Supporting Information for:

## Flexible Direct Synthesis of Phosphorus-Rich CoP<sub>3</sub> on Carbon Black and its Examination in Hydrogen Evolution Electrocatalysis

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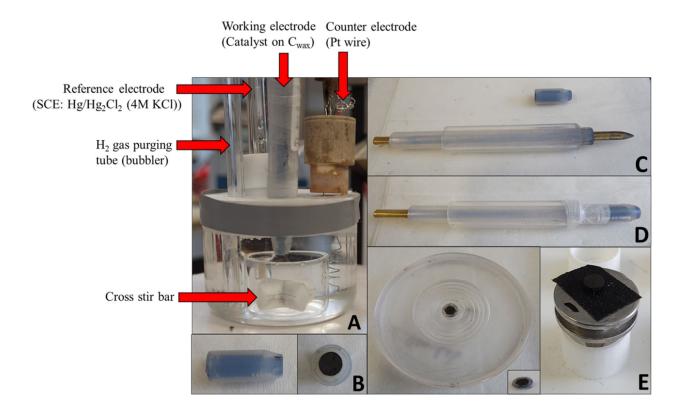
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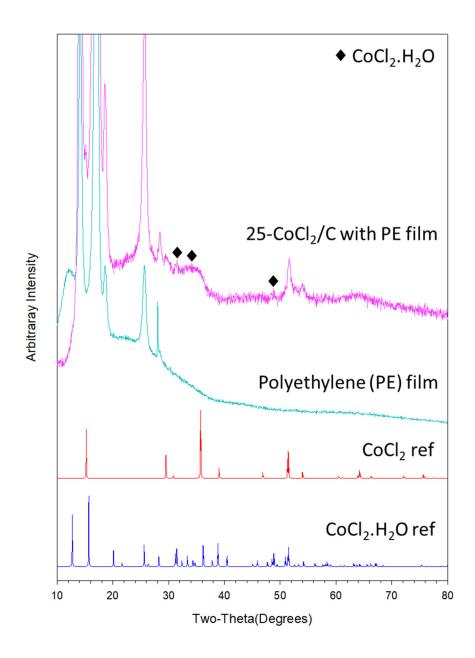
