## **Supplementary Information**

## Combination of float charging and occasional discharging to cause serious LIB degradation analyzed by operando

## neutron diffraction

Tetsuya Omiya<sup>1</sup>, Atsunori Ikezawa<sup>1</sup>, Keita Takahashi<sup>2</sup>, Keiichi Saito<sup>2</sup>, Masao Yonemura<sup>3</sup>, Takashi Saito<sup>3</sup>, Takashi Kamiyama<sup>3</sup>, Hajime Arai<sup>1</sup>

<sup>1</sup> Tokyo Institute of Technology 4259, Nagatsuta, Yokohama, Midori-ku, Kanagawa, 226-0026, Japan

<sup>2</sup> NTT Anode Energy Corporation., Granparktower, 3-4-1 Shibaura, Minato-ku, Tokyo, 105-0023, Japan

<sup>3</sup> High Energy Accelerator Research Organization 203-1, Oazashirane, Tokai, Naka, Ibaraki, 319-1195, Japan

Mode	Ste	р	Parameters	Termination
Cycling	1)	CC Charge	I = 1.675  A (0.5 C)	V = 4.2 V
	2)	Charge Rest (OCV)	OCV	$t > 10 \min$
	3)	CC Discharge	$I = -3.35 \mathrm{A} (1\mathrm{C})$	V = 2.5 V
	4)	Discharge Rest (OCV)	OCV	<i>t</i> > 10 min
Floating	1)	CCCV Charge	<i>I</i> = 1.675 A (0.5C),	
			V = 4.2 V	
Floating-cycling	1)	CCCV Charge	<i>I</i> = 1.675 A (0.5C),	t > 18 h
			V = 4.2 V	
	2)	CC Discharge	$I = -3.35 \mathrm{A} (1\mathrm{C})$	V = 2.5 V
	3)	Discharge Rest (OCV)	OCV	t > 5 h

 Table S1.
 Overview of durability test conditions and parameters.



(b)



Fig.S1 (a) Schematic image of instrumental setup and (b) photo of operando cell in the chamber.

(a)

Mode	Tested days or	Fresh cell capacity	After test capacity	Relative capacity
	cycle number	[Ah]	[Ah]	[%]
Cycling	397	2.83	2.69	95.2
Floating	397	2.84	2.74	96.4
Floating-cycling	397	2.84	2.21	77.9
Floating-cycling (0.3C)	397	2.84	2.16	75.8

 Table S2.
 Discharge capacity values of durability test cell.



**Fig. S2** Trends of (left) 1C discharge capacity of horizontally-set cells as function of tested days (Floating and Floating-cycling) and (right) the corresponding capacity retention at 0.1C before and after durability test.



**Fig. S3** Neutron diffraction patterns of the fresh cell. The reflection corresponding to stage 4 of graphite was observed only for the fresh cell, that also appears in the literature<sup>1</sup>, suggesting that this is one of a stable phase among lithiated-graphite compounds.



Fig.S4 Lattice constant evolution of the positive electrode material (a) *a* and (b) *c* for the fresh cell.



**Fig. S5** The diffraction profiles of the positive electrode 003 reflection of the Floating-cycling cell: (a) Fresh cell, (b) after 160 days, (c) after 397 days.



**Fig. S6** The selected diffraction profiles of the positive electrode 003 reflection of the Floatingcycling cell: (a) Fresh cell, (b) after 160 days, (c) after 397 days.



**Fig. S7** The width at half maximum of the diffraction peaks shown in Fig. S5, obtained by Gaussian distribution fitting.



**Fig. S8** Neutron diffraction patterns and results of Rietveld refinement of the fresh cell in (a) fully charged (b) fully discharged states. The background of diffraction patterns is somewhat large, due to the presence of hydrogen atoms in the electrolyte and separator, but it is monotonous and not too significant to interfere the Rietveld analysis as shown in the figure.



**Fig. S9** Neutron diffraction patterns and results of Rietveld refinement after floating test in (a) fully charged (b) fully discharged states.



