

A simple fluorescent probe for selectively detecting Al³⁺ and F⁻ in cells and tea plant roots

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Fig. s1: ¹H NMR, ¹³C NMR spectra of intermediate **BHMP**.

Fig. s2 (a) Changes of the fluorescence emission of **BHMP** + Al³⁺ on addition of different common anions (1. SO₄²⁻ 2. SO₃²⁻ 3. CO₃²⁻ 4. NO₃⁻ 5. F⁻ 6. Cl⁻ 7. Br⁻ 8. I⁻ 9. S²⁻ 10. S₂O₃²⁻ 11. CH₃COO⁻ 12. PO₄³⁻ 13. HPO₄²⁻). (b) Changes of the fluorescence intensity following the addition of F⁻ at different concentrations to the **BHMP**+Al³⁺.

Fig. s3 (a) Job's plot between **BHMP** and Al³⁺ in CH₃CN. (b) Benesi-Hilderbrand plot of **BHMP** with Al³⁺ in CH₃CN.

Fig. s4 Fluorescence response time of **BHMP** to Al³⁺ in CH₃CN.

Fig. s5 HOMO, LUMO, and energy gap ($\Delta E = E_{\text{HOMO}} - E_{\text{LUMO}}$) of **BHMP** in Enol and Keto form calculated using the B3LYP function with 6-311G(d) basis set.

Fig. s6 the pH stability(a) of **BHMP** in PBS with different pH values and kinetic stability(b) in different solvents.

Fig. s7 MTT assays of Hela cells treated with **BHMP** at different concentrations for 24 h.

Table s1 the crystallographic data of **BHMP**

Table s2 Comparison of parameters between **BHMP** and other reported probes for Al³⁺ detection

Scheme s1 the synthetic routes of target compound **BHMP**.

Fig. s1 ^1H NMR and ^{13}C NMR spectra of intermediate **BHMP**.

