

Host compounds based on the rigid 9,10-dihydro-9,10-ethanoanthracene framework: selectivity behaviour in mixed isomeric dichlorobenzenes

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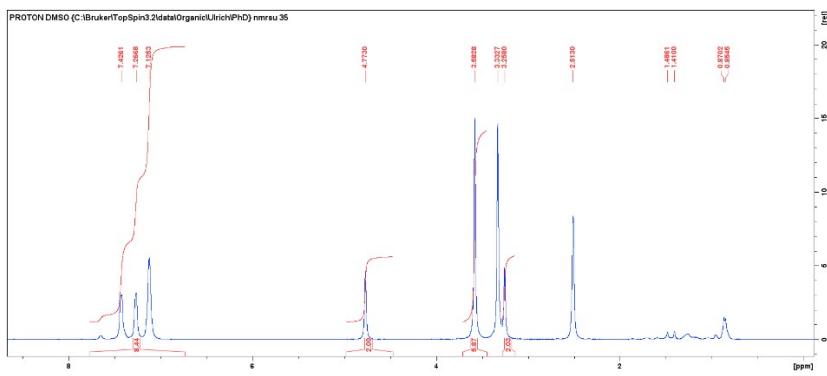
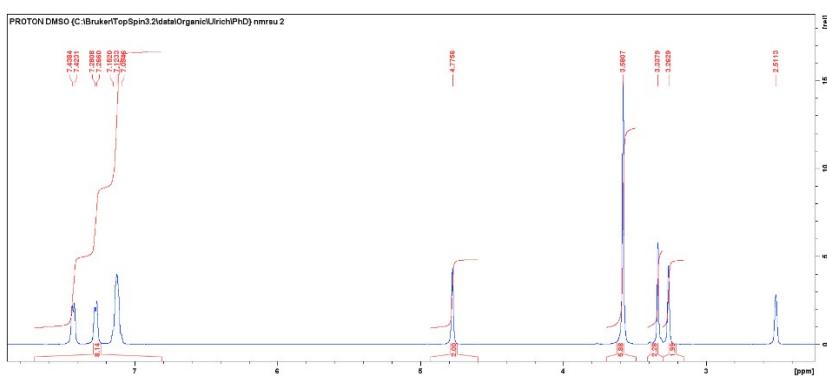
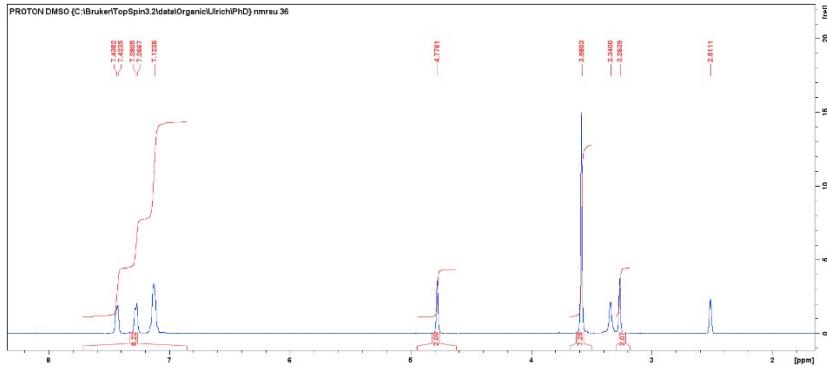
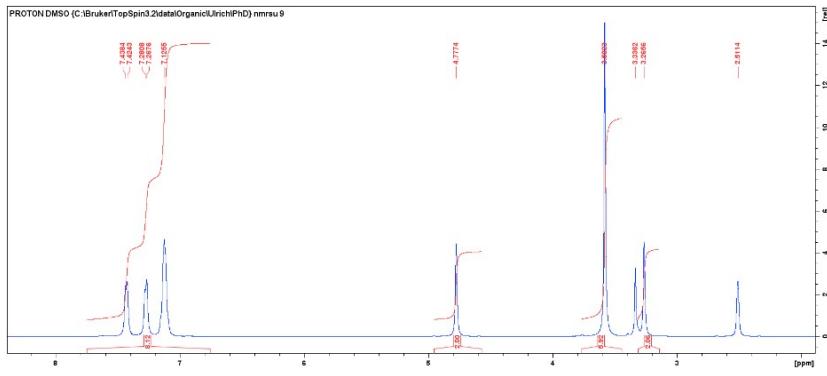
a)**b)**

Figure S1. ^1H -NMR spectra of the duplicated single solvent experiments of *o*-DCB (a) and *m*-DCB (b) with H_1 2

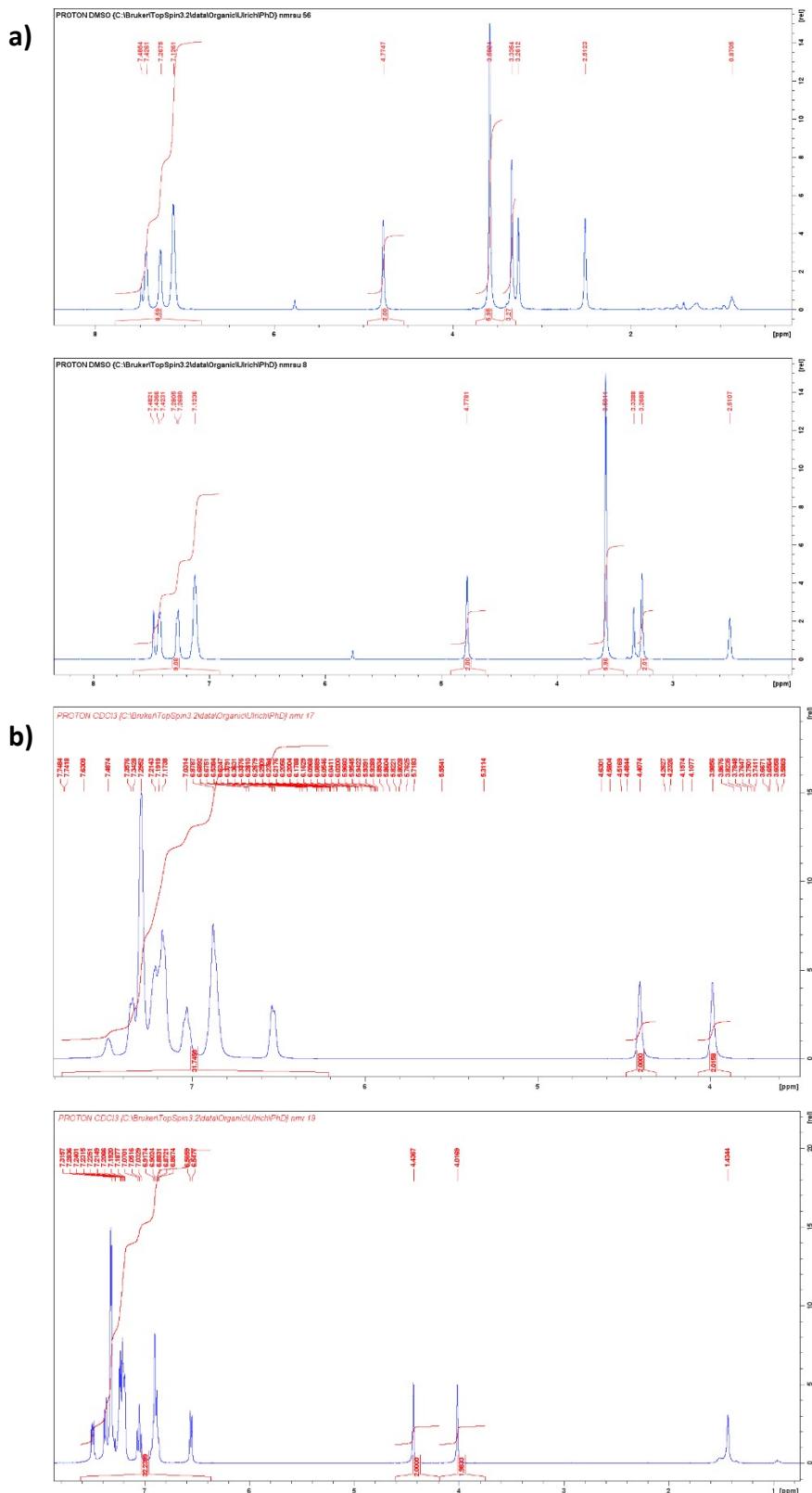


Figure S2. ^1H -NMR spectra of the duplicated single solvent experiments of *p*-DCB (a) with H1 and *o*-DCB (b) with H2

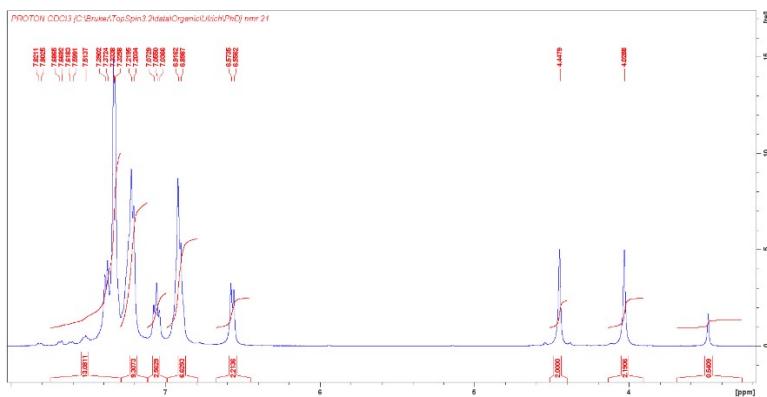
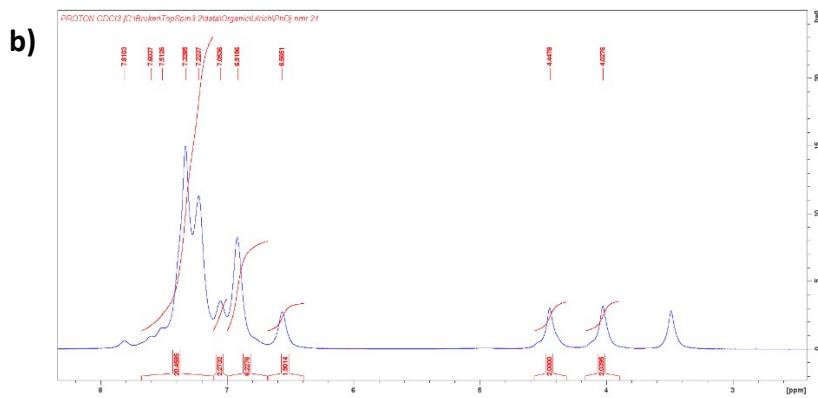
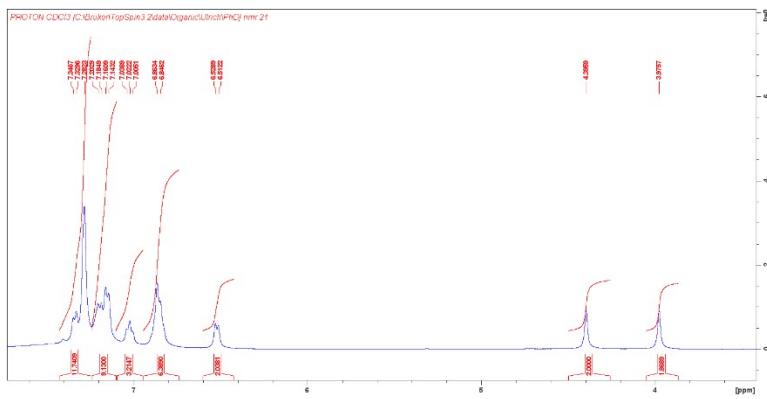
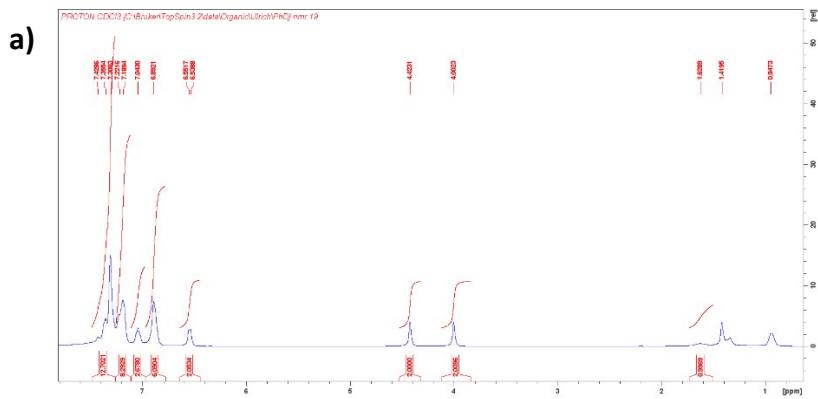


Figure S3. $^1\text{H-NMR}$ spectra of the duplicated single solvent experiments of *m*-DCB (a) and *p*-DCB
(b) with H2