Fig. S10 Neutron diffraction patterns and results of Rietveld refinement after floating-cycling test in (a) fully charged (b) fully discharged states.

Full charged				c [, •]
r'un chargeu	8.26	4.20	2.91	0.509
Full discharged	8.40	4.04	2.94	0.482
Full charged	11.6	3.56	2.74	0.308
Full discharged	9.04	3.28	2.47	0.363
Full charged	9.15	2.79	2.13	0.305
Full discharged	9.49	4.02	3.17	0.424
	Full dischargedFull chargedFull dischargedFull chargedFull chargedFull discharged	Full discharged8.40Full charged11.6Full discharged9.04Full charged9.15Full discharged9.49	Full discharged8.404.04Full charged11.63.56Full discharged9.043.28Full charged9.152.79Full discharged9.494.02	Full discharged8.404.042.94Full charged11.63.562.74Full discharged9.043.282.47Full charged9.152.792.13Full discharged9.494.023.17

**Table S3.** R factors obtained by Rietveld refinement before and after durability tests.



**Fig.S11** The evolution of the positive lattice constant *a* of the Fresh cell during discharging, approximated by the following 6th order function;  $y = 2.8075 - 2.9340 \times 10^{-6}x + 3.9646 \times 10^{-8}x^2 - 5.8252 \times 10^{-11}x^3 + 4.4545 \times 10^{-14}x^4 - 1.5851 \times 10^{-17}x^5 + 1.9993 \times 10^{-21}x^6$ , where *x* is the discharge capacity [mAh] and *y* is the value of the lattice constant *a* [Å], together with the constants *a* of the Floating-cycling cell during discharging (violet circles). Two parameters, the capacity shift caused by incomplete charging and the expansion rate derived from the loss of the positive electrode active material (due to degradation) were optimized to minimize the root-mean-square errors with the approximation curve. The resultant fitted constants *a* are also shown as pink squares.



Fig. S12 Charge-discharge curves and dV/dQ curves of the positive electrode half cells at the 1st and 2nd cycle: (a) charge-discharge curves of fresh cell, (b) charge-discharge curves of the cell after the Floating-cycling test, (c) corresponding dV/dQ curves. All half cells have been stored in the fully discharged states for more than two years after full cell disassembling and reassembling to half-cells.



**Fig. S13** Nyquist plots of the positive electrode half cells in (a) fully charged and (b) fully discharged states. All half cells have been stored in the fully discharged states for more than two years after full cell disassembling and reassembling to half-cells.



Fig. S14 SEM images of positive electrode surface: (a) Fresh cell and (b) Floating-cycling.



Fig. S15 EDX results of positive electrode surface: (a) Fresh cell and (b) Floating-cycling.



Fig. S16 SEM images of negative electrode surface: (a) Fresh cell and (b) Floating-cycling.



Fig. S17 EDX results of negative electrode surface: (a) Fresh cell and (b) Floating-cycling.



**Fig. S18** EDX results of positive electrode cross section after Floating-cycling test: (a) and (b) fluorine mapping, (c) and (d) carbon mapping, (a) and (c) electrolyte side, (b) and (d) current collector side.



**Fig. S19** Cross-sectional TEM images of the positive electrode particle in the fresh cell (a) Overview of TEM image, (b) Particle TEM image of the marked region in (a), (c) High-magnification TEM image in (b).



**Fig. S20** Cross-sectional TEM images of the positive electrode in the cell after 397 days of Floatingcycling: (a) Overview of TEM image, (b) Particle TEM image of the active material surface, (c) particle image in the crack region, (d) High-magnification TEM image in (b), (e) Highmagnification TEM image in (c).

## Reference

1. L. B. Roblin, D. Sheptyakov, P. Borel, C. Tessier, P. Novák, and C. Villevieille, Crystal structure evolution via operando neutron diffraction during long-term cycling of a customized 5 V full Li-ion cylindrical cell LiNi<sub>0.5</sub>Mn<sub>1.5</sub>O<sub>4</sub> *vs.* graphite, *J. Mater. Chem. A*, 2017, **5**, 25574–25582, doi: 10.1039/c7ta07917f.