

Electrochemical and imaging evaluations of electrochemically activated screen-printed gold electrodes

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Equations for calculating the roughness factor (ρ)

$$\text{Charge } (Q_{real}) = \frac{\text{Intergarted area under reduction peak}}{\text{Scan rate of CV}}$$

Eq.1

$$\text{Real electroactive surface } (A_{real}) = \frac{Q_{real}}{Q_{theoretical}} \quad \text{Eq.2}$$

$Q_{theoretical}$ = Theoretical charge density of Au as $390 \mu\text{C cm}^{-2}$.

$$\text{Roughnes factor } (\rho) = \frac{A_{Real}}{A_{Geo}} \quad \text{Eq.3}$$

A_{geo} = Geometric surface area of the SPGE (0.11 cm²)

| Parameter | Value | Fitting error / % |
|------------------|-----------------|-------------------|
| R_u | 34.04 Ω | 0.56 |
| R_{ct1} | 111.17 Ω | 10.15 |
| R_{ct2} | 584.13 Ω | 0.747 |

Table S1: Resistance, fitting error, and convergence fit values for the unactivated SPGE, obtained from the fitted Nyquist and Bode plots using an equivalent model circuit proposed in Figure 2D.

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| Convergence fit value (x ²) = 0.0098 | Parameter | Value | Fitting error / % |
|--|------------------|----------------|-------------------|
| | R_u | 29.45 Ω | 1.66 |
| | R_{ct1} | 10.78 Ω | 3.44 |
| | R_{ct2} | 16.83 Ω | 9.00 |

Table S2: Resistance, fitting error, and convergence fit values for the activated SPGE, obtained from the fitted Nyquist and Bode plots using an equivalent model circuit proposed in Figure 2D.

Convergence fit value (x²) = 0.0026

Equations used for obtaining capacitance, A_{real} , and roughness factor values

$$C_{dl} = \frac{i_a - i_c}{2V} \quad \text{Eq.}$$

$$i_c = A_{Real} C_{dl} \frac{\delta_E}{\delta_t} = A_{real} C_{dl} \nu \quad \text{Eq. 5}$$

$$\rho = \frac{A_{Real}}{A_{Geo}} \quad \text{Eq. 6}$$

C_{dl} = Double layer capacitance
 i_a = Anodic capacitive current
 i_c = Cathodic capacitive current
 ν = CV scan rate
 A_{real} = Real electrode surface area
 A_{geo} = The geometric electrode surface area
 ρ = Roughness factor

The equation used for calculating the heterogeneous electron-transfer rate constant

$$k^o = \frac{RT}{R_{ct} F^2 n A C} \quad \text{Eq. 7}$$

k^o = Heterogeneous electron-transfer rate constant

R = Gas constant

T = Temperature in K at room temperature

R_{ct} = Charge transfer resistance obtained from the fitted Nyquist plot

F = Faradaic constant

C = $[\text{Fe}(\text{CN})_6]^{3-/4-}$ solution in mol cm^{-3}

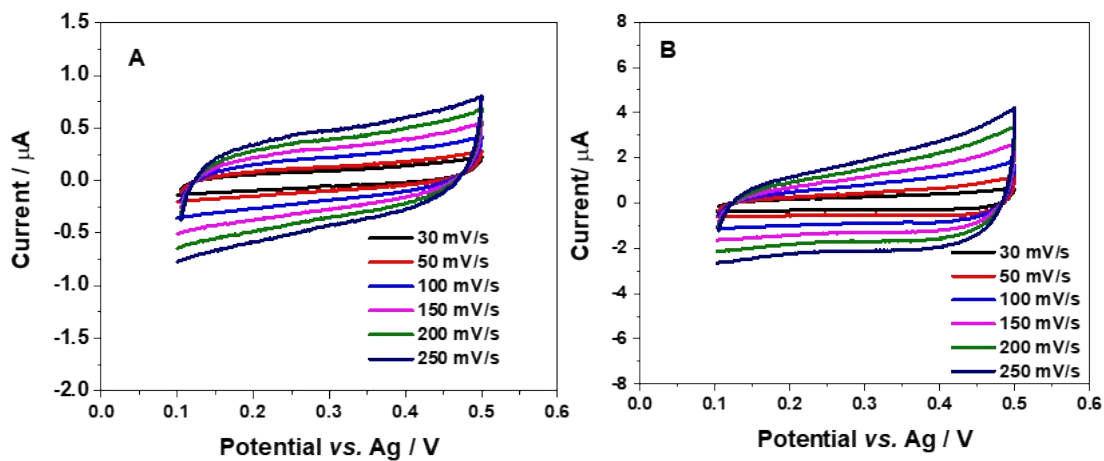


Figure S1: A and B) Cyclic voltammograms for unactivated SPGE and activated SPGE at scan rates of 0.03, 0.05, 0.1, 0.15, 0.2 and 0.25 Vs⁻¹, respectively. The CV measurements were carried out in a 0.1 M KCl solution.

