Supporting Information

External photocatalyst-free photocycloaddition between triplet vinylnitrenes with 1,3biradical character and activated olefins under 420 nm LEDs

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Experimental Section:

General Information: All reactions were achieved in oven-dried glassware. Reagents were bought at the very best commercial quality and used as obtained, unless otherwise certain. Reactions were monitored by using thin layer chromatography (TLC)accomplished on silica gel plates (Merck silica gel 60, f₂₅₄); the spots have been visualized with UV light (254 and 365 nm) and a solution of 5% H₂SO₄-MeOH or vanillin charring solution as developing. Flash column chromatography turned into performed using 230-400 mesh silica gel. Yields confer with isolated yields after chromatographic purification.¹H NMR (600 400 and 300 MHz) and ¹³C NMR (150 and 101 MHz) spectra were recorded in CDCl₃ solvent and are reported relative to the residual solvent signal. Chemical shift (ppm), multiplicity, coupling constant (J in Hz), and integration data are presented for ¹H-NMR spectra. Chemical shift is used to describe ¹³C-NMR data, and multiplicity and coupling constant are used to describe compounds containing fluorine (*J*_{C-F} in Hz). The following abbreviations were used to explain NMR peak multiplicities: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet. High-resolution mass spectra (HRMS, m/z) were recorded using ESI (Q-TOF and Orbitrap, positive ion) mode. Infrared (IR) spectra were recorded on Fourier transform infrared spectroscopy; only intense peaks were reported in terms of frequency of absorption (cm⁻¹). Single-crystal X-ray data were recorded in a diffractometer with MoKa radiation. Melting points were determined in open-end capillary tubes and are uncorrected.

Lamp Type	Luzchem LZC-420	Internal surface	Unpolished scattering aluminum
Filter	None	Chamber Temp.	25°C
Filter effect	N/A	Measurement date	March 7, 2016
Photo reactor model	LZC- ICH2	Monitored range	235 to 850 nm
Number of lamps	8 over head lamps, Side lamps not used.	Harmonic peak interference	None observed
Measurement distance	~18 cm lamp to target	Resolved peaks	419 (broad), 434, 546, 578, 812, 842nm







ESI-01: The safety issues for handling azido compounds¹ Sodium azide (NaN₃)

Sodium azide is toxic (LD_{50} oral = 27 mg/kg for rats) and can be absorbed through the skin. Appropriate gloves are necessary when using it. It is decomposed explosively upon heating to above 275 °C. Sodium azide is relatively safe especially in aqueous solution, unless acidified to form HN₃, which is volatile and highly toxic. 2.2.

Organic azide

Organic azides are potentially explosive substances that decompose with the slight input of energy from external sources (heat, light, pressure, etc). When the designed organic azides are used for the project, we consider the following equation. It is noted that this equation takes into account all nitrogen atoms in the organic azide, not just those in the azido group. $\frac{Nc+No}{Nn} \ge 3$ (N: number of the atom, c= carbon, o= oxygen, n=nitrogen)

All organic azides are enough stable to be stored under -20°C for at least 6 months.

SI-02: General scheme for the synthesis of starting materials:

ESI-02-01: Synthetic procedures for vinyl azide:

1. Sodium azide (1.0 eq, 10 mmol) was dissolved in distilled CH₃CN (20 mL) and cooled at - 20 °C, then solution of iodine mono chloride (1.5 eq, 15 mmol) in CH₂Cl₂ (10 mL) was added dropwise. The mixture was stirred at the same temperature. After 90 min, a solution of alkene (1.0 eq, 10 mmol) in CH₂Cl₂ (8 mL) was added slowly, and the mixture was stirred for 1.0 h. The reaction was quenched with saturated aqueous Na₂S₂O₃, and the organic materials were extracted two times with Et₂O. The combined extracts were

¹ (a) See http://www.ehs.ucsb.edu/units/labsfty/labrsc/factsheets/Azides_FS26.pdf. Date accessed: 01-Aug-2017. (b) Kolb, H. C.; Finn, M. G.; Sharpless, K. B. Click Chemistry: Diverse Chemical Function from a Few Good Reactions. *Angew. Chem. Int. Ed.* **2001**, *40*, 2004.

washed with brine and dried over MgSO₄. After the evaporation of solvents, the resulting crude materials were used immediately for the next step without any further purification.



To a solution of the obtained compounds above in Et_2O (25 mL) was added ^tBuOK (1.5 eq,15.0 mmol) at 0 °C, and the mixture was stirred for 1.5 h at the same temperature. The reaction mixture was filtered through a celite pad, and the solvent was removed in vacuo. The resulting crude materials were purified by flash column chromatography (silica gel/*n*-hexane) to give vinyl azides.²

2. To a solution of alkyne (0.5 mmol, 1.0 eq), TMS-N₃ (0.132 mL, 1.0 mmol) and H₂O (18 μ L, 1.0 mmol) in DMSO (2 mL), Ag₂CO₃ (13.8 mg, 0.05 mmol) was added. The mixture was then heated at 80 °C with an oil bath for 1.0-2.0 h until the substrate was consumed as indicated by TLC. The reaction mixture was cooled to room temperature and taken up by dichloromethane. The organic layer was washed with brine, dried over MgSO₄, and concentrated. Purification of the crude product with flash column chromatography (silica gel) and concentration in vacuo.³



1a-e was synthesized by process 1 and 1f was synthesized by process 2

Used vinyl azide:



ESI-02-02: Synthetic procedures for α-Ketomethylthioesters (2a-k, 2u-y):

PPh₃•HBr (2 equiv/mmol) was taken in 25 mL two necked round bottom flask under argon atmosphere with condenser. Dry DMSO (2 mL) was added dropwise into it. Corresponding benzylidene acetone (1equiv) was introduced separately into the reaction mixtures at room temperature, and the mixtures were stirred at the same temperature or heated with stirring After completion of the reaction (TLC), saturated NH₄Cl solution was added, and the product

² Zhang, F.-L.; Feng, Yi.; Wang, Chiba, S. Amide Synthesis by Nucleophilic Attack of Vinyl Azides. *Angew. Chem., Int. Ed.* **2014**, *53*, 4390.

³ Liu, Z. - H.; Liao, P.-Q.; Bi, X.-H. General Silver-Catalyzed Hydroazidation of Terminal Alkynes by Combining TMS-N₃ and H₂O: Synthesis of Vinyl Azides. *Org. Lett.* **2014**, *16*, 3668.



Used conjugated polarised alkenes:

was extracted with EtOAc. The combined organic layers were dried over anhydrous Na_2SO_4 and filtered, and the filtrate was concentrated under reduced pressure to get a residue. The crude residue was purified over silica gel column chromatography [230-400; eluent: ethyl acetate/*n*-hexane] to obtain **2a-k**, **2u-y**.⁴

As reported in previous literature, **2I-n**⁵, **2o**⁶, **2p**⁷, **2z**⁸ and **2aa**⁹ were synthesized.

ESI-03: General procedure for the synthesis of 3a-3ad:

Under an argon atmosphere, conjugated enone (50 mg, 1 equiv) was added to a solution of vinyl azide (1.2 equiv) in acetonitrile (0.1 M). It was then irradiated with a 420 nm LED at room temperature (25 °C). After the completion reaction (TLC), the crude residue was purified by silica gel column chromatography [230–400 mesh; eluent: ethyl acetate/*n*-hexane] to obtain **3a-3ad**.



ESI-04: General procedure for the synthesis of 4a-4ae:

Under an argon atmosphere, conjugated enone (1 equiv) was added to a solution of vinyl azide (1.2 equiv) in acetonitrile (0.1 M). It was then irradiated with a 420 nm LED at room temperature (25 °C). After completion of the reaction (TLC), 2,3-dichloro-5,6-dicyano-1,4-benzoquinone, DDQ (1.2 eq) was added to the same reaction mixture and stirred at room temperature, maintaining time as mentioned. After completion of the reaction (TLC), the crude residue was purified by silica gel column chromatography [230–400 mesh; eluent: ethyl acetate/n-hexane] to obtain **4a-4ae**.

⁴ Mal, K.; Sharma, A.; Maulik, P. R. ; Das, I. PPh₃·HBr-DMSO Mediated Expedient Synthesis of γ-Substituted β , γ-Unsaturated α -Ketomethylthioesters and α -Bromo Enals: Application to the Synthesis of 2-Methylsulfanyl-3(*2 H*)-furanones. *Chem. Eur. J.* **2014**, *20*, 662–667.

⁵ Modrocká, V.; Veverková, E.; Mečiarová, M.; Šebesta, R. Bifunctional Amine-Squaramides as Organocatalysts in Michael/Hemiketalization Reactions of β, γ-Unsaturated α-Ketoesters and α, β-Unsaturated Ketones with 4-Hydroxycoumarins. J. Org. Chem. **2018**, 83, 13111–13120.

⁶ Liu, J.; Das, D. K.; Zhang, G.; Yang, S.; Zhang, H.; Fang, X. N-Heterocyclic Carbene-Catalyzed Umpolung of β, γ-Unsaturated 1,2-Diketones. *Org. Lett.* **2018**, *20*, 64–67.

⁷ Goudedranche, S.; Bugaut, X.; Constantieux, T.; Bonne, D.; Rodriguez, J. α, β-Unsaturated Acyl Cyanides as New Bis-Electrophiles for Enantioselective Organocatalyzed Formal [3+3] Spiroannulation. *Chem. Eur.* J. **2014**, 20, 410-415.

⁸ Bleger, D.; Hecht, S. Visible-Light-Activated Molecular Switches. *Angew. Chem., Int. Ed.* **2015**, *54*, 11338–11349.

⁹ Crotti, S.; Belletti, G.; Di Iorio, N.; Marotta, E.; Mazzanti, A.; Righi, P.; Bencivenni, G. Asymmetric Vinylogous Aldol Addition of Alkylidene Oxindoles on Trifluoromethyl-α, β-Unsaturated Ketones. *RSC Adv.* **2018**, *8*, 33451-33458.



ESI-05: Optimization of the reaction conditions: ESI-05-01: Screening of solvents and light:



^aReaction Conditions: **1a** (0.13 mmol) and **2a** (0.11 mmol) in solvent 0.1 M concentration was irradiated under a LED at room temperature (25 °C) under argon atmosphere. ^bIsolated yield. ^cDecomposition. ^dUnreacted **2a** isolated.

ESI-05-02: Screening of Concentration:

Entries	Volume	Concentration	Yeild (%) ^b
1	1 ml	0.1 M	94
2	2 ml	54 mM	93
3	3 ml	36 mM	85
4	4 ml	27 mM	81
5	5 ml	21 mM	76

^aReaction Conditions: **1a** (0.13 mmol) and **2a** (0.11 mmol) in mentioned solvent was irradiated under a LED at room temperature (25 °C) under argon atmosphere. ^bIsolated yeild.

Sun light Reaction



ESI-06: Analytical and spectral data of 3a



S-methyl



2-(2-(3,5-dichlorophenyl)-5-phenyl-3,4-dihydro-2*H*-pyrrol-3-yl)-2-oxoethanethioate 3a: Prepared according to the general procedure discussed above: reaction time, 10 h; *R*_f = 0.2; eluent, EtOAc/*n*-hexane (5%); light yellow liquid (67 mg, 94%); ¹H NMR (400 MHz, CDCl₃): δ = 7.92 - 7.90 (m, 2 H), 7.51 - 7.43 (m, 3 H), 7.27 (t, *J* = 2.0 Hz, 1 H), 7.17 (d, *J* = 2.0 Hz, 2 H), 5.55 (dt, *J* = 6.0, 2.0 Hz, 1 H), 3.90 (ddd, *J* = 10.0, 6.4, 5.6 Hz, 1 H), 3.50 (ddd, *J* = 17.2, 10.0, 2.0 Hz, 1 H), 3.34 (ddd, *J* = 17.2, 6.4, 1.2 Hz, 1 H), 2.40 (s, 3 H) ppm; ¹³C{¹H} NMR (101

MHz, $CDCl_3$): δ = 193.5, 191.6, 172.5, 145.7, 135.3, 133.0, 131.6, 128.8 (2 CH), 128.2 (3 CH), 127.9, 125.4 (2 CH), 76.3, 52.1, 38.5, 11.6 ppm; HRMS (ESI-QTOF): *m/z* calcd for $C_{19}H_{16}Cl_2NO_2S [M + H]^+$: 392.0279; found: 392.0277.









Supporting Information



ESI-07: Analytical and spectral data of 3b



S-methyl 2-(2-(4-chlorophenyl)-5-phenyl-3,4-dihydro-2H-pyrrol-3-yl)-2-oxoethanethioate



3b: Prepared according to the general procedure discussed above: reaction time, 12 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (5%); Light yellow liquid (55 mg, 74%). ¹H NMR (400 MHz, CDCl₃): $\delta = 7.93 - 7.91$ (m, 2 H), 7.50 - 7.43 (m, 3 H), 7.31 (d, J = 8.8 Hz, 2 H), 7.23 (d, J = 8.4 Hz, 2 H).5.58 (d, J = 5.6 Hz, 1 H), 3.95 - 3.90 (m, 1 H), 3.46 (ddd, J = 17.2, 9.6,

2.0 Hz, 1 H), 3.37 (ddd, J = 17.2, 6.4, 1.2 Hz, 1 H), 2.39 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta = 193.9$, 191.7, 171.9, 140.8, 133.5, 133.3, 131.4, 128.9 (2 CH), 128.8 (2 CH), 128.1 (4 CH), 77.0, 52.23, 38.3, 11.5 ppm; HRMS (ESI-QTOF): m/z calcd for C₁₉H₁₇ClNO₂S [M + H]⁺: 358.0668; found: 358.0668.

^1H (400 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl3) NMR spectra of 3b











S-methyl 2-(2-(4-nitrophenyl)-5-phenyl-3,4-dihydro-2H-pyrrol-3-yl)-2-oxoethanethioate



3c: Prepared according to the general procedure discussed above: reaction time, 8 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (15%); yellow gum (47 mg, 64%); ¹H NMR (400 MHz, CDCl₃): $\delta = 8.21$ (d, J = 8.8 Hz, 2 H), 7.94 (d, J = 8.2 Hz, 2 H), 7.55 – 7.49 (m, 3 H), 7.48 – 7.45 (m, 2 H), 5.71 (d, J = 5.6 Hz, 1 H), 3.93 (ddd, J = 9.6, 6.8, 6.0 Hz, 1 H), 3.51 (ddd, J = 17.2, 9.6, 2.0 Hz, 1 H), 3.41 (ddd, J = 17.2, 9.6, 9.8 Hz, 9.8 H

17.2, 6.8, 1.6 Hz, 1 H), 2.41 (s, 3 H) ppm; ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃): δ = 193.5, 191.7, 172.8, 149.6, 147.5, 133.0, 131.7, 128.9 (2 CH), 128.2 (2 CH), 127.7 (2 CH), 124.1 (2 CH), 76.7, 52.1, 38.5, 11.6 ppm; HRMS (ESI-QTOF): m/z calcd for C₁₉H₁₇N₂O₄S [M + H]⁺: 369.0909; found: 369.0908.

