Electronic Supplementary Information

For

Article

Crystal structure of Febuxostat marketed polymorph determined by electron diffraction and reinforced by X-ray crystallography

Doriana T. Ungur^{a, b}, Arianna Lanza^c, Danny Stam^d, Carmen Guguta^e, Coca Iordache^b, Victor Fruth^f, Gustavo Santiso-Quinones^{d,g}, Mihaela M. Pop^{*b}

- ^a Doctoral School in Integrative Biology, Faculty of Biology and Geology, "Babeș-Bolyai" University, Cluj-Napoca, Romania.
- ^b TeraCrystal SRL, 67-103 Donat, 400293 Cluj-Napoca, Romania.
- ^c University of Copenhagen, Universitetsparken 5, 2100 Copenhagen Ø, Denmark.
- ^d ELDICO Scientific AG, Park Innovaare, 5234 Villigen, Switzerland.
- ^e Technobis Crystallization Systems BV, Pyrietstraat 2, 1812 SC Alkmaar, The Netherlands.
- ^f "Ilie Murgulescu" Institute of Physical Chemistry, Romanian Academy, 202 Splaiul Independenței, Bucharest, 060021, Romania.
- ^g Crystallise! AG, Hegenheimermattweg 167A, 4123 Allschwil, Switzerland.
- * Author to whom correspondence should be addressed <u>mihaela.pop@teracrystal.com</u>; <u>santiso@crystallise.ch</u>

Solvent used Concentration of FEB [mg/mL] Crystallization method Form by X-ray powder diffraction 30 60 Form A + Q 2-Ethoxyethanol 120 Form A 240 Petermination of MSZW Form A 30 0 Form A 30 0 Form A	Table S1. Crystallization experiments targetting form A.				
30 Form A + Q 2-Ethoxyethanol 120 240 Form A 360 Determination of MSZW 30 Form F3	Solvent used	Concentration of FEB [mg/mL]	Crystallization method	Form by X-ray powder diffraction	
60 Form A 2-Ethoxyethanol 120 240 Form A 360 Determination of MSZW 30 60	2-Ethoxyethanol	30		Form A + Q	
2-Ethoxyethanol 120 240 360 Determination of MSZW 60 Form A Form A Form A		60			
240 Form A 360 Determination of MSZW 30 60		120		Form A	
360 Determination of MSZW 30 60		240	-		
		360	Determination of MSZW		
Ethanol 60 Form F3	Ethanol	30			
		60		Form F3	
120		120			
180		180			
Acetonitrile/Water 5 Anti-solvent vapor diffusion Form A	Acetonitrile/Water	5	Anti-solvent vapor diffusion	Form A	
2-Ethoxyethanol - Schwart drag grindling Form Q	2-Ethoxyethanol	-	Solvent-drop grinding	Form Q	
THF - Solvent-drop grinning Form A + Q	THF	-		Form A + Q	
2-Ethoxyethanol 125 Recrystallization of solvent-drop grinding materials Form A	2-Ethoxyethanol	125	Recrystallization of solvent-drop grinding materials	Form A	
2-Ethoxyethanol 100 Form A	2-Ethoxyethanol	100	Cooling evaporative crystallization	Form A	
THF75Cooling evaporative crystallizationForm A + Q	THF	75		Form A + Q	
Ethanol 35 Form F3	Ethanol	35		Form F3	

 Table S2. Calculation settings of the force fields used in the molecular mechanics calculations.

Variable	Setting			
Forcite				
Optimization algorithm	Smart			
Convergence energy	2.0 × 10 ⁻⁵ kcal/mol			
Convergence force	0.001 kcal/(mol Å)			
Convergence stress	0.001 GPa			
Conv. displacement	1.0 × 10 ⁻⁵ Å			
Max. iterations	1000			
External pressure	0			
Optimize cell	yes			
E ⁻ static sum method	Ewald			
Ewald accuracy	1.0 × 10 ⁻⁵ kcal/mol			
Buffer width	0.5 Å			
vdW sum method	Ewald			
Ewald accuracy	1.0 × 10 ⁻⁵ kcal/mol			
Buffer width	0.5 Å			
Repulsive cutoff	6.0 Å			
H-bond sum method	atom based			
Cutoff distance	6.0 Å			
Spline width	0.5 Å			
Buffer width	0.5 Å			



Figure S1. Experimental and calculated X-ray powder patterns of form A based on the SC-XRD and ED data.



Figure S2. Overlays of the FEB molecules from the CSD structures with the unique conformers identified by the Boltzmann method: (a) HIQQAB02→molecule 1 (red) *versus* conformer F1 (blue); (b) HIQQAB02→molecule 2 (red) *versus* conformer F2 (blue); (c) UREQOY (red) *versus* conformer F2 (blue); (d) HIQQIJ (red) *versus* conformer F2 (blue); (e) HIQQUV (red) *versus* conformer F2 (blue) and (f) HIQQEF (red) *versus* conformer F5 (blue). All values are in Å.