Electronic Supporting Information

Crystal engineering of monometallic lanthanide(III) supramolecular systems

within the N₃-tridentate hydrazone Schiff-base ligand

Dominika Prętka^{a*}, Dawid Marcinkowski^a, Agnieszka Siwiak^a, Maciej Kubicki^a, Giuseppe Consiglio^b, Violetta Patroniak^a, Adam Gorczyński^{a*}

^aFaculty of Chemistry, Adam Mickiewicz University in Poznań, Uniwersytetu Poznańskiego 8,

61-614 Poznań, Poland;

^bDipartimento di Scienze Chimiche, Università di Catania, I-95125 Catania, Italy;

Table of content

SI No.	Titles	Page
		No.
	Synthetic procedures details and characterization of the lanthanide complexes	3-9
	Supporting Tables	
Table S1.	Crystallographic data and structure refinement details of ligand LH·Br ⁻ .	9
Table S2.	Crystallographic data and structure refinement details of complexes 1-25 .	9-11
Table S3.	Crystallographic data and structure refinement details of complexes Tb-perchlorate and Eu-hydrogen .	12
Table S4.	Selected Bond Distances (Å) for complexes 1-6 .	12
Table S5.	Selected Bond Distances (Å) for complexes 7-11 .	12
Table S6.	Selected Bond Distances (Å) for complexes 2a and 7a .	13
Table S7.	Selected Bond Distances (Å) for complexes 13, 15-18, 22 and 24.	13
Table S8.	Selected Bond Distances (Å) for complexes 14 and 19-21.	13
Table S9.	Selected Bond Distances (Å) for complexes 23 and 25 .	13
	Supporting Figures	
Figure S1.	Representative FT-IR spectra of complexes from the triflate and nitrate series along	14
	their schematic coordination modes.	
Figure S2.	Representative mass spectra of complexes from the triflate and nitrate series along	14
F : C2	their schematic coordination modes.	45
Figure S3.	and Lu(III) nitrate (25) complexes.	15
Figure S4.	Perspective view one of the isostructural: (a) Pca2 ₁ complexes (4), (b) P2 ₁ /c complexes	15
	(9), (c) P2 ₁ /n (10) complexes and of the single complexes (d) P2 ₁ (2a), (e) $P\overline{1}$ (11);	
	ellipsoids are drawn at the 50% probability level, hydrogen atoms are depicted as	
Figure S5.	Crystal packing of complexes 2 (left) and 2a (right)	16
Figure S6.	Supramolecular structure created by molecule 10 as an example of motifs present in	16
	dicationic complexes from triflate series as seen along a axis. Thin blue lines show hydrogen bonds.	10
Figure S7.	Perspective view of the representative examples from the nitrate series of (a) Nd(III) complex (15), (b) Dy(III) complex (20), (c) Er(III) complex (22) and (d) Tm(III) complex (23); ellipsoids are drawn at the 50% probability level, hydrogen atoms are depicted as spheres of arbitrary radii.	17
Figure S8.	Crystal structure of complex 25 with lutetium(III) metal ion in the nitrate series.	17
Figure S9.	A comparison of the supramolecular structures created by molecules of 13 (a), 20 (b), 14 (c) and 24 (c) as even along a scient Thin block.	18
5. 645	14 (c) and 24 (d) as seen along a axis. Thin blue lines show hydrogen bonds.	
Figure S10.	Representative example of the supramolecular structure created by complex 16 .	19
Figure S11.	crystal structure of complex Tb-perchlorate with terbium(III) metal ion and perchlorate counterions.	19

Synthetic procedures details and characterization of the lanthanide complexes

Complexes obtained from Ln(OTf)³ **salts.** To a solution of **L** (40.0 mg, 0.18 mmol) in the mixture of MeOH/MeCN (1:1 v:v) appropriate metal salt (0.09 mmol) was added (La(CF₃SO₃)₃ - **1**, Pr(CF₃SO₃)₃ - **2**, Nd(CF₃SO₃)₃ - **3**, Sm(CF₃SO₃)₃ - **4**, Eu(CF₃SO₃)₃ - **5**, Gd(CF₃SO₃)₃ - **6**, Tb(CF₃SO₃)₃ - **7**, Dy(CF₃SO₃)₃ - **8**, Ho(CF₃SO₃)₃ - **9**, Er(CF₃SO₃)₃ - **10**, Yb(CF₃SO₃)₃ - **11** and Lu(CF₃SO₃)₃ - **12**. Upon mixing, the reaction mixtures instantly produced yellow solutions, which were stirred for 24 hours at room temperature. Subsequently, the solvents were evaporated under reduced pressure, resulting in residues that were dissolved in the smallest possible amount of a mixture containing methanol and acetonitrile in a 1:1 volume ratio. The dissolved residues were then precipitated by adding an excess of Et₂O. The resulting yellow solids were filtered using suction filtration and subsequently dried under vacuum.

[LaL₂(OTf)₃] (1)

Yield: 55.0 mg, 58.2%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v(C-H)_{arom} 3126; v_{as}(CH₃)2960; v_s(CH₃) 2894; v(C=C) 1610, 1591, 1571, 1502; v(C=N) 1442, 1331; δ (CH₃) 1382; v_{as}(SO₃) 1310; v_{as}(CF₃) 1245, 1220; v_s(CF₃) 1164; v_s(SO₃) 1035; γ (C-H)_{arom} 1003, 888, 770. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 652 (10) [LaL(CF₃SO₃)₂]⁺, 867 (10) [LaL₂(CF₃SO₃)₂]⁺.

[PrL₂(OTf)₃] (2)

Yield: 53.0 mg, 56.0%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v(C-H)_{arom} 3127; v_{as}(CH₃) 2956; v_s(CH₃) 2894; v(C=C) 1610, 1585, 1502; v(C=N) 1439, 1327; δ (CH₃) 1380; v_{as}(SO₃) 1305; v_{as}(CF₃) 1246, 1220; v_s(CF₃) 1158; v_s(SO₃) 1036; γ(C-H)_{arom} 1002, 961, 888, 772. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 536 (100) [PrL(CF₃SO₃)(CH₃OH)–H⁺]⁺, 869 (10) [PrL₂(CF₃SO₃)₂]⁺.

[PrL₂(OTf)₃](MeCN) (2a)

Yield: 54.2 mg, 56.0%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v(C-H)_{arom} 3144; v_{as}(CH₃) 2959; v_s(CH₃) 2896; v(C=C) 1610, 1587, 1506; v(C=N) 1447, 1327; δ (CH₃) 1378; v_{as}(SO₃) 1301; v_{as}(CF₃) 1247, 1221; v_s(CF₃) 1163; v_s(SO₃) 1035; γ(C-H)_{arom} 1002, 964, 891, 776. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 869 (10) [PrL₂(CF₃SO₃)₂]⁺.

3

[NdL₂(OTf)₃] (**3**)

Yield: 62.5 mg 65.8%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v(C-H)_{arom} 3129; v_{as}(CH₃) 2957; v_s(CH₃) 2899; v(C=C) 1611, 1588, 1502; v(C=N) 1445, 1329; δ (CH₃) 1380; v_{as}(SO₃) 1303; v_{as}(CF₃) 1247, 1219; v_s(CF₃) 1163; v_s(SO₃) 1036; γ(C-H)_{arom} 1004, 889, 772. ESI-MS(+) m/z (%): 216 (90) [LH]⁺, 655 (15) [NdL(CF₃SO₃)₂]⁺.