^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 3c

SM-02-47A-01 single_pulse



ESI-09: Analytical and spectral data of 3d



S-methyl 2-(2-(3-bromophenyl)-5-phenyl-3,4-dihydro-2H-pyrrol-3-yl)-2-oxoethanethioate



3d: Prepared according to the general procedure discussed above: reaction time, 12 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (5%); Light yellow liquid (54 mg, 74%); ¹H NMR (400 MHz, CDCl₃): $\delta = 7.93 - 7.91$ (m, 2 H), 7.50 - 7.40 (m, 5 H), 7.23 - 7.21 (m, 2 H), 5.59 (dt, J = 5.6, 1.6 Hz, 1H), 3.94 (ddd, J = 9.6, 6.4, 5.6 Hz, 1 H), 3.50 (ddd, J = 17.6, 10.0, 2.4 Hz, 1H), 3.35 (ddd, J = 17.2, 6.4, 1.2 Hz, 1 H), 2.40 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta = 193.8$, 191.7, 172.1, 144.6, 133.2, 131.5, 130.8,

130.4, 129.75, 128.8 (2 CH), 128.2 (2 CH), 125.5, 122.9, 76.9, 52.2, 38.4, 11.5 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₁₉H₁₇BrNO₂S [*M* + H]⁺: 402.0163; found: 402.0161.

^1H (400 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl3) NMR spectra of 3d



ESI-10: Analytical and spectral data of 3e



S-methyl



2-oxo-2-(5-phenyl-2-(p-tolyl)-3,4-dihydro-2H-pyrrol-3-yl)ethanethioate 3e: Prepared according to the general procedure discussed above: reaction time, 12 h; $R_f = 0.2$; eluent, EtOAc/nhexane (5%); Light yellow liquid (53 mg, 70 %). ¹H NMR (400 MHz, CDCl₃): δ = 7.96 – 7.90 (m, 2 H), 7.49 – 7.40 (m, 3 H), 7.18 – 7.13 (m, 4 H), 5.59 (d, J = 5.6 Hz, 1 H), 3.98 (ddd, J = 9.2, 6.0, 5.2 Hz, 1 H), 3.46 (ddd, J = 17.2, 9.6, 2.4 Hz, 1 H), 3.36 (ddd, J = 17.2, 6.4, 1.6 Hz, 1 H), 2.39 (s, 3 H),

2.34 (s, 3 H) ppm; ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃): δ = 194.2, 191.8, 171.3, 139.3, 137.4, 133.5, 131.2, 129.5 (2 CH), 128.7 (2 CH), 128.1 (2 CH), 126.6 (2 CH), 77.7, 52.2, 38.2, 21.2, 11.5 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₀H₂₀NO₂S [M + H]⁺: 338.1214; found: 338.1217.

^1H (400 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl3) NMR spectra of 3d

JAY-2-34B single_pulse



JAY-2-34B single pulse decoupled gated NOE









S-methyl



2-(2-(3-methoxyphenyl)-5-phenyl-3,4-dihydro-2*H*-pyrrol-3-yl)-2-oxoethanethioate 3f: Prepared according to the general procedure discussed above: reaction time, 16 h; *R*_f = 0.2; eluent, EtOAc/*n*-hexane (5%); Light yellow liquid (60 mg, 81%). ¹H NMR (400 MHz, CDCl₃): δ = 7.95 - 7.92 (m, 2 H), 7.50 - 7.42 (m, 3 H), 7.26 (t, *J* = 8.0 Hz, 1 H), 6.88 - 6.81 (m, 3 H), 5.61 (dt, *J* = 5.6, 2.0 Hz, 1 H), 3.99 (ddd, *J* = 10.0, 6.4, 5.6 Hz, 1 H), 3.80 (s, 3 H), 3.48 (ddd, *J* = 17.2, 10.0, 2.0 Hz, 1 H), 3.35 (ddd, *J* = 17.2, 6.4, 1.6 Hz, 1 H), 2.39 (s, 3 H) ppm; ¹³C{¹H} NMR

(101 MHz, CDCl₃): δ = 194.1, 191.7, 171.5, 160.0, 143.9, 133.4, 131.3, 129.8, 128.7 (2 CH), 128.2 (2 CH), 119.0, 113.0, 112.5, 77.6, 55.3, 52.1, 38.4, 11.5 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₀H₂₀NO₃S [M + H]⁺: 354.1164; found: 354.1158.

^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 3f



ESI-12: Analytical and spectral data of 3g



S-methyl 2-oxo-2-(5-phenyl-2-(o-tolyl)-3,4-dihydro-2*H*-pyrrol-3-yl)ethanethioate 3g: Prepared according to the general procedure discussed above: reaction time, 12 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (5%); Light yellow liquid (56 mg, 73%). ¹H NMR (400 MHz, CDCl₃): δ = 7.93 - 7.90 (m, 2 H), 7.49 - 7.42 (m, 3 H), 7.20 - 7.14 (m, 3 H),7.05 - 7.03 (m, 1 H), 5.92 (dt, J = 5.6, 1.6 Hz, 1H), 4.03 - 3.97 (m, 1 H), 3.59 (ddd, J = 17.6, 10.8, 2.4 Hz, 1H), 3.28 (ddd, J = 17.2, 6.4, 1.2 Hz, 1 H), 2.37 (s, 3 H), 2.35 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 194.6, 191.7, 170.6, 140.4, 135.3, 133.5, 131.2, 130.8, 128.7 (2

CH), 128.1 (2 CH), 127.6, 126.5, 126.5, 74.7, 51.0, 39.7, 19.8, 11.5 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₀H₂₀NO₂S [M + H]⁺: 338.1214; found: 338.1211.

^1H (400 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl3) NMR spectra of 3g









NOESY, COSY, HSQC, and HMBC (400 MHz, CDCl₃) NMR spectra of 3g



MeSOC

ESI-13: Analytical and spectral data of 3h:



S-methyl 2-(2,5-diphenyl-3,4-dihydro-2H-pyrrol-3-yl)-2-oxoethanethioate 3h: Prepared according to the general procedure discussed above: reaction time, 14 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (5%); Light yellow solid (56 mg, 71%); mp 40-45 °C; solvent of crystallization, nhexane/EtOAc = 2 mL:1 mL. ¹H NMR (400 MHz, CDCl₃): δ = 7.95 - 7.93 (m, 2 H), 7.51 - 7.42 (m, 3 H), 7.35 - 7.27 (m, 5 H), 5.63 (d, J = 5.2 Hz, 1 H), 3.99 (ddd, J = 9.6, 6.4, 5.2 Hz, 1 H), 3.48 (ddd, J = 17.2, 9.6, 2.0 Hz, 1 H), 3.37 (ddd, J = 17.2, 6.4, 1.2 Hz,

1 H), 2.39 (s, 3 H) ppm; ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃): δ = 194.1, 191.8, 171.5, 142.2, 133.5, 131.3, 128.8 (2 CH), 128.7 (2 CH), 128.1 (2 CH), 127.7, 126.7 (2 CH), 77.8, 52.2, 38.3, 11.5 ppm; HRMS (ESI-QTOF): m/z calcd for C₁₉H₁₈NO₂S [M + H]⁺: 324.1058; found: 324.1071.



X-ray determined molecular structure of **3h**; **CCDC 2347214**.

Datablock: 3h

Bond precisio	n: C-C	= 0.0026 A	Wavelength=1.54178
Cell:	a=8.7535(9)	b=9.7512(10)	c=11.0588(12)
	alpha=108.448	(4) beta=99.312(4)	gamma=107.231(3)
Temperature:	100 K		
	Calcul	ated	Reported
Volume	820.5	8(15)	820.53(15)
Space group	P -1		P -1
Hall group	-P 1		-P 1
Moiety formu	la C19 H	17 N O2 S	C19 H17 N O2 S
Sum formula	C19 H	17 N O2 S	C19 H17 N O2 S
Mr	323.4)	323.39
Dx,g cm-3	1.309		1.309
Z	2		2
Mu (mm-1)	1.821		1.821
F000	340.0		340.0
F000'	341.5	7	
h,k,lmax	10,12,	13	10,12,13
Nref	3225		3125
Tmin <i>,</i> Tmax	0.729,	0.721	0.570,0.754
Tmin'	0.662		
Correction me AbsCorr = MU	thod= # Report LTI-SCAN	ed T Limits: Tmin=	0.570 Tmax=0.754
Data complete	eness= 0.969	Theta(ma	x)= 72.101
R(reflections)=	= 0.0439(3036)		wR2(reflections)= 0.1155(3125)
S = 1.053	N	oar= 210	

The following ALERTS were generated.

Alert level C

PLAT369
ALERT 2 C Long C(sp2)-C(sp2) Bond C17
- C18
1.55 Ang.

PLAT767
ALERT 4 C
INS Embedded LIST 6 Instruction Should be LIST 4
Please Check

PLAT906
ALERT 3 C
Large K Value in the Analysis of Variance
2.118 Check

PLAT911
ALERT 3 C
Missing FCF Refl Between Thmin & STh/L=
0.600
49 Report

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49 Report

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8 -8 6, 0 -3 6, -3 -1 7, -9 3 7, -9 3 8, -2 -9 9, 2 1 10, <u>PLAT913 ALERT 3 C</u> Missing # of Very Strong Reflections in FCF 18 Note 2 0 0, 1 -2 1, 2 0 1, 1 1 1, 0 -4 2, -1 -3 2, -1 -2 2, -1 -1 2, 1 -1 2, 2 -1 2, 4 -1 2, -4 2 2, -3 -2 3, -1 -2 3, 4 -1 3, 3 0 3, -2 -4 4, -2 1 4,

Alert level G

PLAT883 ALERT 1 G No Info/Value for _atom_sites_solution_primary .	Please Do !
PLAT910 ALERT 3 G Missing # of FCF Reflection(s) Below Theta(Min).	2 Note
0 1 0, 0 0 1,	
<u>PLAT912_ALERT_4_G</u> Missing # of FCF Reflections Above STh/L= 0.600	49 Note
PLAT961_ALERT_5_G Dataset Contains no Negative Intensities Ple	ase Check
<u>PLAT969_ALERT_5_G</u> The 'Henn et al.' R-Factor-gap value 3.01	Note
Predicted wR2: Based on Sigl**2 3.84 or SHELX Weight 11.35	
PLAT978 ALERT 2 G Number C-C Bonds with Positive Residual Density.	13 Info

ESI-13-01: Gram Scale Reaction:





1	1a	0.84 g	5.82 mmol
2	2h	1 g	4.85 mmol
3	3h	0.98 g	3.030 mmol

Following the standard reaction conditions, the reaction mixture was irradiated and purified by column chromatography after completion. MeCN taken: 40 mL Reaction time: 20 h

Product Yield: 62%

^1H (400 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl₃) NMR spectra of **3h**:



ESI-14: Analytical and spectral data of 3i



S-methyl 2-(2-(naphthalen-2-yl)-5-phenyl-3,4-dihydro-2H-pyrrol-3-yl)-2-oxoethanethioate



3i: Prepared according to the general procedure discussed above: reaction time, 12 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (10%); light yellow gum (55 mg, 76%). ¹H NMR (400 MHz, CDCl₃): $\delta = 7.98$ (d, J = 6.4 Hz, 2 H), 7.83 – 7.808 (m, 3 H), 7.72 (s, 1 H), 7.52 – 7.40 (m, 6 H), 5.80 (d, J = 5.6 Hz, 1 H), 4.07 (dt, J = 9.7, 6.0 Hz, 1 H), 3.56 – 3.37 (m, 2 H), 2.39 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta = 194.2$, 191.8, 171.7, 139.6, 133.5, 133.0, 131.3, 128.8 (4 CH), 128.2

(2 CH), 128.1, 127.8, 126.3, 126.0, 125.4, 124.8, 77.9, 52.2, 38.4, 11.5 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₃H₂₀NO₂S [M + H]⁺: 374.1214; found: 374.1223.

^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 3i







S-methyl 2-oxo-2-(5-phenyl-2-(thiophen-2-yl)-3,4-dihydro-2H-pyrrol-3-yl)ethanethioate



3j: Prepared according to the general procedure discussed above: reaction time, 12 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (5%); light yellow liquid (54 mg, 70%). ¹H NMR (600 MHz, CDCl₃): $\delta = 7.92$ (d, J = 6.6 Hz, 2 H), 7.50 – 7.41 (m, 3 H), 7.32 (dd, J = 4.8, 3.0 Hz, 1 H), 7.18 (d, J = 3.0 Hz, 1 H), 7.10 (d, J = 4.8 Hz, 1 H), 5.68 (d, J = 4.8 Hz, 1 H), 4.04 – 4.01 (m, 1 H), 3.45 – 3.36 (m, 2 H), 2.40 (s, 3 H) ppm; ¹³C{¹H} NMR (151 MHz, CDCl₃): $\delta = 193.4$, 191.3, 170.8,

142.4, 130.7, 128.2 (2 CH), 127.6 (2 CH), 125.9, 125.7, 120.8, 73.5, 50.7, 37.2, 11.0 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₁₇H₁₆NO₂S₂ [*M* + H]⁺: 330.0622; found: 330.0617.

^1H (600 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (151 MHz, CDCl_3) NMR spectra of 3j



S36


NOESY, COSY, HSQC, and HMBC (400 MHz, CDCl₃) NMR spectra of **3j**:







Methyl 2-(2,5-diphenyl-3,4-dihydro-2H-pyrrol-3-yl)-2-oxoacetate 3I: Prepared according to



the general procedure discussed above: reaction time, 14 h; $R_{\rm f}$ = 0.2; eluent, EtOAc/*n*-hexane (10%); yellow liquid (62 mg, 77%). ¹H NMR (400 MHz, CDCl₃): δ = 7.96 – 7.92 (m, 2 H), 7.51 – 7.41 (m, 3 H), 7.38 – 7.27 (m, 5 H), 5.62 (dt, *J* = 6.0, 2.0 Hz, 1 H), 3.97 (ddd, *J* = 9.2, 7.2, 6.0 Hz, 1 H), 3.87 (s, 3 H), 3.47 – 3.43 (m, 2 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 193.1, 171.5, 161.5, 142.3, 133.5, 131.3, 128.8 (2 CH), 128.7 (2 CH), 128.1 (2

CH), 127.8, 126.7 (2 CH), 77.9, 54.8, 53.3, 38.1. ppm; HRMS (ESI-QTOF): m/z calcd for $C_{19}H_{18}NO_3 [M + H]^+$: 308.1286; found: 308.1275.

^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 3I



ESI-17: Analytical and spectral data of 3m



Methyl 2-(2-(3,5-dichlorophenyl)-5-phenyl-3,4-dihydro-2*H*-pyrrol-3-yl)-2-oxoacetate 3m:



Prepared according to the general procedure discussed above: reaction time, 12 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (10%); light yellow liquid (61 mg, 85%). ¹H NMR (400 MHz, CDCl₃): $\delta = 7.93 - 7.91$ (m, 2 H), 7.52 - 7.44 (m, 3 H), 7.28 (t, J = 2.0 Hz, 1 H), 7.19 (d, J = 1.6 Hz, 2 H), 5.58 (dt, J = 6.4, 2.0 Hz, 1 H), 3.91 (s, 3 H), 3.90 - 3.81 (m, 1 H), 3.51 (ddd, J = 17.2, 9.6, 2.0 Hz, 1 H), 3.39 (ddd, J = 17.2, 6.8, 1.6 Hz, 1 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta = 192.4$, 172.4, 161.2,

145.8, 135.4, 133.0, 131.6, 128.8 (2 CH), 128.7, 128.2 (2 CH), 127.9, 125.4 (2 CH), 76.3, 54.6, 53.5, 38.5 ppm; HRMS (ESI-QTOF): m/z calcd for $C_{19}H_{16}Cl_2NO_3 [M + H]^+$: 376.0507; found: 376.0491.