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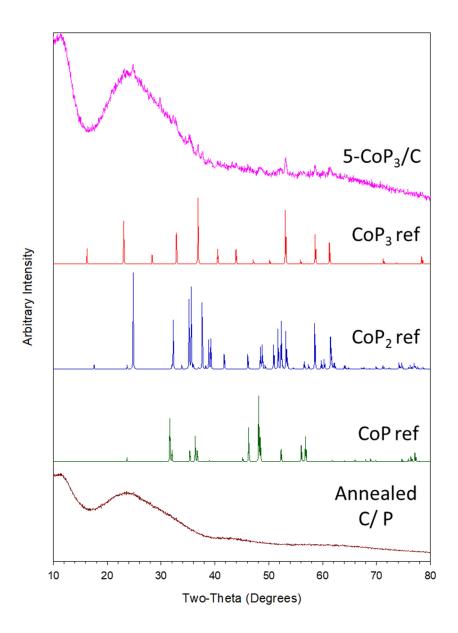
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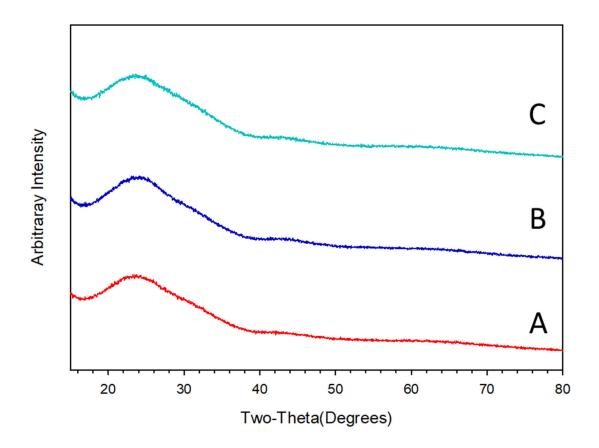
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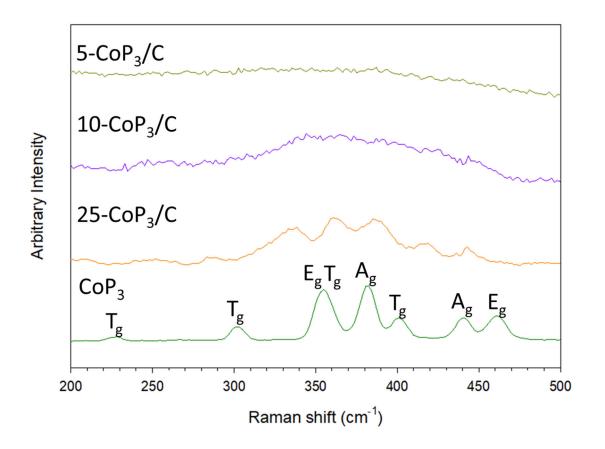
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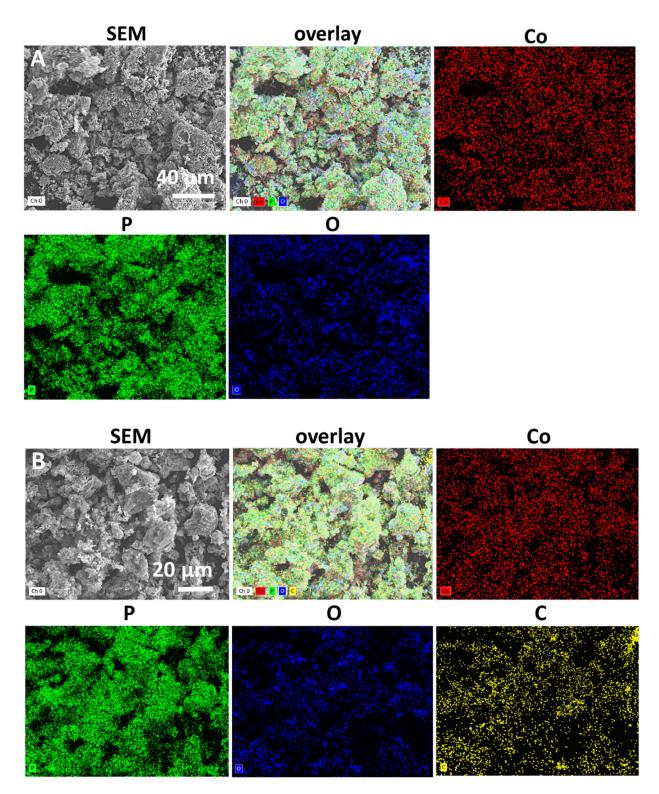
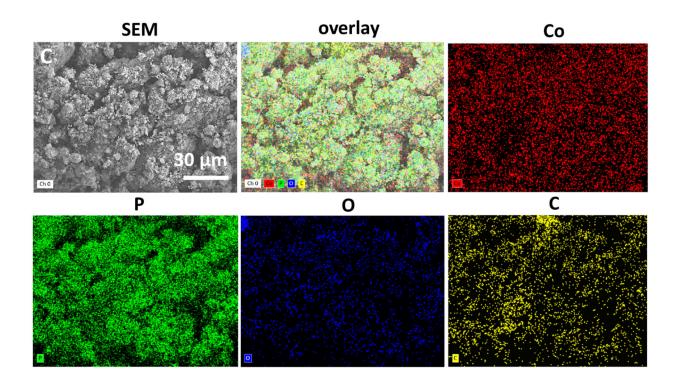