[SmL₂(OTf)₃] (4)

Yield: 60.8 mg, 63.6%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v(C-H)_{arom} 3127; v_{as}(CH₃) 2956; v_s(CH₃) 2896; v(C=C) 1610, 1590, 1502; v(C=N) 1442, 1328; δ (CH₃) 1381; v_{as}(SO₃) 1305; v_{as}(CF₃) 1237, 1220; v_s(CF₃) 1164; v_s(SO₃) 1033; v(C-H)_{arom} 1004, 889, 775. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 547 (90) [SmL₂(CF₃SO₃)(CH₃OH)-H⁺]⁺, 880 (20) [SmL₂(CF₃SO₃)]⁺.

[EuL₂(OTf)₃] (5)

Yield: 72.5 mg, 75.7%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v(C-H)_{arom} 3127; v_{as}(CH₃) 2962; v_s(CH₃) 2924; v(C=C) 1609, 1590, 1497; v(C=N) 1442, 1327; δ (CH₃) 1381; v_{as}(SO₃) 1285; v_{as}(CF₃) 1247, 1224; v_s(CF₃) 1167; v_s(SO₃) 1028; γ(C-H)_{arom} 1004, 963, 885, 771. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 548 (100) [EuL(CF₃SO₃)(CH₃OH)-H⁺]⁺, 881 (10) [EuL₂(CF₃SO₃)₂]⁺.

[GdL₂(OTf)₃] (6)

Yield: 52.9 mg, 55.0%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v(C-H)_{arom} 3127; v_{as}(CH₃) 2964; v_s(CH₃) 2898; v(C=C) 1611, 1589, 1497; v(C=N) 1439, 1323; δ (CH₃) 1378; v_{as}(SO₃) 1280; v_{as}(CF₃) 1252, 1223; v_s(CF₃) 1167; v_s(SO₃) 1029; γ(C-H)_{arom} 1004, 885, 769. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 368 (30) [GdL₂(CF₃SO₃)]²⁺, 886 (20) [GdL₂(CF₃SO₃)₂]⁺.

[TbL2(MeOH)2(OTf)](OTf)2 (7)

Yield: 68.9 mg, 64.4%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v(C-H)_{arom} 3127; v_{as}(CH₃) 2960; v_s(CH₃) 2897; v(C=C) 1610, 1590, 1501; v(C=N) 1440, 1329; δ(CH₃) 1380; v_{as}(SO₃) 1300; v_{as}(CF₃) 1241, 1218; v_s(CF₃) 1167; v_s(SO₃) 1025; γ(C-H)_{arom}

4

1004, 965, 888, 773. ESI-MS(+) m/z (%): 216 (95) [LH]⁺, 554 (20) [TbL(CF₃SO₃)(CH₃OH)-H⁺]⁺, 887 (15) [TbL₂(CF₃SO₃)₂]⁺.

[TbL2(OTf)2(MeCN)](OTf) (7a)

Yield: 64.3 mg, 64.4%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v(C-H)_{arom} 3129; v_{as}(CH₃) 2959; v_s(CH₃) 2938; v(C=C) 1611, 1591, 1499; v(C=N) 1442, 1328; δ (CH₃) 1380; v_{as}(SO₃) 1300; v_{as}(CF₃) 1239, 1221; v_s(CF₃) 1169; v_s(SO₃) 1030; γ(C-H)_{arom} 1006, 966, 886, 772. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 671 (35) [TbL(CF₃SO₃)₂]⁺, 887 (30) [TbL₂(CF₃SO₃)₂]⁺.

[DyL₂(MeOH)₂(OTf)](OTf)₂ (**8**)

Yield: 61.6 mg, 60.0%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad} (O-H)_{methanol} 3404; v(C-H)_{arom} 3127; v_{as} (CH₃) 2960; v_{s} (CH₃) 2900; v(C=C) 1609, 1590, 1497; v(C=N) 1438, 1326; δ (CH₃) 1379; v_{as} (SO₃) 1303; v_{as} (CF₃) 1231, 1222; v_{s} (CF₃) 1169; v_{s} (SO₃) 1029; γ (C-H)_{arom} 1004, 963, 888, 768. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 559 (60) [DyL₂(CF₃SO₃)₃+2K⁺]²⁺, 676 (20) [DyL(CF₃SO₃)₂]⁺, 892 (10) [DyL₂(CF₃SO₃)₂]⁺.

[HoL₂(OTf)₂(MeCN)](OTf) (9)

Yield: 63.3 mg, 62.8%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹):; v(C-H)_{arom} 3136; v_{as}(CH₃) 2963; v_s(CH₃) 2919; v(C=C) 1664, 1614, 1591, 1500; v(C=N) 1444, 1327; δ (CH₃) 1379; v_{as}(SO₃) 1291; v_{as}(CF₃) 1242, 1221; v_s(CF₃) 1168; v_s(SO₃) 1025; γ (C-H)_{arom} 1007, 966, 888, 768. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 560 (30) [HoL₂(CF₃SO₃)₃+2K⁺]²⁺, 678 (15) [HoL(CF₃SO₃)₂]⁺, 893 (10) [HoL₂(CF₃SO₃)₂]⁺.

[ErL₂(OTf)₂(MeOH)₂](OTf)₂ (10)

Yield: 63.3 mg, 54.2%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad} (OH)_{methanol} 3438; v(C-H)_{arom} 3025; v_{as} (C-H)_{aliph} 2983; v_{s} (C-H)_{aliph} 2947; v(C=C) 1677, 1614, 1504; v(C=N) 1443, 1328; δ (CH₃) 1383; v_{as} (SO₃) 1293; v_{as} (CF₃) 1256, 1245; v_{s} (CF₃) 1172; v_{s} (SO₃) 1030; γ (C-H)_{arom} 1023, 997, 889, 776. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 431 (15) [(HL)₂]⁺, 679 (10) [ErL(CF₃SO₃)₂]⁺.

[YbL₂(OTf)₂](OTf) (**11**)

Yield: 54.7 mg, 56.0%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad} (OH)_{methanol} 3467; v(C-H)_{arom} 3147; v_{as} (C-H)_{aliph} 2984; v_{s} (C-H)_{aliph} 2936; v(C=C) 1602, 1566; v(C=N) 1466, 1318; δ (CH₃) 1363; v_{as} (SO₃) 1289; v_{as} (CF₃) 1247, 1221; v_{s} (CF₃) 1161; v_{s} (SO₃) 1027; γ (C-H)_{arom} 986, 869, 799, 749. ESI-MS(+) m/z (%): 216 (100) [HL]⁺, 687 (30) [YbL(CF₃SO₃)₂]⁺, 902 (40) [YbL₂(CF₃SO₃)₂]⁺.