^1H (400 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl3) NMR spectra of 3m





ESI-18: Analytical and spectral data of 3n



Hexyl 2-(2,5-diphenyl-3,4-dihydro-2H-pyrrol-3-yl)-2-oxoacetate 3n: Prepared according to



the general procedure discussed above: reaction time, 14 h; $R_{\rm f}$ = 0.2; eluent, EtOAc/*n*-hexane (10%); yellow liquid (52 mg, 71%). ¹H NMR (400 MHz, CDCl₃): δ = 7.94 – 7.92 (m, 2 H), 7.49 – 7.41 (m, 3 H), 7.36 – 7.32 (m, 2 H), 7.30 – 7.25 (m, 3 H), 5.59 (dt, *J* = 6.0, 2.0 Hz, 1 H), 4.29 – 4.12 (m, 2 H), 3.95 (td, *J* = 8.0, 5.6 Hz, 1 H), 3.44 (dd, *J* = 8.0, 1.6 Hz, 2 H), 1.81 –

1.57 (m, 2 H), 1.37 – 1.19 (m, 6 H), 1.01 – 0.76 (m, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = ¹³C NMR (101 MHz, CDCl₃): δ = 193.5, 171.6, 161.3, 142.3, 133.4, 131.3, 128.8 (2 CH), 128.7 (2 CH), 128.1 (2 CH), 127.8, 126.8 (2 CH), 77.9, 67.0, 54.8, 38.1, 31.4, 28.3, 25.5, 22.6, 14.1 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₂₄H₂₈NO₃ [*M* + H]⁺: 378.2069; found: 378.2071.

^1H (400 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl₃) NMR spectra of 3n

SM-3-1398 single_pulse C₈H₁₃00C



ESI-19: Analytical and spectral data of 3q



(2,5-Diphenyl-3,4-dihydro-2H-pyrrol-3-yl) (phenyl)methanone 3q: Prepared according to



the general procedure discussed above: reaction time, 16 h; $R_{\rm f}$ = 0.3; eluent, EtOAc/*n*-hexane (5%); Colourless liquid (41 mg, 52%). ¹H NMR (400 MHz, CDCl₃): δ = 7.96 (dt, *J* = 6.8, 1.6 Hz, 2 H), 7.82 (dt, *J* = 6.8, 1.6 Hz, 2 H), 7.60 – 7.51 (m, 1 H), 7.51 – 7.37 (m, 5 H), 7.37 – 7.25 (m, 3 H), 7.24 – 7.20 (m, 2 H), 5.62 (dt, *J* = 6.0, 2.0 Hz, 1 H), 4.18 (ddd, *J* = 13.2, 6.8, 2.4 Hz, 1 H), 3.66 – 3.45 (m, 2 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 199.6, 142.9, 136.3, 133.6 (2 CH), 131.2, 128.9 (2 CH), 128.8 (2 CH), 128.8 (2

CH), 128.7 (2 CH), 128.2 (2 CH), 127.7 (2 CH), 127.0 (2 CH), 79.6, 53.7, 40.1 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₂₃H₂₀NO [*M* + H]⁺: 326.1545; found: 326.1543.

^{1}H (400 MHz, CDCl_3) and $^{13}\text{C}\{^{1}\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 3q $_{\text{single_pulse}}^{\text{SM-2-117A}}$





ESI-20: Analytical and spectral data of 3r



(2-(4-Methoxyphenyl)-5-phenyl-3,4-dihydro-2H-pyrrol-3-yl) (phenyl) methanone 3r:



Prepared according to the general procedure discussed above: reaction time, 18 h; $R_f = 0.3$; eluent, EtOAc/*n*hexane (15%); Colourless liquid (43 mg, 58%). ¹H NMR (600 MHz, CDCl₃): δ = 7.97 (d, *J* = 7.2 Hz, 2 H), 7.85 (d, *J* = 7.2 Hz, 2 H), 7.59 (t, *J* = 7.2 Hz, 1 H), 7.51 – 7.43 (m, 5 H), 7.17 (d, *J* = 9.0 Hz, 2 H), 6.88 (d, *J* = 9.0 Hz, 2 H), 5.56 (d, *J* = 6.0 Hz, 1 H), 4.18 (dt, *J* = 9.6, 6.6 Hz, 1 H), 3.82 (s, 3 H), 3.62 (ddd, *J* = 16.8, 6.6, 1.8 Hz, 1 H), 3.52 (ddd, *J* =

17.4, 10.2, 2.4 Hz, 1 H) ppm; ${}^{13}C{}^{1}H$ NMR (151 MHz, CDCl₃): δ = 199.3, 170.7, 158.6, 135.8, 134.7, 133.3, 133.0, 130.5, 128.4 (2 CH), 128.2 (2 CH), 128.1 (2 CH), 127.6 (2 CH), 127.6 (2 CH), 113.6 (2 CH), 78.9, 54.9, 53.4, 39.4 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₄H₂₂NO₂ [M + H]⁺: 356.1650; found: 356.1664.

^1H (600 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (151 MHz, CDCl₃) NMR spectra of 3r:



ESI-21: Analytical and spectral data of 3s





(2-(4-Chlorophenyl)-5-phenyl-3,4-dihydro-2*H*-pyrrol-3-yl) (phenyl)methanone 3s: Prepared according to the general procedure discussed above: $R_f = 0.3$; eluent, EtOAc/*n*-hexane (10%); Colourless liquid (47 mg, 63%). ¹H NMR (400 MHz, CDCl₃): $\delta = 7.94 - 7.91$ (m, 2 H), 7.84 - 7.81 (m, 2 H), 7.60 - 7.55 (m, 1 H), 7.49 - 7.40 (m, 5 H), 7.28 (d, J = 8.4 Hz, 2 H), 7.16 (d, J = 8.4Hz, 2 H), 5.61 (dt, J = 6.4, 2.4 Hz, 1 H), 4.11 (td, J = 8.4, 6.4

Hz, 1 H), 3.52 (dd, J = 8.4, 2.0 Hz, 2 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta = 199.5$, 171.7, 141.7, 136.3, 133.7, 133.7, 133.3, 131.2, 128.9 (2 CH), 128.9 (4 CH), 128.7 (2 CH), 128.3 (2 CH), 128.1 (2 CH), 78.7, 53.9, 40.4 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₃H₁₉ClNO [M + H]⁺: 360.1155; found: 360.1150.

^1H (400 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl₃) NMR spectra of 3s:

JAY-3-131B single_pulse





JAY-3-131B single pulse decoupled gated NOE











MeSOC

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2-(2-(naphthalen-2-yl)-5-(*p*-tolyl)-3,4-dihydro-2*H*-pyrrol-3-yl)-2oxoethanethioate 3u: Prepared according to the general procedure discussed above: reaction time, 14 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (10%); yellow liquid (54 mg, 72%). ¹H NMR (400 MHz, CDCl₃): $\delta = 7.88 - 7.80$ (m, 5 H), 7.72 (s, 1 H), 7.47 - 7.40 (m, 3 H), 7.28 (d, J =7.6 Hz, 2 H), 5.79 (d, J = 5.6 Hz, 1 H), 4.05 (ddd, J = 10.0, 6.4, 5.6 Hz, 1 H), 3.50 (ddd, J = 17.2, 9.6, 2.0 Hz, 1 H), 3.38 (ddd, J = 17.2, 6.4, 1.2 Hz, 1 H), 2.43 (s, 3 H), 2.38 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta = 194.2$,

191.78, 171.7, 141.8, 139.7, 133.5, 133.0, 130.7, 129.5 (2 CH), 128.7, 128.2 (2 CH), 128.1, 127.8, 126.2, 126.0, 125.4, 124.8, 77.8, 52.2, 38.4, 21.6, 11.5 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₄H₂₂NO₂S [M + H]⁺: 388.1371; found: 388.1365.

^1H (400 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl3) NMR spectra of 3u



ESI-23: Analytical and spectral data of 3v





6.4, 1.2 Hz, 1 H), 2.41 (s, 3 H), 2.38 (s, 3 H) ppm; ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃): δ = 194.2, 191.8, 171.4, 142.4, 141.7, 130.8, 129.4 (2 CH), 128.8 (2 CH), 128.1 (2 CH), 127.7, 126.7 (2 CH), 77.7, 52.3, 38.3, 21.6, 11.5 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₀H₂₀NO₂S [M + H]⁺: 338.1214; found: 338.1211.

^1H (400 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl3) NMR spectra of 3v



ESI-24: Analytical and spectral data of 3w



S-methyl 2-oxo-2-(2-(o-tolyl)-5-(p-tolyl)-3,4-dihydro-2*H*-pyrrol-3-yl) ethanethioate 3w: Prepared according to the general procedure discussed above: reaction time, 12 h; $R_f = 0.2$; eluent, EtOAc/*n*hexane (5%); light yellow liquid (55 mg, 69%). ¹H NMR (600 MHz, CDCl₃): $\delta = 7.82$ (d, J = 7.80 Hz, 2 H), 7.26 (d, J =8.4 Hz, 2 H), 7.21 – 7.16 (m, 3 H), 7.06 – 7.04 (m, 1 H), 5.92 (d, J = 6.0 Hz, 1 H), 4.01 – 3.98 (m, 1 H), 3.58 (ddd, J = 17.4, 10.8, 2.4 Hz, 1 H), 3.28 (ddd, J = 17.4, 6.6, 1.2 Hz, 1 H), 2.42

(s, 3 H), 2.38 (s, 3 H), 2.36 (s, 3 H) ppm; ${}^{13}C{}^{1}H$ NMR (151 MHz, CDCl₃): δ = 194.0, 191.2, 170.0, 141.1, 140.0, 134.7, 130.2, 130.2, 128.9 (2 CH), 127.5 (2 CH), 127.0, 126.0 (2 CH), 74.0, 50.5, 39.1, 21.1, 19.2, 11.0 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₁H₂₂NO₂S [M + H]⁺: 352.1371; found: 352.1362.

^1H (600 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (151 MHz, CDCl_3) NMR spectra of 3w





ESI-25: Analytical and spectral data of 3x



S-methyl



2-oxo-2-(5-(p-tolyl)-2-(4-(trifluoromethyl)phenyl)-3,4-dihydro-2H-pyrrol-3-yl) ethanethioate 3x: Prepared according to the general procedure discussed above: reaction time, 12 h; $R_{\rm f}$ = 0.2; eluent, EtOAc/n-hexane (5%); light yellow liquid; (53 mg, 72%). ¹H NMR (400 MHz, CDCl₃): δ = 7.82 (d, J = 8.4 Hz, 2 H), 7.60 (d, J = 8.4 Hz, 2 H), 7.41 (d, J = 8.4 Hz, 2 H), 7.26 (d, J = 8.0 Hz, 2 H), 5.65 (d, J = 5.6 Hz, 1 H), 3.92 (ddd, J = 9.6, 6.4, 5.6 Hz, 1 H), 3.47 (ddd, J =

17.2, 9.6, 2.0 Hz, 1 H), 3.36 (ddd, J = 17.2, 6.4, 1.2 Hz, 1 H), 2.42 (s, 3 H), 2.39 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 193.8, 191.7, 172.2, 146.4, 142.0, 130.5, 130.0, 129.7, 129.5 (2 CH), 128.1 (2 CH), 127.1 (2 CH), 126.9, 125.7 (q, J = 4.1 Hz, 1 C), 76.9, 52.1, 38.4, 21.6, 11.5 ppm; ¹⁹F NMR (376 MHz, CDCl₃): $\delta = -62.4$ ppm. HRMS (ESI-QTOF): *m/z* calcd for C₂₁H₁₉F₃NO₂S [*M* + H]⁺: 406.1088; found: 406.1080.

 ^1H (400 MHz, CDCl_3), $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3), and ^{19}F (376 MHz, CDCl_3) NMR spectra of 3x $_{\text{single_pulse}}^{\text{SM-2-50A-02}}$





Supporting Information

SM-02-50A-02 single pulse decoupled gated NOE





ESI-26: Analytical and spectral data of 3y



S-methyl



2-(2-(3,5-dibromophenyl)-5-(*p*-tolyl)-3,4-dihydro-2*H*-pyrrol-3-yl)-2-oxoethanethioate 3y: Prepared according to the general procedure discussed above: *R*_f = 0.2; eluent, EtOAc/*n*-hexane (5%); light yellow liquid; (58 mg, 85%).
¹H NMR (400 MHz, CDCl₃): δ = 7.80 - 7.78 (m, 2 H), 7.57 (t, *J* = 2.0 Hz, 1 H), 7.35 (d, *J* = 1.6 Hz, 2 H), 7.26 (s, 1 H), 7.24 (d, *J* = 2.8 Hz, 1 H), 5.52 (d, *J* = 6.0 Hz, 1 H), 3.91 - 3.84 (m, 1 H), 3.48 (ddd, *J* = 17.2, 10.0, 2.1 Hz, 1 H), 3.30 (ddd, *J* = 17.2, 6.4, 1.2 Hz, 1 H), 2.41 (s, 3 H), 2.40 (s, 3 H)

ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 193.6, 191.7, 172.4, 146.4, 142.1, 133.3, 130.3, 129.5 (2 CH), 128.7 (2 CH), 128.8 (3 CH), 123.3, 76.2, 52.3, 38.5, 21.6, 11.6 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₂₀H₁₈Br₂NO₂S [*M* + H]⁺: 493.9425; found: 493.9427.

^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 3y

SM-02-46B single_pulse





ESI-27: Analytical and spectral data of 3z





S-methyl

B

MeSOC



Hz, 1 H), 3.49 (ddd, J = 17.6, 10.8, 2.4 Hz, 1 H), 3.28 (ddd, J = 17.6, 6.8, 1.6 Hz, 1 H), 2.41 (s, 3 H), 2.37 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta = 194.3$, 191.6, 172.0, 142.0, 139.6, 133.6, 132.2, 130.5, 130.4, 129.5 (2 CH), 129.4, 128.1 (2 CH), 121.7, 75.8, 49.8, 40.4, 21.6, 11.6 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₀H₁₈BrClNO₂S [M + H]⁺: 449.9930; found: 449.9945

^1H (400 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl3) NMR spectra of 3z



S63

ESI-28: Analytical and spectral data of 3aa



S-methyl 2-oxo-2-(2-phenyl-5-(*m*-tolyl)-3,4-dihydro-2*H*-pyrrol-3-yl)ethanethioate 3aa:



Prepared according to the general procedure discussed above: reaction time, 15 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (5%); light yellow liquid (56.2 mg, 69%). ¹H NMR (400 MHz, CDCl₃): $\delta = 7.80$ (s, 1 H), 7.69 (d, J = 7.2 Hz, 1 H), 7.37 – 7.31 (m, 4 H), 7.30 – 7.27 (m, 3 H), 5.62 (dt, J = 5.6, 2.0 Hz, 1 H), 3.98 (ddd, J = 9.6, 6.4, 5.2 Hz, 1 H), 3.47 (ddd, J = 17.2, 9.6, 2.0 Hz, 1 H), 3.35 (ddd, J = 17.2, 6.4, 1.6 Hz, 1 H), 2.41 (s, 3

H), 2.39 (s, 3 H) ppm; ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃): δ = 194.1, 191.8, 171.7, 142.3, 138.5, 133.3, 132.1, 128.8 (2 CH), 128.7, 128.6, 127.7, 126.7 (2 CH), 125.4, 77.7, 52.2, 38.3, 21.4, 11.5 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₀H₂₀NO₂S [M + H]⁺: 338.1214; found: 338.1203.