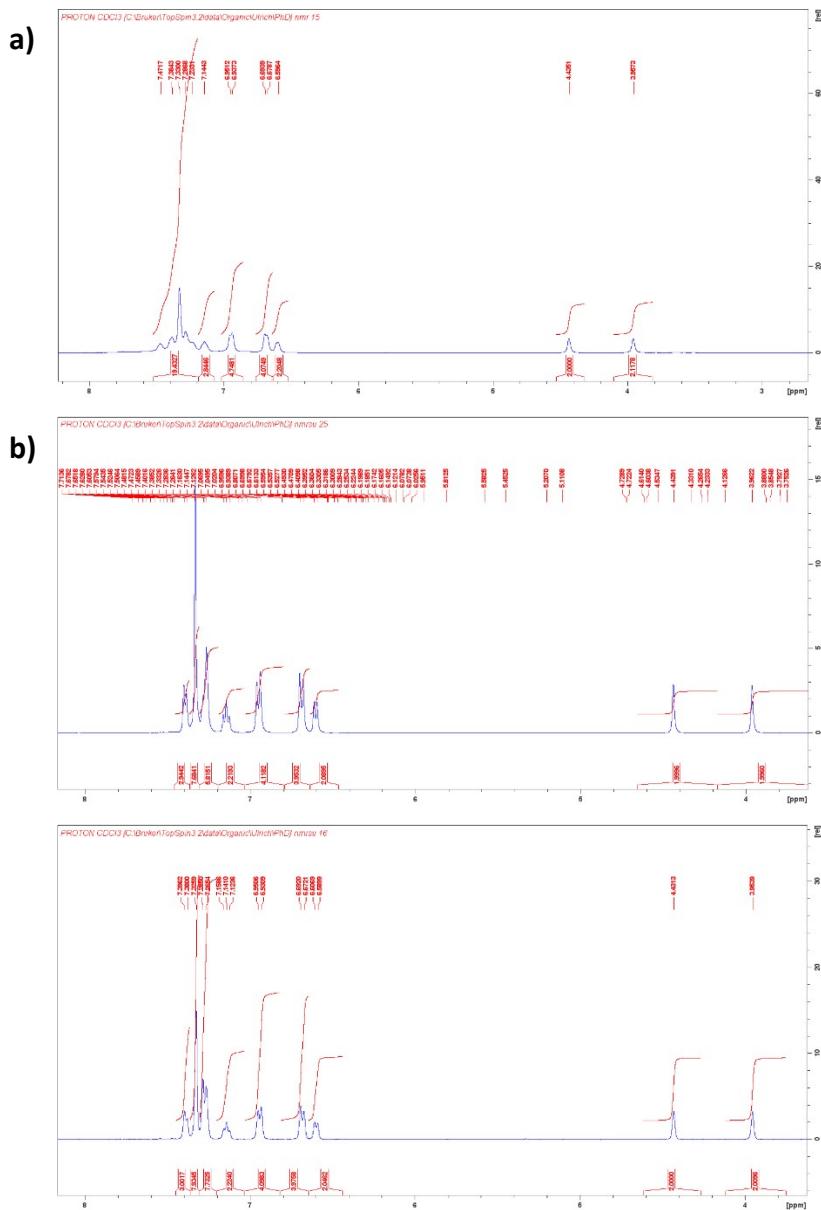


Figure S4. ¹H-NMR spectra of the duplicated single solvent experiments of *o*-DCB (a) and *m*-DCB (b) with H3

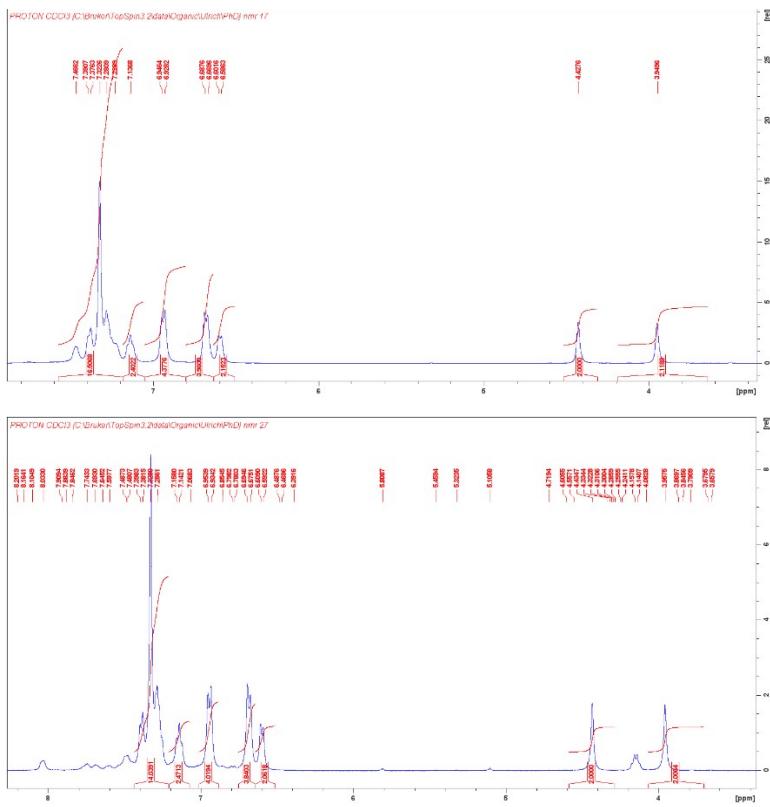


Figure S5. ¹H-NMR spectra of the duplicated single solvent experiment of *p*-DCB with H3

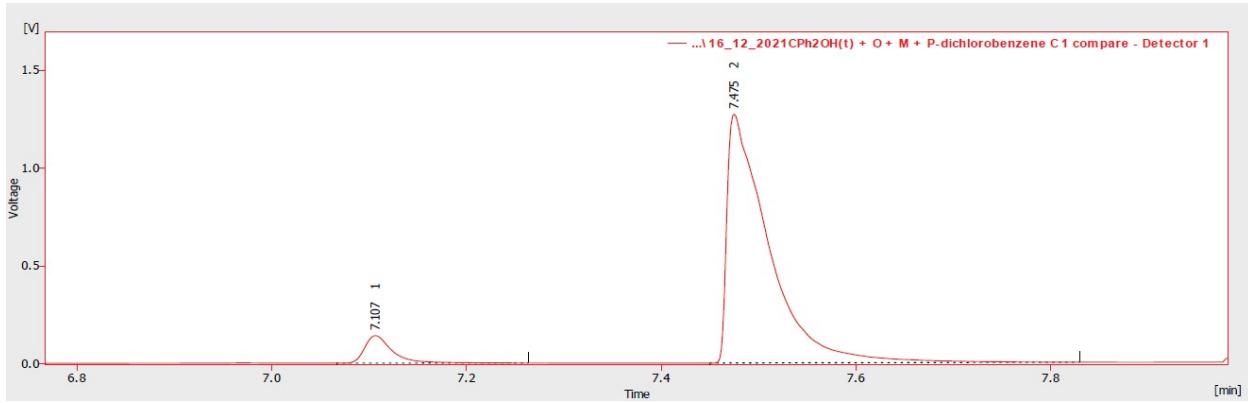
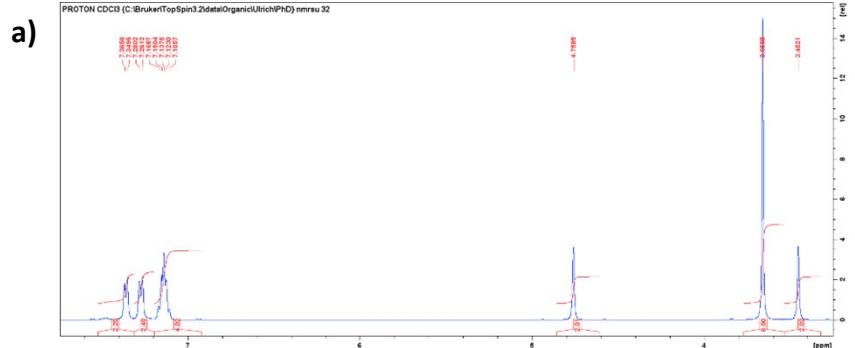
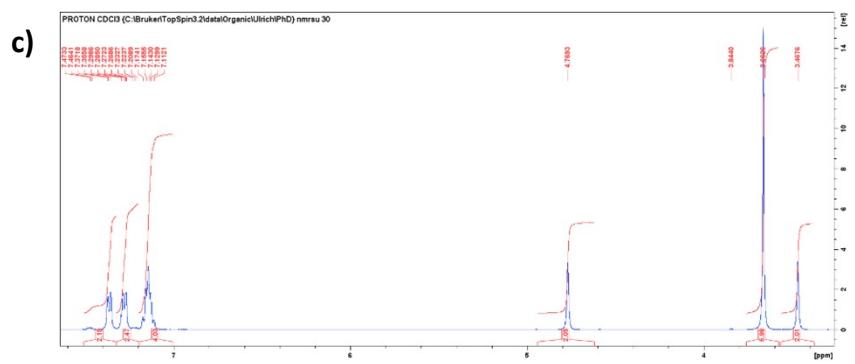
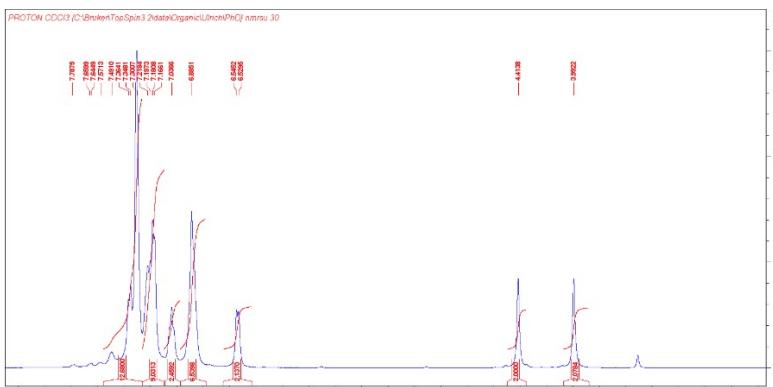
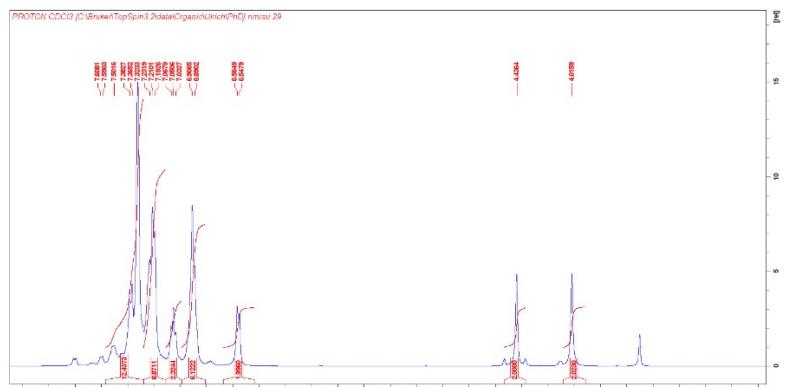
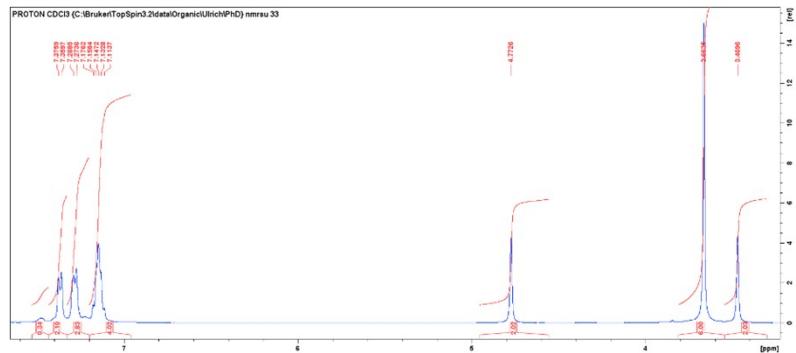
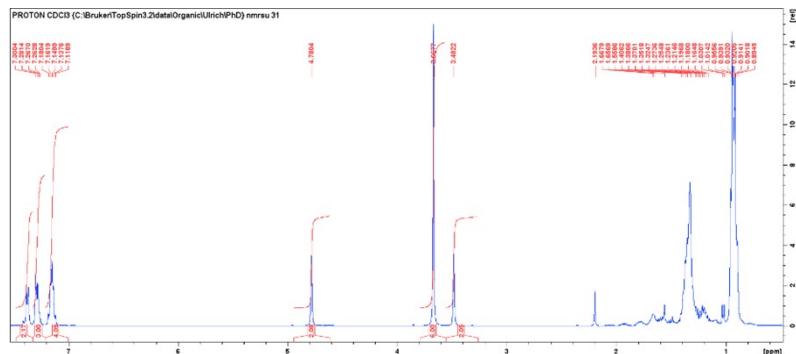


Figure S6. *o*-DCB/*m*-DCB/*p*-DCB ternary mixture of the three dichlorobenzene isomers; eluting in the order of *m*-DCB, *o*-DCB and *p*-DCB (the latter two peak overlapped)



b)





