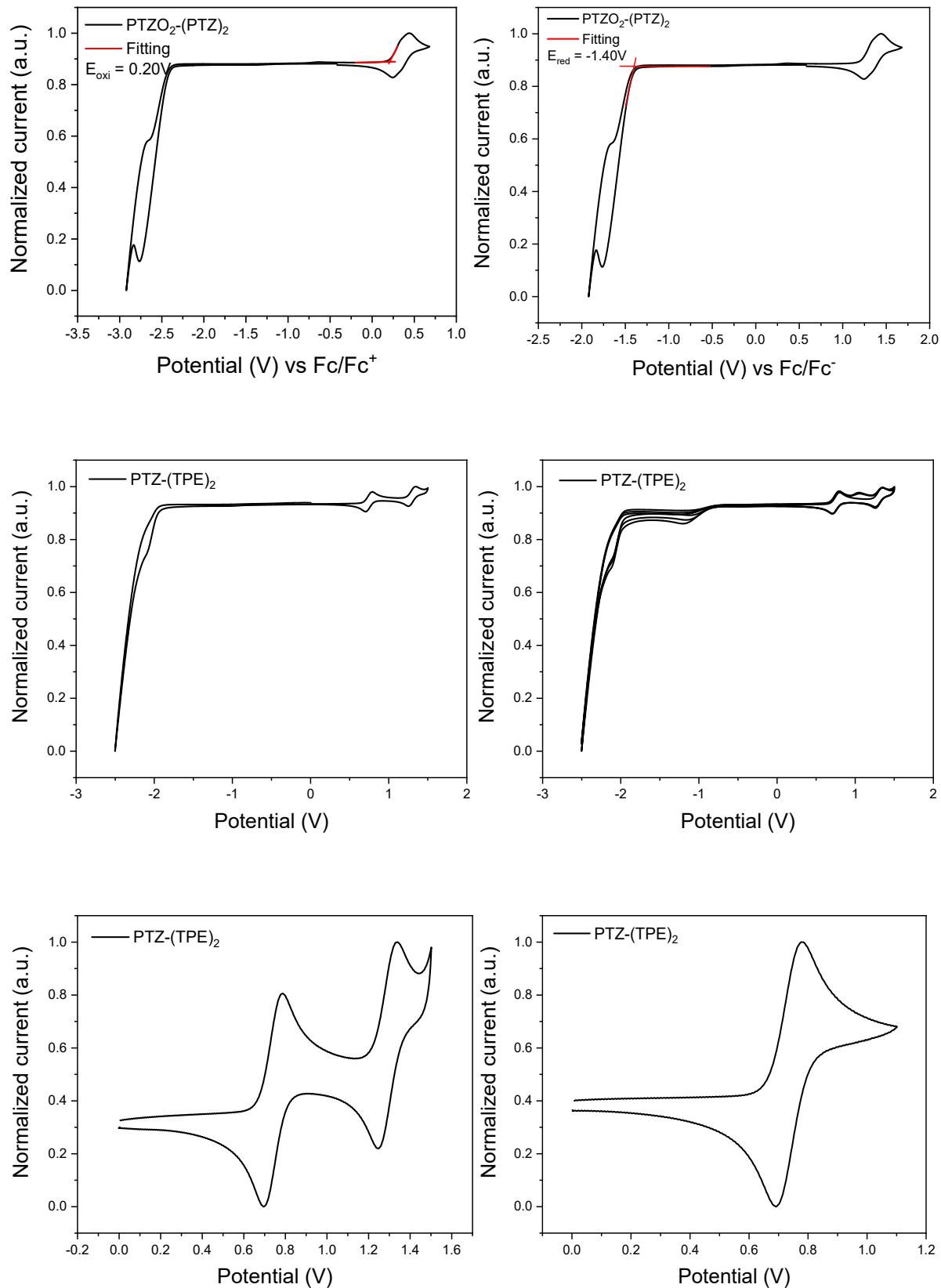
By subtracting IP_{CV} from EA_{CV} it is possible to calculate the gap (E_{gCV}). The CV spectra of all the compounds were ran for three cycles, the results indicate reversibility and the stability of the samples in solution.

Employing different methods of IP measurements caused small differences between the obtained values that can be explained by different environments in the solution and the solid state. Such small margin of error indicates collaborative electron-donor effect of PTZ donor moiety on electron releasing energy. The absorption band gap of the derivatives has linear effect on the IP values.









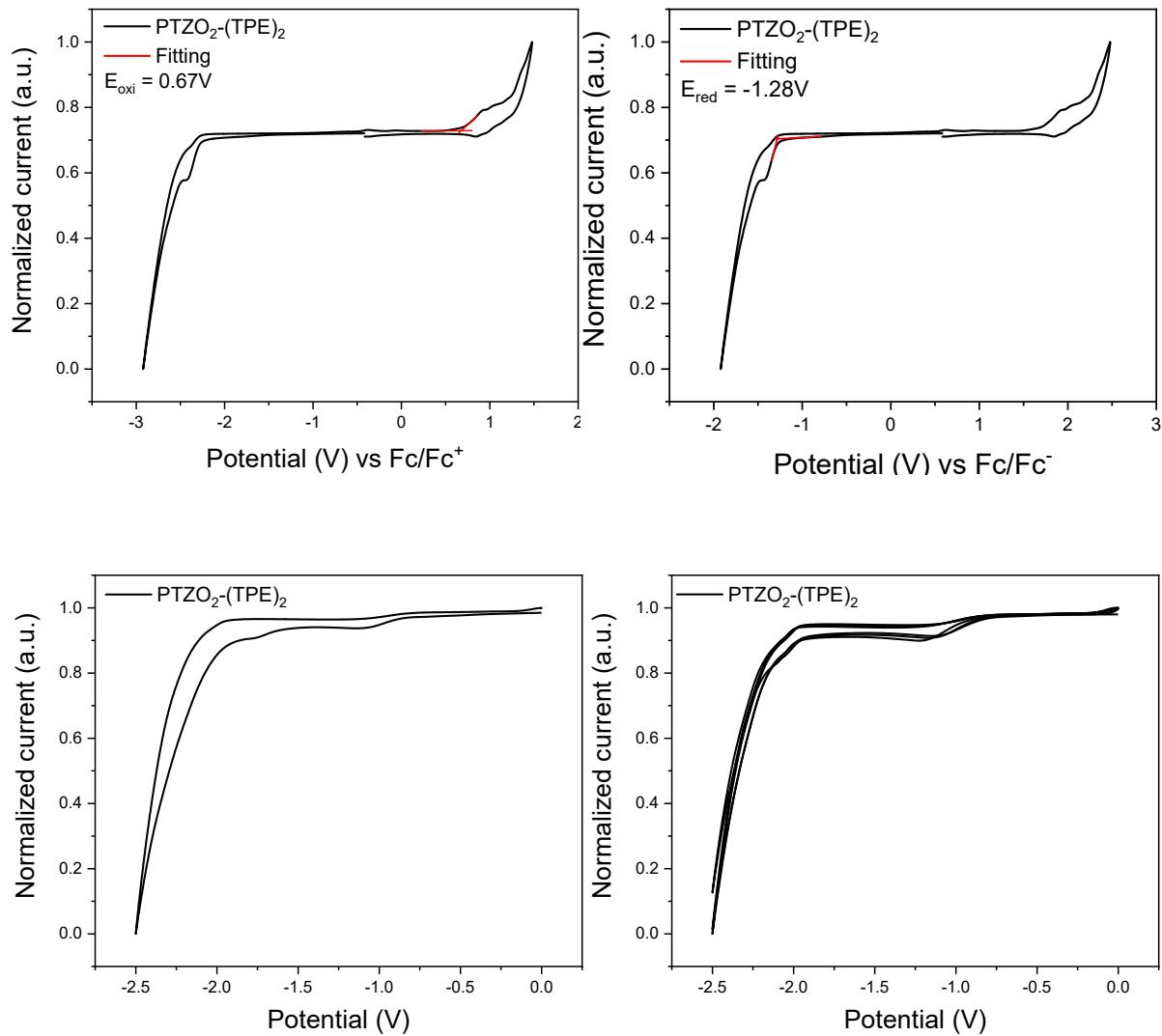


Fig. S9 Cyclic voltammograms of the Ferrocene and the dilute solutions of $(\text{PTZ})_3$, $\text{PTZO}_2\text{-}(\text{PTZ})_2$, $\text{PTZ}\text{-}(\text{TPE})_2$, and $\text{PTZO}_2\text{-}(\text{TPE})_2$ in dichloromethane recorded as sweep rate of 100 mV/s.

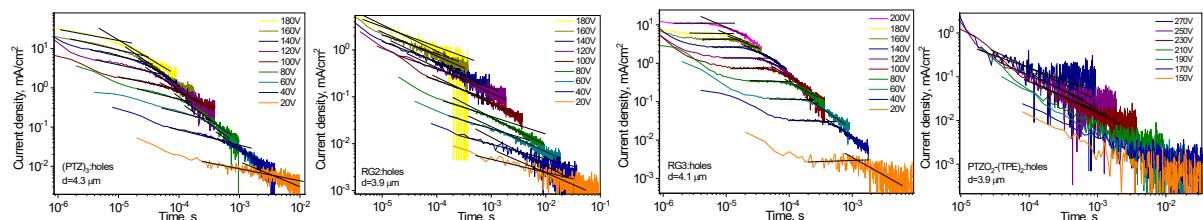


Fig. S10 Time of flight (TOF) current transients of holes for deposited layers of $(\text{PTZ})_3$, $\text{PTZO}_2\text{-}(\text{PTZ})_2$, $\text{PTZ}\text{-}(\text{TPE})_2$, and $\text{PTZO}_2\text{-}(\text{TPE})_2$.