[LuL₂(OTf)₃] (12)

Yield: 55.9 mg, 57.8%. IR (KBr, cm⁻¹): v(C-H)_{arom} 3130; v_{as}(C-H)_{aliph} 2967; v_s(C-H)_{aliph} 2926; v(C=C) 1616, 1591, 1570, 1501; v(C=N) 1440, 1330; δ(CH₃) 1380; v_{as}(SO₃) 1278; v_{as}(CF₃) 1223; v_s(CF₃) 1167; v_s(SO₃) 1030; γ(C-H)_{arom} 1007, 861, 885, 770. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 570 (10) [LuL(CF₃SO₃)(CH₃OH)–H⁺]⁺, 269 (5) [LuL(CF₃SO₃)]²⁺. ¹H NMR (600 MHz, Acetonitrile-*d*₃) δ, ppm: 8.61 (1H), 8.02 (1H), 7.97 (1H), 7.38 (1H), 7.29 (1H), 7.27 (1H), 7.17 (1H), 3.92 (3H), 3.63 (3H). 3.61 (d, 3H).

Complexes obtained from Ln(NO₃)₃ salts. For complexes 13-25 the molar ratio of the ligand to the corresponding salt was 1:1. To a solution of L (40.0 mg, 0.18 mmol) in the mixture of MeOH/MeCN (1:1 v:v) appropriate metal salt (0.18 mmol) was added (La(NO₃)₃·6H₂O - 13, Pr(NO₃)₃·6H₂O - 14, Nd(NO₃)₃·6H₂O - 15, Sm(NO₃)₃·6H₂O - 16, Eu(NO₃)₃·5H₂O - 17, Gd(NO₃)₃·6H₂O - 18, Tb(NO₃)₃·5H₂O - 19, Dy(NO₃)₃·6H₂O - 20, Ho(NO₃)₃·5H₂O - 21, Er(NO₃)₃·5H₂O - 22, Tm(NO₃)₃·5H₂O - 23, Yb(NO₃)₃·5H₂O - 24 and Lu(NO₃)₃·8DMSO - 25. Subsequently, the solvents were evaporated under reduced pressure, resulting in residues that were dissolved in the smallest possible amount of a mixture containing methanol and acetonitrile in a 1:1 volume ratio. The dissolved residues were then precipitated by adding an excess of Et₂O or iPr₂O. The resulting yellow solids were filtered using suction filtration and subsequently dried under vacuum.

[LaL(NO₃)₃(MeOH)](MeCN) (13)

Yield: 70.7 mg, 62.0%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad} (O-H)_{methanol} 3387; v(C-H)_{arom} 3118; v_{as} (CH₃) 2974; v_{s} (CH₃) 2924; v(C=C) 1606, 1580; v_{as} (NO₂) 1497, 1453; v(C=N) 1433, 1318; δ(CH₃) 1384; $v_{s,broad}$ (NO₂) 1284, 1232; v(NO) 1042; γ(C-H)_{arom} 1002, 955, 885, 768; δ(NO) 817. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 239 (10) [LaL(NO₃)₂+H⁺]²⁺, 478 (10) [LaL(NO₃)₂]⁺.

[PrL(NO₃)₃(MeOH)](MeOH) (14)

Yield: 65.4 mg, 59.6%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad}(O-H)_{methanol} 3430; v(C-H)_{arom} 3127 v_{as}(CH₃) 2957; v_s(CH₃) 2923; v(C=C) 1632, 1609, 1583; v_{as}(NO₂) 1496 ; v(C=N) 1436, 1318; δ (CH₃) 1385; v_{s,broad}(NO₂) 1288, 1233; v(NO) 1060; γ(C-H)_{arom} 1034, 1002, 885, 773; δ (NO) 817. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 480 (90) [PrL(NO₃)₂]⁺, 695 (20) [PrL₂(NO₃)₂]⁺.

[NdL(NO₃)₃(MeOH)](MeCN) (15)

Yield: 80.6 mg, 70.1%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad}(O-H)_{methanol} 3432; v(C-H)_{arom} 3128 v_{as}(CH₃) 2960; v_s(CH₃) 2927; v(C=C) 1671, 1610, 1586; v_{as}(NO₂) 1497 ; v(C=N) 1436, 1322; δ (CH₃) 1387; v_{s,broad}(NO₂) 1278 1234 ; v(NO) 1051; γ(C-H)_{arom} 1034, 1006, 883, 771; δ (NO) 816. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 251 (10) [NdL(NO₃)₂+Na⁺]²⁺, 481 (10) [NdL(NO₃)₂]⁺.

[SmL(NO₃)₃(H₂O)](MeOH) (16)

Yield: 73.1 mg, 65.3%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad}(O-H)_{methanol} 3426; v(C-H)_{arom} 3127 v(CH₃) 2962 v_s(CH₃) 2930 v(C=C) 1672, 1633, 1610, 1590; v_{as}(NO₂) 1497; v(C=N) 1440, 1322; δ (CH₃) 1382; v_{s,broad}(NO₂) 1286, 1232; v(NO) 1049; γ(C-H)_{arom} 1034, 1005, 881, 769; δ (NO) 812. ESI-MS(+) m/z (%): 214 (25) [SmL(NO₃)]²⁺, 216 (100) [LH]⁺, 491 (20) [SmL(NO₃)₂]⁺.

[EuL(NO₃)₃(H₂O)](MeOH) (**17**)

Yield: 88.5 mg, 78.9%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad}(O-H)_{methanol} 3428; v(C-H)_{arom} 3122 v_{as}(CH₃) 2962; v_s(CH₃) 2926; v(C=C) 1671, 1635, 1610, 1587; v_{as}(NO₂) 1499 ; v(C=N) 1438, 1321; δ (CH₃) 1384; v_{s,broad}(NO₂) 1283, 1236; v(NO) 1060; γ(C-H)_{arom} 1034, 1003, 884, 770; δ (NO) 814. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 429 (10) [EuL(NO₃)-H⁺]⁺, 492 (45) [EuL(NO₃)₂]⁺.

[GdL(NO₃)₃(MeOH)](MeCN) (18)

Yield: 72.7 mg, 61.9%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad}(O-H)_{methanol} 3428; v(C-H)_{arom} 3126; v_{as}(CH₃) 2954; v_s(CH₃)

2931; v(C=C) 1672, 1609; v_{as}(NO₂) 1497, 1460; v(C=N) 1437, 1322; δ(CH₃) 1384; v_{s,broad}(NO₂) 1286, 1237; v(NO) 1046; γ(C-H)_{arom} 1007, 885, 773, 743; δ(NO) 815. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 249 (10) [GdL(NO₃)₂+H⁺]²⁺, 497 (25) [GdL(NO₃)₂]⁺.

[TbL(NO₃)₃(H₂O)](MeOH) (**19**)

Yield: 82.5 mg, 72.7%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad} (O-H)_{methanol} 3391; v(C-H)_{arom} 3147; v_{as} (CH₃) 2962; v_{s} (CH₃) 2923; v(C=C) 1665, 1611, 1586; v_{as} (NO₂) 1497; v(C=N) 1435, 1325; δ (CH₃) 1384; $v_{s,broad}$ (NO₂) 1289, 1234; v(NO) 1051; γ (C-H)_{arom} 1027, 1004, 883, 772; δ (NO) 811. ESI-MS(+) m/z (%): 216 (70) [LH]⁺, 315 (100) [TbL(NO₃)₃(H₂O)₂(CH₃OH)+2H⁺]²⁺, 498 (45) [TbL(NO₃)₂]⁺.