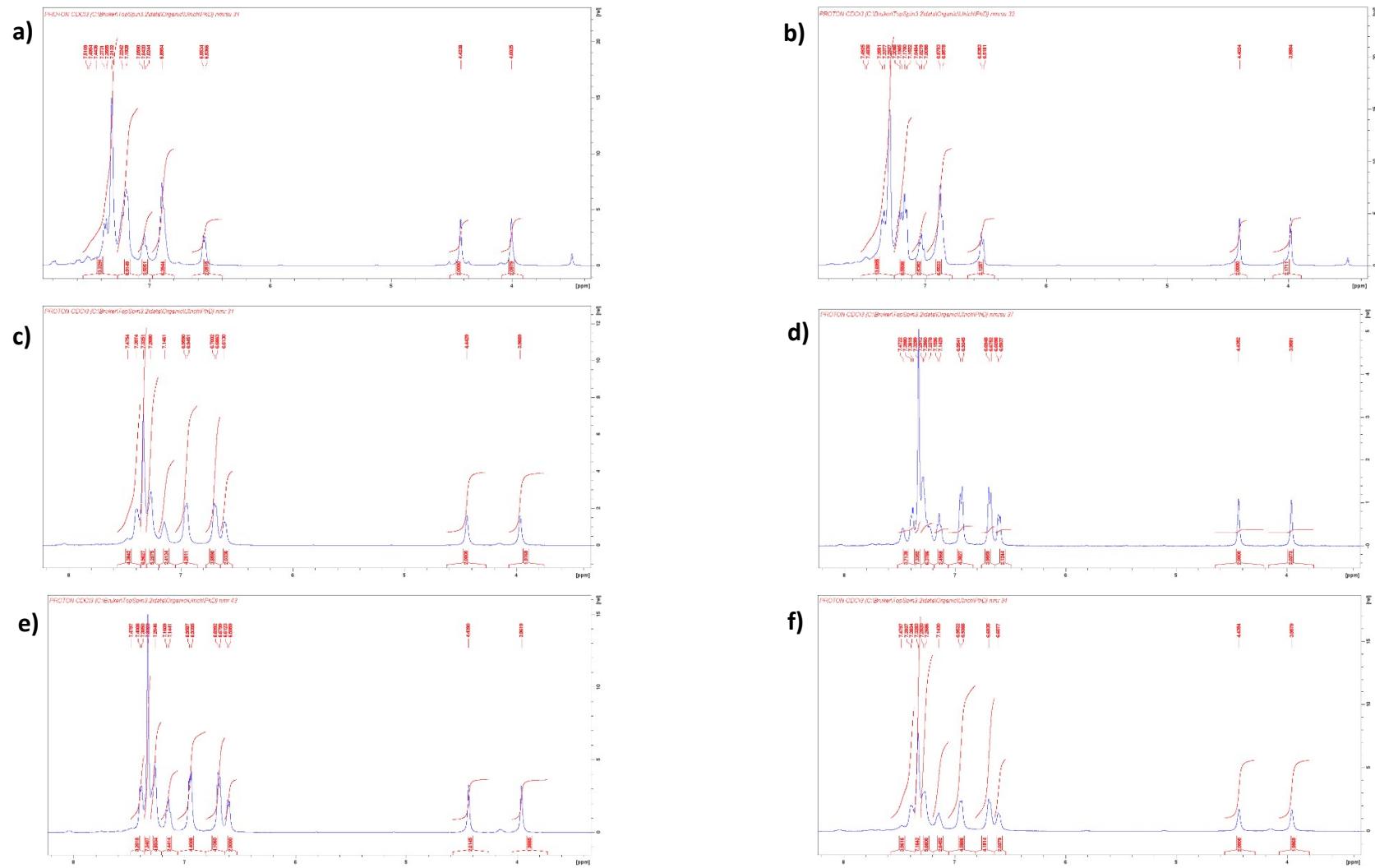


Figure S8. Overall H:G ratios for the *m*-DCB/*p*-DCB (a), *o*-DCB/*m*-DCB/*p*-DCB (b) with H2, *o*-DCB/*m*-DCB (c), *o*-DCB/*p*-DCB (d), *m*-DCB/*p*-DCB (e), *o*-DCB/*m*-DCB/*p*-DCB (f) competition experiments with H3

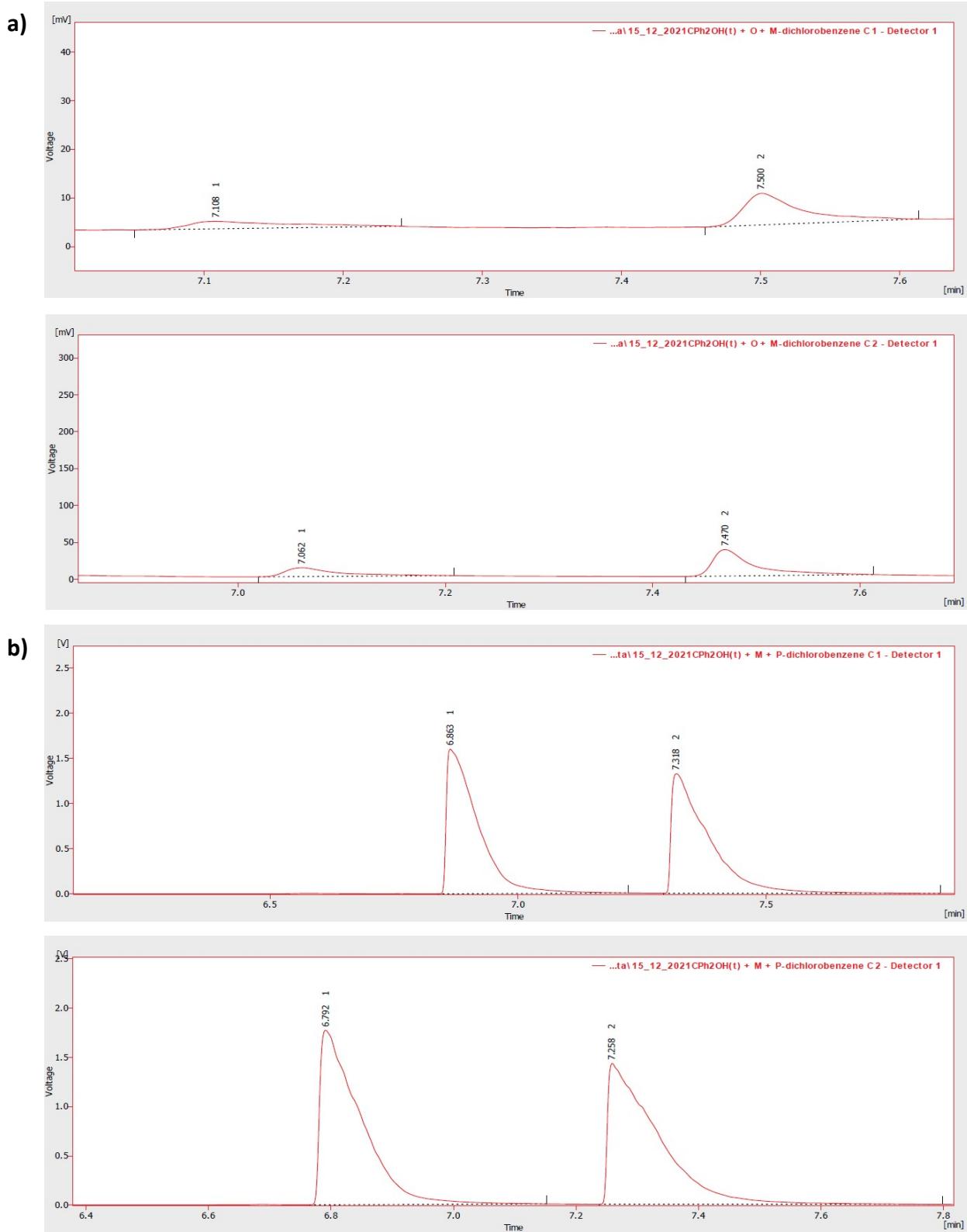
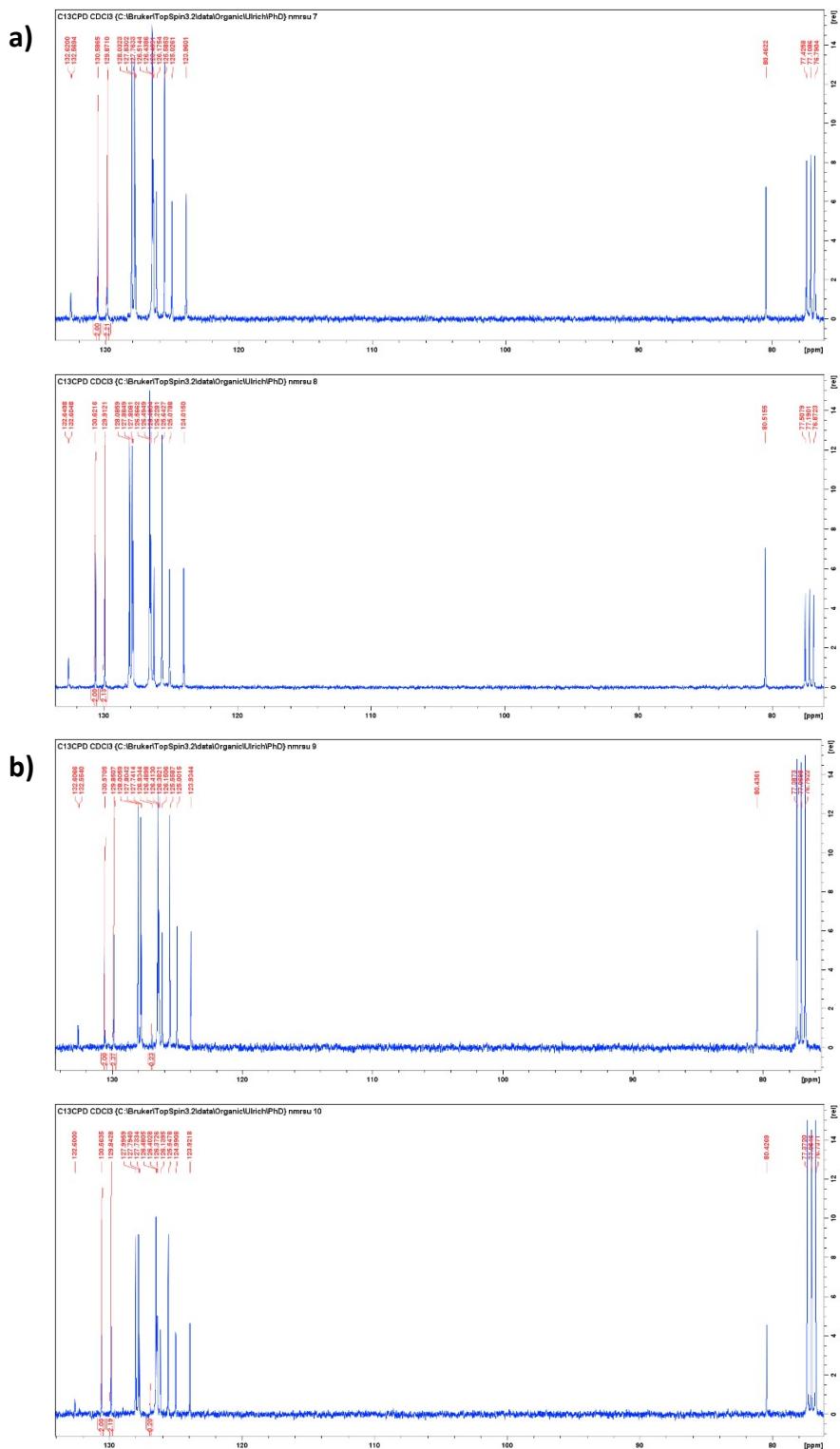


Figure S9. GC-FID chromatograms of the *o*-DCB/*m*-DCB (a) and *m*-DCB/*p*-DCB (b) binary guest mixtures with H2



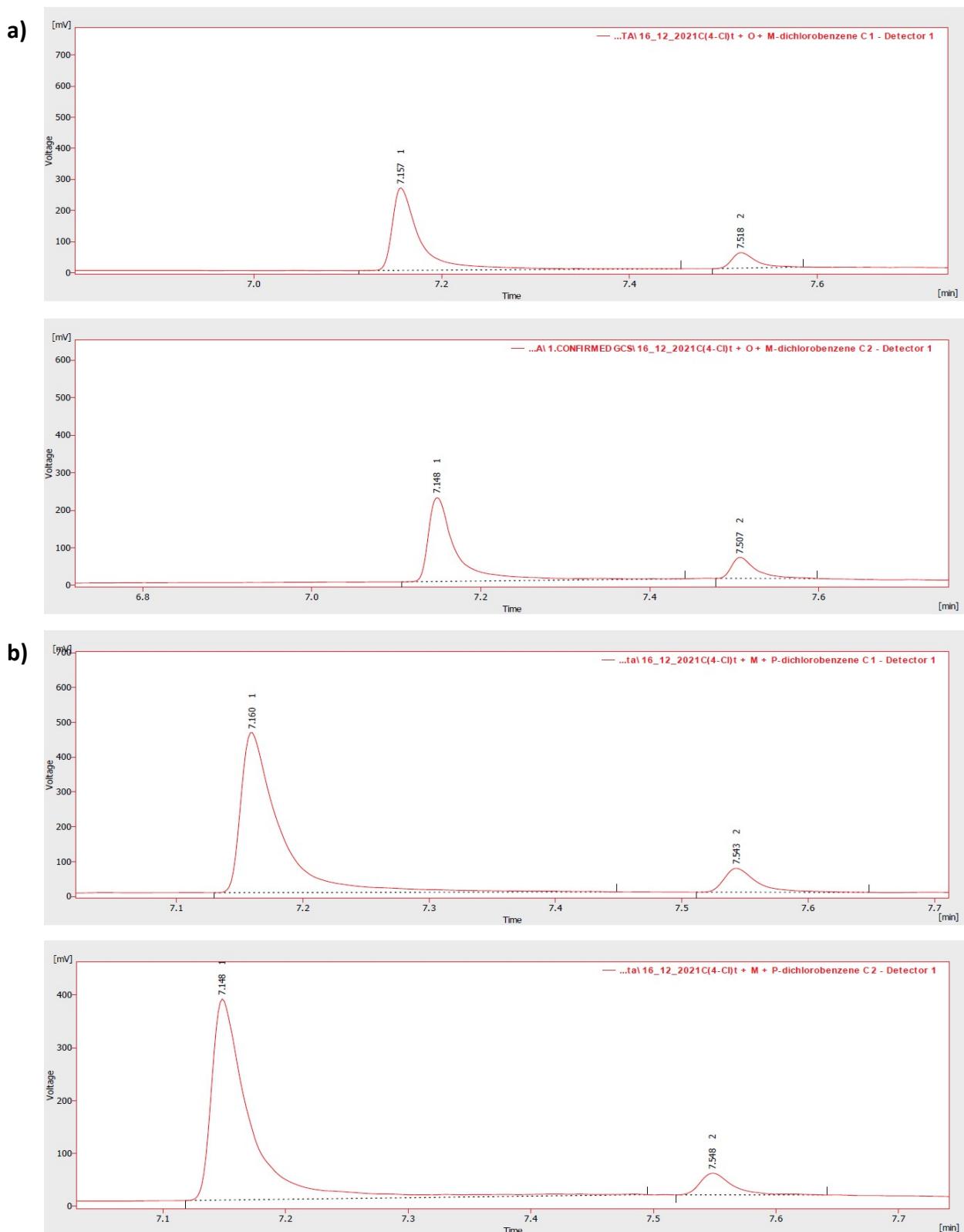


Figure S11. GC-FID chromatograms of the *o*-DCB/*m*-DCB (a) and *m*-DCB/*p*-DCB (b) binary guest mixtures with H3