Device

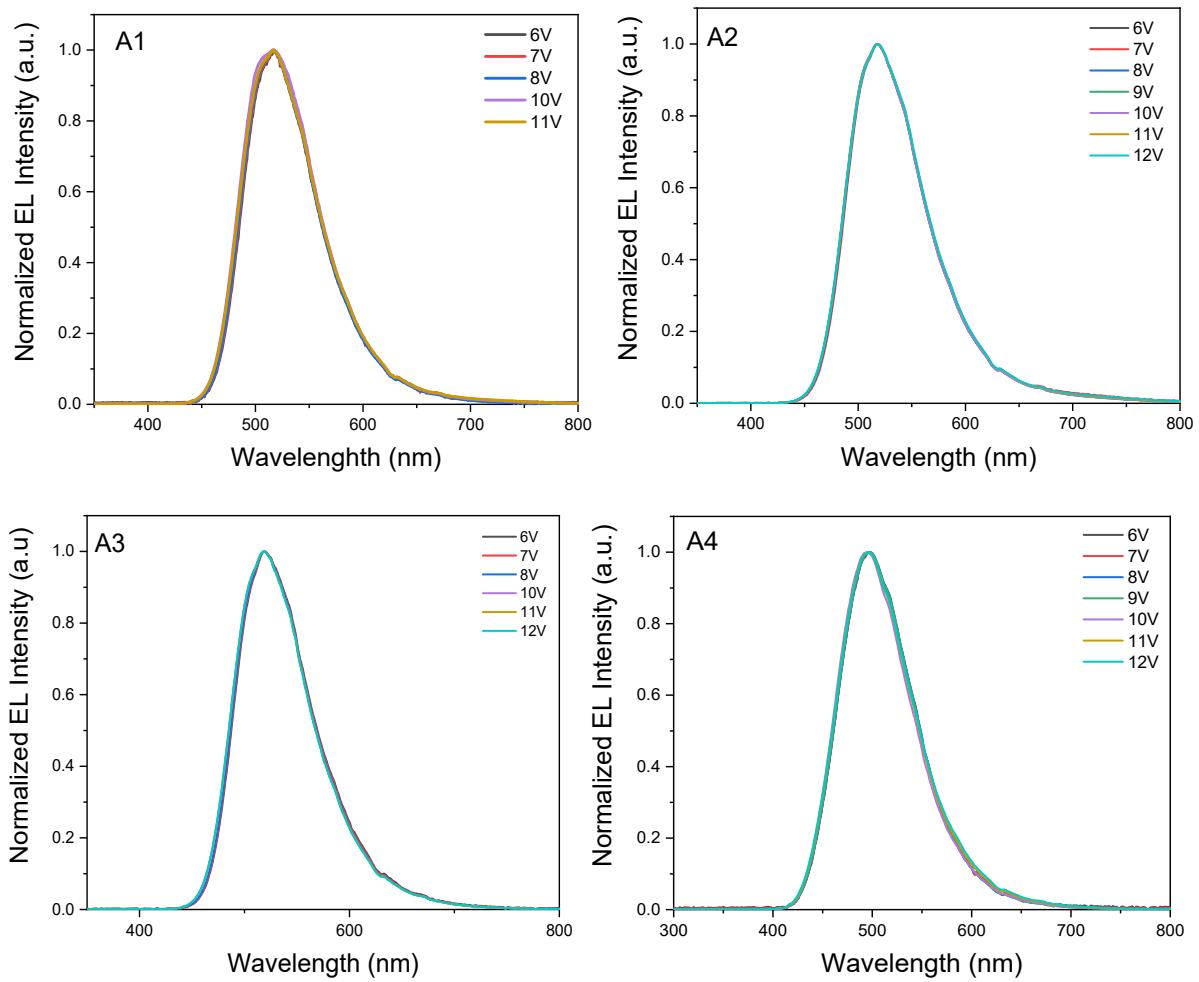


Fig. S11 EL spectra of devices A1-A4 at different voltages.

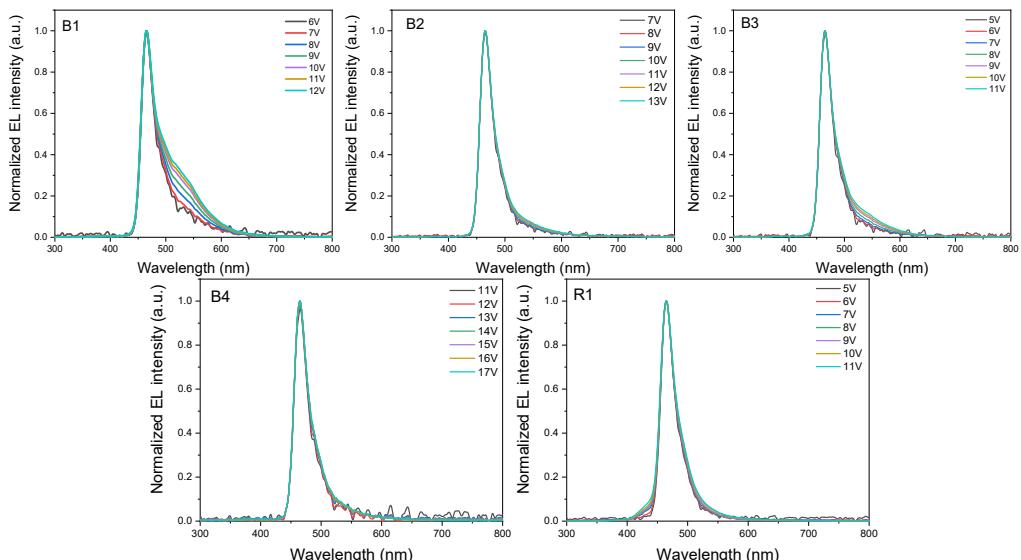


Fig. S12 EL spectra of devices B1-B4 and R1 at different voltages.

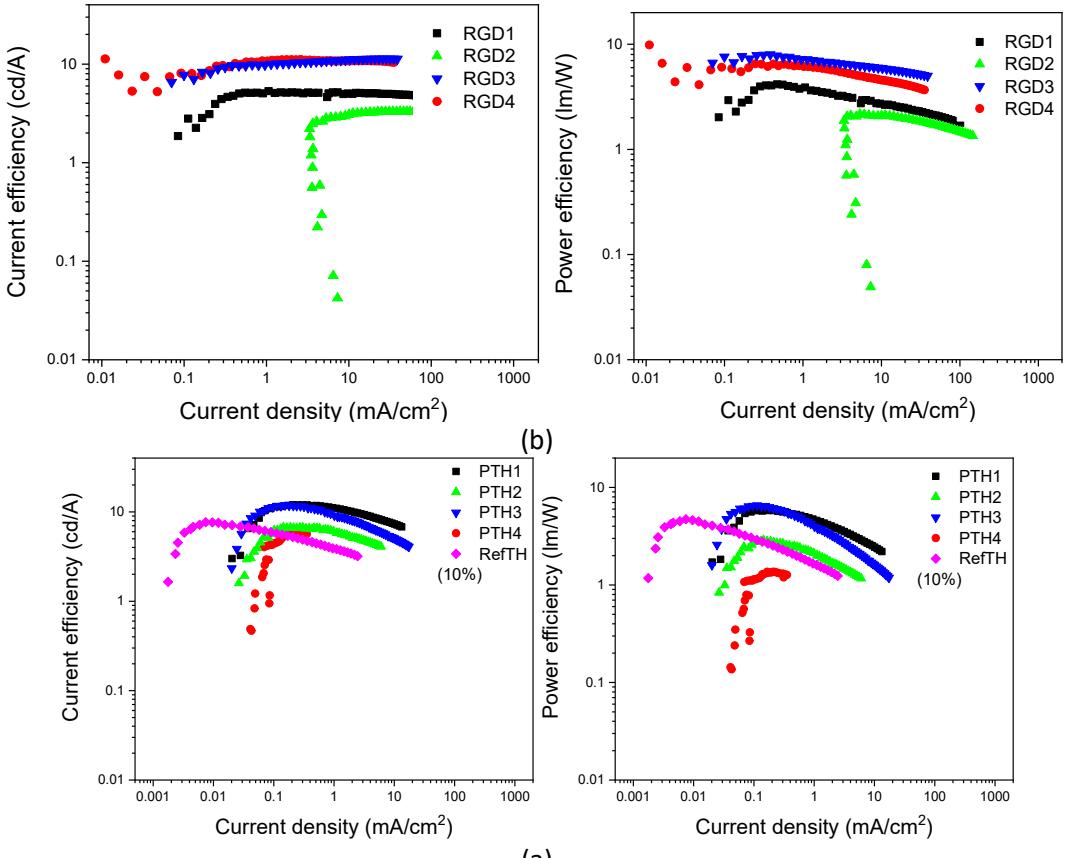


Fig. S13 Current and power efficiencies of devices A1-A4 (a) and B1-B4, R1 (b).