[DyL(NO₃)₃(H₂O)](MeOH) (20)

Yield: 73.3 mg, 64.2%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad} (O-H)_{methanol} 3374; v(C-H)_{arom} 3118; v_{as} (CH₃) 2959; v_{s} (CH₃) 2932; v(C=C) 1615, 1589, 1570; v_{as} (NO₂) 1497; v(C=N) 1437, 1322; δ (CH₃) 1384; $v_{s,broad}$ (NO₂) 1291, 1238; v(NO) 1049; v(C-H)_{arom} 1002, 885, 769; δ (NO) 812. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 317 (30) [DyL(NO₃)₃(H₂O)₂(CH₃OH)+2H⁺]²⁺, 502 (20) [DyL(NO₃)₂]⁺.

[HoL(NO₃)₃(MeOH)](MeCN) (21)

Yield: 78.5 mg, 66.0%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad} (O-H)_{methanol} 3397; v(C-H)_{arom} 3143; v_{as} (CH₃) 2959; v_{s} (CH₃) 2924; v(C=C) 1667, 1611, 1588; v_{as} (NO₂) 1497; v(C=N) 1436, 1325; δ (CH₃) 1384; $v_{s,broad}$ (NO₂) 1290, 1236; v(NO) 1051; γ (C-H)_{arom} 1007, 881, 772, 746; δ (NO) 811. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 321 (45) [HoL(NO₃)₃(CH₃CN)(CH₃OH)+2H⁺]²⁺, 504 (20) [HoL(NO₃)₂]⁺.

[ErL(NO₃)₃(H₂O)] (**22**)

Yield: 67.4 mg, 62.0%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad}(OH)_{methanol} 3426; v(C-H)_{arom} 3147; v_{as}(C-H)_{aliph} 2985; v_s(C-H)_{aliph} 2972; v(C=C) 1670, 1613, 1540; v_{as}(NO₂) 1460, 1450; v(C=N) 1439, 1325; δ (CH₃) 1384; v_s, _{broad}(NO₂) 1295, 1237; v(NO) 1051; γ(C-H)_{arom} 1110, 1030, 1006, 997, 882, 773; δ (NO) 811. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 199 (10) [ErL(H₂O)-H⁺]²⁺, 230 (10) [ErL(NO₃)(H₂O)]²⁺, 253 (10) [ErL(NO₃)₂+H⁺]²⁺.

[TmL(NO₃)₃(MeOH)](NO₃)(L) (23)

Yield: 101.5 mg, 62.0%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad}(O-H)_{methanol} 3396; v(C-H)_{arom} 3144 v_{as}(CH₃) 2955; v_s(CH₃) 2923; v(C=C) 1668, 1615, 1589; v_{as}(NO2) 1500 ; v(C=N) 1438, 1322; δ(CH₃) 1383; v_{s,broad}(NO₂) 1298, 1238; v(NO) 1050; γ(C-H)_{arom} 1031, 1007, 881, 772; δ(NO) 810. ESI-MS(+) m/z (%): 216 (70) [LH]⁺, 362 (5) [TmL₂(NO₃)₂+H⁺]²⁺, 508 (10) [TmL(NO₃)₂]⁺.

[YbL(NO₃)₃(MeOH)](MeCN) (24)

Yield: 77.4 mg, 65.0%. Crystals suitable for X-ray analysis were obtained *via* slow diffusion methods in MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹): v_{broad} (OH)_{methanol} 3422; v(C-H)_{arom} 3091; v_{as} (C-H)_{aliph} 2992; v_{s} (C-H)_{aliph} 2973; v(C=C) 1668, 1614, 1503; v_{as} (NO₂) 1461, 1450; v(C=N) 1439, 1325; δ (CH₃) 1384; $v_{s, broad}$ (NO₂) 1297, 1238; v(NO) 1051; γ (C-H)_{arom} 1111, 1031, 1007, 995, 881, 773, 713; δ (NO) 811. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 242 (20) [YbL(NO₃)(MeOH)]²⁺, 273 (10) [YbL(NO₃)₂(MeOH)+H]²⁺.

[LuL(NO3)3(DMSO)](MeCN) (25)

Yield: 74.5 mg, 61.2%. Crystals suitable for X-ray analysis were obtained via slow diffusion methods in MeOH, MeCN/iPr₂O system at 4°C. IR (KBr, cm⁻¹) v(C-H)_{arom} 3121; v_{as}(C-H)_{aliph} 2960; v_s(C-H)_{aliph} 2925; v(C=C) 1616, 1588; v_{as}(NO2) 1506; v(C=N) 1440, 1331; δ (CH3) 1387; v_{s,broad}(NO2) 1288, 1243; v(NO) 1047; γ(C-H)_{arom} 1031, 1003, 886, 765; δ (NO) 814. ESI-MS(+) m/z (%): 216 (100) [LH]⁺, 514 (20) [LuL(NO₃)₂]⁺, 278 (10) [LuL(NO₃)(MeOH)₂(MeCN)]²⁺. ¹H NMR (600 MHz, Acetonitrile-*d*₃) δ , ppm: 8.60 – 8.56 (1H), 7.97 – 7.93 (1H), 7.90 (1H), 7.22 (1H), 7.19 (1H), 7.13 – 7.11 (2H), 3.87 (3H), 3.62 – 3.60 (3H).

Supporting Tables

Table S1. Crystallographic data and structure refinement details of ligand LH·Br⁻.

LH·Br⁻: C₁₁H₁₄N₅⁺·Br⁻·CH₃OH, M_r=328.22, triclinic, P1, a=7.1754(4)Å, b=9.5801(5)Å, c=11.4950(6)Å, V=704.82(7)Å³, Z=2, d_x=1.547 g·cm⁻³, F(000)=336, μ =3.990mm⁻¹, 4875 reflections collected, of which 2784 unique (R_{int}=0.0555), 2751 with I>2 σ (I), R(F)[I>2 σ (I)]=0.0464, wR(F²)(I>4 σ (I))=0.1253, R(F)[all data]=0.0468, wR(F²)[all data]=0.1258, S=1.093, max/min Δp in the final ΔF map: 1.18/-1.16 e·Å⁻³.

 Table S2. Crystallographic data and structure refinement details of complexes 1-25.

1: C₂₅H₂₆F₉LaN₁₀O₉S₃, M_r=1016.65, orthorhombic, Pca₂₁, a=10.9243(2)Å, b=17.2695(4)Å, c=19.9168(4)Å, V=3763.64(13)Å³, Z=4, d_x=1.79 g·cm⁻³, F(000)=2016, μ =11.31mm⁻¹, 10708 reflections collected, of which 6116 unique (R_{int}=0.029), 5432 with I>2σ(I), R(F)[I>2σ(I)]=0.046, wR(F²)(I>4σ(I))=0.122, R(F)[all data]=0.052, wR(F²)[all data]=0.128, S=1.05, max/min Δp in the final ΔF map: 0.86/-0.46 e·Å⁻³.

