Supporting Information for

Catalyst-Free Ring Opening of Azlactones in Water Microdroplets++

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Table of contents

Entry		Page No.
1	General information	S2
2	Optimization of reaction conditions	S3
3	Reaction in Bulk phase: Control experiment	S3
4	Microdroplet synthesis of N-benzoyl derivatives 2a-g	S4
5	Synthesis of dipeptide 3a	S7
6	HPLC chromatogram of racemic and chiral dipeptide 3a	S 8
7	Crystallographic data	S10
8	References	S15
9	NMR spectra	S16
10	Polarimeter data	S21
11	MS and MS ² data of $2a$	S24

1. General information

All the amino acids were purchased from Avra synthesis Pvt. Ltd., India and used without further purification. All the solvents were purchased from Merck, India and distilled using standard procedures.^{1a} For the microdropet spray used, the fused silica gel capillary for transfer of solution to the spray source (i.d. of 100 µm and o.d. of 360 µm from Ploymicro Technologies, Az, USA). TLC plate was performed on pre-coated 0.25 mm thick aluminium-backed silica gel plates purchased from Merck KGaA, Germany, and visualized with a UV lamp ($\lambda = 254$ nm) or ninhydrin stain. Flash chromatography was performed on Merck silica gel (230-400 mesh). Single crystal X-ray diffraction analysis (SC-XRD) analysis was performed on the Bruker D8 Quest model. ¹H-NMR and ¹³C-NMR spectra were recorded on JEOL ECZ500R/S1 instrument. All proton NMR chemical shifts are reported in ppm relative to tetramethylsilane (0.00 ppm) or the residual solvent peak (Chloroform- $d \delta$ 7.26 ppm and DMSO-d₆ δ 2.50). Multiplets are abbreviated as follows: s = singlet, d = doublet, t = triplet, q = quartet, m =multiplet. ¹³C NMR spectra were recorded at 126 MHz, and data are reported as follows: a chemical shift in ppm from tetramethylsilane with the solvent as an internal indicator (Chloroform- $d \delta$ 77.16 ppm and DMSO-d₆ δ 39.52). HRMS data were obtained in ESI mode by Agilent LC/Q-TOF instrument. Optical rotations were measured in an Rudolph Polarimeter AUTOPOL-V PLUS. The polarimeter optical rotation is consistent with that reported in the literature.²⁻⁴ Enantiomeric excesses of the sample were determined on an Agilent HPLC instrument using the Dacel chiralpak columns (AD-H). All the azlactones (1a-1e) were prepared according to reported procedures, and the proton NMR data is consistent with that reported in the literature.1b&c

2. Optimization of reaction conditions:

General procedure I:

Azlactone (\pm -1a) was dissolved in a water-ACN or water-dioxane, or water-acetone solvent (1 mL, 1:1) and the solution loaded into an airtight glass syringe. The solution was delivered with a syringe pump (NE-300, New Era Pump Systems, Inc. Farmingdale, NY, USA) at a flow rate of 30 µl/min. to a fused silica gel glass capillary (i.d. of 100 µm and o.d. of 360 µm, capillary length: 20 cm). Dry nitrogen, which served as the sheath gas, was operated at 120 psi pressure or, as mentioned in Table S1. The distance between spray source and collecting surface kept as 40 mm. The microdroplets were sprayed for 33.3 minutes and collected. The collected crude material was then assayed by crude ¹H NMR to assess the conversion and purified through flash column chromatography (95:5→80:20, CH₂Cl₂/MeOH), afforded the title compound racemic *N*-benzoyl alanine (2a).Table S1: Optimization of reaction conditions.^{*a*}



Entry	Solvent	Pressure	Flow rate	Concentration	Conversion ^b	Yield
	rauo (1:1)	(psi)	(µL/min.)	(mol L ⁻¹)	(%)	(%) ^c
1	H ₂ O:	120	30	0.15	45	31
	Dioxane					
2	H_2O :	120	30	0.15	100	82
	Acetone					
3	H ₂ O: ACN	120	30	0.15	100	90
4	H ₂ O: ACN	120	50	0.1	64	56
5	H ₂ O: ACN	120	30	0.1	100	94
6	H ₂ O: ACN	120	30	0.2	38	30
7	H ₂ O: ACN	50	30	0.1	42	36
8	H ₂ O: ACN	0	30	0.1	n.d.	n.d.