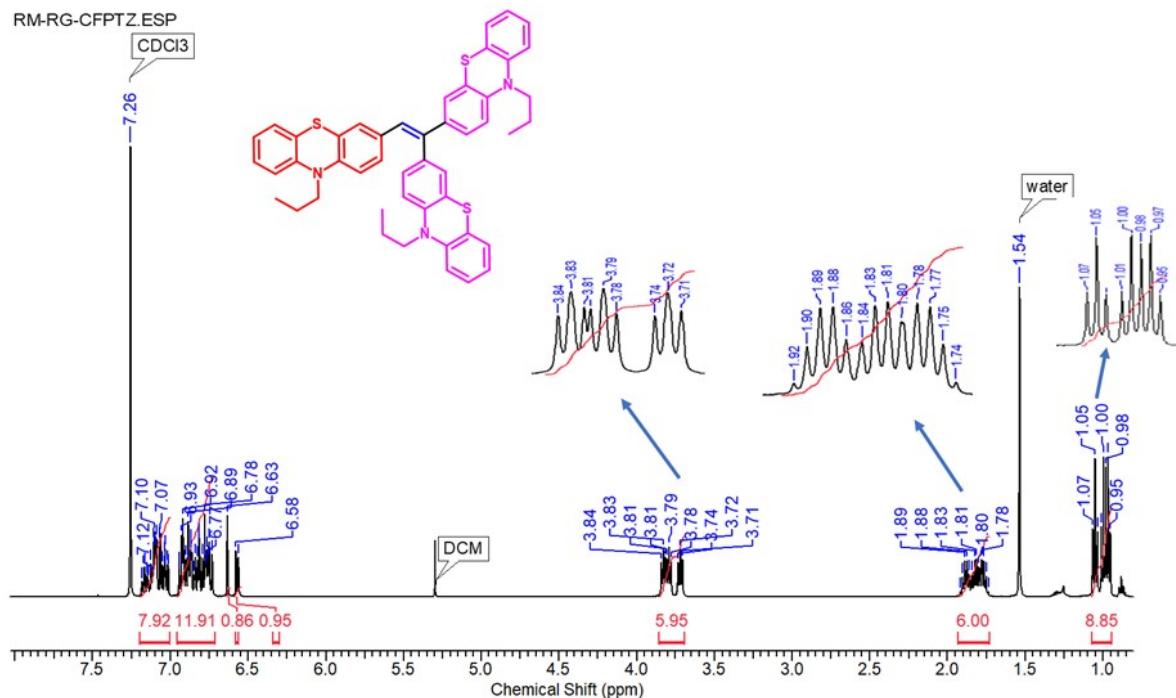


Fig. S14 ^1H NMR spectra of $(\text{PTZ})_3$ (CDCl_3 , 500 MHz).

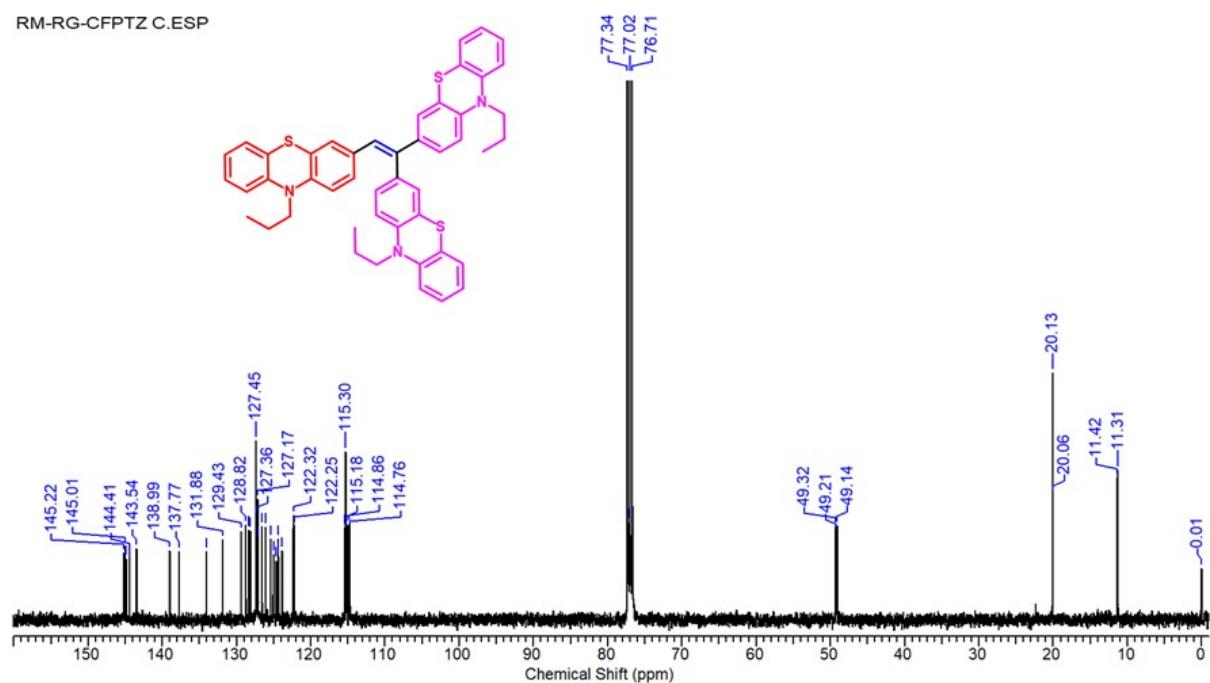


Fig. S15. $^{13}\text{C} \{^1\text{H}\}$ NMR spectra of $(\text{PTZ})_3$ (CDCl_3 , 126 MHz).

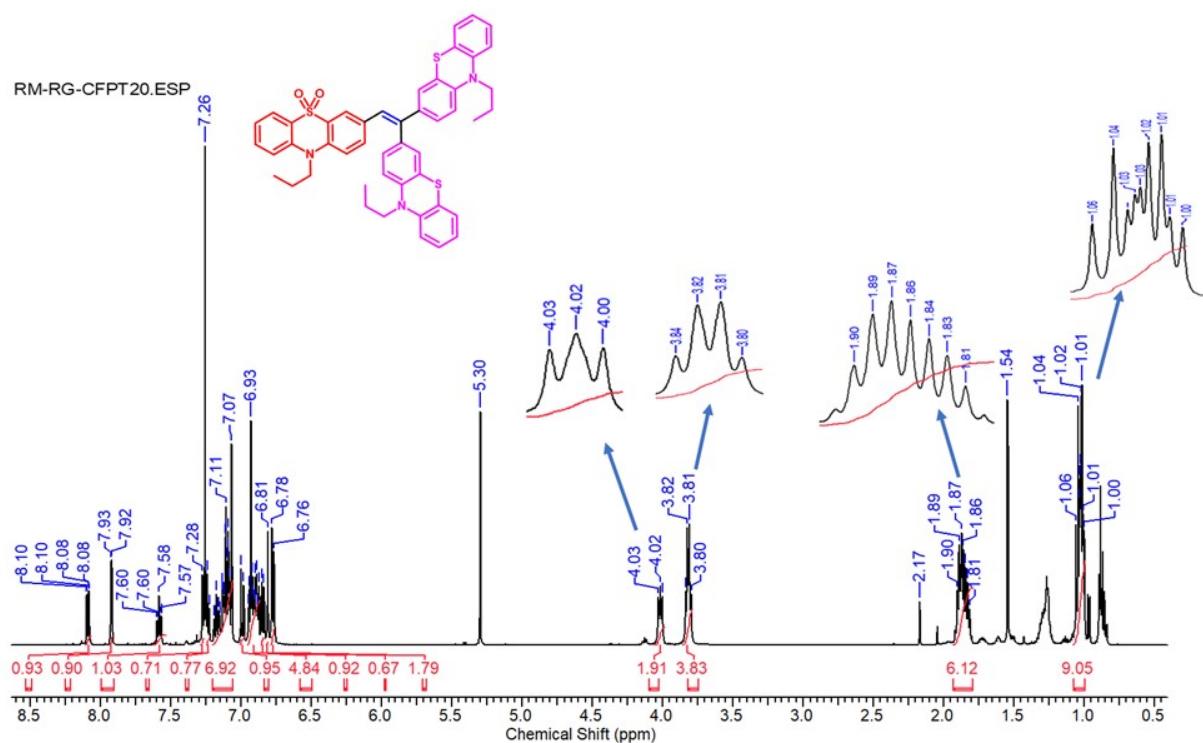


Fig. S16. ^1H NMR spectra of $\text{PTZO}_2\text{-}(\text{PTZ})_2$ (CDCl_3 , 500 MHz).

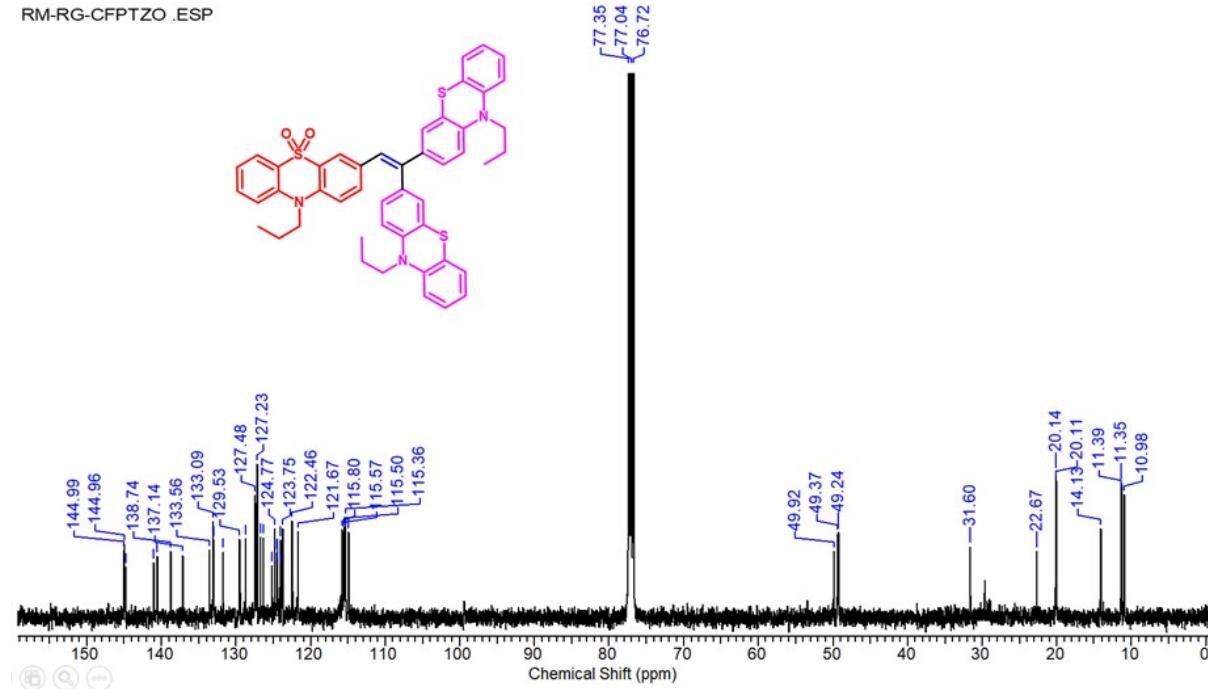


Fig. S17. ^{13}C $\{{}^1\text{H}\}$ NMR spectra of $\text{PTZO}_2\text{-}(\text{PTZ})_2$ (CDCl_3 , 126 MHz).