Figure S6. EDS elemental maps of (A) CoP<sub>3</sub> and (B) 25-CoP<sub>3</sub>/C.



SEM

overlay

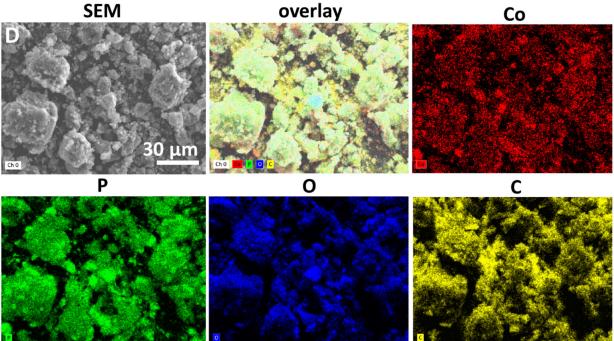


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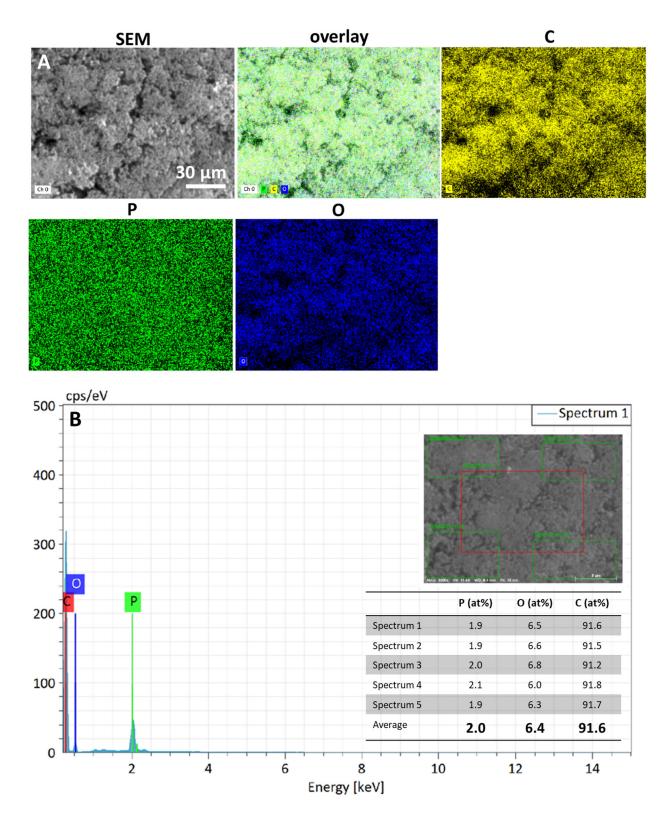
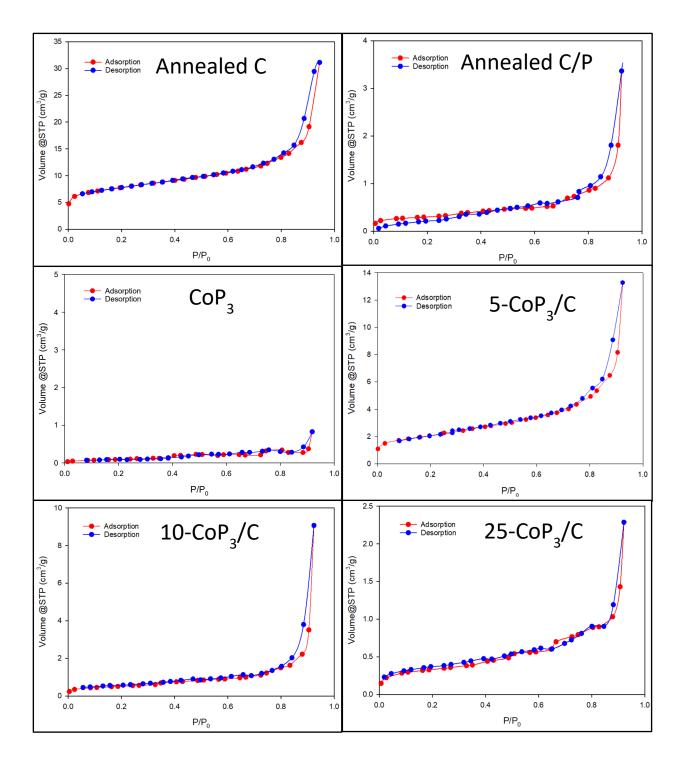
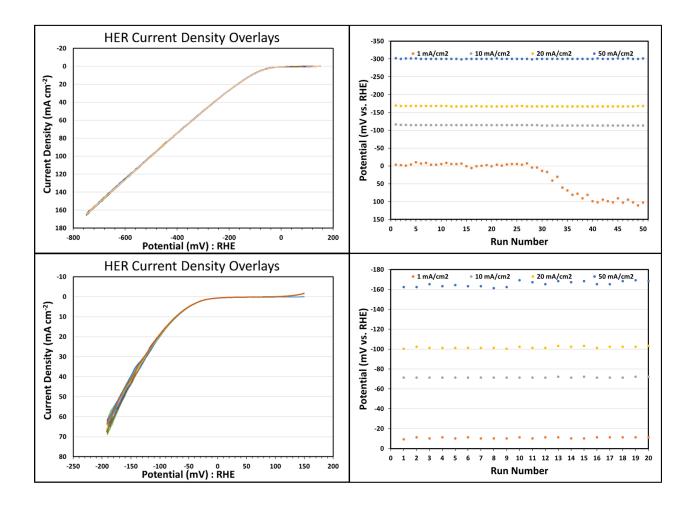