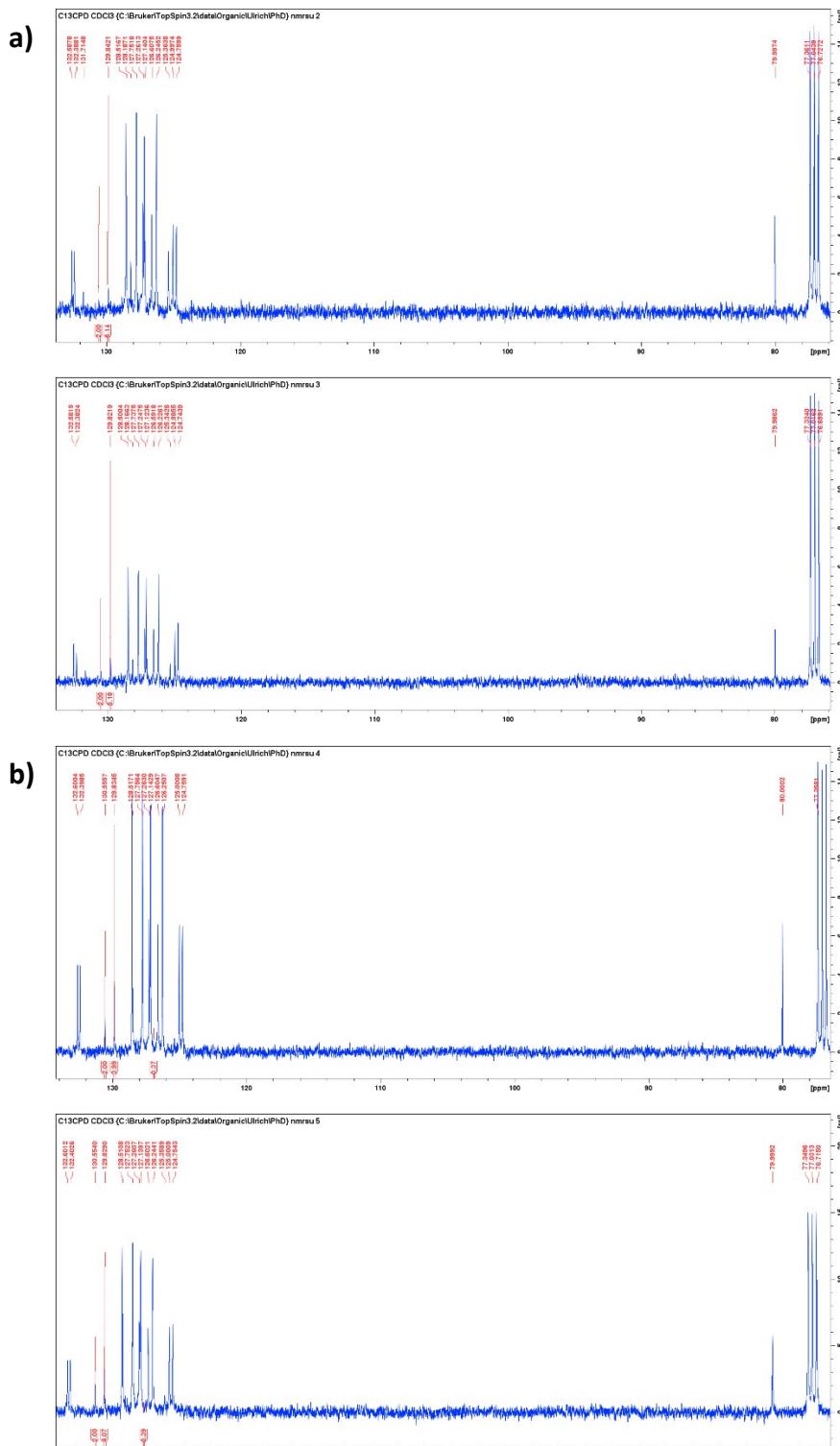


Figure S12. ^{13}C -NMR spectra of the *o*-DCB/*p*-DCB (a) binary and *o*-DCB/*m*-DCB/*p*-DCB ternary guest mixtures with H3

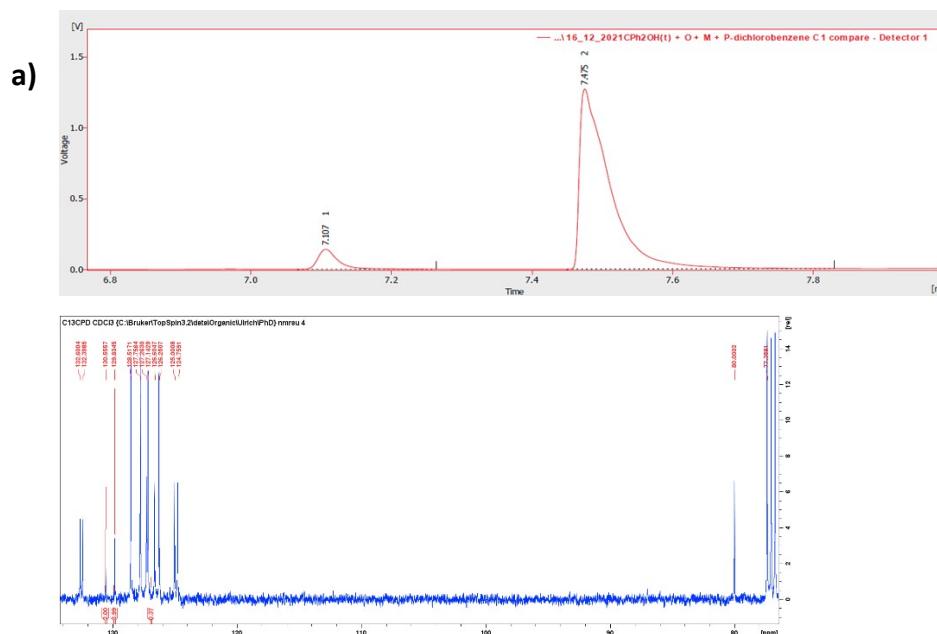


Figure S13. GC-FID chromatogram (a) and ^{13}C -NMR spectrum (b) of the *o*-DCB/*m*-DCB/*p*-DCB ternary guest mixtures with H2; showing the close agreement of the obtained G:G ratios [(6.8% *m*-DCB: 93.2% *o*-DCB/*p*-DCB from GC-FID) and (6.7% *m*-DCB: 58.6% *o*-DCB: 34.7% *p*-DCB from ^{13}C -NMR spectroscopy)]

Table S1. Duplicate values obtained for the binary and ternary mixtures for H2 and H3, and toluene, ethylbenzene and cumene guests (H1 experiments resulted in the obtainment of vial containing only the apohost)

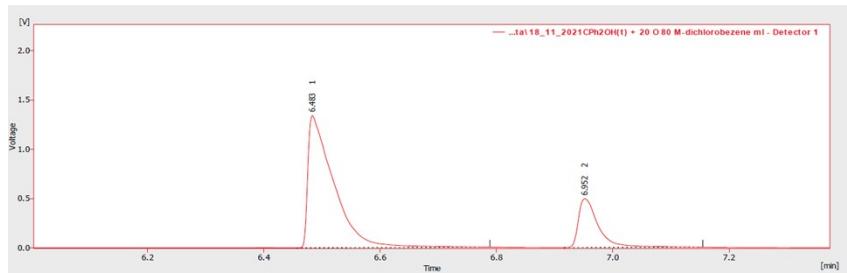
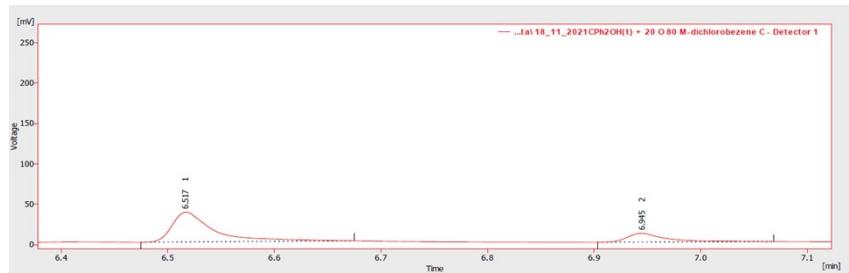
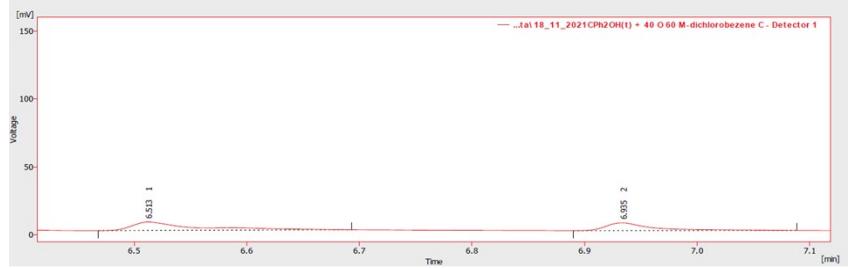
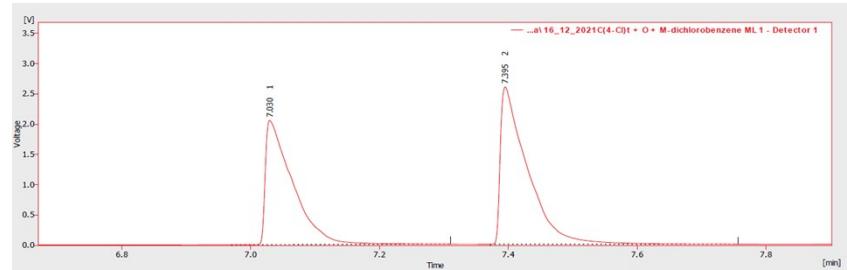
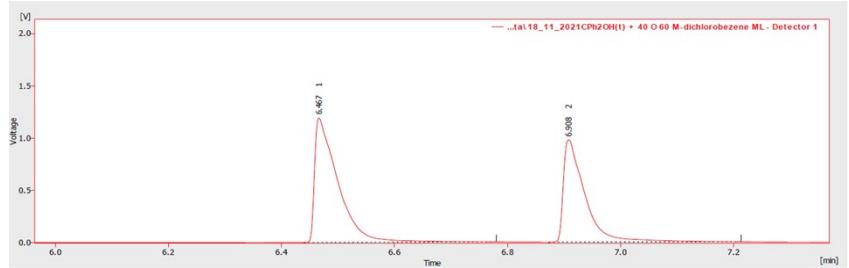
			H2			H3		
<i>o</i> -DCB	<i>m</i> -DCB	<i>p</i> -DCB	Duplicate #1	Duplicate #2	e.s.d.s (%)	Duplicate #1	Duplicate #2	e.s.d.s (%)
X	X^b		71.1:28.9	70.0:30.0	(0.6)(0.6)	13.8:86.2	17.7:82.3	(2.0)(2.0)
X		X^c	64.4:35.6	65.3:34.7	(0.5)(0.5)	43.8:56.2	43.6:56.4	(0.1)(0.1)

X	X^b	49.3:50.7	49.7:50.3	(0.2)(0.2)	88.8:11.2	91.7:8.3	(1.5)(1.5)
X	X^c	58.6:6.7:34.7	60.7:6.1:33.2	(1.1)(0.3)(0.8)	45.8:45.7:8.5	45.2:46.0:8.8	(0.3)(0.1)(0.2)

^aApohost was obtained, the guests were not included

^bThe *o*-DCB/*m*-DCB and *m*-DCB/*p*-DCB experiments were analyzed using GC-FID

^cThe *o*-DCB/*p*-DCB and *o*-DCB/*m*-DCB/*p*-DCB experiments were analyzed using ¹³C-NMR spectroscopy