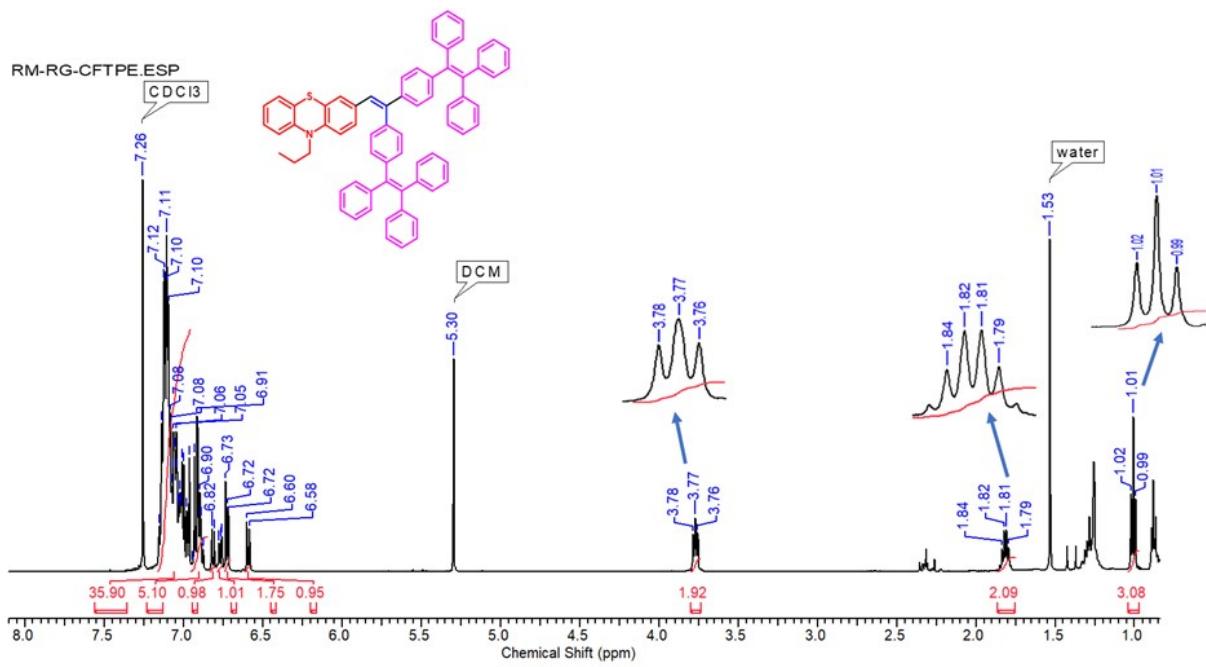


Fig. S18. ¹H NMR spectra of PTZ-(TPE)₂ (CDCl₃, 500 MHz).

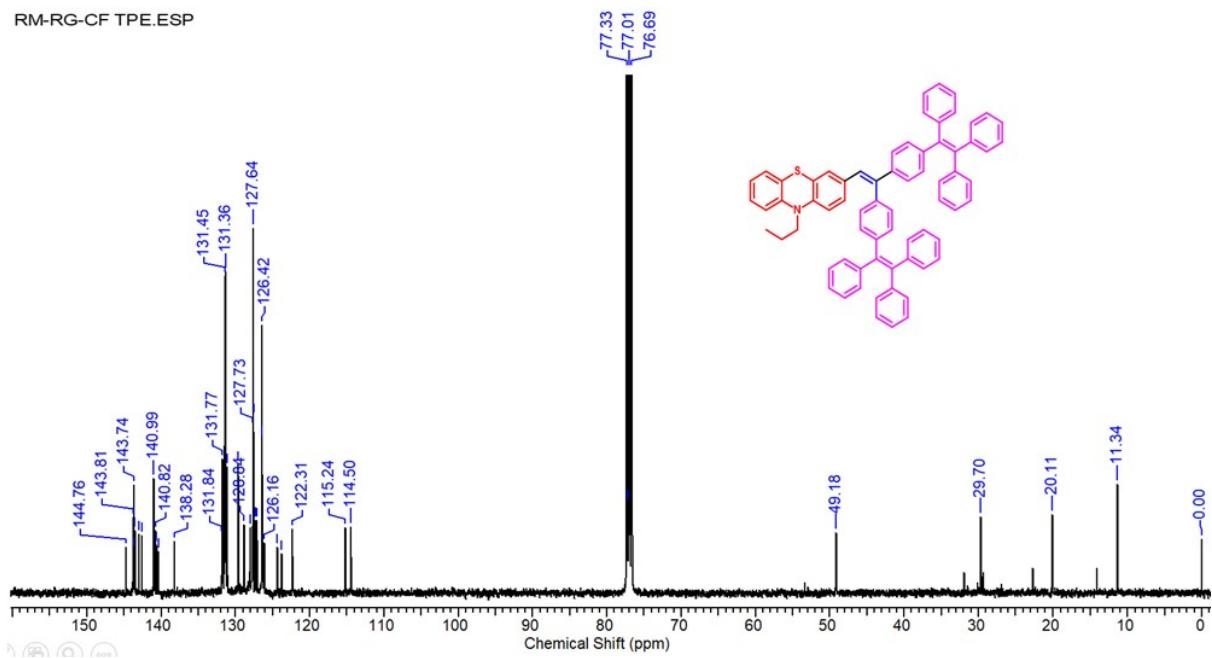


Fig. S19. ¹³C {¹H} NMR spectra of PTZ-(TPE)₂ (CDCl₃, 126 MHz).

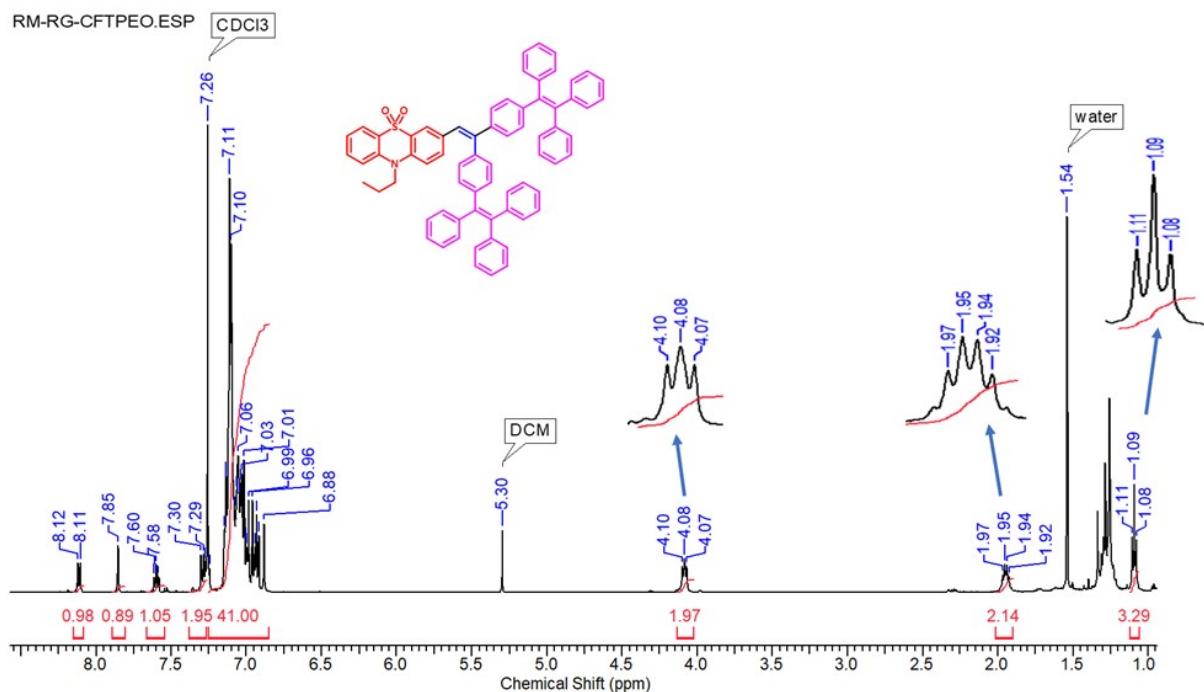


Fig. S20. ^1H NMR spectra of PTZO₂-(TPE)₂ (CDCl₃, 500 MHz).

