

Supporting Information

Bimetallic metal-organic framework-derived cobalt selenide-based composites as bifunctional electrocatalysts for both hydrogen evolution and mono-alcohol oxidation

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Fig. S1 SEM image of CuCo-MOF.

Fig. S2 TEM image of Cu-Co_{0.85}Se@NC.

Fig. S3 XRD pattern of Co-MOF and CuCo-MOF.

Fig. S4 Cu 2p spectra of Co_{0.85}Se@NC and Cu-Co_{0.85}Se@NC materials.

Fig. S5 C 1s spectra of Co_{0.85}Se@NC and Cu-Co_{0.85}Se@NC materials.

Fig. S6 N 1s spectra of Co_{0.85}Se@NC and Cu-Co_{0.85}Se@NC materials.

Fig. S7 (a) HER long-term stability test for the Cu-Co_{0.85}Se@NC. (b) LSV curves of Cu-Co_{0.85}Se@NC before and after the stability test.

Fig. S8 XRD pattern of Cu-Co_{0.85}Se@NC before and after the HER stability test.

Fig. S9 (a) HER polarization curves, (b) HER Tafel slopes, (c) OER polarization curves and (d) OER Tafel slopes for Cu-Co_{0.85}Se@NC, Cu-Co_{0.85}Se@NC-1 and Cu-Co_{0.85}Se@NC-2 electrocatalysts.

Fig. S10 CV curves of (a) Co_{0.85}Se@NC and (b) Cu-Co_{0.85}Se@NC at different scan rates of 20, 50, 100, 150 and 200 mV s⁻¹.

Fig. S11 Nyquist plots of Co_{0.85}Se@NC and Cu-Co_{0.85}Se@NC electrodes.

Fig. S12 Performance comparison of the mono-alcohol oxidation reaction (M-AOR) with different concentrations of electrolytes.

Fig. S13 (a) CV curves at various scan rates from 20 mV s⁻¹ to 200 mV s⁻¹ for Cu-Co_{0.85}Se@NC. (b) The corresponding C_{dl} value comparison of Cu-Co_{0.85}Se@NC in 1.0 M KOH with and without methanol.

Fig. S14 (a) MOR long-term stability test for the Cu-Co_{0.85}Se@NC. (b) LSV curves of Cu-Co_{0.85}Se@NC before and after the stability test.

Fig. S15 Raman spectrum of Co_{0.85}Se@NC and Cu-Co_{0.85}Se@NC catalysts after HER test.

Fig. S16 The contact angles of Cu-Co_{0.85}Se@NC in (a) n-propanol, (b) isopropanol and (c) ethanol.

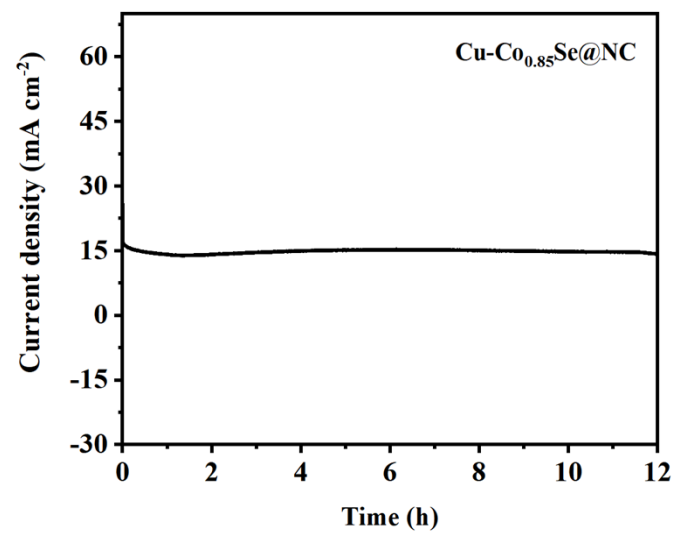


Fig. S17 The i-t curve of the Cu-Co_{0.85}Se@NC catalyst.