Table S3: Capacitance, A_{real} , and roughness factor values for unactivated and activated SPGEs, calculated from cyclic voltammetric measurements as shown in Figures S1A, S1B, and 3B.

| Parameters | E8 | | E9 | | E14 | |
|------------------------------------|-------------|-----------|-------------|-----------|-------------|-----------|
| | Unactivated | Activated | Unactivated | Activated | Unactivated | Activated |
| Capacitance (slope)/ μF | 1.75 | 8.06 | 1.64 | 8.50 | 1.59 | 5.32 |
| $A_{\text{real}}/ \text{cm}^2$ | 1.19 | 1.20 | 1.20 | 1.12 | 1.12 | 1.20 |
| Roughness factor | 10.8 | 10.9 | 10.9 | 10.2 | 10.2 | 10.9 |

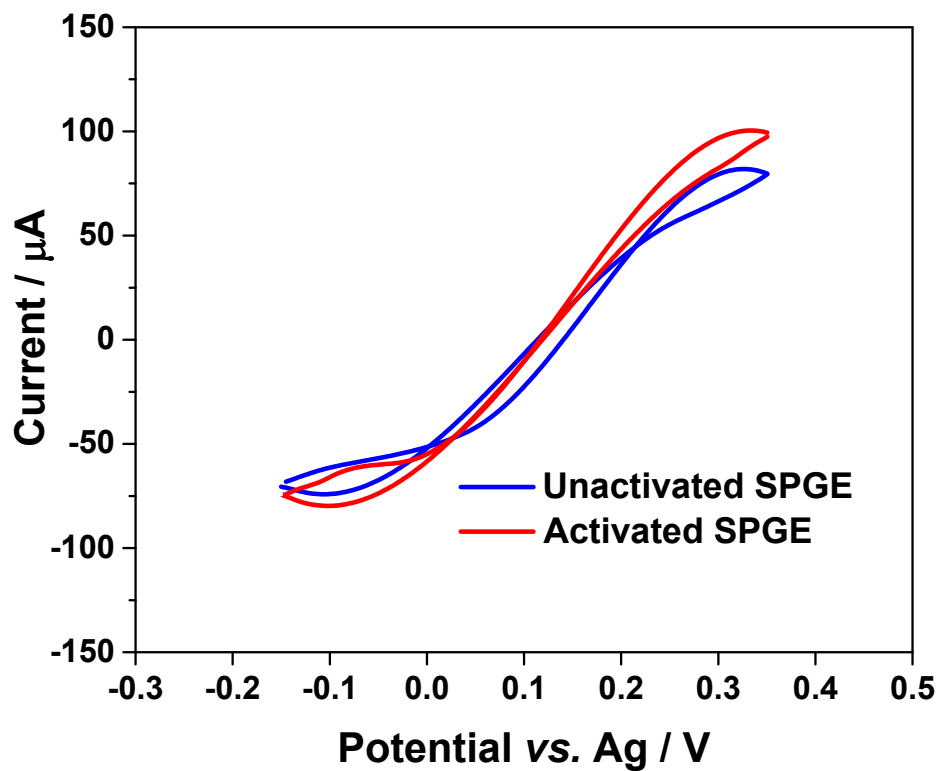
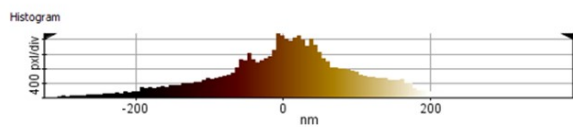
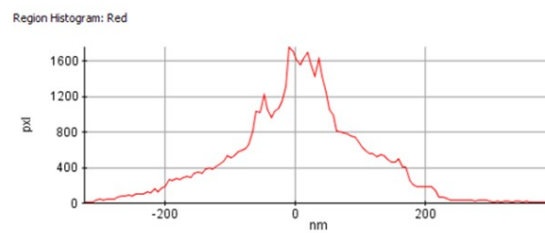
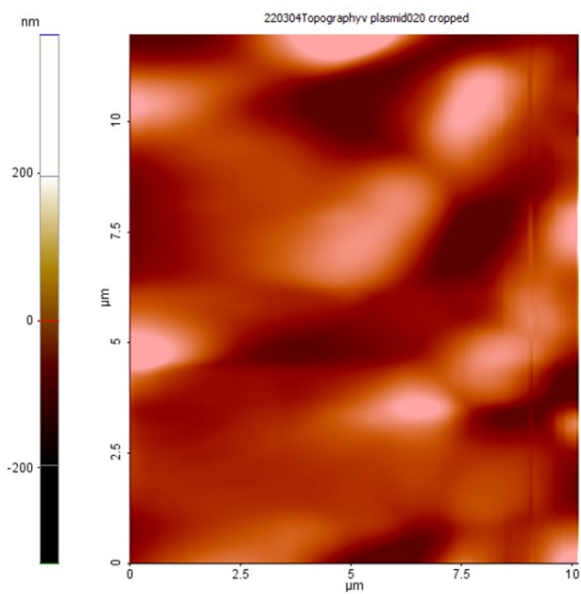