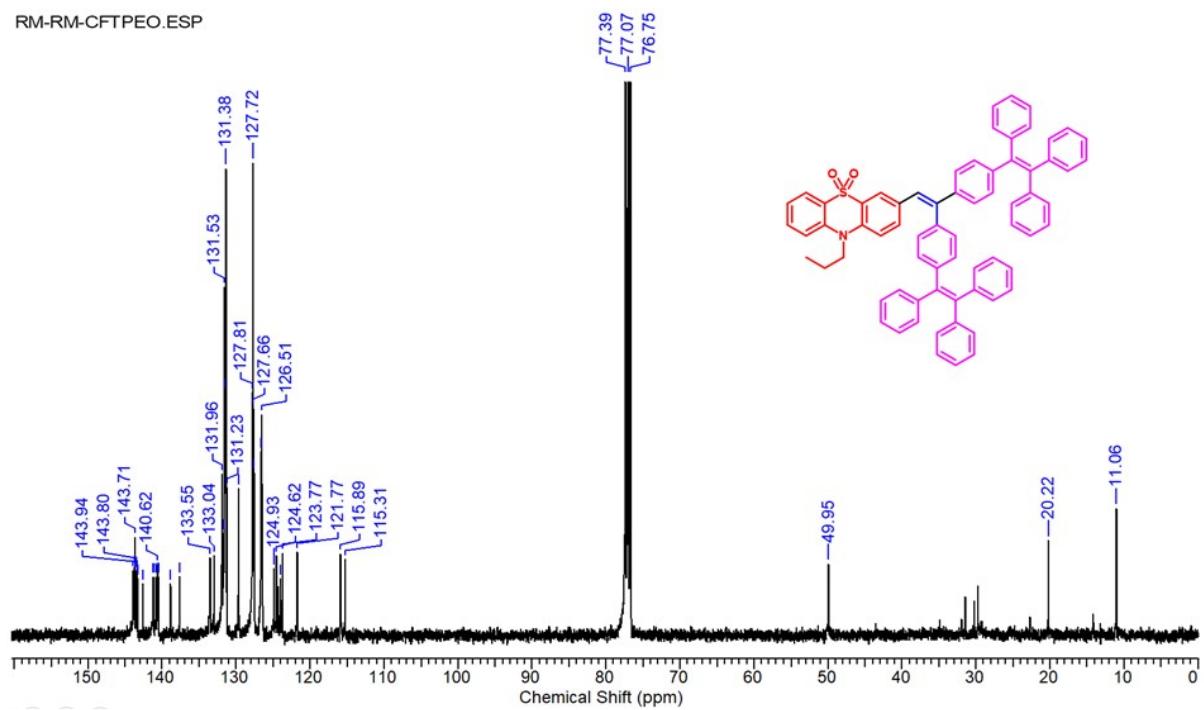


Fig. S21. ^{13}C { ^1H } NMR spectra of PTZO₂-(TPE)₂ (CDCl₃, 126 MHz).

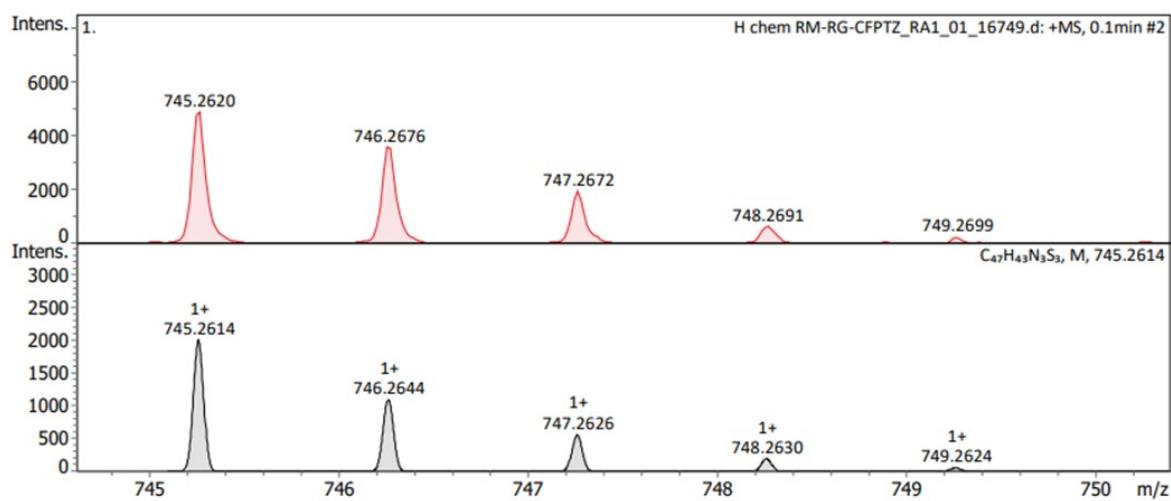


Fig. S22. HRMS of $(\text{PTZ})_3$.

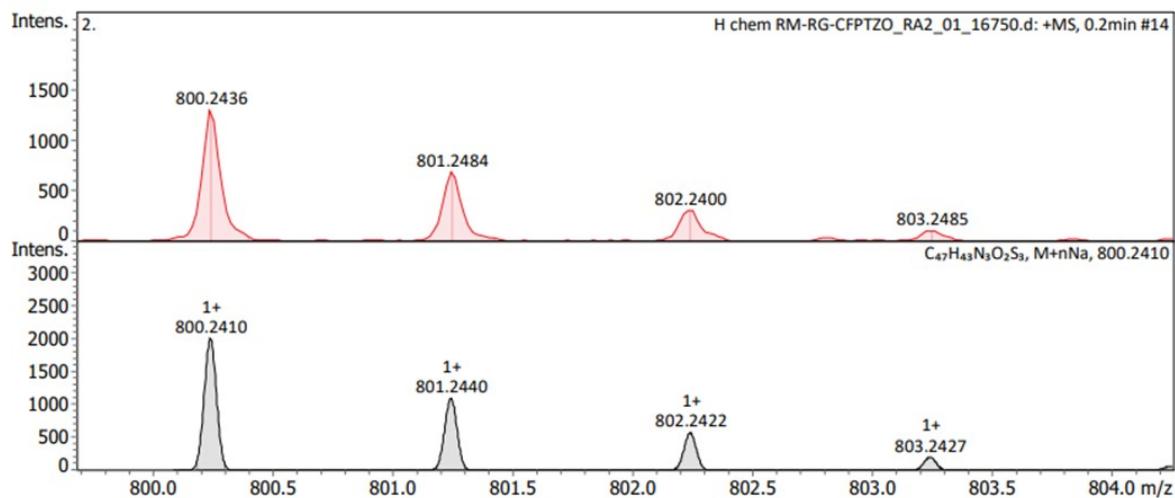


Fig. S23. HRMS of $\text{PTZO}_2\text{-(PTZ)}_2$.

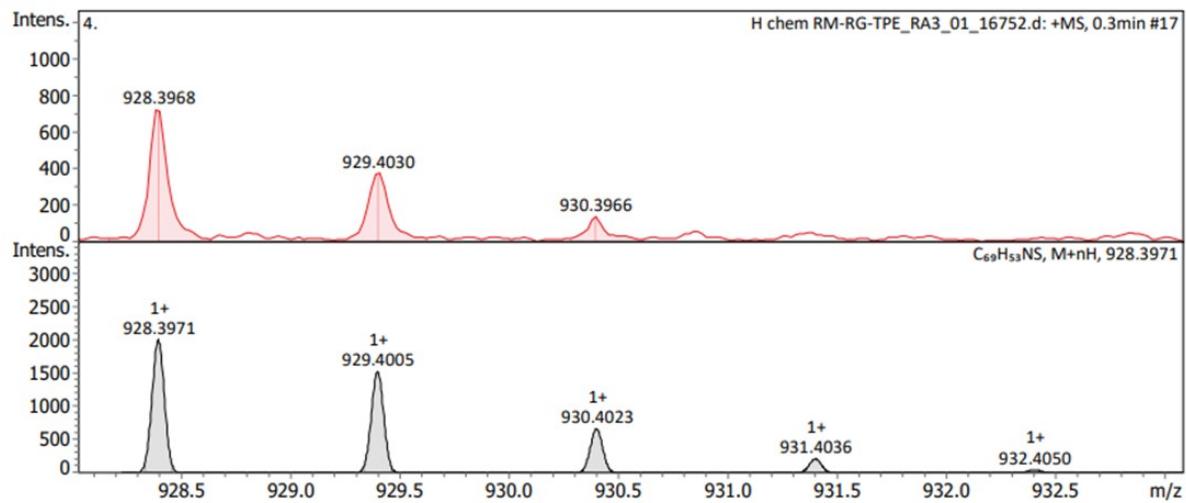


Fig. S24. HRMS of PTZ-(TPE)_2 .