^1H (400 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl₃) NMR spectra of **3aa**:



ESI-29: Analytical and spectral data of 3ab



S-methyl 2-(5-(naphthalen-2-yl)-2-phenyl-3,4-dihydro-2H-pyrrol-3-yl)-2-oxoethanethioate



3ab: Prepared according to the general procedure discussed above: reaction time, 14 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (10%); light yellow liquid (66 mg, 73%). ¹H NMR (400 MHz, CDCl₃): $\delta = 8.27$ (s, 1 H), 8.18 (dd, J = 8.8, 1.6 Hz, 1 H), 7.93 – 7.87 (m, 3 H), 7.59 – 7.51 (m, 2 H), 7.39 – 7.29 (m, 5 H), 5.69 (dt, J = 5.6, 1.6 Hz, 1 H), 4.04

(ddd, J = 9.6, 6.4, 5.2 Hz, 1 H), 3.62 (ddd, J = 17.2, 9.6, 2.0 Hz, 1 H), 3.49 (ddd, J = 17.2, 6.4, 1.6 Hz, 1 H), 2.40 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta = 194.1$, 191.8, 171.4, 142.3, 134.8, 133.0, 131.0, 128.9 (2 CH), 128.8 (2 CH), 128.5, 127.9, 127.7, 127.6, 126.8 (2 CH), 126.7, 124.7, 77.9, 52.4, 38.4, 11.5 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₃H₂₀NO₂S [M + H] ⁺: 374.1214; found: 374.1201.

1H (400 MHz, CDCl₃) and $^{13}C\{^1H\}$ (101 MHz, CDCl₃) NMR spectra of **3ab**



ESI-30: Analytical and spectral data of 3ac





1 H), 2.41 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 193.4, 191.6, 171.5, 145.5, 135.4, 132.1 (2 CH), 131.9, 129.7 (3 CH), 128.0, 126.32 (2 CH), 125.3, 76.4, 52.1, 38.4, 11.6 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₁₉H₁₅BrCl₂NO₂S [*M* + H] ⁺: 469.9384; found: 469.9379.

^1H (400 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl₃) NMR spectra of **3ac:**



CI

CI

ESI-31: Analytical and spectral data of 4a



2-(2-(3,5-dichlorophenyl)-5-phenyl-1H-pyrrol-3-yl)-2-oxoethanethioate S-methyl 4a: Prepared according to the general procedure discussed above: reaction time, 30 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (5%); red solid (61 mg, 86%); mp 130–133 °C. ¹H NMR (400 MeSOC MHz, CDCl₃): δ = 9.03 (s, 1 H), 7.53 (dd, J = 8.4, 1.2 Hz, 1 H), 7.45 (d, J = 2.4 Hz, 1 H), 7.41 (t, J = 8.4 Hz, 2 H), 7.36 (d, J = NH 2.8 Hz, 1 H), 7.37 – 7.26 (m, 2 H), 2.33 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 194.0, 180.3, 136.0, 133.6, 132.6,

132.3, 132.2, 132.0, 131.0, 130.8, 130.5, 129.3 (2 CH), 127.9,

124.5 (2 CH), 118.2, 109.3, 11.6 ppm; HRMS (ESI-QTOF): m/z calcd for C19H14Cl2NO2S [M + H]⁺: 390.0122; found: 390.0109.

^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 4a







S-methyl 2-(2-(4-chlorophenyl)-5-phenyl-1H-pyrrol-3-yl)-2-oxoethanethioate 4b: Prepared



according to the general procedure discussed above: reaction time, 36 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (10 %); red solid (56 mg, 76%), mp 140–145 °C. ¹H NMR (600 MHz, CDCl₃): δ = 8.88 (s, 1 H), 7.61 (d, *J* = 8.4 Hz, 2 H), 7.56 (d, *J* = 8.4 Hz, 2 H), 7.45 (dt, *J* = 7.8, 3.6 Hz, 4 H), 7.37 (d, *J* = 3.0 Hz, 1 H), 7.34 (t, *J* = 7.2 Hz, 1 H), 2.40 (s, 3 H) ppm;

¹³C{¹H} NMR (151 MHz, CDCl₃): δ = 194.1, 180.2, 139.9, 134.9, 132.7, 130.3, 129.7 (2 CH), 129.1, 128.7 (2 CH), 128.3 (2 CH), 127.3, 123.8 (2 CH), 115.7, 109.8, 11.1 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₁₉H₁₄ClNO₂SNa [*M* + Na]⁺: 378.0332; found: 378.0343.
^1H (600 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (151 MHz, CDCl3) NMR spectra of 4b







S-methyl 2-(2-(4-nitrophenyl)-5-phenyl-1H-pyrrol-3-yl)-2-oxoethanethioate 4c: Prepared



according to the general procedure discussed above: reaction time, 14 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (15%); red solid (48 mg, 67%), mp 163-168 °C; ¹H NMR (400 MHz, CDCl₃): $\delta = 9.09$ (s, 1 H), 8.23 (d, J = 8.8 Hz, 2 H), 7.78 (d, J = 8.8 Hz, 2 H), 7.58 – 7.53 (m, 2 H), 7.45 – 7.40 (m, 2 H), 7.36 (d, J = 2.4 Hz, 1 H), 7.35 – 7.31 (m, 1 H), 2.38 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta =$

194.2, 180.9, 147.7, 138.2, 137.4, 134.5, 130.4, 129.7 (2 CH), 129.3 (2 CH), 128.2, 124.6 (2 CH), 123.8 (2 CH), 117.4, 111.1, 11.7 ppm; HRMS (ESI-QTOF): m/z calcd for C₁₉H₁₄N₂O₄SK [M + K]⁺: 405.0311; found: 405.0317.

^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 4c



ESI-34: Analytical and spectral data of 4d



S-methyl



2-(1-(2-(3-bromophenyl)-5-phenyl-1*H*-pyrrol-3-yl)ethan-1-one)-2 oxoethanethioate 4d: Prepared according to the general procedure discussed above: reaction time, 36 h; *R*_f = 0.3; eluent, EtOAc/*n*-hexane (10 %); red solid (53 mg, 76%), mp 72-75 °C; ¹H NMR (400 MHz, CDCl₃): δ = 8.92 (s, 1H), 7.73 (t, *J* = 1.6 Hz, 1H), 7.60 - 7.56 (m, 1H), 7.55 - 7.50 (m, 3H), 7.43 - 7.39 (m, 2 H), 7.34 - 7.26 (m, 3 H), 2.36 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 194.5, 181.0, 139.7, 133.5, 133.2, 132.2,

131.6, 130.8, 130.0, 129.3 (2 CH), 127.87 (2 CH), 124.4 (2 CH), 122.5, 116.6, 110.3, 11.6 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₁₉H₁₅BrNO₂S [*M* + H]⁺: 400.0007; found: 399.9988.

^1H (400 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl₃) NMR spectra of 4d









S-methyl 2-oxo-2-(5-phenyl-2-(*p*-tolyl)-1*H*-pyrrol-3-yl)ethanethioate 4e: Prepared according to the general procedure discussed above: reaction time, 38 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (10%); red solid (55 mg, 72%), mp 122–127 °C. ¹H NMR (400 MHz, Acetone- d_6): $\delta = 11.27$ (s, 1 H), 7.85 – 7.74 (m, 2 H), 7.65 – 7.55 (m, 2 H), 7.50 – 7.37 (m, 2 H), 7.33 – 7.20 (m, 4 H), 2.39 (s, 3 H), 2.33 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, Acetone- d_6): $\delta = 194.5$, 181.2, 142.3, 138.8, 133.2,

131.5, 129.4 (2 CH), 128.9 (2 CH), 128.7, 128.6 (2 CH), 127.2, 124.6 (2 CH), 115.5, 109.6, 20.5, 10.4 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₀H₁₇NO₂SNa [M + Na]⁺: 358.0878; found: 358.0883.

¹H (400 MHz, Acetone- d_6) and ¹³C{¹H} (101 MHz, Acetone- d_6) NMR spectra of **4e**

JAY-462A single_pulse











above: reaction time, 42 h; $R_{\rm f}$ = 0.2; eluent, EtOAc/*n*-hexane (10%); red solid (60 mg, 81%); mp 92–94 °C. ¹H NMR (400 MHz, CDCl₃): δ = 8.98 (s, 1 H), 7.53 – 7.50 (m, 2 H), 7.41 – 7.37 (m, 2 H), 7.33 – 7.26 (m, 3 H), 7.18 – 7.14 (m, 2 H), 6.93 (ddd, *J* = 8.4, 2.4, 1.2 Hz, 1 H), 3.82 (s, 3 H), 2.33 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, Acetone- d_6): δ = 194.3, 181.5, 159.4, 141.8, 133.4, 132.7, 131.5, 129.1, 128.9 (2 CH), 127.2, 124.6 (2 CH), 121.8,

115.9, 114.9, 114.6, 109.6, 54.8, 10.5 ppm; HRMS (ESI-QTOF): *m*/*z* calcd for C₂₀H₁₇NO₃SNa [*M* + Na]⁺: 374.0827; found: 374.0817.

^1H (400 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, Acetone- $d_6)$ NMR spectra of 4f







S-methyl 2-oxo-2-(5-phenyl-2-(o-tolyl)-1*H*-pyrrol-3-yl)ethanethioate 4g: Prepared according to the general procedure discussed above: reaction time, 36 h; $R_f = 0.3$; eluent, EtOAc/n-hexane (10%); red solid (56 mg, 74%); mp 122-127 °C. ¹H NMR (400 MHz, CDCl₃): $\delta =$ 8.74 (s, 1 H), 7.52 – 7.49 (m, 2 H), 7.43 – 7.36 (m, 3 H), 7.34 (dd, J = 7.2, 1.6 Hz, 1 H), 7.31 – 7.26 (m, 3 H), 7.22 (d, J = 8.0 Hz, 1 H), 2.28 (s, 3 H), 2.24 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta =$ 194.5, 180.5, 141.5, 138.3, 132.6, 131.5, 131.1,

130.3, 130.1, 129.5, 129.2 (2 CH), 127.6, 125.7, 124.2 (2 CH), 117.6, 108.8, 19.9, 11.5 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₂₀H₁₈NO₂S [*M* + H]⁺: 336.1058; found: 336.1055.

^1H (400 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl3) NMR spectra of 4g

JAY-3-131C single_pulse







S-methyl 2-(2,5-diphenyl-1*H*-pyrrol-3-yl)-2-oxoethanethioate 4h: Prepared according to the general procedure discussed above: reaction time, 38 h; $R_{\rm f}$ = 0.2; eluent, EtOAc/*n*-hexane (5%); red solid (57 mg, 73%); mp 80-85 °C. ¹H NMR (400 MHz, CDCl₃): δ = 8.93 (s, 1 H), 7.63 – 7.60 (m, 2 H), 7.54 – 7.51 (m, 2 H), 7.45 – 7.38 (m, 5 H), 7.31 – 7.27 (m, 2 H), 2.33 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, Acetone- d_6): δ = 194.4, 181.3, 142.1, 133.5, 131.5, 131.5, 129.5

(2 CH), 128.9 (2 CH), 128.8, 128.0 (2 CH), 127.2, 124.6 (2 CH), 115.7, 109.6, 10.4 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₁₉H₁₆NO₂S [*M* + H]⁺: 322.0901; found: 322.0901.

¹H (400 MHz, CDCl₃) and ¹³C{¹H} (101 MHz, Acetone- d_6) NMR spectra of **4h**



ESI-39: Analytical and spectral data of 4i



S-methyl 2-(2-(naphthalen-2-yl)-5-phenyl-1*H*-pyrrol-3-yl)-2-oxoethanethioate 4i: Prepared according to the general procedure discussed above: reaction time, 36 h; $R_f = 0.3$; eluent, EtOAc/*n*-hexane (10%); red solid (64 mg, 73%); mp 175–180 °C; solvent of crystallization, MeCN/DCM (2 mL:1 mL, v/v) at room temperature. ¹H NMR (400 MHz, CDCl₃): δ = 9.00 (s, 1 H), 8.05 (d, *J* = 2.0 Hz, 1 H), 7.87 – 7.82 (m, 3 H), 7.69 (dd, *J* = 8.4, 2.0 Hz, 1 H), 7.57 – 7.53 (m, 2 H), 7.52 – 7.48 (m, 2 H), 7.44 – 7.39 (m, 2 H), 7.32 (d, *J* = 2.8 Hz, 1 H), 7.32 – 7.28 (m,

1 H), 2.30 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 194.7, 181.0, 141.9, 133.5, 133.2, 133.0, 131.0, 129.2 (2 CH), 128.8, 128.4, 128.2, 128.1, 127.9, 127.7, 127.0, 126.7, 126.5, 124.4 (2 CH), 116.4, 110.3, 11.6 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₂₃H₁₇NO₂SNa [*M* + Na]⁺: 394.0878; found: 394.0874.



X-ray determined molecular structure of 4i; CCDC 2347215.

Datablock: 4i

Bond precision:		C-C = 0.0041 A			Wavelength=1.54178			
Cell:	a=7.4754	4(5)	b=9.5803(7)	c=	25.2169(18)			
	alpha=90)	beta=90	ga	mma=90			
Temperature:	130 K							
		Calculated	ł		Reported			
Volume		1806.0(2)			1805.9(2)			
Space group		P 21 21 21			P 21 21 21			
Hall group		P 2ac 2ab			P 2ac 2ab			
Moiety formula		C23 H17 N O2 S			C23 H17 N O2 S			
Sum formula		C23 H17 N O2 S			C23 H17 N O2 S			
Mr		371.44			371.43			
Dx,g cm-3		1.366			1.366			
Z		4			4			
Mu (mm-1)		1.734			1.734			
F000		776.0			776.0			
F000'		779.42						
h,k,lmax		9,11,31			9,11,31			
Nref	Vref 3		3695[2141]		3606			
Tmin <i>,</i> Tmax	,Tmax 0.768,0.969		59		0.435,0.754			
Tmin'		0.673						
Correction me AbsCorr = MU	thod= # I LTI-SCAN	Reported ⁻	T Limits: Tmin=	0.435	Tmax=0.754			
Data complete	eness= 1.	68/0.98	Theta(ma	ax)= 74	.468			
R(reflections)= 0.0404(3548)					wR2(reflections)= 0.1056(3606)			
S = 1.090		Npar=	245					
The following ALERTS were generated.								