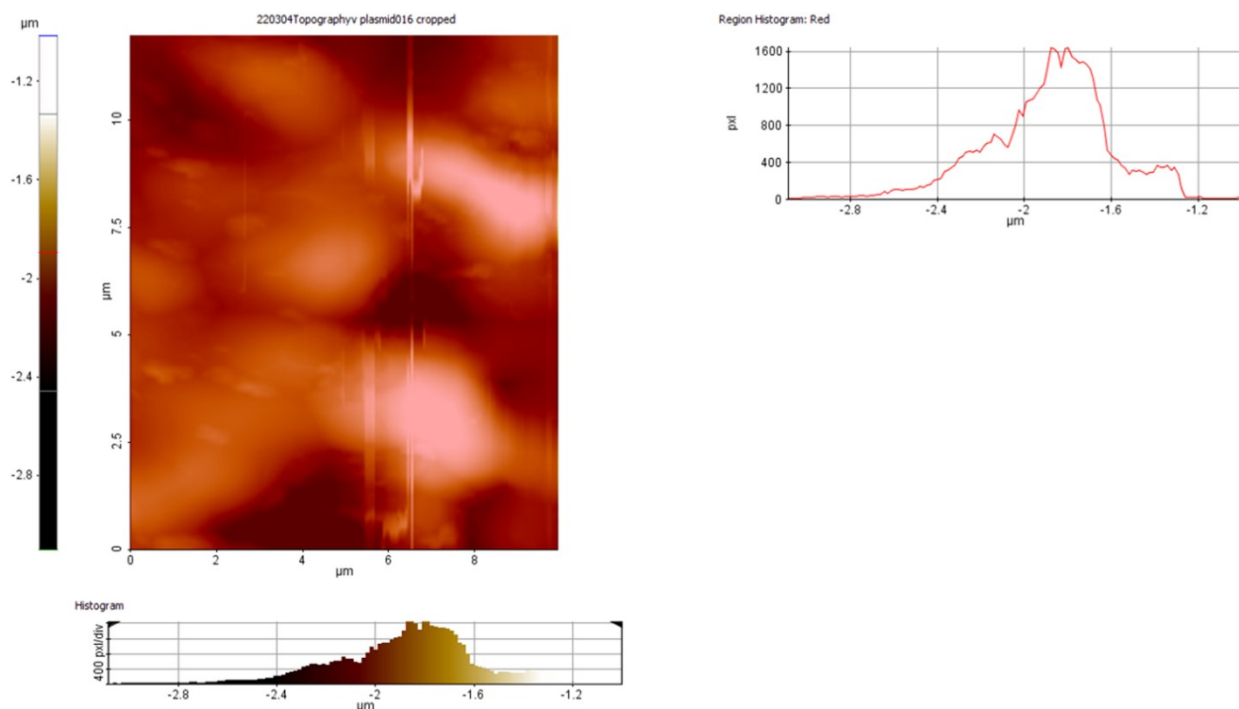
Figure S2: Created baseline for capacitive currents of unactivated SPGE and activated SPGE in a mixture of 10 mM ferrocyanide/ferricyanide at 100 mV/s. The created baselines for both electrodes were obtained from CV responses shown in Figure 2A. The baselines were created using Origin software.



Statistics

| Region | Min(μm) | Max(μm) | Mid(μm) | Mean(μm) | Rpv(μm) | Rq(μm) | Ra(μm) | Rz(μm) | Rsk | Rku |
|---|----------------------|----------------------|----------------------|-----------------------|----------------------|---------------------|---------------------|---------------------|-------|-------|
| <input checked="" type="checkbox"/> Red | -0.330 | 0.387 | 0.029 | 0.000 | 0.717 | 0.100 | 0.076 | 0.690 | 0.140 | 3.447 |

Figure S3: AFM topographical analysis for unactivated SPGE. Analysis was made at 10 x 12 μM .