^{*a*}All reactions were performed with *rac*-4-methyl-2-phenyloxazol-5(4*H*)-one for 33.3 minutes. ^{*b*}Conversion was determined by crude ¹H NMR analysis. ^cYields of isolated products. n.d. = not detected.

3. Reaction in Bulk phase: Control experiment:

A 5 ml glass vial was charged with azlactone (\pm)-1a (0.1 mmol) and dissolved in 1 mL of wateracetonitrile (v:v = 1:1). The resulting mixture was stirred at room temperature for 6 hours and the organic solvent was evaporated *in vacuo*, and the reaction mixture was extracted with ethyl acetate (3×3 ml). The organic extracts were combined and dried over MgSO₄, and the organic solvent was concentrated *in vacuo* to afford corresponding crude (\pm)-2a. To determine the conversion, the crude sample was analyzed by using ¹H NMR and desired *N*-benzoyl alanine (2a) was not observed.

4. Microdroplet synthesis of *N*-benzoyl derivatives (2a-g):

General procedure II



0.1 mmol of azlactone (1) was dissolved in a water-acetonitrile in 1 mL of water/acetonitrile solvent (v:v = 1:1), and the solution was loaded into an airtight glass syringe. The solution was delivered with a syringe pump (NE-300, New Era Pump Systems, Inc. Farmingdale, NY, USA) at a flow rate of 30 μ L/min to a fused silica gel capillary (i.d. of 100 μ m and o.d. of 360 μ m, capillary length: 20 cm). The end of the capillary was equipped with a sheath-gas-assisted spray emitter. Dry nitrogen, which served as the sheath gas, was operated at 120 psi. The distance between spray source and collecting surface kept as 40 mm. The microdroplets were sprayed for 33.3 minutes and collected. The collected crude material was purified through flash column chromatography to afford the corresponding *N*-benzoyl amino acid (**2**).

N-benzoyl-L-alanine (2a)



The title compound was prepared according to the general procedure II, using (S)-4-methyl-2-phenyloxazol-5(4*H*)-one **1a** (17.5 mg, 0.1 mmol). The crude product was subjected to flash column chromatography (95:5 \rightarrow 80:20, CH₂Cl₂/MeOH), afforded the title compound **2a** as a white solid (18.2 mg, 94% yield). The structure of

the compound was determined by using ¹H NMR data.

¹H NMR (DMSO-d₆, 400 MHz): δ 8.67 (d, J = 7.3 Hz, 1H), 7.88 (m, 2H), 7.52 (m, 1H), 7.47 (m, 2H), 4.42 (qn, J = 7.3 Hz, 1H), 1.39 (d, J = 7.3 Hz, 3H). The ¹H NMR data is consistent with that reported in the literature.²

HRMS: Calculated for $C_{10}H_{10}NO_3^{-}[M-H]$ 192.0666, found 192.0670. [α] p^{22} : +24.1° (*c* 1.0, CHCl₃), lit. [α] p^{22} : +29.1° (*c* 1.0, CHCl₃)³

N-benzoyl-L-valine (2b)



The title compound was prepared according to the general procedure II, using (S)-4-isopropyl-2-phenyloxazol-5(4*H*)-one **1b** (20.3 mg, 0.1 mmol). The crude product was subjected to flash column chromatography (95:5 \rightarrow 85:15 CH₂Cl₂/MeOH), afforded the title compound **2b** as a white solid (17.9 mg, 81% yield). The structure of

the compound was determined by using ¹H NMR data.

¹H NMR (DMSO-d₆, 400 MHz): δ 8.41 (d, J = 8.1 Hz, 1H), 7.91 – 7.84 (m, 2H), 7.56 – 7.50 (m, 1H), 7.49 – 7.44 (m, 2H), 4.29 (t, J = 7.6 Hz, 1H), 2.31 – 2.11 (m, 1H), 0.97 (t, J = 6.8 Hz, 6H). The ¹H NMR data is consistent with that reported in the literature.² [a]p²⁰: +10.4 (c 1.09, MeOH), lit. [a]p²⁰: +10.0 (c 1.03, MeOH)² **N-benzoyl-L-leucine (2c)**



The title compound was prepared according to the general procedure II, using (S)-4-isobutyl-2-phenyloxazol-5(4H)-one **1c** (21.7 mg, 0.1 mmol). The crude product was subjected to flash column chromatography (95:5 \rightarrow 85:15 CH₂Cl₂/MeOH), afforded the title compound **2c** as a light yellow solid (20 mg, 85% yield). The structure of the compound was determined by using ¹H NMR data.