2: C₂₅H₂₆F₉N₁₀O₉PrS₃, M_r=1018.65, orthorhombic, Pca₂₁, a=10.9258(3)Å, b=17.1974(4)Å, c=19.8714(6)Å, V=3733.74(18)Å³, Z=4, d_x=1.81 g·cm⁻³, F(000)=2024, μ=1.58mm⁻¹, 12868 reflections collected, of which 5629 unique (R_{int}=0.025), 4715 with I>2σ(I), R(F)[I>2σ(I)]=0.031, wR(F²)(I>4σ(I))=0.077, R(F)[all data]=0.043, wR(F²)[all data]=0.082, S=1.04, max/min Δρ in the final ΔF map: 0.56/-0.49 e·Å⁻³. Flack parameter: 0.430(10). **2a**: C₂₅H₂₆F₉N₁₀O₉PrS₃ C₂H₃N, M_r=1059.70, monoclinic, P2₁, a=8.9742(4)Å, b=14.0340(5)Å, c=15.8971(6)Å, β=92.838(4)°, V=1999.69(14)Å³, Z=2, d_x=1.76g·cm⁻³, F(000)=1056, μ=1.48mm⁻¹, 8138 reflections collected, of which 6021 unique (R_{int}=0.043), 5784 with I>2σ(I), R(F)[I>2σ(I)]=0.037, wR(F²)(I>4σ(I))=0.087, R(F)[all data]=0.039, wR(F²)[all data]=0.090, S=1.07, max/min Δρ in the final ΔF map: 2.42/-1.07e·Å⁻³.

3: C₂₅H₂₆F₉N₁₀NdO₉S₃, M_r=1021.98, orthorhombic, Pca2₁, a=10.7762(2)Å, b=17.0625(2)Å, c=19.6569(3)Å, V=3614.29(10)Å³, Z=4, d_x=1.88 g·cm⁻³, F(000)=2028, μ =1.72mm⁻¹, 25948 reflections collected, of which 6680 unique (R_{int}=0.021), 6553 with I>2σ(I), R(F)[I>2σ(I)]=0.021, wR(F²)(I>4σ(I))=0.052, R(F)[all data]=0.022, wR(F²)[all data]=0.052, S=1.06, max/min Δρ in the final ΔF map: 0.75/-0.54 e·Å⁻³. Flack parameter: 0.182(5).

4: $C_{25}H_{26}F_9N_{10}O_9S_3Sm$, M_r =1028.09, orthorhombic, Pca2₁, a=10.8093(5)Å, b=17.0375(5)Å, c=19.6847(5)Å, V=3625.2(2)Å³, Z=4, dx=1.88 g·cm⁻³, F(000)=2036, μ=1.90mm⁻¹, 8618 reflections collected, of which 4954 unique (R_{int}=0.020), 4899 with I>2σ(I), R(F)[I>2σ(I)]=0.025, wR(F²)(I>4σ(I))=0.066, R(F)[all data]=0.025, wR(F²)[all data]=0.066, S=1.05, max/min Δp in the final ΔF map: 1.02/-0.79 e·Å⁻³. Flack parameter: 0.323(7).

5: C₂₅H₂₆ EuF₉N₁₀O₉S₃, M_r=1029.70, orthorhombic, Pca2₁, a=10.8953(2)Å, b=17.1060(3)Å, c=19.8121(5)Å, V=3692.24(13)Å³, Z=4, d_x=1.85 g·cm⁻³, F(000)=2040, μ =1.98mm⁻¹, 51185 reflections collected, of which 8358 unique (R_{int}=0.014), 7868 with I>2σ(I), R(F)[I>2σ(I)]=0.026, wR(F²)(I>4σ(I))=0.066, R(F)[all data]=0.029, wR(F²)[all data]=0.068, S=1.04, max/min Δρ in the final ΔF map: 1.04/-0.51 e·Å⁻³. Flack parameter: 0.265(2).

6: C₂₅H₂₆F₉GdN₁₀O₉S₃, M_r=1034.99, orthorhombic, Pca2₁, a=10.9002(6)Å, b=17.0907(12)Å, c=19.8139(8)Å, V=3691.2(4)Å³, Z=4, d_x=1.86 g·cm⁻³, F(000)=2044, μ =2.08mm⁻¹, 9469 reflections collected, of which 5469 unique (R_{int}=0.041), 4714 with I>2σ(I), R(F)[I>2σ(I)]=0.055, wR(F²)(I>4σ(I))=0.150, R(F)[all data]=0.065, wR(F²)[all data]=0.157, S=1.07, max/min Δp in the final ΔF map: 1.09/-0.80 e·Å⁻³.

7: C₂₅H₃₄F₃N₁₀O₅STb 2(CF₃O₃S), M_r=1100.74, monoclinic, P2₁/n, a=10.9638(4)Å, b=15.0249(7)Å, c=26.8438(15)Å, β=99.705(5)°, V=4358.7(4)Å³, Z=4, d_x=1.68 g·cm⁻³, F(000)=2192, μ =1.87mm⁻¹, 18343 reflections collected, of which 7646 unique (R_{int}=0.032), 6990 with I>2σ(I), R(F)[I>2σ(I)]=0.096, wR(F²)(I>4σ(I))=0.206, R(F)[all data]=0.103, wR(F²)[all data]=0.209, S=1.13, max/min Δρ in the final ΔF map: 9.53/-3.60 e·Å⁻³.

7a: C₂₆H₂₆F₆N₁₁O₆S₂Tb CF₃O₃S, M_r=1074.69, monoclinic, P2₁/c, a=19.0497(5)Å, b=10.3198(2)Å, c=21.3425(5)Å, β=102.668(2)°, V=4093.57(17)Å³, Z=4, d_x=1.74 g·cm⁻³, F(000)=2124, μ =10.86mm⁻¹, 16653 reflections collected, of which 7381 unique (R_{int}=0.023), 6405 with I>2σ(I), R(F)[I>2σ(I)]=0.069, wR(F²)(I>4σ(I))=0.206, R(F)[all data]=0.076, wR(F²)[all data]=0.211, S=1.45, max/min Δρ in the final ΔF map: 1.71/-1.26 e·Å⁻³.

8: C₂₅H₃₄DyF₃N₁₀O₅S 2(CF₃O₃S), M_r=1104.32, monoclinic, P2₁/n, a=10.9170(3)Å, b=14.9827(3)Å, c=26.8953(10)Å, β=100.184(3)°, V=4329.9(2)Å³, Z=4, d_x=1.69 g·cm⁻³, F(000)=2196, μ =1.97mm⁻¹, 38766 reflections collected, of which 7596 unique (R_{int}=0.081), 5550 with I>2σ(I), R(F)[I>2σ(I)]=0.087, wR(F²)(I>4σ(I))=0.176, R(F)[all data]=0.118, wR(F²)[all data]=0.183, S=1.71, max/min Δρ in the final ΔF map: 5.13/-2.64 e·Å⁻³.

9: C₂₆H₂₆F₆HoN₁₁O₆S₂ CF₃O₃S, M_r=1083.72, monoclinic, P2₁/c, a=18.7750(7)Å, b=10.2574(2)Å, c=21.0038(6)Å, β =101.760(3)°, V=3960.1(2)Å³, Z=4, d_x=1.82 g·cm⁻³, F(000)=2144, μ =2.26mm⁻¹, 14768 reflections collected, of which 6967 unique (R_{int}=0.017), 6291 with I>2σ(I), R(F)[I>2σ(I)]=0.053, wR(F²)(I>4σ(I))=0.139, R(F)[all data]=0.059, wR(F²)[all data]=0.141, S=1.97, max/min Δρ in the final ΔF map: 2.87/-2.26 e·Å⁻³.