Statistics

| Region | Min(μm) | Max(μm) | Mid(μm) | Mean(μm) | Rpv(μm) | Rq(μm) | Ra(μm) | Rz(μm) | Rsk | Rku |
|---|---------|---------|---------|----------|---------|--------|--------|--------|-------|-------|
| <input checked="" type="checkbox"/> Red | -3.102 | -1.016 | -2.059 | -1.896 | 2.087 | 0.286 | 0.218 | 2.066 | 0.497 | 3.799 |

Figure S4: AFM topographical analysis for activated SPGE. Analysis was made at 10 x 12 μM.

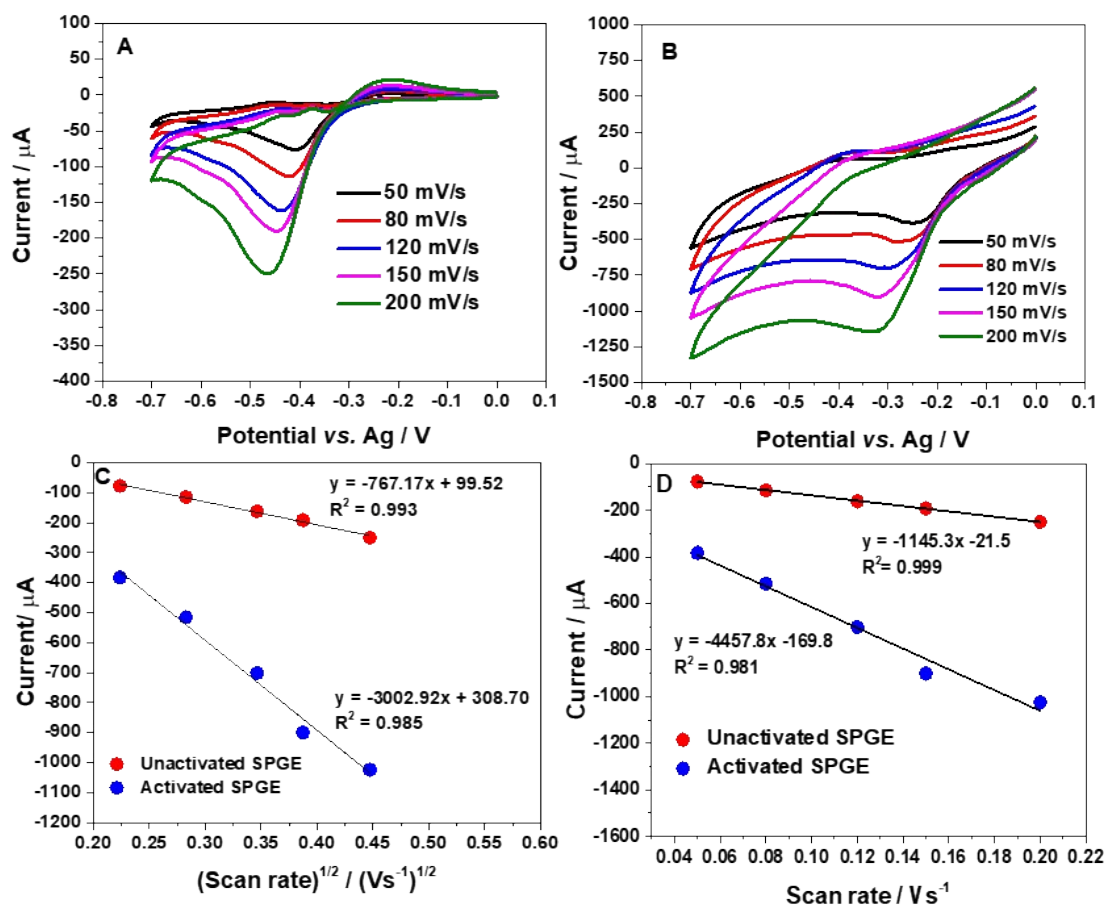


Figure S5: (A) Cyclic voltammograms of unactivated SPGE at different scan rates of 0.05, 0.08, 0.12, 0.15, and 0.2 Vs⁻¹, measured in 0.1 M KCl at saturated O₂. (B) Cyclic voltammograms of activated SPGE at different scan rates of 0.05, 0.08, 0.12, 0.15, and 0.2 Vs⁻¹, measured in 0.1 M KCl at saturated O₂. (C) plots of cathodic peak currents versus square root of scan rates ($v^{1/2}$) and (D) plots of cathodic peak currents versus different scan rates (v).