

## Supporting Information

### Effect of Sn Valence State in situ-growth of FeOOH Precursor on Performance of $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> Photoanode

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## Supporting Information

### ■ Results and discussion

**Figure S1.** Tauc's plots of the synthesized Fe<sub>2</sub>O<sub>3</sub>, Sn<sup>2+</sup>-Fe<sub>2</sub>O<sub>3</sub> and Sn<sup>4+</sup>-Fe<sub>2</sub>O<sub>3</sub> films.

Table S1. Relative percentage of Sn for Fe<sub>2</sub>O<sub>3</sub>, Sn<sup>2+</sup>-Fe<sub>2</sub>O<sub>3</sub> and Sn<sup>4+</sup>-Fe<sub>2</sub>O<sub>3</sub> films.

Samples	Fe	O	Sn
	At%	At%	At%
Fe <sub>2</sub> O <sub>3</sub>	24.07	75.93	*
Sn <sup>2+</sup> -Fe <sub>2</sub> O <sub>3</sub>	20.82	68.74	10.44
Sn <sup>4+</sup> -Fe <sub>2</sub> O <sub>3</sub>	21.45	77.90	0.65

Table S2. Relative percentage of O<sub>L</sub> and O<sub>V</sub> components for Fe<sub>2</sub>O<sub>3</sub>, Sn<sup>2+</sup>-Fe<sub>2</sub>O<sub>3</sub> and Sn<sup>4+</sup>-Fe<sub>2</sub>O<sub>3</sub> films.

Samples	O <sub>L</sub>		O <sub>V</sub>	
	BE	At%	BE	At%
Fe <sub>2</sub> O <sub>3</sub>	529.63	51.50	531.44	48.50
Sn <sup>2+</sup> -Fe <sub>2</sub> O <sub>3</sub>	529.93	60.98	531.51	39.02

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$\text{Sn}^{4+}\text{-Fe}_2\text{O}_3$	529.47	43.35	531.38	56.65
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Table S3. Comparison of PEC performance and morphology for different hematite reported previously.

Material/ Reference	Electrolyte	IPCE (%)	Photocurrent density( $\text{mA}/\text{cm}^2$ )	Annealing temperature/Time
$\text{Sn-Fe}_2\text{O}_3$ nanostructures <sup>[1]</sup>	1M NaOH	23.7 ( $1.4V_{\text{RHE}}$ )	1.63 ( $1.4V_{\text{RHE}}$ )	$800^\circ\text{C}/10\text{min}$
E-I- $\text{Sn-Fe}_2\text{O}_3$ NWs <sup>[2]</sup>	1M KOH	27 ( $0.23V_{\text{Ag}/\text{AgCl}}$ )	1.36 ( $0.23V_{\text{Ag}/\text{AgCl}}$ )	$800^\circ\text{C}/20\text{min}$ (HF Etching $\text{SiO}_2$ encapsulation)
$\text{Sn-Fe}_2\text{O}_3$ nanocorals <sup>[3]</sup>	1M NaOH	19.8 ( $1.23V_{\text{RHE}}$ )	1.86 ( $1.23V_{\text{RHE}}$ )	$800^\circ\text{C}/20\text{min}$
$\text{Sn-Fe}_2\text{O}_3$ <sup>[4]</sup>	1M NaOH	17 ( $1.23V_{\text{RHE}}$ )	0.86 ( $1.23V_{\text{RHE}}$ )	$800^\circ\text{C}/20\text{min}$
$\text{Sn-Fe}_2\text{O}_3$ nanorod <sup>[5]</sup>	1M NaOH		1.35 ( $1.23V_{\text{RHE}}$ )	$800^\circ\text{C}/10\text{min}$
$\text{Sn-Fe}_2\text{O}_3$ nanorod <sup>[6]</sup>	1M NaOH		1.00 ( $1.23V_{\text{RHE}}$ )	$800^\circ\text{C}/10\text{min}$
$\text{Sn-Fe}_2\text{O}_3$ nanorod <sup>[7]</sup>	1M KOH	22 ( $1.23V_{\text{RHE}}$ )	0.93 ( $1.23V_{\text{RHE}}$ )	$800^\circ\text{C}/3\text{min}$
$\text{Sn-Fe}_2\text{O}_3$ nanotubes <sup>[8]</sup>	1M NaOH		0.65 ( $1.23V_{\text{RHE}}$ )	$750^\circ\text{C}/30\text{min}$
$\text{Sn-Fe}_2\text{O}_3$ <sup>[9]</sup>	1M NaOH		0.96 ( $1.23V_{\text{RHE}}$ )	$750^\circ\text{C}/15\text{min}$
$\text{Sn}(8\%)\text{-Fe}_2\text{O}_3$ <sup>[10]</sup>	1M KOH	9.9 ( $1.6V_{\text{RHE}}$ )	0.298 ( $1.6V_{\text{RHE}}$ )	$600^\circ\text{C}/240\text{min}$

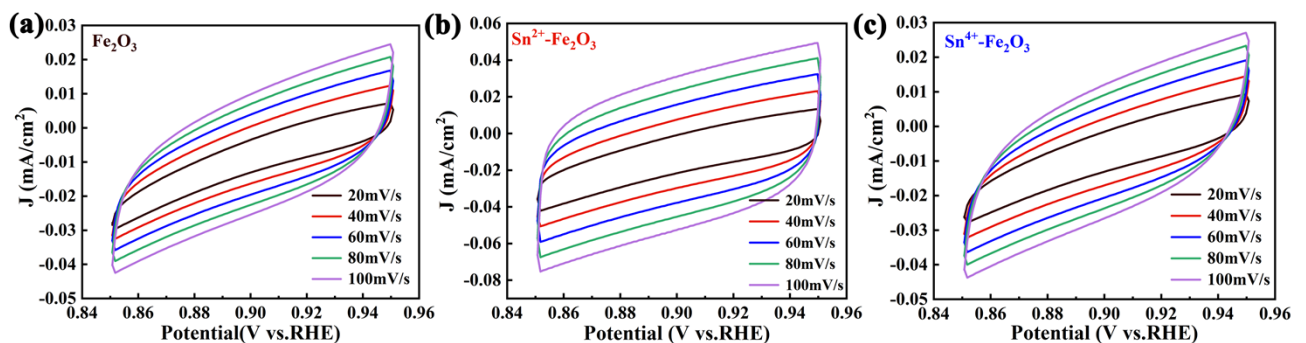


Figure S2. Cyclic voltammetry (CV) for  $\text{Fe}_2\text{O}_3$ ,  $\text{Sn}^{2+}\text{-Fe}_2\text{O}_3$  and  $\text{Sn}^{4+}\text{-Fe}_2\text{O}_3$  photoanodes at various scan rates.

Figure S3. EIS Nyquist for  $\text{Fe}_2\text{O}_3$ ,  $\text{Sn}^{2+}\text{-Fe}_2\text{O}_3$  and  $\text{Sn}^{4+}\text{-Fe}_2\text{O}_3$  photoanodes.

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