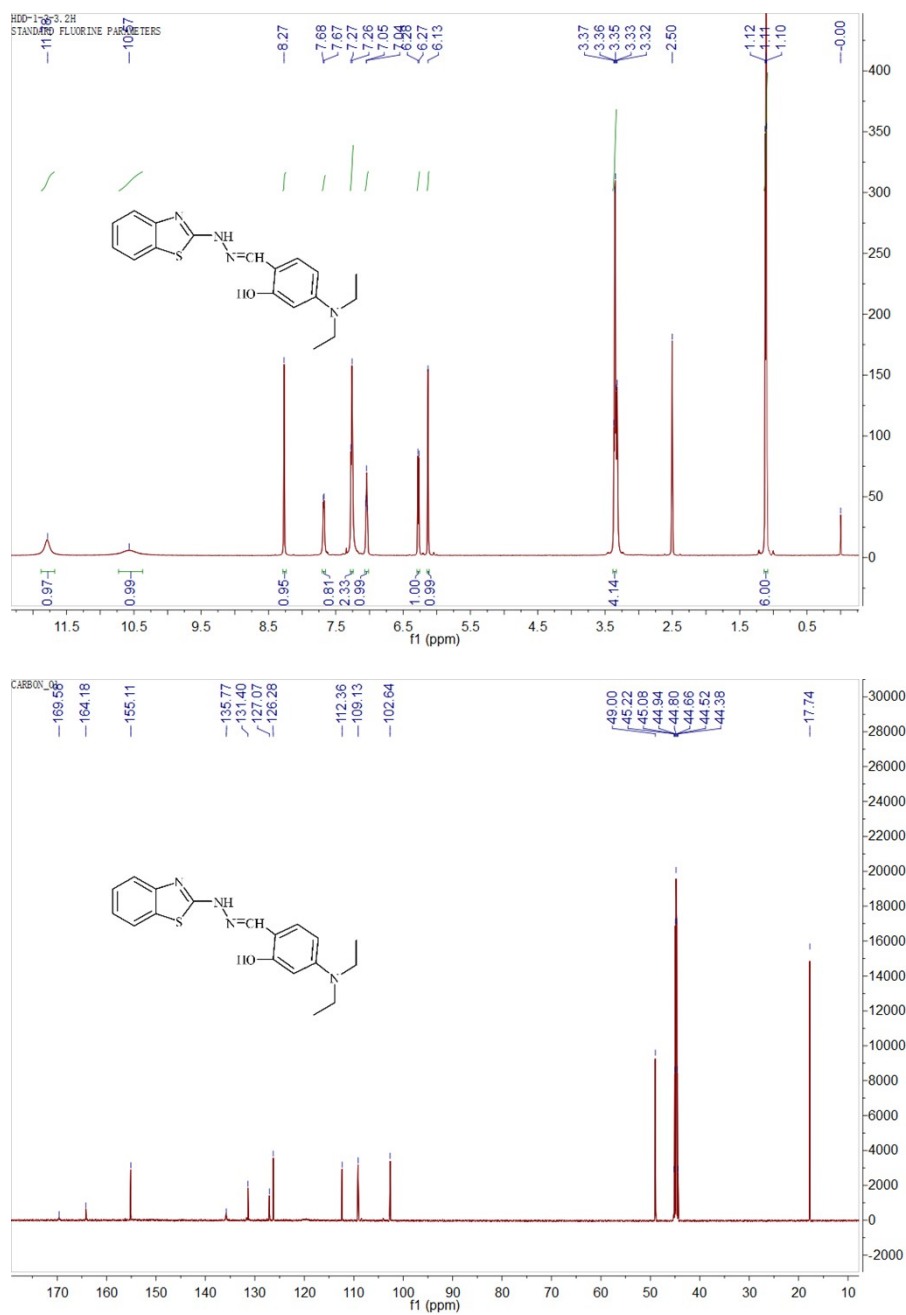


Fig. s2 (a) Changes of the fluorescence emission of **BHMP** + Al^{3+} on addition of different common anions (1. SO_4^{2-} 2. SO_3^{2-} 3. CO_3^{2-} 4. NO_3^- 5. F^- 6. Cl^- 7. Br^- 8. I^- 9. S^{2-} 10. $\text{S}_2\text{O}_3^{2-}$ 11. CH_3COO^- 12. PO_4^{3-} 13. HPO_4^{2-}). (b) Changes of the fluorescence intensity following the addition of F^- at different concentrations to the **BHMP**+ Al^{3+} .

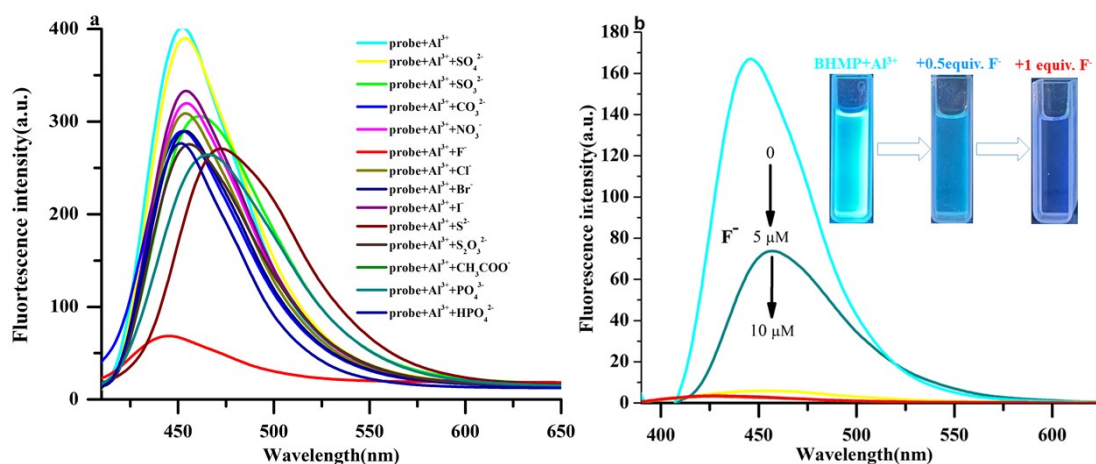


Fig. s3(a) Job's plot between **BHMP** and Al^{3+} in CH_3CN . (b) Benesi-Hilderbrand plot of **BHMP** with Al^{3+} in CH_3CN .

