

SUPPLEMENTARY MATERIAL

Lab-made CO₂ laser-engraved electrochemical sensors for ivermectin determination

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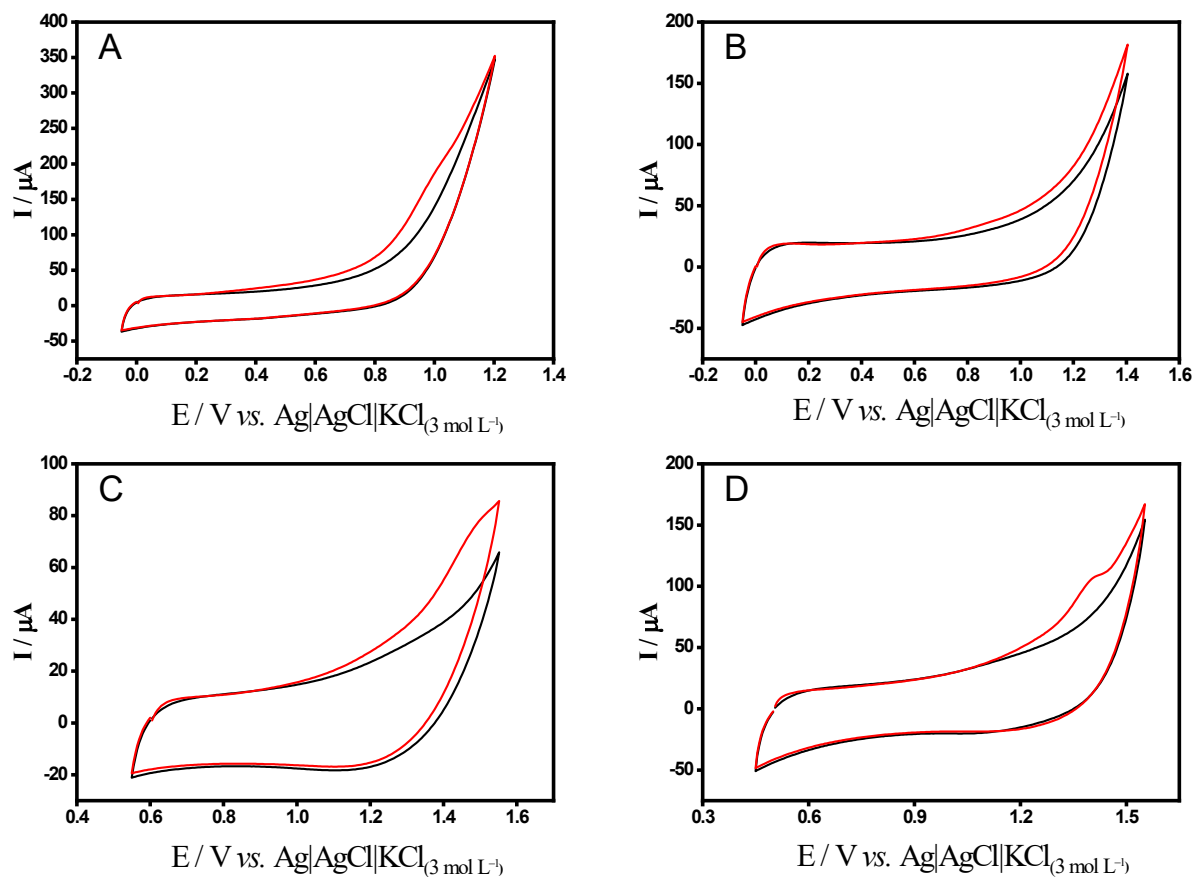


Fig. S1 Cyclic voltammograms recorded using $50 \mu\text{mol L}^{-1}$ IVM in 0.12 mol L^{-1} BR (30% ethanol), (A) pH 12.0, (B) pH 7.0, (C) pH 2.0 and (D) in 0.1 mol L^{-1} H_2SO_4 (30% ethanol) for the LIG electrode. Instrumental conditions: scan rate = 50 mV s^{-1} and step potential = 5 mV s^{-1}