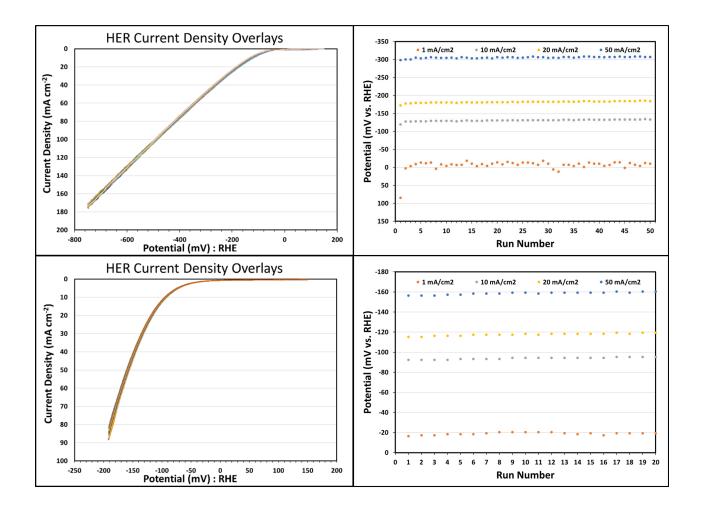
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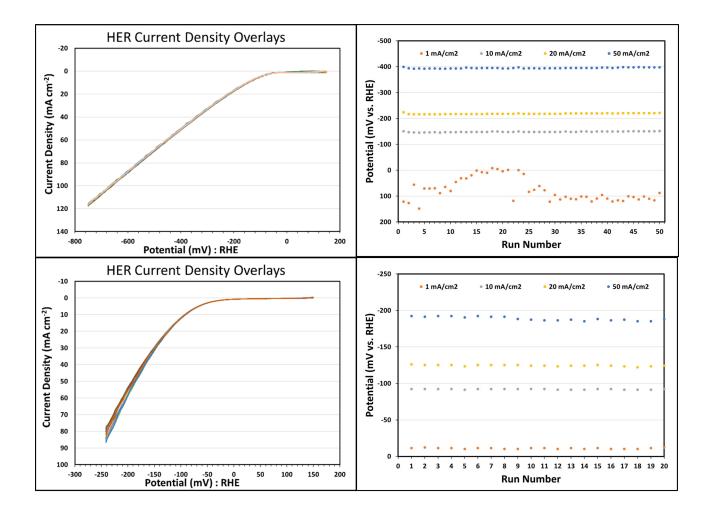
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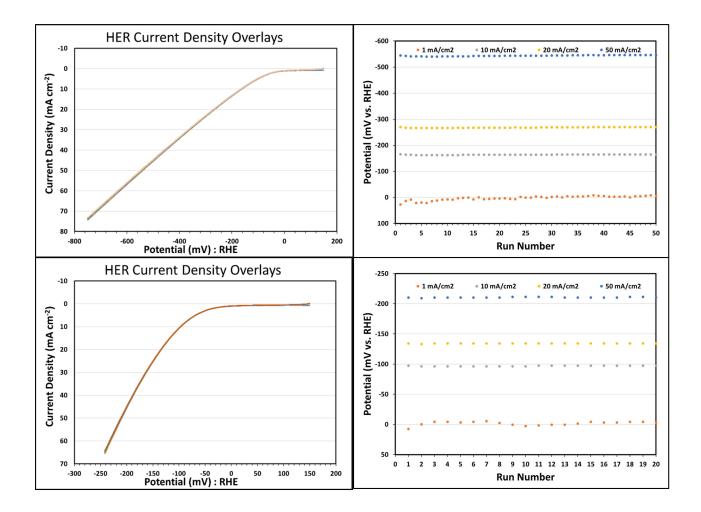
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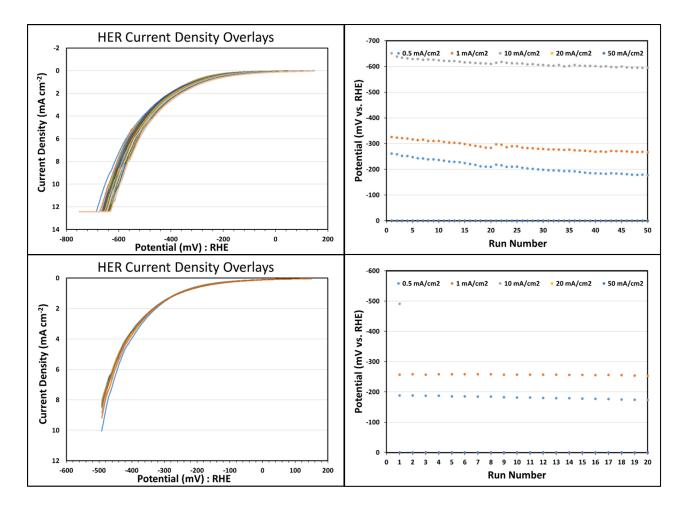
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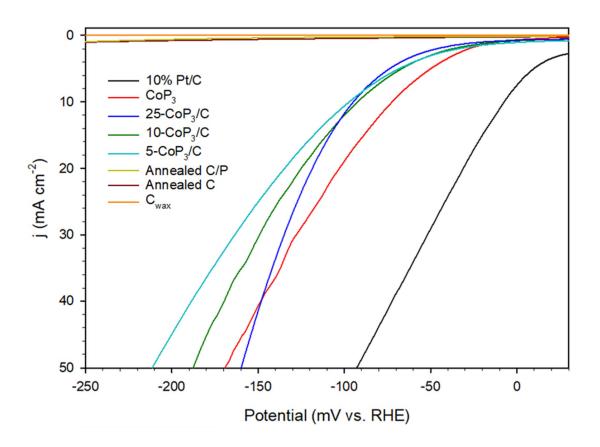
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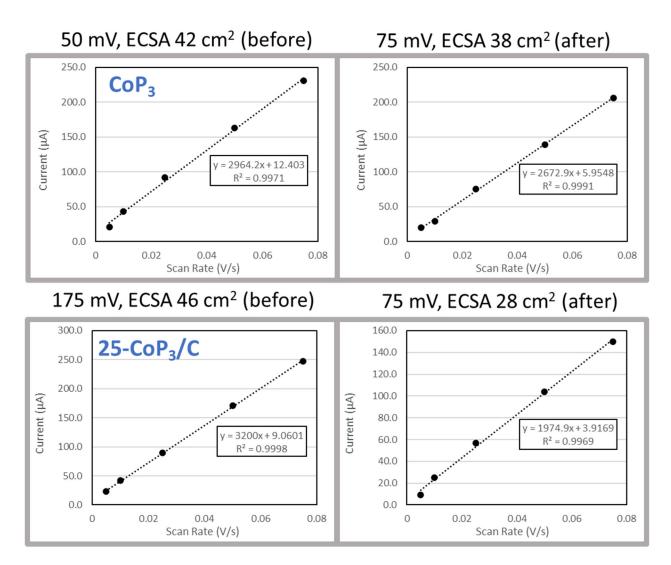