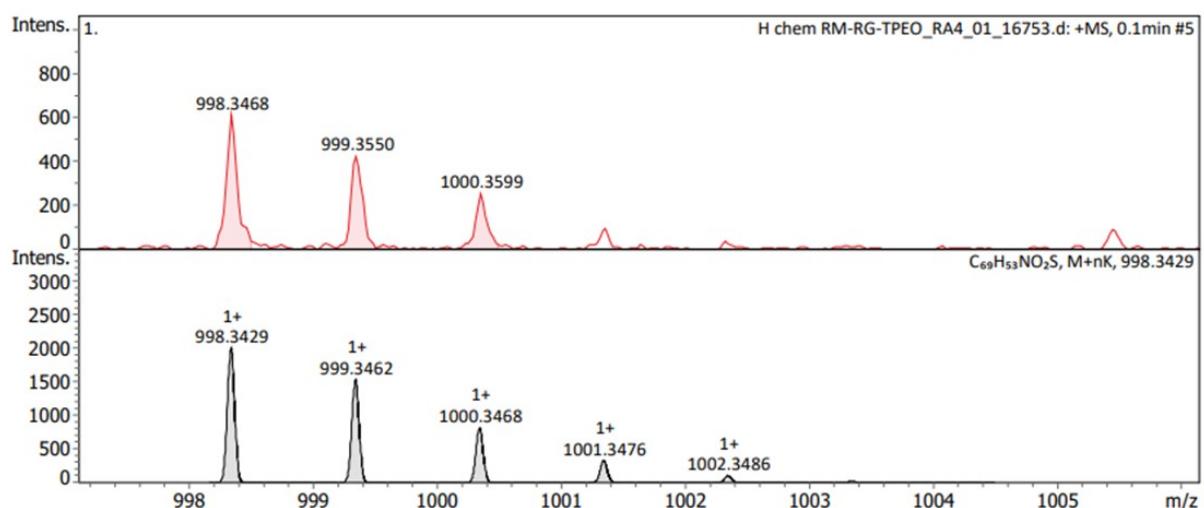


Fig. S25. HRMS of $\text{PTZO}_2\text{-(TPE)}_2$.

DFT

(PTZ)₃

Standard orientation:

Center	Atomic Number	Atomic Number	Type	Coordinates (Angstroms)		
	X	Y	Z			
1	6	0	7.447878	-5.844508	0.892078	
2	6	0	6.371145	-5.659044	0.024154	
3	6	0	8.013233	-4.738764	1.524132	
4	6	0	5.888922	-4.376625	-0.237395	
5	1	0	5.906248	-6.508359	-0.467121	
6	6	0	7.510020	-3.457538	1.295624	
7	1	0	8.843428	-4.867439	2.212193	
8	6	0	6.450027	-3.252123	0.399204	
9	1	0	7.948493	-2.614970	1.817769	
10	7	0	5.924804	-1.963828	0.129746	
11	6	0	3.753004	-2.784092	-0.620032	
12	6	0	4.519460	-1.818156	0.059019	
13	6	0	6.760549	-0.783725	0.363217	
14	6	0	2.368977	-2.670831	-0.697318	
15	6	0	3.833742	-0.751848	0.665017	
16	1	0	7.036177	-0.677900	1.424708	
17	6	0	8.015749	-0.757471	-0.520304	
18	1	0	6.153803	0.088884	0.114645	
19	1	0	1.807834	-3.446912	-1.210518	
20	6	0	2.454569	-0.618749	0.545381	
21	1	0	4.378983	-0.024673	1.256397	

22	1	0	7.697217	-0.805776	-1.567741
23	1	0	8.623028	-1.649693	-0.339726
24	1	0	1.965530	0.215121	1.034208
25	6	0	1.684898	-1.567535	-0.151994
26	6	0	8.856863	0.499217	-0.276806
27	1	0	9.739126	0.511692	-0.923547
28	1	0	8.284465	1.411595	-0.479986
29	1	0	9.207023	0.553178	0.760421
30	16	0	4.594273	-4.127493	-1.438532
31	6	0	0.222350	-1.530848	-0.269155
32	6	0	-0.645559	-0.485010	-0.229478
33	1	0	-0.223570	-2.514242	-0.407775
34	1	0	7.833127	-6.842598	1.074619
35	6	0	-0.220087	0.943427	-0.266621
36	6	0	-0.704168	1.867171	0.674360
37	6	0	0.629602	1.424480	-1.272836
38	6	0	-0.296690	3.198857	0.653142
39	1	0	-1.391478	1.537871	1.447301
40	6	0	1.003984	2.764868	-1.325948
41	1	0	0.986022	0.741892	-2.037133
42	6	0	0.575113	3.674181	-0.346007
43	16	0	-0.820225	4.320173	1.938517
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46	6	0	-0.964870	5.791328	0.940259
47	6	0	-0.021157	6.013010	-0.081715
48	6	0	2.210523	5.424486	-1.011429
49	6	0	-1.964035	6.728258	1.204041
50	6	0	-0.110258	7.202081	-0.821390

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53	1	0	2.809467	4.518841	-1.123341
54	6	0	-2.019021	7.922107	0.483667
55	1	0	-2.688402	6.521958	1.986086
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65	1	0	4.142134	7.266496	-1.899804
66	6	0	-2.108095	-0.750354	-0.202674
67	6	0	-2.635419	-1.869347	0.467172
68	6	0	-3.018185	0.087689	-0.867348
69	6	0	-4.001963	-2.128570	0.483046
70	1	0	-1.972114	-2.527037	1.019469
71	6	0	-4.382699	-0.187435	-0.885710
72	1	0	-2.649879	0.952031	-1.409559
73	6	0	-4.909321	-1.288612	-0.192705
74	16	0	-4.645796	-3.484643	1.447739
75	1	0	-5.038643	0.458271	-1.458309
76	7	0	-6.292621	-1.578196	-0.148829
77	6	0	-5.981903	-3.944742	0.359389
78	6	0	-6.695688	-2.927905	-0.304552
79	6	0	-7.262017	-0.505751	-0.386103

80	6	0	-6.343287	-5.283670	0.210098
81	6	0	-7.781106	-3.297582	-1.113230
82	1	0	-7.610238	-0.490060	-1.431425
83	6	0	-8.458555	-0.554422	0.574700
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85	6	0	-7.445972	-5.633590	-0.570191
86	1	0	-5.763469	-6.047391	0.719398
87	6	0	-8.160856	-4.635240	-1.229224
88	1	0	-8.335647	-2.538975	-1.653475
89	1	0	-8.074983	-0.515984	1.600454
90	1	0	-8.985623	-1.508384	0.476089
91	6	0	-9.431485	0.602097	0.326233
92	1	0	-7.735686	-6.675220	-0.665178
93	1	0	-9.013401	-4.892691	-1.850603
94	1	0	-10.272825	0.562658	1.024296
95	1	0	-8.941025	1.573900	0.453028
96	1	0	-9.842860	0.569267	-0.689186

Rotational constants (GHZ): 0.0413659 0.0266382 0.0168548

PTZO₂-(PTZ)₂

Standard orientation:

Center Atomic Atomic Coordinates (Angstroms)

Number	Number	Type	X	Y	Z
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2      6      0      6.572722 -5.050012  0.053635

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6	6	0	7.580290	-2.869510	1.484014
7	1	0	9.027937	-4.246114	2.246757
8	6	0	6.461687	-2.671125	0.651374
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10	7	0	5.801529	-1.431093	0.586694
11	6	0	3.746301	-2.345870	-0.397522
12	6	0	4.412532	-1.370954	0.369732
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14	6	0	2.362252	-2.360189	-0.547509
15	6	0	3.602096	-0.385573	0.969633
16	1	0	6.917723	-0.196160	1.911505
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18	1	0	5.859302	0.632559	0.786984
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21	1	0	4.041504	0.351046	1.631852
22	1	0	7.300797	0.043316	-1.124554
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53	1	0	1.626583	5.916581	-2.043942
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55	1	0	2.438528	4.708985	-1.062180
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70	6	0	-3.198449	0.013330	-0.847929
71	6	0	-4.101274	-2.210346	0.547976
72	1	0	-2.063896	-2.500560	1.124050
73	6	0	-4.547883	-0.325925	-0.880490
74	1	0	-2.861158	0.878820	-1.408040
75	6	0	-5.035217	-1.430497	-0.163894
76	16	0	-4.697228	-3.563743	1.545661
77	1	0	-5.222855	0.272398	-1.481557
78	7	0	-6.404242	-1.777666	-0.129292
79	6	0	-6.001965	-4.112224	0.460297
80	6	0	-6.749898	-3.147886	-0.242573
81	6	0	-7.416159	-0.755392	-0.408981
82	6	0	-6.306125	-5.469148	0.351703
83	6	0	-7.811322	-3.587969	-1.047702
84	1	0	-7.755835	-0.789770	-1.456445
85	6	0	-8.617142	-0.822552	0.545032
86	1	0	-6.933755	0.214182	-0.273536

87	6	0	-7.385675	-5.889412	-0.426055
88	1	0	-5.700134	-6.191294	0.890269
89	6	0	-8.134675	-4.943238	-1.123022
90	1	0	-8.391346	-2.871011	-1.617301
91	1	0	-8.243730	-0.733965	1.571448
92	1	0	-9.102821	-1.800647	0.474900
93	6	0	-9.635341	0.283386	0.251776
94	1	0	-7.631055	-6.944673	-0.489813
95	1	0	-8.969899	-5.255560	-1.742700
96	1	0	-10.479577	0.231885	0.945506
97	1	0	-9.186635	1.278482	0.349169
98	1	0	-10.037301	0.199864	-0.764498

Rotational constants (GHZ): 0.0399669 0.0253272 0.0162355

PTZ-(TPE)₂

Standard orientation:

Center	Atomic Number	Atomic Number	Type	Coordinates (Angstroms)		
	X	Y	Z			
1	6	0	6.704010	-8.133584	-0.018669	
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3	6	0	7.521366	-7.242736	0.674267	
4	6	0	5.378551	-6.269244	-0.811667	
5	1	0	4.962613	-8.319256	-1.282569	
6	6	0	7.261612	-5.872122	0.644006	