¹H NMR (DMSO-d₆, 400 MHz): δ 8.57 (d, J = 8.0 Hz, 1H), 7.88 (d, J = 7.3 Hz, 2H), 7.53 (t, J = 7.3 Hz, 1H), 7.46 (t, J = 7.3 Hz, 2H), 4.44 (ddd, J = 10.8, 8.0, 4.3 Hz, 1H), 1.83 – 1.65 (m, 2H), 1.58 (ddd, J = 12.8, 8.8, 4.3 Hz, 1H), 0.91 (d, J = 6.4 Hz, 3H), 0.87 (d, J = 6.4 Hz, 3H). The ¹H NMR data is consistent with that reported in the literature.² [α] p^{22} : -9.0 (*c* 1.08, MeOH), lit. [α] p^{20} : -12.8 (*c* 1.08, MeOH)²

N-benzoyl-L-phenylalanine (2d)



The title compound was prepared according to the general procedure II, using (S)-4-benzyl-2-phenyloxazol-5(4*H*)-one **1d** (17.5 mg, 0.1 mmol). The crude product was subjected to flash column chromatography (95:5 \rightarrow 80:20 CH₂Cl₂/MeOH), afforded the title compound **2d** as a white solid (23.7 mg, 88% yield). The structure

of the compound was determined by using ¹H NMR data.

¹H NMR (DMSO-d₆, 400 MHz): δ 8.59 (d, J = 8.1 Hz, 1H), 7.80 – 7.75 (m, 2H) 7.51 (t, J = 7.3 Hz, 1H), 7.44 (t, J = 7.6 Hz, 2H), 7.30 (d, J = 6.9 Hz, 2H), 7.24 (t, J = 7.6 Hz, 2H), 7.16 (t, J = 7.1 Hz, 1H), 4.59 (m, 1H), 3.21 (dd, J = 13.7, 4.3 Hz, 1H), 3.07 (dd, J = 13.7, 10.2 Hz, 1H). The ¹H NMR data is consistent with that reported in the literature.² [α] p^{20} : -32.0 (c 1.09, MeOH), lit. [α] p^{20} : -31.4 (c 1.09, MeOH)²

N-benzoyl-L-tryptophan (2e)



The title compound was prepared according to the general procedure II, using (S)-4-((1H-indol-3-yl)methyl)-2-phenyloxazol-5(4*H*)-one **1e** (29 mg, 0.1 mmol). The crude product was subjected to flash column chromatography (95:5 \rightarrow 80:20 CH₂Cl₂/MeOH), afforded the title compound **2e** as a brown solid (24 mg, 78% yield).

^{2e} ¹H NMR (DMSO-d₆, 400 MHz): δ 10.82 (d, J = 2.5 Hz, 1H), 8.63 (d, J = 7.9 Hz, 1H), 7.82 (dd, J = 7.2, 1.7 Hz, 2H), 7.61 (d, J = 7.5 Hz, 1H), 7.52 (t, J = 7.3 Hz, 1H), 7.44 (t, J = 7.3 Hz, 2H), 7.33 (d, J = 7.9 Hz, 1H), 7.22 (d, J = 2.3 Hz, 1H), 7.06 (t, J = 7.3 Hz, 1H), 6.99 (t, J = 7.6 Hz, 1H), 4.67 (m, 1H), 3.33 (dd, J = 14.6, 4.6 Hz, 1H), 3.22 (dd, J = 14.6, 9.9 Hz, 1H). The ¹H NMR data is consistent with that reported in the literature.²

HRMS: Calculated for $C_{18}H_{17}N_2O_3^+$ [M+H] 309.1234, found 309.1240. [α] p^{20} : -32.3 (c = 1.08, MeOH), lit. [α] p^{20} : -35.7 (c 1.08, MeOH)²

N-benzoyl-L-tyrosine (2f)



The title compound was prepared according to the general procedure II, using (S)-4-(4-hydroxybenzyl)-2-phenyloxazol-5(4H)-one **1f** (26.7 mg, 0.1 mmol). The crude product was subjected to flash column chromatography (95:5 \rightarrow 80:20 CHCl₃/MeOH), afforded the title compound **2d** as a white solid (21.4 mg, 75% yield). The structure of the compound was

determined by using ¹H NMR data.