10: $C_{25}H_{34}\text{Er}F_{3}N_{10}O_5S$ 2(CF₃O₃S), M_r=1109.08 , monoclinic, P2₁/n, a=10.9049(4)Å, b=15.0238(4)Å, c=27.3563(9)Å, β=100.350(4)°, V=4408.9(3)Å³, Z=4, dx=1.67 g·cm⁻³, F(000)=2204, μ=2.15mm⁻¹, 18023 reflections collected, of which 9184 unique (R_{int}=0.046), 8398 with I>2σ(I), R(F)[I>2σ(I)]=0.070, wR(F²)(I>4σ(I))=0.170, R(F)[all data]=0.075, wR(F²)[all data]=0.173, S=1.12, max/min Δρ in the final ΔF map: 5.67/-3.19e·Å⁻³.

11: C₂₄H₂₆F₆N₁₀O₆S₂Yb CF₃O₃S, M_r=1050.78, triclinic, P1, a=11.3018(5)Å, b=13.3845(6)Å, c=14.5613(4)Å, α =96.292(3)°, β=99.214(3)°, γ=107.796(4)°, V=2040.55(15)Å³, Z=2, dx=1.71g·cm⁻³, F(000)=1034, μ=2.54mm⁻¹, 21875 reflections collected, of which 7183 unique (R_{int}=0.039), 6382 with I>2σ(I), R(F)[I>2σ(I)]=0.056, wR(F²)(I>4σ(I))=0.159, R(F)[all data]=0.062, wR(F²)[all data]=0.163, S=1.31, max/min Δρ in the final ΔF map: 3.23/-1.11e·Å⁻³.

13: $C_{12}H_{17}LaN_8O_{10}$ C_2H_3N , M_r =613.30, triclinic, $P\overline{1}$, a=8.2733(4)Å, b=11.4426(7)Å, c=13.5163(7)Å, α =109.919(6)°, β =106.972(4)°, γ =93.483(4)°, V=1132.21(12)Å³, Z=2, d_x=1.80g·cm⁻³, F(000)=608, μ =1.96mm⁻¹, 5084 reflections collected, of which 3393 unique (R_{int}=0.019), 3317 with I>2 σ (I), R(F)[I>2 σ (I)]=0.022,

wR(F²)(I>4 σ (I))=0.068, R(F)[all data]=0.023, wR(F²)[all data]=0.069, S=1.12, max/min $\Delta\rho$ in the final Δ F map: 0.66/-1.02e·Å⁻³.

14: C₁₂H₁₇N₈O₁₀Pr·½(CH₄O), M_r=590.27, triclinic, P1̄, a=8.1238(3)Å, b=10.5711(7)Å, c=12.6360(5)Å, α =99.359(4)°, β=97.224(4)°, γ=100.751(4)°, V=1038.18(9)Å³, Z=2, d_x=1.89g·cm⁻³, F(000)=586, μ=2.42mm⁻¹, 6734 reflections collected, of which 4006 unique (R_{int}=0.019), 3893 with I>2σ(I), R(F)[I>2σ(I)]=0.023, wR(F²)(I>4σ(I))=0.062, R(F)[all data]=0.024, wR(F²)[all data]=0.063, S=1.08, max/min Δρ in the final ΔF map: 1.07/-0.63e·Å⁻³.

15: $C_{12}H_{17}N_8NdO_{10}$ C_2H_3N , M_r =618.63, triclinic, $P\overline{1}$, a=8.3168(5)Å, b=11.3880(6)Å, c=13.4939(8)Å, α=110.119(5)°, β=107.600(5)°, γ=92.758(5)°, V=1127.23(12)Å³, Z=2, d_x=1.82g·cm⁻³, F(000)=614, μ=2.37mm⁻¹, 7629 reflections collected, of which 4261 unique (R_{int}=0.012), 4103 with I>2σ(I), R(F)[I>2σ(I)]=0.016, wR(F²)(I>4σ(I))=0.038, R(F)[all data]=0.017, wR(F²)[all data]=0.038, S=1.08, max/min Δρ in the final ΔF map: 0.47/-0.45e·Å⁻³.

16: C₁₁H₁₅N₈O₁₀Sm·CH₄O, M_r=601.70, triclinic, P $\overline{1}$, a=8.1537(3)Å, b=10.4239(4)Å, c=12.3673(3)Å, α=96.710(3)°, β=98.034(3)°, γ=100.381(4)°, V=1012.84(6)Å³, Z=2, d_x=1.97g·cm⁻³, F(000)=594, μ=2.98mm⁻¹, 7077 reflections collected, of which 4169 unique (R_{int}=0.020), 3879 with I>2σ(I), R(F)[I>2σ(I)]=0.021, wR(F²)(I>4σ(I))=0.045, R(F)[all data]=0.024, wR(F²)[all data]=0.046, S=1.04, max/min Δρ in the final ΔF map: 0.62/-0.50e·Å⁻³.

17: C₁₁H₁₅EuN₈O₁₀·CH₄O, M_r=603.31, triclinic, P $\overline{1}$, a=8.1426(4) Å, b=10.4679(8) Å, c=12.4743(7) Å, α=98.188(6) °, β=97.531(4)°, γ=100.576(5)°, V=1020.95(11) Å³, Z=2, d_x=1.96g·cm⁻³, F(000)=596, μ=3.14mm⁻¹, 7292 reflections collected, of which 4237 unique (R_{int}=0.023), 3940 with I>2σ(I), R(F)[I>2σ(I)]=0.035, wR(F²)(I>4σ(I))=0.081, R(F)[all data]=0.039, wR(F²)[all data]=0.083, S=1.07, max/min Δρ in the final ΔF map: 2.47/-1.89e·Å⁻³.

18: C₁₂H₁₇GdN₈O₁₀ C₂H₃N, M_r=631.64, triclinic, P1, a=8.2937(17)Å, b=11.3355(15)Å, c=13.441(3)Å, α=110.013(16)°, β=107.641(18)°, γ=92.604(14)°, V=1115.7(4)Å³, Z=2, d_x=1.88g·cm⁻³, F(000)=622, μ=3.04mm⁻¹, 7965 reflections collected, of which 4601 unique (R_{int}=0.024), 4217 with I>2σ(I), R(F)[I>2σ(I)]=0.024, wR(F²)(I>4σ(I))=0.055, R(F)[all data]=0.028, wR(F²)[all data]=0.056, S=1.05, max/min Δρ in the final ΔF map: 1.19/-0.77e·Å⁻³.

19: C₁₁H₁₅N₈O₁₀TbCH₄O, M_r=610.27, triclinic, P1, a=8.2400(4)Å, b=10.5276(6)Å, c=12.3755(6)Å, α=98.043(4)°, β=96.749(4)°, γ=100.129(5)°, V=1035.23(9)Å³, Z=2, dx=1.96g·cm⁻³, F(000)=600, μ=17.49mm⁻¹, 7534 reflections collected, of which 4161 unique (R_{int}=0.047), 4034 with I>2σ(I), R(F)[I>2σ(I)]=0.045, wR(F²)(I>4σ(I))=0.130, R(F)[all data]=0.045, wR(F²)[all data]=0.131, S=1.09, max/min Δρ in the final ΔF map: 1.87/-2.25e·Å⁻³.