7	1	0	8.361431	-7.610435	1.255899
8	6	0	6.195741	-5.360033	-0.111702
9	1	0	7.895018	-5.199609	1.210894
10	7	0	5.912862	-3.973083	-0.178196
11	6	0	3.571993	-4.272561	-0.799222
12	6	0	4.562129	-3.562676	-0.093407
13	6	0	6.976707	-3.008297	0.110715
14	6	0	2.236269	-3.890918	-0.731698
15	6	0	4.146346	-2.474940	0.692869
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17	6	0	8.148439	-3.091853	-0.877937
18	1	0	6.534960	-2.013962	0.024785
19	1	0	1.496971	-4.473552	-1.274474
20	6	0	2.816695	-2.069146	0.719867
21	1	0	4.865481	-1.946136	1.308187
22	1	0	7.753446	-2.943858	-1.889335
23	1	0	8.588437	-4.093683	-0.860508
24	1	0	2.539301	-1.227894	1.343276
25	6	0	1.826645	-2.756340	-0.005286
26	6	0	9.225214	-2.047439	-0.569046
27	1	0	10.047501	-2.108685	-1.287889
28	1	0	8.821679	-1.029447	-0.613189
29	1	0	9.649458	-2.191834	0.431243
30	16	0	4.068422	-5.633047	-1.841047
31	6	0	0.398588	-2.419862	0.014816
32	6	0	-0.226216	-1.235560	0.247026
33	1	0	-0.250284	-3.268914	-0.193255
34	1	0	6.900134	-9.200650	0.009801

35	6	0	0.486900	0.071289	0.352011
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37	6	0	1.320060	0.528647	-0.681730
38	6	0	0.969123	2.124835	1.567047
39	1	0	-0.336245	0.575253	2.279441
40	6	0	1.955179	1.763810	-0.598040
41	1	0	1.467690	-0.094906	-1.557806
42	6	0	1.789127	2.592118	0.524901
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46	6	0	2.493949	4.833007	-0.359663
47	6	0	3.208258	4.113981	1.939019
48	6	0	3.406156	6.017131	-0.364444
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53	6	0	4.787037	5.874831	-0.146595
54	6	0	2.067087	4.926743	-2.841002
55	6	0	0.195631	4.539835	-1.369959
56	6	0	4.674953	3.284268	3.698491
57	1	0	4.163392	2.182423	1.922663
58	6	0	3.664943	5.463837	3.913346
59	1	0	2.385562	6.072490	2.294758
60	6	0	3.754977	8.407546	-0.678216
61	1	0	1.846946	7.428006	-0.839691
62	6	0	5.636098	6.979222	-0.189678

63	1	0	5.192131	4.888773	0.055393
64	6	0	1.214915	4.866807	-3.942237
65	1	0	3.127386	5.109024	-2.988223
66	6	0	-0.658674	4.491998	-2.470211
67	1	0	-0.204571	4.418983	-0.368684
68	6	0	4.494705	4.463818	4.423350
69	1	0	5.315310	2.497970	4.088314
70	1	0	3.507809	6.380256	4.475030
71	6	0	5.123547	8.251328	-0.449905
72	1	0	3.346118	9.392287	-0.886206
73	1	0	6.701191	6.844462	-0.023831
74	6	0	-0.152228	4.648740	-3.761668
75	1	0	1.618843	4.994767	-4.942531
76	1	0	-1.723007	4.336913	-2.317759
77	1	0	4.990669	4.599309	5.380058
78	1	0	5.785360	9.111926	-0.481999
79	1	0	-0.817915	4.608701	-4.618873
80	6	0	-1.708730	-1.198469	0.354541
81	6	0	-2.431045	-2.256755	0.936836
82	6	0	-2.444907	-0.098892	-0.124676
83	6	0	-3.819026	-2.226821	1.014283
84	1	0	-1.892406	-3.097585	1.362716
85	6	0	-3.833328	-0.076694	-0.060208
86	1	0	-1.919043	0.743270	-0.562569
87	6	0	-4.555034	-1.144732	0.500062
88	1	0	-4.345423	-3.049272	1.489057
89	1	0	-4.372455	0.780839	-0.448840
90	6	0	-6.042943	-1.108161	0.617238

91	6	0	-6.858529	-0.769957	-0.426365
92	6	0	-6.572660	-1.472558	1.967664
93	6	0	-8.313467	-0.475214	-0.249849
94	6	0	-6.367806	-0.668916	-1.834632
95	6	0	-6.020236	-0.903778	3.128710
96	6	0	-7.589414	-2.430439	2.118014
97	6	0	-9.268214	-1.056377	-1.102934
98	6	0	-8.761580	0.428149	0.728602
99	6	0	-6.708612	0.438144	-2.631824
100	6	0	-5.606606	-1.696114	-2.417843
101	6	0	-6.490142	-1.255767	4.392707
102	1	0	-5.219213	-0.176855	3.033420
103	6	0	-8.051465	-2.791625	3.382446
104	1	0	-8.015553	-2.893118	1.233994
105	6	0	-10.624793	-0.770648	-0.960607
106	1	0	-8.939003	-1.740685	-1.879075
107	6	0	-10.117027	0.723449	0.863395
108	1	0	-8.037373	0.900203	1.384184
109	6	0	-6.275322	0.530429	-3.953223
110	1	0	-7.314097	1.232466	-2.205753
111	6	0	-5.182939	-1.609508	-3.742714
112	1	0	-5.349287	-2.566816	-1.823854
113	6	0	-7.508133	-2.202154	4.525171
114	1	0	-6.057300	-0.793821	5.275575
115	1	0	-8.835001	-3.538372	3.474679
116	6	0	-11.055453	0.121240	0.023534
117	1	0	-11.346094	-1.241703	-1.622445
118	1	0	-10.439770	1.428728	1.623995

119	6	0	-5.510122	-0.493474	-4.514845
120	1	0	-6.539954	1.401197	-4.546514
121	1	0	-4.599210	-2.417943	-4.173632
122	1	0	-7.869572	-2.482703	5.510228
123	1	0	-12.111813	0.350208	0.130130
124	1	0	-5.178239	-0.425774	-5.546794

Rotational constants (GHZ): 0.0276765 0.0165274 0.0121838

PTZO₂-(TPE)₂

Standard orientation:

Center	Atomic Number	Atomic Number	Coordinates (Angstroms)		
		Type	X	Y	Z

1	6	0	6.971095	-7.624615	0.203597
2	6	0	5.860741	-7.159949	-0.491773
3	6	0	7.773094	-6.708365	0.891233
4	6	0	5.568353	-5.796142	-0.494809
5	1	0	5.213550	-7.830702	-1.047227
6	6	0	7.465127	-5.351720	0.913031
7	1	0	8.643690	-7.056576	1.439216
8	6	0	6.337698	-4.860838	0.226147
9	1	0	8.087694	-4.676991	1.487899
10	7	0	5.962362	-3.507397	0.298385
11	6	0	3.670198	-3.887708	-0.495487

12	6	0	4.606909	-3.136235	0.239934
13	6	0	6.994503	-2.488864	0.548366
14	6	0	2.310173	-3.592417	-0.492948
15	6	0	4.096227	-2.046104	0.973890
16	1	0	7.462902	-2.638555	1.530707
17	6	0	8.055957	-2.428454	-0.559502
18	1	0	6.492096	-1.523019	0.589806
19	1	0	1.643620	-4.234300	-1.061064
20	6	0	2.747244	-1.726576	0.944231
21	1	0	4.750796	-1.470734	1.617924
22	1	0	7.545502	-2.261472	-1.514434
23	1	0	8.564793	-3.393743	-0.646159
24	1	0	2.399965	-0.895147	1.545725
25	6	0	1.812283	-2.483738	0.205461
26	6	0	9.081887	-1.321940	-0.298306
27	1	0	9.828349	-1.286945	-1.096817
28	1	0	8.605490	-0.336368	-0.247014
29	1	0	9.614829	-1.482312	0.646030
30	16	0	4.276688	-5.182269	-1.546653
31	8	0	3.241357	-6.214469	-1.724503
32	8	0	4.894103	-4.572235	-2.738073
33	6	0	0.367284	-2.228346	0.191210
34	6	0	-0.318031	-1.069812	0.376487
35	1	0	-0.229752	-3.117388	-0.001766
36	1	0	7.214256	-8.681756	0.206446
37	6	0	0.334274	0.270912	0.442594
38	6	0	0.115848	1.130095	1.532453

39	6	0	1.156044	0.726472	-0.600931
40	6	0	0.720556	2.382326	1.589499
41	1	0	-0.527051	0.805492	2.345417
42	6	0	1.738486	1.989437	-0.555575
43	1	0	1.334645	0.081337	-1.455398
44	6	0	1.529242	2.847398	0.537259
45	1	0	0.556472	3.017854	2.454546
46	1	0	2.361920	2.321147	-1.379108
47	6	0	2.181314	4.189598	0.611694
48	6	0	2.140319	5.085214	-0.419640
49	6	0	2.878793	4.473167	1.903352
50	6	0	2.998636	6.307867	-0.464038
51	6	0	1.229427	4.917037	-1.593154
52	6	0	3.739869	3.519268	2.473605
53	6	0	2.650535	5.666087	2.609882
54	6	0	2.444770	7.556149	-0.798822
55	6	0	4.383017	6.236853	-0.233693
56	6	0	1.717675	5.062541	-2.903392
57	6	0	-0.141981	4.665121	-1.419889
58	6	0	4.374650	3.762213	3.690299
59	1	0	3.912523	2.583243	1.950721
60	6	0	3.276014	5.904607	3.832373
61	1	0	1.976203	6.407138	2.193642
62	6	0	3.240925	8.697885	-0.867760
63	1	0	1.380482	7.626900	-1.001817
64	6	0	5.181475	7.376219	-0.315097
65	1	0	4.831022	5.278643	0.007651