¹H NMR (DMSO-d₆, 500 MHz): δ 9.21 (s, 1H), 8.62 (d, J = 8.0 Hz, 1H), 7.79 (d, J = 8.6 Hz, 2H), 7.52 (t, J = 7.3 Hz, 1H), 7.45 (t, J = 7.4 Hz, 2H), 7.10 (d, J = 8.6 Hz, 2H), 6.64 (d, J = 8.6 Hz, 2H), 4.52 (ddd, J = 10.5, 8.0, 4.5 Hz, 1H), 3.06 (dd, J = 13.8, 4.5 Hz, 1H), 2.94 (dd, J = 13.8, 10.5 Hz, 1H). The ¹H NMR data is consistent with that reported in the literature.⁴ [a]p²⁵: -33.4 (c 1.00, MeOH), lit. [a]p²³: -36.0 (c 1.1, MeOH)⁴

N-benzoyl-*L*-methionine (2g)



The title compound was prepared according to the general procedure II, using (S)-4-(2-(methylthio)ethyl)-2-phenyloxazol-5(4*H*)-one **1g** (23.5 mg, 0.1 mmol). The crude product was subjected to flash column chromatography (95:5 \rightarrow 85:15 CH₂Cl₂/MeOH), afforded the title compound **2g** as a white solid (20.2 mg, 80% yield). The

structure of the compound was determined by using ¹H NMR data.

¹H NMR (DMSO-d₆, 500 MHz): δ 8.64 (d, J = 7.8 Hz, 1H), 7.88 (d, J = 7.0 Hz, 2H), 7.59 – 7.50 (m, 1H), 7.48 (t, J = 7.3 Hz, 2H), 4.58 – 4.47 (m, 1H), 2.65 – 2.49 (m, 2H), 2.12 – 2.04 (m, 2H), 2.05 (s, 3H). The ¹H NMR data is consistent with that reported in the literature.² [α] ρ ²⁵: -14.5° (*c* 1.00, MeOH), lit. [α] ρ ²⁰:-21.0 (*c* 1.03, MeOH)²

5. Synthesis of dipeptide 3a:^{5a}



In a 25 mL round bottom flask charged with benzoyl-*L*-alanine (19.3 mg, 0.1 mmol), methyl glycinate hydrochloride (12.5 mg, 0.1 mmol), oxymapure (0.11 mmol., 1.1 equiv.), and DMF (2 ml) at room temperature in a nitrogen atmosphere. DIC (18.9 mg, 0.15 mmol, 1.5 equiv.) and NEt₃ (3.0 equiv.) were added to the reaction mixture at room temperature and stirred for 10 h. Quenched the reaction mixture with cold water and extracted with dichloromethane (3×5 ml), dried over anhydrous Na₂SO₄ and concentrated. The crude product was purified by column chromatography (CH₂Cl₂: MeOH = 40:1), afforded the title compound (*S*)-**3a** as a white solid (15.8 mg, 60% yield). For HPLC analysis, the racemic sample of **3a** has been synthesised from racemic **2a**.

¹**H NMR (CDCl₃, 500 MHz):** δ 7.80 (d, *J* = 8.3 Hz, 2H), 7.58 – 7.42 (m, 2H), 7.43 – 7.30 (m, 3H), 4.83 (qn, *J* = 7.1 Hz, 1H), 4.01 (d, *J* = 4.7 Hz, 2H), 3.69 (s, 3H), 1.47 (d, *J* = 6.9 Hz, 3H).

¹³C NMR (CDCl₃, 126 MHz): δ 173.2, 170.3, 167.5, 133.7, 131.9, 128.6, 127.5, 52.4, 49.1, 41.3, 18.4. The NMR data is consistent with that reported in the literature.^{5a}

HRMS: Calculated for $C_{13}H_{17}N_2O_4^+$ [M+H] 265.1183, found 265.1187.