Alert level C

<u>PLAT987 ALERT 1 C</u> The Flack x is >> 0 - Do a BASF/TWIN Refinement Please Check

Alert level G	
PLAT007 ALERT 5 G Number of Unrefined Donor-H Atoms	1 Report
H1	
PLAT033 ALERT 4 G Flack x Value Deviates > 3.0 * sigma from Zero . 0.	.052 Note
PLAT883 ALERT 1 G No Info/Value for _atom_sites_solution_primary .	Please Do !
PLAT910 ALERT 3 G Missing # of FCF Reflection(s) Below Theta(Min).	2 Note
0 1 1, 0 0 2,	
PLAT912 ALERT 4 G Missing # of FCF Reflections Above STh/L= 0.600	13 Note
PLAT961_ALERT_5_G Dataset Contains no Negative Intensities Plea	ise Check
PLAT969_ALERT_5_G The 'Henn et al.' R-Factor-gap value 3.16	Note
Predicted wR2: Based on SigI**2 3.35 or SHELX Weight 10.04	
PLAT978 ALERT 2 G Number C-C Bonds with Positive Residual Density.	2 Info

^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 4i







S-methyl 2-oxo-2-(5-phenyl-2-(thiophen-2-yl)-1*H*-pyrrol-3-yl)ethanethioate 4j: Prepared according to the general procedure discussed above: reaction time, 36 h; $R_f = 0.3$; eluent, EtOAc/*n*-hexane (10%); red gummy liquid (52 mg, 68%). ¹H NMR (400 MHz, CDCl₃): δ = 8.93 (s, 1 H), 8.01 (dd, J = 2.8, 1.2 Hz, 1 H), 7.52 (dd, J = 8.4, 1.2 Hz, 2 H), 7.44 (dd, J = 5.2, 1.6 Hz, 1 H), 7.43 – 7.37 (m, 3 H), 7.31 (d, J = 2.8 Hz, 1 H), 7.31 – 7.27 (m, 1 H), 2.38 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 194.9, 180.6, 136.9, 132.5, 131.4, 130.9, 129.2

(2 CH), 127.7, 127.4, 126.3, 126.0, 124.4 (2 CH), 116.0, 110.3, 11.7 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₁₇H₁₄NO₂S₂ [*M* + H]⁺: 328.0466; found: 328.0453.

^1H (600 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (151 MHz, CDCl_3) NMR spectra of 4j





ESI-41: Analytical and spectral data of 4I



Methyl 2-(2,5-diphenyl-1H-pyrrol-3-yl)-2-oxoacetate 4I: Prepared according to the general



procedure discussed above: reaction time, 38 h; $R_{\rm f}$ = 0.2; eluent, EtOAc/*n*-hexane (10%); yellow solid (58 mg, 72%); mp 112-116 °C. ¹H NMR (400 MHz, CDCl₃): δ = 9.01 (s, 1 H), 7.55 – 7.51 (m, 4 H), 7.44 – 7.38 (m, 5 H), 7.31 – 7.25 (m, 1 H), 7.07 (d, *J* = 2.8 Hz, 1 H), 3.56 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 181.5, 164.9, 141.1, 133.2, 131.1, 130.9, 129.4, 129.2 (2 CH), 129.0 (2 CH), 128.7 (2 CH), 127.7, 124.4 (2 CH),

118.7, 109.0, 52.3 ppm; HRMS (ESI-QTOF): m/z calcd for C₁₉H₁₅NO₃Na [M + Na]⁺: 328.0950; found: 328.0943.

^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of **4I:**







Methyl 2-(2-(3,5-dichlorophenyl)-5-phenyl-1*H*-pyrrol-3-yl)-2-oxoacetate 4m: Prepared according to the general procedure discussed above: reaction time, 34 h; $R_f = 0.3$; eluent, EtOAc/*n*-hexane (10%); yellow solid; (53 mg, 74%); mp 133-138 °C. ¹H NMR (400 MHz, Acetone- d_6): $\delta = 11.47$ (s, 1 H), 7.79 – 7.77 (m, 2 H), 7.71 (d, J = 1.9 Hz, 2 H), 7.52 (t, J = 1.9 Hz, 1 H), 7.43 – 7.39 (m, 2 H), 7.29 (ddt, J = 8.2, 6.8, 1.2 Hz, 1 H), 7.11 (d, J = 2.8 Hz, 1 H), 3.76 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, Acetone-

 d_6): δ = 181.5, 165.1, 136.7, 134.4, 134.3, 134.3, 134.3, 131.0, 129.0 (2 CH), 128.2, 128.0 (2 CH), 127.6, 124.8 (2 CH), 118.7, 109.6, 51.8 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₁₉H₁₄Cl₂NO₃ [*M* + H]⁺: 374.0350; found: 374.0366.

¹H (400 MHz, Acetone- d_6) and ¹³C{¹H} (101 MHz, Acetone- d_6) NMR spectra of **4m**

SM-3-138A single_pulse MeOOC



ESI-43: Analytical and spectral data of 4n



Hexyl 2-(2,5-diphenyl-1*H*-pyrrol-3-yl)-2-oxoacetate 4n: Prepared according to the general procedure discussed above: reaction time, 38 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (10%); light yellow solid (50 mg, 70%), mp 54-56 °C. ¹H NMR (400 MHz, Acetone- d_6): $\delta = 11.27$ (s, 1 H), 7.82 – 7.77 (m, 2 H), 7.63 – 7.60 (m, 2H), 7.44 – 7.38 (m, 5 H), 7.29 – 7.25 (m, 1 H), 7.06 – 7.05 (m, 1 H), 3.93 (t, J = 6.8 Hz, 2 H), 1.56 – 1.05 (m, 2 H), 1.29 – 1.26 (m, 6 H), 0.87 – 0.83 (m, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, Acetone- d_6): $\delta = 182.1$, 165.2, 140.8, 133.5,

131.4, 131.3, 129.6 (3 CH), 128.9 (2 CH), 128.2 (2 CH), 127.3, 124.6 (2 CH), 118.3, 108.5, 65.2, 31.3, 28.2, 25.4, 22.4, 13.4 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₄H₂₆NO₃ [M + H]⁺: 376.1912; found: 376.1904.

¹H (400 MHz, Acetone- d_6) and ¹³C{¹H} (101 MHz, Acetone- d_6) NMR spectra of **4n**

SM=3-143B single_pulse





ESI-44: Analytical and spectral data of 2o and 4o



(*E*)-4-(4-chlorophenyl)-1-phenylbut-3-ene-1,2-dione; 2o: Prepared according to the literature⁶: $R_{\rm f}$ = 0.2; eluent, EtOAc/*n*-hexane (5%); Yellow solid (486 mg, 58%) mp 78–81 °C; ¹H NMR (400 MHz, CDCl₃): δ = 8.07 – 7.99 (m, 2 H), 7.71 – 7.60 (m, 2 H), 7.57 – 7.45 (m, 4 H), 7.42 – 7.34 (m, 2 H), 7.11 (d, *J* = 16.4 Hz, 1 H) ppm; ¹³C{¹H}

NMR (101 MHz, CDCl₃ δ = 193.0, 192.3, 147.2, 137.7, 134.8, 132.8, 132.6, 130.3 (2 CH), 130.1 (2 CH), 129.5 (2 CH), 129.0 (2 CH), 122.8 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₁₆H₁₂ClO₂ [*M* + H]⁺: 271.0526; found: 271.0518.

1-(2-(4-chlorophenyl)-5-phenyl-1H-pyrrol-3-yl)-2-phenylethane-1,2-dione 4o: Prepared



according to the general procedure discussed above: reaction time, 12 h; $R_{\rm f}$ = 0.2; eluent, EtOAc/*n*-hexane (10%); red solid (51 mg, 72%); mp 194–198 °C. ¹H NMR (400 MHz, CDCl₃): δ = 8.92 (s, 1 H), 7.90 (dd, *J* = 8.0, 1.2 Hz, 2 H), 7.62 – 7.54 (m, 3 H), 7.49 – 7.42 (m, 4 H), 7.40 – 7.33 (m, 3 H), 7.32 – 7.25 (m, 2 H), 6.88 (d, *J* = 2.8 Hz, 1 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 194.4, 189.9, 138.6, 135.5,

134.5, 133.4, 133.2, 130.7, 130.5 (2 CH), 130.0 (2 CH), 129.3, 129.2 (2 CH), 128.9 (2 CH), 128.8 (2 CH), 127.9, 124.4 (2 CH), 119.1, 109.7 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₂₄H₁₇CINO₂ [*M* + H]⁺: 386.0948; found: 386.0944.

 ^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 2o $_{\text{single_pulse}}^{\text{SM-3-123C}}$





^1H (400 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl₃) NMR spectra of 4ab

SM-3-125A single_pulse





ESI-45: Analytical and spectral data of 2p and 4p:



Cinnamoyl cyanide 2p: Prepared according to the literature procedure⁷ discussed above: $R_{\rm f}$ = 0.2; eluent, EtOAc/*n*-hexane (15%); white solid (0.65 g, 80%); mp 110-114 °C. ¹H NMR (400 MHz, CDCl₃): δ = 8.00 (d, *J* = 16.4 Hz, 1 H), 7.66 - 7.59 (m, 2 H), 7.56 - 7.43 (m, 3 H), 6.85 (d, *J* = 16.4 Hz, 1 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 167.6, 155.1, 133.2, 132.9, 129.6 (4 CH), 125.4, 112.5 ppm; HRMS (EI): *m/z* calcd for C₁₀H₇NO

2,5-diphenyl-1*H*-**pyrrole-3-carbonyl cyanide 4p:** Prepared according to the general procedure discussed above: reaction time, 16 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (10%); yellow solid (44 mg, 52%); mp 145-148 °C. ¹H NMR (400 MHz, CDCl₃): $\delta = 9.00$ (s, 1 H), 7.71 – 7.61 (m, 2 H), 7.60 – 7.50 (m, 2 H), 7.53 – 7.40 (m, 5 H), 7.40 – 7.30 (m, 1 H), 7.16 (d, J = 3.2 Hz, 1 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta = 178.7$, 161.3, 133.9, 130.3, 130.2, 129.6, 129.4 (2 CH), 129.0,

128.9 (2 CH), 128.4 (2 CH), 124.5 (2 CH), 120.3, 114.3, 110.1 ppm; HRMS (ESI-QTOF): m/z calcd for C₁₈H₁₃N₂O [M + H]⁺: 273.1028; found: 273.1017.

[[]*M*]⁺: 157.0528; found: 157.0520.

 ^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 2p $_{\text{single_pulse}}^{\text{SM-3-146B}}$





 ^{1}H (400 MHz, CDCl_3) and $^{13}\text{C}\{^{1}\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 4p $_{\text{single_pulse}}^{\text{SM-3-151C}}$





ESI-46: Analytical and spectral data of 4u





128.2, 128.2, 128.1, 127.9, 127.0, 126.7, 126.5, 124.3 (2 CH), 116.4, 109.7, 21.3, 11.6 ppm; HRMS (ESI-QTOF):*m/z*calcd for C₂₄H₂₀NO₂S [*M*+ H]⁺: 386.1214; found: 386.1217.

^1H (400 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl3) NMR spectra of 4u



ESI-47: Analytical and spectral data of 4v



S-methyl 2-oxo-2-(2-phenyl-5-(*p*-tolyl)-1*H*-pyrrol-3-yl)ethanethioate 4v: Prepared according to the general procedure discussed above: reaction time, 38 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (10%); red solid (57 mg, 70%); mp 112-117°C; ¹H NMR (400 MHz, Acetone- d_6): $\delta = 11.25$ (s, 1 H), 7.68 – 7.63 (m, 4 H), 7.43 – 7.38 (m, 3 H), 7.23 – 7.20 (m, 3 H), 2.32 (s, 3 H), 2.30 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, Acetone- d_6): δ = 194.4, 181.2, 141.8, 137.0, 133.6, 131.6, 129.5 (3 CH),

128.7 (2 CH), 128.0 (2 CH), 124.5 (2 CH), 115.7, 109.0 (2 CH), 20.3, 10.4 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₀H₁₈NO₂S [M + H]⁺: 336.1058; found: 336.1047.

¹H (400 MHz, Acetone- d_6) and ¹³C{¹H} (101 MHz, Acetone- d_6) NMR spectra of 4v

SM-3-132D single_pulse



ESI-48: Analytical and spectral data of 4w



S-methyl 2-oxo-2-(2-(o-tolyl)-5-(p-tolyl)-1*H*-pyrrol-3-yl)ethanethioate 4w: Prepared according to the general procedure discussed above: reaction time, 36 h; $R_f = 0.2$; eluent, EtOAc/n-hexane (5%); red solid (56 mg, 71%), mp 62–66 °C. ¹H NMR (400 MHz, Acetone- d_6): $\delta = 11.22$ (s, 1 H), 7.66 (d, J = 8.4 Hz, 2 H), 7.33 – 7.18 (m, 7 H), 2.31 (s, 3 H), 2.21 (s, 3 H), 2.18 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, Acetone- d_6): $\delta = 194.2$, 181.0, 141.4, 138.1, 136.8, 133.3, 132.1, 130.6, 129.7,

129.6 (2 CH), 128.9, 128.9, 125.2, 124.3 (2 CH), 117.2, 107.4, 20.3, 19.1, 10.3 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₂₁H₂₀NO₂S [*M* + H]⁺: 350.1214; found: 350.1213.
¹H (400 MHz, Acetone- d_6) and ¹³C{¹H} (101 MHz, Acetone- d_6) NMR spectra of **4w**











4x: Prepared according to the general procedure discussed above: reaction time, 36 h; $R_f = 0.3$; eluent, EtOAc/*n*-hexane (5%); red solid (53 mg, 72%); mp 83-87 °C. ¹H NMR (400 MHz, CDCl₃): δ = 8.86 (s, 1 H), 7.75 (d, *J* = 8.0 Hz, 2 H), 7.69 (d, *J* = 8.0 Hz, 2 H), 7.43 (d, *J* = 8.0 Hz, 2 H), 7.31 (d, *J* = 3.2 Hz, 1 H), 7.23 (d, *J* = 7.6

Hz, 2 H), 2.37 (s, 3 H), 2.36 (s, 3 H) ppm; ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃): δ = 194.5, 180.8, 139.3, 138.0, 133.9, 131.2, 130.0 (2 CH), 129.3 (2 CH), 127.9, 125.5 (q, *J* = 4.0 Hz, 1 C), 124.4 (2 CH), 116.7, 110.1 (2 CH), 21.3, 11.6 ppm; ${}^{19}F$ NMR (376 MHz, CDCl₃): δ = -62.7 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₂₁H₁₆F₃NO₂SNa [*M* + Na]⁺: 426.0752; found: 426.0751.

 ^1H (400 MHz, CDCl_3), $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3), and ^{19}F (376 MHz, CDCl_3) NMR spectra of 4x $_{\text{sm-3-129D}}$



Supporting Information

SM-3-129D single pulse decoupled gated NOE





0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -190 -20 f1 (ppm)





S-methyl 2-(2-(3,5-dibromophenyl)-5-(p-tolyl)-1H-pyrrol-3-yl)-2-oxoethanethioate 4y:



Prepared according to the general procedure discussed above: reaction time, 34 h; $R_f = 0.3$; eluent, EtOAc/*n*-hexane (10%); red solid (55.5 mg, 82%); mp 145-150 °C. ¹H NMR (400 MHz, CDCl₃): δ = 8.84 (s, 1 H), 7.70 – 7.68 (m, 3 H), 7.42 (d, *J* = 8.4 Hz, 2 H), 7.25 (d, *J* = 2.4 Hz, 1 H), 7.22 (d, *J* = 7.6 Hz, 2 H), 2.37 (s, 6 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): δ = 194.3, 181.0, 138.1, 137.3, 134.6, 134.5, 134.1, 130.6 (2 CH), 130.0 (2 CH),

127.7, 124.4 (2 CH), 122.9, 117.0, 110.0 (2 CH), 21.3, 11.6 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₂₀H₁₅Br₂NO₂SNa [*M* + Na]⁺: 513.9088; found: 513.9075.