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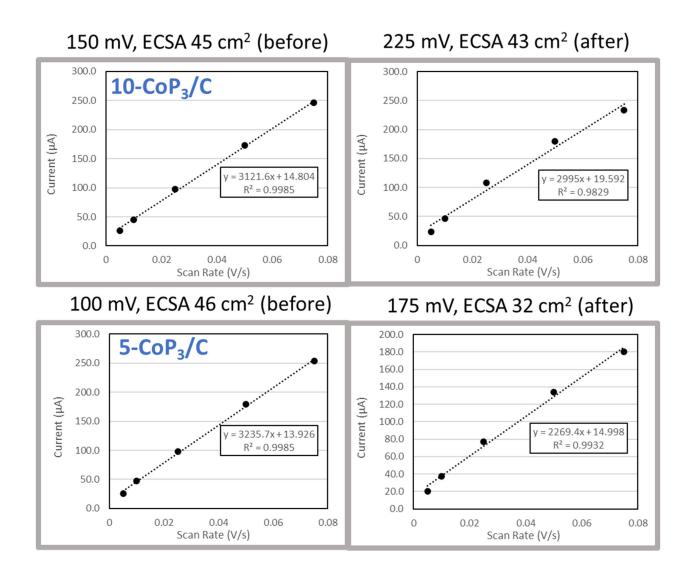
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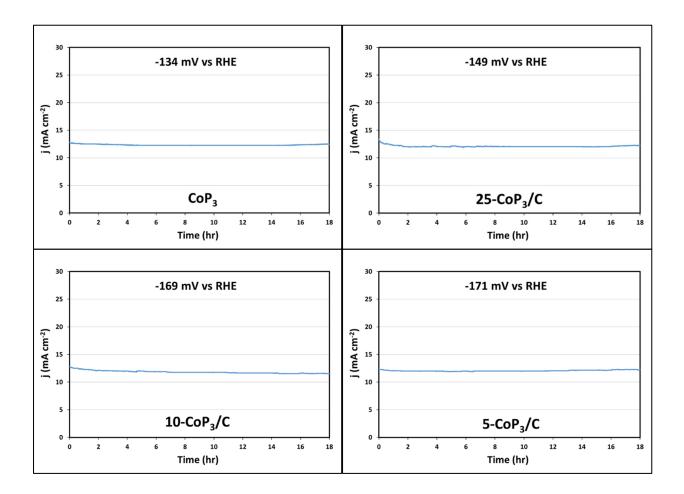
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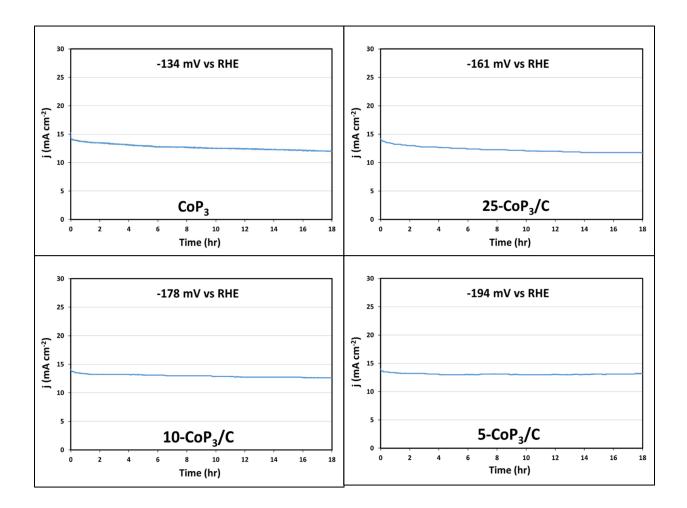
**Figure S16.** Analysis of scan rate data from CV runs to calculate ECSA values before and after 50 LSV scans (*iR* uncompensated) for CoP<sub>3</sub> and 25-CoP<sub>3</sub>/C materials from Table 2.