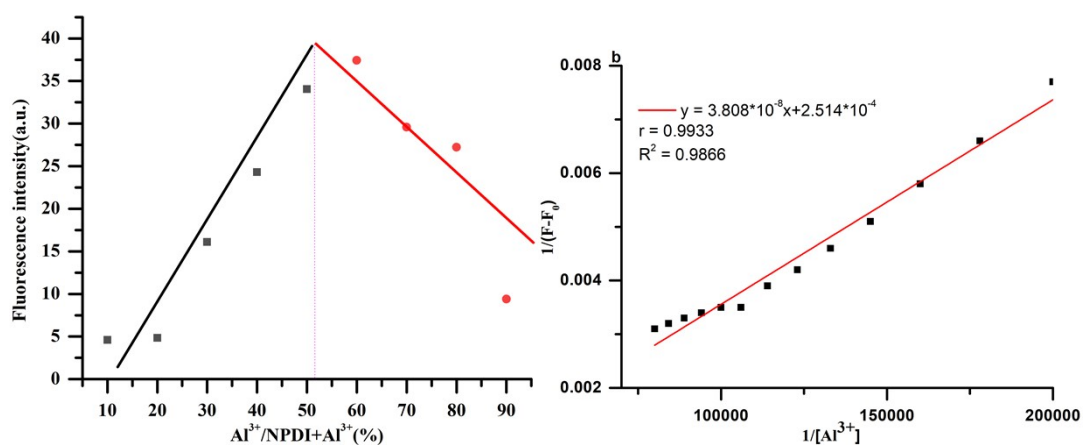


Fig. s4 Fluorescence response time of **BHMP** to Al^{3+} in CH_3CN

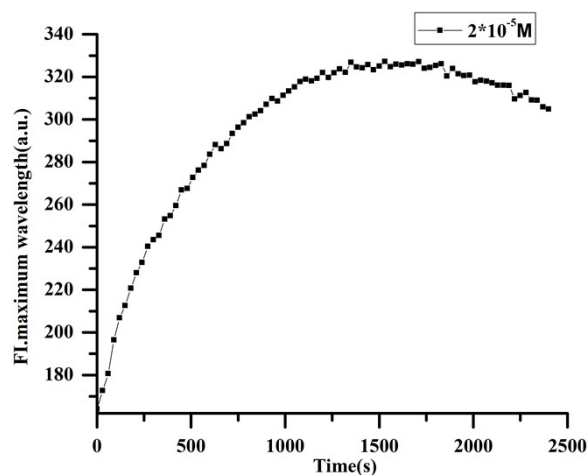


Fig. s5 HOMO, LUMO, and energy gap ($\Delta E = E_{\text{HOMO}} - E_{\text{LUMO}}$) of **BHMP** in Enol and Keto form calculated using the B3LYP/6-311G(d) basis set.

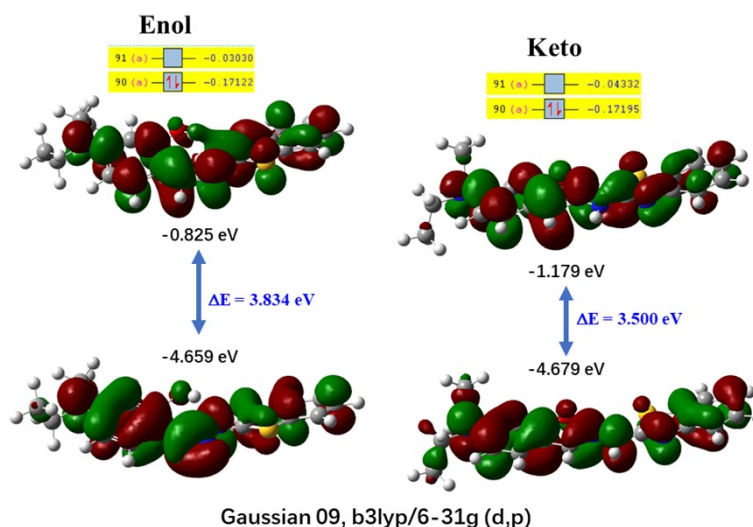


Fig. s6 the pH stability(a) of **BHMP** in PBS with different pH values and kinetic stability(b) of **BHMP** in different solvents.