^{1}H (400 MHz, CDCl_3) and $^{13}\text{C}\{^{1}\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 4y $_{^{JAY-3+1288}}$ single-pulse



ESI-51: Analytical and spectral data of 4z



S-methyl 2-(2-(4-bromo-2-chlorophenyl)-5-(p-tolyl)-1H-pyrrol-3-yl)-2-oxoethanethioate



4z: Prepared according to the general procedure discussed above: reaction time, 35 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (10%); red solid (54.7 mg, 78%); mp 168–173 °C. ¹H NMR (600 MHz, Acetone- d_6): $\delta = 11.48$ (s, 1H), 7.77 (d, J = 2.4 Hz, 1 H), 7.69 (d, J = 8.4 Hz, 2 H), 7.64 – 7.63 (m, 1 H), 7.50 (d, J = 8.4 Hz, 1 H), 7.34 (d, J = 3.0 Hz, 1 H), 7.27 (d, J = 7.8 Hz, 2 H), 2.36 (s, 3 H), 2.33

(s, 3 H) ppm; ¹³C{¹H} NMR (151 MHz, Acetone- d_6): δ = 193.4, 179.7, 136.6, 136.0, 134.8, 133.4, 133.2, 131.2, 130.5, 129.4, 129.1 (2 CH), 128.0, 123.9 (2 CH), 122.2, 117.2, 107.4, 19.8, 9.9 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₀H₁₅BrClNO₂S [M + H]⁺: 447.9773; found: 447.9776.

¹H (600 MHz, Acetone- d_6) and ¹³C{¹H} (151 MHz, Acetone- d_6) NMR spectra of 4z

19w-JAY-133B.1.1.1r JAY-133B 1H-NMR in Acetone-d6





ESI-52: Analytical and spectral data of 4aa



S-methyl 2-oxo-2-(2-phenyl-5-(*m*-tolyl)-1*H*-pyrrol-3-yl)ethanethioate 4aa: Prepared according to the general procedure discussed above: reaction time, 38 h; $R_f = 0.3$; eluent, EtOAc/*n*-hexane (5%); red gum (58 mg, 71%). ¹H NMR (600 MHz, Acetone- d_6): $\delta =$ 11.34 (s, 1 H), 7.71 – 7.66 (m, 3 H), 7.62 (d, J = 7.8 Hz, 1 H), 7.47 – 7.43 (m, 3 H), 7.34 – 7.30 (m, 2 H), 7.14 (d, J = 7.8 Hz, 1 H), 2.39 (s, 3 H), 2.34 (s, 3 H) ppm; ¹³C{¹H} NMR (151 MHz, Acetone- d_6): $\delta =$ 193.9, 180.7, 141.4, 137.9, 133.0, 131.0,

130.8, 129.0 (2 CH), 128.3, 128.2, 127.4 (2 CH), 127.4, 124.7, 121.2, 115.1, 109.0, 20.1, 9.9 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₂₀H₁₈NO₂S [*M* + H]⁺: 336.1058; found: 336.1067.

¹H (600 MHz, Acetone- d_6) and ¹³C{¹H} (151 MHz, Acetone- d_6) NMR spectra of **4aa**:



ESI-53: Analytical and spectral data of 4ab





 $\begin{array}{c} \text{nexane (10\%); re} \\ & ^{1}\text{H NMR (400 MH} \\ & 1\text{H}), 7.93 (d, J = 3 \\ & 7.69 (m, 2\text{H}), 7. \end{array}$

Prepared according to the general procedure discussed above: reaction time, 38 h; $R_f = 0.3$; eluent, EtOAc/*n*hexane (10%); red solid (68.4 mg, 76%), mp 165–168 °C. ¹H NMR (400 MHz, Acetone- d_6): $\delta = 11.49$ (s, 1H), 8.31 (s, 1H), 7.93 (d, J = 1.6 Hz, 2H), 7.80 – 7.86 (m, 2H), 7.71 – 7.69 (m, 2H), 7.50 – 7.42 (m, 6H), 2.32 (s, 3H) ppm;

¹³C{¹H} NMR (101 MHz, Acetone- d_6): δ = 194.4, 181.2, 142.4, 133.9, 133.4, 132.7, 131.6, 129.6 (2 CH), 128.9 (2 CH), 128.7, 128.0 (3 CH), 127.8, 126.7, 126.0, 123.3, 122.5, 115.9, 110.4, 10.5 ppm; HRMS (ESI-QTOF): m/z calcd for C₂₃H₁₈NO₂S [M + H] ⁺: 372.1058; found: 372.1053.

¹H (400 MHz, Acetone- d_6) and ¹³C{¹H} (101 MHz, Acetone- d_6) NMR spectra of **4ab**

SM-3-140B single_pulse









S-methyl 2-(5-(4-bromophenyl)-2-(3,5-dichlorophenyl)-1*H*-pyrrol-3-yl)-2oxoethanethioate 4ac: Prepared according to the general procedure discussed above:



reaction time, 32 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (5%); yellow solid (66 mg, 78%); mp 140-145 °C. ¹H NMR (400 MHz, Acetone- d_6): $\delta = 11.74$ (s, 1 H), 7.73 (d, J = 8.4 Hz, 2 H), 7.70 (d, J = 2.0 Hz, 2 H), 7.58 (d, J = 8.8 Hz, 2 H), 7.50 (t, J = 2.0 Hz, 1 H), 7.33 (d, J = 2.8 Hz, 1 H), 2.33 (s, 3 H) ppm; ¹³C{¹H} NMR (101 MHz, Acetone- d_6): $\delta = 194.0$, 181.2, 138.4, 134.5, 134.2, 133.1, 132.0 (2 CH), 130.4, 128.2, 128.0 (2 CH), 126.6 (2 CH), 125.1, 120.7,

116.7, 110.8, 10.5 ppm; HRMS (ESI-QTOF): *m*/*z* calcd for C₁₉H₁₂BrCl₂NO₂SNa [*M* + Na] ⁺: 489.9047; found: 480.9052.

¹H (400 MHz, Acetone- d_6) and ¹³C{¹H} (101 MHz, Acetone- d_6) NMR spectra of **4ac**



CI

CI

ESI-55: Analytical and spectral data of 4ad



S-methyl 2-(5-cyclohexyl-2-(3,5-dichlorophenyl)-1H-pyrrol-3-yl)-2-oxoethanethioate 4ad: Prepared according to the general procedure discussed above: reaction time, 36 h; $R_f = 0.2$; eluent, EtOAc/*n*-hexane MeSOC (5%); red solid (39.6 mg, 55%); mp 96-100 °C. ¹H NMR (400 MHz, CDCl₃): δ = 8.46 (s, 1 H), 7.44 (d, J = 2.0 Hz, 2 H), 7.34 NH (t, J = 1.6 Hz, 1 H), 6.70 (dd, J = 2.8, 0.8 Hz, 1 H), 2.59 - 2.52

(m, 1 H), 2.34 (s, 3 H), 2.03 – 1.98 (m, 2 H), 1.85 – 1.81 (m, 2 H), 1.76 – 1.70 (m, 1 H), 1.46 – 1.37 (m, 4 H), 1.35 – 1.31 (m, 1 H) ppm; ${}^{13}C{}^{1}H$ NMR (101 MHz, Acetone- d_6): δ = 194.4,

181.3, 141.0, 135.8, 135.1, 134.2, 127.7, 127.5 (3 CH), 115.1, 108.3, 36.5, 32.7 (2 CH₂), 26.1 (2 CH₂), 25.8, 10.4 ppm; HRMS (ESI-QTOF): *m*/*z* calcd for C₁₉H₂₉Cl₂NO₂SNa [*M* + Na]⁺: 418.0412; found: 418.0401.

 ^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, Acetone- d_6) NMR spectra of 4ad $_{\text{SM-3-17B}}$ single_pulse





ESI-56: Analytical and spectral data of 4ae



S-methyl (E)-2-(2-(4-bromo-2-chlorostyryl)-5-(p-tolyl)-1H-pyrrol-3-yl)-2-oxoethanethioate



4ae: Prepared according to the general procedure discussed above: reaction time, 28 h; $R_{\rm f}$ = 0.2; eluent, EtOAc/*n*-hexane (10%); red solid (27.2 mg, 40%); mp 212-215 °C. ¹H NMR (400 MHz, CDCl₃): δ = 9.15 (s, 1 H), 7.88 (d, *J* = 16.8 Hz, 1 H), 7.66 (d, *J* = 8.4 Hz, 1 H), 7.58 (d, *J*

= 2.4 Hz, 1 H), 7.46 (d, 2 H), 7.32 – 7.26 (m, 3 H), 7.24 – 7.21 (m, 2 H), 2.40 (s, 3 H), 2.38 (s, 3 H) ppm; $^{13}C{^{1}H}$ NMR (101 MHz, CDCl₃): δ = 194.3, 180.8, 138.5, 138.1, 134.7, 134.5, 134.5, 132.8, 129.9 (2 CH), 128.2, 127.9, 127.7, 127.0, 124.6 (2 CH), 124.3, 120.6, 117.9, 110.1, 21.3, 11.7 ppm; HRMS (ESI-QTOF): *m/z* calcd for C₂₂H₁₈BrClNO₂S [*M* + H]⁺: 473.9930; found: 473.9926.

^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 4ae $_{\text{single_pulse}}^{\text{SM-3-3A}}$



ESI-57: General procedure for the synthesis of 5a-c¹⁰

Aniline, glycine methyl ester, or cyclohexyl amine (2 equiv/mmol) was added into a solution of **4h** (50 mg, 0.16 mmol, 1 equiv) or **4a** (50 mg, 0.12 mmol, 1 equiv) in MeCN (2 mL) and 30 mol% of copper acetate [Cu(OAc)₂]. The mixture was stirred at room temperature for 8 h. After the reaction (TLC), a saturated ammonium chloride solution was added, and the product was extracted with DCM. The combined organic layers were dried over anhydrous Na₂SO₄ and filtered, and the filtrate was concentrated under reduced pressure to get a residue. The crude residue was passed through a short pad of silica gel column [230– 400 mesh; eluent: ethyl acetate/*n*-hexane] to obtain the desired **5a-c**.



2-(2,5-Diphenyl-1*H***-pyrrol-3-yl)-2-oxo-***N***-phenylacetamide 5a: Prepared according to the general procedure discussed above: R_{\rm f} = 0.2; eluent, EtOAc/***n***-hexane (25%); Yellow solid (42 mg, 74%); mp 140–145 °C; ¹H NMR (400 MHz, CDCl₃): \delta = 9.10 (s, 1 H), 8.90 (s, 1 H), 7.82 (d,** *J* **= 2.8 Hz, 1 H), 7.70 – 7.60 (m, 4 H), 7.57 – 7.53 (m, 2 H), 7.49 – 7.34 (m, 7 H), 7 .31 – 7.26 (m, 1 H), 7.20 – 7.11 (m, 1 H).; ¹³C{¹H} NMR (101 MHz, CDCl₃\delta = 181.8, 160.3, 142.3, 137.0, 132.6, 131.9, 131.2,129.3, 129.3 (2 CH), 129.2 (2 CH), 129.0 (2 CH), 128.5 (2 CH), 127.5, 125.0, 124.3 (2 CH), 119.8 (2 CH), 117.6, 111.5 ppm; HRMS (ESI-QTOF):** *m/z* **calcd for C₂₄H₁₈N₂O₂Na**

[*M* + Na]⁺: 389.1266; found: 389.1260.

(Methyl (2-(2,5-diphenyl-1*H*-pyrrol-3-yl)-2-oxoacetyl)glycinate 5b: Prepared according to MeO_2C HN HN HN HN HN O Ph HN HN Ph HN HNR (400 MHz, CDCl₃ δ = 8.93 (s, 1 H), 7.65 (d, J = 2.8 Hz, 1 H), 7.65 - 7.56 (m, 3 H), 7.56 - 7.48 (m, 2 H), 7.48 - 7.33 (m, 5H), 7.28 (d, J = 7.2 Hz, 1 H), 4.09 (d, J = 5.6 Hz, 2 H), 3.76 (s, 3 H) ppm; $^{13}C{^{1}H}$ NMR (101 MHz, CDCl₃): δ = 181.4, 169.6, 163.1, 141.8, 132.5, 131.8, 131.2, 129.2, 129.1 (2 CH), 129.0 (2 CH), 128.5 (2 CH), 127.5,124.3 (2 CH), 117.7, 111.3, 52.6, 41.2 ppm; HRMS (ESI-

QTOF): m/z calcd for C₂₁H₁₈N₂O₄Na [M + Na]⁺: 385.1165; found: 385.1164.

¹⁰ Maity, R.; Ghosh, S.; Das, I. Integrating Regioselective $E \rightarrow Z$ Isomerization of Trienones with Cascade Sequences under Photosensitizer-Free Direct Irradiation at 390 nm. *Chem. Eur. J.* 2023, **29**, e202300421.

N-cyclohexyl-2-(5-(3,5-dichlorophenyl)-2-phenyl-1*H*-pyrrol-3-yl)-2-oxoacetamide; 5c:



Prepared according to the general procedure discussed above: R_f = 0.2; eluent, EtOAc/*n*-hexane (10 %); Yellow solid (45.1 mg, 80%); mp 215–220 °C; ¹H NMR (400 MHz, DMSO- d_6): δ = 12.21 (s, 1 H), 8.44 (d, J = 8.0 Hz, 1 H), 7.75 – 7.72 (m, 2 H), 7.68 (d, J = 2..0 Hz, 2 H), 7.62 (t, J = 2.0 Hz, 1 H), 7.39 (t, J = 7.6 Hz, 2 H), 7.25 (t, J = 7.6 Hz, 1 H), 7.10 (d, J = 2.4 Hz, 1 H), 3.54 – 3.47 (m, 1 H), 1.71 – 1.51 (m, 4 H), 1.53 (d, J = 12.4 Hz, 1 H), 1.30 – 1.03 (m, 5 H) ppm; ¹³C{¹H} NMR (101 MHz DMSO- d_6): δ = 186.4, 165.2, 136.5, 135.0, 134.0, 133.8 (2 CH), 131.5, 129.4 (2 CH), 128.5 (2 CH), 128.1, 127.7, 125.2 (2 CH), 119.2, 110.7, 48.2, 32.5 (2 CH₂), 25.6, 25.2 (2 CH₂) ppm; HRMS (ESI-QTOF): m/z calcd for C₂₄H₂₃Cl₂N₂O₂

[*M* + H]⁺: 441.1136; found: 441.1129.

^1H (400 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl₃) NMR spectra of 5a



^1H (400 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl₃) NMR spectra of 5b



^1H (400 MHz, CDCl_3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 5c





ESI-58: Mechanistic Studies:

ESI-58-01: UV-vis Spectrum:

1. UV-vis absorption data for compound **2a** is reported for 10 different concentrations of acetonitrile solvent. Interestingly, it can be observed from the spectrum that increasing the concentration leads to an elongation of the absorption tail towards 550 nm.



Entry	Concentration (mM)	Absorption
1	50	0.007
2	100	0.011
3	150	0.032
4	200	0.12
5	300	0.42
6	500	1.05



Molar Absorptivity (ε) at 420 nm = 164 M⁻¹cm⁻¹

2. UV-vis absorption data for compounds **2a**, **2c**, **2h**, **2i**, **2m**, **2o**, and **2q** in acetonitrile solvent at 15 mM concentration are reported.



3. UV-vis absorption data for compounds olefin **2a**, vinyl azide **1a**, and their mixture in acetonitrile solvent at 40 mM concentration are reported.

Entry	Compound	$\lambda_{max}(nm)$
1	2a	318
2	1a	254
3	1a+2a	318

The **absence** of a new peak in the absorption spectra of the mixture 2a + 1a provides conclusive evidence that the formation of any donor-acceptor complex can be ruled out.¹¹



4. UV-vis absorption spectra for olefins 2k, 2t, 2y, 2z, and 2aa in MeCN solvent at 100 mM concentration are reported. None of them absorbed in the visible region or their tail show any such absorption, except 2y. So, olefins 2k, 2t, 2z, and 2aa did not undergo photocycloaddition. Although 2y absorbs in the visible region, but it did not undergo photocycloaddition due to the steric hindrance from the phenyl rings.