HPLC: 90% *ee* (95:5 er) (Daicel CHIRALPAK AD-H, Hexane: ^{*i*}PrOH (90:10) 1.0 mL/min; $T_{\text{major}} = 19.6 \text{ min}, T_{\text{minor}} = 12.8 \text{ min}, 254 \text{ nm}$ absorbance)

[α]**D**²⁵: -23.7 (*c* 1.00, CHCl₃)



6. HPLC chromatogram of racemic and chiral dipeptide 3a

Signal: DAD1B,Sig=254,4 Ref=off

Name	Peak Retention Time	Area	Area%	Height
	12.885	12821282	50.06	4946854.90
	19.647	12788340	49.94	3516759.59



Signal: DAD1B,Sig=254,4 Ref=off

Name	Peak Retention Time	Area	Area%	Height
	12.877	1801156	5.00	716442.20
	19.606	34222462	95.00	9428735.54



Figure S1: A picture of the experimental setup for the collection of the sprayed microdroplets in a round bottom flask (100 ml), connected to a condenser and solvent trap. The reaction duration calculated based on applied pressure, and the distance between spray source and collecting surface (40 mm). The calculated reaction duration found to be 483 μ s (less than 0.5 ms).^{5b}

7. X-ray crystallographic data:



Table S2. Crystal data and structure refinement of product $3a^6$

Identification code	EG_VSR_DIPEP_1292
Empirical formula	$C_{13}H_{16}N_2O_4$
Formula weight	264.28
Temperature/K	304.0
Crystal system	monoclinic
Space group	$P2_1/c$
a/Å	13.5013(11)
b/Å	11.4480(10)
c/Å	9.0667(8)
α/°	90
β/°	100.556(4)
γ/°	90
Volume/Å ³	1377.7(2)
Ζ	4
$\rho_{calc}g/cm^3$	1.274
µ/mm ⁻¹	0.095
F(000)	560.0

_0ma_a

Crystal size/mm ³	$0.146 \times 0.049 \times 0.021$			
Radiation	MoKa ($\lambda = 0.71073$)			
2Θ range for data collection/	°4.698 to 50.092			
Index ranges	$\text{-16} \le h \le 16, \text{-13} \le k \le 13, \text{-10} \le l \le 10$			
Reflections collected	33644			
Independent reflections	2435 [$R_{int} = 0.0721, R_{sigma} = 0.0336$]			
Data/restraints/parameters	2435/0/174			
Goodness-of-fit on F ²	1.031			
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0398, wR_2 = 0.0862$			
Final R indexes [all data]	$R_1 = 0.0682, wR_2 = 0.1037$			
Largest diff. peak/hole / e Å ⁻³ 0.14/-0.14				

Table S3 Fractional Atomic Coordinates (×10⁴) and Equivalent Isotropic Displacement Parameters ($Å^2 \times 10^3$) for EG_VSR_DIPEP_1292_0ma_a. U_{eq} is defined as 1/3 of of the trace of the orthogonalised U_{IJ} tensor.

Atom x		У	ζ	U(eq)	
01	1878.1(10)	6571.2(12)	1902.8(14)	60.8(4)	
02	6612.7(11)	5882.5(12)	5947.3(15)	68.4(4)	
03	4029.3(10)	6740.6(11)	5382.7(13)	54.4(4)	
O4	6175.0(12)	7749.0(13)	5556.3(16)	71.4(4)	
N1	2173.0(11)	7742.5(14)	3914.5(16)	51.2(4)	
N2	4614.7(12)	7027.9(14)	3252.0(16)	52.3(4)	
C1	597.7(14)	6667.3(17)	3380(2)	50.3(5)	
C2	105.5(16)	7341.3(19)	4287(2)	61.3(6)	
C3	-820.9(17)	7001(2)	4574(3)	78.2(7)	
C4	-1263.5(18)	5989(3)	3962(3)	85.6(8)	
C5	-787.8(19)	5322(2)	3060(3)	83.3(7)	
C6	139.4(16)	5658.8(19)	2756(2)	67.3(6)	
C7	1597.8(14)	6986.2(16)	3015(2)	48.4(5)	
C8	3142.1(14)	8120.1(16)	3607(2)	50.4(5)	

