Supporting Information

## Gas-fed Photoelectrochemical Reactions Sustained by Phosphotungstic Acid as an Inorganic Surface Electrolyte

Fumiaki Amano<sup>1\*</sup>, Keisuke Tsushiro<sup>1,2</sup>, Chiho Akamoto<sup>2</sup>

<sup>1</sup> Department of Applied Chemistry for Environment, Tokyo Metropolitan University, 1-1 Minami-osawa, Hachioji, Tokyo 192-0397, Japan

<sup>2</sup> Department of Chemical and Environmental Engineering, The University of Kitakyushu, 1-1 Hibikino, Wakamatsu-ku, Kitakyushu, Fukuoka 808-0135, Japan

\*Corresponding author

Fumiaki Amano: f.amano@tmu.ac.jp

Photoanode	IPCE (%) <sup>c</sup>	H <sub>2</sub> FE (%) $^d$	O <sub>2</sub> FE (%) <sup>e</sup>
PFSA/WO3 <sup>b</sup>	2.4	93.6	84.8
PWA/WO <sub>3</sub> <sup>b</sup>	2.8	99.4	94.8

Table S1. Vapor-fed PEC water splitting at 1.2 V (vs. Pt-CB) under visible light for 20 h<sup>a</sup>

 $^a$  Wavelength 453 nm, Irradiance 12 mW cm  $^{-2}$  , and area 16 cm  $^2$ 

<sup>b</sup> Surface electrolyte loading 5 wt%

<sup>c</sup> Incident photon-to-current conversion efficiency, which was calculated by subtracting the dark current from the current density just before the light is turned off.

<sup>d</sup> Faradaic efficiency of H<sub>2</sub> evolution on the cathode side

<sup>e</sup> Faradaic efficiency of O<sub>2</sub> evolution on the photoanode side

Photoanode	Reaction	C (mass%)	O (mass%)	F (mass%)	S (mass%)	Ti (mass%)	W (mass%)
PFSA/WO3 <sup>b</sup>	Before	14.2	10.3	36.5	0.18	22.1	16.7
	After	9.3	17.5	18.5	0.13	21.8	32.8
Photoanode	Reaction	0	Р	Ti	W		
		(mass%)	(mass%)	(mass%)	(mass%)		
PWA/WO <sub>3</sub> <sup>b</sup>	Before	29.0	0.29	25.8	44.9		
	After	25.6	0.31	20.5	53.6		

**Table S2.** SEM-EDS analysis of the functionalized WO<sub>3</sub> photoanodes before and after the long-term vapour-fed PEC water splitting at 1.2 V (vs. Pt-CB) under visible light for 20 h<sup>a</sup>

 $^a$  Wavelength 453 nm, Irradiance 12 mW cm  $^{-2}$  , and area 16 cm  $^2$ 

<sup>b</sup> Surface electrolyte loading 5 wt%

**Table S3.** Gas-phase PEC methane conversion at 1.2 V (vs. Pt-CB) under visible light irradiation(Fig.7)  $^{a}$ 

	IPCE	H <sub>2</sub> FE	FE in photoanode (%)				Selectivity (%, C-basis)			
Photoanode	(%)	(%)	O <sub>2</sub>	C <sub>2</sub> H <sub>6</sub>	CO <sub>2</sub>	СО	Sum	C <sub>2</sub> H <sub>6</sub>	CO <sub>2</sub>	СО
PFSA/WO <sub>3</sub> <sup>b</sup>	6.3	99.5	15.7	8.6	63.2	2.7	90.2	50.5	46.7	2.8
PWA/WO <sub>3</sub> <sup>c</sup>	7.1	101	2.5	10.5	71.3	9.2	93.9	50.0	42.3	7.6

 $^{\it a}$  Wavelength 453 nm, Irradiance 22 mW cm $^{-2}$  , and area 2 cm $^2$ 

<sup>b</sup> PFSA loading 12 wt%

<sup>c</sup> PWA loading 5 wt%



Figure S1. SEM images of the porous WO<sub>3</sub> electrode functionalized by PWA coating (PWA/WO<sub>3</sub>).



**Figure S2.** Cyclic voltammetry curves of the Pt-CB electrocatalyst in 0.1 M H<sub>2</sub>SO<sub>4</sub> electrolyte (pH 1.0) at a scan rate of 10 mV s<sup>-1</sup> with and without *iR* compensation. The series resistance between the working and reference electrodes was measured by electrochemical impedance spectroscopy analysis.



**Figure S3.** SEM-EDS mapping images of the PWA/WO<sub>3</sub> photoanode after the long-term vapour-fed PEC water splitting for 20 h.



**Figure S4.** Raman spectra of (a) bare WO<sub>3</sub>, (b) PWA/WO<sub>3</sub>, and (c) PWA/WO<sub>3</sub> after the long-term vapor-fed PEC water splitting for 20 h.



**Figure S5.** Vapour-fed PEC water splitting of PFSA/WO<sub>3</sub> and PWA/WO<sub>3</sub> photoanodes under UV light (wavelength 385 nm, irradiance 60 mW cm<sup>-2</sup>, irradiation area 2 cm<sup>2</sup>) at an applied voltage of 1.2 V (vs. Pt-CB cathode). Carrier gas: Ar, Flow rate: 20 mL min<sup>-1</sup>, Water vapour: 3 kPa, Temperature: 25 °C.