¹¹ Rai, P.; Maji, K.; Jana, S. K.; Maji, B. Intermolecular dearomative [4 + 2] cycloaddition of naphthalenes via visible-light energy-transfer-catalysis. *Chem. Sci.* **2022**, *13*, 12503–12510.

ESI-58-02: Fluorescence quenching experiments¹²

An experiment on fluorescence quenching was conducted using an Agilent Technologies Cary Eclipse fluorescence spectrophotometer, with an excitation wavelength of 425 nm and an emission wavelength of 500 nm. The experiment was performed in a solution of 10 mM of thioester **2a**. The maximum emission of vinyl azide **1a** (10 mM in CH₃CN) was observed at 500 nm. Upon the gradual addition of **1a** (1.0-6.0 equiv.), the fluorescence intensity gradually decreased, as demonstrated in the data below.



ESI-58-03: Cyclic Voltammetry Analysis:

All cyclic voltammetry studies were measured using the Admiral Squidstat Solo instrument at room temperature in acetonitrile solvent (5.0 mL). Before collecting CV data, the solution was degassed with argon gas. $^{n}Bu_{4}NPF_{6}$ (0.1 M) was used as the supporting electrolyte, and a glassy carbon electrode was used as the working electrode while the counter electrode

¹² Li, L.; Li, J.-Z.; Sun, Y.-B.; Luo, C.-M.; Qiu, H.; Tang, K.; Liu, H.; Wei, W.-T. Visible-Light-Catalyzed Tandem Radical Addition/ 1,5-Hydrogen Atom Transfer/Cyclization of 2-Alkynylarylethers with Sulfonyl Chlorides. *Org. Lett.* **2022**, *24*, 4704-4709.

was a platinum plate. The reference was an Ag/AgCl electrode submerged in a saturated aqueous KCl solution. The scan rate was 100 mV/s ranging from -2.0 V to +2.0 V.

Discussion: The CV of thioester **2a** (0.01 M) showed a reduction potential -1.6 V (vs Ag/AgCl) and Vinyl azide **1a** reduction potential -1.25 V (vs Ag/AgCl). So, in this reaction vinyl azide **1a** preferentially undergo reduction to form diradical to initiate the reaction. During the CV, the overall reduction potential of mixture value changes -1.36 V (vs. Ag/AgCl), so there might be some interaction between **2a** and **1a**.



5.0 mL of CH₃CN solvent: (i) containing 0.1 M of ${}^{n}Bu_{4}NPF_{6}$ (black line); (ii) containing 0.1 M of ${}^{n}Bu_{4}NPF_{6}$ with 0.01 M of **1a** (red line); (iii) containing 0.1 M of ${}^{n}Bu_{4}NPF_{6}$ with 0.01 M of **2a** (pink line); (iv) containing 0.1 M of ${}^{n}Bu_{4}NPF_{6}$ with 0.01 M of **2a** (blue line).

ESI-58-04: Effects of triplet and singlet quenchers on the reaction efficiency:



To understand whether a singlet or triplet diradical involved in the reaction, we performed control experiments in the presence of singlet and triplet quenchers. The reaction efficiency and product yield were significantly affected when **2a (0.1 mmol)** and **1a** (0.12 mmol) were exposed to irradiation in the presence of 0.1 M acetonitrile solvent medium either in an

oxygen atmosphere, or using 1,3-cycloheaxadiene (1 equiv) as a triplet quencher. The expected product **3a** was obtained in very less yeild, with 38 % and 47% yields, respectively. While the addition of azulene (1 equiv), a singlet and triplet quencher, had a detrimental effect on the yield (33%) with an increased reaction time (15 h). With these observations, we proposed that the reaction might proceed through a triplet state mechanism¹³.

ESI-58-05: Analytical and spectral data of 8a

To verify that compound **2** forms its triplet excited state, we studied the potential *E-Z* isomerization of the olefin, since triplet intermediates are known to be capable of *trans-cis* isomerization¹⁴. When we exposed olefin **2h** (50 mg, 0.24 mmol), dissolved into acetonitrile (2 mL, 0.12 M), to 420 nm irradiation for 6 hr at room temperature, the *Z*-**2h** isomer was produced (*E/Z* ratio = 3.3:1) along with the very less [2+2] cycloaddition product, and after completion of the reaction(TLC), the [2+2] cycloaddition product **8a** can be isolated with 42% yield by purifying through column chromatography [230–400 mesh; eluent: ethyl acetate/*n*-hexane].



S,S'-dimethyl 2,2'-(3,4-diphenylcyclobutane-1,2-diyl)bis(2-oxoethanethioate) 8a: Prepared



according to the general procedure discussed above: $R_f = 0.2$; eluent, EtOAc/*n*-hexane (5%); Light yellow liquid (21 mg, 42%). ¹H NMR (400 MHz, CDCl₃) $\delta = 7.36 - 7.26$ (m, 4 H), 7.26 (d, J = 1.3 Hz, 3 H), 7.24 (d, J = 1.6 Hz, 3 H), 4.23 - 4.15 (m, 2 H), 3.97 - 3.87 (m, 2 H), 2.32 (s, 6 H) ppm; ¹³C{¹H} NMR (101 MHz, CDCl₃): $\delta = 192.4$ (2 CO), 191.35 (2 CO), 140.3 (2 CH), 128.8 (4 CH), 127.5 (2 CH), 127.0 (4 CH), 46.6 (2 CH), 45.7 (2

CH), 11.4 (2 CH) ppm; HRMS (ESI-QTOF): *m*/*z* calcd for C₂₂H₂₀O₄S₂Na [*M* + Na]⁺: 435.0701; found: 435.0693.

¹³ Le, T. M. T.; Brégent, T.; Jubault, P.; Poisson, T. Photocatalytic $E \rightarrow Z$ Contra-Thermodynamic Isomerization of Vinyl Silanes with Lewis Base. Chem. Eur. J. **2022**, 28, e202201514.

¹⁴ Wang, J. S.; Wu, K.; Yin, C. Z.; Li, K.; Huang, Y. H.; Ruan, J.; Feng, X. M.; Hu, P.; Su, C. Y. Cage-confined photocatalysis for wide-scope unusually selective 2 + 2 cycloaddition through visible-light triplet sensitization. *Nat. Commun.* **2020**, *11*, 4675.

¹H (400 MHz, CDCl₃) spectra of crude reaction after 6 hr.



^1H (400 MHz, CDCl3) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl3) NMR spectra of 8a



ESI-58-06: Analytical and spectral data of 6a and 6b:

Under an argon atmosphere, vinyl azide 1 (50 mg) was dissolved into acetonitrile (2 mL, 0.16 M) and 5 mol % of our alkene compound 2a was added to it. It was then irradiated with a 420 nm LED at room temperature (25–30 °C) for 12 hr. After the completion reaction (TLC), we immediately added DDQ (1.2 equiv) and stirred for another 24 hours. After the completion (TLC), the crude residue was purified by silica gel column chromatography [230-400 mesh; eluent: ethyl acetate/n-hexane] to obtain **6a** and **6b**.¹⁵



2,5-di-p-tolylpyrazine 6a: Prepared according to the general procedure discussed above: R_f



= 0.2; eluent, EtOAc/n-hexane (5%); yellow solid (15 mg, 36%); mp 178–182 °C. ¹H NMR (400 MHz, CDCl₃): δ = 9.03 (s, 2 H), 7.96 (d, J = 8.2 Hz, 4 H), 7.33 (d, J = 8.0 Hz, 4 H), 2.44 (s, 6 H) ppm; ¹³C{¹H} NMR (101 MHz, $CDCI_3$): δ = 150.4 (2 CH), 141.0 (2 CH), 139.9 (2 CH), 133.6 (2 CH), 129.8 (4 CH), 126.6 (4 CH), 21.39 (2 CH₃) ppm; HRMS (ESI-QTOF): m/z calcd for C₁₈H₁₇N₂ [M +

H]⁺: 261.1392; found: 261.1384.



2,5-di-m-tolylpyrazine 6b: Prepared according to the general procedure discussed above: Rf = 0.2; eluent, EtOAc/n-hexane (5%); yellow solid (6 mg, 20%); mp 50-56 °C. ¹H NMR (600 MHz, CDCl₃): δ = 9.08 (s, 2H), 7.91 (s, 2 H), 7.86 (d, J = 7.8 Hz, 2 H), 7.44 (t, J = 7.2 Hz, 2 H), 7.32 (d, J = 7.8 Hz, 2 H), 2.49 (s, 6 H) ppm; ${}^{13}C{}^{1}H$ NMR (151 MHz, CDCl₃): δ = 150.3 (2 CH), 140.8 (2 CH), 138.4 (2 CH), 135.8 (2 CH), 130.1 (2 CH), 128.5 (2 CH), 127.0 (2 CH), 123.4 (2 CH), 21.1 (2 CH₃)

ppm; HRMS (ESI-QTOF): m/z calcd for $C_{18}H_{17}N_2$ [M + H]⁺: 261.1392; found: 261.1384.

¹⁵ Chen, Z.; Ye, D.; Xu, G.; Ye, M.; Liu, L. Highly efficient synthesis of 2,5-disubstituted pyrazines from (Z)- β haloenol acetates. Org. Biomol. Chem. 2013, 11, 6699-6702.

^1H (400 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl₃) NMR spectra of **6a**

SM-3-148B single_pulse





^1H (600 MHz, CDCl₃) and $^{13}\text{C}\{^1\text{H}\}$ (151 MHz, CDCl₃) NMR spectra of 6b



ESI-58-07: Analytical and spectral data of 2y, 2z, and 2aa:

Me



S-methyl 2-oxo-4,4-diphenylbut-3-enethioate 2y⁴: R_f = 0.2; eluent, EtOAc/n-hexane (5%); Yellow Liquid (17 mg, 27%); ¹H NMR (600 MHz, CDCl₃) δ = 7.50 - 7.35 (m, 8 H), 7.32 (s, 1 H), 7.27 - 7.21 (m, 1 H), 2.30 (s, 3 H) ppm; ¹³C NMR (151 MHz, CDCl₃) δ = 193.9, 182.8, 162.8, 140.7, Ο 138.4, 130.8, 129.6 (2 CH), 129.2 (3 CH), 128.6 (2 CH), 128.3 (2

Ô C₁₇H₁₄O₂SNa [*M* + Na]⁺: 305.0612; found: 305.0611. (E)-5-(m-tolyl)pent-4-ene-2,3-dione 2z: Prepared according to the literature⁸, R_f = 0.2; eluent, EtOAc/*n*-hexane (5%); Yellow Liquid (17 mg, 27%); ¹H NMR (400 MHz, CDCl₃) δ = 7.80 (d, J = 16.4 Hz, 1H), 7.45 – 7.39 (m, 3H), Me 7.29 (t, J = 7.2 Hz, 1H), 7.26 – 7.20 (m, 1H), 2.44 (s, 3H), 2.37 (s, 3H) II ppm; ¹³C NMR (101 MHz, CDCl₃) δ = 199.10, 186.92, 148.12, O 138.86, 134.46, 132.40, 129.59, 129.02, 126.39, 117.82, 24.49, 21.36 ppm; HRMS (ESI-QTOF): m/z calcd for $C_{12}H_{13}N_2O_2 [M + H]^+$:

Me

(E)-4-(4-bromophenyl)-1,1,1-trifluorobut-3-en-2-one 2aa: Prepared according to the



literature procedure⁹: $R_f = 0.2$; eluent, EtOAc/*n*-hexane (5%); Yellow solid (0.15 mg, 54%); mp 57-58 °C. ¹H NMR (400 MHz, CDCl₃) δ = 7.87 (d, J = 16.0 Hz, 1H), 7.62 – 7.53 (m, 2H), 7.53 – 7.44 (m, 2H), 6.99 (dd, J = 16.0, 1.2 Hz, 1H) ppm; ¹³C NMR (101 MHz, CDCl₃) δ = 180.0 (q, J_{C-F} = 35.6 Hz), 148.7, 132.7, 132.3,

CH), 116.2, 11.6 ppm; HRMS (ESI-QTOF): m/z calcd for

130.6, 127.1, 116.1 (q, J_{C-F} = 290.7 Hz) ppm; ¹⁹F NMR (376 MHz, CDCl₃) δ = -77.6 ppm. HRMS (EI): *m*/*z* calcd for C₁₀H₆BrF₃O [*M*]⁺: 277.9554; found: 277.9559.

^{189.0916;} found: 189.0915.

¹H (600 MHz, CDCl₃) and ¹³C{¹H} (151 MHz, CDCl₃) NMR spectra of **2y**:

05-SM-3-23B.1.1.1r SM-3-23B 1H-NMR in CDCl3



SM-3-23B single pulse decoupled gated NOE






 ^{1}H (400 MHz, CDCl_3) and $^{13}\text{C}\{^{1}\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 2z: $^{\text{SM-03-138C}}_{\text{single_pulse}}$

 ^1H (400 MHz, CDCl₃), $^{13}\text{C}\{^1\text{H}\}$ (101 MHz, CDCl₃), and ^{19}F (376 MHz, CDCl₃) NMR spectra of **2aa:** $_{\text{SM-3-149B}}^{\text{SM-3-149B}}$



SM-3-149B single pulse decoupled gated NOE

o ∥ CF3 Br′



Supporting Information

SM-3-149B single pulse decoupled gated NOE



ESI-58-08: Analytical and spectral data of 7a:



3-(Naphthalen-2-yl)-2*H***-azirine 7a:** Prepared according to the general procedure discussed in literature¹⁶: batch size, 200 mg; $R_f = 0.2$; eluent, EtOAc/*n*-hexane (5%); colorless oil (52 mg, 30%); ¹H NMR (400 MHz, CDCl₃) $\delta = 8.35$ (s, 1 H), 8.02 – 7.97 (m, 3 H), 7.93 – 7.89 (m, 1 H), 7.68 – 7.54 (m, 2 H), 1.88 (s, 2 H) ppm; ¹³C NMR (101 MHz, CDCl₃) $\delta = 166.0$, 135.6, 133.0, 132.0, 129.2 (2 CH), 128.6, 128.2, 127.2, 124.5, 123.0, 20.0 ppm; HRMS

(ESI-QTOF): m/z calcd for $C_{12}H_{10}N [M + H]^+$: 168.0813, Found: 168.0806.

¹⁶ Jiang, Y.; Park, C. M.; Loh, T. P. Transition-metal-free synthesis of substituted pyridines via ring expansion of 2-allyl-2*H*-azirines. *Org. Lett.* **2014**, *16*, 3432– 3435

 ^{1}H (400 MHz, CDCl_3) and $^{13}\text{C}\{^{1}\text{H}\}$ (101 MHz, CDCl_3) NMR spectra of 7a: $_{\text{SM-3-15B}}^{\text{SM-3-15B}}$





ESI-58-09: Light On-Off Experiments

On/off visible-light irradiation experiments proceeded to investigate the effect of photoirradiation on the system, and the result further clearly indicates that visible-light irradiation is necessary for the reaction.



ESI-58-10: Determination of Quantum Yield¹⁷:

(A) Determination of photon flux of Luzchem light λ = 420 nm (Intensity = 4300 lx):

We light up all the lamps in the photoreactor with λ =420 nm and measure the intensity using a Luxmeter, which reads 4300 lx. Then we calculate the photon flux.

