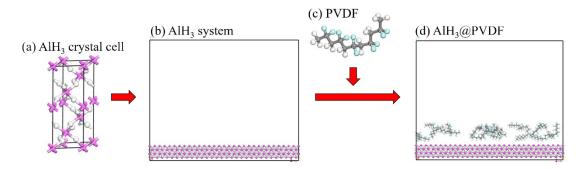
Supporting information

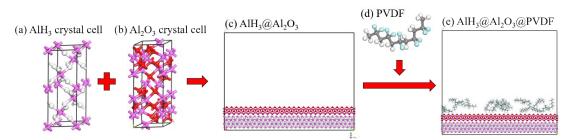
Reactivity and stabilization mechanisms of AlH₃ crystals by

coating of polyvinylidene difluoride

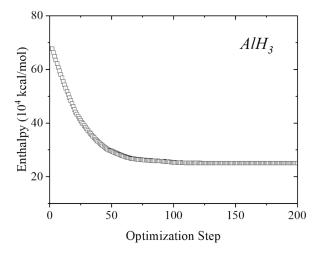
Yiran Zhang, Haorui Zhang, Minghui Yu, Qi-Long Yan* National Key Laboratory of Solid Rocket Propulsion, Northwestern Polytechnical University, Xi'an 710072, China



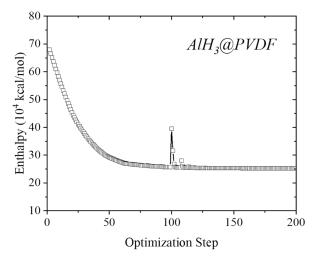
Figs. S1. Detailed building process of AlH₃@PVDF



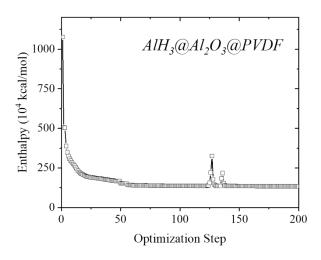
Figs. S2. Detailed building process of AlH₃@Al₂O₃@PVDF



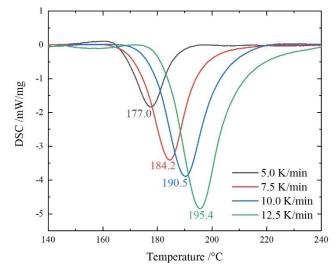
Figs. S3. Energy change of the AlH₃ system during relaxation



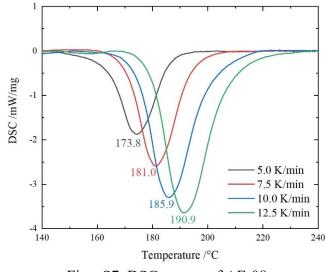
Figs. S4. Energy change of the AlH₃@PVDF system during relaxation

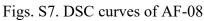


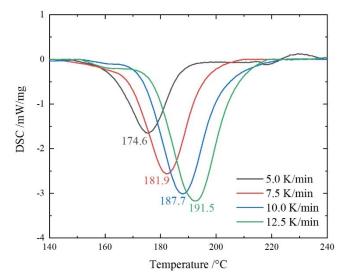
Figs. S5. Energy change of the AlH₃@Al₂O₃@PVDF system during relaxation

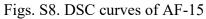


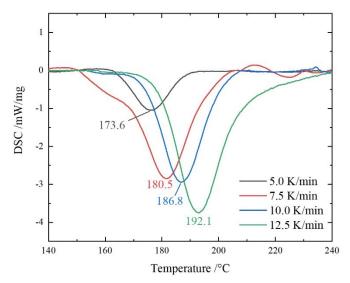
Figs. S6. DSC curves of AlH₃











Figs. S9. DSC curves of AF-20

Samples	R	$T_{\rm i}$	T_{p}	ΔH
AlH ₃	5.0	167.8	177.0	313.7
	7.5	173.5	184.2	393.8
	10.0	179.1	190.5	370.8
	12.5	183.9	1954	301.5
AF-08	5.0	165.3	173.8	313.1
	7.5	171.3	181.0	322.9
	10.0	176.2	185.9	312.6
	12.5	180.1	190.9	285.0
AF-15	5.0	163.6	174.6	298.3
	7.5	169.5	181.9	318.8
	10.0	174.1	187.7	296.2
	12.5	178.1	191.5	244.5
AF-20	5.0	167.2	173.6	191.0
	7.5	180.5	180.5	278.2
	10.0	176.1	186.8	275.4
	12.5	179.5	192.1	256.9

Table S1. Detailed DSC data for AlH₃ and composites.

Notes: *R*, the heating rate, in °C min⁻¹; T_i , the onset temperature, in °C; T_p , the peak temperature, in °C; ΔH , heat absorption, in J g⁻¹.