

Supporting Information

Li₂NbHO₂: a new transition-metal oxyhydride with rock-salt-type structure

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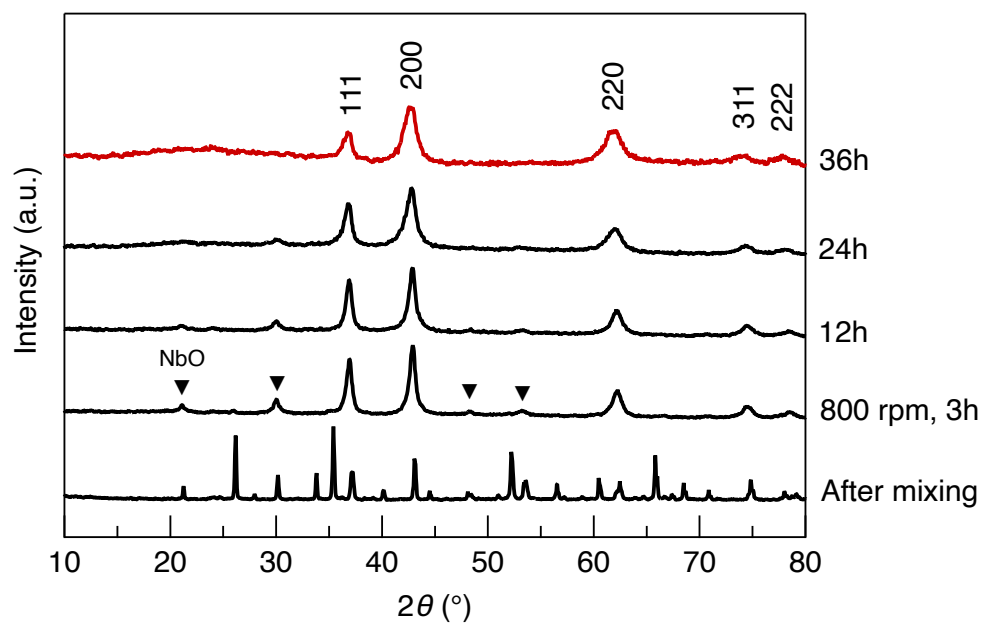


Figure S1. X-ray diffraction patterns of raw materials after hand-mixing and obtained powders by ball milling at 800 rpm for several reaction time.

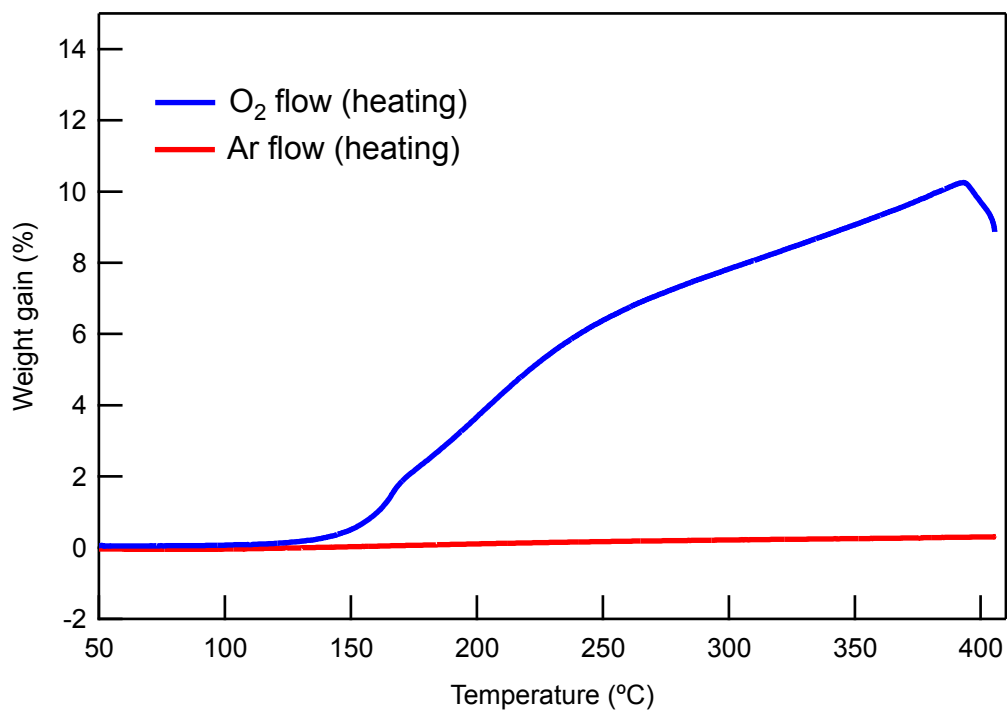


Figure S2. Thermogravimetric data of Li_2NbHO_2 under O_2 and Ar gas flowing at $100 \text{ mL} \cdot \text{min}^{-1}$, using a Rigaku Thermo Plus II apparatus.

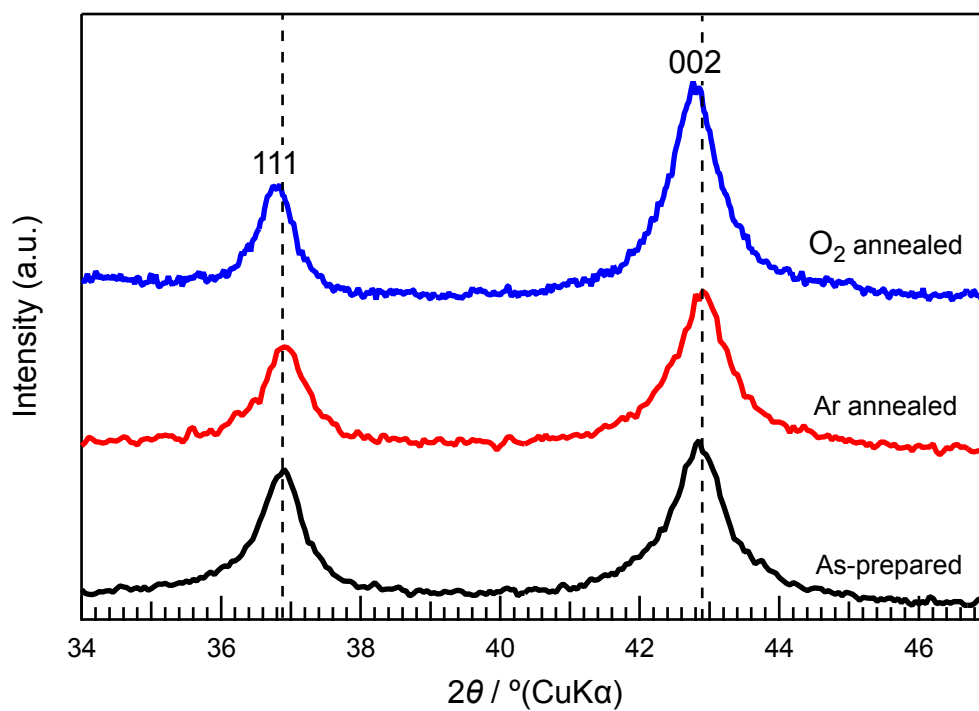


Figure S3. X-ray diffraction patterns of Li_2NbHO_2 before and after TG measurements.

As shown in Fig. S2, a weight gain suggesting ion exchange between H^- and O^{2-} was detected from approximately 125 °C under dry O_2 gas flow, while no significant weight gain was observed under Ar flow. In addition to comparing the XRD patterns before and after the TG measurements (Fig. S3), there was no significant change after the measurement under Ar atmosphere, whereas a significant peak shift to lower angles while maintaining symmetry was observed after the measurement under dry O_2 atmosphere, implying an anion exchange reaction from H^- to O^{2-} . These results indicate that Li_2NbHO_2 is stable at battery operating temperatures and atmospheres.