The photon flux of the spectrophotometer was determined by standard ferrioxalate actinometry. The following solutions were prepared and stored in the dark:

Potassium ferrioxalate solution: 0.737 g of potassium ferrioxalate trihydrate was dissolved in 10 mL H_2SO_4 (0.05 M) and stored in the dark.

Buffered solution: Buffer solution was prepared by dissolving 2.5 g of sodium acetate and 0.5 mL of H_2SO_4 (95-98%) in 50 mL of distilled water.

General Protocol to assess the photon flux of the 420 nm blue LEDs: To a 10 mL Schlenk flask containing a stirring bar, 1 mL of the actinometer solution was added. Then, the solution was irradiated for 60 s. Immediately, a 100 μ L aliquot was added to a 10 mL volumetric flask containing 15 mg of 1, 10-phenanthroline in 3 mL of the buffer solution. The flask was filled with distilled water. The absorbance of this solution was then measured at 510 nm by UV/Vis spectrophotometry. In a similar manner, this procedure is repeated with the actinometer

¹⁷ (*a*) S. K. Hota, G. Singh and S. Murarka *Chem. Commun.*, 2024,**60**, 6268-6271; (*b*) X. Huang, X. Li, X. Xie, K. Harms, R. Riedel and E. Meggers. *Nat Commun*, 2017, **8**, 2245; (*c*) X. Huang, T. R. Quinn, K. Harms, R. D. Webster, L. Zhang, O. Wiest and E. Meggers, *J. Am. Chem. Soc.*, 2015, **139**, 9120 —9123; (*d*) C. Wang, K. Harms and E. Meggers, *Angew. Chem. Int. Ed.* 2016, **55**, 13495-13498.

solution stored in the dark. Using then the Beer's Law, the number of moles of Fe²⁺ produced by light irradiation is obtained by:

mol Fe²⁺=
$$\frac{v_1 \times v_3 \times \Delta A(510 nm)}{10^3 \times v_2 \times l \times \epsilon}$$

Where:

 v_1 = Irradiated volume (1 mL).

 v_2 = The aliquot of the irradiated solution taken to estimate Fe⁺ions (0.100 mL).

 v_3 = Final volume of the solution after complexation with 1, 10-phenanthroline (10 mL).

 ϵ (510 nm) = Molar extinction coefficient of [Fe (Phen)₃]²⁺ complex (11100 L mol⁻¹ cm⁻¹).

I = Optical path length of the cuvette (1 cm).

 ΔA (510 nm) = 0.576 (absorbance difference between the irradiated solution and the solution stored in the dark).

mol Fe²⁺=
$$\frac{1 \times 10 \times 0.576}{10^3 \times 0.1 \times 1 \times 11100}$$

= 5.19 x 10⁻⁶ mol

The photon flux (F) is obtained by using the following equation at 420 nm:

$$F_{4300 \text{ lx}}(420 \text{ nm}) = \frac{mol of Fe^{2+}}{\Phi.t.f_{4300 \text{ lx}}}$$

Where Φ (420 nm) is the quantum yield for the ferrioxalate actinometer (1.04 here), *t* is the irradiated time, and *f* is the fraction of light absorbed at λ = 420 nm. The measurement of the fraction of the light at 420 nm for the ferrioxalate solution was shown in UV below. The absorbance of the ferrioxalate solution at 420 nm is 0.734.

$$f_{4300 \text{ lx}} = 1 - 10^{-\text{A} (420 \text{ nm})}$$
$$= 1 - 10^{-0.734}$$
$$= 0.815$$

The photon flux can be calculated as follows:

$$F_{4300 \text{ lx}}(420 \text{ nm}) = \frac{5.19 \times 10^{-6}}{1.04 \times 60 \times 0.815}$$
$$= 1.02 \text{ x } 10^{-7} \text{ einsteins s}^{-1}$$



(B) Determination of photon flux of Luzchem light at λ =420 nm (Intensity = 1400 lx):

We reduce the light intensity by using only four lamps, which produce an intensity of 1400 lx. Then, at this intensity, we calculate the photon flux following the previous step.

Where:

 v_1 = Irradiated volume (1 mL).

 v_2 = The aliquot of the irradiated solution taken to estimate Fe⁺ions (0.100 mL).

 v_3 = Final volume of the solution after complexation with 1, 10-phenanthroline (10 mL).

 ε (510 nm) = Molar extinction coefficient of [Fe (Phen)₃]²⁺ complex (11100 L mol⁻¹ cm⁻¹).

I = Optical path length of the cuvette (1 cm).

 ΔA (510 nm) = 0.271 (absorbance difference between the irradiated solution and the solution stored in the dark).

mol Fe²⁺=
$$\frac{1 \times 10 \times 0.271}{10^3 \times 0.1 \times 1 \times 11100}$$
$$= 2.44 \times 10^{-6} \text{ mol}$$
$$f_{1400 \text{ /x}} = 1 - 10^{-\text{A} (420 \text{ nm})}$$
$$= 1 - 10^{-0.578}$$
$$= 0.735$$

The photon flux $(F_{1400 \text{ lx}})$ is obtained by using the following equation at 420 nm:

$$F_{1400 lx} = \frac{2.44 \times 10^{-6}}{1.04 \times 60 \times 0.735}$$
$$= 5.3 \times 10^{-8} \text{ einsteins s}^{-1}$$



(B) Quantum Yield Calculation:



S-methyl (*E*)-4-(3,5-dichlorophenyl)-2-oxobut-3-enethioate (**2a**) (30 mg, 1 eq), (1-azidovinyl) benzene (**1a**) (19 mg, 0.13 mmol, 1.2 equiv), were added in a predried 10 mL two different reaction tube. Then, CD₃CN (1.1 mL) was added to each tube under an argon atmosphere. Then, the first reaction tube was irradiated for 30 minutes using a Luzchem blue LED (420 nm) lamp when the light intensity was at 4300 lux, and the second tube was also irradiated for 30 minutes using the same lamp but at 1400 lux intensity. After 30 min, a 0.5 mL reaction aliquot from each reaction tube was taken out by a syringe and ¹H NMR was carried out. The NMR yield of product **3a** was determined.

The quantum yield for 4300 lx was calculated as follows:

$$\Phi = \frac{mol \ product}{F_{4300 \ lux} \times t \times f_{4300 \ lx}}$$
$$\Phi_{4300 \ lx} = \frac{1.63 \times 10^{-5}}{1.02 \times 10^{-7} \times 1800 \times 0.815}$$
$$\Phi_{4300 \ lx} = 0.109$$

Where the mol of the product (**3a**) formation is 1.63×10^{-5} mol, flux [F _{4300 lx}] is the photon flux determined by ferrioxalate actinometry (1.02×10^{-7} Einstein/s), *t* is the time (1800 s),

and $f_{4300 \text{ lx}}$ is the fraction of light absorbed by **1a** and **2a** at 420 nm under the reaction condition mentioned above (0.815).

And
$$\Phi_{1400\,lx} = \frac{6.2 \times 10^{-6}}{5.63 \times 10^{-8} \times 1800 \times 0.735}$$

 $\Phi_{1400\,lx} = 0.0832$

Where the mol of the product (**3a**) formation is 6.2 x 10^{-6} mol, flux [F $_{1400 \text{ lux}}$] is the photon flux determined by ferrioxalate actinometry (5.3 x 10^{-8} Einstein/s), *t* is the time (1800 s), and $f_{1400 \text{ lx}}$ is the fraction of light absorbed by **1a** and **2a** at 420 nm under the reaction condition mentioned above.

So, by decreasing the light intensity, the quantum yield(Φ) of the reaction lowers down.

(C) The relation between concentration and the quantum yield:



S-methyl (*E*)-4-(3,5-dichlorophenyl)-2-oxobut-3-enethioate (**2a**) (30 mg, 1 eq), (1-azidovinyl) benzene (**1a**) (19 mg, 0.13 mmol, 1.2 equiv), were added in five predried 10 mL reaction tubes. Then, CD₃CN was added in different volumes under an argon atmosphere in each tube to make different concentrations. Then, the mixture was irradiated for 30 minutes using a Luzchem blue LED (420 nm, 4300 lx) lamp. After 30 min, a 0.5 mL reaction aliquot was removed by a syringe, and ¹H NMR was carried out. The NMR yield of product **3a** was determined.

After the yield determination, we calculate the Quantum Yield (Φ)

Entry	Concentration (mM)	NMR Yield	Mol of 3a	Quantum
				$Yield(\Phi)$
1	100	15 %	1.63 x 10⁻⁵	0.109
2	54	14 %	1.5 x 10⁻⁵	0.100
3	36	11 %	1.21 x 10⁻⁵	0.080
4	27	10 %	1.1 x 10 ⁻⁵	0.073
5	21	8 %	8.8 x 10⁻ ⁶	0.059



(D) The relation between viscosity and the quantum yield:



S-methyl (*E*)-4-(3,5-dichlorophenyl)-2-oxobut-3-enethioate (**2a**) (30 mg, 1 eq), (1-azidovinyl) benzene (**1a**) (19 mg, 0.13 mmol, 1.2 equiv), were added in four predried 10 mL reaction tube. Then, each tube added different deuterated solvents (in 0.1 M) under an argon atmosphere. Then, the mixture was irradiated for 30 minutes using a Luzchem blue LED (420 nm, 4300 lx) lamp. After 30 min, a 0.5 mL reaction aliquot was removed by a syringe, and ¹H NMR was carried out. The NMR yield of product **3a** was determined.

After the yield determination, we calculate the Quantum Yield (Φ)

Entry	Solvent	Viscosity (20 °C)	Yield	mol of 3a	Quantum Yield(Φ)
1	DMSO-d ₆	2.24	3 %	3.3 x 10⁻ ⁶	0.022
2	CDCl₃	0.57	6 %	6.5 x 10⁻ ⁶	0.043
3	CDCl ₂	0.45	9 %	9.8 x 10⁻ ⁶	0.065
4	CD₃CN	0.37	15 %	1.63 x 10⁻⁵	0.109



ESI-58-10: Plausible Mechanism



ESI-59: Kinetic Study:

Time-dependent reaction progress monitoring during the formation of 3a from the reactants (1a and 2a) by ¹H NMR analysis (400 MHz, CD₃CN)

1a and 2a were mixed in CD₃CN in a round bottom flask under argon. Eleven equal portions of the reaction mixture were divided into septum-sealed screw-cap vials. The aliquots were exposed to 420 nm irradiation at r.t. for 0, 30, 90, 210, 390, and 510 minutes. After exposure, those aliquots were directly transferred in an NMR tube (diluted in additional CD₃CN if required), and then the NMR data were recorded as shown below.



6.0 5.5 f1 (ppm)

Supporting Information





ESI-60: Green chemistry metrics and Eco-scale calculations¹⁸:



Reaction mass efficiency (%) = $\frac{\text{Mass of desired product}}{\text{Mass of all reactants}} \times 100$ (RME)

1	Reactant 1	1a	0.145 g	1 mmol	FW 145.165
2	Reactant 2	2h	0.206 g	1 mmol	FW 206.259
3	Reactant 3	DDQ	0.227 g	1 mmol	FW 227.000
4	Additive solvent	CH ₃ CN (10 mL)	7.86 g		
5	Recycled solvent	CH₃CN (7 mL)	5.5 g		
6	Product	3h	0.210 g		FW 321.394

Product Yield = 65%

 $E- \text{ factor} = \frac{0.145+0.206+0.227+7.86-(5.5+0.21)}{0.21} = 12.99 \text{ Kg waste/ 1 Kg product}$ Atom economy = $\frac{321.394}{206.259+145.165+227.0} \times 100 = 55\%$ Atom efficiency = $(55\% \times 65\%)/100 = 35.75\%$ Carbon Efficiency = $\frac{19}{11+8+8} \times 100 = 70\%$ Reaction mass efficiency = $\frac{0.210 \text{ g}}{0.145 \text{ g}+0.206 \text{ g}+0.227 \text{ g}} \times 100 = 36.33\%$

¹⁸ Dam, B.; Sahoo, A. K.; Patel, B. K. Visible-Light-Mediated Synthesis of β-keto Sulfones using g-C₃N₄ as a Recyclable Photocatalyst under Sustainable Conditions. *Green Chem.* **2022**, *24*, 7122–7130.

ESI-61: Computational Studies:

All the molecules (**1a**, **2h**, ³**1a**, and ³**2h**(*E*), ³**2h**(*Z*)) were optimized at B3LYP/6-31G(d)¹⁹ level of theory using Q-Chem quantum chemistry software.²⁰ These structures were further used for TDDFT calculations to obtain the singlet and triplet excited states. The basis set was kept as 6-31G(d). The transition state (TS) was found using UB3LYP/6-31G(d) level of theory using the Gaussian 16 program package.²¹ The final product (**3h**) was also optimized at the same levels of theory.

FSI-61-01 · Excitation er	nergies of the sr	ecies. TD-DFT	excitation energy	ries (B3LYP	(4)) A-31G
LJI-UI-UI. LACILALIUII EI	iergies of the sp		excitation energ		/0-310(u))

State	2h	1a
Excited singlet (Bright state)	3.60	4.82
T1	2.16	2.52

The experimental irradiation of 420 nm corresponds to \sim 3 eV, which agrees reasonably with the absorption energy of the bright state of **2h**.



¹⁹ (a) Becke, A. D. Density-Functional Thermochemistry. III. The Role of Exact Exchange. *J. Chem. Phys.* **1993**, *98*, 5648–5652. (b) Lee, C.; Yang, W.; Parr, R. G. Development of the Colle-Salvetti Correlation-Energy Formula into a Functional of the Electron Density. *Phys. Rev. B.* **1988**, *37*, 785–789.

²⁰ Shao Y, Gan Z, Epifanovsky E, Gilbert ATB, Wormit M, Kussmann J, Lange AW, Behn A, Deng J, Feng X et al (**2015**) Advances in molecular quantum chemistry contained in the q-chem 4 program package. Mol Phys **113**, 184–215.

²¹ Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Petersson, G. A.; Nakatsuji, H.; Li, X.; Caricato, M.; Marenich, A. V.; Bloino, J.; Janesko, B. G.; Gomperts, R.; Mennucci, B.; Hratchian, H. P.; Ortiz, J. V.; Izmaylov, A. F.; Sonnenberg, J. L.; Williams; ; Ding, F.; Lipparini, F.; Egidi, F.; Goings, J.; Peng, B.; Petrone, A.; Henderson, T.; Ranasinghe, D.; Zakrzewski, V. G.; Gao, J.; Rega, N.; Zheng, G.; Liang, W.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Throssell, K.; Montgomery, J. A., Jr; Peralta, J. E.; Ogliaro, F.; Bearpark, M. J.; Heyd, J. J.; Brothers, E. N.; Kudin, K. N.; Staroverov, V. N.; Keith, T. A.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A. P.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Millam, J. M.; Klene, M.; Adamo, C.; Cammi, R.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Farkas, O.; Foresman, J. B.; Fox, D. J. *Gaussian 16 Rev. C.01*; Gaussian Inc.: **2016**.





Spin Density Diagram: The purple-colored blobs are the spin densities accumulated on the atoms. (calculated in B3LYP/6-31G(d) level of theory)









ESI-61-02: Cartesian Coordinates:

1	а
-	~

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С	0.5160023402	0.1671364031	0.0202214831
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³1a'

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3h'

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