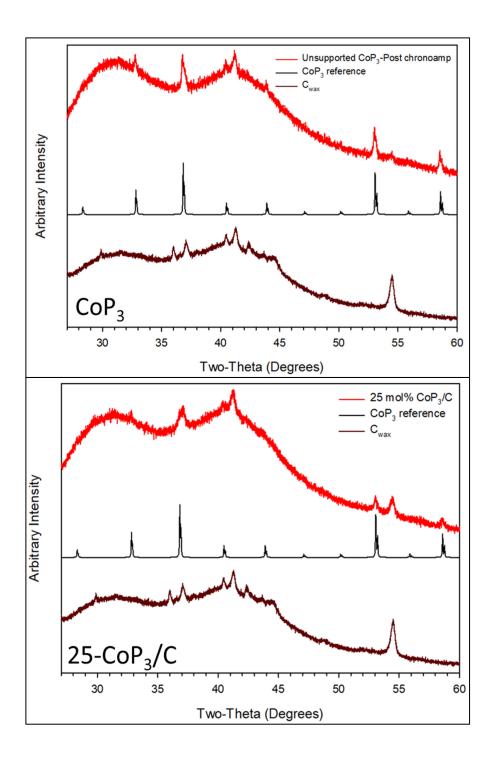
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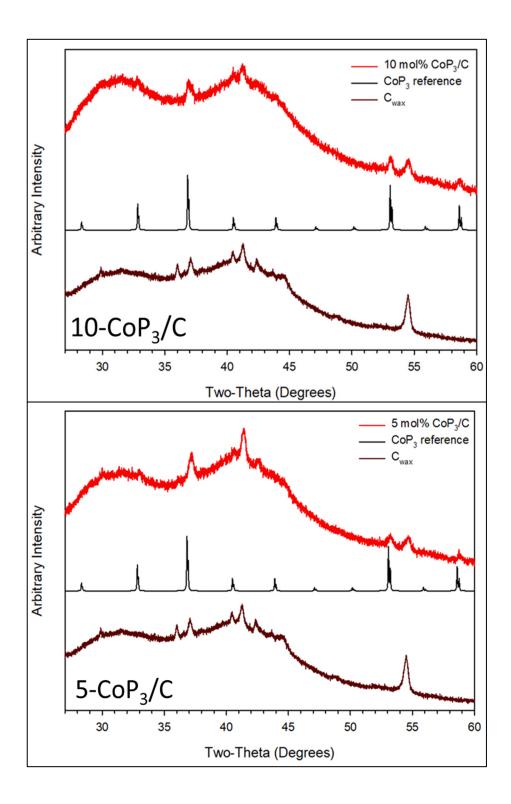
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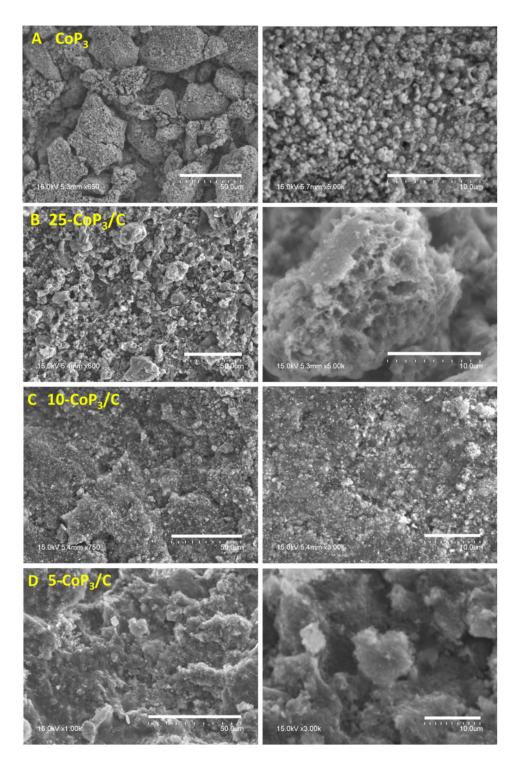
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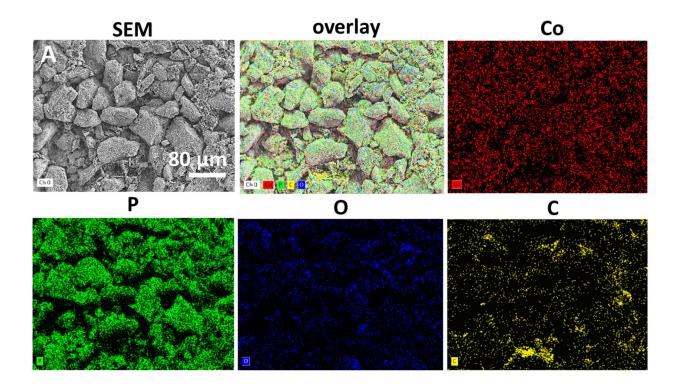
**Figure S20**. XRD results of  $CoP_{3}$ , and 25-CoP<sub>3</sub>/C materials embedded on  $C_{wax}$  tips before and after 18-hour constant potential chronoamperometry (CA) HER experiments.