C9	3953.6(14)	7220.0(16)	4149.4(19)	46.4(5)
C10	5478.7(15)	6295.3(17)	3744(2)	53.4(5)
C11	6111.4(14)	6747.7(18)	5168(2)	50.6(5)
C12	7257.4(19)	6202(2)	7353(3)	86.2(8)
C13	3451.0(16)	9274.3(17)	4388(2)	67.4(6)

TableS4AnisotropicDisplacementParameters $(Å^2 \times 10^3)$ forEG_VSR_DIPEP_1292_0ma_a.The Anisotropic displacement factor exponent takes theform: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+...].$

Aton	n U11	U22	U33	U23	U13	U12
01	67.5(9)	69.6(9)	47.0(8)	-4.6(7)	15.2(7)	0.4(7)
02	76(1)	59.6(9)	64.8(9)	-2.3(7)	0.1(8)	13.3(8)
03	67.1(9)	62.4(8)	35.9(7)	5.1(6)	15.0(6)	2.7(7)
O4	85.1(11)	51.6(9)	72(1)	-4.6(8)	0.1(8)	-2.0(8)
N1	49.6(10)	63.6(10)	41.7(8)	-2.4(8)	11.9(7)	-3.6(8)
N2	57.6(10)	66.3(11)	34.7(8)	3.5(7)	13.3(8)	1.8(8)
C1	46.2(11)	57.6(12)	44.7(10)	6.6(9)	2.2(9)	0(1)
C2	53.0(13)	73.5(14)	57.3(12)	-1.2(11)	9.7(10)	-2.3(11)
C3	53.9(14)	107(2)	76.7(16)	1.4(14)	18.6(12)	3.1(14)
C4	52.0(15)	106(2)	97.8(19)	10.4(17)	10.3(14)	-11.2(15)
C5	60.6(16)	79.8(17)	104(2)	-4.7(15)	1.8(14)	-16.9(13)
C6	58.2(14)	66.3(14)	75.1(15)	-4.4(12)	6.0(11)	-2.1(11)
C7	51.7(12)	52.2(11)	40.2(10)	6.2(9)	5.2(9)	4.2(9)
C8	48.4(12)	61.5(13)	41.7(10)	3.9(9)	9.7(9)	-4.3(10)
C9	50.0(12)	53.5(11)	36(1)	-5.3(9)	8.9(9)	-7.5(9)
C10	59.9(13)	57.8(12)	45.6(11)	-3.6(9)	17.8(10)	0.9(10)
C11	50.7(12)	53.1(13)	51.2(11)	1.5(10)	17.9(9)	0.9(10)
C12	86.4(18)	91.7(18)	70.0(15)	-3.2(13)	-13.3(13)	19.3(14)
C13	66.7(14)	55.2(13)	79.5(15)	4.1(11)	11.4(12)	-5.8(11)

Atom	Atom	Length/Å	Atom	Atom	Length/Å
01	C7	1.235(2)	C1	C6	1.381(3)
02	C11	1.328(2)	C1	C7	1.493(3)
02	C12	1.452(3)	C2	C3	1.379(3)
O3	C9	1.233(2)	C3	C4	1.373(3)
O4	C11	1.198(2)	C4	C5	1.363(3)
N1	C7	1.337(2)	C5	C6	1.385(3)
N1	C8	1.453(2)	C8	C9	1.519(3)
N2	C9	1.332(2)	C8	C13	1.521(3)
N2	C10	1.440(2)	C10	C11	1.504(3)
C1	C2	1.384(3)			

Table S5 Bond Lengths for EG_VSR_DIPEP_1292_0ma_a.

Table S6 Bond Angles for EG_VSR_DIPEP_1292_0ma_a.