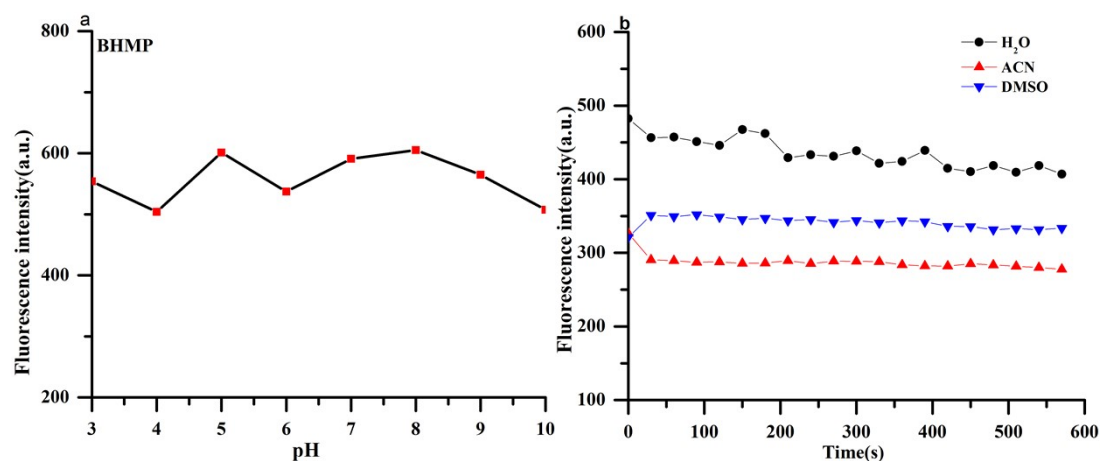


Fig. s7 MTT assays of HeLa cells treated with **BHMP** at different concentrations for 24 h.

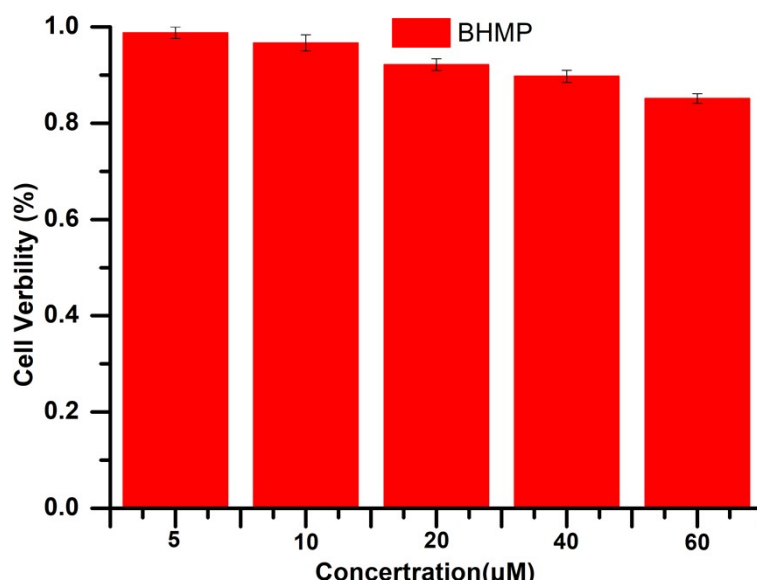


Table s1: The crystallographic data of **BHMP**

Compounds	BHMP
Empirical formula	C ₁₈ H ₂₀ N ₄ OS
Formula weight	340.44
Crystal system	Orthorhombic
Space group	Pccn
a[Å]	18.512(3)
b[Å]	27.014(4)
c[Å]	7.4371(12)
α[°]	90
β[°]	90
γ[°]	90
V[Å ³]	3719.2(10)
Z	8
T[K]	296
D _{calcd} [g·cm ⁻³]	1.216
F(000)	1440.0
μ[mm ⁻¹]	0.185
θrange[°]	2.20-27.53
Total no. data	4014
No. unique data	1934
R _{int}	0.1512
R ₁	0.0757
wR ₂	0.2646
GOF	1.052

Table s2: Comparison of parameters between **BHMP** and other probes for Al³⁺ detection

Probe	λ _{ex} (nm)	Medium (v:v)	Practical application	LOD (mol.L ⁻¹)	Ref.
Flu-py	523	Tris-HCl	Human cervical carcinoma cells	0.92×10 ⁻⁷ 1.12×10 ⁻⁷	[8]
AHN	425	CH ₃ CN		1.148×10 ⁻⁶ 1.44×10 ⁻⁶	[9]
SPBH	478	DMF:H ₂ O(1:1)	Human non-small cell lung cancer H460 cells	1.1×10 ⁻⁷ 1.47×10 ⁻⁶	[10]
Probe1	540	CH ₃ CN:H ₂ O(9:1)	Water containing surfactant SDBS	3.2×10 ⁻⁸ 7.5×10 ⁻⁷	[12]
H-2L	607	DMSO:H ₂ O(1:5)	A549 cells	1.05×10 ⁻⁸ 0.98×10 ⁻⁸	[18]
BHMP	490	CH ₃ CN	Hela cells Tea plant roots	1.04×10 ⁻⁸	This work

Scheme s1: The synthetic routes of target compound **BHMP**.