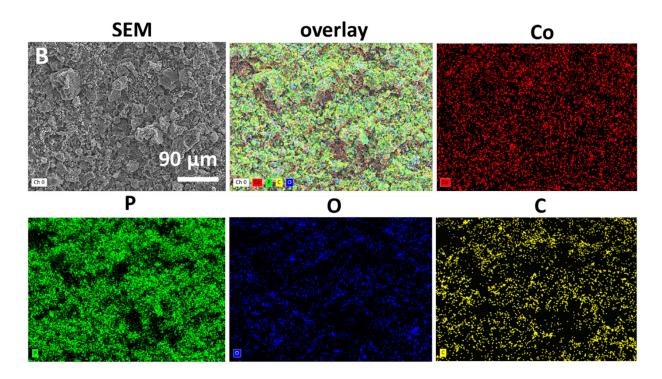


**Figure S21**. XRD results of 10-CoP<sub>3</sub>/C and 5-CoP<sub>3</sub>/C materials embedded on  $C_{wax}$  tips before and after 18-hour constant potential chronoamperometry (CA) HER experiments.

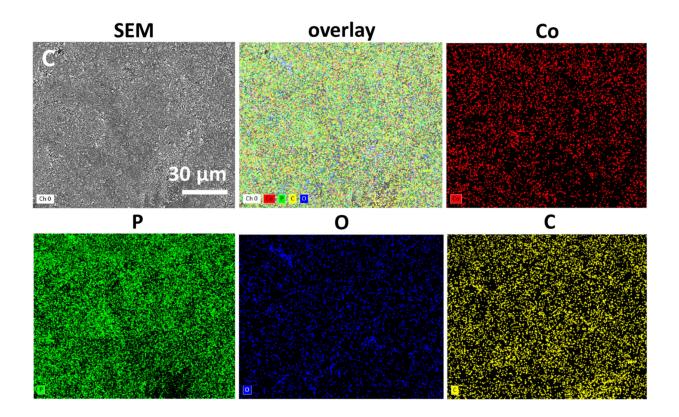


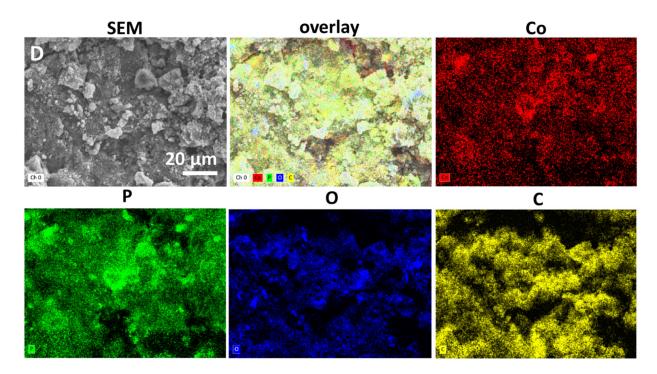
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**Figure S23**. EDS maps of (A) CoP<sub>3</sub> and (B) 25-CoP<sub>3</sub>/C embedded on C<sub>wax</sub> electrode tips after 18-hour CA HER experiments.