a)**b)****c)****d)**

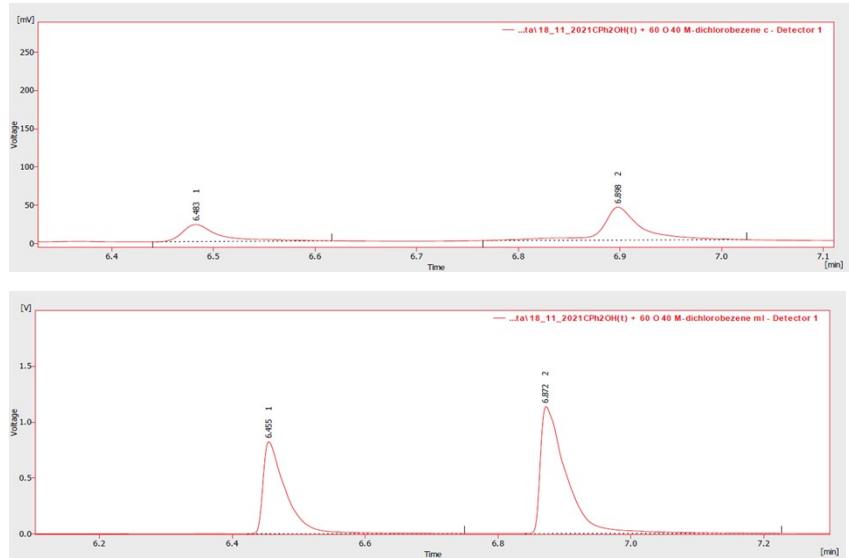
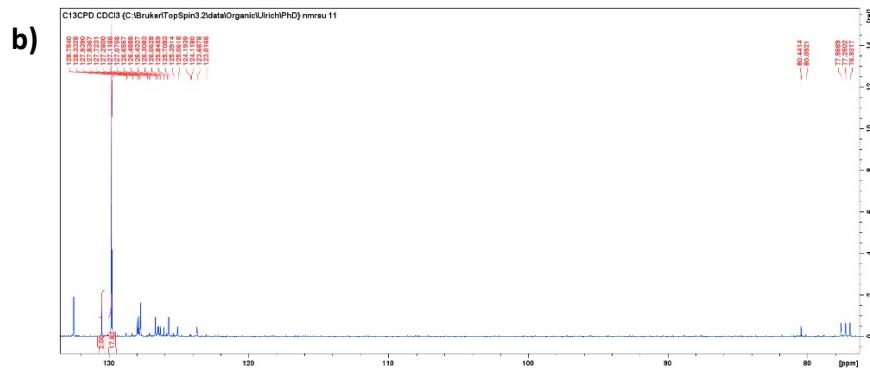
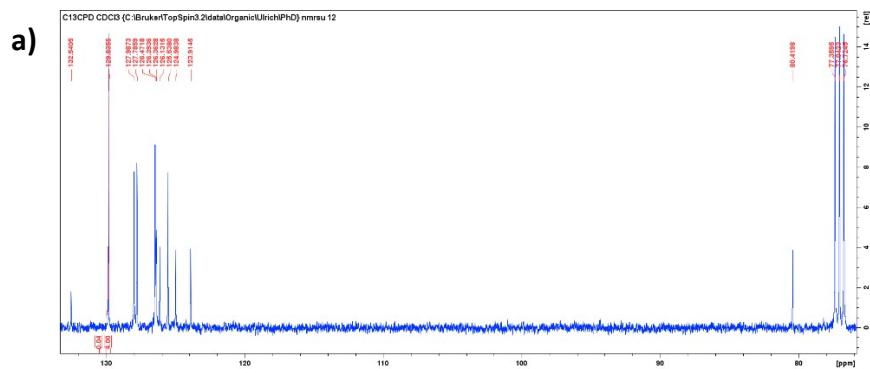
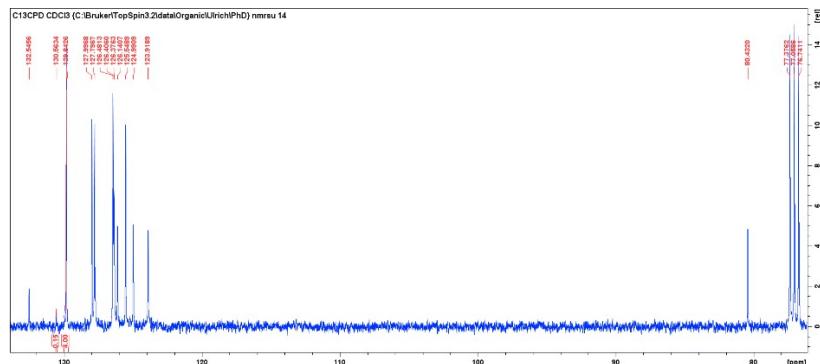
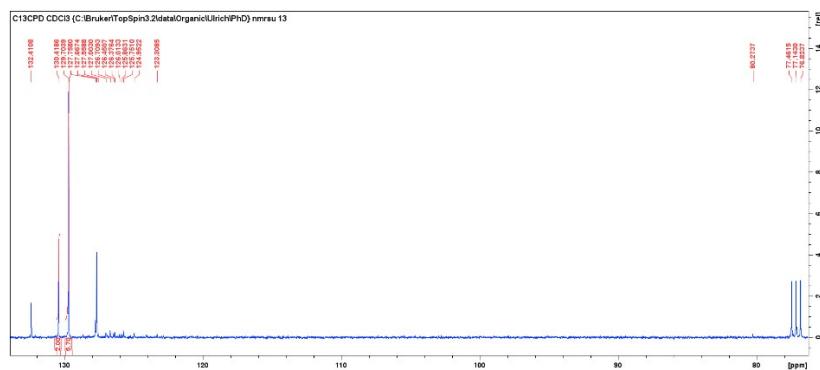


Figure S14. GC-FID chromatograms of the crystals (left) and mother liquor (right) for the 20:80 (a), 40:60 (b), 50:50 (c), and 60:40 (d) *o*-DCB/*m*-DCB binary experiments with H2





c)



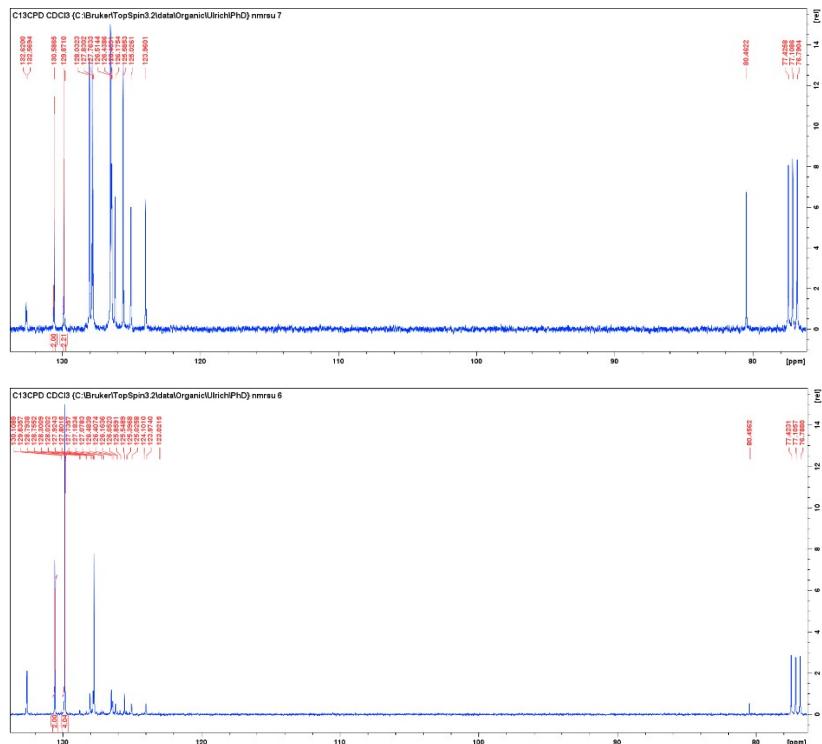
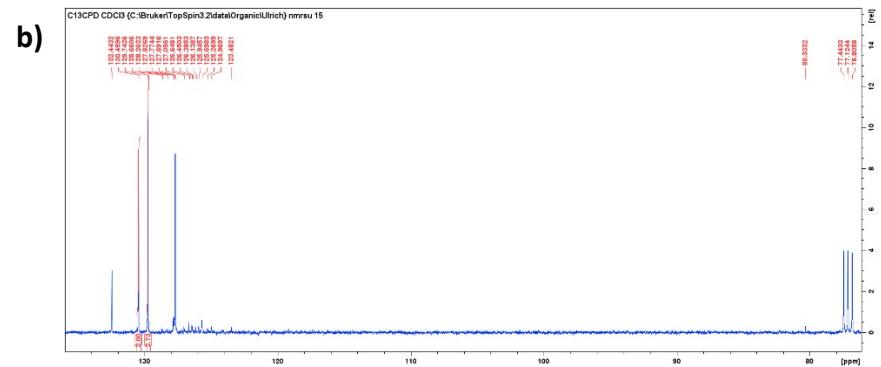
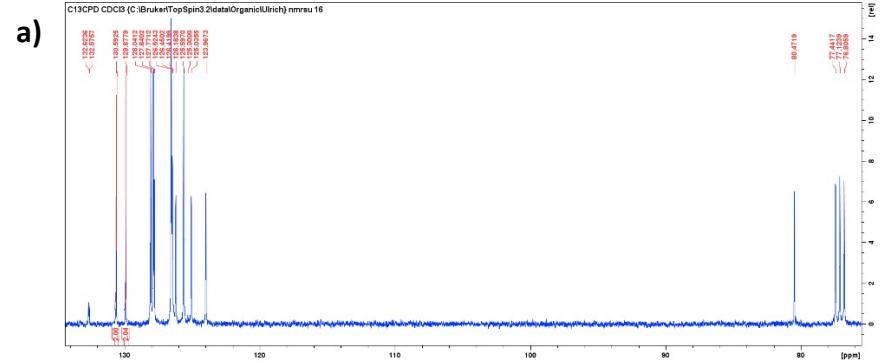


Figure S15. ^{13}C -NMR spectra of the crystals (left) and mother liquor (right) for the 20:80 (a), 40:60 (b), and 50:50 (c) *o*-DCB/*p*-DCB binary experiments with H2



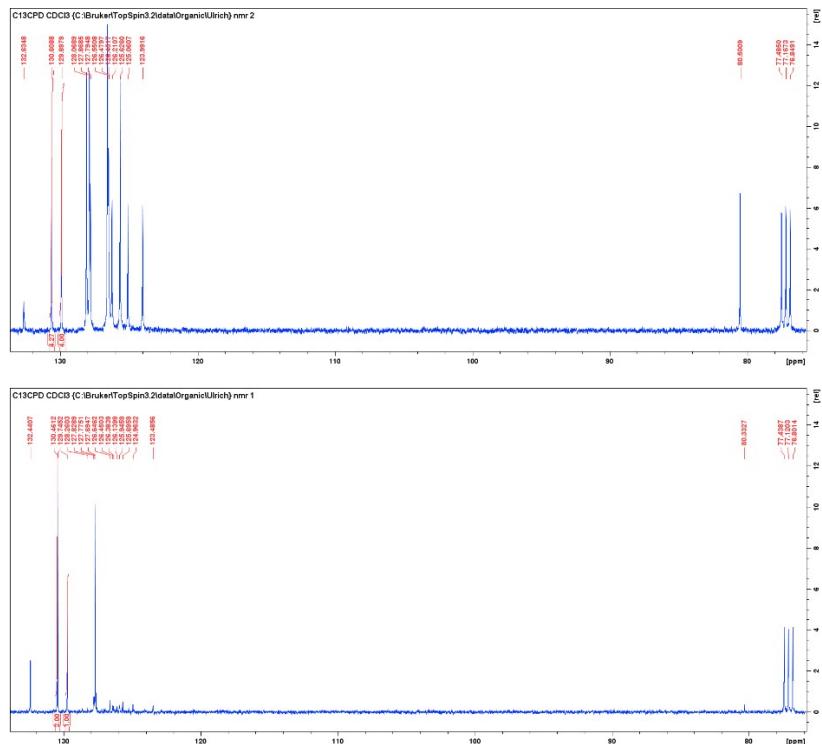
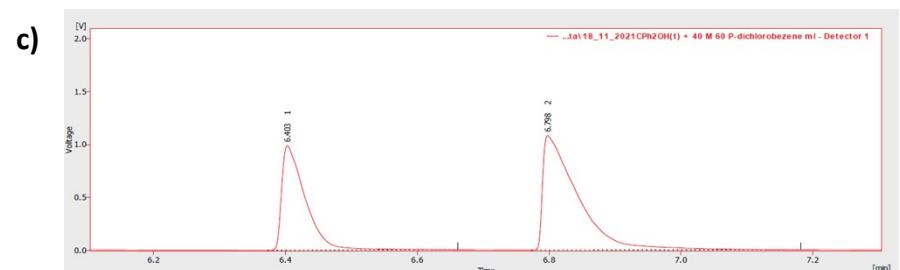
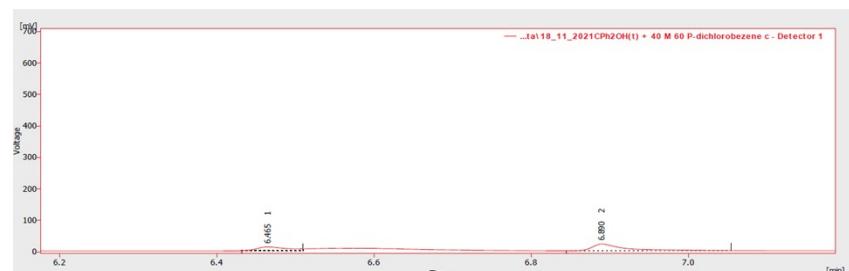
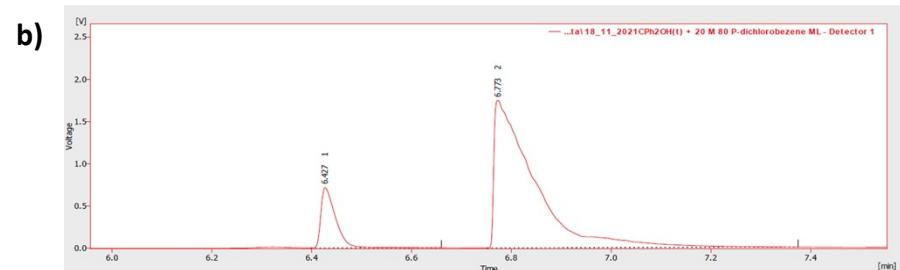
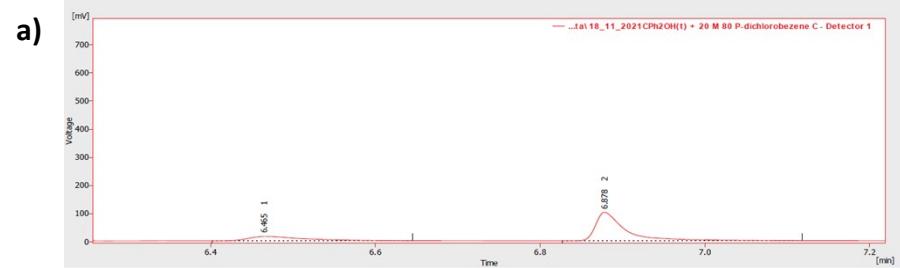


Figure S16. ¹³C-NMR spectra of the crystals (left) and mother liquor (right) for the 60:40 (a) and 80:20 (b) *o*-DCB/*p*-DCB binary experiments with H2