66	6	0	0.871587	4.924174	-4.002204
67	1	0	2.770694	5.279039	-3.056507
68	6	0	-0.990886	4.539621	-2.518308
69	1	0	-0.540009	4.571184	-0.414812
70	6	0	4.144867	4.956642	4.375697
71	1	0	5.046025	3.016237	4.106377
72	1	0	3.080748	6.831888	4.363384
73	6	0	4.613564	8.612755	-0.626841
74	1	0	2.789495	9.654317	-1.115787
75	1	0	6.250455	7.296577	-0.139095
76	6	0	-0.486779	4.661634	-3.814367
77	1	0	1.273127	5.025435	-5.006450
78	1	0	-2.048977	4.350455	-2.361085
79	1	0	4.632936	5.143953	5.327690
80	1	0	5.235992	9.500640	-0.689020
81	1	0	-1.147757	4.560142	-4.670093
82	6	0	-1.801576	-1.099707	0.462667
83	6	0	-2.482115	-2.177997	1.058189
84	6	0	-2.579820	-0.046374	-0.052140
85	6	0	-3.870845	-2.211919	1.113714
86	1	0	-1.912091	-2.983586	1.510493
87	6	0	-3.968558	-0.088438	-0.009110
88	1	0	-2.087001	0.809469	-0.501617
89	6	0	-4.648263	-1.177553	0.563196
90	1	0	-4.365612	-3.048282	1.597889
91	1	0	-4.540849	0.733864	-0.425299
92	6	0	-6.137960	-1.210769	0.654068

93	6	0	-6.949693	-0.930407	-0.409575
94	6	0	-6.673779	-1.579152	2.000967
95	6	0	-8.420112	-0.704513	-0.263572
96	6	0	-6.438043	-0.830126	-1.810483
97	6	0	-6.177656	-0.959472	3.161279
98	6	0	-7.638780	-2.589422	2.149442
99	6	0	-9.329406	-1.337805	-1.129179
100	6	0	-8.929887	0.185025	0.697319
101	6	0	-6.815902	0.245550	-2.633523
102	6	0	-5.618269	-1.829878	-2.361078
103	6	0	-6.653127	-1.314462	4.422377
104	1	0	-5.416868	-0.190237	3.067570
105	6	0	-8.105825	-2.953140	3.411324
106	1	0	-8.020558	-3.090553	1.266071
107	6	0	-10.700531	-1.115585	-1.015693
108	1	0	-8.953368	-2.012488	-1.892282
109	6	0	-10.300128	0.416460	0.803462
110	1	0	-8.242256	0.697970	1.361398
111	6	0	-6.362531	0.335431	-3.948340
112	1	0	-7.466323	1.017370	-2.233059
113	6	0	-5.174025	-1.746101	-3.679386
114	1	0	-5.332172	-2.677476	-1.747193
115	6	0	-7.619482	-2.313788	4.552926
116	1	0	-6.265088	-0.813151	5.304600
117	1	0	-8.848386	-3.740731	3.502548
118	6	0	-11.192285	-0.236985	-0.048392
119	1	0	-11.385455	-1.626087	-1.686819

120	1	0	-10.670727	1.112360	1.550757
121	6	0	-5.539181	-0.660389	-4.477309
122	1	0	-6.656949	1.182068	-4.562147
123	1	0	-4.544838	-2.533179	-4.084987
124	1	0	-7.984963	-2.596301	5.535913
125	1	0	-12.260205	-0.057712	0.035711
126	1	0	-5.191477	-0.594929	-5.504146

Rotational constants (GHZ): 0.0262281 0.0161129 0.0117663

TD-DFT

(PTZ)₃

Excitation energies and oscillator strengths:

Excited State 1: Singlet-A 3.0572 eV 405.55 nm f=0.5688 <S**2>=0.000
197 -> 198 0.69607

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -3175.68858448

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 3.3267 eV 372.69 nm f=0.0794 <S**2>=0.000
196 -> 198 0.68590

Excited State 3: Singlet-A 3.4258 eV 361.91 nm f=0.0247 <S**2>=0.000
195 -> 198 0.68463

Excited State 4: Singlet-A 3.7283 eV 332.55 nm f=0.0218 <S**2>=0.000
194 -> 198 0.16289

195 -> 199 0.10151

197 -> 199 0.63624

Excited State 5: Singlet-A 3.8030 eV 326.02 nm f=0.0115 <S**2>=0.000

195 -> 200 0.12948

196 -> 199 0.16244

196 -> 200 0.12868

196 -> 201 0.19770

197 -> 200 0.51404

197 -> 201 0.27769

Excited State 6: Singlet-A 3.8580 eV 321.37 nm f=0.0069 <S**2>=0.000

194 -> 198 -0.10996

195 -> 199 0.23662

195 -> 200 -0.12541

196 -> 200 0.19202

197 -> 200 -0.22056

197 -> 201 0.45226

197 -> 202 -0.19745

197 -> 203 -0.11893

SavETr: write IOETrn= 770 NScale= 10 NData= 16 NLR=1 NState= 6 LETran= 118.

PTZO₂-(PTZ)₂

Excitation energies and oscillator strengths:

Excited State 1: Singlet-A 3.0606 eV 405.10 nm f=0.5077 <S**2>=0.000

205 -> 206 0.69892

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -3326.06923826

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 3.2418 eV 382.46 nm f=0.0922 <S**2>=0.000
204 -> 206 0.69717

Excited State 3: Singlet-A 3.4877 eV 355.49 nm f=0.0220 <S**2>=0.000
203 -> 207 0.12046
204 -> 207 0.10778
205 -> 207 0.66990

Excited State 4: Singlet-A 3.6586 eV 338.89 nm f=0.5148 <S**2>=0.000
203 -> 206 0.68264

Excited State 5: Singlet-A 3.7872 eV 327.37 nm f=0.0406 <S**2>=0.000
204 -> 207 0.61010
204 -> 208 -0.23619
205 -> 207 -0.11951
205 -> 208 -0.12195
205 -> 209 -0.11073

Excited State 6: Singlet-A 3.8104 eV 325.39 nm f=0.0053 <S**2>=0.000
202 -> 206 0.11135
204 -> 210 -0.12671
205 -> 208 -0.35953
205 -> 210 0.45025
205 -> 211 -0.16269
205 -> 212 -0.20282
205 -> 214 -0.11360

SavETr: write IOETrn= 770 NScale= 10 NData= 16 NLR=1 NState= 6 LETran= 118.

PTZ-(TPE)₂

Excitation energies and oscillator strengths:

Excited State 1: Singlet-A 2.9453 eV 420.96 nm f=0.5884 <S**2>=0.000

245 -> 246 0.69657

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -3114.17269158

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 3.1865 eV 389.09 nm f=0.2043 <S**2>=0.000

245 -> 247 0.69204

Excited State 3: Singlet-A 3.3790 eV 366.93 nm f=0.2047 <S**2>=0.000

243 -> 246 -0.19117

244 -> 246 0.65524

Excited State 4: Singlet-A 3.4482 eV 359.57 nm f=0.1927 <S**2>=0.000

243 -> 246 0.65637

244 -> 246 0.20184

Excited State 5: Singlet-A 3.5410 eV 350.13 nm f=0.0674 <S**2>=0.000

245 -> 248 0.66817

Excited State 6: Singlet-A 3.6642 eV 338.37 nm f=0.0289 <S**2>=0.000

243 -> 246 0.11877

243 -> 247 0.14951

244 -> 247 0.66292

SavETr: write IOETrn= 770 NScale= 10 NData= 16 NLR=1 NState= 6 LETran= 118.

PTZO₂-(TPE)₂

Excitation energies and oscillator strengths:

Excited State 1: Singlet-A 3.0914 eV 401.06 nm f=0.9157 <S**2>=0.000

253 -> 254 0.70067

This state for optimization and/or second-order correction.

Total Energy, E(TD-HF/TD-KS) = -3264.54798096

Copying the excited state density for this state as the 1-particle RhoCI density.

Excited State 2: Singlet-A 3.3250 eV 372.88 nm f=0.3138 <S**2>=0.000

252 -> 254 0.59157

253 -> 255 0.37011

Excited State 3: Singlet-A 3.3958 eV 365.11 nm f=0.0426 <S**2>=0.000

252 -> 254 -0.37289

253 -> 255 0.58693

Excited State 4: Singlet-A 3.6113 eV 343.33 nm f=0.0796 <S**2>=0.000

251 -> 254 0.65862

253 -> 257 0.23072

Excited State 5: Singlet-A 3.6355 eV 341.04 nm f=0.0323 <S**2>=0.000

251 -> 256 -0.15018

252 -> 255 -0.10519

253 -> 256 0.66049

Excited State 6: Singlet-A 3.7023 eV 334.89 nm f=0.3243 <S**2>=0.000
252 -> 255 0.66970
253 -> 257 -0.12772
SavETr: write IOETrn= 770 NScale= 10 NData= 16 NLR=1 NState= 6 LETran= 118.

References

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- S2. J. Yang, H. Gao, Y. Wang, Y. Yu, Y. Gong, M. Fang, D. Ding, W. Hu, B. Z. Tang and Z. Li, *Mater. Chem. Front.*, 2019, **3**, 1391–1397.
- S3. J. Huang, X. Yang, J. Wang, C. Zhong, L. Wang, J. Qin and Z. Li, *J. Mater. Chem.*, 2012, **22**, 2478–2484.
- S4. F. Khan, A. Ekbote, S. M. Mobin and R. Misra, *J. Org. Chem.*, 2021, **86**, 1560–1574.