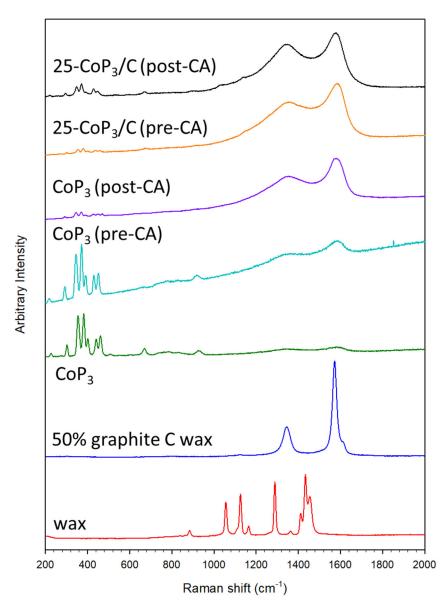




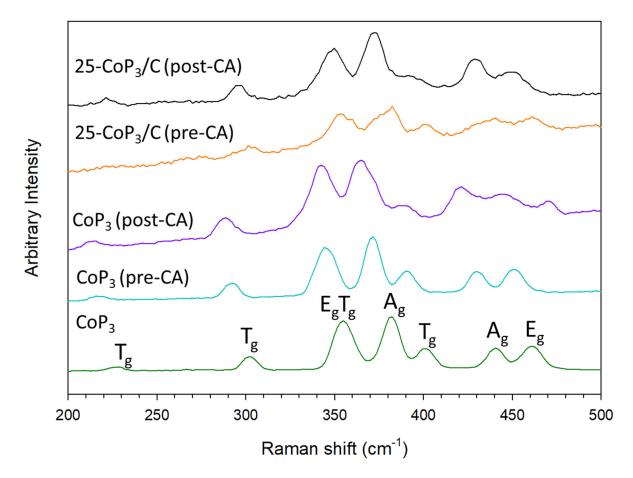
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**Table S1.** EDS compositional analysis of  $CoP_3$ , and x-CoP<sub>3</sub>/C materials embedded on  $C_{wax}$  tips after 18-hour constant potential chronoamperometry (CA) HER experiments.

Sample	Co (at%)	P (at%)	O (at%)	C (at%)
CoP <sub>3</sub>	6.8	13.8	34.6	44.8
25-CoP <sub>3</sub> /C	6.2	12.2	30.5	51.1
10-CoP <sub>3</sub> /C	3.8	8.4	21.0	66.8
5-CoP <sub>3</sub> /C	1.0	2.3	8.7	88.0



**Figure S25**. Raman spectra of  $CoP_3$  and 25- $CoP_3/C$  materials embedded on  $C_{wax}$  tips before 18-hour constant potential chronoamperometry (pre-CA) and after CA (post-CA) HER experiments in the region of 200-2000cm<sup>-1</sup>.



**Figure S26**. Raman spectra of  $CoP_3$  and 25- $CoP_3/C$  materials embedded  $C_{wax}$  tips before 18-hour constant potential chronoamperometry (pre-CA) and after CA (post-CA) HER experiments in a zoomed in region of 200-500 cm<sup>-1</sup>.

**Table S2**. Literature comparison table for carbon-supported cobalt phosphides HER in 0.5 M H<sub>2</sub>SO<sub>4</sub>. All current densities are normalized for the geometrical surface area of the electrode. a) Values estimated from plots and graphical data in reference. GCE = Glassy carbon electrode, NRs = Nanorods, NPs = Nanoparticles, CNTs = Carbon nano tubes, NCNT = N-doped CNTs, NCNW = N-doped bundled carbon nanowires, PNC = P, N co-doped carbon, NWs = Nanowires, RGO = Reduced Graphene Oxide, NAs = nanoarrays, CP = Concave Polyhedrons, CFP = Carbon Fiber Paper.

Supported Cobalt	Electrode,	10 mA/cm <sup>2</sup>	20 mA/cm <sup>2</sup>	Tafel	Ref.
phosphide	Geometric	(mV)	(mV)	slope	
	area (cm <sup>2</sup> )			(mV/dec)	
Co <sub>2</sub> P NRs	Ti foil (0.5)	-134	-167	52	1
Co <sub>2</sub> P NPs	Ti foil (0.2)	-95	-109	45	2
Co <sub>2</sub> P/CNT	GCE (0.09)	-195	-219	74	3
Co <sub>2</sub> P/NCNT	GCE (0.09)	-170 <sup>a</sup>	-190 <sup>a</sup>	62	3
CoP/NCNWs	GCE (0.09)	-95	-120 <sup>a</sup>	50	4
CoP@PNC	GCE (0.09)	-84	-110 <sup>a</sup>	57	4
CoP/CNT	GCE (0.09)	-122	-180 <sup>a</sup>	54	3
CoP/CNT	GCE (0.09)	-165	-198	68	3
CoP/NCNT	GCE (0.09)	-79	-99	49	3
CoP NPs	Ti foil (0.2)	-75	-85	50	5
CoP NWs	GCE (0.09)	-110	-142	54	6
CoP	CC (6)	-67	-100	51	7
CoP/RGO	GCE (0.09)	-157	-190 <sup>a</sup>	70	8
CoP <sub>2</sub>	GCE (0.09)	-120	-150 <sup>a</sup>	73	9
CoP <sub>2</sub> /RGO	GCE (0.09)	-88	-106 <sup>a</sup>	50	9
CoP <sub>3</sub> NAs	CFP (5.4)	-65	-100 <sup>a</sup>	46	10
CoP <sub>3</sub> CPs	CFP (0.25)	-78	-110 <sup>a</sup>	53	11

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