Atom Atom Angle/°				Atom Atom Atom Angle/°			
C11	02	C12	116.35(16)	01	C7	C1	120.78(18)
C7	N1	C8	121.24(15)	N1	C7	C1	118.09(17)
С9	N2	C10	120.18(15)	N1	C8	C9	110.95(15)
C2	C1	C7	123.22(18)	N1	C8	C13	110.42(16)
C6	C1	C2	118.81(19)	C9	C8	C13	108.59(15)
C6	C1	C7	117.96(18)	03	С9	N2	121.81(17)
C3	C2	C1	120.4(2)	03	С9	C8	122.06(16)
C4	C3	C2	120.3(2)	N2	С9	C8	116.02(16)
C5	C4	C3	119.8(2)	N2	C10	C11	111.47(15)
C4	C5	C6	120.4(2)	02	C11	C10	110.73(17)
C1	C6	C5	120.3(2)	O4	C11	02	123.77(19)
01	C7	N1	121.13(18)	O4	C11	C10	125.50(19)

Table S7 Hydrogen Atom Coordinates (Å×10⁴) and Isotropic Displacement Parameters (Å²×10³) for EG_VSR_DIPEP_1292_0ma_a.

Atom	X	У	Ζ	U(eq)
H1	1965.48	8010.59	4690.2	61
H2	4525.29	7342.32	2376.19	63
H2A	401.01	8027.3	4706.58	74
Н3	-1147.42	7459.9	5183.49	94
H4	-1886.03	5759.35	4163.1	103
Н5	-1087.51	4636.29	2646.37	100
Н6	454.74	5203.93	2128.08	81
H8	3087.96	8224.72	2522.23	60
H10A	5884.2	6261.66	2965.54	64
H10B	5254.98	5508.8	3909	64
H12A	7643.61	5534.85	7760.49	129
H12B	6849.63	6466.03	8050.08	129
H12C	27705.08	6817	7176.68	129
H13A	4101.93	9499.52	4201.9	101
H13B	3479.61	9187.34	5447.86	101
H13C	2966.26	9864.65	4006.48	101

Experimental

Single crystals of $C_{13}H_{16}N_2O_4$ [EG_VSR_DIPEP_1292_0ma_a] were []. A suitable crystal was selected and [] on a 'Bruker APEX-II CCD' diffractometer. The crystal was kept at 304.0 K during data collection. Using Olex⁷, the structure was solved with the XT⁸ structure solution program using Intrinsic Phasing and refined with the XL⁹ refinement package using Least Squares minimisation.

Crystal structure determination of [EG_VSR_DIPEP_1292_0ma_a]

Crystal Data for C₁₃H₁₆N₂O₄ (M =264.28 g/mol): monoclinic, space group P2₁/c (no. 14), a = 13.5013(11) Å, b = 11.4480(10) Å, c = 9.0667(8) Å, $\beta = 100.556(4)^{\circ}$, V = 1377.7(2) Å³, Z = 4, T = 304.0 K, μ (MoK α) = 0.095 mm⁻¹, *Dcalc* = 1.274 g/cm³, 33644 reflections measured (4.698° $\leq 2\Theta \leq 50.092^{\circ}$), 2435 unique ($R_{int} = 0.0721$, $R_{sigma} = 0.0336$) which were used in all calculations. The final R_1 was 0.0398 (I > 2 σ (I)) and wR_2 was 0.1037 (all data).

8. References:

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9. NMR Spectra:













10. Polarimeter data of chiral 2 and 3a:



IIT Roorkee Thursday, 29-AUG-2024 Ph 0 HO This sample was measured on an Autopol V Plus, Serial #85232 Manufactured by Rudolph Research Analytical, Hackettstown, NJ, USA. \cap 2d Set Temperature : 20.0 Temperature Correction : Sucrose Average Std.Dev. Maximum Minimum n 5 -31.247 -32.016 0.5289 -32.671 Sample ID OR °Arc S.No Time Result Scale WLG Conc. Temp. 1 N-Bz-PhAla 03:33:35 PM -32.671 SR -0.356 589 1.090 20.4 1.090 20.4 2 N-Bz-PhAla 03:33:46 PM -32.293 SR -0.352 589 3 03:33:57 PM -32.001 -0.349 589 1.090 20.3 N-Bz-PhAla SR 4 N-Bz-PhAla 03:34:08 PM -31.247 SR -0.341 589 1.090 20.3 5 N-Bz-PhAla 03:34:19 PM -31.869 SR -0.347 589 1.090 20.3 IIT Roorkee Thursday, 29-AUG-2024 This sample was measured on an Autopol V Plus, Serial #85232 Manufactured by Rudolph Research Analytical, Hackettstown, NJ, USA. HO н Set Temperature : 20.0 ö Temperature Correction : Sucrose 2e n Average Std.Dev. Maximum Minimum 5 -32.250 0.0909 -32.363 -32.116 OR °Arc S.No Sample ID Time Result Scale WLG Conc. Temp. N-BZ-Trp 03:24:56 PM -32.225 SR -0.348 589 1.080 20.4 1 2 N-BZ-Trp 03:25:07 PM -32.291 SR -0.349 589 1.080 20.4 3 -0.350 1.080 20.3 N-BZ-Trp 03:25:18 PM -32.363 SR 589 4 03:25:29 PM -32.116SR -0.347 589 1.080 20.3 N-BZ-Trp 5 03:25:40 PM -32.253 SR -0.348 589 1.080 20.2 N-BZ-Trp **IIT Roorkee** HO Thursday, 10-0CT-2024 HO This sample was measured on an Autopol V Plus, Serial #85232 Ĥ Manufactured by Rudolph Research Analytical, Hackettstown, NJ, USA. 2f Set Temperature : 25.0 Temperature Correction : Sucrose Std.Dev. Maximum Minimum n Average 5 -33.410 1.4177 -35.775 -32.324 S.No Sample ID Time Result Scale OR ºArc WLG Conc. Temp. 1 11:00:39 AM -32.999 SR -0.330 589 1.000 25.4 N Bz Tyro 2 11:00:49 AM -33.575 SR -0.336 589 1.000 25.3 N Bz Tyro 3 11:01:00 AM -35.775 SR -0.358 589 1.000 25.3 N Bz Tyro 4 N.Bz.Tyro 11:01:10 AM -32.375 SR -0.324 589 1.000 25.2 5 N.Bz.Tyro 11:01:20 AM -32.324 SR -0.323 589 1.000 25.2

IIT Roorkee									
Thursday, 10-0CT-2024									
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This sample was measured on an Autopol V Plus, Serial #85232									
Manufactured by Rudolph Research Analytical, Hackettstown, NJ, USA. 2g									
Set Temperature : 25.0									
Temperature Correction : Sucrose									
n	Average	Std.Dev.	Maximum		Minimum				
5	-14.501	0.0635	-14.562		-14.410				
Sillo	Sample ID	Time		Result	Scale	OR ºArc	WLG	Conc.	Temp.
1	N_BZ_Meth	10:48:53	3 AM	-14.486	SR	-0.145	589	1.000	25.2
2	N_BZ_Meth	10:49:04	AM 1	-14.562	SR	-0.146	589	1.000	25.2
3	N_BZ_Meth	10:49:14	AM I	-14.562	SR	-0.146	589	1.000	25.2
4	N_BZ_Meth	10:49:24	AM I	-14.487	SR	-0.145	589	1.000	25.1
5	N_BZ_Meth	10:49:34	AM 1	-14.410	SR	-0.144	589	1.000	25.1
III Roorkee									
Saturday, 31-AUG-2024									
Sucuru	uy, 51 Acc 201							N_N	
This sample was measured on an Autopol V Plus, Serial #85232									
Manufactured by Rudolph Research Analytical, Hackettstown, NJ, USA.									
Set Temperature : 25.0									
Temperature Correction : Sucrose									
n	Average	Std.Dev.	Ma	XIMUM	Minimum				
5	-23.740	0.4370	-24.	209	-23.168				
S No	Sample TD	Time		Result	Scale	OR ºArc	WI G	Conc	Temp
1	Me BE7-L-AL	AGLY 12:29:36	PM	-23,168	SR	-0.232	589	1.000	25.1
2	Me BEZ-L-ALA	AGLY 12:29:46	PM	-23.398	SR	-0.234	589	1.000	25.0
3	Me BEZ-L-ALA	AGLY 12:30:01	PM	-23,954	SR	-0.240	589	1.000	25.0
4	Me BEZ-L-ALA	AGLY 12:30:11	PM	-24.209	SR	-0.242	589	1.000	24.9
5	Me BEZ-L-ALA	AGLY 12:30:21	PM	-23.972	SR	-0.240	589	1.000	24.9
1									

11. MS and MS² data of 2a:

