# Highly Conductive and Homogeneous NiO<sub>x</sub> Nanoparticles for Stable

## and Efficient Flexible Perovskite Solar Cells

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## **Experimental section**

#### Materials

Ammonium persulphate (APS) and Tetrabutylammonium hydroxide (TBAOH) was purchased from Innochem. Nickel nitrate hexahydrate (Ni(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O), Isopropyl alcohol (IPA, 99.7%), anhydrous ethanol (99.5%), toluene and Bathocuproine (BCP, 96%) were obtained from Alfa Aesar. DMF (99.9%), DMSO (99.9%) and chlorobenzene (CB, 99.9%) were purchased from Sigma-Aldrich. Flexible ITO/PET (25-35  $\Omega$ /cm<sup>2</sup>, 125 µm), Lead iodide (PbI<sub>2</sub>, 99.999%), formamidinium iodide (FAI), cesium iodide (CsI), and phenyl-C61-butyric acid methyl ester (PC<sub>61</sub>BM, 99.9%) were bought from Advanced Election Technology. Lead bromide (PbBr<sub>2</sub>, 99.999%), methylammonium bromide (MABr), and PTAA were purchased from Xi'an Yuri Solar Co., Ltd. All starting materials were obtained from commercial suppliers without further purification.

## Synthesis of NiO<sub>x</sub> and NiO<sub>x</sub>-APS nanoparticles (NPs)

The Ni(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O (50 mmol) was dissolved in 50 mL deionized water. Subsequently, 50 mL TBAOH was added drop wisely under vigorous stirring. The resulting precipitates were collected by centrifugation and washed with deionized water 4 times to remove residual TBAOH. The washed precipitates were vacuum-dried at 100°C for 12 h, followed by sintering at 270°C for 2 h to obtain NiO<sub>x</sub> NPs.

The Ni(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O (50 mmol) and APS (0.15 mmol (0.3 mol%)) was dissolved in

50 mL deionized water. Subsequently, 50 mL TBAOH was added drop wisely under vigorous stirring. The resulting precipitates were collected by centrifugation and washed with deionized water 4 times to remove residual TBAOH, APS and its by-products. The washed precipitates were vacuum-dried at 100°C for 12 h, followed by sintering at 270°C for 2 h to obtain NiO<sub>x</sub>-APS NPs.

#### Solar cell fabrication

First, the NiO<sub>x</sub> NPs or NiO<sub>x</sub>-APS NPs were dispersed in deionized water and IPA (4:1) to prepare 5 mg/mL NiO<sub>x</sub> or NiO<sub>x</sub>-APS solution and filtered with a 0.45  $\mu$ m nylon filter. The NiO<sub>x</sub> or NiO<sub>x</sub>-APS solution was spin-coated on the ITO/PET substrates at 4000 rpm for 30 s and annealed on a hotplate at 100°C for 30 min in air. PTAA (0.5 mg/mL in Toluene) was coated on NiO<sub>x</sub> or NiO<sub>x</sub>-APS films at 6000 rpm for 40 s and then annealed at 100°C for 10 min. For deposition of perovskite layer, The (1.4 M) perovskite precursor solution was constructed by mixing PbI<sub>2</sub>, PbBr<sub>2</sub>, FAI, MABr, and CsI in a mixed solvent (DMF/DMSO=4/1) according to a formula of  $Cs_{0.05}(FA_{0.95}MA_{0.05})_{0.95}Pb(I_{0.95}Br_{0.05})_3$ . The precursor solution was dripped onto HTLcoated substrates and spin-coated at 5000 rpm for 30 s, 150 µL CB as anti-solvent was dripped on the film at 18 s from the start and then annealed at 100°C for 30 min. After that, the PC<sub>61</sub>BM (20 mg/mL in CB) was coated on the perovskite surface at 5000 rpm for 30 s and annealed at 100°C for 10 min, then the BCP solution (0.5 mg/mL in IPA) was spin-coated at 4000 rpm for the 30 s on the PCBM layer. Finally, 200 nm thickness of Ag electrode was deposited by thermal evaporation as an electrode using a shadow mask.

