## **Electronic Supplementary Information (ESI)**

## Self-reversible mechanofluorochromic AIEgens with tunable solid-state fluorescence: effectof acceptor and intermolecular interactions





Scheme S1. Synthesis of carbazole derivatives with different acceptor units.



 $^{1}H$  and  $^{13}C$  NMR of **Cz-MN** in CDCl<sub>3</sub>.



 $^{1}H$  and  $^{13}C$  NMR of **Cz-ECA** in CDCl<sub>3</sub>.







 $^{1}H$  and  $^{13}C$  NMR of **Cz-CAA** in CDCl<sub>3</sub>.



 $^{1}H$  and  $^{13}C$  NMR of **Cz-NBA** in CDCl<sub>3</sub>.



HRMS of Cz-MN calculated: 349.1; found 349.1.



*HRMS* of **Cz-ECA** calculated: 396.1; found 396.5.



HRMS of Cz-CAA calculated: 368.1; found 368.1.



*HRMS* of **Cz-CA** calculated: 367.1; found 367.2.



*HRMS* of **Cz-NBA** calculated: 439.1; found 439.0.



Fig. S1. Absorption spectra of (a) Cz-MN, (b) Cz-ECA, (c) Cz-CA, (d) Cz-CAA and (e) Cz-NBA in different solvents.



Fig. S2. Solid-state absorption spectra.



Fig. S3. AIE studies of Cz-CAA.



Fig. S4. AIE studies of Cz-NBA.

Table S1. Optical band gap of carbazole derivatives obtained from the theoretical calculation.

Compound	НОМО	LUMO	band gap(eV)
Cz-MN	-6.16	-2.64	3.52
Cz-ECA	-5.89	-2.33	3.56
Cz-CA	-5.90	-2.22	3.68
Cz-CAA	-5.96	-2.46	3.50
Cz-NBA	-5.92	-2.70	3.22



Fig. S5. Self-reversible fluorescence switching cycle of (a) Cz-MN and (b) Cz-ECA.



Fig. S6. PXRD of Cz-ECA before and after crushing.



Fig. S7. Molecular confroamtion of **Cz-MN**, **Cz-ECA and Cz-CA** in the crystal lattice.

Table S2. Moelcular twist between the methoxy phenyl and carbazole in the crystal lattice. measured by torsion angle  $(\tau)$  in the



Compound	$\tau_1$	$ au_2$
Cz-MN	107.08	95.15
Cz-ECA	55.00	56.89
Cz-CA-a	54.11	51.07
Сz-СА-b	63.75	52.19



Fig. S8. Molecular packing in the crystal lattice of (a) **Cz-MN**, (b) **Cz-ECA** and (c) **Cz-CA**.