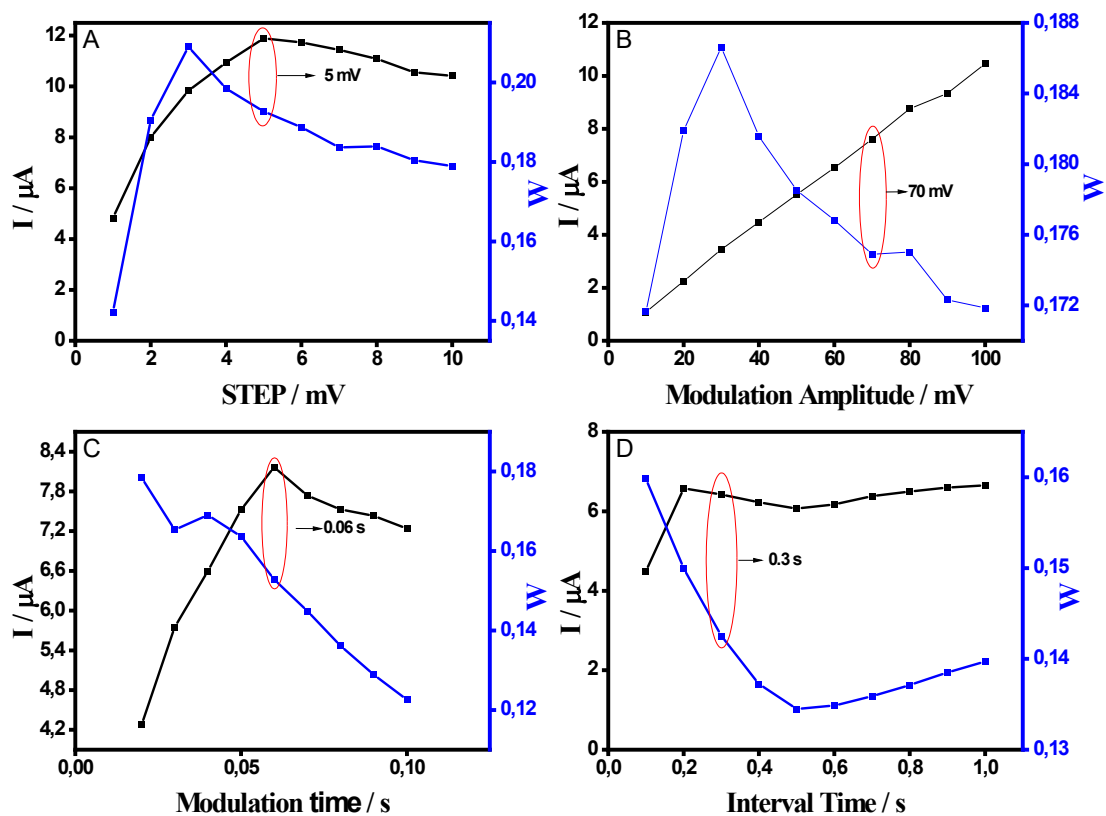


Fig. S2 Optimization of the DPV parameters: (A) Potential Step (1.0 to 10.0 mV), (B) Modulation Amplitude (10.0 to 100.0 mV), (C) Modulation Time (0.02 to 0.10 s), and (D) Interval Time (0.10 to 1.00 s), recorded using $50 \mu\text{mol L}^{-1}$ IVM in 0.1 mol L^{-1} H_2SO_4 (30% ethanol), for the LIG electrode.

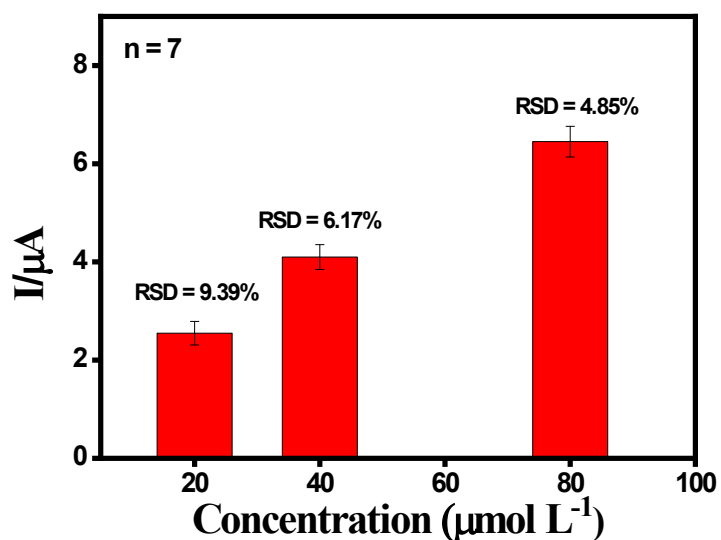


Fig. S3 Repeatability study with consecutive measurements ($n=7$) at different concentrations of IVM (20, 40, and 80 $\mu\text{mol L}^{-1}$). DPV conditions: Potential step: 0.5 mV, modulation amplitude: 70 mV, modulation time: 0.06 s, and interval time: 0.3 s, supporting electrolyte: 0.1 mol L⁻¹ H₂SO₄ solution (30% ethanol, v/v).

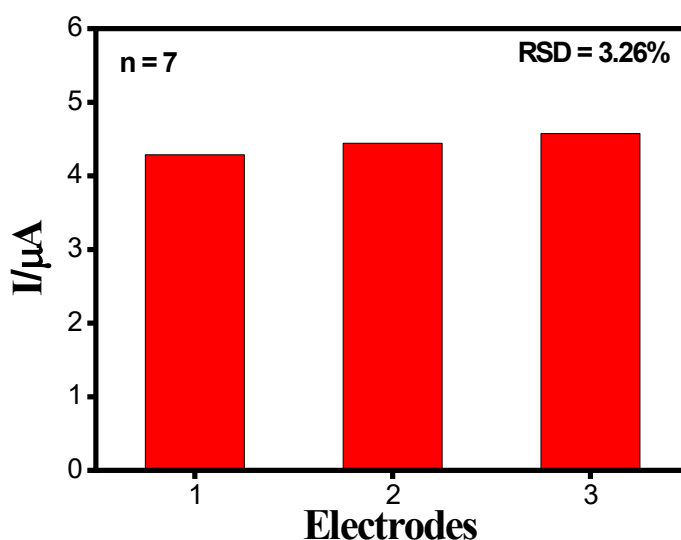


Fig. S4 Reproducibility study with consecutive measurements ($n=7$) at different electrodes, recorded using 40 $\mu\text{mol L}^{-1}$ IVM. DPV conditions: Potential step: 0.5 mV, modulation amplitude: 70 mV, modulation time: 0.06 s, and interval time: 0.3 s, supporting electrolyte: 0.1 mol L⁻¹ H₂SO₄ solution (30% ethanol, v/v).