20: C₁₁H₁₅DyN₈O₁₀ CH₄O, M_r=613.85, triclinic, P1, a=8.2029(7)Å, b=10.4813(9)Å, c=12.4024(7)Å, α =97.826(6)°, β=96.891(6)°, γ=100.152(7)°, V=1028.45(14)Å³, Z=2, d_x=1.98g·cm⁻³, F(000)=602, μ=3.71mm⁻¹, 7224 reflections collected, of which 4207 unique (R_{int}=0.017), 3955 with I>2σ(I), R(F)[I>2σ(I)]=0.021, wR(F²)(I>4σ(I))=0.048, R(F)[all data]=0.024, wR(F²)[all data]=0.049, S=1.06, max/min Δρ in the final ΔF map: 0.60/-0.47e·Å⁻³.

21: C₁₂H₁₇HoN₈O₁₀ C₂H₃N, M_r=639.32, triclinic, P1, a=8.2808(4)Å, b=11.2432(6)Å, c=13.2986(9)Å, α =68.357(6)°, β=72.061(5)°, γ=87.716(4)°, V=1091.08(12)Å³, Z=2, dx=1.95g·cm⁻³, F(000)=628, μ=3.70mm⁻¹, 8017 reflections collected, of which 4528 unique (R_{int}=0.026), 4302 with I>2σ(I), R(F)[I>2σ(I)]=0.026, wR(F²)(I>4σ(I))=0.064, R(F)[all data]=0.028, wR(F²)[all data]=0.066, S=1.09, max/min Δρ in the final ΔF map: 1.47/-1.50e·Å⁻³.

22: C₁₁H₁₅ErN₈O₁₀, M_r=586.57, triclinic, P1̄, a=8.3705(7)Å, b=9.3039(7)Å, c=12.7182(9)Å, α=102.222(6)°, β=103.692(7)°, γ=102.763(6)°, V=901.39(13)Å³, Z=2, d_x=2.16g·cm⁻³, F(000)=570, μ =4.73mm⁻¹, 5555 reflections collected, of which 3145 unique (R_{int}=0.068), 2826 with I>2σ(I), R(F)[I>2σ(I)]=0.057, wR(F²)(I>4σ(I))=0.149, R(F)[all data]=0.065, wR(F²)[all data]=0.154, S=1.23, max/min Δρ in the final ΔF map: 4.16/-2.24e·Å⁻³.

23: C₁₂H₁₇N₈O₁₀TmC₁₁H₁₄N₅·NO₃, M_r=880.55, triclinic, PĪ, a=8.3950(4)Å, b=12.7332(6)Å, c=16.9061(9)Å, α =71.362(5)°, β=80.611(4)°, γ=76.595(4)° V=1657.96(15)Å³, Z=2, dx=1.76g·cm⁻³, F(000)=880, μ=5.76mm⁻¹, 11642 reflections collected, of which 6651 unique (R_{int}=0.043), 6370 with I>2σ(I), R(F)[I>2σ(I)]=0.059, wR(F²)(I>4σ(I))=0.163, R(F)[all data]=0.061, wR(F²)[all data]=0.164, S=1.10, max/min Δρ in the final ΔF map: 1.72/-3.09e·Å⁻³.

24: $C_{12}H_{17}N_8O_{10}Yb$ C_2H_3N , M_r =647.43, triclinic, $P\overline{1}$, a=8.1752(5)Å, b=11.5352(9)Å, c=13.1045(10)Å, α =66.989(7)°, β =72.978(6)°, γ =85.535(5)°, V=1086.69(15)Å³, Z=2, d_x=1.98g·cm⁻³, F(000)=634, μ =4.38mm⁻¹, 4347 reflections collected, 3960 with I>2 σ (I), R(F)[I>2 σ (I)]=0.049, wR(F²)(I>4 σ (I))=0.159, R(F)[all data]=0.058,

wR(F²)[all data]=0.162, S=1.39, max/min $\Delta \rho$ in the final ΔF map: 3.27/-5.07e·Å⁻³.

25: C₁₃H₁₉LuN₈O₁₀S·CH₃CN, M_r=695.44, orthorhombic, Pna2, a=23.9529(11)Å, b=11.3128(4)Å, c=9.2768(4)Å, α =90°, β=90°, γ=90°, V=2513.77(18)Å³, Z=4, d_x=1.838g·cm⁻³, F(000)=1368, μ=8.900mm⁻¹, 9005 reflections collected, of which 4510 unique (R_{int}=0.0523), 4342 with I>2σ(I), R(F)[I>2σ(I)]=0.0531, wR(F²)(I>4σ(I))=0.1517, R(F)[all data]=0.0547, wR(F²)[all data]=0.1540, S=1.084, max/min Δρ in the final ΔF map: 2.22/-1.80e·Å⁻³.

Table S3. Crystallographic data and structure refinement details of complexes Tb-perchlorate and Euhydrogen.

Tb-perchlorate: C₂₄H₃₁ClN₁₁O₅Tb²⁺ ·2(ClO₄⁻), M_r=946.87, triclinic, PĪ, a=9.0968(9)Å, b=11.9590(8)Å, c=17.1192(12)Å, α=86.084(6)°, β=75.282(7)°, γ=83.710(7)° V=1788.8(3)Å³, Z=2, d_x=1.758g·cm⁻³, F(000)=944, μ=2.277mm⁻¹, 11546 reflections collected, of which 6235 unique (R_{int}=0.1416), 2982 with I>2σ(I), R(F)[I>2σ(I)]=0.0745, wR(F²)(I>4σ(I))=0.1365, R(F)[all data]=0.1362, wR(F²)[all data]=0.1654, S=0.905, max/min Δρ in the final ΔF map: 1.47/-1.21e·Å⁻³.

Eu-hydrogen: C₂₅H₃₀EuF₉N₁₀O_{11⁺}·CF₃SO₃⁻·CH₃OH, M_r=1065.73, monoclinic, P₂₁/c, a=13.59487(17)Å, b=15.49254(18)Å, c=17.9886(3)Å, α=90°, β=92.7582(12)°, γ=90° V=3784.35(9)Å³, Z=4, d_x=1.871g·cm⁻³, F(000)=2120, μ =1.936mm⁻¹, 38298 reflections collected, of which 8448 unique (R_{int}=0.0369), 7065 with I>2σ(I), R(F)[I>2σ(I)]=0.0269, wR(F²)(I>4σ(I))=0.0517, R(F)[all data]=0.0393, wR(F²)[all data]=0.0553, S=1.044, max/min Δp in the final ΔF map: 0.66/-0.57e·Å⁻³.

Table S4. Selected Bond Distances (Å)) for complexes 1-6 .
---------------------------------------	------------------------------

Bond	1	2	3	4	5	6
Ln-N14A	2.64(1)	2.575(8)	2.537(4)	2.522(6)	2.508(6)	2.50(2)
Ln-N14B	2.56(1)	2.551(9)	2.551(4)	2.515(6)	2.518(7)	2.51(2)
Ln-N1A	2.640(8)	2.589(5)	2.573(3)	2.542(4)	2.549(4)	2.54(1)
Ln-N1B	2.65(1)	2.594(6)	2.564(4)	2.553(5)	2.532(5)	2.53(1)
Ln-N8A	2.727(7)	2.676(5)	2.650(3)	2.637(4)	2.621(4)	2.61(1)
Ln-N8B	2.735(8)	2.691(5)	2.652(3)	2.628(3)	2.625(4)	2.61(1)
Ln-O13C	2.561(7)	2.528(4)	2.504(2)	2.502(3)	2.504(3)	2.507(9)
Ln-O13D	2.53(1)	2.505(6)	2.471(3)	2.457(4)	2.441(4)	2.43(1)
Ln-O13E	2.53(1)	2.485(6)	2.483(3)	2.447(4)	2.446(4)	2.44(1)