#### **Characterizations**

The TEM images were taken on JEM-2100 (HR), JEOL Co.Ltd. The XRD patterns were obtained with Smartlab 9kW X-ray diffractometer and with Cu K $\alpha$  ( $\lambda$  = 1.5406 Å) radiation source. The Raman spectrum was prepared by Raman spectrometer (New Mills, UK). The XPS and UPS characterizations were performed on a Thermo Fisher ESCALAB 250XI (USA), with -5 V bias for UPS measurement. The atomic force microscopy (AFM, Bruker Dimension ICON) instrumented c-AFM and KPFM of NiO<sub>x</sub> films. The morphology of samples was acquired by field-emission scanning electron

microscope (FE-SEM, Hitachi SU8010). The UV-Vis absorption spectra were obtained from a UV-2600 spectrophotometer (Shimadzu, Japan). The contact angle of the surface was measured by a video-based automatic contact angle measuring instrument (OCA15, Data Physics). The electrochemical impedance spectra (EIS) were recorded by electrochemical workstation (Zahner, Germany). The current density-voltage (*J-V*) curves of the perovskite solar cells were obtained by a Keithley 2400 source under AM 1.5 G simulated irradiation (100 mW cm<sup>-2</sup>) from a solar simulator (Newport 94023A) and then obtain the indoor *J-V* curve by altering the illuminance to 1000, 500 and 200 lux using the illuminance meter (ENLITECH) with fluorescent lamps (ILS-30).



Figure S1. *J-V* curves of F-PSCs based on different concentrations of APS assisted synthesized NiO<sub>x</sub>-APS.



Figure S2. The statistical crystal grain size distribution of  $NiO_x$  (a) and  $NiO_x$ -APS (b).



Figure S3. XRD patterns (a) and Raman spectrum (b) of the  $NiO_x$  and  $NiO_x$ -APS NPs.



Figure S4. Photographs of Ni(OH)<sub>2</sub> (a) and Ni(OH)<sub>2</sub>-APS (b).



Figure S5. XPS S 2P spectra of Ni(OH)<sub>2</sub>-APS NPs.



Figure S6. SEM images of  $NiO_x$  (a) and  $NiO_x$ -APS (b) films.



Figure S7. XPS O 1s spectra of NiO<sub>x</sub> and NiO<sub>x</sub>-APS films.



Figure S8. a) UV-vis absorption spectra of  $NiO_x$  and  $NiO_x$ -APS films. b) Tauc plot showing an optical bandgap of  $NiO_x$  and  $NiO_x$ -APS films.



Figure S9. Photovoltaic parameters statistics distribution of  $V_{OC}$ ,  $J_{SC}$ , PCE, and FF for the F-PSCs based on NiO<sub>x</sub> and NiO<sub>x</sub>-APS HTLs.



Figure S10. The contact angle of perovskite precursor solution on the surface of  $NiO_x$  and  $NiO_x$ -APS films.



Figure S11. Top-view and cross-section SEM images of the perovskite films deposited on  $NiO_x$  and  $NiO_x$ -APS films. (Due to the difficulty in preparing cross-sectional samples for flexible films, the cross-sectional sample was prepared on an ITO/glass substrate.)



Figure S12. Nyquist plots of the F-PSCs based on  $NiO_x$  and  $NiO_x$ -APS HTLs, measured with a bias voltage of 1 V in dark condition. Inset: equivalent circuit model.



Figure S13. The spectra of 3000 K LED light at 1000 lux, 500 lux, and 200 lux illuminances.

Table S1 Summary of device performance for F-PSCs based on different concentrations of APS assisted synthesized NiO<sub>x</sub>-APS.

Device	$V_{\rm OC}({ m V})$	$J_{\rm SC}({\rm mA/cm^2})$	PCE(%)	FF(%)
NiO <sub>x</sub>	1.110	23.02	19.63	76.82
NiO <sub>x</sub> -APS-1%	1.119	23.35	20.86	79.83
NiO <sub>x</sub> -APS-3%	1.149	23.70	22.68	83.28
NiO <sub>x</sub> -APS-5%	1.136	23.59	21.76	81.19

Table S2 Photovoltaic parameters for champion F-PSCs based on  $NiO_x$  and  $NiO_x$ -APS HTLs.