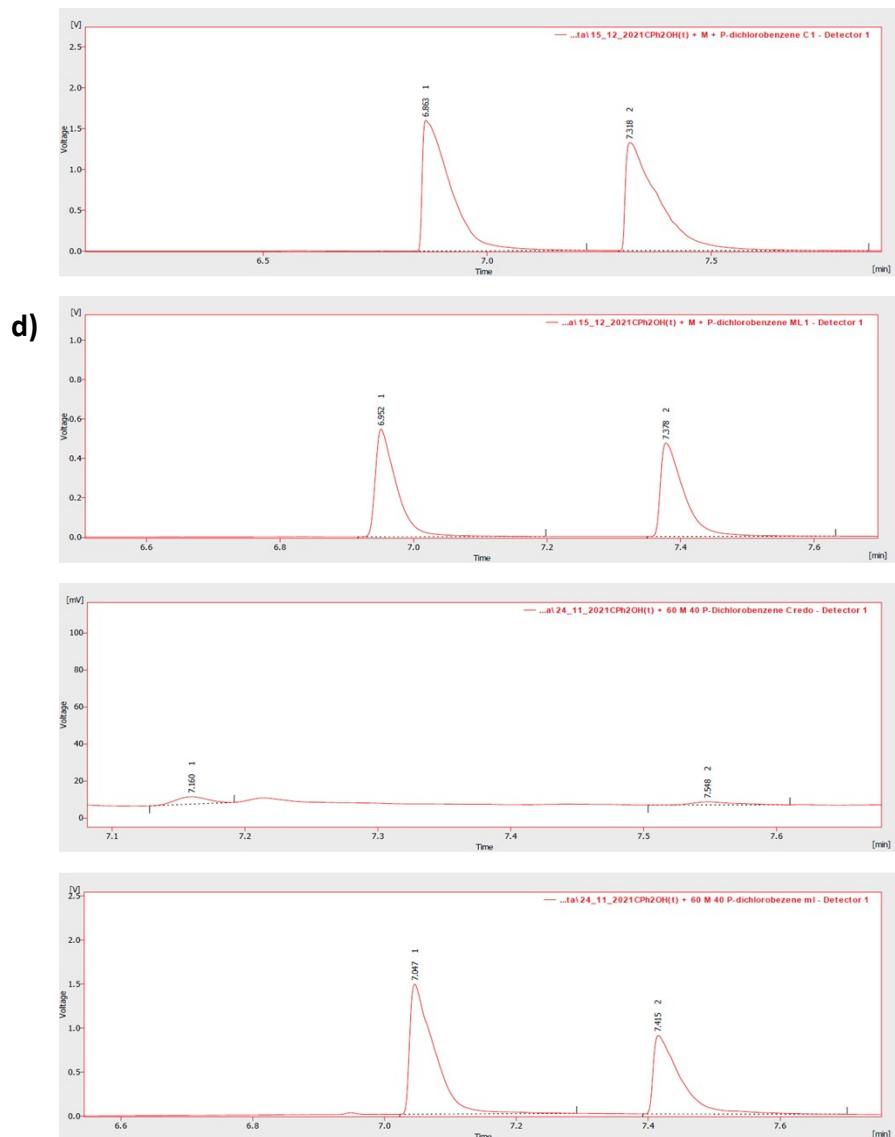


Figure S17. GC-FID chromatograms of the crystals (left) and mother liquor (right) for the 20:80 (a), 40:60 (b), 50:50 (c), and 60:40 (d)
m-DCB/*p*-DCB binary experiments with H2

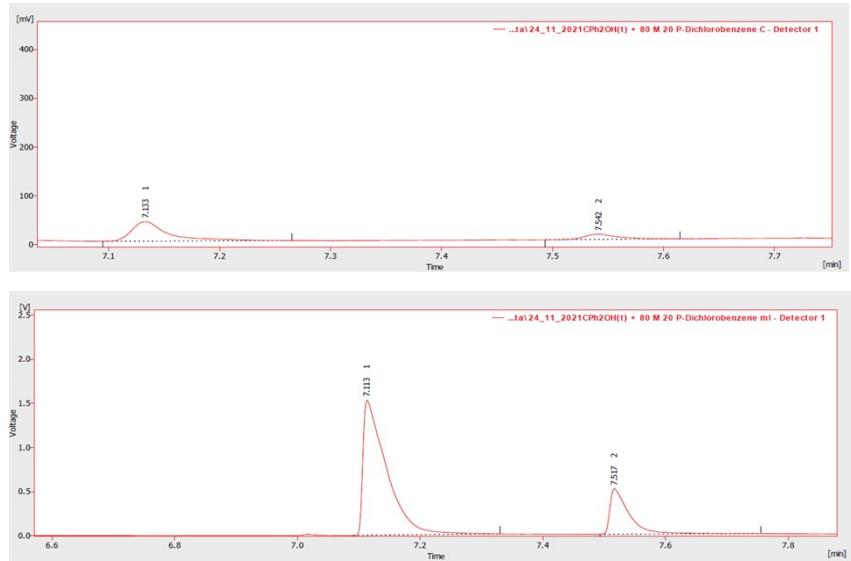
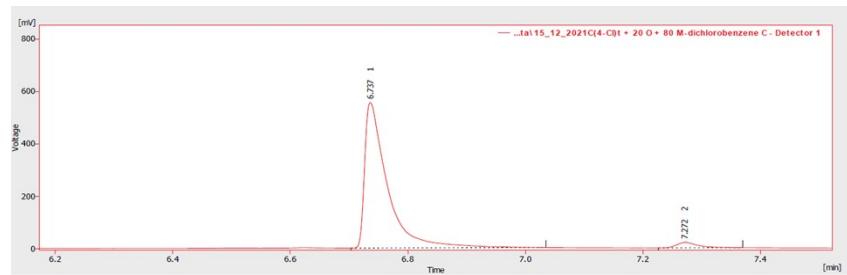


Figure S18. GC-FID chromatograms of the crystals (left) and mother liquor (right) for the 80:20 *m*-DCB/*p*-DCB binary experiments with H₂

a)



b)

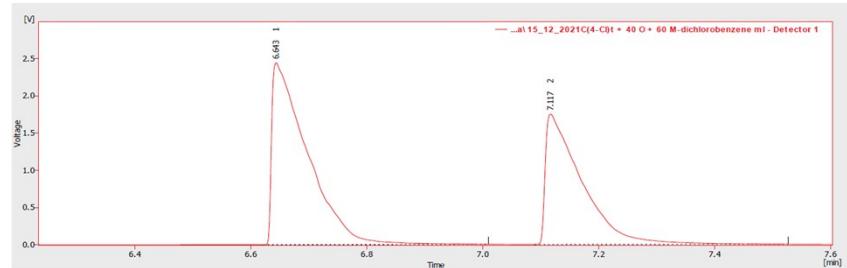
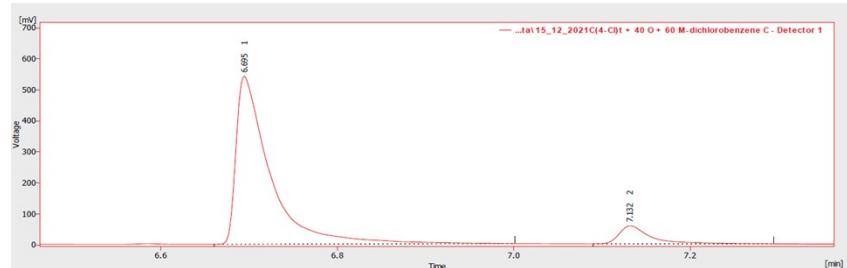
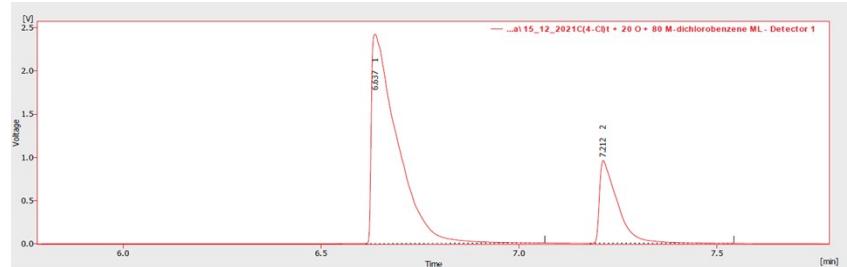
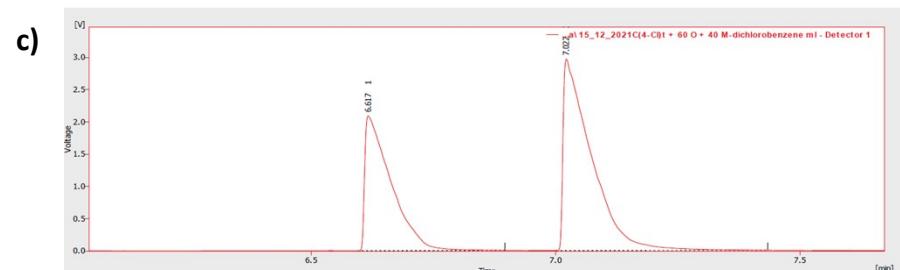
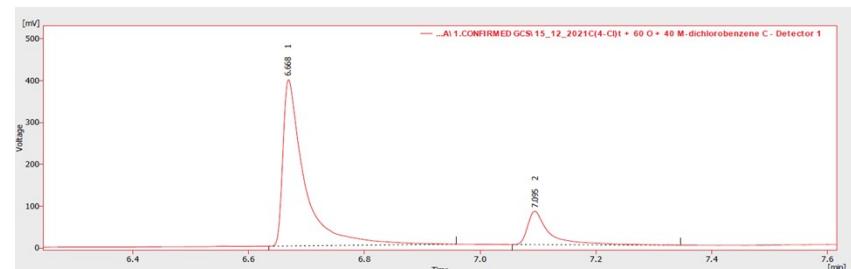
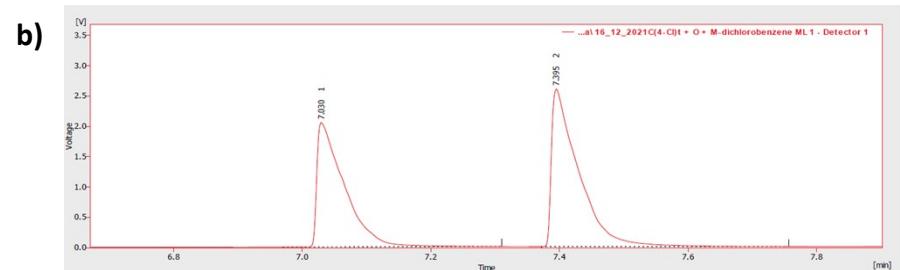
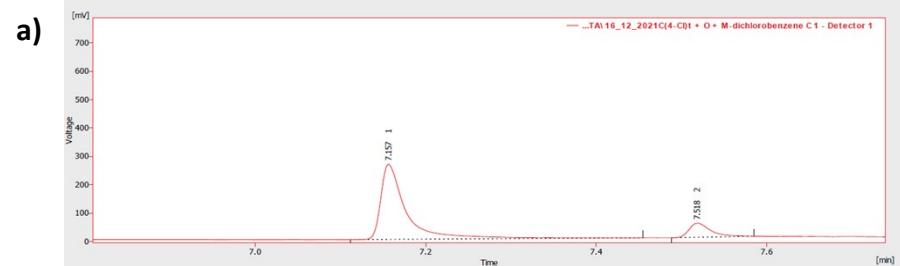


Figure S19. GC-FID chromatograms of the crystals (left) and mother liquor (right) for the 20:80 (a) and 40:60 (b) *o*-DCB/*m*-DCB binary experiments with