Table S5. Selected Bond Distances (Å)) for	complexes 7	′-11 .
---------------------------------------	-------	-------------	---------------

Bond	7	8	10	Bond	9	11
Ln-N14A	2.49(1)	2.478(8)	2.465(6)	Ln-N14A	2.454(5)	2.390(9)
Ln-N14B	2.484(9)	2.468(9)	2.445(6)	Ln-N14B	2.447(5)	2.402(6)
Ln-N1A	2.559(8)	2.539(7)	2.520(7)	Ln-N1A	2.545(5)	2.475(8)
Ln-N1B	2.563(9)	2.534(8)	2.515(6)	Ln-N1B	2.522(5)	2.470(7)
Ln-N8A	2.607(9)	2.592(8)	2.572(6)	Ln-N1E	2.543(6)	-
Ln-N8B	2.576(9)	2.570(8)	2.557(5)	Ln-N8A	2.552(4)	2.480(7)
Ln-O13C	2.355(9)	2.361(8)	2.331(6)	Ln-N8B	2.555(7)	2.511(7)
Ln-O1D	2.432(8)	2.441(8)	2.390(5)	Ln-O11C	2.344(4)	2.255(6)
Ln-O1E	2.461(9)	2.411(7)	2.434(6)	Ln-O11D	2.328(5)	2.235(5)

Bond	2a	Bond	7a
Ln-N14A	2.605(7)	Ln-N14A	2.480(5)
Ln-N14B	2.523(5)	Ln-N14B	2.468(6)
Ln-N1A	2.572(7)	Ln-N1A	2.577(6)
Ln-N1B	2.580(7)	Ln-N1B	2.544(6)
Ln-N8A	2.667(6)	Ln-N1D	2.572(8)
Ln-N8B	2.675(6)	Ln-N8A	2.577(6)
Ln-O13C	2.484(5)	Ln-N8B	2.579(8)
Ln-O13D	2.524(3)	Ln-O11E	2.368(5)
Ln-O13E	2.486(5)	Ln-O13C	2.366(5)

Table S6. Selected Bond Distances (Å) for complexes 2a and 7a.

Table S7. Selected Bond Distances (Å) for complexes 13, 15-18, 22 and 24.

Bond	13	15	16	17	18	22	24
Ln-N14A	2.613(3)	2.558(2)	2.515(2)	2.511(4)	2.510(3)	2.41(1)	2.418(5)
Ln-N1A	2.661(2)	2.608(2)	2.556(2)	2.548(4)	2.565(2)	2.45(1)	2.460(8)
Ln-N8A	2.688(2)	2.617(2)	2.572(2)	2.555(4)	2.570(3)	2.506(8)	2.486(8)
Ln-O11B	2.617(2)	2.613(2)	2.502(2)	2.476(4)	2.539(2)	2.460(8)	2.440(5)
Ln-O11C	2.596(3)	2.531(2)	2.507(2)	2.494(3)	2.474(3)	2.426(9)	2.451(7)
Ln-O11D	2.669(2)	2.565(1)	2.578(2)	2.572(3)	2.580(3)	2.302(9)	-
Ln-O12B	2.592(2)	2.553(1)	2.578(2)	2.567(4)	2.501(2)	2.382(7)	2.402(6)
Ln-O12C	2.656(2)	2.603(1)	2.576(2)	2.549(4)	2.574(2)	2.436(8)	2.382(7)
Ln-O12D	2.606(2)	2.544(1)	2.526(2)	2.519(4)	2.510(2)	-	2.251(5)
Ln-O1E	2.504(3)	2.449(2)	2.418(2)	2.426(3)	2.400(3)	2.357(6)	2.331(5)

 Table S8. Selected Bond Distances (Å) for complexes 14 and 19-21.

Bond	14	19	20	21
Ln-N14A	2.567(3)	2.490(5)	2.469(3)	2.468(2)
Ln-N1A	2.595(2)	2.538(5)	2.508(3)	2.523(3)
Ln-N8A	2.615(2)	2.535(4)	2.524(2)	2.500(3)
Ln-O12B	2.600(2)	2.562(5)	2.563(2)	2.540(4)
Ln-O12C	2.593(2)	2.573(5)	2.567(3)	2.447(2)
Ln-O12D	2.540(2)	2.490(4)	2.466(2)	2.422(2)
Ln-O13B	2.583(2)	2.485(5)	2.474(2)	2.453(3)
Ln-O13C	2.532(2)	2.440(5)	2.442(3)	2.503(3)
Ln-O13D	2.576(2)	2.550(5)	2.559(2)	2.530(2)
Ln-O1E	2.492(2)	2.398(4)	2.377(2)	2.358(2)

Table S9. Selected Bond Distances (Å) for complexes 23 and 25.

David	22	Devel	25
Bond	23	Bona	25
Ln- N14A	2.486(6)	Ln-N1A	2.40(1)
Ln-N1A	2.389(5)	Ln- N10A	2.43(1)
Ln-N8A	2.545(5)	Ln-N7A	2.53(1)
Ln- O12B	2.413(5)	Ln- 01C	2.563(9)
Ln- 012C	2.345(4)	Ln- 02C	2.371(9)
Ln- O13B	2.462(5)	Ln- O2B	2.38(1)
Ln- 013C	2.465(6)	Ln- 01B	2.46(1)
Ln- 011C	2.279(9)	Ln- 01D	2.28(1)
Ln- 01D	2.345(5)	Ln-O1E	2.240(8)

Supporting Figures



Figure S1. Representative FT-IR spectra of complexes from the triflate and nitrate series along their schematic coordination modes.



Figure S2. Representative mass spectra of complexes from the triflate and nitrate series along their schematic coordination modes.



Figure S3. The comparison of ¹H NMR spectra of ligand L and the diamagnetic Lu(III) triflate (**12**) and Lu(III) nitrate (**25**) complexes.



Figure S4. Perspective view one of the isostructural: (a) $Pca2_1$ complexes (4), (b) $P2_1/c$ complexes (9), (c) $P2_1/n$ (10) complexes and of the single complexes (d) $P2_1(2a)$, (e) $P\overline{1}$ (11); ellipsoids are drawn at the 50% probability level, hydrogen atoms are depicted as spheres of arbitrary radii.



Figure S5. Crystal packing of complexes 2 (left) and 2a (right).



Figure S6. Supramolecular structure created by molecule **10** as an example of motifs present in dicationic complexes from triflate series as seen along a axis. Thin blue lines show hydrogen bonds.



Figure S7. Perspective view of the representative examples from the nitrate series of (a) Nd(III) complex (15), (b) Dy(III) complex (20), (c) Er(III) complex (22) and (d) Tm(III) complex (23); ellipsoids are drawn at the 50% probability level, hydrogen atoms are depicted as spheres of arbitrary radii.



Figure S8. Crystal structure of complex 25 with lutetium(III) metal ion in the nitrate series.



Figure S9. A comparison of the supramolecular structures created by molecules of 13 (a), 20 (b), 14 (c) and 24 (d) as seen along a axis. Thin blue lines show hydrogen bonds.



Figure S10. Representative example of the supramolecular structure created by complex 16.



Figure S11. Crystal structure of complex Tb-perchlorate with terbium(III) metal ion and perchlorate counterions.