Device	$V_{\rm OC}({ m V})$	$J_{\rm SC}({\rm mA/cm^2})$	PCE(%)	FF(%)
NiO <sub>x</sub> _RS	1.110	23.02	19.63	76.82
NiO <sub>x</sub> _FS	1.090	23.02	18.54	73.89
NiO <sub>x</sub> -APS_RS	1.149	23.70	22.68	83.28
NiO <sub>x</sub> -APS_FS	1.149	23.70	22.54	82.77

Device architecture		$[\text{mA cm}^{-2}]$	PCE [%]	FF [%]	Ref.
$\frac{PEN/ITO/NiO_{x}/Cs_{0.05}(MA_{0.15}FA_{0.85})_{0.95}Pb(Br_{0.1})_{5I_{0.85}}}{_{5}I_{0.85}}_{3}/PC_{61}BM/BCP/Ag}$	1.11	23.50	21.08	80.76	1
$\frac{PET/ITO/NiO_{x}/Cs_{0.05}MA_{0.05}\ FA_{0.9}PbI_{0.95}Br_{0.05}}{/PC_{61}BM/BCP/Ag}$	1.12	21.79	19.17	79	2
$\label{eq:PET/ITO/NiO_x/Poly(LA)/Cs_{0.05}(FA_{0.92}MA_{0.08})_{0.95}Pb(I_{0.92}Br_{0.08})_3/PC_{61}BM/BCP/Ag$	1.06	23.78	19.03	76	3
$\begin{array}{l} PET/ITO/NiO_{x}/Cs_{0.05}(FA_{0.77}MA_{0.23})_{0.95}Pb(I_{0.77}B\\ r_{0.23})_{3}/PC_{61}BM/BCP/Ag \end{array}$	1.230	20.90	21.02	81.76	4
$\begin{array}{l} PET/ITO/NiO_x/PTAA/Cs_{0.05}(FA_{0.95}MA_{0.05})_{0.95}P\\ b(I_{0.95}Br_{0.05})_3/PC_{61}BM/BCP/Ag \end{array}$	1.172	23.89	23.72	84.64	5
CPI/ITO/Sp-NiO <sub>x</sub> /MeO- 2PACz/Cs <sub>0.04</sub> (FA <sub>0.83</sub> MA <sub>0.17</sub> ) <sub>0.96</sub> Pb(I <sub>0.83</sub> Br <sub>0.17</sub> ) <sub>3</sub> /C <sub>60</sub> /Cr/Cu	1.07	19.60	15.52	71.96	6
PEN/ITO/NiO <sub>x</sub> /2PACz&MeO-2PACz /(CsPbI <sub>3</sub> ) <sub>0.13</sub> (FAPbI <sub>3</sub> ) <sub>0.85</sub> (MAPbBr <sub>3</sub> ) <sub>0.02</sub> /C <sub>60</sub> /BC P/Au	1.153	23.27	21.96	81.83	7
$\label{eq:petriform} \begin{split} PET/ITO/NiO_x/Cs_{0.05}(FA_{0.92}MA_{0.08})_{0.95}Pb(I_{0.92}B\\ r_{0.08})_3/PC_{61}BM/BCP/Ag \end{split}$	1.097	22.62	20.42	82.33	8
$\begin{array}{l} PET/ITO/NiO_{x}/PTAA/Cs_{0.05}(FA_{0.95}MA_{0.05})_{0.95}P\\ b(I_{0.95}Br_{0.05})_{3}/PC_{61}BM/BCP/Ag \end{array}$	1.149	23.70	22.68	83.28	This work

# Table S3 Summary of F-PSCs with $NiO_x$ HTMs in recent years.

Table S4 Statistics of performance parameters of champion devices based on NiO<sub>x</sub>-APS HTLs at different light intensities.

Device	Light	Pin	$V_{\rm OC}({\rm V})$	$J_{sc}(mA/cm^2)$	PCE(%)	FF(%)
	source	$(\mu W/cm^2)$	100(1)	• 30( )		
	1000 lux	312	0.926	0.1472	35.59	81.46
Target	500 lux	158	0.858	0.0755	32.92	80.18
	200 lux	62	0.812	0.0284	28.28	76.02

## **Supplementary References**

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