H3



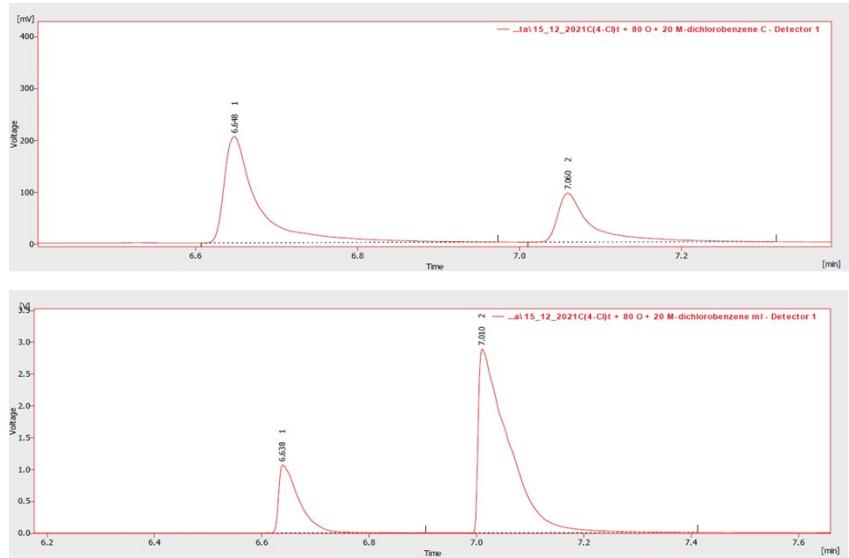
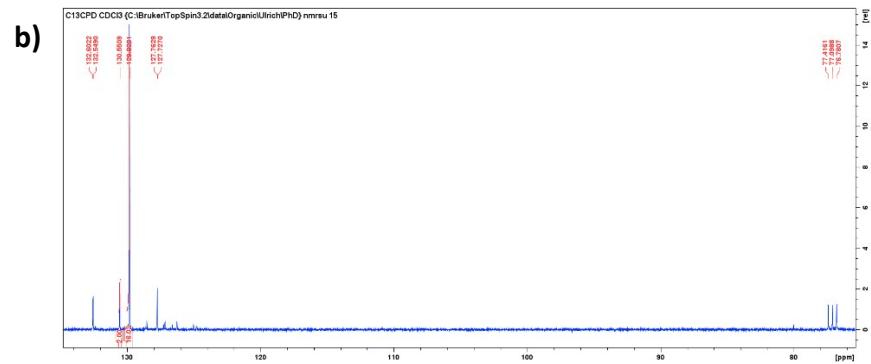
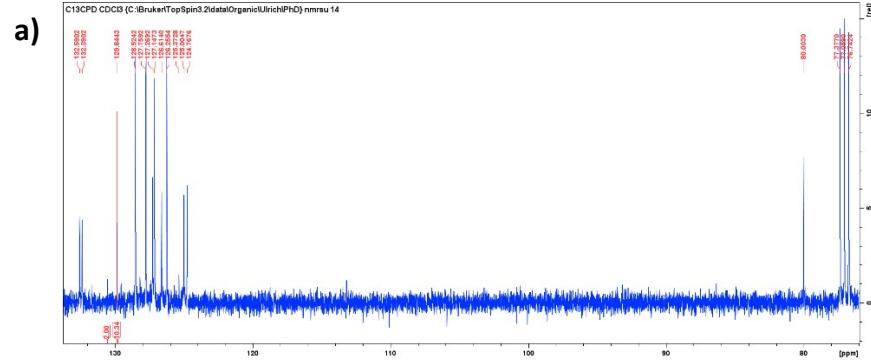
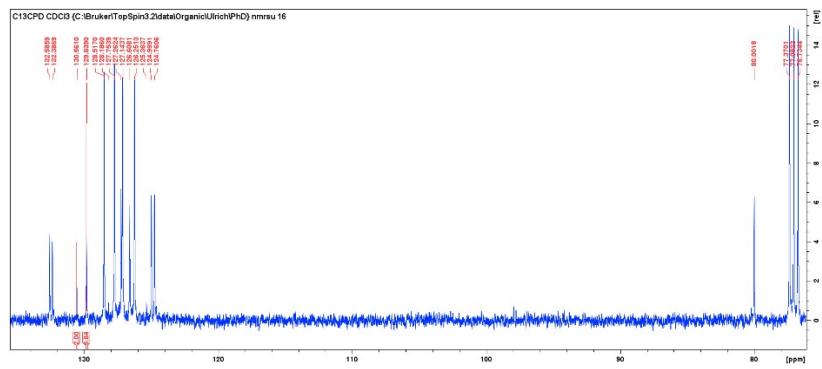
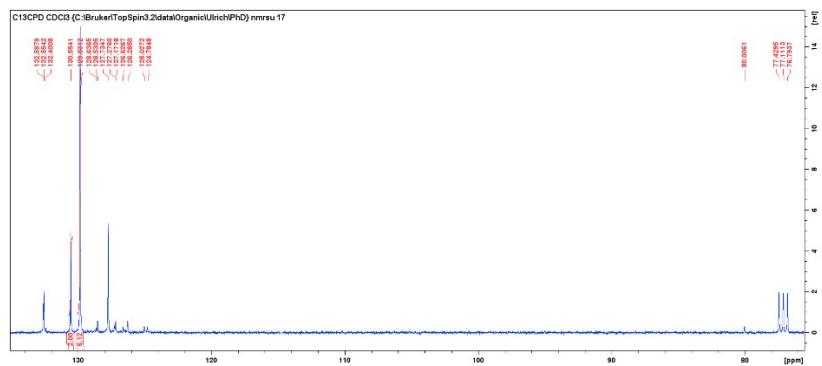


Figure S20. GC-FID chromatograms of the crystals (left) and mother liquor (right) for the 50:50 (a), 60:40 (a), and 80:20 (b) *o*-DCB/*m*-DCB binary experiments with H3





c)



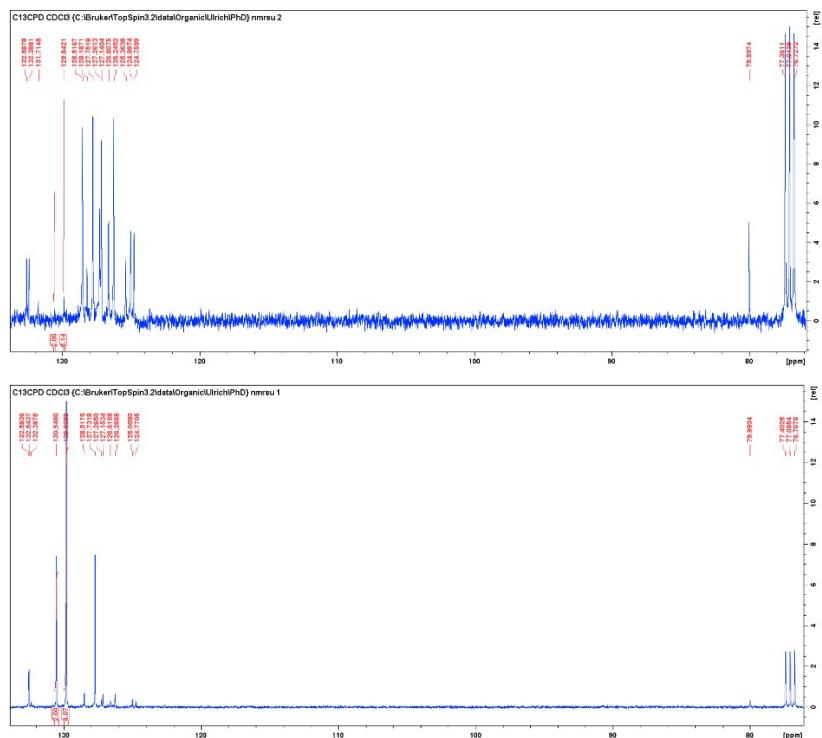


Figure S21. ^{13}C -NMR spectra of the crystals (left) and mother liquor (right) for the 20:80 (a), 40:60 (b), and 50:50 (c) o -DCB/ p -DCB binary experiments with H3

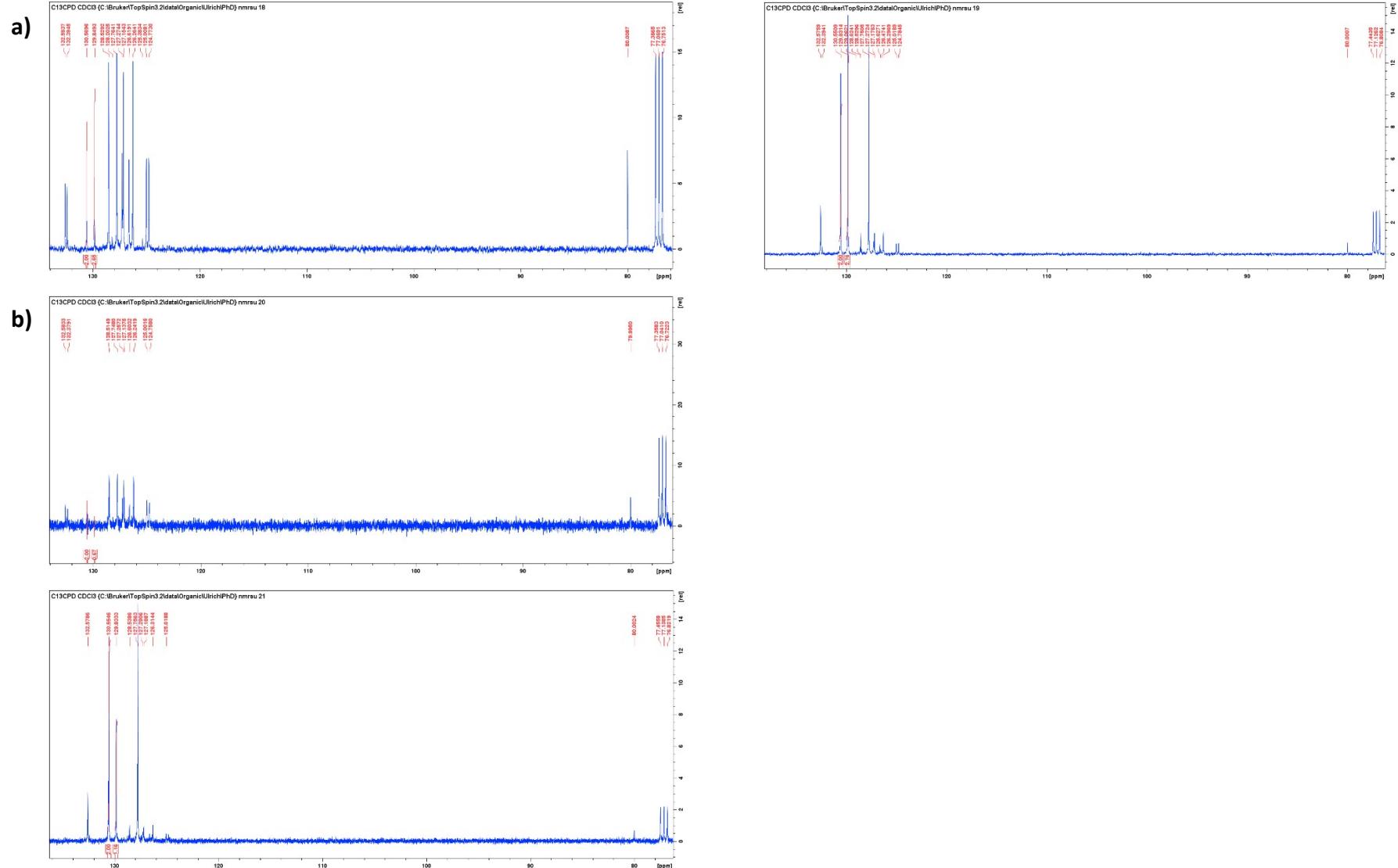
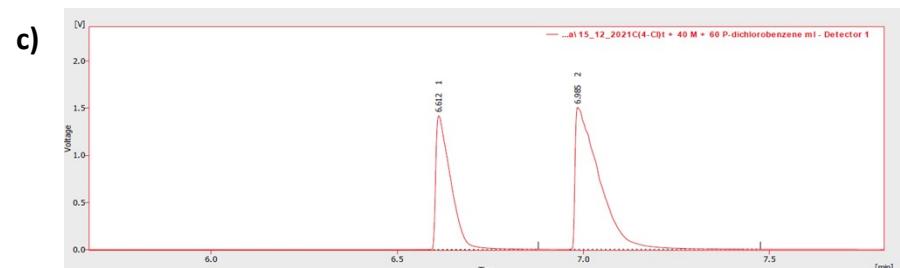
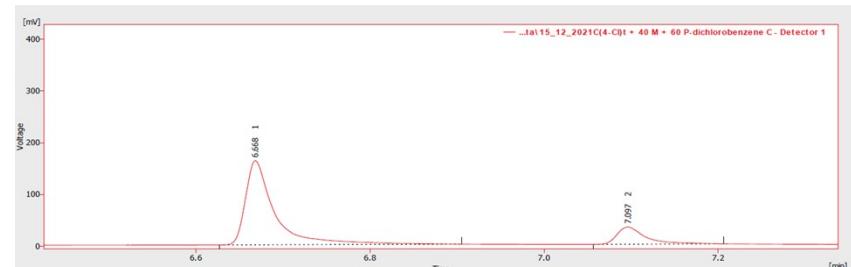
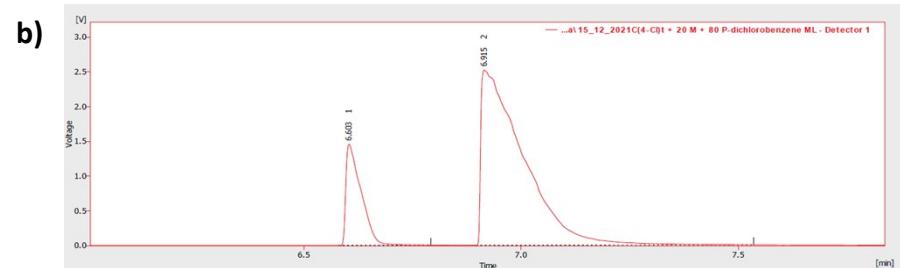
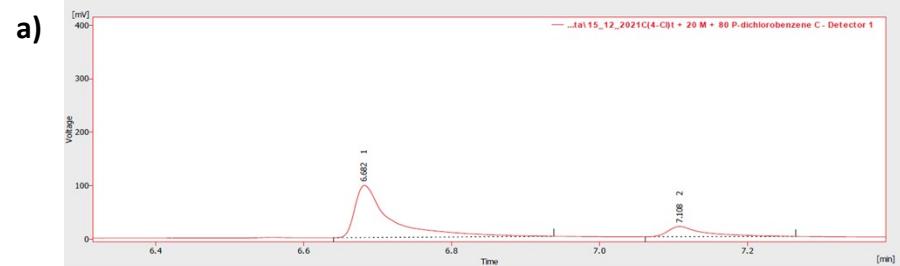


Figure S22. ^{13}C -NMR spectra of the crystals (left) and mother liquor (right) for the 60:40 (a) and 80:20 (b) *o*-DCB/*p*-DCB binary experiments

with

H3



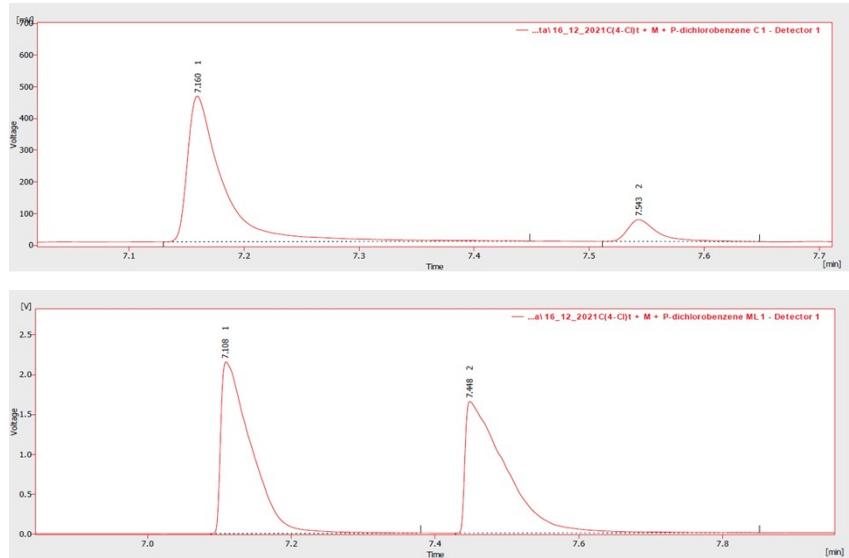


Figure S23. GC-FID chromatograms of the crystals (left) and mother liquor (right) for the 20:80 (a), 40:60 (b), and 50:50 (c) *m*-DCB/*p*-DCB binary experiments with H3

