Supplementary Information

Simultaneous detection of tyrosine and uric acid in sweat by CoWO4@CNT with hydrogel modified electrochemical biosensor

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Materials and instruments

Instruments

The morphology and structure of the synthesized materials were analyzed using the FEI Quattro S (ESEM). The composition and chemical states of the materials were determined using the Thermo Fisher Scientific ESCALAB 250Xi multifunctional X-ray photoelectron spectroscopy (XPS). The electrochemical performance was evaluated using the CHI 660E electrochemical workstation (CH Instruments, Shanghai, China) in a three-electrode configuration through cyclic voltammetry (CV), chronoamperometry (i-t), and differential pulse voltammetry (DPV) methods.

Materials

Uric acid (UA), tyrosine (Tyr) and were purchased from Solarbio (Beijing, China). Glucose (Glu), urea, sodium chloride (NaCl, 99%), potassium chloride (KCl, 99%), calcium chloride (CaCl₂, 99%), methanol, acetone, and carboxylated multi-walled carbon nanotubes (CNT-COOH) were provided by Aladdin. Cobalt nitrate hexahydrate (Co(NO₃)₂-6H₂O), sodium tungstate dihydrate (Na₂WO₄-2H₂O) were obtained from Thermo Fisher Scientific (China). Phosphate buffered saline (PBS) was purchased from Beijing Dingguo Changsheng Biotechnology Co.

	Time (Min)	_					Sweat absorption rate at
DA/DM	Weight	0	10	20	30	40	40 minutes (%)
	(g)						+0 minutes (70)
	8	0.60	0.96	1.11	1.22	1.29	215.47
	4	0.92	1.64	1.87	1.99	2.08	226.04
	2	0.85	1.29	1.42	1.55	1.70	200.57
	0	0.78	1.16	1.33	1.47	1.55	198.47

Table S1 Sweat absorption rate of different DA/DM Hydrogels



Figure S1 HR-TEM images of CoWO₄ (bar: 100 nm)



Figure S2 XPS spectra corresponding to C 1S.



Figure S3 Current at different temperatures accordingly.



Figure S4 Swelling of different DA/DM hydrogels from 0 to 40 minutes of sweat absorption.