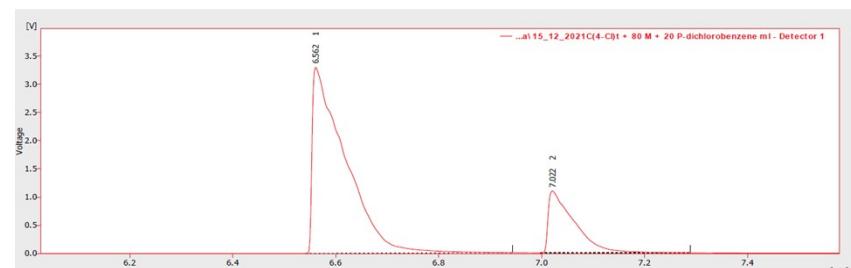
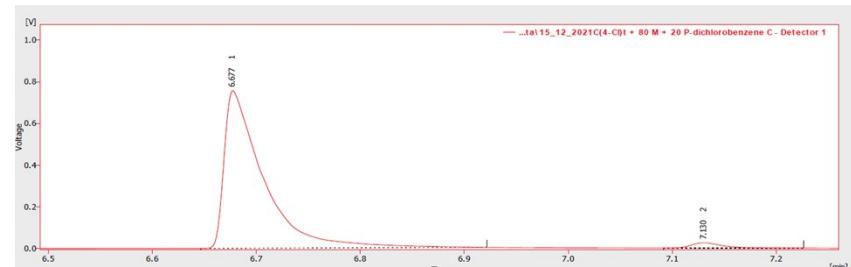
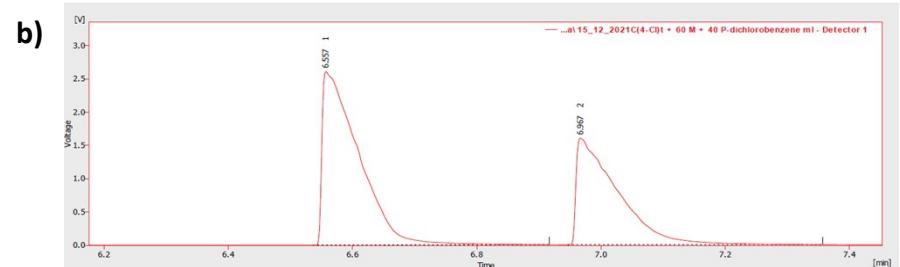
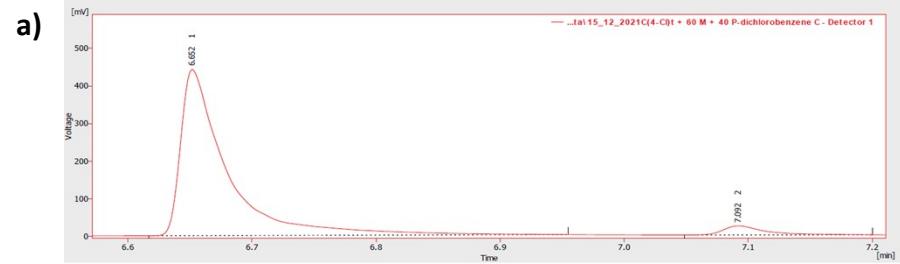


Figure S24. GC-FID chromatograms of the crystals (left) and mother liquor (right) for the 60:40 (a) and 80:20 (b) *m*-DCB/*p*-DCB binary experiments with H3

Table S2. Data points obtained for the guest/guest (dichlorobenzene) mixtures with corresponding K values for H2

Binary mixture	Crystal G _A :G _B	Mother liquor G _A :G _B	K values
20:80 <i>o</i> -DCB/ <i>m</i> -DCB	0.21:0.79	0.22:0.78	1.04
40:60 <i>o</i> -DCB/ <i>m</i> -DCB	0.41:0.59	0.43:0.57	1.12
50:50 <i>o</i> -DCB/ <i>m</i> -DCB	0.68:0.32	0.61:0.39	1.31
60:40 <i>o</i> -DCB/ <i>m</i> -DCB	0.71:0.29	0.62:0.38	1.51
80:20 <i>o</i> -DCB/ <i>m</i> -DCB	^a	^a	—
20:80 <i>o</i> -DCB/ <i>p</i> -DCB	0.02:0.98	0.18:0.82	10.98
40:60 <i>o</i> -DCB/ <i>p</i> -DCB	0.07:0.93	0.41:0.59	9.31
50:50 <i>o</i> -DCB/ <i>p</i> -DCB	0.64:0.36	0.50:0.50	1.82
60:40 <i>o</i> -DCB/ <i>p</i> -DCB	0.66:0.34	0.59:0.41	1.34
80:20 <i>o</i> -DCB/ <i>p</i> -DCB	0.68:0.32	0.80:0.20	1.87
20:80 <i>m</i> -DCB/ <i>p</i> -DCB	0.22:0.78	0.13:0.87	1.91
40:60 <i>m</i> -DCB/ <i>p</i> -DCB	0.36:0.64	0.38:0.62	1.11
50:50 <i>m</i> -DCB/ <i>p</i> -DCB	0.49:0.51	0.49:0.51	1.01
60:40 <i>m</i> -DCB/ <i>p</i> -DCB	0.62:0.38	0.60:0.40	1.10
80:20 <i>m</i> -DCB/ <i>p</i> -DCB	0.80:0.20	0.79:0.21	1.12

^aA gel remained in the vial

Table S3. Data points obtained for the guest/guest (dichlorobenzene) mixtures with corresponding K values for H3

Binary mixture	Crystal G _A :G _B	Mother liquor G _A :G _B	K values
20:80 <i>o</i> -DCB/ <i>m</i> -DCB	0.03:0.97	0.22:0.78	7.83
40:60 <i>o</i> -DCB/ <i>m</i> -DCB	0.10:0.90	0.42:0.58	6.80
50:50 <i>o</i> -DCB/ <i>m</i> -DCB	0.14:0.86	0.56:0.44	7.89
60:40 <i>o</i> -DCB/ <i>m</i> -DCB	0.17:0.83	0.61:0.39	7.53
80:20 <i>o</i> -DCB/ <i>m</i> -DCB	0.31:0.69	0.80:0.20	8.90
20:80 <i>o</i> -DCB/ <i>p</i> -DCB	0.35:0.65	0.20:0.80	2.19
40:60 <i>o</i> -DCB/ <i>p</i> -DCB	0.41:0.59	0.40:0.60	1.05
50:50 <i>o</i> -DCB/ <i>p</i> -DCB	0.44:0.56	0.50:0.50	1.26
60:40 <i>o</i> -DCB/ <i>p</i> -DCB	0.60:0.40	0.59:0.41	1.04
80:20 <i>o</i> -DCB/ <i>p</i> -DCB	0.86:0.14	0.78:0.22	1.74
20:80 <i>m</i> -DCB/ <i>p</i> -DCB	0.83:0.17	0.17:0.83	24.01
40:60 <i>m</i> -DCB/ <i>p</i> -DCB	0.83:0.17	0.37:0.63	8.63
50:50 <i>m</i> -DCB/ <i>p</i> -DCB	0.89:0.11	0.50:0.50	8.09
60:40 <i>m</i> -DCB/ <i>p</i> -DCB	0.95:0.05	0.59:0.41	13.97
80:20 <i>m</i> -DCB/ <i>p</i> -DCB	0.97:0.03	0.79:0.21	9.78

Table S4. Host···host forces for the H₂·*o*-DCB, 3(H₂)·*m*-DCB and H₃·*m*-DCB^a

Interactions	H ₂ · <i>o</i> -DCB	3(H ₂)· <i>m</i> -DCB	H ₃ · <i>m</i> -DCB	Symmetry
(host)π···π(host)	3.511(1) – 3.733(1) Å (2 contacts)	3.572(2) Å (1 contact)	3.561(2) Å (1 contact)	
C–H···π	(H···Cg, X–H···Cg) 2.89 Å, 136°	(H···Cg, X–H···Cg) 2.79 Å, 137°	(H···Cg, X–H···Cg) None	–x, –y, 1–z 1–x, y, 1–z 1–x, y, 1–z
(host)C–H···π(host) ^b		2.93 Å, 137°		
O–H···π				
(host)O–H···π(host) ^c	2.67 Å, 153°			x, y, z
(host)O–H···π(host) ^c	2.58 Å, 161°			x, y, z
(host)O–H···π(host) ^c		2.55 Å, 154°		x, y, z
(host)O–H···π(host) ^c		2.57 Å, 165°		x, y, z
(host)O–H···π(host) ^c			2.54 Å, 155°	x, y, z
(host)O–H···π(host) ^c			2.53 Å, 166°	x, y, z
C–Cl···π	(X···Cg, Y–X···Cg)	(X···Cg, Y–X···Cg)	(X···Cg, Y–X···Cg)	
(host)C–Cl···π(host) ^b	–	–	3.511(1) Å, 94.7(1)°	3/2–x, y, 1–z
(host)C–Cl···π(host) ^b			3.644(1) Å, 170.1(1)°	3/2–x, y, 1–z
(host)C–Cl···π(host) ^b			3.418(1) Å, 160.4(1)°	1/2+x, y, –z
(host)C–Cl···π(host) ^b			3.517(1) Å, 93.8(1)°	1/2+x, y, –z
Short contacts				
(host)C–H···O(host) ^c	2.33 Å, 112°, <<			
(host)C–H···O(host) ^c	2.55 Å, 106°, <<			
(host)C–H···O(host) ^c	2.41 Å, 103°, <<			
(host)C–H···O(host) ^c	2.29 Å, 102°, <<			
(host)C–H···O(host) ^c	2.41 Å, 102°, <<			
(host)C–H···O(host) ^c	2.29 Å, 102°, <<			
(host)C–H···O(host) ^c		2.38 Å, 111°, <<		
(host)C–H···O(host) ^c		2.48 Å, 109°, <<		
(host)C–H···O(host) ^c		2.43 Å, 103°, <<		
(host)C–H···O(host) ^c		2.32 Å, 102°, <<		
(host)C–H···O(host) ^c		2.38 Å, 103°, <<		
(host)C–H···O(host) ^c		2.30 Å, 103°, <<		
(host)C–C···Cl–C(host) ^b			3.443 Å, 103°, <	1/2–x, y, –z
(host)C–H···O–C(host) ^b			2.65 Å, 144°, <	1–x, 1/2+y, 1/2–z
(host)C–H···H–C(host) ^b			2.18 Å, 156°, <<	1/2+x, –1/2+y, 1/2+z
(host)C–H···O(host) ^c			2.34 Å, 111°, <<	
(host)C–H···O(host) ^c			2.44 Å, 108°, <<	
(host)C–H···O(host) ^c			2.37 Å, 103°, <<	
(host)C–H···O(host) ^c			2.29 Å, 102°, <<	
(host)C–H···O(host) ^c			2.32 Å, 102°, <<	
(host)C–H···O(host) ^c			2.35 Å, 104°, <<	
Hydrogen bonding	(H···A, D–H···A)	(H···A, D–H···A)	(H···A, D–H···A)	
	None	None	None	

^a< denotes contacts less than the sum of the van der Waals radii and << contacts less than this sum minus 0.2 Å

^bIntermolecular interaction

^cIntramolecular

interaction

Table S5. Host···guest and guest···guest forces for the H₂·*o*-DCB, 3(H₂)·*m*-DCB and H₃·*m*-DCB^a

Interactions	H ₂ · <i>o</i> -DCB	3(H ₂)· <i>m</i> -DCB	H ₃ · <i>m</i> -DCB	Symmetry
(host)π···π(guest)	Not significant	Not significant	Not significant	
(guest)π···π(guest)	None	None	None	
C–H···π	(H···Cg, X–H···Cg)	(H···Cg, X–H···Cg)	(H···Cg, X–H···Cg)	
(host)C–H···π(guest) ^b	2.91 Å, 145°		None	1–x, 1–y, 1–z
(host)C–H···π(guest) ^b		2.75 Å, 160°		x, 1–y, 1–z
C–Cl···π	(X···Cg, Y–X···Cg)	(X···Cg, Y–X···Cg)	(X···Cg, Y–X···Cg)	
(guest)C–Cl···π(host) ^b	None	None	3.864(2) Å, 134.2(1)°	1/2+x, 1–y, z
Short contacts				
(host)C–H···C–Cl(guest) ^b	2.85 Å, 146°, <			x, y, z
(guest)C–H···C–C(host) ^b	2.74 Å, 133°, <			1/2–x, 1/2+y, 3/2–z
(host)C–H···H–C(guest) ^b		2.35 Å, 161°, <		x, y, z
(host)C–H···C–Cl(guest) ^b		2.61 Å, 154°, <<		1–x, y, 1–z
(host)C–H···Cl–C(guest) ^b		2.75 Å, 155°, <<		1–x, 1–y, z
(host)C–H···C–C(guest) ^b		2.84 Å, 134°, <		x, 1–y, 1–z
(guest)C–H···C–C(guest) ^b		2.45 Å, 154°, <<		x, 1–y, 1–z
(guest)C–Cl···C–Cl(guest) ^b		2.51(3) Å, 156(2)°, <<		1–x, y, 2–z
(guest)C–Cl···C–Cl(guest) ^b		2.65(3) Å, 141(2)°, <<		x, 1–y, 2–z
(guest)C–H···C–C(guest) ^b		2.52 Å, 144°, <<		1–x, y, 1–z
(host)C–H···Cl(guest) ^b		2.75 Å, 123°, <<		1–x, y, 1–z
(guest)C–H···H–C(guest) ^b			2.24 Å, 148°, <	3/2–x, y, 1–z

^a< denotes contacts less than the sum of the van der Waals radii and << contacts less than this sum minus 0.2 Å

^bIntermolecular interaction